

1 **Electronic Supplementary Information**

2 **Ce-MOF@Polydopamine Composite Nanozyme Efficiently Scavenges Reactive**
3 **Oxygen Species and Iron in Thalassemia Disease**

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17 **1. Supplementary Methods**

18 **1.1 Materials and reagents**

19 No further purification was required for all chemical reagents. Cerium (IV) ammonium nitrate, 1,4-
20 benzenedicarboxylic acid (H₂BDC), hydroethidine (HE), sodium chloride (NaCl) and N,N-dimethylformamide
21 (DMF) were purchased from Beijing HWRK Chem Co., Ltd. Hydrogen peroxide (H₂O₂, 30 wt%), penicillin-
22 streptomycin, phosphate buffered saline (PBS), 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide
23 (MTT), fetal bovine serum and DMEM medium and 1-diphenyl-2-picrylhydrazyl (DPPH) were acquired from
24 Shanghai Macklin Biochemical Co., Ltd. Dextran/Fe was purchased Jiangxi Bolai Pharmacy Co., Ltd.

25 2',7'-dichlorofluorescein diacetate (DCFH-DA) was obtained from Beyotime Institute of Biotechnology. SOD
26 assay kit-WST and mouse ferritin enzyme-linked immunosorbent assay kit were obtained from Sigma-Aldrich.
27 Ethanol (EtOH) and acetone were acquired from Sinopharm Chemical Reagent Co., Ltd. RAW264.7 cells were
28 obtained from Stem Cell Bank, Chinese Academy of Sciences. The female Balb/C mice were acquired from
29 Hunan SJA laboratory animal Co., Ltd. Ultrapure water (Millipore 18.2 MΩ•cm) used throughout the whole
30 experiments.

31 **1.2. Instruments and characterization**

32 Scanning electron microscope (SEM) was performed with a Quanta 200 FEG SEM (Philips, Netherland).
33 Transmission electron microscopy (TEM) photos were obtained by a FEI Tecnai G2 20 with a field emission gun
34 operating at 200 kV. X-ray diffraction (XRD) patterns were characterized on a D/max 2550 VB/PC diffractometer
35 (Rigaku, Japan) using Cu K α radiation ($\lambda = 0.15418$ nm). Fourier transform infrared (FT-IR) was carried out on a
36 spectrometer (PE, USA). X-ray photoelectron spectroscopy (XPS) data were gotten from a Thermo ESCALAB
37 250XI electron spectrometer. The UV-Vis absorption spectra were recorded on the Cary 60 spectrophotometer
38 (Agilent, USA). The concentrations of Fe were detected by the inductively coupled plasma optical emission
39 spectrometer (ICP-OES) (Flexar/NexION300X, USA.). The electron paramagnetic resonance (EPR) were detected
40 by JES FA200(JEOL, Japan)).

41 **1.3. SOD mimetic activity assay**

42 The superoxide anion scavenging activity was assessed with a SOD assay kit (Sigma-Aldrich, USA). First, 20
43 μ L different nano concentration was mixed with 160 μ L of a 2-(4-iodophenyl)-3-(4-nitrophenyl)-5-(2,4-
44 disulfophenyl)-2H-tetrazolium sodium salt (WST-1) working solution. Then, 20 μ L of a xanthine oxidase solution
45 was added to each microplate well. After incubating at 37 °C for 20 min, the absorbance at 450 nm was measured
46 using a multiple plate reader (Victor X4). The inhibition rate of the superoxide was calculated by measuring the
47 reduction in color development.

48 **1.4. Catalase-like activity assay.**

49 Fluorescent method H₂O₂ can decompose into \cdot OH, which would react with terephthalic acid (TA) to produce
50 fluorescent 2-hydroxyterephthalic acid with an excitation wavelength of 320 nm and an emission peak at 425 nm.
51 In the presence of catalase (or catalase mimics), H₂O₂ would decompose into H₂O and O₂ and could not generate
52 fluorescent 2-hydroxyterephthalic acid. Therefore, by monitoring the fluorescent signal of 2-hydroxyterephthalic
53 acid, the elimination of H₂O₂ could be investigated. After 800 μ L phosphate buffer (10 mM, pH 7.2) containing
54 H₂O₂ (1 mM) and 100 μ L different concentrations of nanoparticle was vortexed vigorously and incubated at room
55 temperature for 5 h, 100 μ L TA (0.5 mM) was added. Reaction 1 h, then the fluorescence of the mixture was
56 measured.

57 **1.5. Elimination of iron assay(ICP)**

58 0.1 g Ce-MOF@PDA or 0.1 g Ce-MoF was added to 10 mL PBS solution (pH=7.2) containing 2 mM Fe³⁺

59 and incubated at room temperature for 11 h. Incubation at 37 °C for predetermined time, 100 µL of the supernatant
60 was taken out for analysis and 100 µL of fresh media was added into the well to keep a constant volume. The Fe³⁺
61 concentration in the supernatant was determined by using inductively coupled plasma (ICP) assay.

62 **1.6. In vitro cytotoxicity studies**

63 The in vitro cytotoxicity of nanoparticles was tested using 3-(4,5-dimethylthiazol-2-yl)-2,5-
64 diphenyltetrazolium bromide MTT assay. Specifically, RAW264.7 cells were seeded in 96-well plate and were
65 incubated for 24 h. Series dilutions of Ce-MOF@PDA and Ce-MOF ranging from 10 to 100 µg/mL were added
66 into each well and were cultured for another 24 hours. After this, 100 µL MTT solution was added to each well.
67 Incubating with cells for 4 h. MTT was removed, and 100 µL DMSO was added into each well using dissolving
68 the purple formazan crystal. The cell viability was measured through reading the absorbance at 570 nm and
69 samples were normalized to non-treated cells.

70 **1.7. In vitro ferritin reduction assay**

71 RAW264.7 cells were seeded in 6-well plates and were incubated for 24 h. The cells were treated with
72 culture medium containing 100 µM Ferric ammonium citrate (FAC) for 24 h to induce IO as described above.
73 Subsequently, cells were washed with PBS and treated with Ce-MOF@PDA and Ce-MOF at 10 and 100 µg/mL.
74 After 48 h incubation, cells were lysed with cell lysis buffer (150 mM NaCl, 10 mM Tris, 1% Triton X-100 and
75 protease inhibitor cocktail, pH 7.4). Total protein concentration was measured with the BCA protein assay kit
76 (Sigma-Aldrich, USA) and cellular ferritin concentration was measured with a mouse ferritin ELISA kit (Sigma-
77 Aldrich, USA). The results are plotted as the ratio of ng of ferritin per µg total protein concentration.

78 **1.8. Measuring iron-mediated oxidation stress levels in the presence of iron Ce-MOF@PDA.**

79 The DCFDA cellular ROS detection assay kit was used to determine the capability of Ce-MOF@PDA and Ce-
80 MOF to reduce iron-mediated oxidative stress in iron overload (IO) cells. Briefly, RAW264.7 cells were seeded in
81 96-well plates at a density of 10,000 cells/well incubated at 37 °C, 5% CO₂ and 100% humidity with DMEM
82 complete medium for 24 h prior to treating with culture medium containing 100 µM ferric ammonium citrate (FAC)
83 for 24 h to induce IO. Subsequently, cells were washed with PBS and treated with 50 µM DFO, 0.1 mg/mL Ce-
84 MOF@PDA and 0.1 mg/mL Ce-MOF for 8 h; Next, non-fluorescent DCFDA solution was added to each well at a
85 final concentration of 20 µM for 30 min at 37 °C. Cells were washed, and H₂O₂ was added to each well at a final
86 concentration of 50 µM; the fluorescence change was measured at indicated times using a SpectraMax Gemini EM
87 microplate reader by exciting at 485 nm and measuring emission at 535 nm at 37 °C. NIO cells that had not been
88 treated with materials but had been treated with DCFDA served as control blanks.

89 **1.9. In vivo elimination studies**

90 All the animal studies were approved by the Committee for Experimental Animals Welfare and School of
91 Chemistry and Pharmaceutical Science of Guangxi Normal University.. Female Balb/C mice, 6 weeks old, were
92 housed in cages in a room maintained at 20 ± 1 °C and with 12 h light and dark cycles. Feed and water were
93 available ad libitum. Mice were IO by a single tail vein injection of Dextran/Fe (150 mg/kg of Fe, 10 µL/g BW in
94 normal saline) on Day 1. On Day 8, mice were housed into each metabolic chamber and started on iron-deficient
95 powder diet. The following 5 treatments were administered intravenously to animals on Day 8 and every other day
96 for a total of 5 doses: Group 1=NIO control mice received saline; Group 2=IO mice received saline; Group 3=IO
97 mice received 150 mg/kg DFO; Group 4=IO mice received 500 mg/kg Ce-MOF@PDA; Group 5=IO mice
98 received 500 mg/kg Ce-MOF. Mice were monitored daily and bw were recorded on alternate days, and necropsied
99 7 days following administration of the last treatment dose.

100 To quantify elimination of iron from animals, feces and urine were collected daily from metabolic chambers and
101 weighed throughout the duration of the study. Fecal material was homogenized in distilled water, iron was
102 extracted from the homogenate by the addition of 5% Trichloroacetic acid and 1.5 M HCl, and the extract was

103 heated to 70 °C for 90 min. Final iron concentration in urine and feces was measured by standard Inductively
104 coupled plasma mass spectrometry (ICP-MS, Agilent, USA).

105 Mice were euthanized by CO₂ overdose and blood was directly collected via cardiac puncture and
106 added to microcentrifuge tubes to separate out the serum for ferritin measurements by ELISA assay based on the
107 manufacturer's instructions. The lungs, heart, spleen, kidneys, brain and liver of animals were subsequently
108 harvested, rinsed with fresh PBS, blotted dry with Whatman filter paper, and then weighed (note that organ weight
109 is reported as mg of total organ weight per g of animal BW, mg/g). Next, organs were sectioned into different
110 portions and snap-frozen in liquid nitrogen or fixed in neutral buffered formalin.

111 Determination of total iron content in liver: The liver tissue was homogenized, and was digested with mixed
112 acid (nitric acid:hydrogen peroxide:perchloric acid=1:1:3). After the white smoke disappeared, the concentration
113 of iron in the solution was determined by ICP.

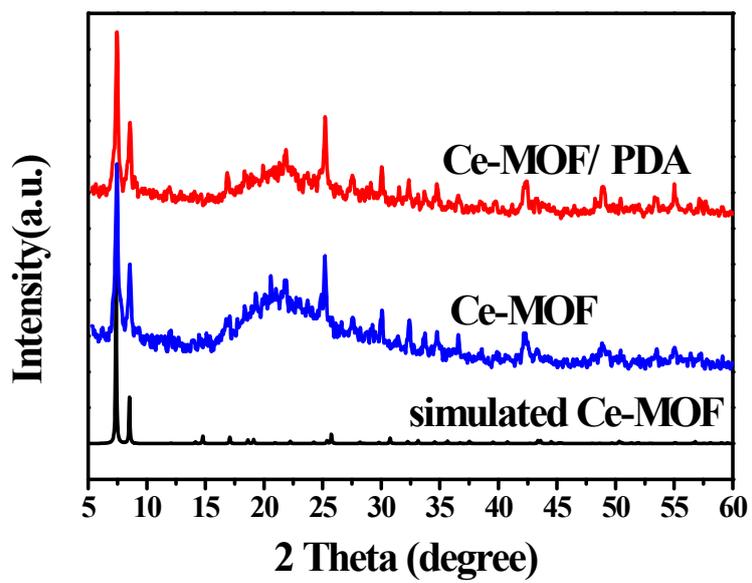
114 Determination of ROS content in liver: The liver tissue was homogenized, centrifuge the supernatant, adding
115 ROS detection probe DCF (the final concentration of DCF is 5 μM) and incubate at 37 °C for 1 hour in the dark,
116 and then use a microplate reader to detect fluorescence strength.

117 Determination of SOD activity in liver: The liver tissue was homogenized, centrifuge the supernatant, The
118 following steps are consistent with the previous method of detecting SOD mimetic activity simulation.

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121 2. Supplementary Figures and Table

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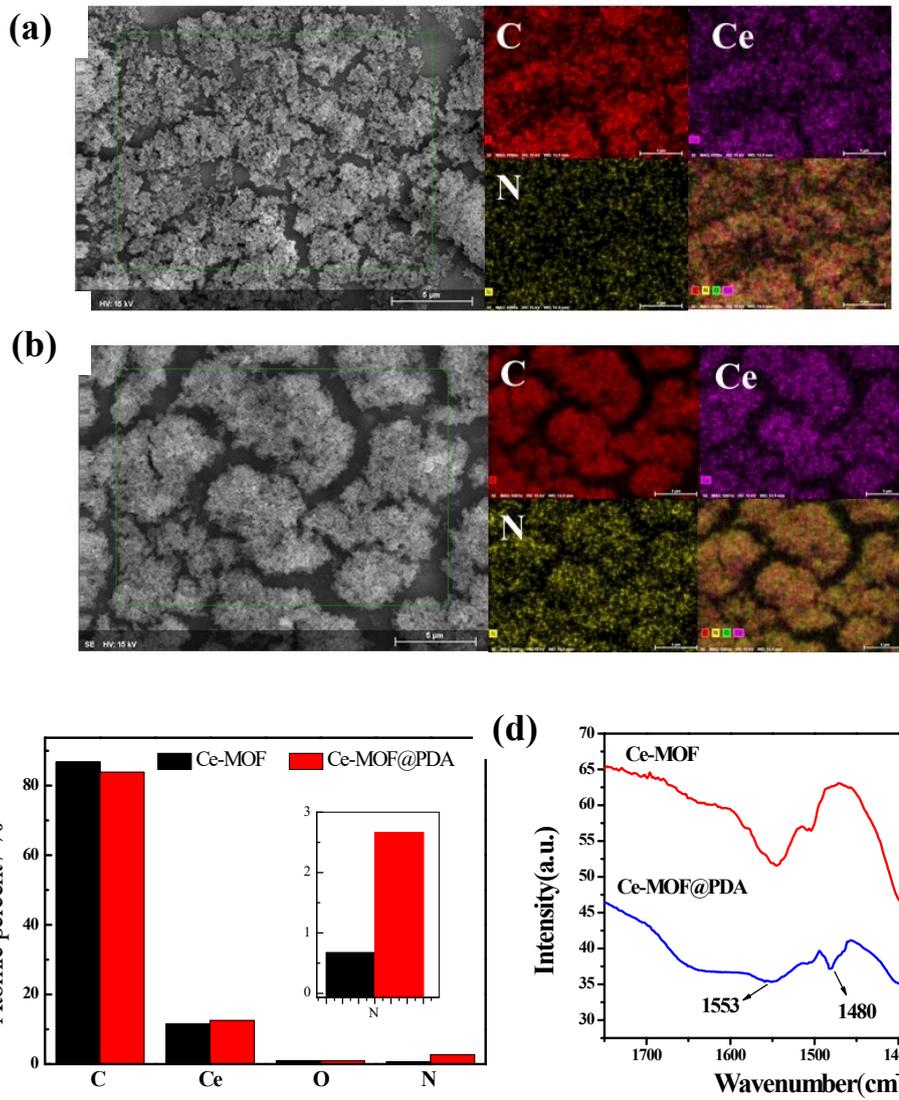


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125 **Fig. S1** XRD pattern of Ce-MOF and Ce-MOF@PDA

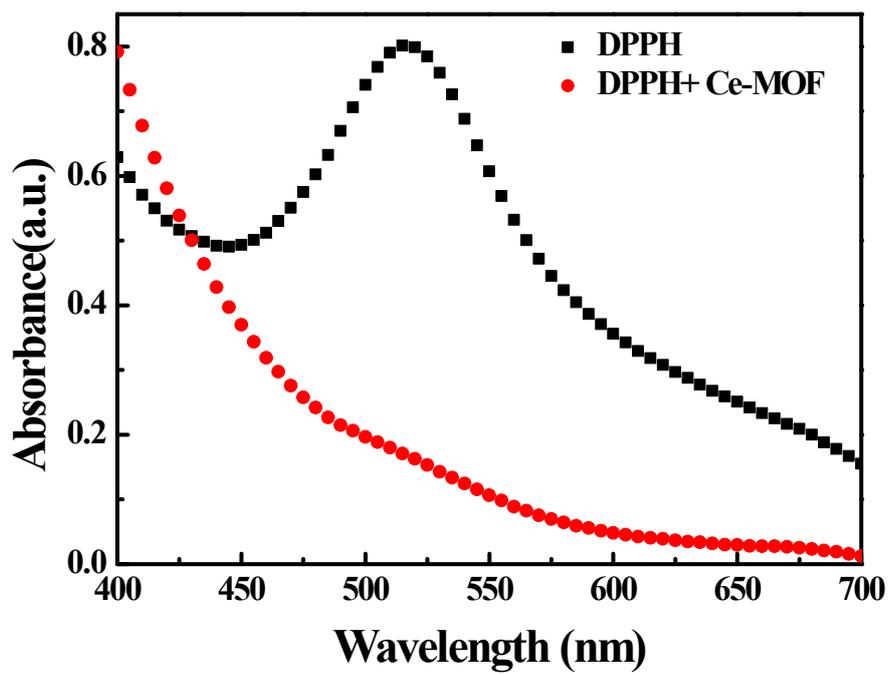
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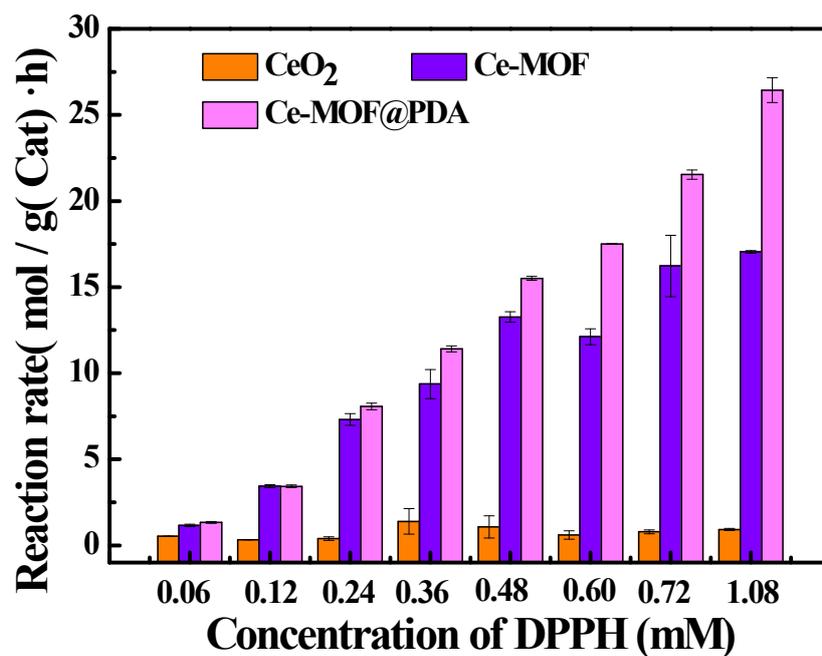
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133 **Fig. S2** Energy-dispersive X-ray spectroscopy (EDX) elements mapping of Ce, N and C of Ce-
134 MOF(a) and Ce-MOF@PDA(b); (c) Overlay map showing the distribution of C, Ce, O and N
135 elements in the Ce-MOF and Ce-MOF@PDA; (d) FT-IR spectra of Ce-MOF and Ce-MOF@PDA



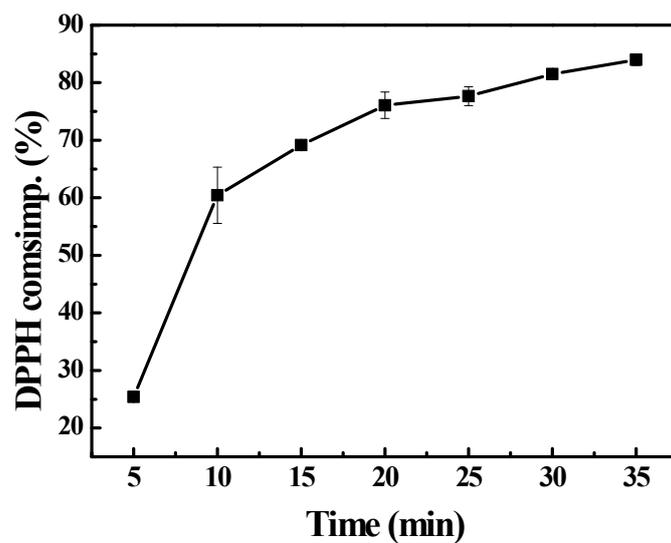
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138 **Fig. S3** The UV-Vis spectrogram of ethanolic DPPH solution and discolored after exposure (0.5
139 h) to the Ce-MOF.



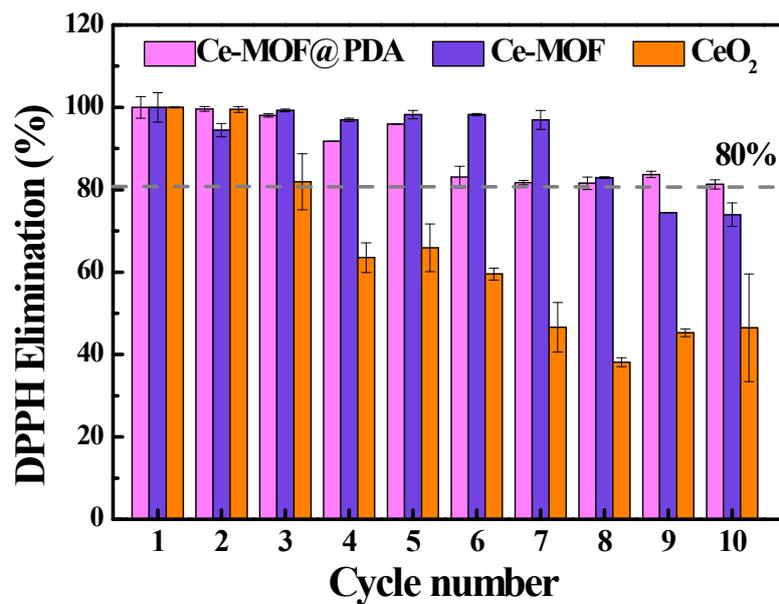
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141 **Fig. S4** Reaction rate of the elimination of DPPH and concentrations of DPPH (The material
 142 dosage is 50 $\mu\text{g/mL}$)



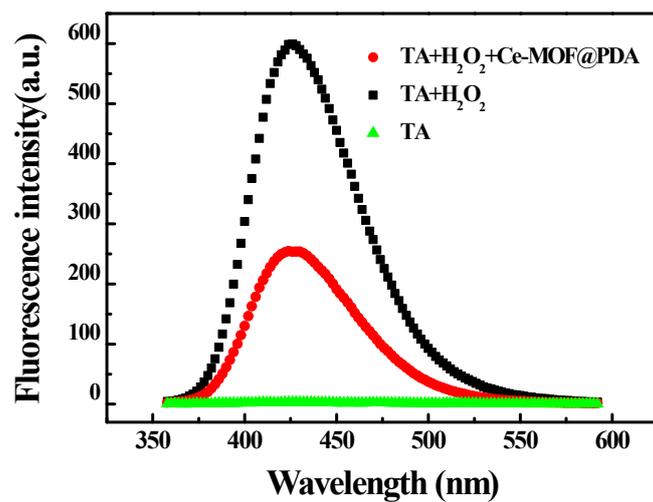
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144 **Fig. S5** Effect of time on DPPH elimination rate (concentrations of DPPH is 0.36 mM, the
145 material dosage is 50 $\mu\text{g/mL}$)



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147 **Fig. S6** Reuse performance Of CeO₂ NPs, Ce-MOF and Ce-MOF@PDA

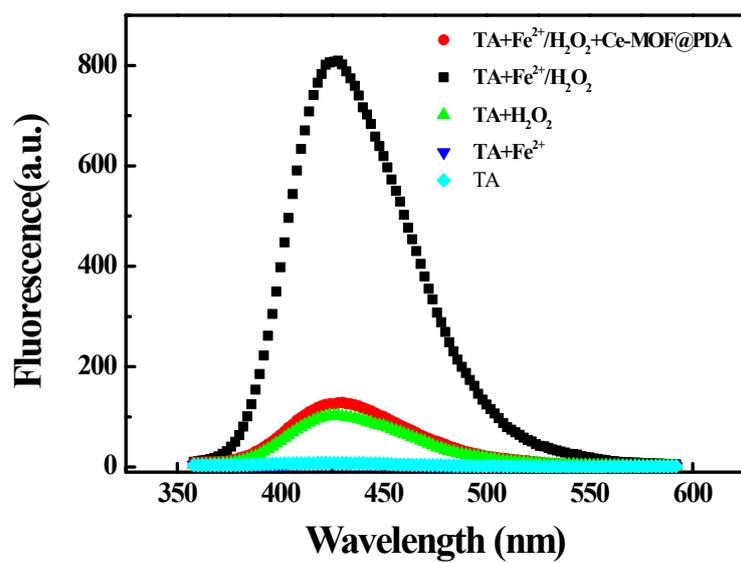


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150 **Fig. S7** Fluorescent spectra of TA after reaction with H₂O₂, in the absence and presence of Ce-
151 MOF@PDA.

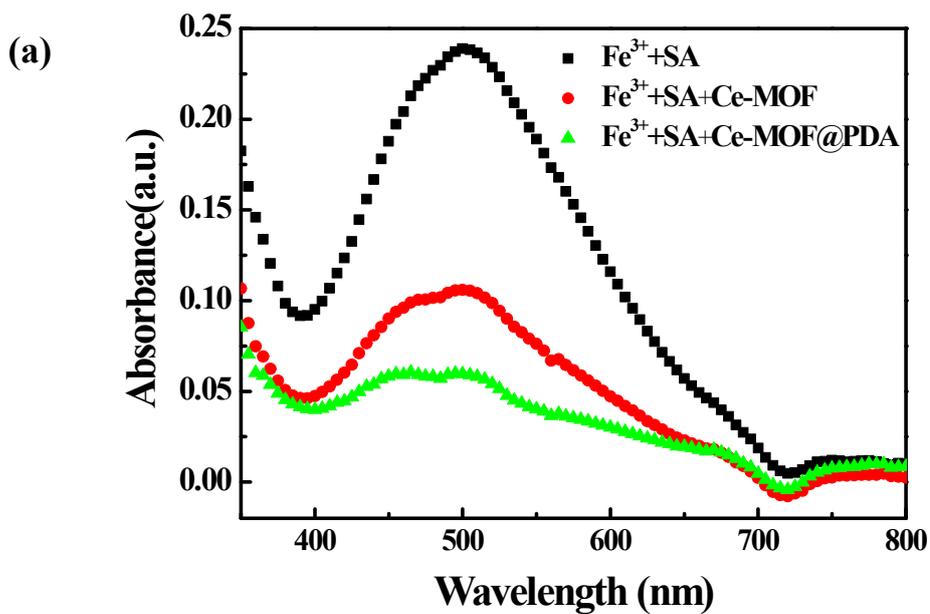
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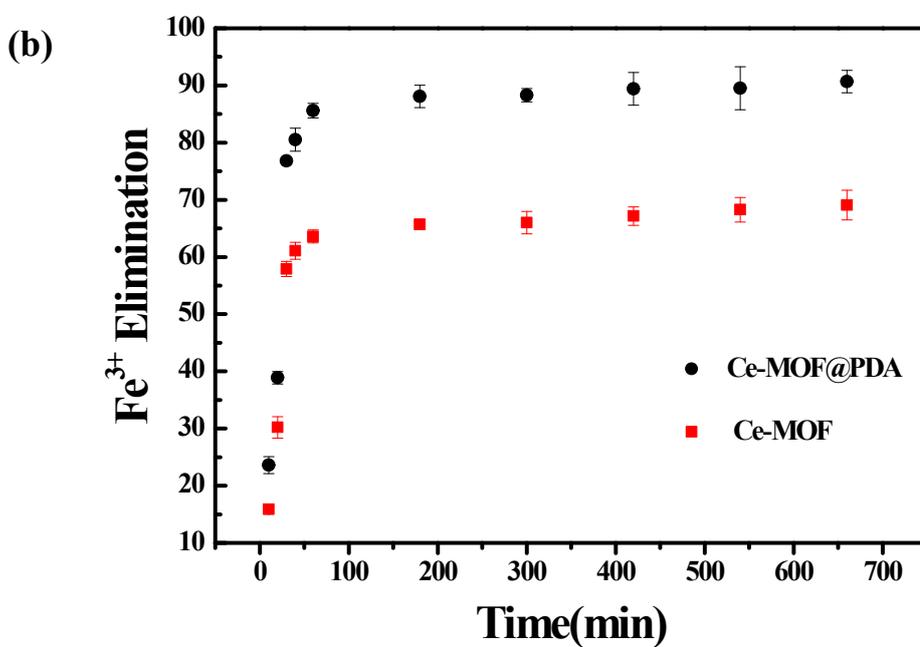


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156 **Fig. S8** Fluorescent spectra of TA after reaction with Fe²⁺/H₂O₂, in the absence and presence of
157 Ce-MOF@PDA. TA alone, and TA reacted with Fe²⁺ or H₂O₂ were used as control.



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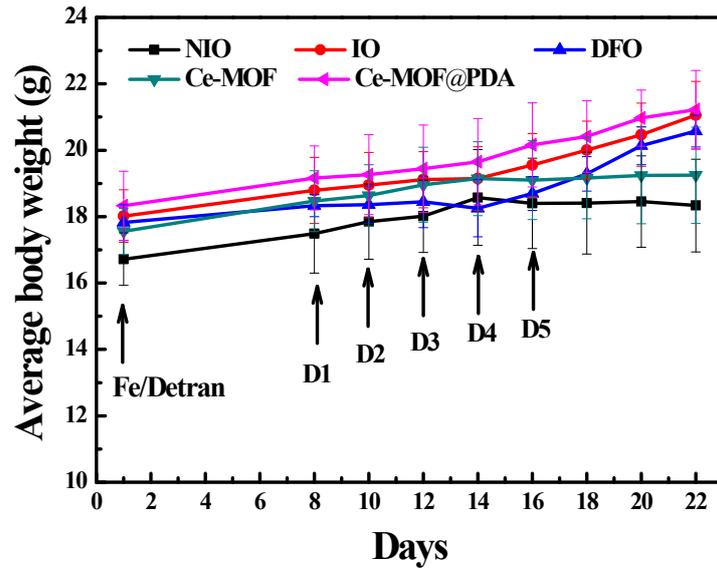


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161 **Fig. S9** (a) Absorption spectrum of SA after reaction with Fe^{3+} , in the absence and presence of
 162 Ce-MOF@PDA and Ce-MOF; (b) Effect of contact time on the elimination of Fe^{3+} onto Ce-
 163 MOF@PDA and Ce-MOF (2 mM Fe^{3+} , 100 $\mu\text{g}/\text{mL}$ Ce-MOF@PDA or Ce-MOF).

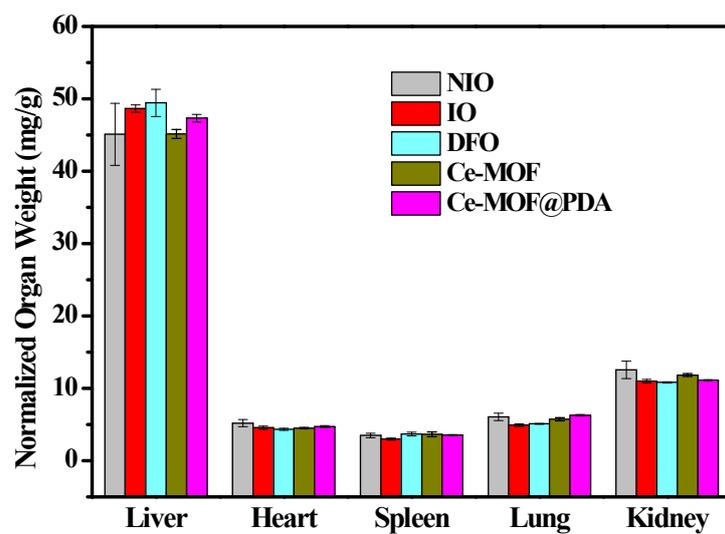
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168 **Fig. S10** BW of mice receiving Ce-MOF@PDA, Ce-MOF and DFO treatments



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171 **Fig. S11** The normalized organ weights with respect to animal body weight (mg/g)

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173 **Table S1** Summary comparing histological lesions in major organs after H&E staining, as
 174 assessed by a board certified pathologist.

	Liver	Spleen	Heart	Kidney	Lung
NIO	0	0	0	0	0
IO	++	0	0	0	0
DFO	++	0	0	0	0
Ce-MOF	++	0	0	0	0
Ce-MOF@PDA	+	0	0	0	0

175 Focal inflammatory focus around hemosiderin-laden cells: +; Multifocal inflammatory sites
 176 around hemosiderin-laden cells: ++; Numerous inflammatory foci around hemosiderin-laden cells:
 177 +++; Large number of inflammatory foci around hemosiderin-laden cells: ++++.