

## Supplemental Files

### Engineering the crystal facets of $\alpha$ -MnO<sub>2</sub> nanorods for electrochemical energy storage: experiment and theory

Yifan Wang<sup>a, b, #</sup>, Zhengwei Lu<sup>a, b, #</sup>, Peipei Wen<sup>a, b</sup>, Yinyan Gong<sup>a, b</sup>,

Can Li<sup>a, b</sup>, Lengyuan Niu<sup>\* a, b</sup>, Shiqing Xu<sup>a, b</sup>

<sup>a</sup> Key Laboratory of Rare Earth Optoelectronic Materials and Devices of Zhejiang Province,  
China Jiliang University, Hangzhou 310020, Zhejiang, China

<sup>b</sup> College of Optical and Electronic Technology, China Jiliang University, Hangzhou 310020,  
Zhejiang, China

---

<sup>#</sup> These authors contributed equally to this work.

<sup>\*</sup> Corresponding authors, E-mail: [niulengyuan@163.com](mailto:niulengyuan@163.com)  
Fax: +86-571-86872363, Tel.: +86-571-86872363.

## 1. The EDS results of $\alpha$ -MnO<sub>2</sub>

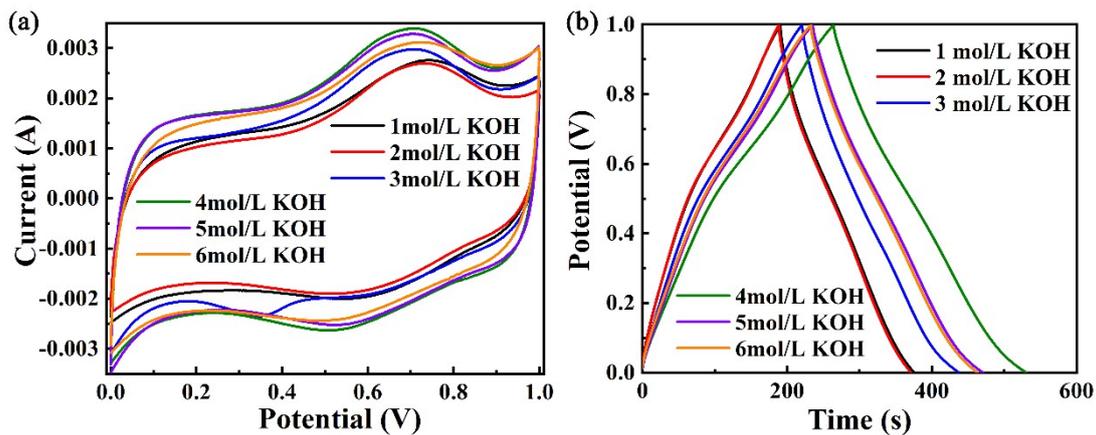
The EDS was used to analyze the elemental composition of the as-prepared samples. The weight percent of Mn, O and K (deduct the C element) is shown in **Table S1**. The Mn and O content is nearly close, which is in accordance with the XPS results and further indicates the successful preparation of MnO<sub>2</sub>.

**Table S1** The weight percent (%) of Mn, O and K for  $\alpha$ -MnO<sub>2</sub> based on EDS results.

Samples	Mn	O	K
$\alpha$ -MnO <sub>2</sub> -200	68.85	24.94	6.14
$\alpha$ -MnO <sub>2</sub> -110	71.84	22.23	5.92
$\alpha$ -MnO <sub>2</sub> -310	69.86	21.82	8.32

## 2. The effect of KOH concentrations on the electrochemical performance of $\alpha$ -MnO<sub>2</sub>-310

KOH was applied as an alkaline etchant to improve the electrochemical performance of  $\alpha$ -MnO<sub>2</sub>-310. **Fig.S1a** shows the CV curves of as-prepared  $\alpha$ -MnO<sub>2</sub>-310 at the scan rate of 10 mV/s. Compared to  $\alpha$ -MnO<sub>2</sub>-310 treated with other concentrations of KOH, the  $\alpha$ -MnO<sub>2</sub>-310 after 4 M KOH treatment displays the larger area and broader region. The GCD curves of  $\alpha$ -MnO<sub>2</sub>-310 with different concentrations of KOH at 0.5 A/g are depicted in **Fig. S1b**. Among, the  $\alpha$ -MnO<sub>2</sub>-310 treated with 4 M KOH displays the highest specific capacitance of 133 F/g, which is better than their counterparts in terms of specific capacitance values. Thus, we finally choose 4 M as the optimum concentration for the treatment of  $\alpha$ -MnO<sub>2</sub>-310.



**Fig. S1** Electrochemical properties of as-prepared  $\alpha$ -MnO<sub>2</sub>-310 treated with different concentrations of KOH: (a)

CV curves at a scan rate of 10 mV/s, (b) GCD curves at 0.5 A/g.

### 3. The electrochemical performances in the two-electrode case

In order to further explore the practical application of as-prepared  $\alpha$ -MnO<sub>2</sub>, the electrochemical performances in the two-electrode case were evaluated, while using the  $\alpha$ -MnO<sub>2</sub>-310 as the positive electrode, the activated carbon (AC) as the negative electrode, and 0.5 M Na<sub>2</sub>SO<sub>4</sub> as electrolyte. **Fig.S2a** shows the CV curves of MnO<sub>2</sub>//AC asymmetric supercapacitor at different scan rates. It can be seen that all the CV curves exhibit a quasi-rectangular shaped capacitive behavior, which indicate the good reversibility and pseudocapacitive behavior. Meanwhile, the GCD curves with the current densities from 0.5 A g<sup>-1</sup> to 5 A g<sup>-1</sup> (**Fig.S2b**) show a symmetrical triangle shape, indicating that the assembled asymmetric supercapacitor has excellent reversibility and high coulombic efficiency. The calculated specific capacitance was 29.17 F/g at a current density of 0.5 A g<sup>-1</sup>. Besides, the capacity retention ratio was 47.14% when the current density increasing to 5 A g<sup>-1</sup> (**Fig.S2c**). The energy density can reach 5.83 Wh kg<sup>-1</sup> at a power density of 300 W kg<sup>-1</sup>, which is better than the commercial AC//AC symmetric supercapacitors (**Fig.S2d**).<sup>1-3</sup>

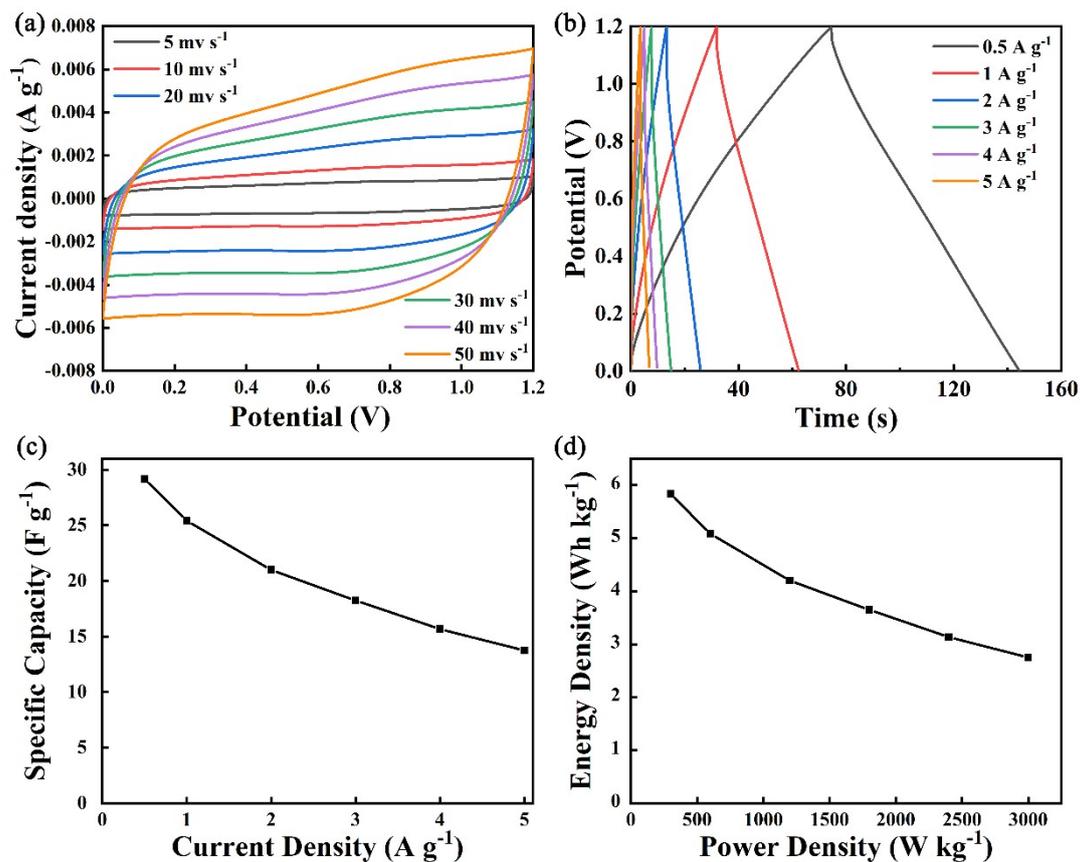


Fig. S2 Electrochemical properties of MnO<sub>2</sub>//AC asymmetric supercapacitor: (a) CV curves at different scan rates, (b) GCD curves at different current densities, (c) the calculated specific capacity at various current densities and (d)

Ragone plots.

## References

1. M. Wang, J. Yang, K. Jia, S. Liu, C. Hu and J. Qiu, *Chemistry – A European Journal*, 2020, 26, 2897-2903.
2. Z. Fang, L. Cao, F. Lai, D. Kong, X. Du, H. Lin, Z. Lin, P. Zhang and W. Li, *Composites Science and Technology*, 2019, 183.
3. J. Hao, X. Wang, Y. Wang, X. Lai, Q. Guo, J. Zhao, Y. Yang and Y. Li, *Nanoscale Advances*, 2020, 2, 878-887.