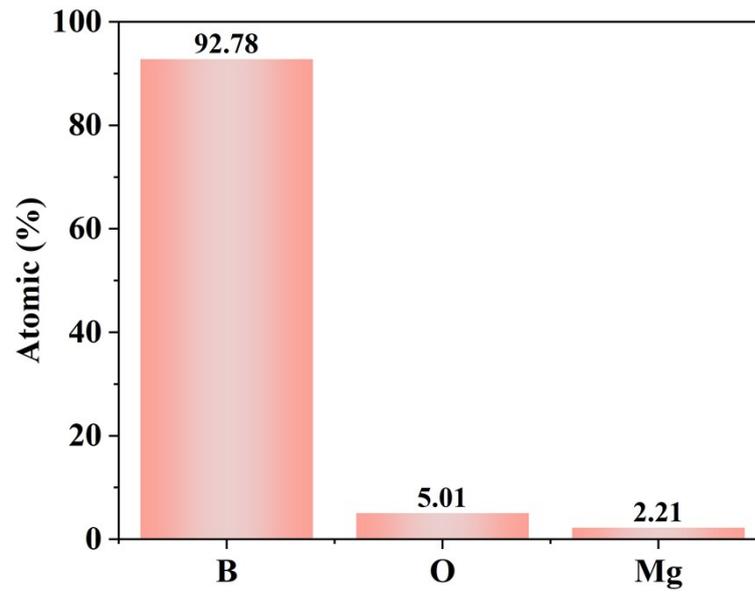
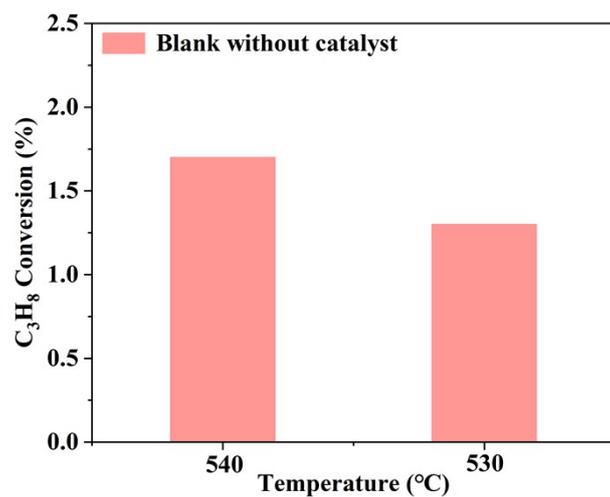


**Figure S1.** (a) SEM image of Mg-BNSs. (b-d) EDX mapping of Mg-BNSs.

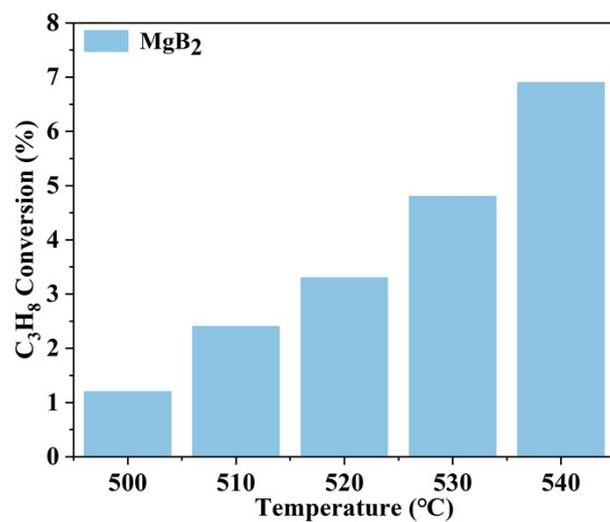
Scale bars, 200 nm.



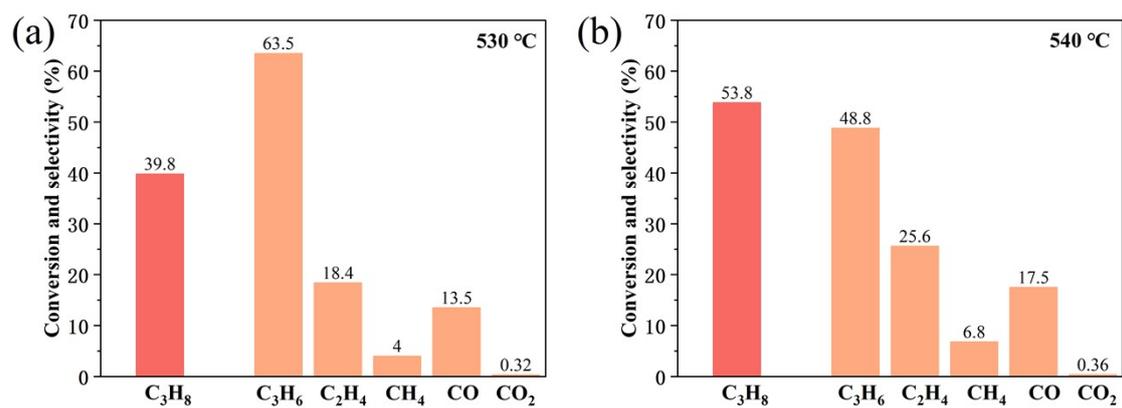
**Figure S2.** EDX analysis results of Mg-BNSs.



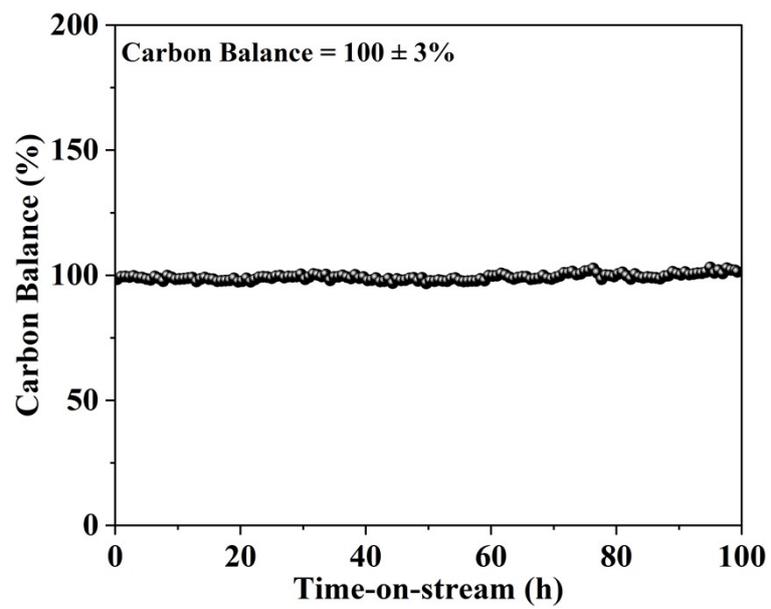
**Figure S3.** The blank without catalyst test for ODHP. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 :1 :3, WHSV = 24000 ml/g/h.



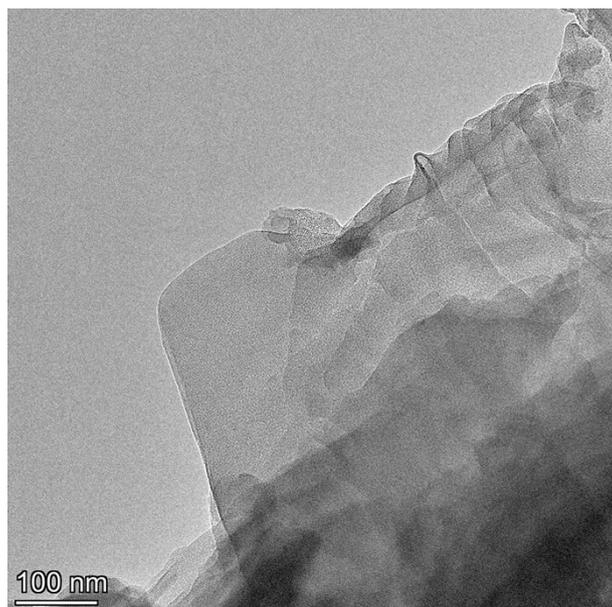
**Figure S4.** Catalytic performance of MgB<sub>2</sub> at different temperatures. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 : 1 : 3, WHSV = 24000 ml/g/h.



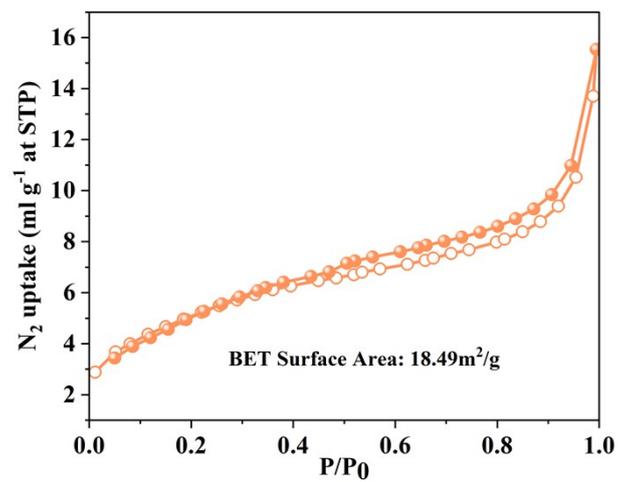
**Figure S5.** Propane conversions and product selectivities of Mg-BNSs at 530 °C and 540 °C. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 : 1 : 3, WHSV = 24000 ml/g/h.



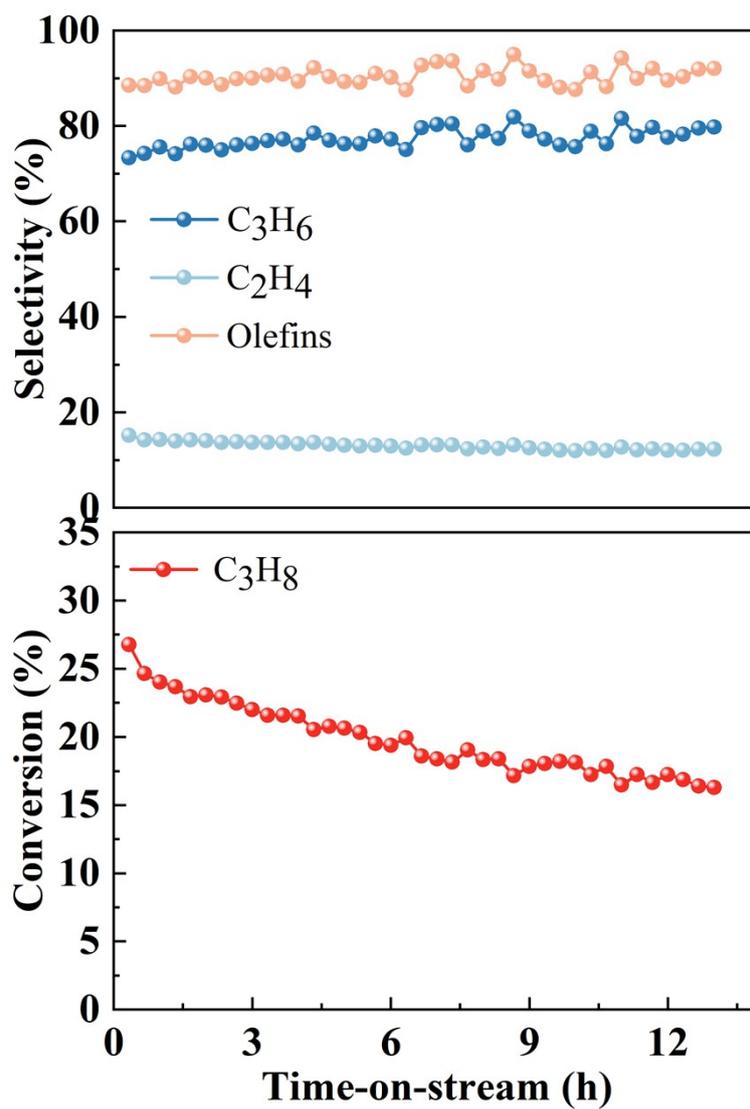
**Figure S6.** Carbon balance of Mg-BNSs during the stability test at 530 °C.



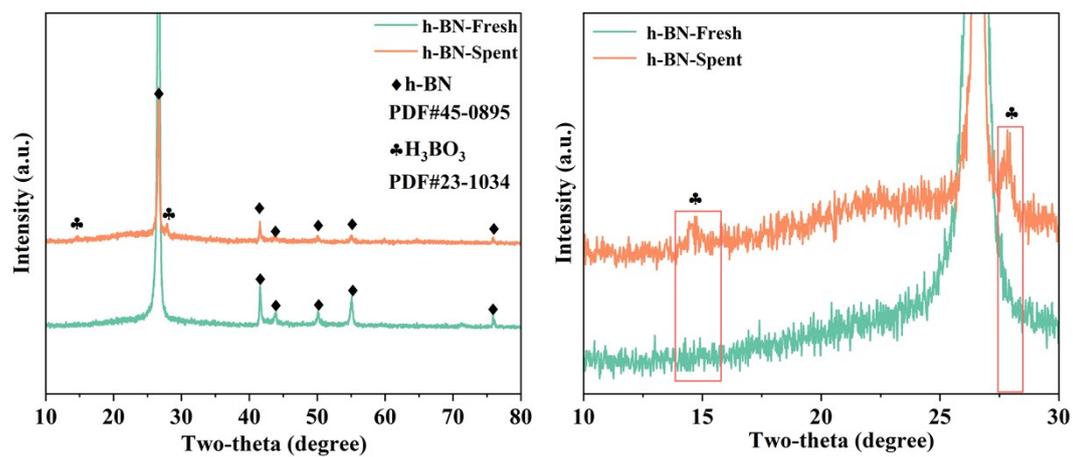
**Figure S7.** TEM image of Mg-BNSs after the catalytic test for 100 h.



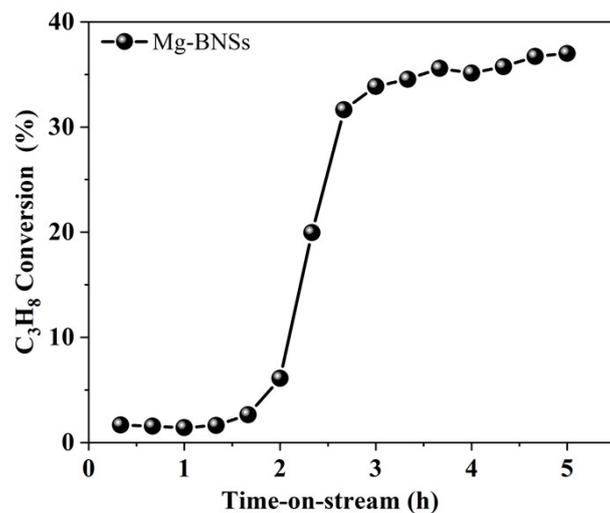
**Figure S8.** N<sub>2</sub> sorption isotherms of Mg-BNSs after the catalytic test for 100 h.



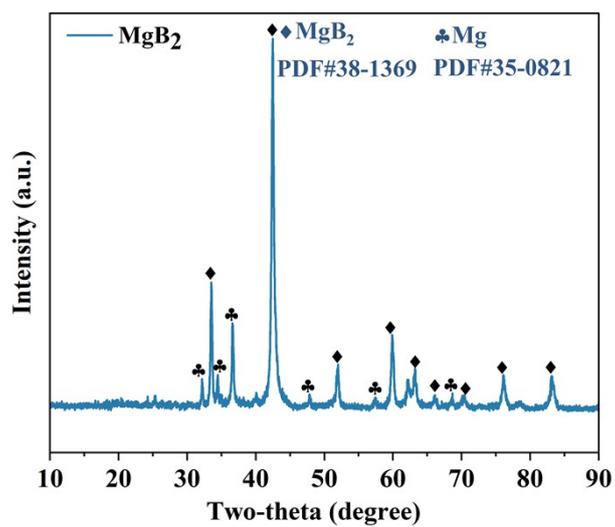
**Figure S9.** Long-term stability test of h-BN at 530 °C. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 :1 :3, WHSV = 24000 ml/g/h.



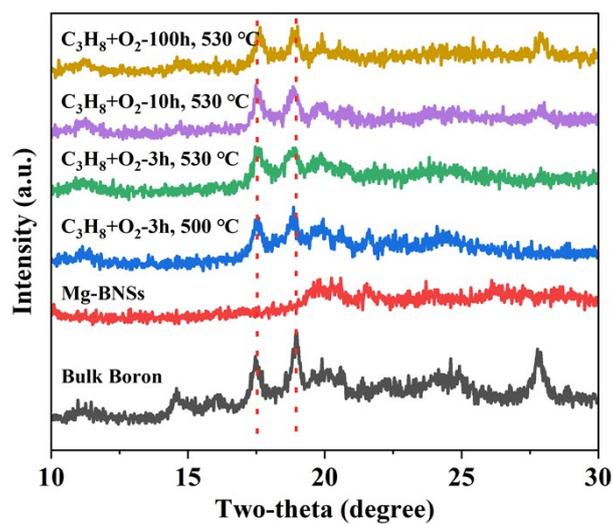
**Figure S10.** XRD patterns of the fresh h-BN and the spent h-BN.



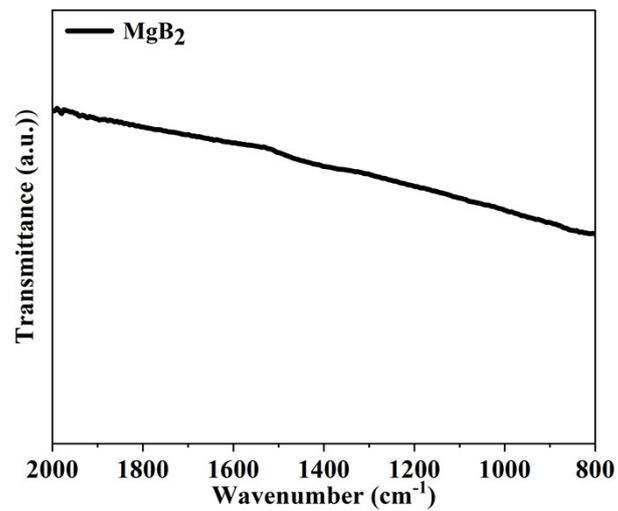
**Figure S11.** Propane conversion as a function of time on stream on the fresh Mg-BNSs at 530 °C. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 :1 :3, WHSV = 24000 ml/g/h.



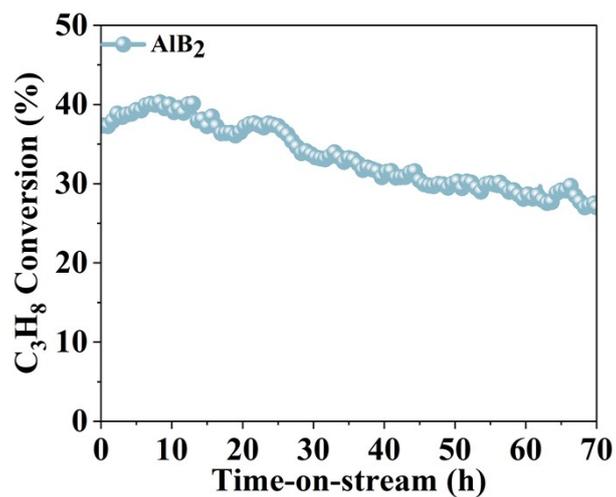
**Figure S12.** XRD pattern of commercial MgB<sub>2</sub>.



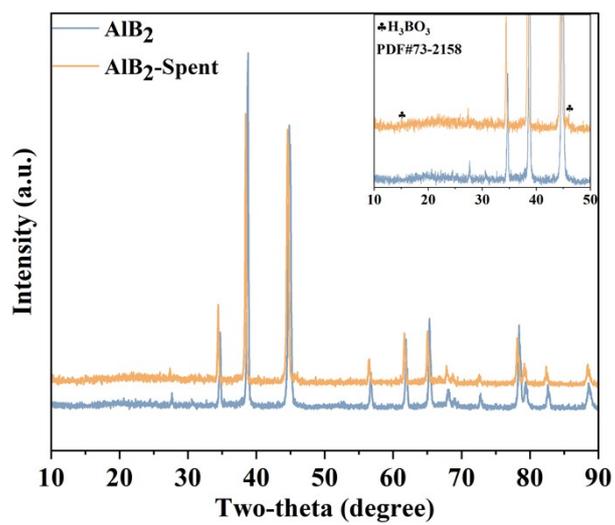
**Figure S13.** Selected regions of the XRD patterns of bulk boron, Mg-BNSs and their spent forms.



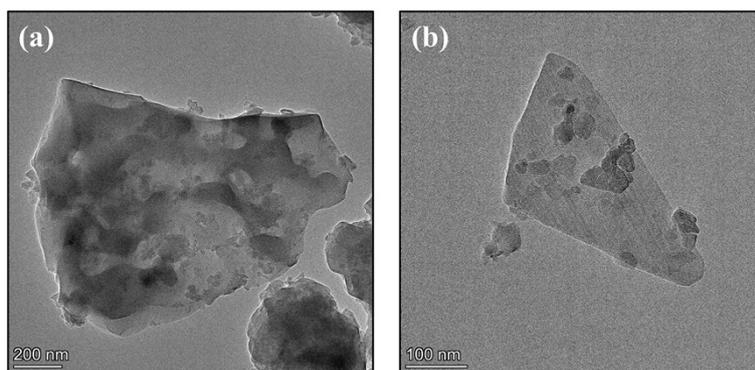
**Figure S14.** FT-IR spectrum of MgB<sub>2</sub>.



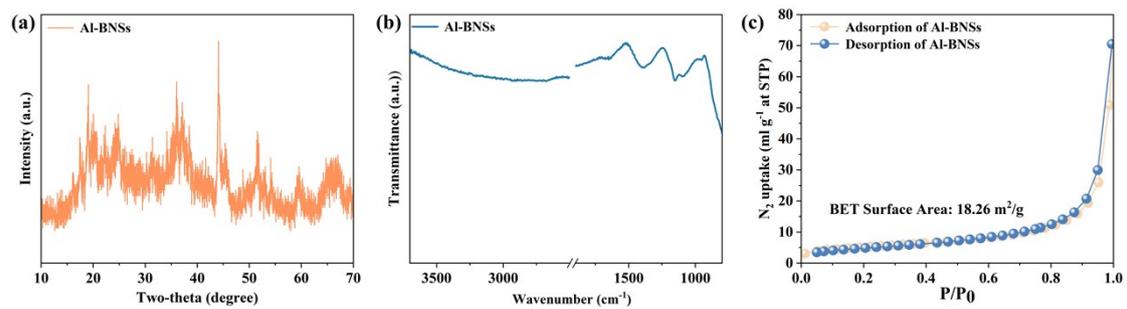
**Figure S15.** Propane conversion as a function of time on stream on the pristine AlB<sub>2</sub> at 530 °C. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 :1 :3, WHSV = 24000 ml/g/h.



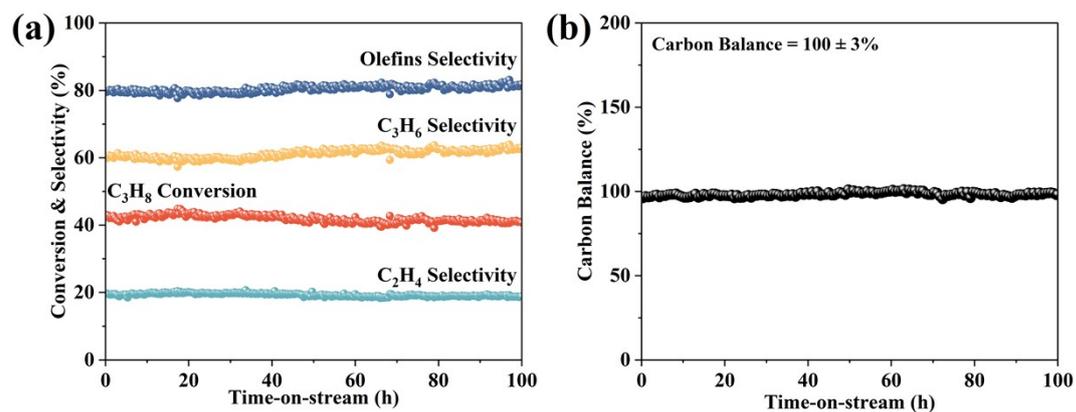
**Figure S16.** XRD pattern of AlB<sub>2</sub> and the spent AlB<sub>2</sub>.



**Figure S17.** TEM images of (a) AlB<sub>2</sub> and (b) Al-BNSs.



**Figure S18.** (a) XRD pattern, (b) FT-IR spectrum and (c) N<sub>2</sub> adsorption and desorption studies of Al-BNSs.



**Figure S19.** (a) Stability test of Al-BNSs at 530 °C over 100 h. (b) Carbon balance of Al-BNSs during the stability test at 530 °C. Reaction conditions: atmospheric pressure, C<sub>3</sub>H<sub>8</sub>/O<sub>2</sub>/N<sub>2</sub> ratio = 1 : 1 : 3, WHSV = 24000 ml/g/h.

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**Table S1.** ICP-OES analysis results of Mg-BNSs.

<b>Sample</b>	<b>ICP-AES value (mg/L)</b>		<b>Stoichiometric ratio</b>
	<b>Mg</b>	<b>B</b>	<b>Mg:B</b>
<b>MgB<sub>2</sub>-HCl</b>	<b>1.16</b>	<b>8.57</b>	<b>Mg<sub>0.12</sub>B<sub>2</sub></b>

Note: the stoichiometric ratio was calculated by using the following formula:

$$\frac{\text{Elemental concentration of Mg}}{\text{Atomic weight of Mg (24.3)}} : \frac{\text{Elemental concentration of B}}{\text{Atomic weight of B (10.8)}}$$

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**Table S2.** Elements compositions of MgB<sub>2</sub>, Mg-BNSs, and the spent Mg-BNSs based on XPS analysis results.

<b>Samples</b>	<b>B (at.%)</b>	<b>Mg (at.%)</b>	<b>O (at.%)</b>
MgB <sub>2</sub>	17.81	16.73	65.46
Mg-BNSs-Fresh	80.57	1.61	17.82
Mg-BNSs-Spent	42.64	2.40	54.96

**Table S3.** Catalytic performance of Mg-BNSs in comparison with the reported boron-based catalysts in ODHP.

No.	Catalysts	Temp. [°C]	Conv. [%]	Selectivity [%]			Productivity [ $\frac{\text{g}_{\text{olefin}}}{\text{g}_{\text{cat}} \cdot \text{h}^{-1}}$ ]	Ref.
				C <sub>3</sub> <sup>=</sup>	C <sub>2</sub> <sup>=</sup>	C <sub>2,3</sub> <sup>=</sup>		
1	WB	500	2.5	87.9	7.3	95.2	0.13	1
2	NiB	500	6.1	85.4	9.3	94.7	0.40	1
3	Ti <sub>2</sub> B	500	5.8	85.4	9.1	94.5	0.50	1
4	B <sub>4</sub> C	500	7.0	84.2	9.3	93.5	0.60	1
5	h-BN	490	14	79.0	12.0	91.0	0.50	2
6	High surface area BN	525	24	69.0	-	-	0.04	3
7	BS-1	540	23.8	55.4	27.2	82.6	0.12	4
8	B <sub>2</sub> O <sub>3</sub> /Al <sub>2</sub> O <sub>3</sub>	550	24.1	42.6	12.5	55.1	0.35	5
9	BOS-10	450	14.8	73.3	14.1	87.4	1.1	6
10	B <sub>2</sub> O <sub>3</sub> @BPO <sub>4</sub>	550	24.7	66.4	18.4	84.8	0.79	7
11	SiB <sub>6</sub>	535	19.2	82.2	12.2	94.4	1.49	8
12	B-MWW	540	29.9	72.5	15.3	87.8	1.11	9
13	Mg-BNSs	530	39.8	63.5	18.4	81.9	2.48	This work
14	Mg-BNSs	540	53.8	48.8	25.6	74.4	2.91	This work

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**Table S4.** The prices of MgB<sub>2</sub> and AlB<sub>2</sub> from different resources.

Samples	Yield of boron nanosheet (wt%)	Cost (CNY/g)
MgB <sub>2</sub>	27.0	1.2
AlB <sub>2</sub>	24.3	9.5

Note: The price was calculated based on the price of MgB<sub>2</sub> from RHAWN (30.73 CNY/25g) as of when this work was submitted. And the price of AlB<sub>2</sub> from Acmec (47.55 CNY/5g) as of when this work was submitted.

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