

Electronic Supplementary Information

First-principles study of the oxidation susceptibility of WS₂, WSe₂, and WTe₂ monolayers

Ashima Rawat,^{1*} Lokanath Patra², Ravindra Pandey¹ and Shashi P. Karna^{3*‡}

¹*Department of Physics, Michigan Technological University, Houghton, MI, 49931, USA*

²*Department of Mechanical Engineering, University of California, Santa Barbara, Santa Barbara, CA 93106-5070, USA.*

³*DEVCOM Army Research Laboratory, Weapons, and Materials Research Directorate, ATTN: FCDD-RLW, Aberdeen Proving Ground, Aberdeen, Maryland 21005-5069, USA*

Corresponding Authors: arawat2@mtu.edu, shashi.p.karna.civ@army.mil

1. Pristine WX₂ monolayers

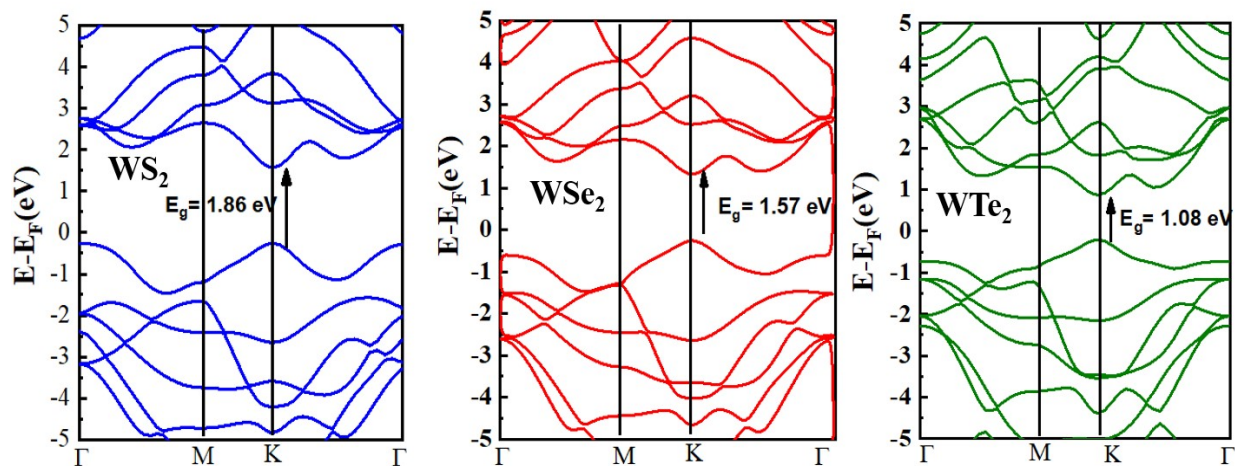


Figure S1. Band Structure of pristine WX₂ monolayers

2. O₂ and O₃ molecules

Table S1 Bond Lengths and Bond Angles for the O₂ and O₃ molecules

Molecule	R _{O-O} (O ₂)	R _{O-O} (O ₃)	A _{O-O-O} (O ₃)
Bond-Length (R _{O-O})/ Bond Angle (A _{O-O-O})	1.24 Å	1.28 Å	118.15°
	1.24 Å (DFT) ⁴¹ 1.21 Å (expt.) ⁴²	1.28 Å (expt.) ⁴³ 1.28 Å (DFT) ⁴⁴	116.49° (expt.) ⁴³ 117.90° (DFT) ⁴⁴

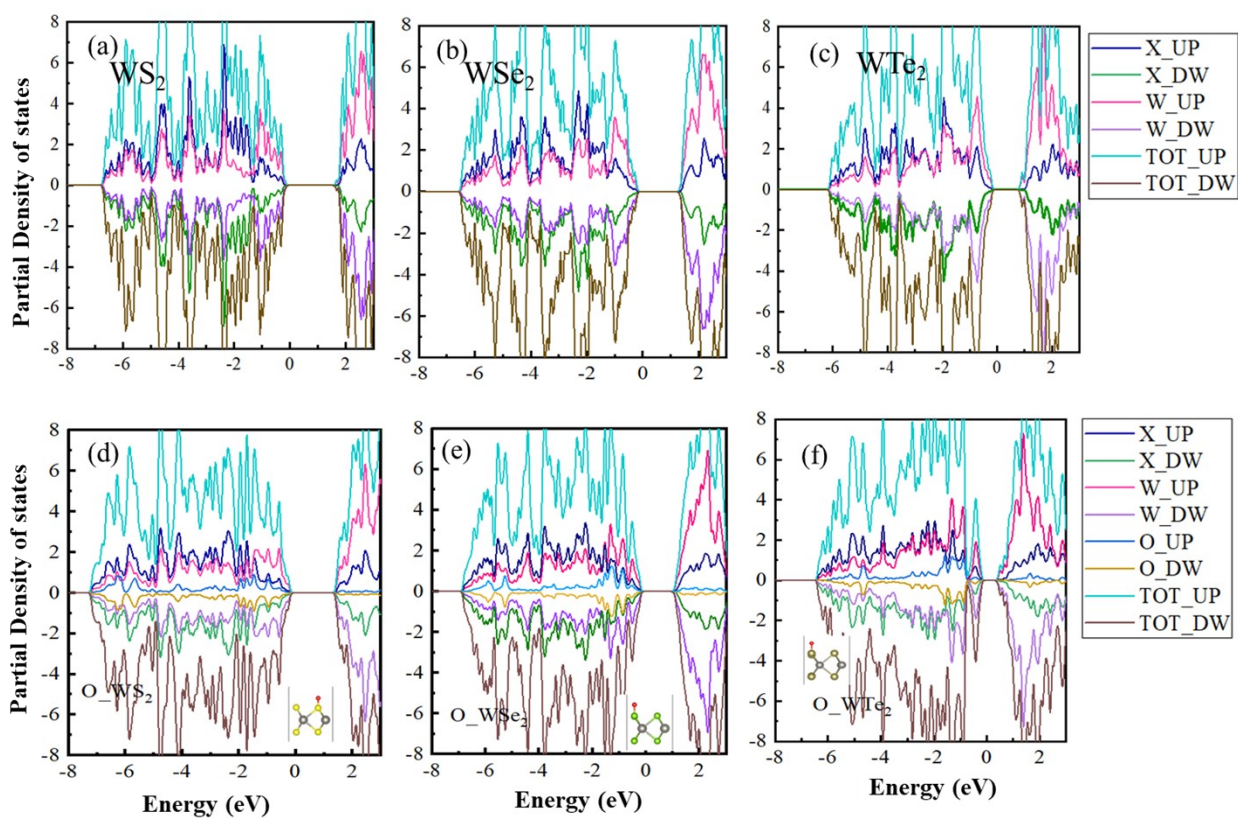


Figure S2. Calculated Total DOS and PDOS for (a)-(c) pristine and (d)-(f) oxygen adsorbed WX_2 monolayers.