

Electronic supporting information

Application of hindered ether solvents for palladium catalyzed Suzuki-Miyaura, Sonogashira and cascade Sonogashira cross-coupling reactions

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Table S1. Boiling point (BP), KAT parameters and Hansen solubility parameters of hindered ether solvents and traditional solvents.

| Solvent | BP /°C | α | β | π^* | δ_D /MPa ^{1/2} | δ_P /MPa ^{1/2} | δ_H /MPa ^{1/2} |
|-------------------------|--------|----------|---------|---------|--------------------------------|--------------------------------|--------------------------------|
| TMO ¹ | 112 | 0.00 | 0.77 | 0.35 | 15.4 | 2.4 | 2.1 |
| DEDMO ² | 162 | 0.00 | 0.62 | 0.32 | 16.2 | 3.1 | 2.2 |
| Eucalyptol ³ | 172 | 0.00 | 0.61 | 0.36 | 16.6 | 2.5 | 2.5 |
| THF ¹ | 66 | 0.00 | 0.58 | 0.59 | 16.8 | 5.7 | 8.0 |
| Toluene ¹ | 111 | 0.00 | 0.10 | 0.51 | 18.0 | 1.4 | 2.0 |

Key: BP, boiling point; α , hydrogen bond donating ability; β , hydrogen bond accepting ability; π^* , dipolarity; δ_D , Hansen dispersion force solubility parameter; δ_P , Hansen polarity solubility parameter; δ_H , Hansen hydrogen bonding solubility parameter.

Table S2. Yields of Sonogashira reaction using halobenzene performed in TMO.

| Entry | X | R | Compound | Yields ^a |
|-------|----|-------------------|----------|---------------------|
| 1 | Cl | H | 6a, 6'a | 7% |
| 2 | Br | 2-OMe | 6'h | trace |
| 3 | Br | 2-NO ₂ | 6'i | trace |
| 4 | I | 2-NH ₂ | 6'j | trace |
| 5 | Br | 4-OMe | 6b, 6'b | 78% |
| 6 | I | 4-CH ₃ | 6'k | 73% |

Table S3. Optimization of Sonogashira reaction using iodobenzene and phenylacetylene.

| Entry | Catalyst (mol%) | Base | Yield ^b |
|-------|----------------------------|---------------------------------|--------------------|
| 1 | Pd(dba) ₂ (10%) | Cs ₂ CO ₃ | 20% |
| 2 | Pd(OAc) ₂ (10%) | Cs ₂ CO ₃ | 25% |
| 3 | PdCl ₂ (10%) | Cs ₂ CO ₃ | 42% |
| 4 | PdCl ₂ (10%) | Cs ₂ CO ₃ | 34% ^c |
| 5 | PdCl ₂ (20%) | Cs ₂ CO ₃ | 56% |
| 6 | PdCl ₂ (10%) | KO <i>t</i> -Bu | 91% |
| 7 | PdCl ₂ (5%) | KO <i>t</i> -Bu | 98% |
| 8 | PdCl ₂ (2.5%) | KO <i>t</i> -Bu | 99% |
| 9 | PdCl ₂ (1%) | KO <i>t</i> -Bu | 98% |

^a Iodobenzene = 1 mmol, Phenylacetylene = 1.5 mmol, base = 2 mmol, PPh₃ = 0.2 mmol, 100°C, 24 h, 3 mL of TMO as a solvent, ^b Isolated yield and ^c 10 h.

Table S4. Optimization of cascade Sonogashira reaction using iodobenzene and phenylacetylene.

| Entry | Catalyst (mol%), PPh ₃ (mol%) | Time (h) | Temp. (°C) | Yield ^b |
|-------|---|----------|------------|--------------------|
| 1 | PdCl ₂ (10%), PPh ₃ (20%) | 2 | 100 | 3% |

| | | | | |
|---|---|----|-----|-----|
| 2 | PdCl ₂ (10%), PPh ₃ (20%) | 5 | 100 | 8% |
| 3 | PdCl ₂ (10%), PPh ₃ (20%) | 24 | 100 | 52% |
| 4 | PdCl ₂ (10%), PPh ₃ (20%) | 24 | 110 | 43% |

^a 2-Iodophenol = 1 mmol, Phenylacetylene = 1.5 mmol, KO^t-Bu = 2 mmol, 3 mL of TMO as a solvent.

^b Isolated yield.

S5. Materials used for green metric calculation

Table S5. Materials used for green metric calculation.^{2,4,5}

| Chemicals | Suzuki reaction | | | | | Sonogashira reaction | | | | |
|--|---|--------|------------|--------|--------|---|--------|------------|--------|--------|
| | TMO | DEDMO | Eucalyptol | THF | Tol | TMO | DEDMO | Eucalyptol | THF | Tol |
| Solvent | 1.60 | 1.67 | 1.85 | 1.77 | 1.73 | 2.41 | 2.51 | 2.78 | 2.65 | 2.60 |
| Product | 0.1372 | 0.1326 | 0.1326 | 0.1311 | 0.1357 | 0.1622 | 0.1444 | 0.1355 | 0.1515 | 0.1426 |
| Iodobenzene | 1.00 mmol x 204.01 g mol ⁻¹ = 0.2040 g | | | | | 1.00 mmol x 204.01 g mol ⁻¹ = 0.2040 g | | | | |
| Phenyl boronic acid / Phenyl acetylene | 1.25 mmol x 121.93 g mol ⁻¹ = 0.1524 g | | | | | 1.50 mmol x 102.13 g mol ⁻¹ = 0.1532 g | | | | |
| Pd(dba) ₂ / PdCl ₂ | 0.025 mmol x 575 g mol ⁻¹ = 0.0144 g | | | | | 0.10 mmol x 177.33 g mol ⁻¹ = 0.0177 g | | | | |
| PPh ₃ | 0.15 mmol x 262.29 g mol ⁻¹ = 0.0393 g | | | | | 0.20 mmol x 262.29 g mol ⁻¹ = 0.0524 g | | | | |
| Reaction water | 0.6667 mL x 1 g mL ⁻¹ = 0.6667 g | | | | | - | | | | |
| K ₂ CO ₃ / KO ^t -Bu | 2 mmol x 138.21 g mol ⁻¹ = 0.2764 g | | | | | 2 mmol x 112.21 g mol ⁻¹ = 0.2244 g | | | | |
| H ₂ O | 10 mL x 1 g mL ⁻¹ = 10.00 g | | | | | 10 mL x 1 g mL ⁻¹ = 10.00 g | | | | |
| EtOAc | 30 mL x 0.902 g mL ⁻¹ = 27.06 g | | | | | 30 mL x 0.902 g mL ⁻¹ = 27.06 g | | | | |
| MgSO ₄ | 0.10 g | | | | | 0.10 g | | | | |
| Hexane | 50 ml x 0.655 g mL ⁻¹ = 32.75 g | | | | | 50 ml x 0.655 g mL ⁻¹ = 32.75 g | | | | |
| silica | 10.00 g | | | | | 10.00 g | | | | |

total mass in process

Process mass intensity (PMI) = $\frac{\text{mass of product}}{\text{total mass in process}}$

mass of isolated product

Reaction mass efficiency (RME) = $\frac{\text{mass of isolated product}}{\text{total mass of reactants}} \times 100$

molecular weight of product

Atom economy (AE) = $\frac{\text{molecular weight of product}}{\text{total molecular weight of reactants}} \times 100$

RME

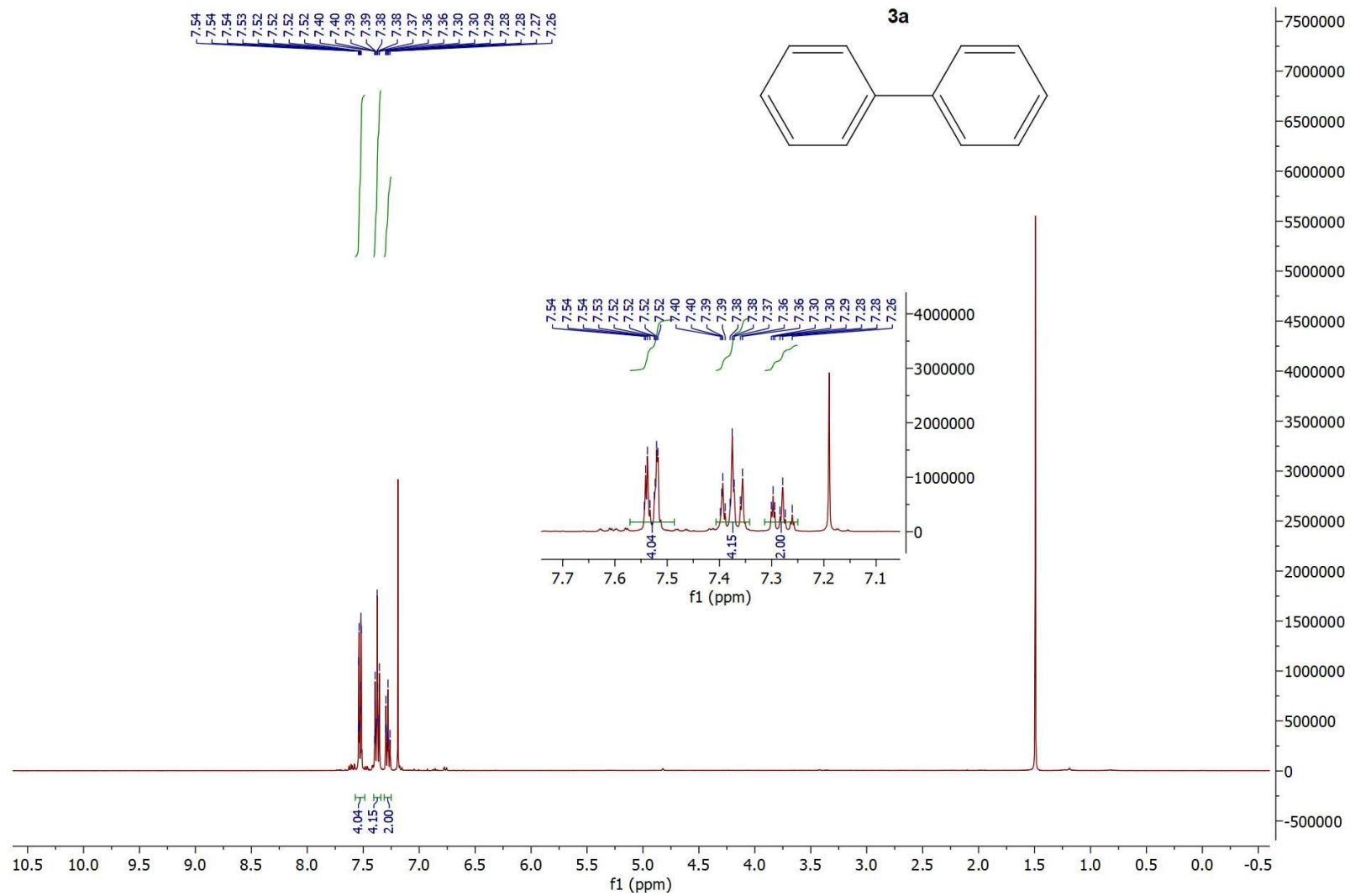
Optimum efficiency (OE) = $\frac{\text{RME}}{\text{AE}} \times 100$

Table S6. Green metric assessment of Suzuki-Miyaura and Sonogashira reactions.

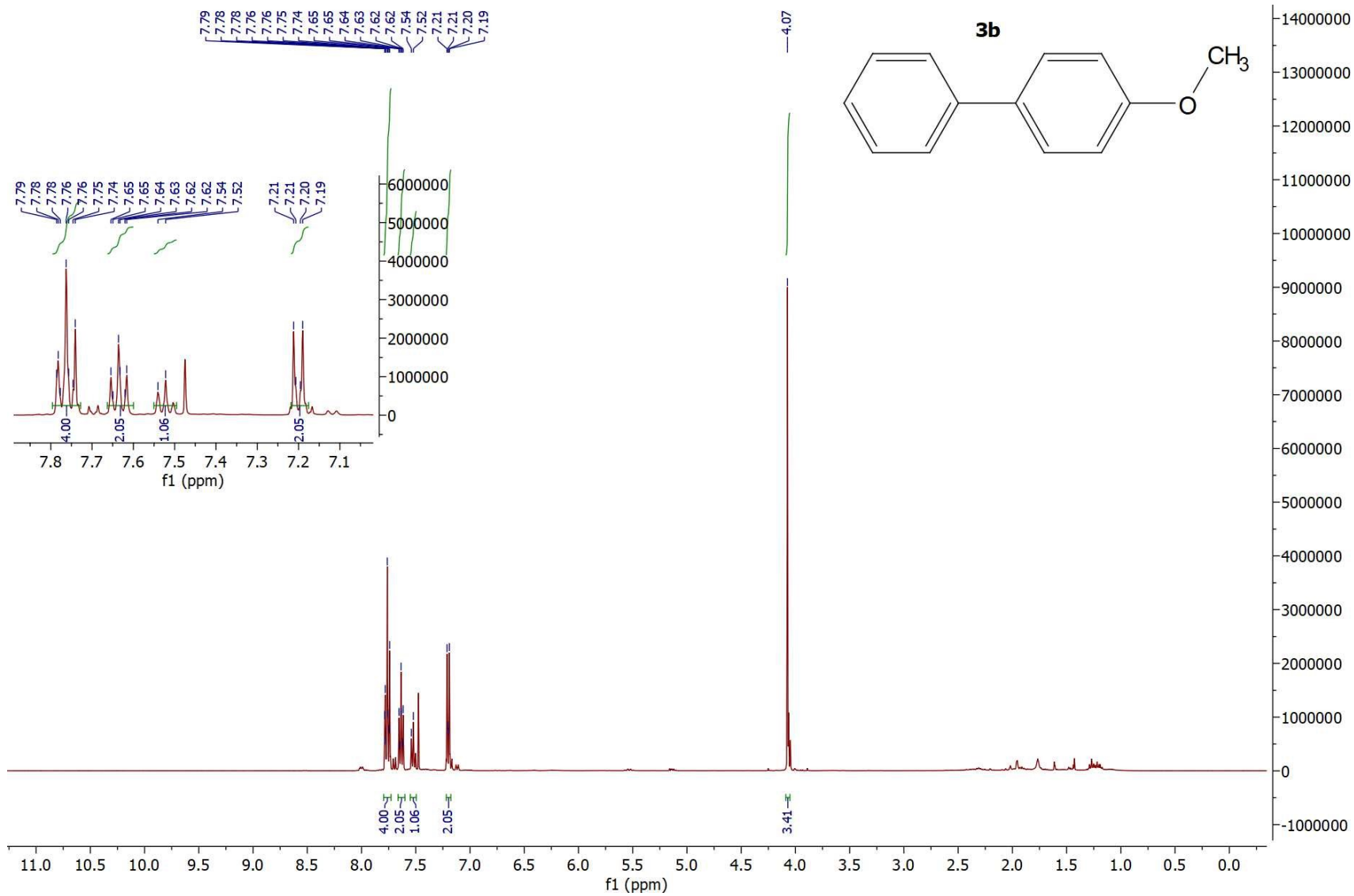
| Solvent | Suzuki-Miyaura reaction (Iodobenzene + Phenylboronic acid) | | | | | Sonogashira reaction (Iodobenzene + Phenylacetylene) | | | | |
|-----------|---|-------|------------|-----|-----|---|-------|------------|-----|-----|
| | TMO | DEDMO | Eucalyptol | THF | Tol | TMO | DEDMO | Eucalyptol | THF | Tol |
| Yield (%) | 89 | 86 | 86 | 85 | 88 | 91 | 81 | 76 | 85 | 80 |
| PMI | 604 | 625 | 627 | 633 | 612 | 512 | 575 | 615 | 549 | 583 |
| RME (%) | 38 | 37 | 37 | 36 | 38 | 45 | 40 | 38 | 42 | 40 |
| AE (%) | 47 | | | | | 55 | | | | |
| OE (%) | 81 | 79 | 79 | 77 | 81 | 82 | 73 | 69 | 76 | 73 |

S7. ¹H and ¹³C NMR Spectra of Synthesized Compounds

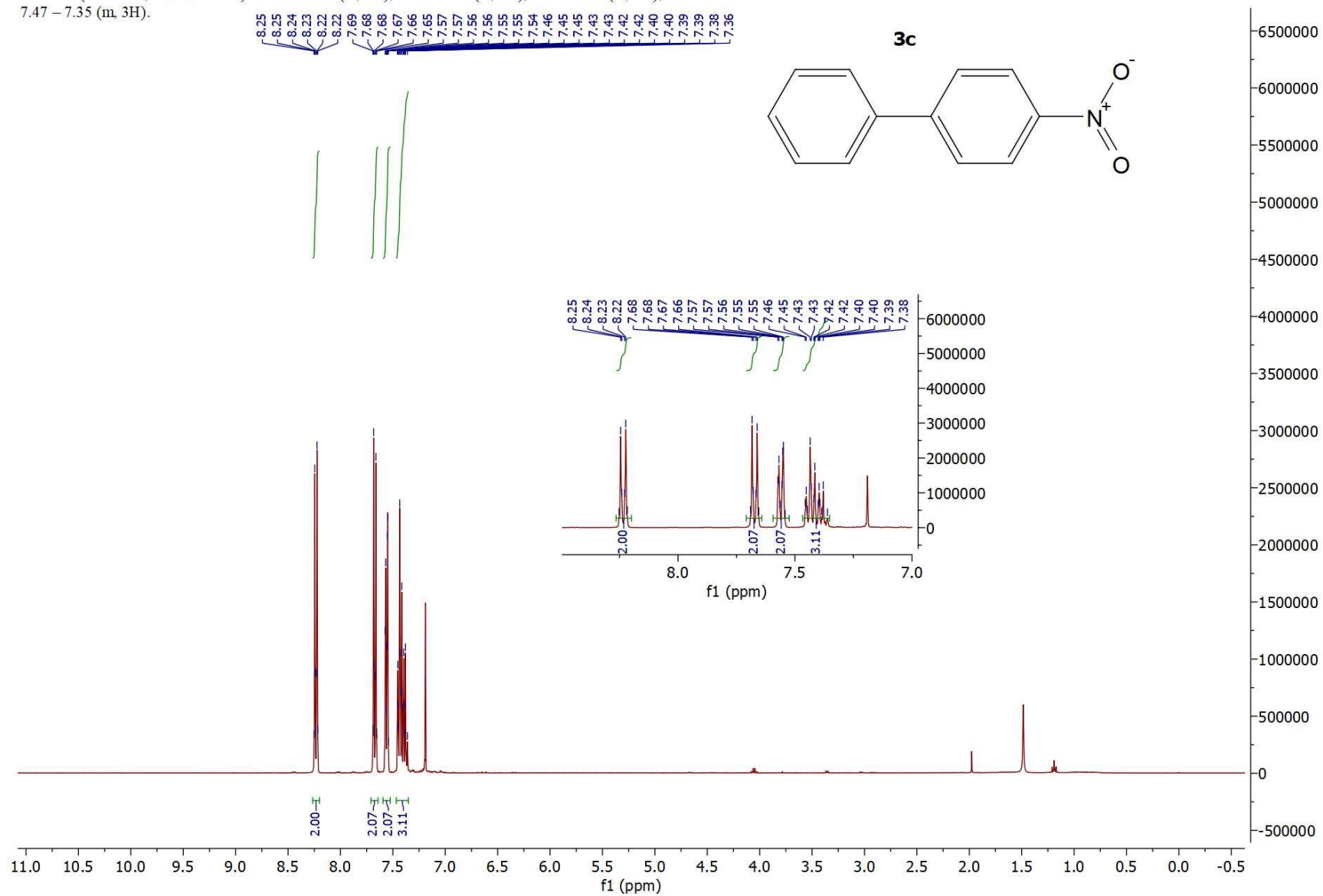
¹H NMR (400 MHz, Chloroform-*d*) δ 7.56 – 7.50 (m, 4H), 7.41 – 7.34 (m, 4H), 7.31 – 7.25 (m, 2H).



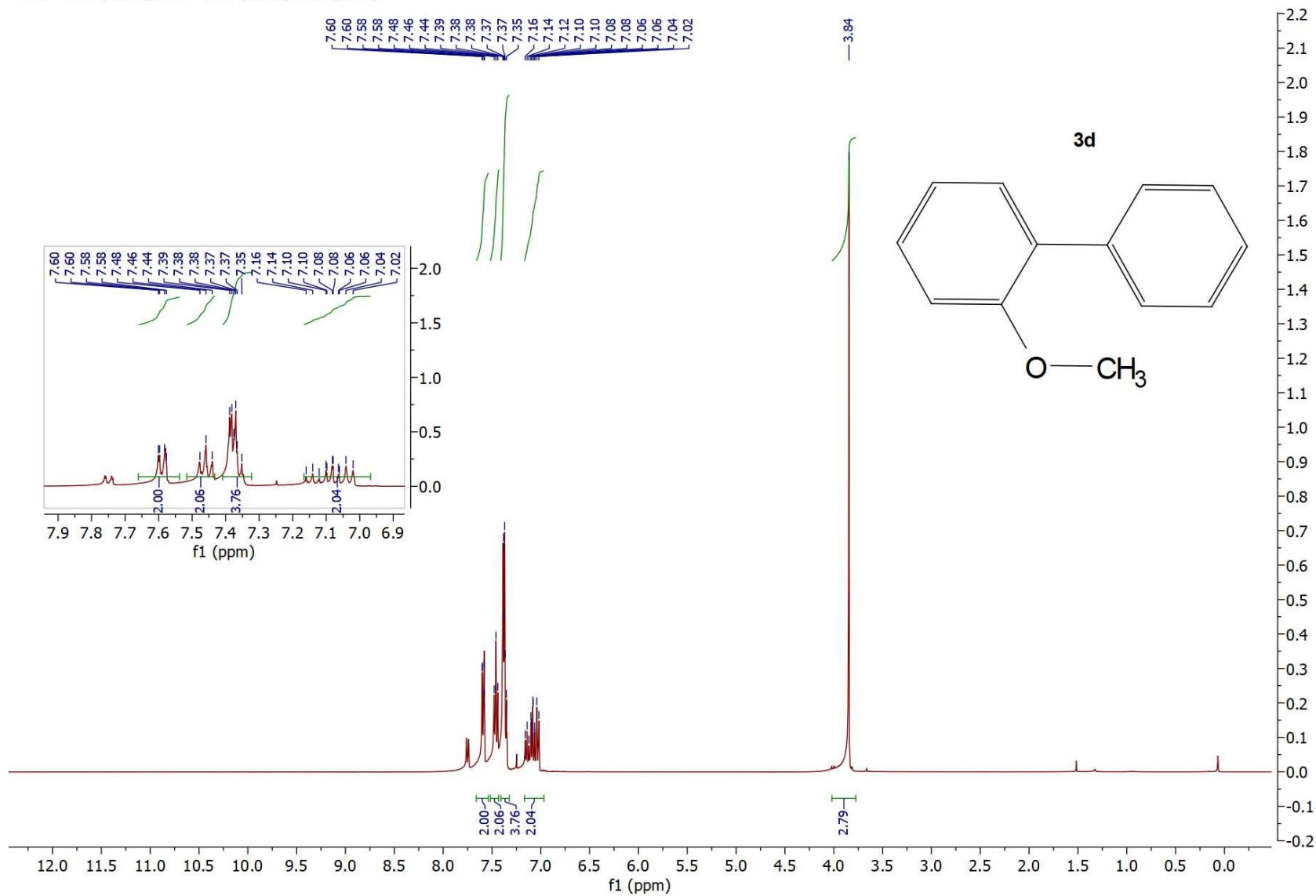
^1H NMR (400 MHz, Chloroform- d) δ 7.80 – 7.73 (m, 4H), 7.63 (dd, J = 8.4, 6.9 Hz, 2H), 7.53 (d, J = 7.4 Hz, 1H), 7.23 – 7.17 (m, 2H), 4.07 (s, 3H).



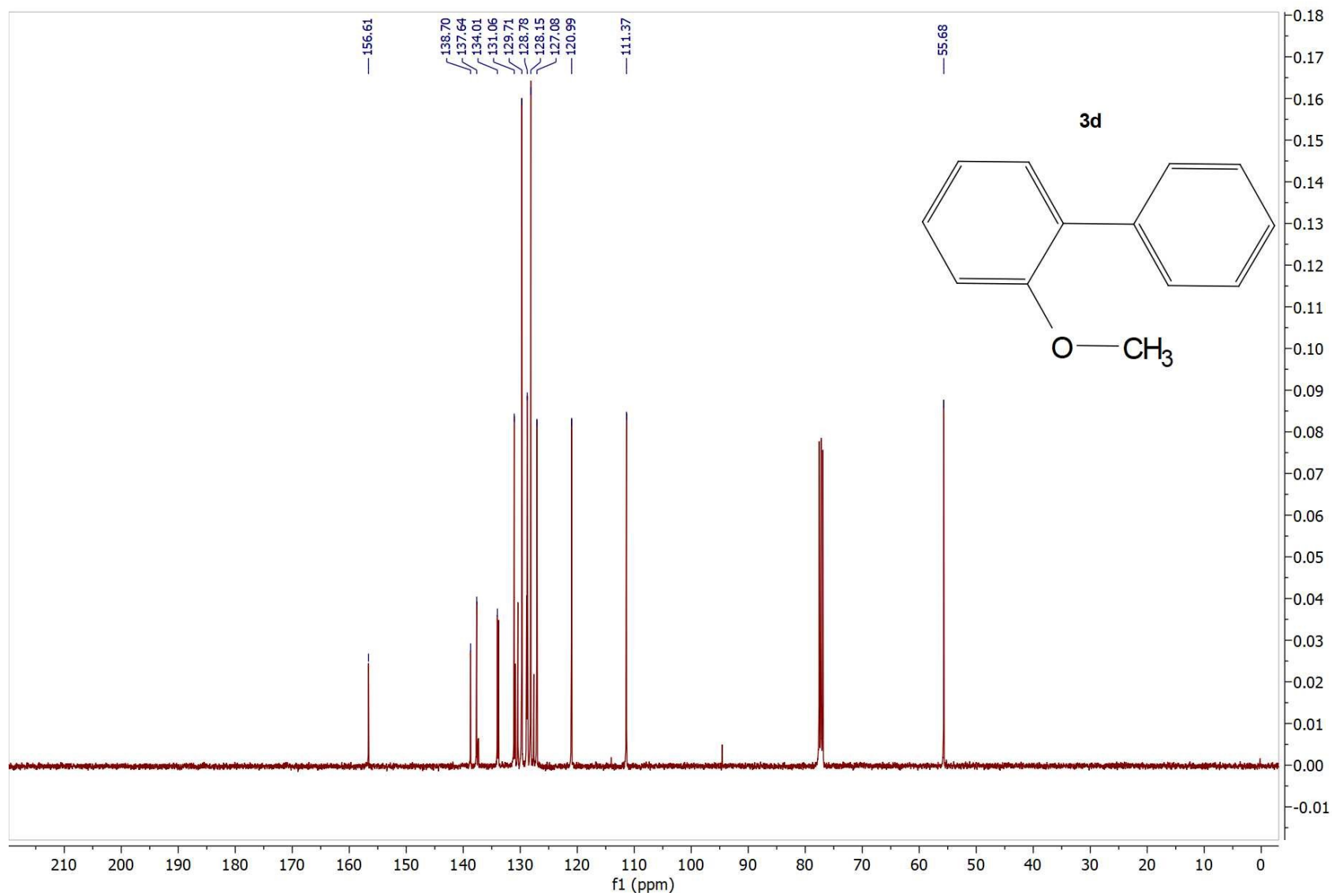
^1H NMR (400 MHz, Chloroform-*d*) δ 8.27 – 8.21 (m, 2H), 7.70 – 7.65 (m, 2H), 7.58 – 7.54 (m, 2H), 7.47 – 7.35 (m, 3H).



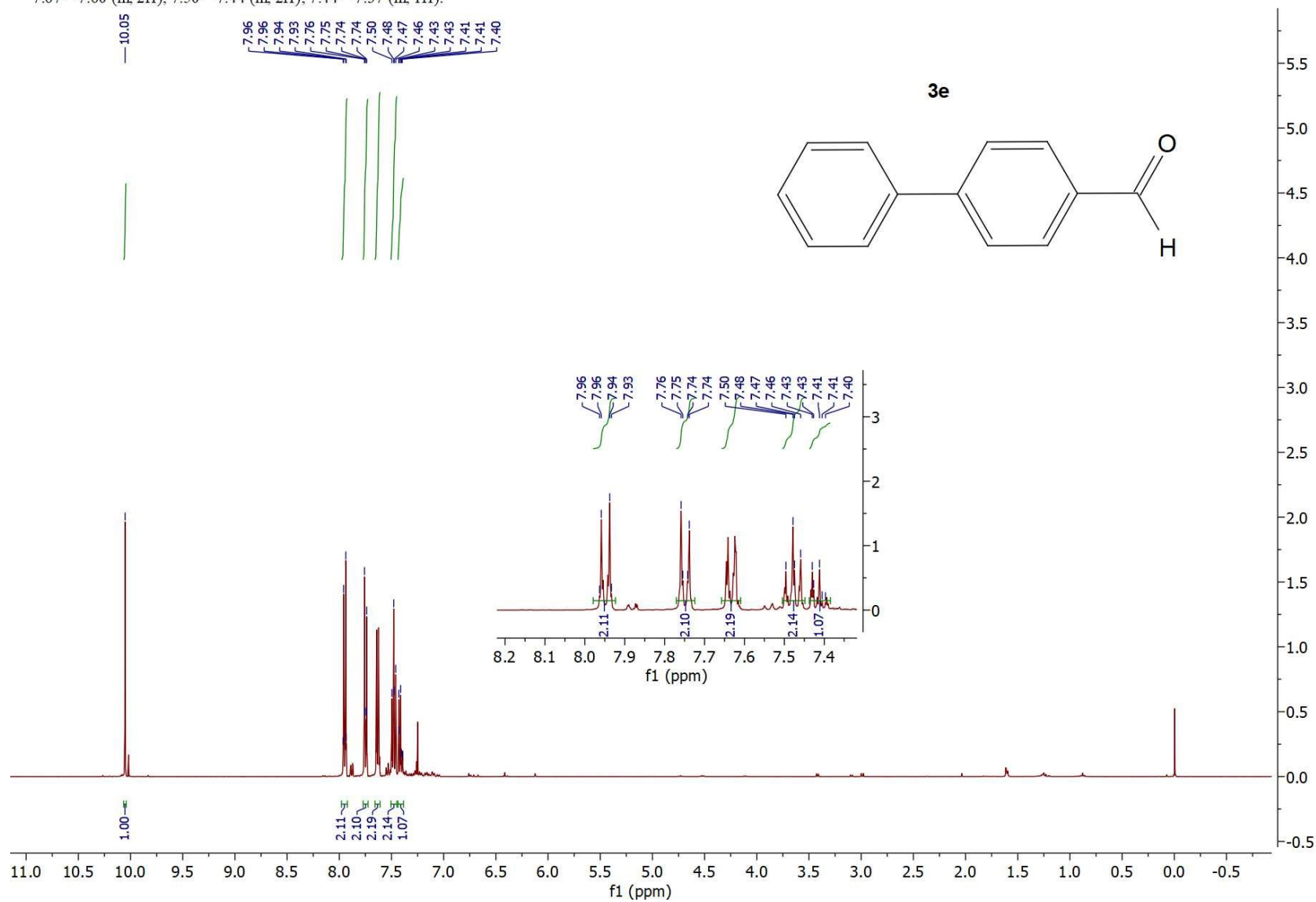
^1H NMR (400 MHz, Chloroform-*d*) δ 7.59 (dd, $J = 7.8, 1.8$ Hz, 2H), 7.46 (t, $J = 7.3$ Hz, 2H), 7.41 – 7.34 (m, 3H), 7.17 – 6.99 (m, 2H), 3.84 (s, 3H).



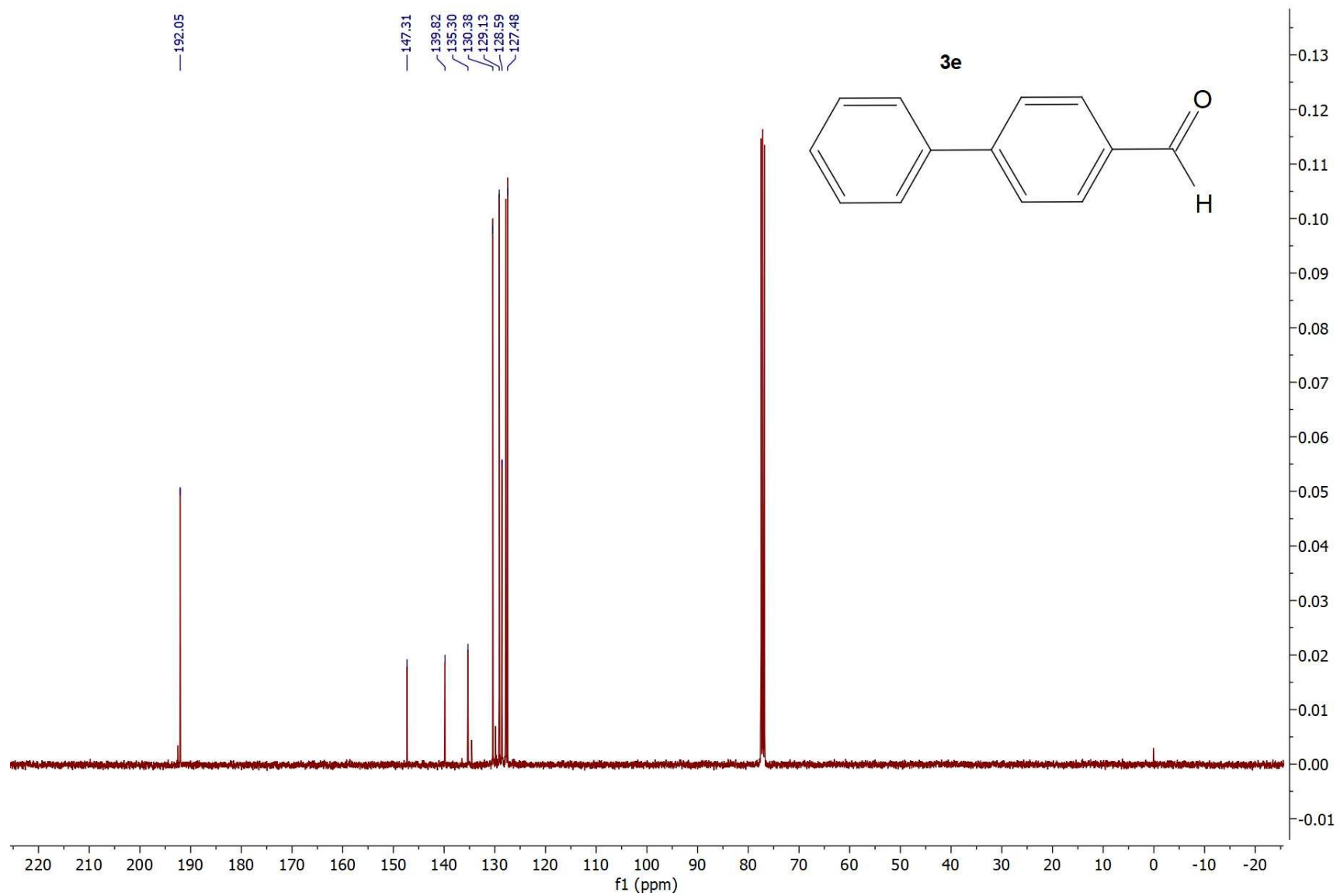
^{13}C NMR (101 MHz, CHLOROFORM- D) δ 156.61, 138.70, 137.64, 134.01, 131.06, 129.71, 128.78, 128.15, 127.08, 120.99, 111.37, 55.68.



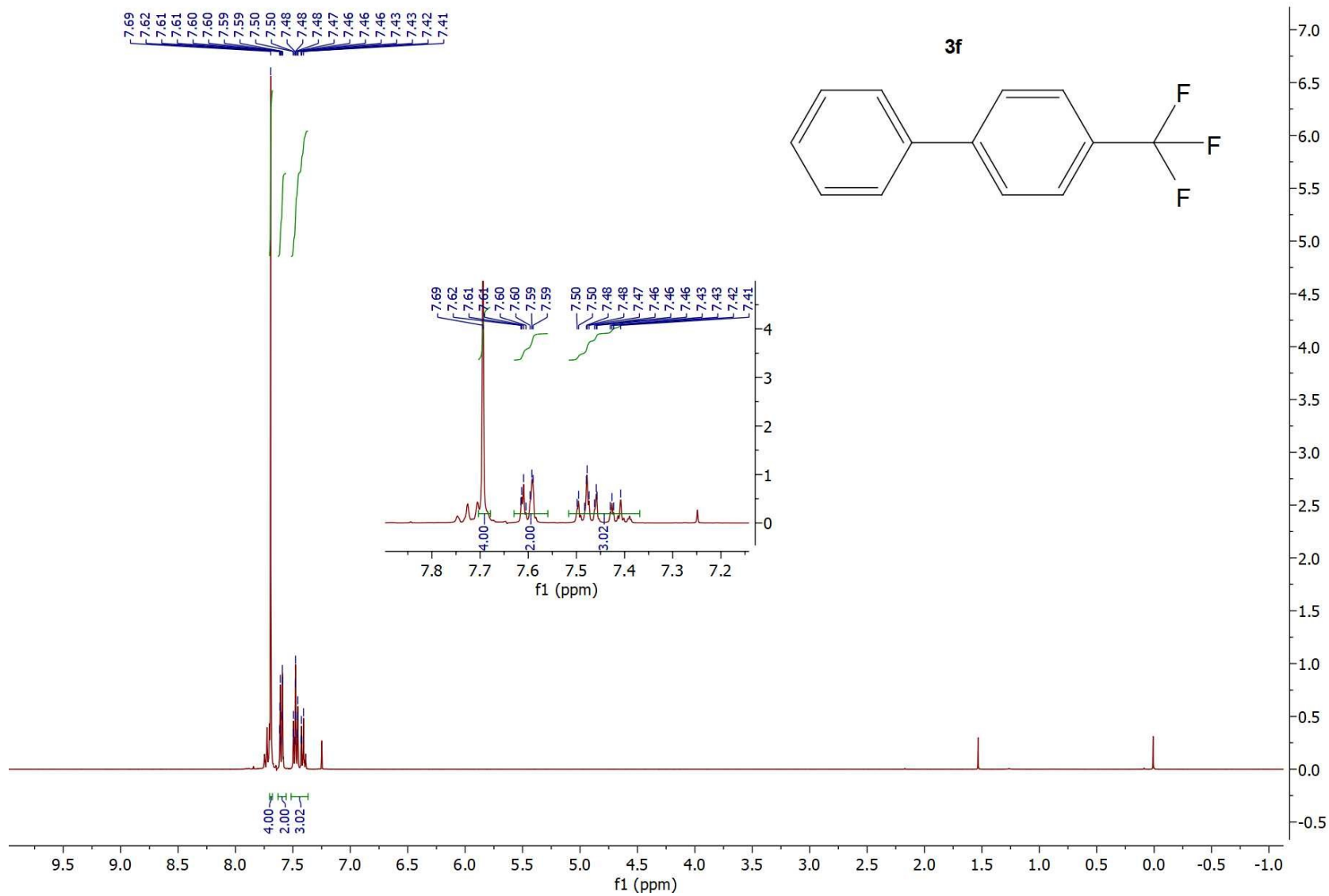
^1H NMR (400 MHz, Chloroform-*d*) δ 10.05 (s, 1H), 7.98 – 7.92 (m, 2H), 7.78 – 7.72 (m, 2H), 7.67 – 7.60 (m, 2H), 7.50 – 7.44 (m, 2H), 7.44 – 7.37 (m, 1H).



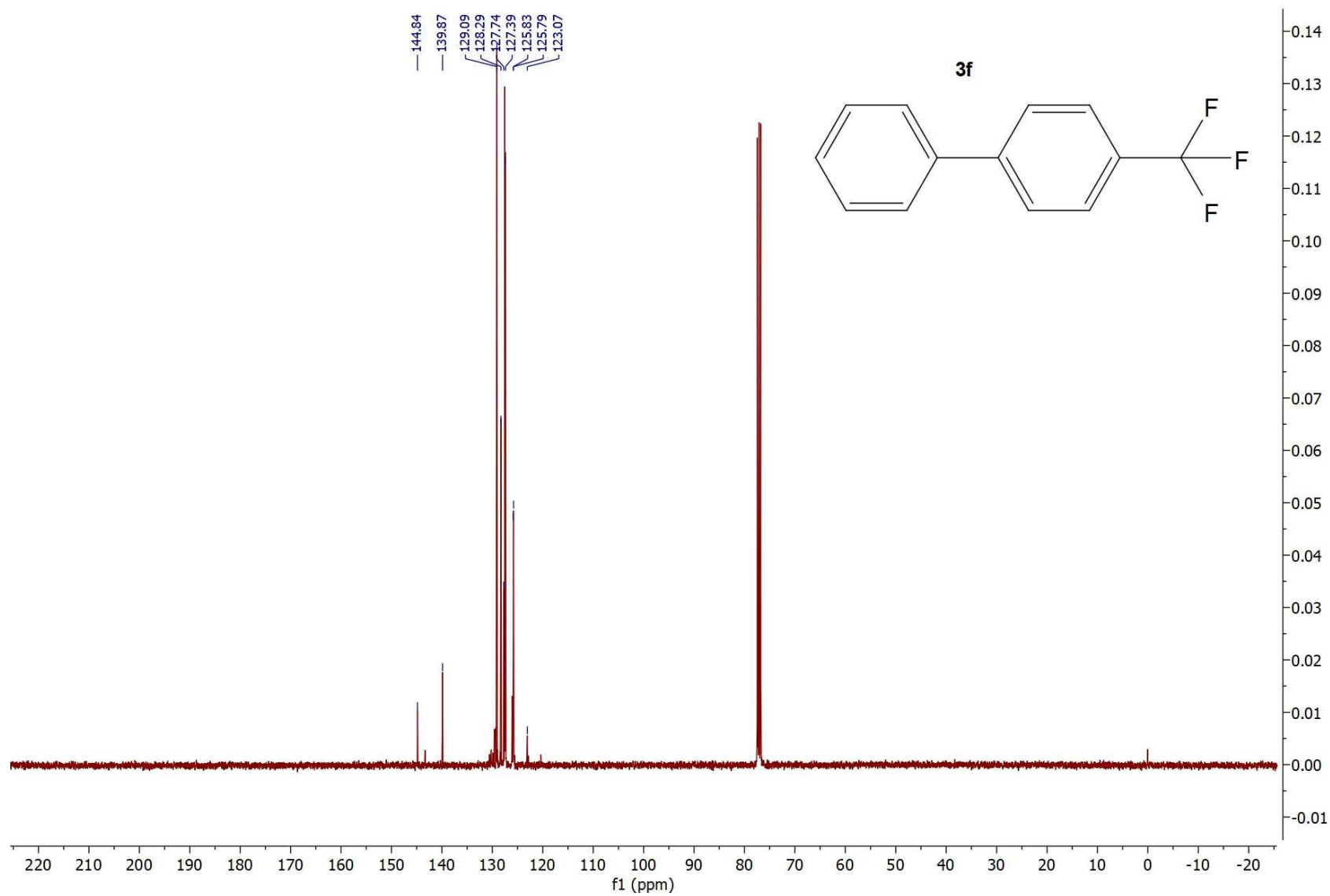
^{13}C NMR (101 MHz, CHLOROFORM-*D*) δ 192.05, 147.31, 139.82, 135.30, 130.38, 129.13, 128.59, 127.48.



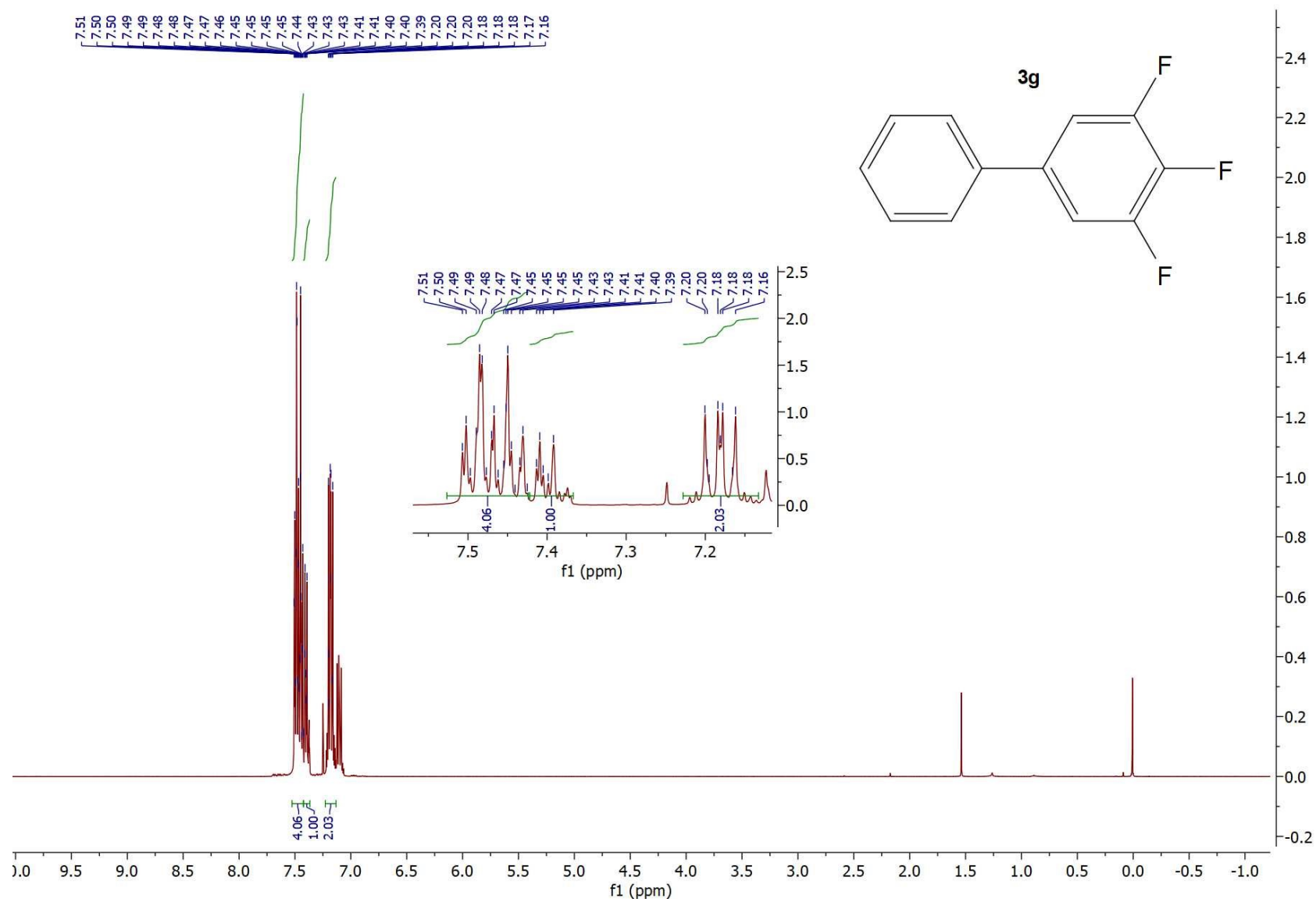
^1H NMR (400 MHz, Chloroform-*d*) δ 7.69 (s, 4H), 7.63 – 7.58 (m, 2H), 7.51 – 7.38 (m, 3H).



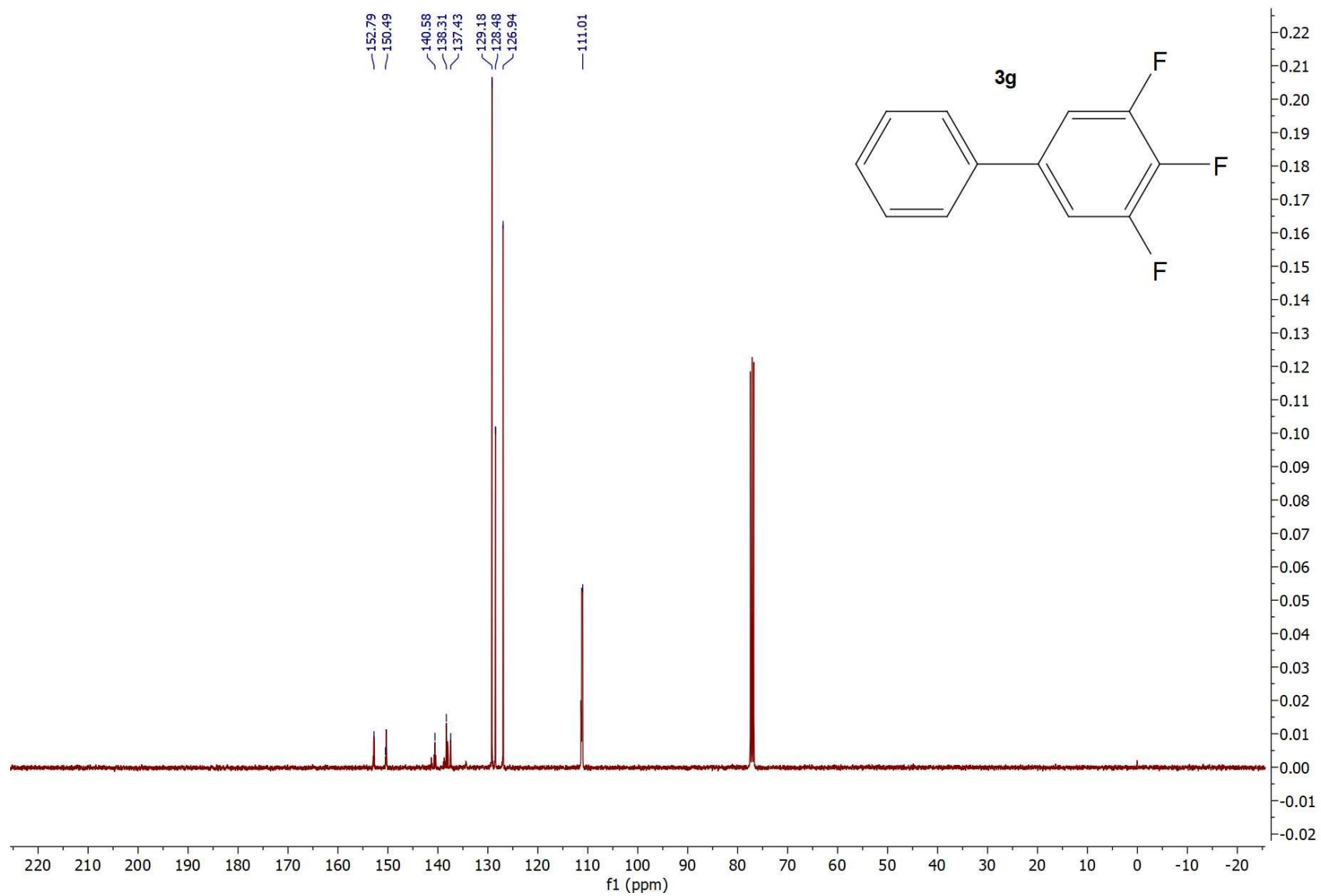
^{13}C NMR (101 MHz, CHLOROFORM-*D*) δ 144.84, 139.87, 129.09, 128.29, 127.74, 127.39, 125.83, 125.79, 123.07.



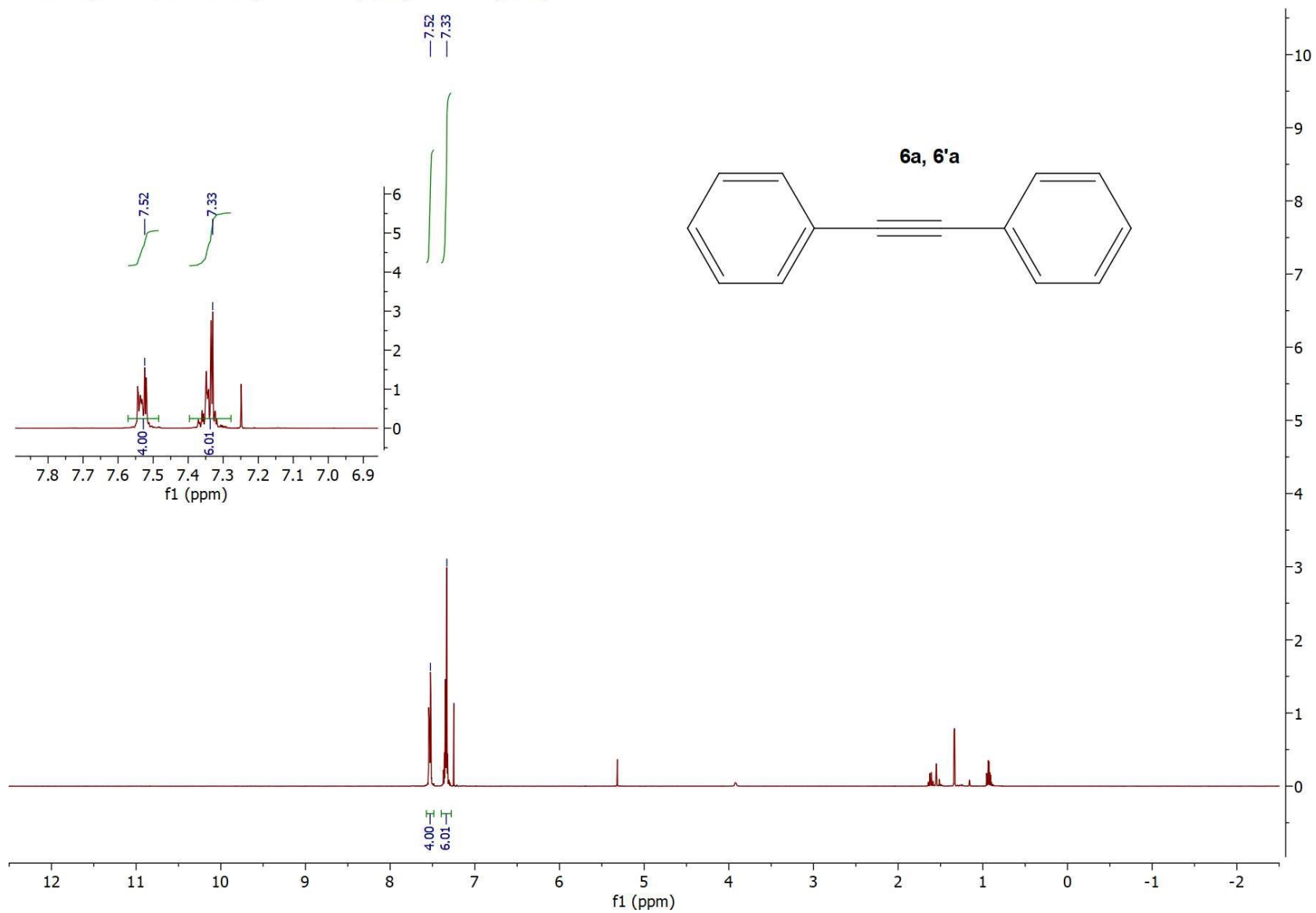
^1H NMR (400 MHz, Chloroform-*d*) δ 7.53 – 7.42 (m, 4H), 7.42 – 7.37 (m, 1H), 7.23 – 7.13 (m, 2H).



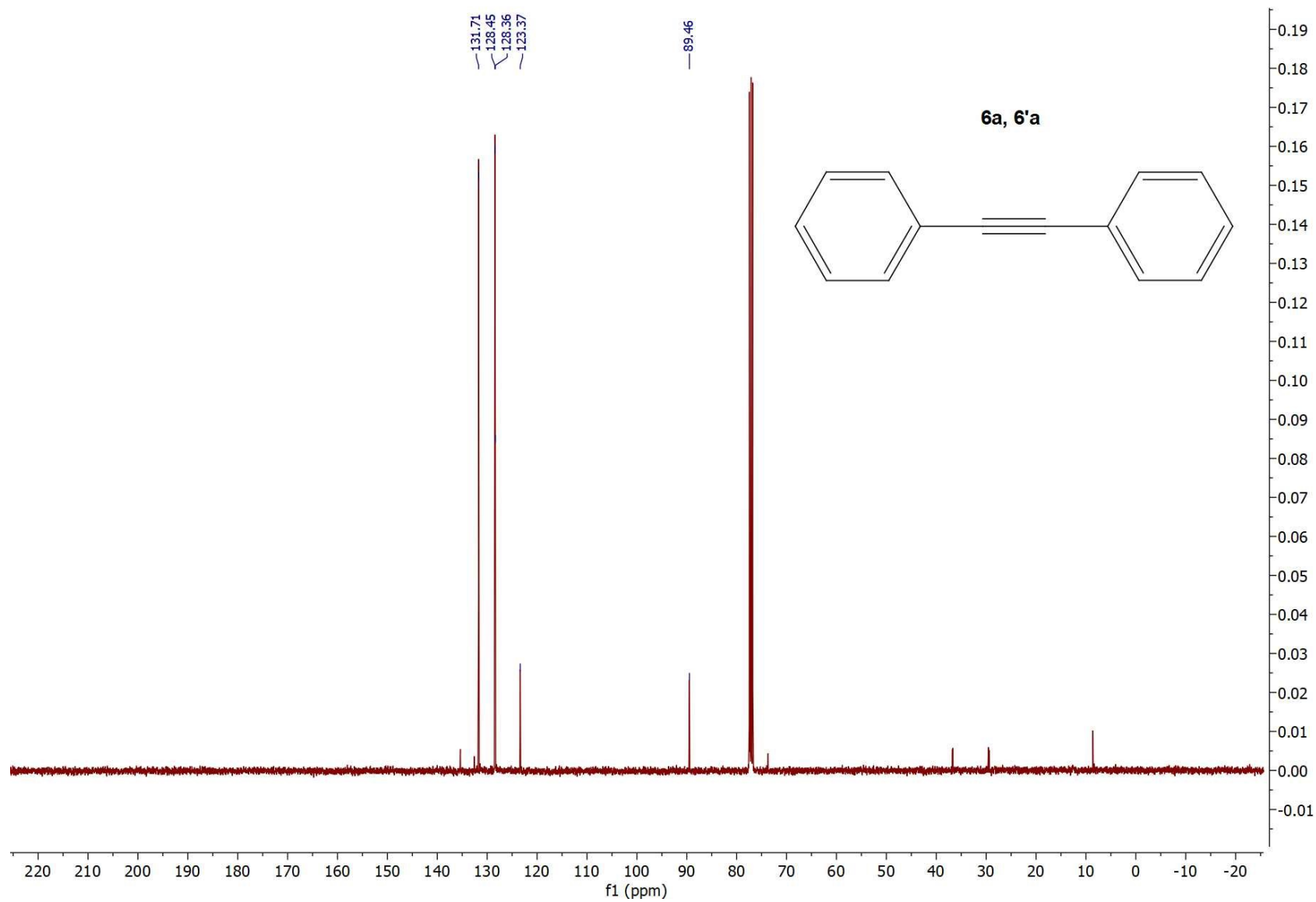
^{13}C NMR (101 MHz, CHLOROFORM-*D*) δ 152.79, 150.49, 140.58, 138.31, 137.43, 129.18, 128.48, 126.94, 111.01.



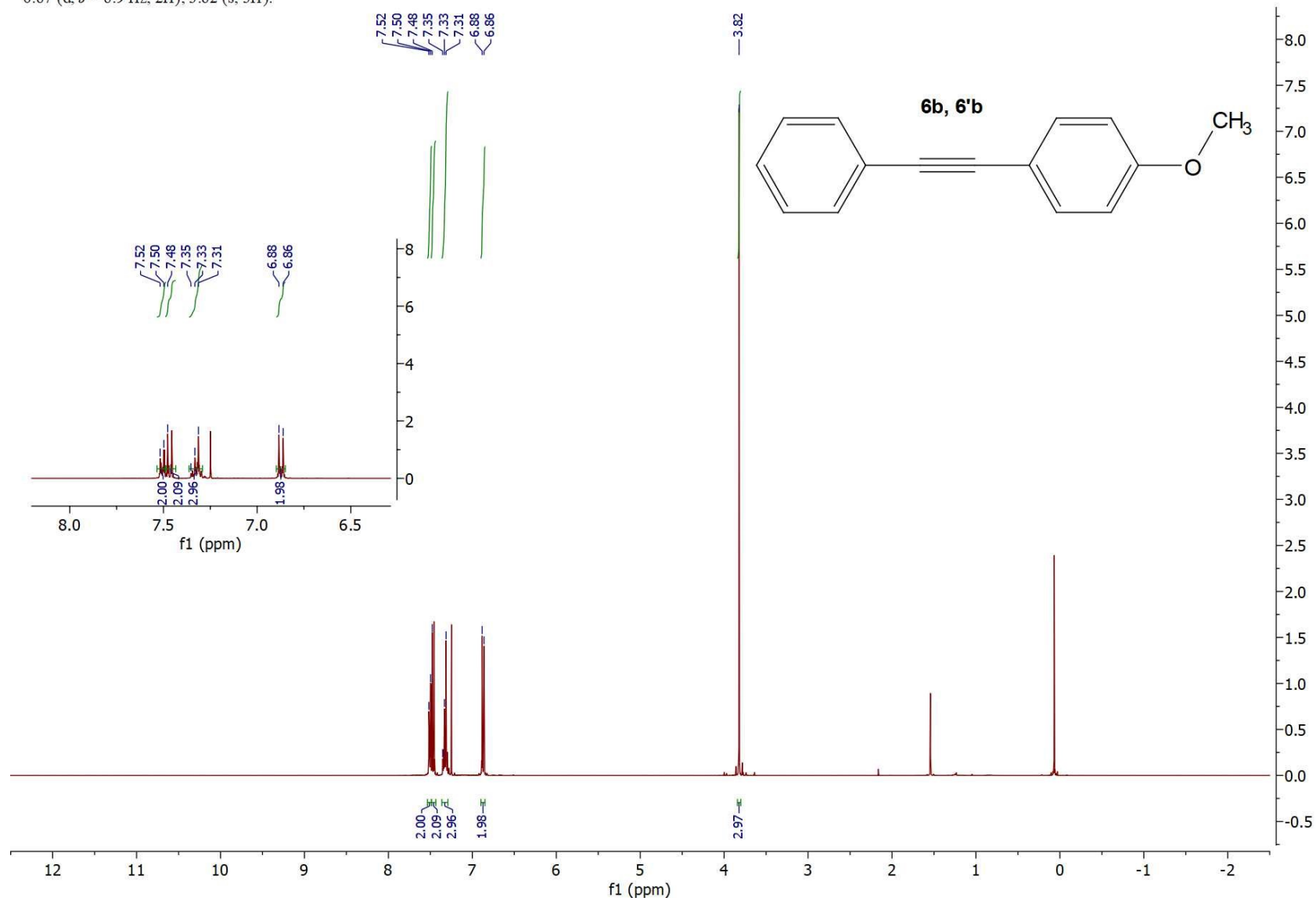
^1H NMR (400 MHz, Chloroform-*d*) δ 7.57 – 7.48 (m, 4H), 7.40 – 7.28 (m, 6H).



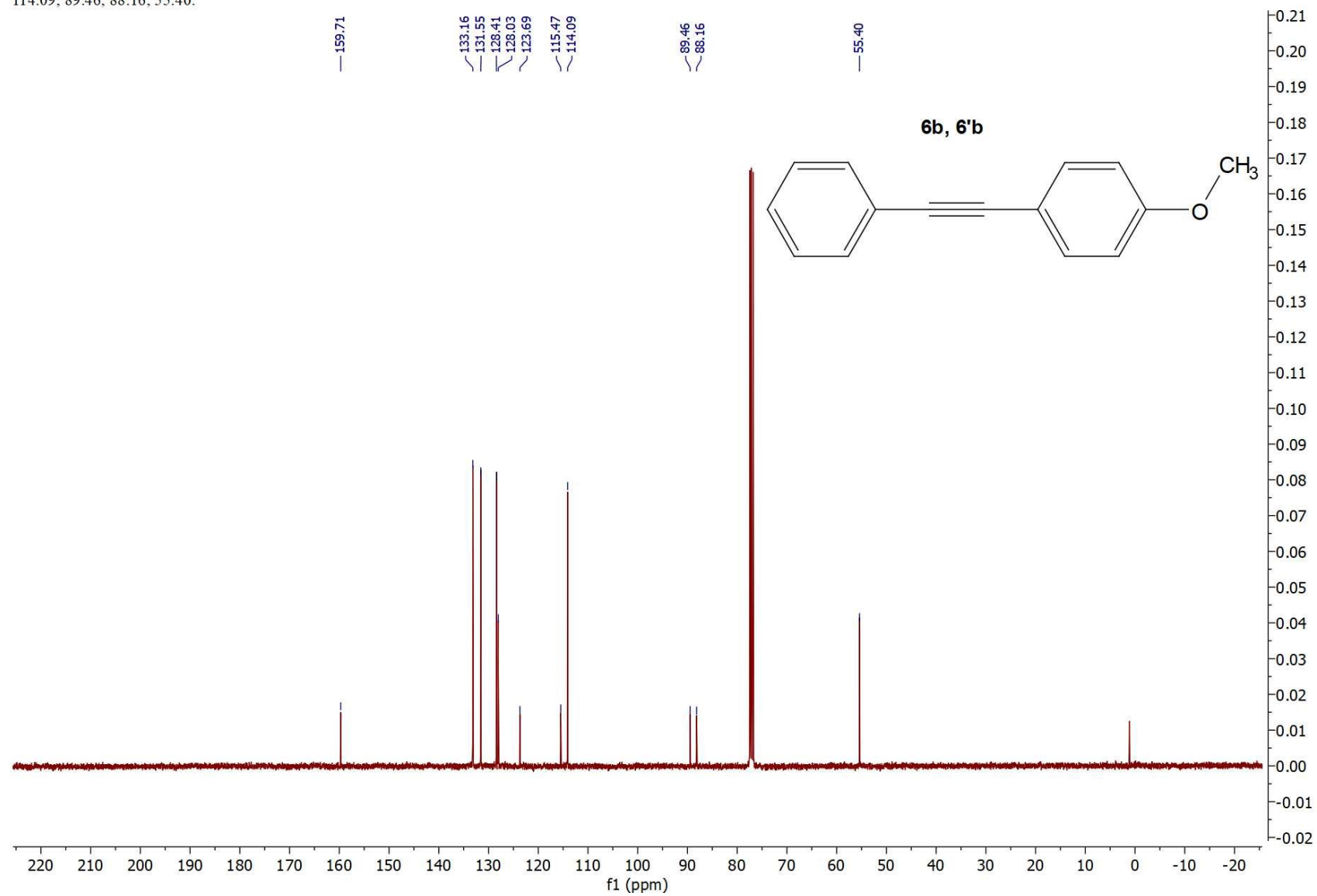
^{13}C NMR (101 MHz, Chloroform-*d*) δ 131.71, 128.45, 128.36, 123.37, 89.46



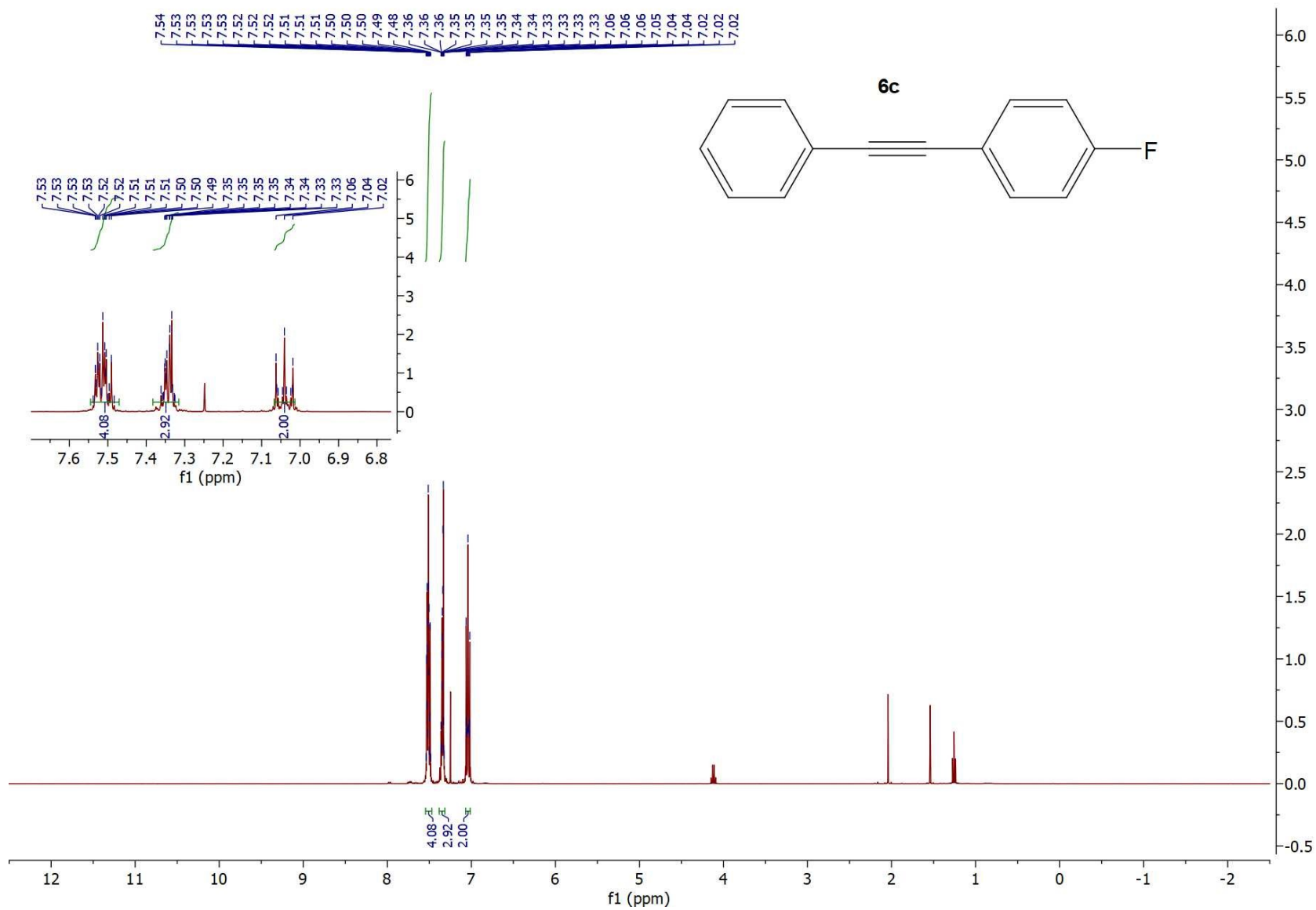
^1H NMR (400 MHz, Chloroform-*d*) δ 7.51 (d, $J = 8.1$ Hz, 2H), 7.48 (s, 2H), 7.36 – 7.29 (m, 3H), 6.87 (d, $J = 8.9$ Hz, 2H), 3.82 (s, 3H).



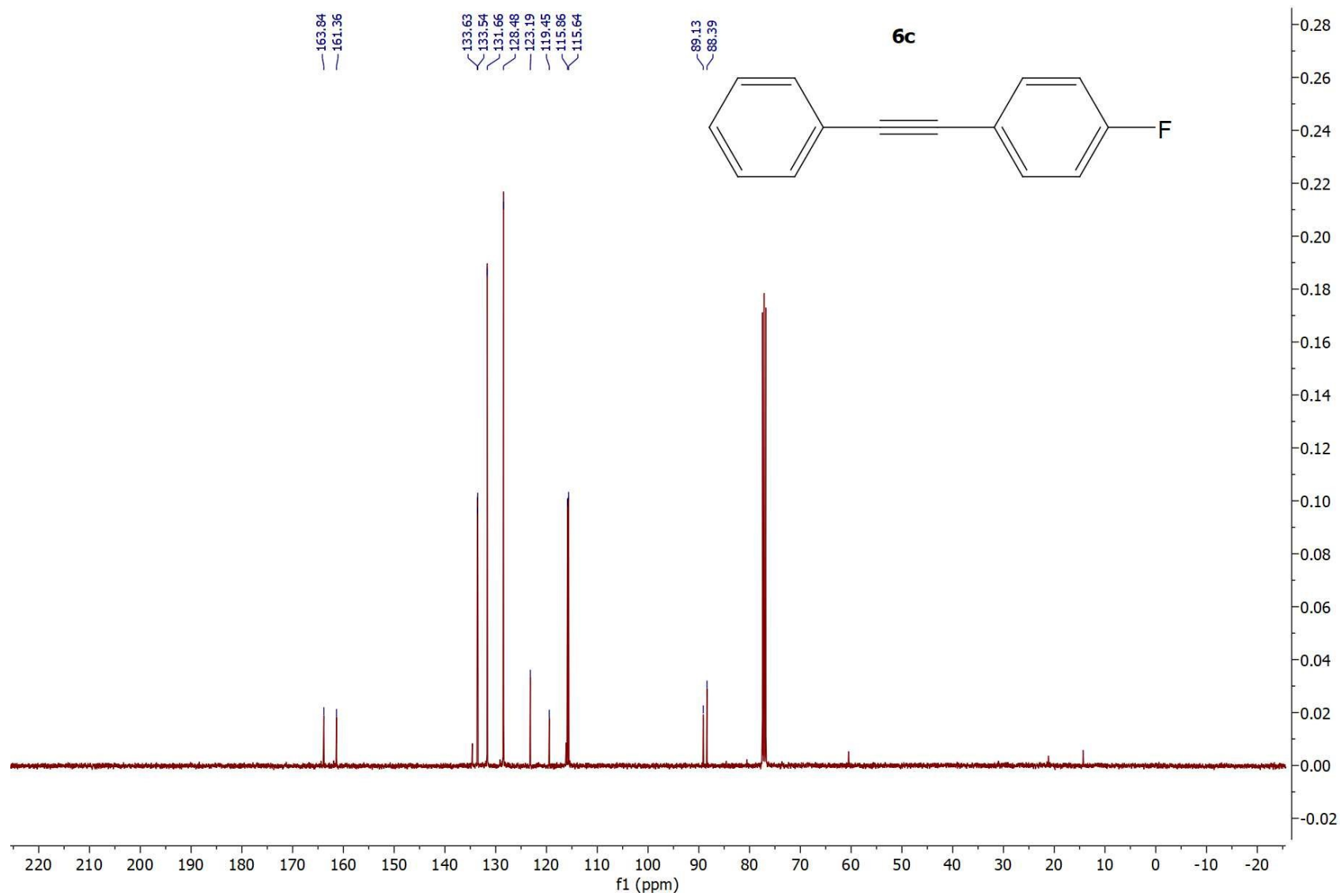
¹³C NMR (101 MHz, Chloroform-*d*) δ 159.71, 133.16, 131.55, 128.41, 128.03, 123.69, 115.47, 114.09, 89.46, 88.16, 55.40.



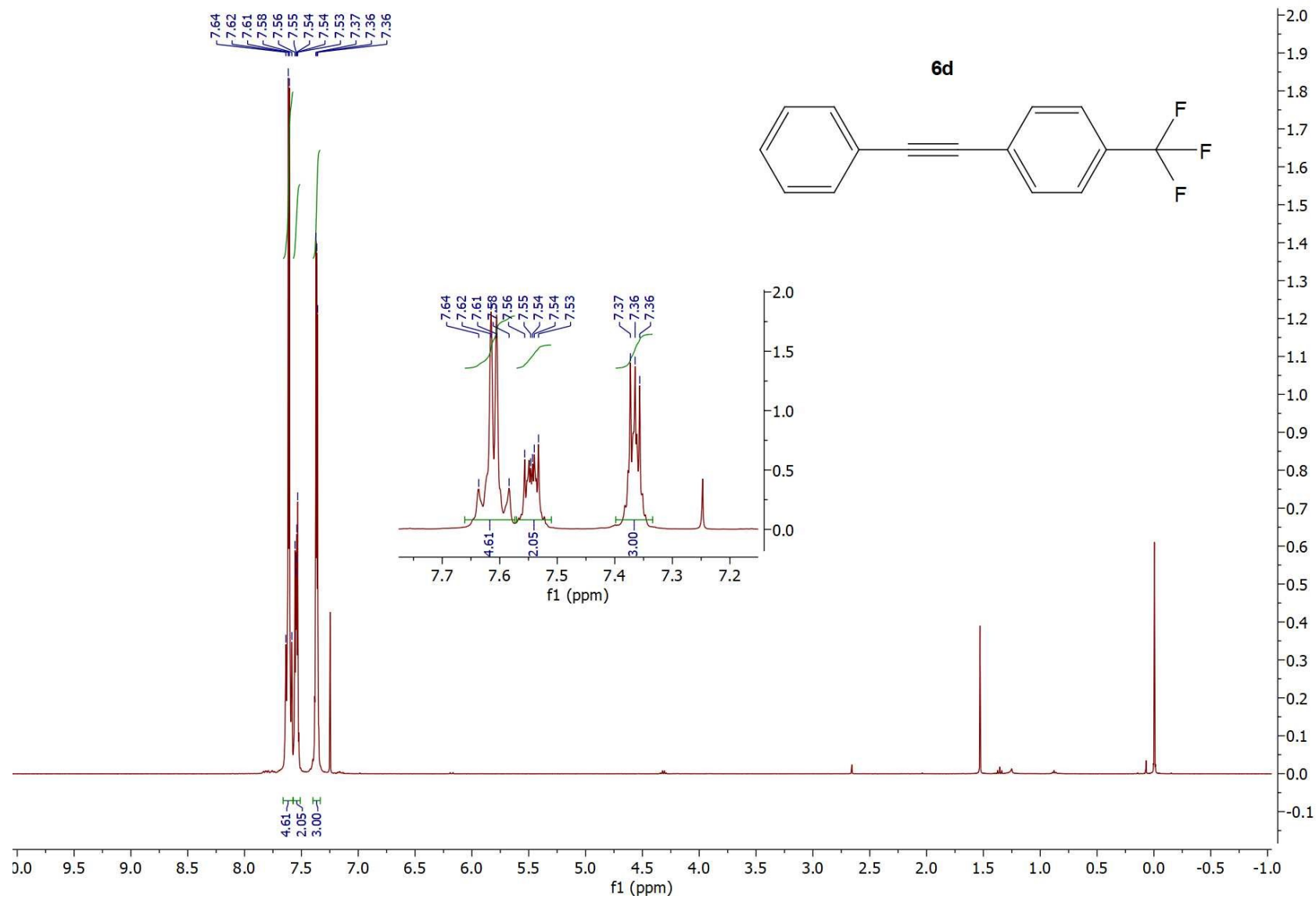
^1H NMR (400 MHz, Chloroform-*d*) δ 7.54 – 7.48 (m, 4H), 7.36 – 7.29 (m, 3H), 7.07 – 6.99 (m, 2H).



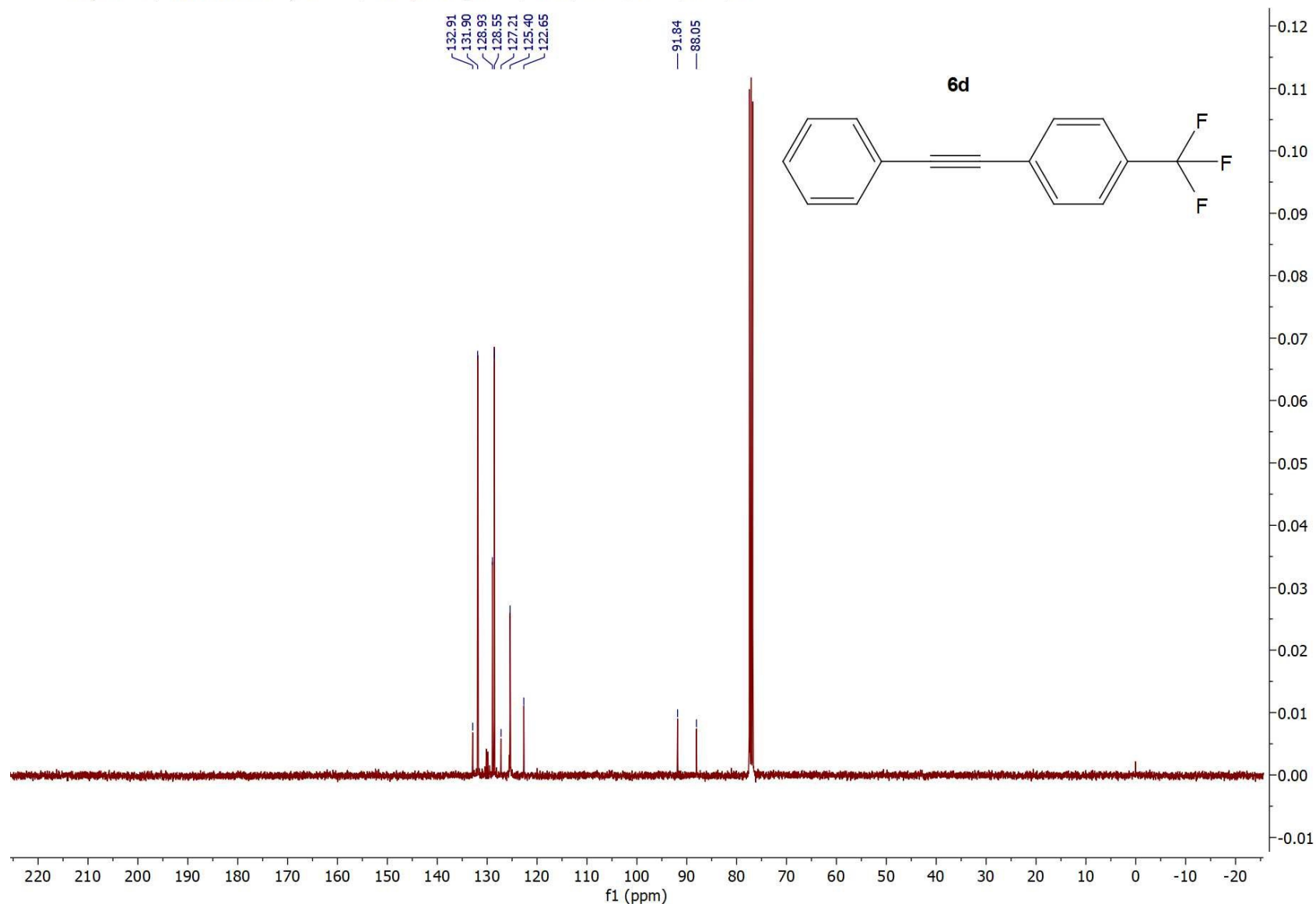
^{13}C NMR (101 MHz, Chloroform-*d*) δ 163.84, 133.63, 133.54, 131.66, 128.48, 123.19, 119.45, 115.86, 115.64, 89.13, 88.39.



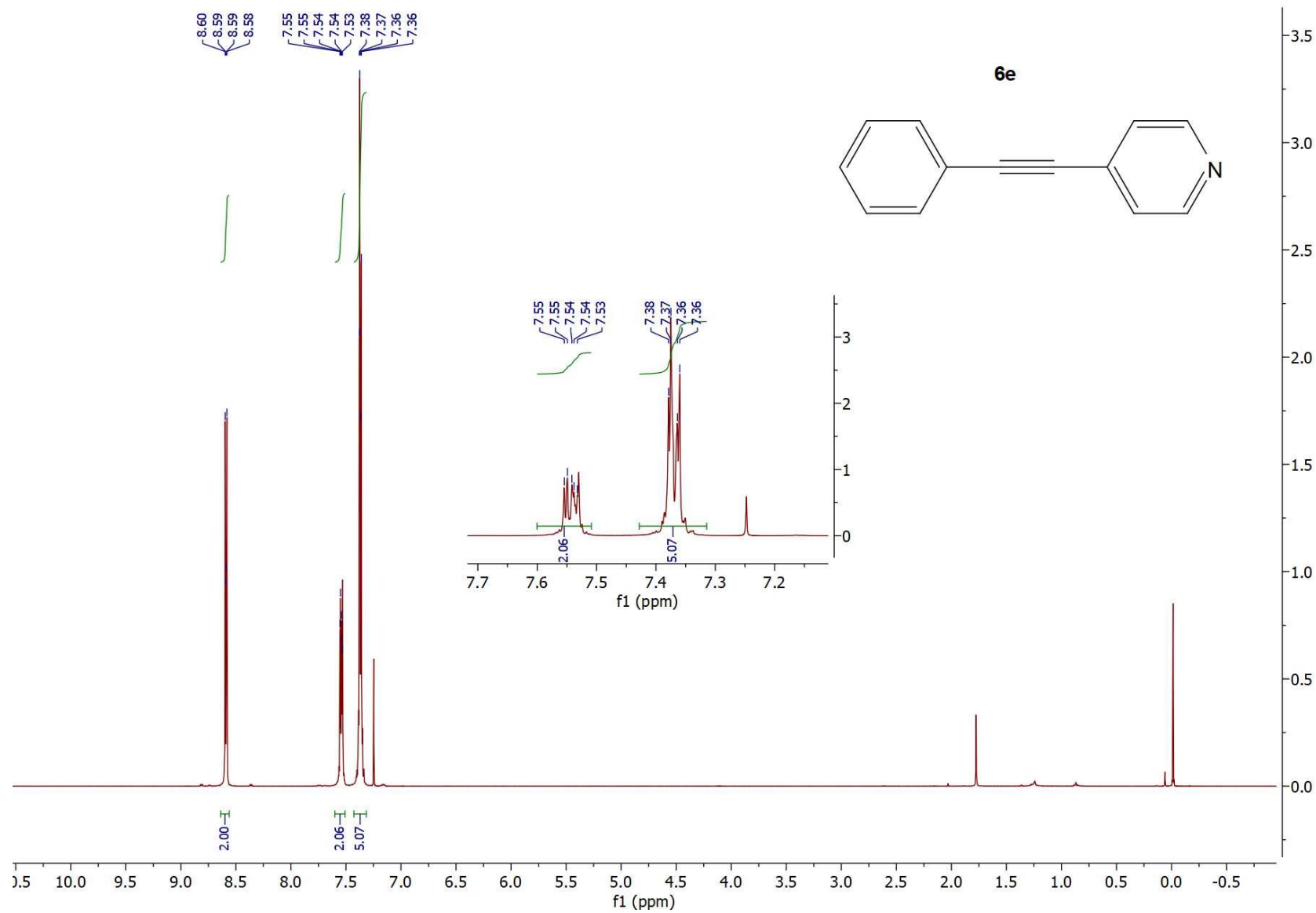
^1H NMR (400 MHz, Chloroform-*d*) δ 7.65 – 7.58 (m, 4H), 7.57 – 7.51 (m, 2H), 7.40 – 7.33 (m, 3H).



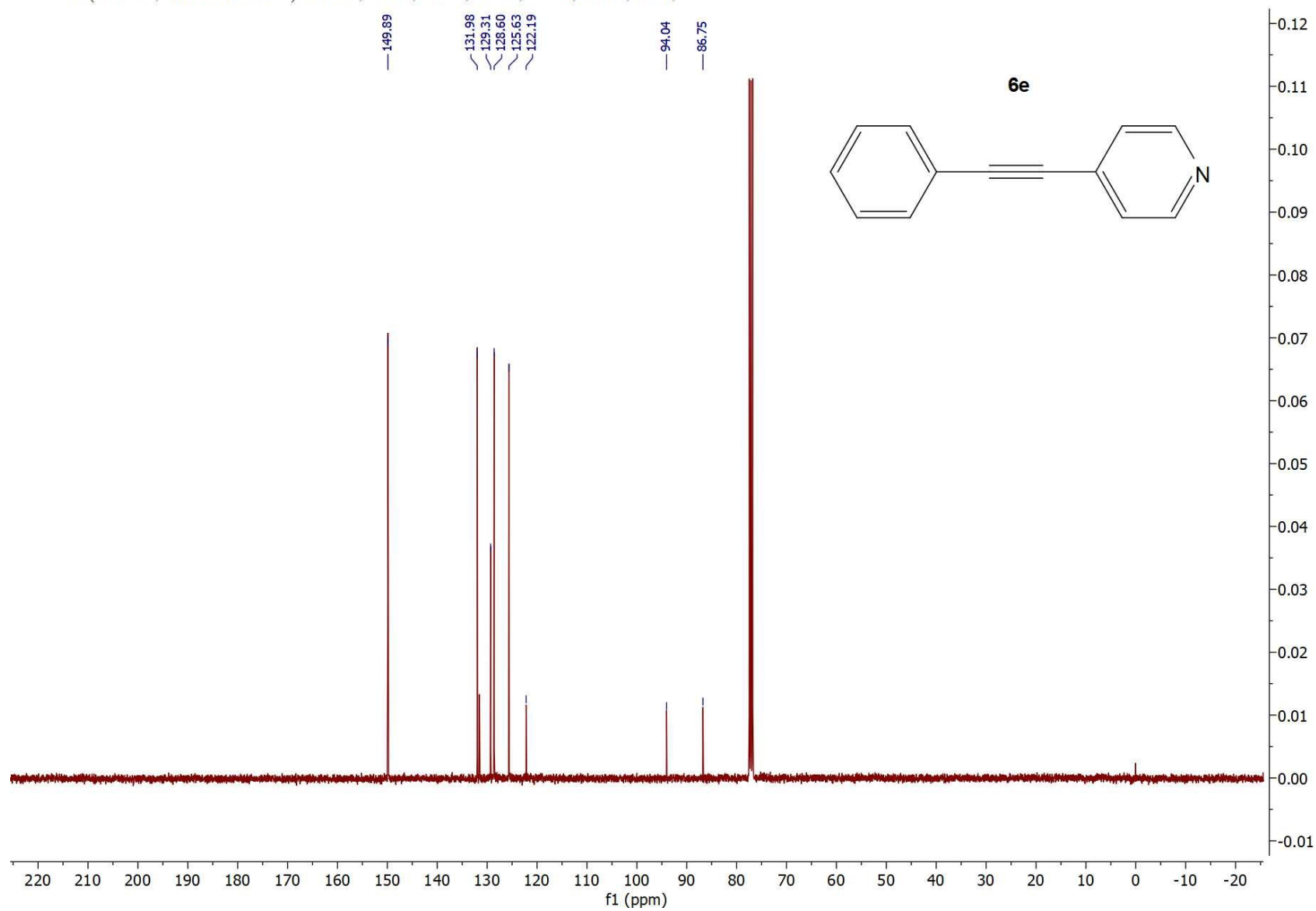
^{13}C NMR (101 MHz, CHLOROFORM-*D*) δ 132.91, 131.90, 128.93, 128.55, 127.21, 125.40, 122.65, 91.84, 88.05.



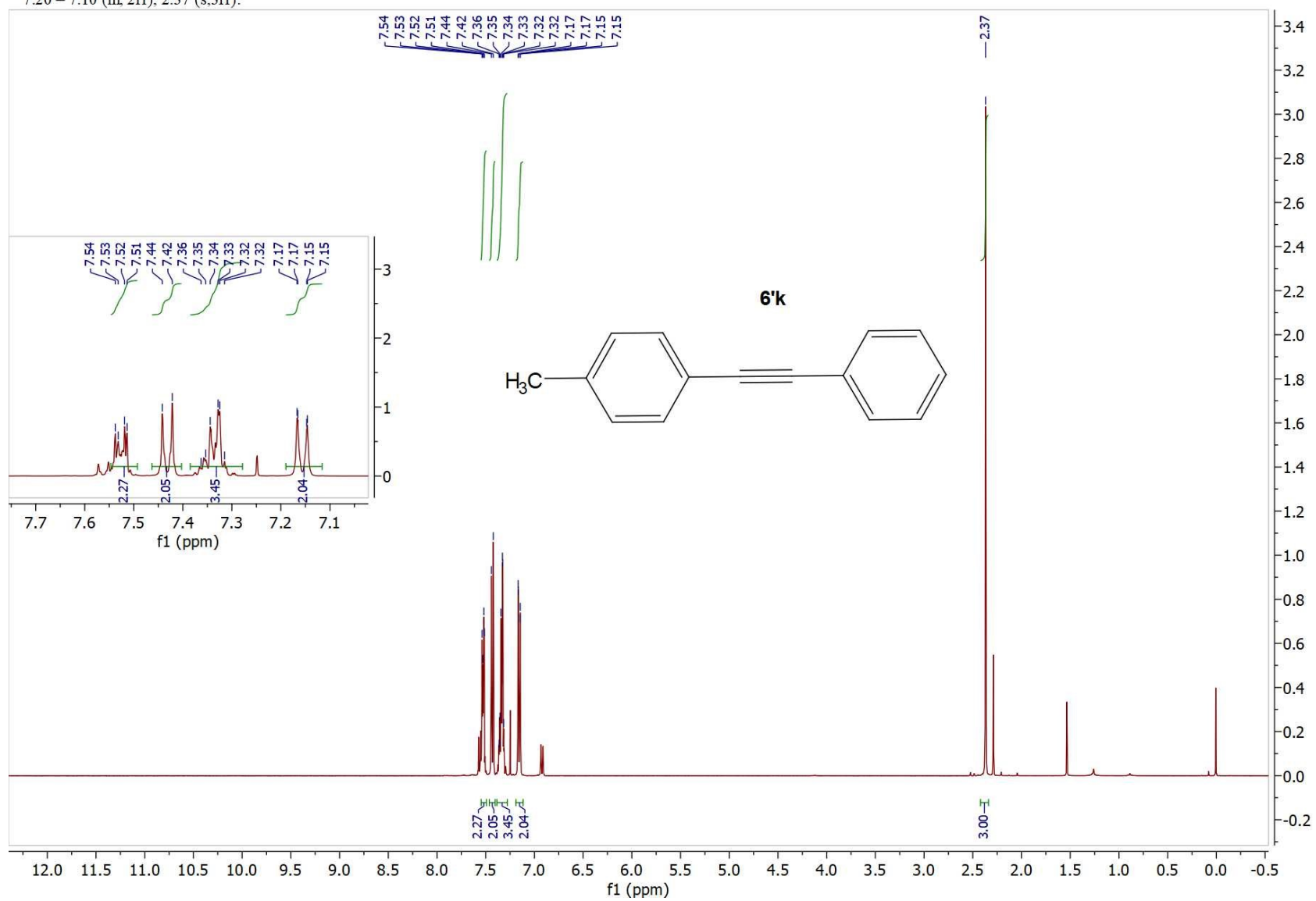
^1H NMR (400 MHz, Chloroform-*d*) δ 8.62 – 8.56 (m, 2H), 7.58 – 7.50 (m, 2H), 7.41 – 7.31 (m, 5H).



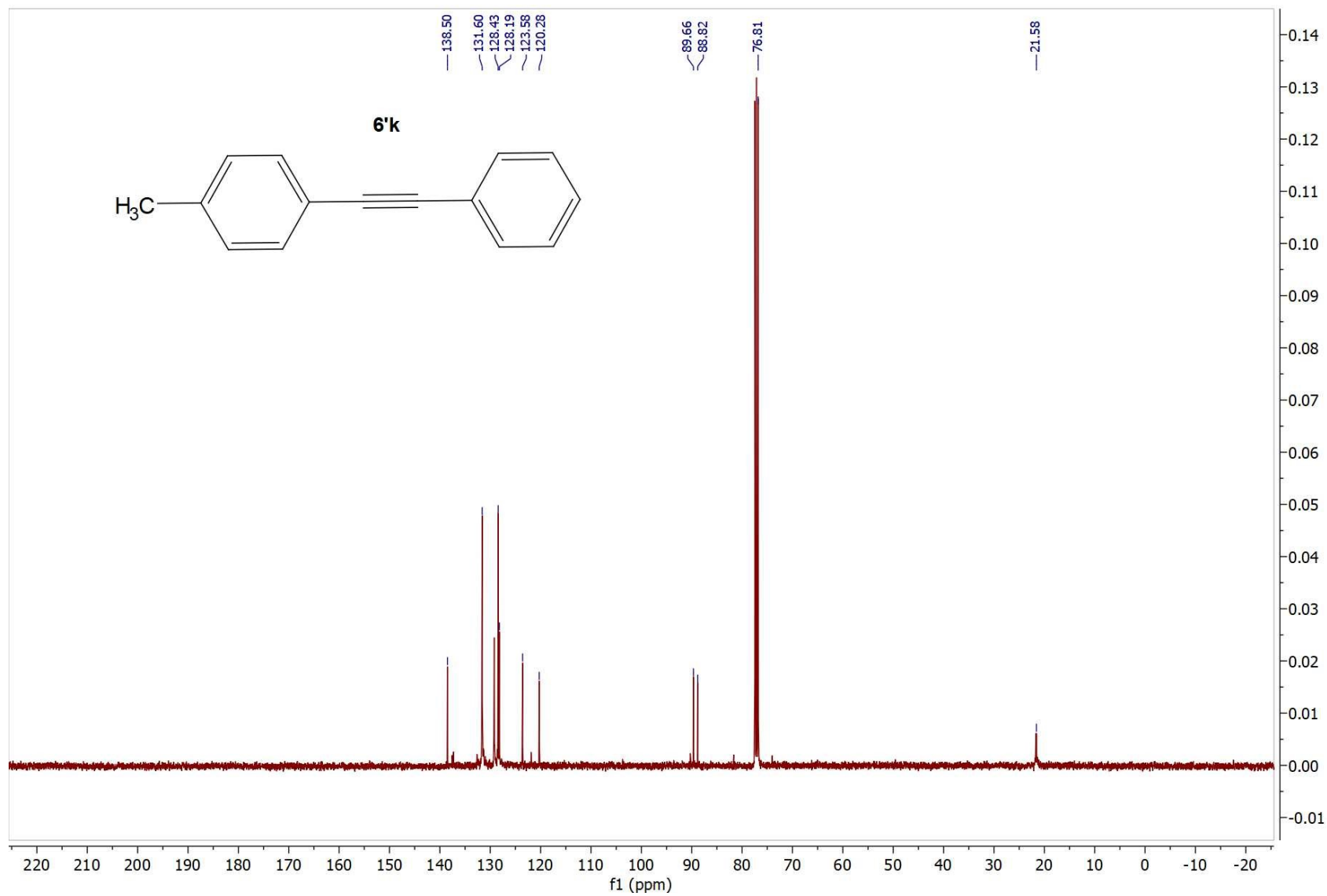
¹³C NMR (101 MHz, CHLOROFORM-D) δ 149.89, 131.98, 129.31, 128.60, 125.63, 122.19, 94.04, 86.75.



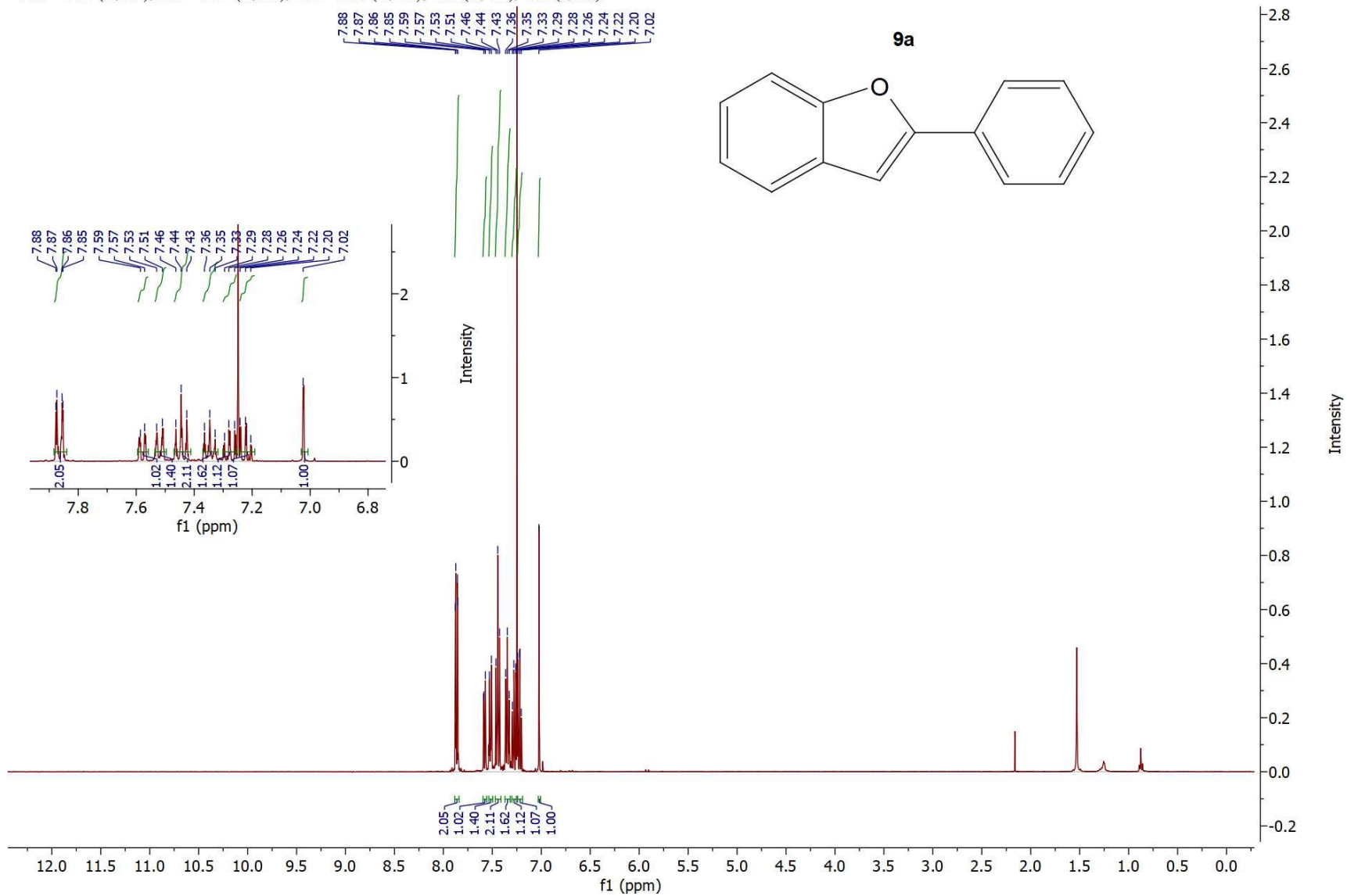
^1H NMR (400 MHz, Chloroform- d) δ 7.56 – 7.50 (m, 2H), 7.47 – 7.38 (m, 2H), 7.37 – 7.27 (m, 3H), 7.20 – 7.10 (m, 2H), 2.37 (s, 3H).



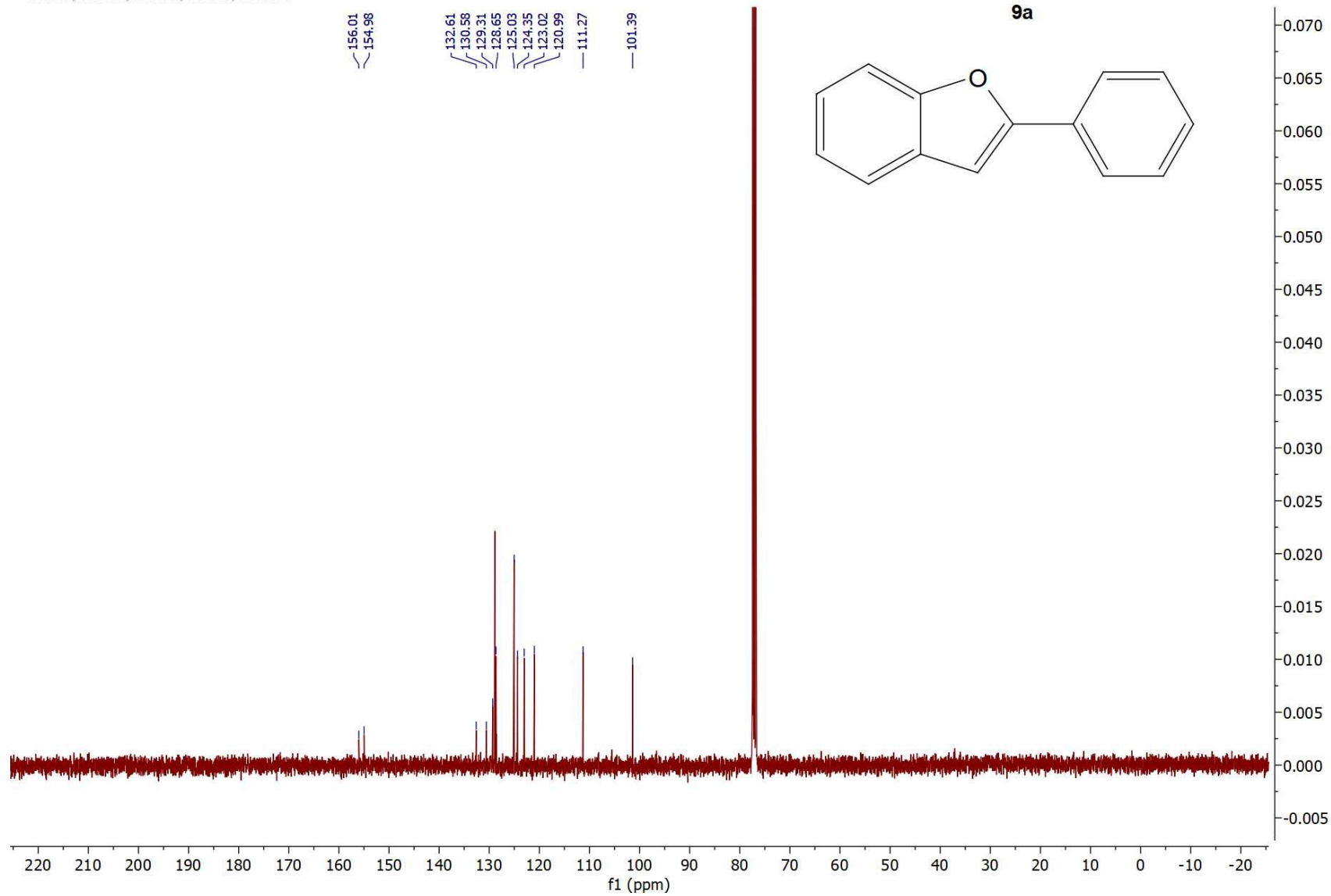
¹³C NMR (101 MHz, Chloroform-*d*) δ 138.50, 131.60, 128.43, 128.19, 123.58, 120.28, 89.66, 88.82, 21.58.



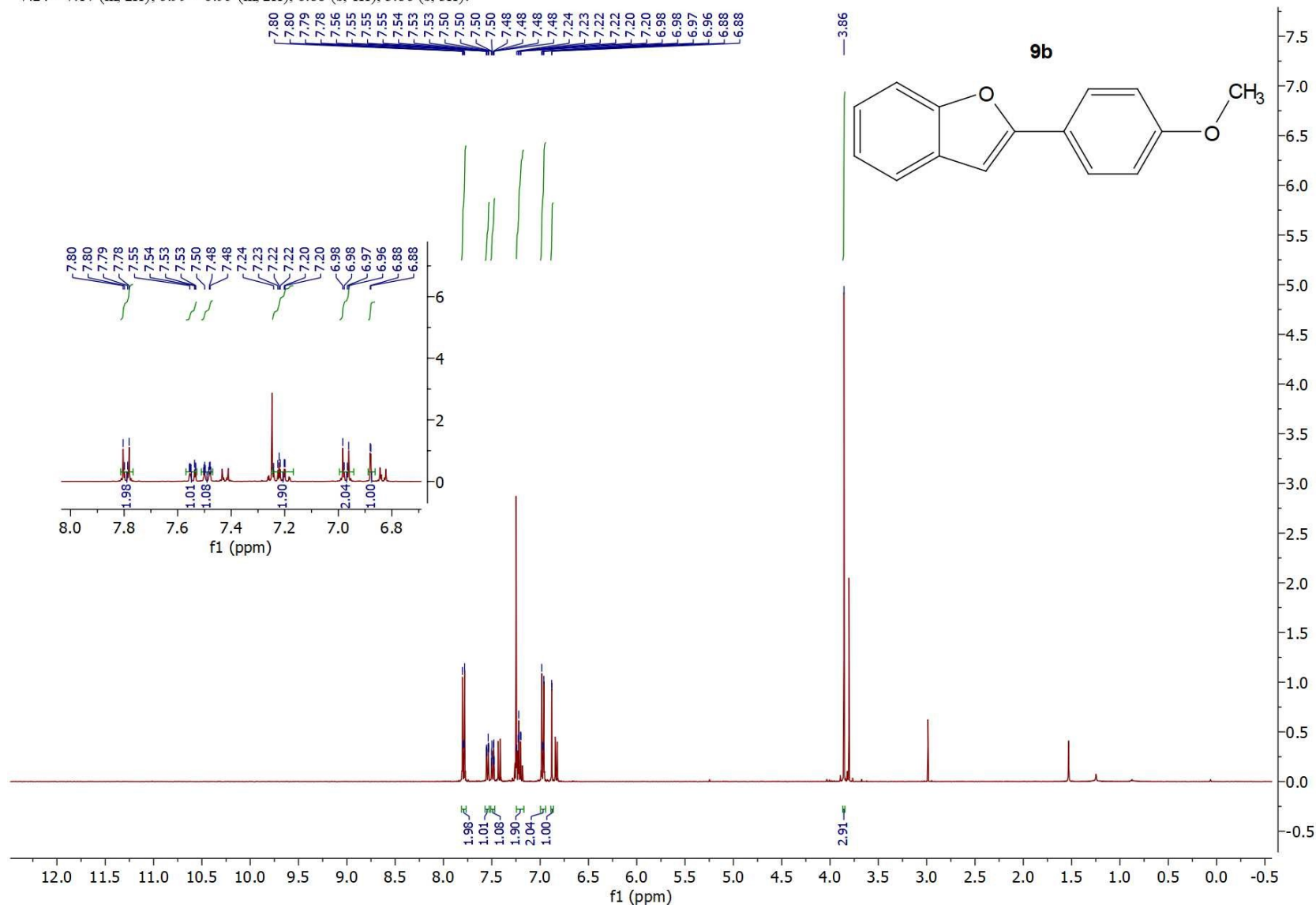
^1H NMR (400 MHz, Chloroform-*d*) δ 7.89 – 7.84 (m, 2H), 7.58 (dd, $J=7.5, 1.5$ Hz, 1H), 7.54 – 7.50 (m, 1H), 7.47 – 7.41 (m, 2H), 7.38 – 7.32 (m, 1H), 7.31 – 7.26 (m, 1H), 7.22 (m, 1H), 7.02 (s, 1H).



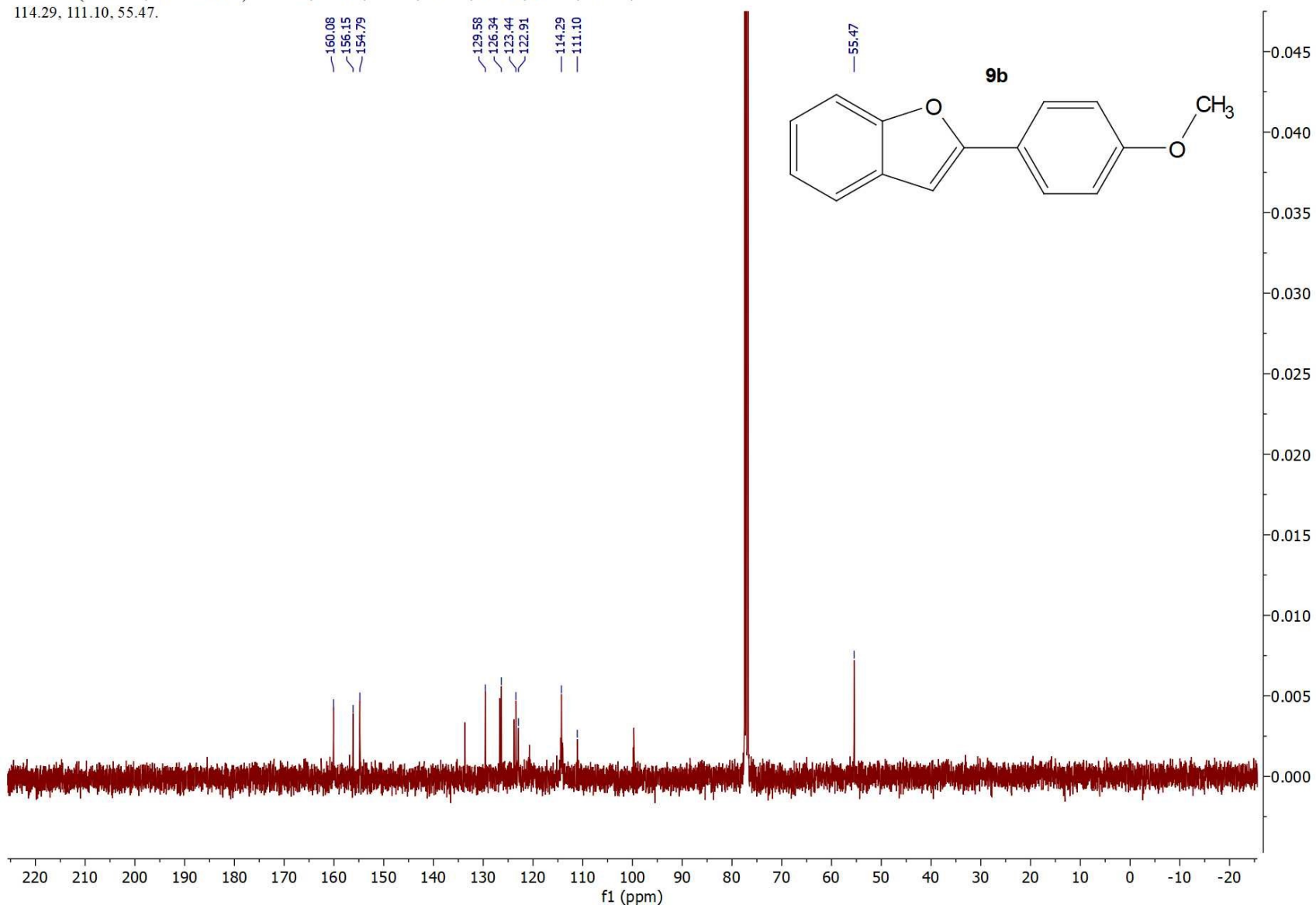
^{13}C NMR (101 MHz, Chloroform-*d*) δ 156.01, 154.98, 132.61, 130.58, 129.31, 128.65, 125.03, 124.35, 123.02, 120.99, 111.27, 101.39.



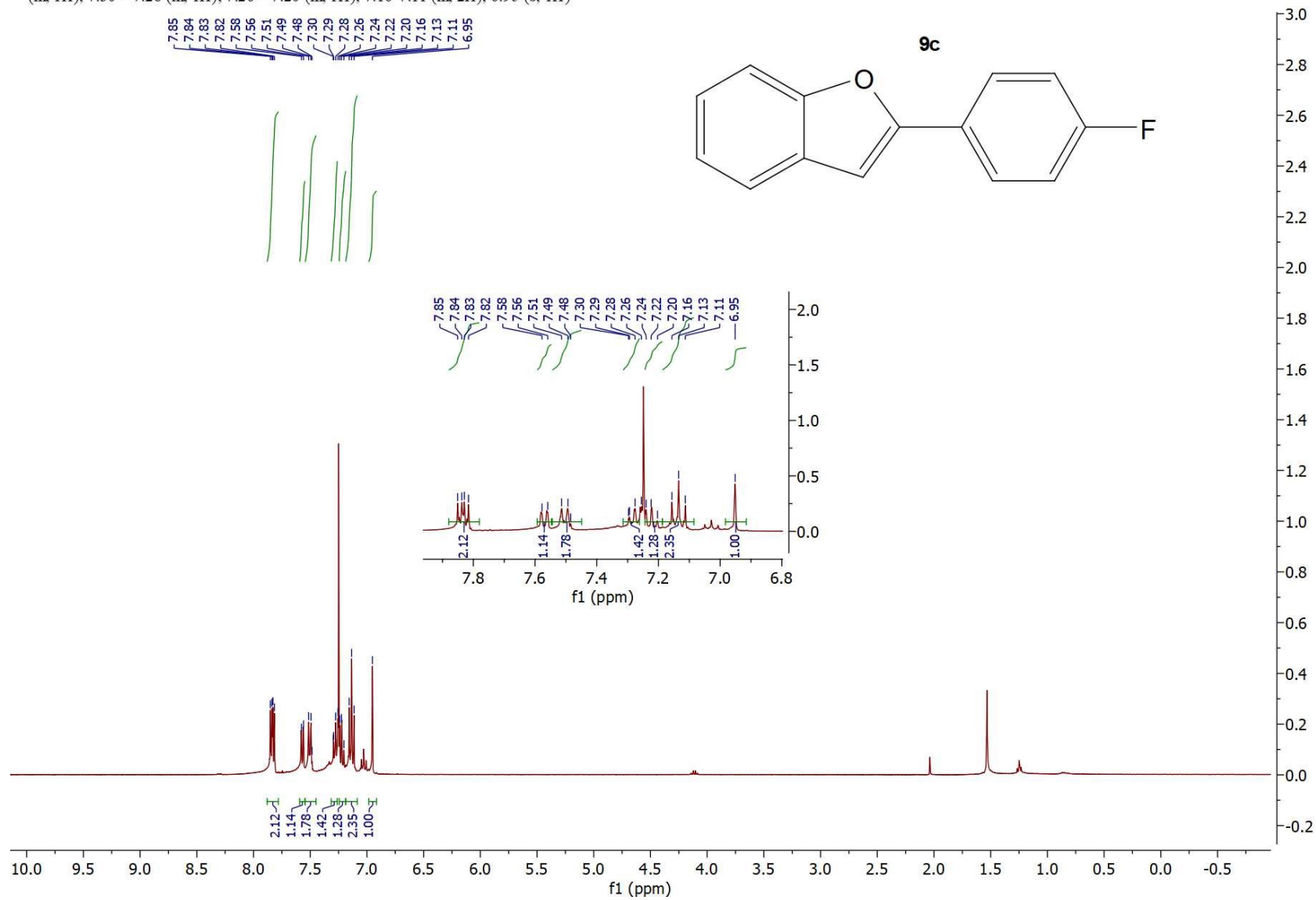
^1H NMR (400 MHz, Chloroform-*d*) δ 7.81 – 7.76 (m, 2H), 7.56 – 7.52 (m, 1H), 7.50-7.48 (m, 1H), 7.24 – 7.17 (m, 2H), 6.99 – 6.95 (m, 2H), 6.88 (s, 1H), 3.86 (s, 3H).



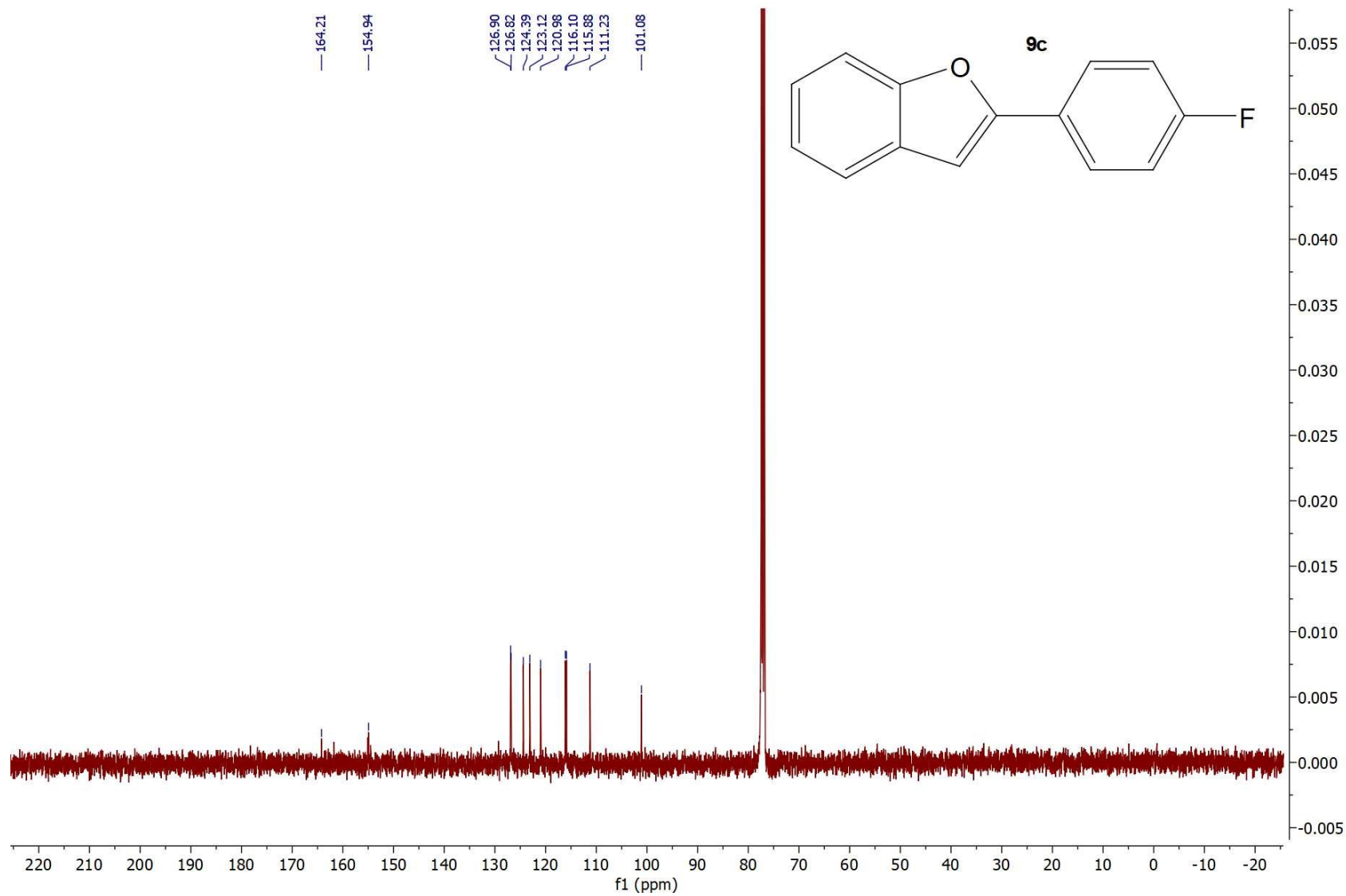
^{13}C NMR (101 MHz, Chloroform-*d*) δ 160.08, 156.15, 154.79, 129.58, 126.34, 123.44, 122.91, 114.29, 111.10, 55.47.



^1H NMR (400 MHz, Chloroform-*d*) δ 7.83 (dd, $J = 8.8, 5.3$ Hz, 2H), 7.57 (d, $J = 7.4$ Hz, 1H), 7.51 – 7.48 (m, 1H), 7.30 – 7.28 (m, 1H), 7.26 – 7.20 (m, 1H), 7.16-7.11 (m, 2H), 6.95 (s, 1H)



^{13}C NMR (101 MHz, Chloroform- d) δ 164.21, 154.94, 126.90, 126.82, 124.39, 123.12, 120.98, 116.10, 115.88, 111.23, 101.08, 77.42.



S8. References

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