Supporting Information

Total Synthesis of Aspidostomide G from a Brominated Tryptamine

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A. General Information

Unless otherwise noted, all reagents and solvents were obtained commercially and used without further purification. 1 H and 13 C NMR spectra were recorded on a Jeol-400 MHz spectrometer. Chloroform-d (δ = 7.24) or DMSO- d_6 (δ = 2.49) or acetone- d_6 (δ = 2.05) was used as internal standard in 1 H NMR spectra. The center peak of chloroform-d (δ = 77.0) or DMSO- d_6 (δ = 39.5) or acetone- d_6 (δ = 29.9) was used as internal standard in 13 C NMR spectra. High-resolution mass spectrometry (HRMS) analyses were determined on a Thermo Scientific Orbitrap LTQ XL mass spectrometer. Melting points were measured on a melting point apparatus with a capillary melting point tube. Thin-layer chromatography (TLC) plates were visualized by exposure to ultraviolet light at 254 nm and/or immersion in a staining solution (phosphomolybdic acid, potassium permanganate, ninhydrin, or p-anisaldehyde) followed by heating on a hot plate. Flash chromatography was carried out utilizing silica gel 60, 70-230 mesh ASTM.

B. Experimental section

OH
$$NH_2$$
 CH_3I, K_2CO_3 NH_2 NO_2 NO_2 NO_2

Compound 19.^{S1}

To the flask was added 2-amino-3-nitrophenol (13) (15.4 g, 100 mmol), potassium carbonate (27.6 g, 200 mmol, 2 equiv) and DMF (100 mL). The mixture was then stirred at room temperature for 3 h. Once the reaction was complete (monitored by TLC), ice water (500 mL) was poured into the mixture and the product was precipitated. The solid was collected by filtration through a Büchner funnel to give the title compound as a yellow solid (15.3 g, 91%).

R_f: 0.2 (hexanes/EtOAc = 3:1). M.p. 62.5–63.2 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.64 (dd, J = 8.9, 1.2 Hz, 1H), 6.82 (d, J = 7.7 Hz, 1H), 6.53 (dd, J = 8.9, 7.7 Hz, 1H), 6.40 (s, 2H), 3.85 (s, 3H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 148.0, 136.9, 131.3, 117.0, 114.4, 113.1, 56.1. HRMS (EI): calculated for C₇H₈N₂O₃ (M⁺), 168.0535, found 168.0543.

$$\begin{array}{c|c} OCH_3 & OCH_3 \\ \hline NH_2 & Br_2, NaOAc \\ \hline NO_2 & HOAc, rt \\ \end{array}$$

Compound 20.^{S2}

To the flask was added compound 19 (16.8 g, 100 mmol), sodium acetate (16.4 g, 200 mmol, 2 equiv) and acetic acid (100 mL). The mixture was stirred for 10 min until all solids were dissolved. A solution of

bromine (5.1 mL, 100 mmol, 1 equiv) in acetic acid (100 mL) was added dropwise into the previous mixture in an ice-water bath. The reaction was warmed to room temperature and stirred for an additional hour. Ice water (500 mL) was poured into the mixture and the product precipitated. The solid was collected by filtration through a Büchner funnel to give the title compound as a red solid (23.6 g, 96%).

R_f: 0.45 (hexanes/EtOAc = 3:1). M.p. 126–127.4 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.85 (d, J = 2.1 Hz, 1H), 6.90 (d, J = 1.9 Hz, 1H), 6.43 (s, 2H), 3.89 (s, 3H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 148.6, 136.3, 131.4, 119.3, 116.5, 106.1, 56.6. HRMS (EI): calculated for C₇H₇BrN₂O₃ (M⁺), 245.9640, found 245.9639.

$$\begin{array}{c} \text{OCH}_3 & \text{1. NaNO}_2, \text{H}_2\text{SO}_4, \text{NaI} \\ \text{NH}_2 & \underline{\text{DMSO/H}_2\text{O}, 0 °C} \\ \text{NO}_2 & 100 °C \\ \end{array} \begin{array}{c} \text{OCH}_3 \\ \text{2. Fe, NH}_4\text{Cl, MeOH/H}_2\text{O} \\ \text{Br} & \text{NH}_2 \\ \end{array}$$

Compound 18.

Compound **20** (37.1 g, 150 mmol) was added to a mixture of DMSO (300 mL) and H₂SO₄ (240 mL, 30% aqueous solution). The resulting mixture was stirred in an ice-water bath for 5 min. The mixture was then treated with a solution of sodium nitrite (31.0 g, 450 mmol, 3 equiv) in water (40 mL). The reaction was stirred at 0 °C for 1.5 h and then treated, in one portion, with a solution of sodium iodide (67.5 g, 450 mmol, 3 equiv) in water (40 mL). After the completion of the reaction by TLC analysis (another 1.5 h), the mixture was then extracted with Et₂O (500 mL × 3). The collected organic layers were washed with saturated aqueous sodium thiosulfate, brine then dried over anhydrous MgSO₄ and concentrated under reduced pressure to give the crude iodide **21** as an orange-brown solid.

Without any purification, the crude iodide was dissolved in MeOH/H₂O (500 mL, 1/1 v/v) and iron (41.9 g, 750 mmol, 5 equiv) and ammonium chloride (40.1 g, 750 mmol, 5 equiv) were added respectively. After stirring under refluxing for 12 h, the reaction mixture was filtered and the filtrate was concentrated. After the dilution of the residue with EtOAc (500 mL), the mixture was washed with brine. The separated aqueous layer was further extracted with EtOAc. The combined organic layers were dried over anhydrous MgSO₄, filtered and concentrated under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate = 40/1] to afford the title compound (38.9 g, 79% over 2 steps) as a pale yellow solid.

R_f: 0.7 (hexanes/EtOAc = 3:1). M.p. 85.9–87.2 °C. ¹H NMR (400 MHz, CDCl₃) δ 6.55 (d, J = 1.9 Hz, 1H), 6.33 (d, J = 1.9 Hz, 1H), 4.26 (s, 2H), 3.82 (s, 3H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 159.0, 148.8, 123.1, 110.2, 103.9, 73.9, 56.5. HRMS (EI): calculated for C₇H₇BrINO (M⁺), 326.8756, found 326.8752.

$$\begin{array}{c} OCH_3 \\ I \\ Br \end{array} \underbrace{TMSA, PdCl_2(Ph_3P)_2, CuI}_{NH_2} \\ \hline Et_2NH, THF, rt \\ Br \\ \hline \\ 23a \end{array}$$

Compound 23a.

A mixture of PdCl₂(PPh₃)₂ (970 mg, 1.38 mmol, 0.069 equiv) and CuI (520 mg, 2.8 mmol, 0.14 equiv) in a flask was vacuumed and then charged with an atmosphere of nitrogen. To this flask were sequentially added anhydrous THF (15 mL), Et₂NH (15 mL), and compound **18** (6.54 g, 20 mmol). The mixture was stirred for 10 min, followed by the addition of TMSA (trimethylsilylacetylene, 3.4 mL, 24 mmol, 1.2 equiv), and then stirred for 16 h under a mixed atmosphere of nitrogen and hydrogen [4/1 (v/v)]. The resulting mixture was filtrated, and the volatiles were removed under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 30/1] to afford the title compound (5.55 g, 93%) as a yellow oil.

 R_f : 0.75 (hexanes/EtOAc = 3:1). ¹H NMR (400 MHz, CDCl₃) δ 6.49 (d, J = 1.5 Hz, 1H), 6.35 (d, J = 1.4 Hz, 1H), 4.34 (s, 2H), 3.81 (s, 3H), 0.26 (s, 9H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 161.3, 150.4, 123.8, 110.0, 105.1, 103.7, 97.3, 96.2, 56.0, 0.2. HRMS (EI): calculated for $C_{12}H_{16}BrNOSi$ (M⁺), 297.0185, found 297.0190.

Compound 22.

Compound **23a** (5.55 g, 18.6 mmol) was dissolved in THF (85 mL) and TBAF (1M solution in THF, 19 mL, 18.6 mmol, 1 equiv) was added. After stirring at room temperature for 0.5 h, the mixture was concentrated and the residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 40/1] to afford the title compound (3.99 g, 95%) as a brown solid.

 R_f : 0.6 (hexanes/EtOAc = 3:1). M.p. 83.2–84.6 °C. ¹H NMR (400 MHz, CDCl₃) δ 6.51 (d, J = 1.6 Hz, 1H), 6.39 (d, J = 1.6 Hz, 1H), 4.34 (s, 2H), 3.84 (s, 3H), 3.64 (s, 1H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 161.7, 150.6, 124.1, 110.1, 103.7, 94.9, 87.3, 76.4, 56.0. HRMS (EI): calculated for C₉H₈BrNO (M⁺), 224.9789, found 224.9797.

$$\begin{array}{c|c} OCH_3 & OCH_3 \\ \hline DMSO, 120 \ ^{\circ}C \\ \hline \end{array}$$

Compound 9.S3

To a solution of compound **22** (5.94 g, 26.3 mmol) in anhydrous DMSO (53 mL) was added sodium amide (5.12 g, 131 mmol, 5 equiv). The reaction mixture was stirred at 120 °C for 16 h. After cooling to room temperature, the reaction mixture was then poured into water (50 mL) and extracted with Et₂O (100 mL × 3). The combined organic layers were washed with saturated aqueous NaHCO₃, dried over anhydrous MgSO₄ and concentrated under reduced pressure to afford the title compound (5.23 g, 88%) as a white solid. R_f: 0.6 (hexanes/EtOAc = 3:1). M.p. 85.3–87.4 °C. ¹H NMR (400 MHz, CDCl₃) δ 8.10 (s, 1H), 7.11 (t, J = 1.0 Hz, 1H), 7.00 (dd, J = 3.2, 2.5 Hz, 1H), 6.69 (d, J = 1.2 Hz, 1H), 6.64 (t, J = 2.3 Hz, 1H), 3.93 (s, 3H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 153.3, 137.2, 123.1, 117.4, 115.5, 107.5, 103.7, 99.8, 55.4. HRMS (EI): calculated for C₉H₈BrNO (M⁺), 224.9789, found 224.9783.

$$\begin{array}{c} \text{OCH}_3\\ \text{Br} & \begin{array}{c} 1. \text{ POCI}_3, \text{ DMF, } 0 \text{ to } 40 \text{ }^{\circ}\text{C} \\ \hline 2. \text{ CH}_3\text{NO}_2, \text{NH}_4\text{OAc, } 100 \text{ }^{\circ}\text{C} \\ \end{array} \\ \begin{array}{c} \text{Br} & \begin{array}{c} \text{NO}_2\\ \text{N}\\ \text{H} \end{array} \end{array}$$

Compound 17.^{S3b}

A solution of POCl₃ (1.24 mL, 13.3 mmol, 1.3 equiv) in anhydrous DMF (12 mL) was stirred in an icewater bath for 30 min. Then a solution of compound **9** (2.3 g, 10.2 mmol) in anhydrous DMF (12 mL) was added dropwise over a period of 10 min to the above mixture. The resulting mixture was stirred for 30 min at the same temperature, then heated to 40 °C for 1 h. The reaction mixture was then chilled in an ice-water bath, and 1N aqueous NaOH solution (24 mL) was added until the mixture became basic. The mixture was placed at low temperature (5 °C) for 20 min. 1N aqueous HCl solution (24 mL) was added until the mixture became acidic, and then was extracted with EtOAc (50 mL × 3). The combined organic layers were dried over anhydrous MgSO₄ and concentrated under reduced pressure to afford the crude product, which was purified by column chromatography [silica gel, hexanes/ethyl acetate 40/1] to afford the aldehyde (2.33 g, 90%) as white solid.

To a solution of the aldehyde intermediate (3.64 g, 14.3 mmol) in nitromethane (29 mL) was added ammonium acetate (1.10 g, 14.3 mmol, 1 equiv) at room temperature. The reaction mixture was then heated to 100 °C for 2 h, then chilled in an ice-water bath for 12 h. The precipitate was filtered, and the solid was washed with toluene (10 mL). The combined organic layers were concentrated to give the title compound quantitatively (4.25 g) as a brown solid.

R_f: 0.3 (hexanes/EtOAc = 3:1). M.p. 226.3–227.9 °C. ¹H NMR (400 MHz, DMSO- d_6) δ 8.49 (d, J = 13.3 Hz, 1H), 8.25 (s, 1H), 8.07 (d, J = 13.3 Hz, 1H), 7.28 (d, J = 1.4 Hz, 1H), 6.86 (d, J = 1.2 Hz, 1H), 3.96 (s, 3H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ 154.1, 139.0, 134.7, 132.9, 132.4, 116.2, 114.4, 108.7, 107.9, 106.0, 56.0. HRMS (EI): calculated for C₁₁H₉O₃N₂Br (M⁺), 294.9797, found 294.9793.

$$\begin{array}{c|c} & \text{NO}_2 & \text{NH}_2 \\ & \text{OCH}_3 & \text{OCH}_3 & \text{OCH}_3 \\ & & \text{THF, 70 °C} & \text{Br} & \text{N} \\ & & \text{15} & \text{H} \end{array}$$

Compound 15.S3b

To a solution of compound 17 (1.54 g, 5.19 mmol) in THF (45 mL) in an ice-water bath was slowly added LiAlH₄ (1.18 g, 31.1 mmol, 6 equiv). The reaction mixture was then heated to 70 °C for 3.5 h. After the reaction, the resulting mixture was chilled in an ice-water bath, and excess LiAlH₄ was quenched by the dropwise addition of water (1 mL), 6N aqueous KOH solution (1 mL) and water (2 mL) until no further gas evolution was observed. The mixture was diluted with EtOAc (150 mL), filtered through Celite, dried over anhydrous MgSO₄ and concentrated to afford the title compound quantitatively (1.40 g) as a brown oil.

 R_f : 0.1 (DCM/MeOH = 10:1). ¹H NMR (400 MHz, DMSO- d_6) δ 10.90 (s, 1H), 7.09 (d, J = 1.5 Hz, 1H), 6.97 (s, 1H), 6.54 (d, J = 1.5 Hz, 1H), 3.84 (s, 3H), 2.85–2.75 (m, 4H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ 154.6, 138.2, 122.3, 115.9, 113.8, 112.7, 107.5, 102.4, 55.4, 42.7, 30.0. HRMS (EI): calculated for $C_{11}H_{13}BrN_2O$ (M⁺), 268.0211, found 268.0201.

Scheme S1. Synthetic route of the trichloromethyl ketone 16

$$\begin{array}{c}
Cl & Cl \\
Cl & Cl
\end{array}$$

$$\begin{array}{c}
Cl_3C \\
N \\
N \\
S1
\end{array}$$

Compound S1.S4

To a solution of pyrrole (3 mL, 43.2 mmol) in anhydrous ether (50 mL) in an ice-water bath was added a solution of trichloroacetyl chloride (5.3 mL, 47.6 mmol, 1.1 equiv) in anhydrous ether (20 mL). The reaction mixture was then stirred at room temperature for 24 h. After the reaction, the reaction mixture was treated with saturated aqueous sodium bicarbonate solution (200 mL). The mixture was extracted with EtOAc (200 mL × 2). The combined organic layers were washed with brine (200 mL), dried over anhydrous

MgSO₄, and concentrated under reduced pressure to afford the title compound quantitatively (10.0 g) as a black solid.

R_f: 0.5 (hexanes/EtOAc = 3:1). ¹H NMR (400 MHz, CDCl₃) δ 9.48 (s, 1H), 7.37 (s, 1H), 7.15 (s, 1H), 6.38–6.36 (m, 1H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 173.3, 127.4, 122.9, 121.3, 111.8, 94.9. HRMS (APCI) m/z: [M – H]⁺ calculated for C₆H₃ONCl₃ 209.9275, found 209.9282.

$$\begin{array}{c|c} Cl_3C & & Br_2 \\ \hline N & HOAc, 0 \ ^{\circ}C \ to \ rt \\ \hline S1 & & 16 \\ \end{array}$$

Compound 16.^{S5}

To a solution of compound S1 (10.0 g, 47.1 mmol) in acetic acid (50 mL) in an ice-water bath was slowly added bromine (5.6 mL, 109 mmol, 2.3 equiv). After addition, the reaction mixture was then stirred at room temperature for 20 h. The reaction was quenched with saturated aqueous sodium hyposulfite solution (100 mL). The resulting mixture was extracted with EtOAc (200 mL × 2). The combined organic layers were washed with saturated aqueous sodium bicarbonate solution (200 mL) and brine (200 mL), dried over anhydrous MgSO₄, and concentrated under reduced pressure. The crude mixture was purified by column chromatography [silica gel, hexanes/ethyl acetate 30/1 (v/v)] to afford the title compound (15.8 g, 90%) as a white solid.

 R_f : 0.7 (hexanes/EtOAc = 3:1). ¹H NMR (400 MHz, CDCl₃) δ 9.71 (s, 1H), 7.31 (d, J = 2.9 Hz, 1H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 172.0, 123.8, 123.1, 113.3, 102.4, 93.8. HRMS (APCI) m/z: [M – H]⁺ calculated for C₆HONBr₂Cl₃ 365.7485, found 365.7483.

Compound 14a.

To a solution of compound **15** (1.09 g, 4.06 mmol) and compound **16** (1.5 g, 4.87 mmol, 1.2 equiv) in EtOAc (9 mL) was added Et₃N (1.41 mL, 10.2 mmol, 2.5 equiv) dropwise. The mixture was stirred at room temperature for 12 h. The volatiles were removed under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 10/1] to afford the title compound (1.25 g, 59%) as a pale yellow solid.

R_f: 0.6 (hexanes/EtOAc = 1:1). M.p. 184.1–185.7 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.93 (s, 1H), 8.02 (s, 1H), 7.16 (s, 1H), 6.88 (s, 1H), 6.65 (s, 1H), 6.30 (s, 1H), 6.10 (s, 1H), 3.92 (s, 3H), 3.71–3.67 (m, 2H), 3.11 (t, J = 6.3 Hz, 2H). ¹³C{¹H} NMR (100 MHz, CDCl₃) δ 159.3, 154.4, 138.3, 127.3, 121.6, 116.3, 116.0, 113.6, 111.3, 107.9, 104.9, 104.1, 99.5, 55.6, 41.0, 26.2. HRMS (ESI) m/z: [M + H]⁺ calculated for C₁₆H₁₅O₂N₃Br₃ 517.8709, found 517.8697.

Compound 14b.

To a solution of compound **14a** (950 mg, 1.83 mmol) in DCM (18 mL) in an ice-water bath was added dropwise of BBr₃ (1M solution in DCM, 18 mL, 10 equiv). After addition, the mixture was then stirred at room temperature for 12 h. After the reaction was completed by TLC analysis, the mixture was extracted with DCM (50 mL) and EtOAc (50 mL). The combined organic layers were dried over anhydrous MgSO₄ and concentrated under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 3/1 (v/v)] to afford the title compound (674 mg, 73%) as a dark green solid. R_f: 0.4 (hexanes/EtOAc = 1:1). ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.60 (s, 1H), 10.75 (s, 1H), 9.84 (s, 1H), 8.09 (t, J = 5.4 Hz, 1H), 6.93 (s, 2H), 6.89 (d, J = 2.6 Hz, 1H), 6.45 (d, J = 1.2 Hz, 1H), 3.48 (dd, J = 12.8, 6.8 Hz, 2H), 2.98 (t, J = 7.0 Hz, 2H). ¹³C{¹H} NMR (100 MHz, DMSO-*d*₆) δ 158.7, 152.5, 138.9, 128.4, 121.8, 115.7, 113.6, 112.4, 112.3, 105.9, 105.6, 104.2, 97.7, 40.1, 26.3. HRMS (EI): calculated for C₁₅H₁₂Br₃N₃O₂ (M⁺) 502.8480, found 502.8470.

Compound 14c.

To a solution of compound 14b (200 mg, 390 μ mol) in DCM (790 μ L) was added acetyl chloride (34 μ L, 468 μ mol, 1.2 equiv) and pyridine (48 μ L, 585 μ mol, 1.5 equiv). The mixture was stirred at room temperature for 3 h. The volatiles were removed under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 3/1 (v/v)] to afford the title compound (187 mg, 87%) as a white solid.

R_f: 0.4 (hexanes/EtOAc = 1:1). ¹H NMR (400 MHz, DMSO- d_6) δ 12.65 (d, J = 2.1 Hz, 1H), 11.19 (d, J = 1.6 Hz, 1H), 8.16 (t, J = 5.4 Hz, 1H), 7.42 (d, J = 1.6 Hz, 1H), 7.17 (d, J = 2.5 Hz, 1H), 6.91 (d, J = 1.6 Hz, 1H), 6.89 (d, J = 2.7 Hz, 1H), 3.43 (q, J = 6.8 Hz, 2H), 2.83 (t, J = 7.2 Hz, 2H), 2.36 (s, 3H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ 169.5, 158.9, 143.9, 139.0, 128.4, 124.9, 119.0, 114.8, 112.5, 112.4, 112.0, 110.9, 104.5, 97.7, 40.3, 26.0, 20.8, HRMS (EI): calculated for C₁₇H₁₄Br₃N₃O₃ (M⁺) 544.8585, found 544.8592.

Compound 25a.

To a solution of compound **14a** (200 mg, 385 μmol) in acetic acid (3.9 mL) and formic acid (1.17 mL). Then NBS (69 mg, 385 μmol, 1 equiv) was added in the resulting mixture and stirred at room temperature for 20 min. The volatiles were removed under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 10/1] and then by TLC plate [dichloromethane/hexanes 15/1] for four developments to afford the title compound (99 mg, 43%) as a colorless solid.

 $R_f = 0.65$ (hexanes/EtOAc = 1:1). M.p. 178.3–179.2 °C. ¹H NMR (400 MHz, CDCl₃) δ 9.98 (s, 1H), 8.14 (s, 1H), 7.07 (d, J = 1.2 Hz, 1H), 6.64 (d, J = 1.2 Hz, 1H), 6.27 (s, 1H), 6.09–6.05 (m, 1H), 3.88 (s, 3H), 3.65 (q, J = 6.0 Hz, 2H), 3.06 (t, J = 6.1 Hz, 2H). 13 C{ 1 H} NMR (100 MHz, CDCl₃) δ 159.4, 153.2, 137.7, 127.3, 116.4, 116.2, 112.8, 111.4, 107.6, 107.3, 104.9, 99.5, 55.8, 40.5, 29.7, 25.5. HRMS (ESI) m/z: [M – H] $^{+}$ calculated for C₁₆H₁₂O₂N₃Br₄ 593.7658, found 593.7676.

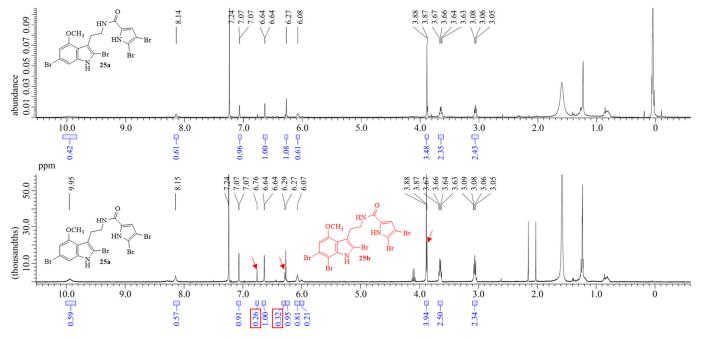


Figure S1. Comparing the clean proton NMR spectrum of 25a with the mixed NMR spectrum of 25a and 25b

Compound 25c.

To a solution of compound 14c (180 mg, 328 µmol) in acetic acid (3.3 mL) and formic acid (0.99 mL). Then NBS (59 mg, 328 µmol, 1 equiv) was added in the resulting mixture and stirred at room temperature for 3 h. The volatiles were removed under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 4/1 (v/v)] to afford the title compound (48 mg, 55%) as a white solid.

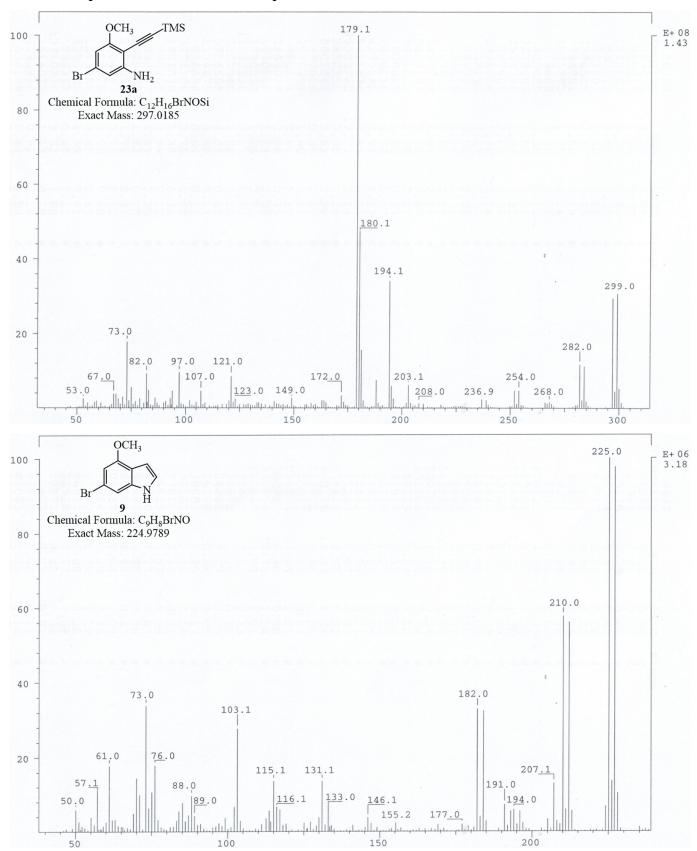
 $R_f = 0.6$ (hexanes/EtOAc = 1:1). ¹H NMR (400 MHz, DMSO- d_6) δ 12.63 (d, J = 2.2 Hz, 1H), 12.12 (s, 1H), 8.20 (t, J = 5.8 Hz, 1H), 7.34 (d, J = 1.6 Hz, 1H), 7.00 (d, J = 1.5 Hz, 1H), 6.86 (d, J = 2.7 Hz, 1H), 2.78 (t, J = 7.0 Hz, 2H), 2.38 (s, 3H). ¹³C{¹H} NMR (100 MHz, DMSO- d_6) δ 170.0 159.4, 143.3, 139.0, 128.8, 119.4, 116.4, 113.5, 113.0, 112.1, 111.8, 110.7, 104.8, 98.2, 40.2, 29.5, 26.3. HRMS (EI): calculated for C₁₇H₁₃Br₄N₃O₃ (M⁺) 622.7690, found 622.7698.

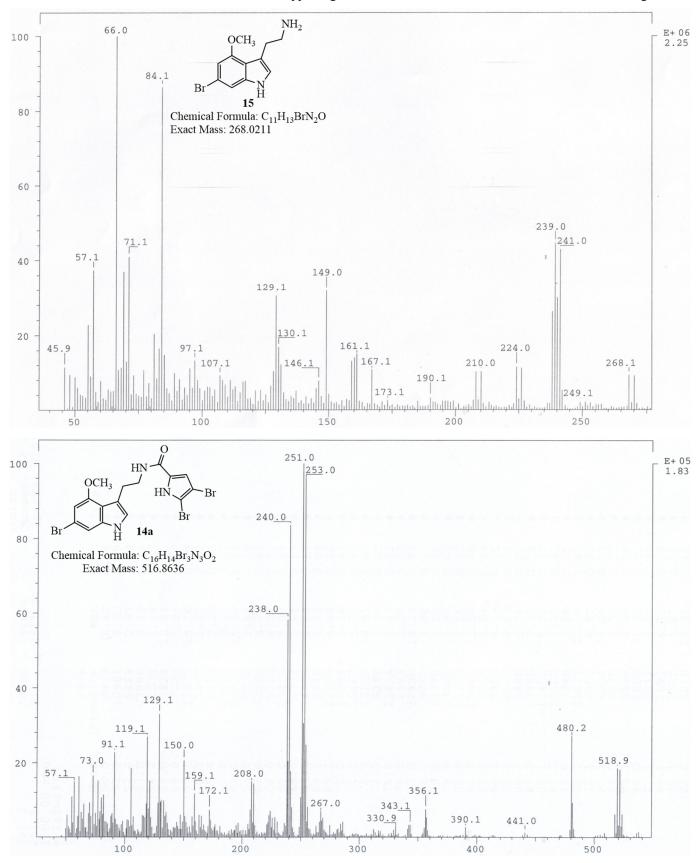
Aspidostomide G (1).^{S6}

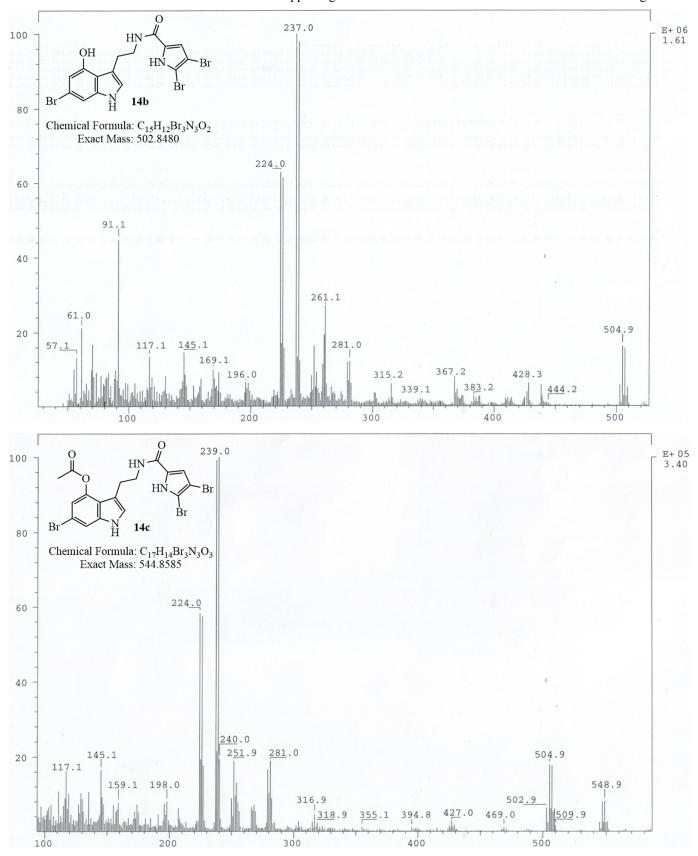
To a solution of compound **25c** (46 mg, 73 μmol) in dry MeOH (730 μL) was added K₂CO₃ (25 mg, 183 μmol, 2.5 equiv). After stirring at 0 °C for 5 h, the reaction was quenched with a mixed solution of saturated aqueous Na₂S₂O₃/saturated aqueous NaHCO₃ (5 mL, 1/1 v/v). The resulting mixture was extracted with EtOAc (5 mL × 3). The combined organic layers were dried over anhydrous MgSO₄ and concentrated under reduced pressure. The residue was purified by column chromatography [silica gel, hexanes/ethyl acetate 4/1 (v/v)] to afford the title compound (33 mg, 77%) as a pale yellow solid.

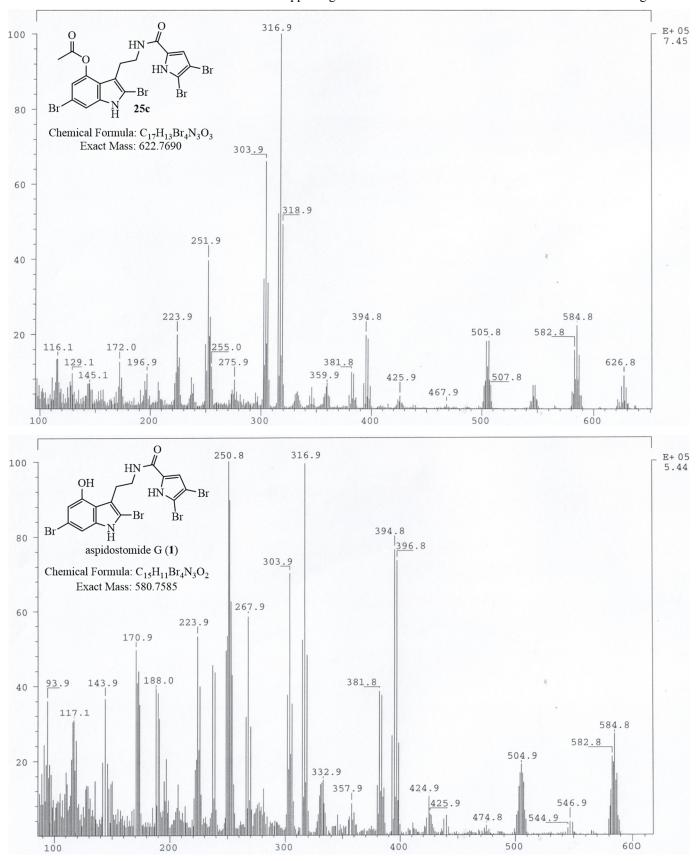
 $R_f = 0.6$ (hexanes/EtOAc = 1:1). ¹H NMR (400 MHz, acetone- d_6) δ 11.63 (s, 1H), 10.54 (s, 1H), 9.07 (s, 1H), 6.90 (d, J = 1.5 Hz, 1H), 6.68 (s, 1H), 6.53 (d, J = 1.5 Hz, 1H), 3.52 (dd, J = 12.4, 6.5 Hz, 2H), 2.95 (t, J = 6.5 Hz, 2H). ¹³C{¹H} NMR (100 MHz, acetone- d_6) δ 159.2, 151.1, 138.9, 128.6, 116.1, 114.9, 112.3, 112.0, 107.8, 107.5, 105.8, 104.1, 98.5, 40.2, 25.8. HRMS (EI): calculated for C₁₅H₁₁Br₄N₃O₂ (M⁺) 580.7585, found 580.7593.

C. LRMS Spectra for brominated compounds

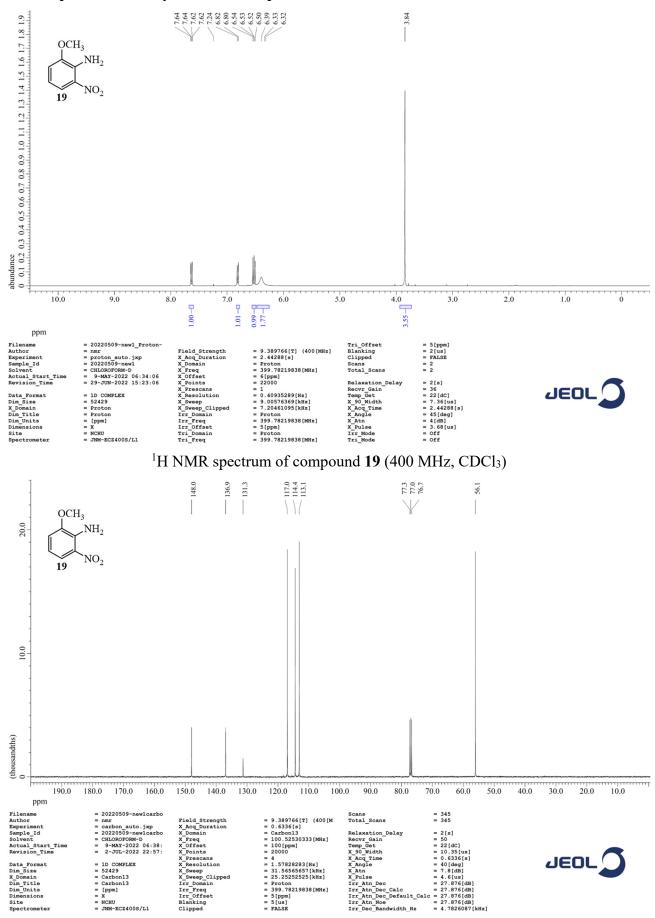




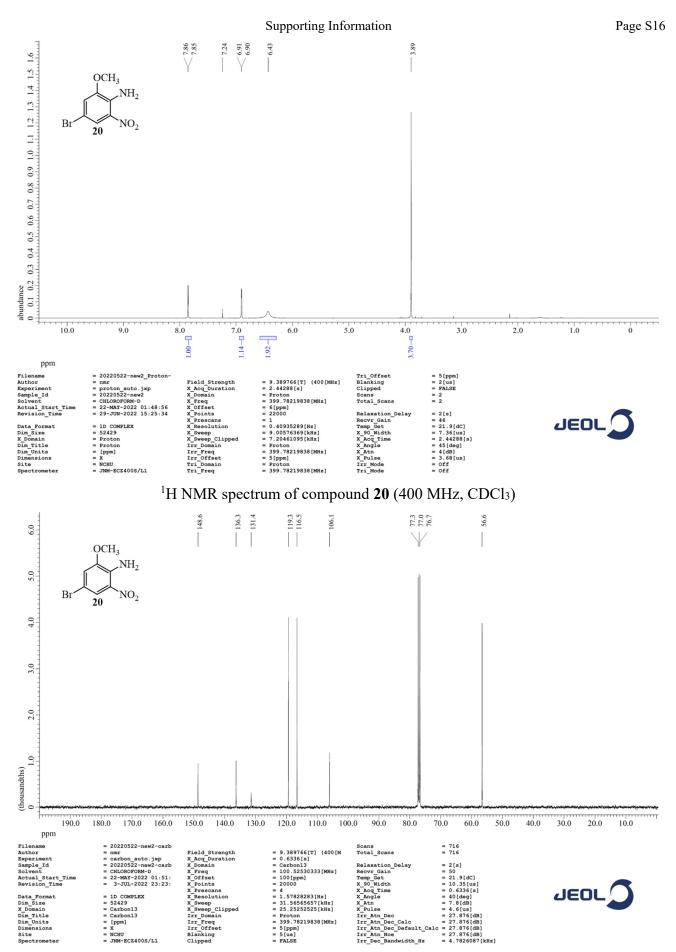




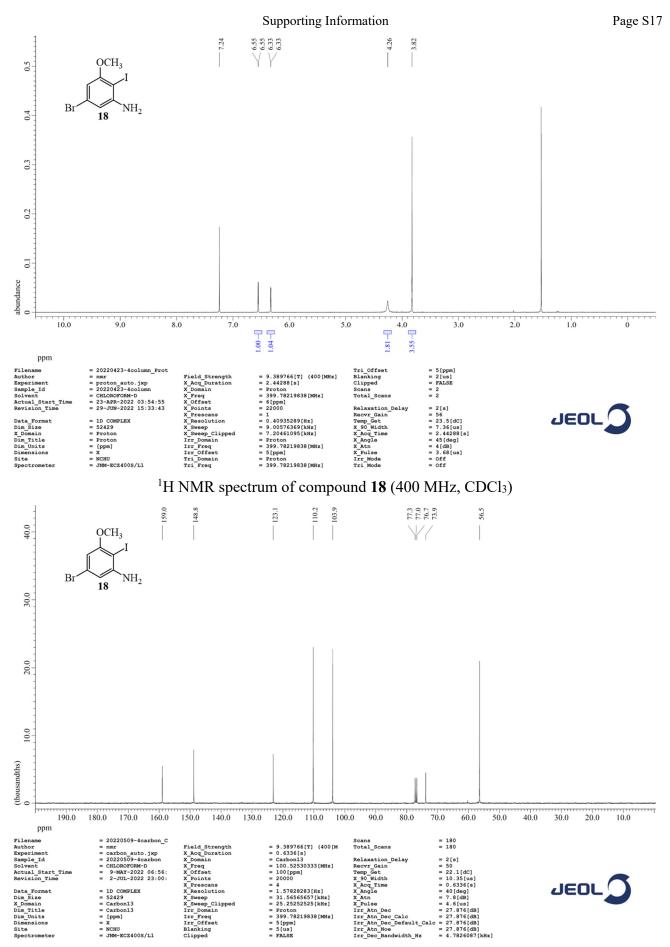
D. NMR Spectra for the synthesized compounds



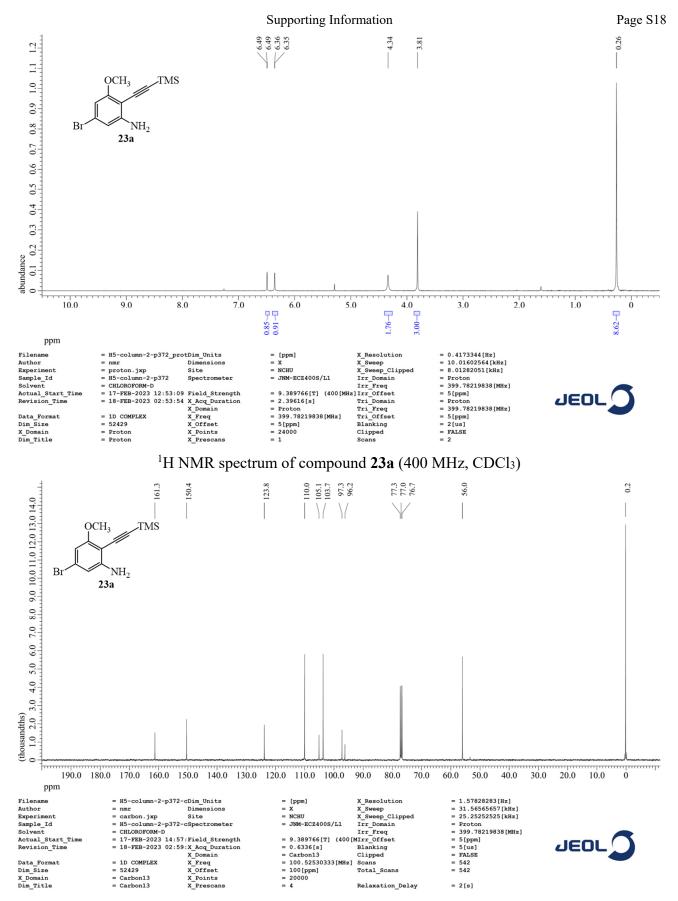
¹³C NMR spectrum of compound **19** (100 MHz, CDCl₃)



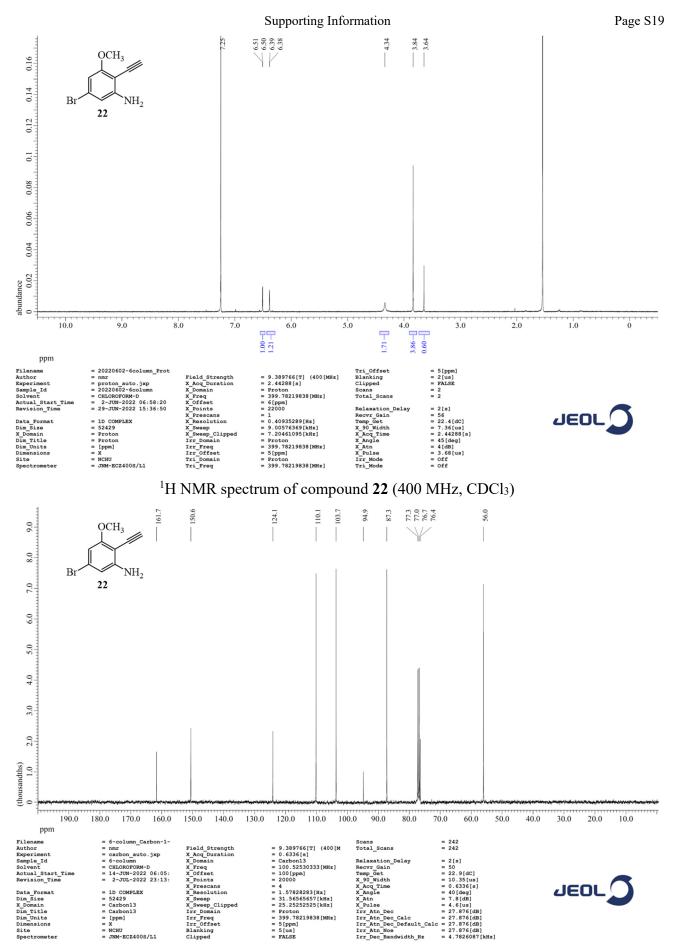
¹³C NMR spectrum of compound **20** (100 MHz, CDCl₃)



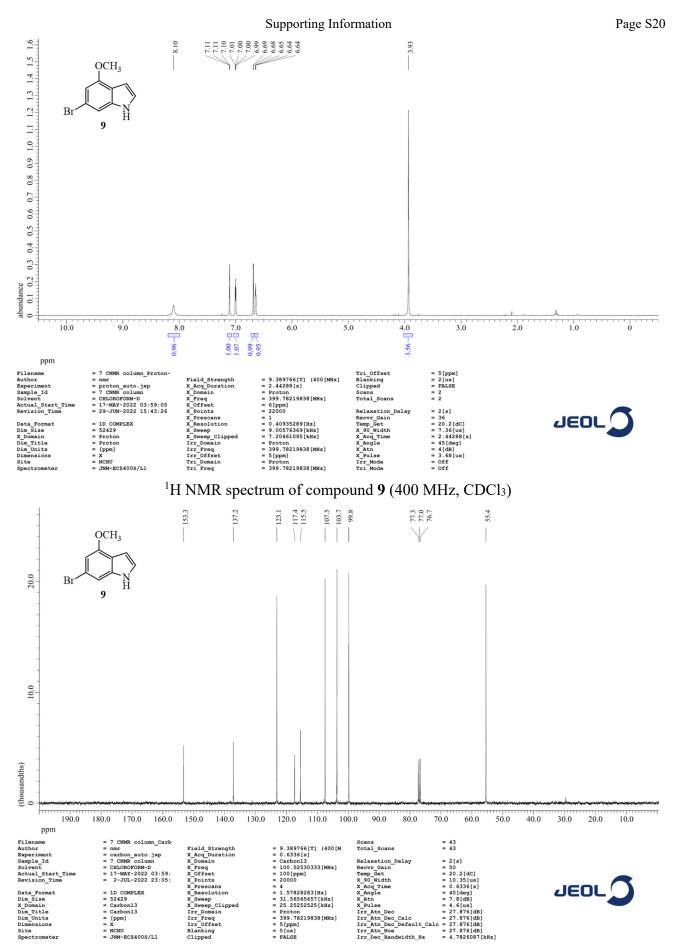
¹³C NMR spectrum of compound **18** (100 MHz, CDCl₃)



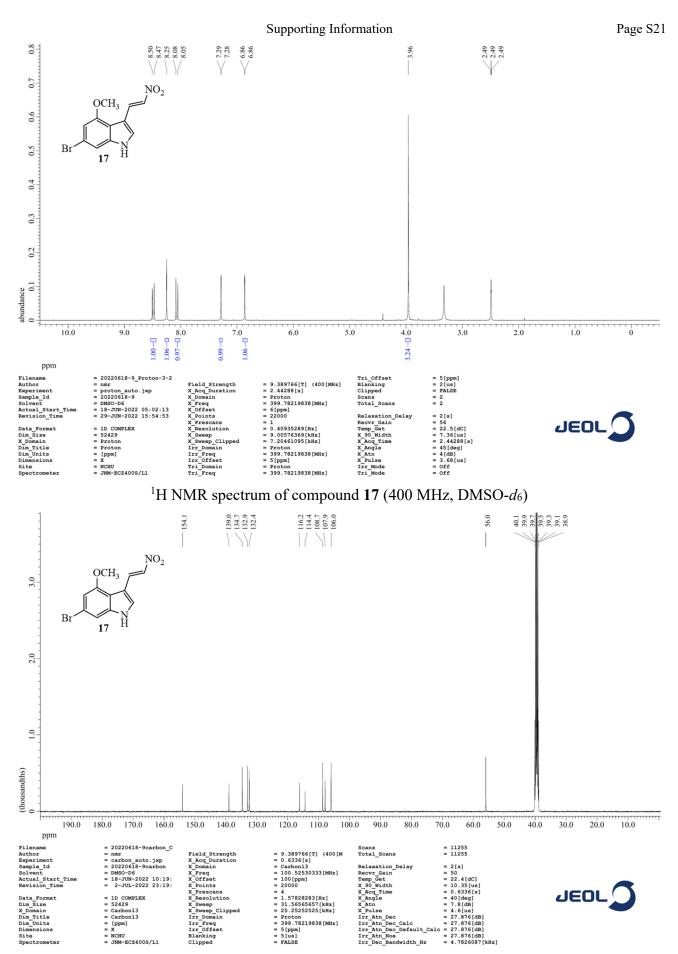
¹³C NMR spectrum of compound **23a** (100 MHz, CDCl₃)



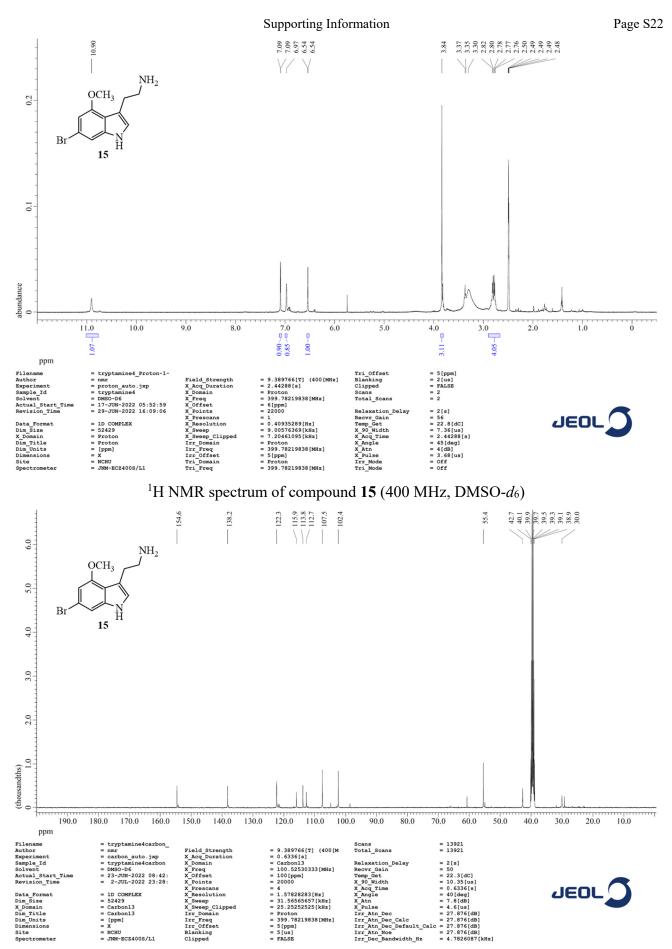
¹³C NMR spectrum of compound **22** (100 MHz, CDCl₃)



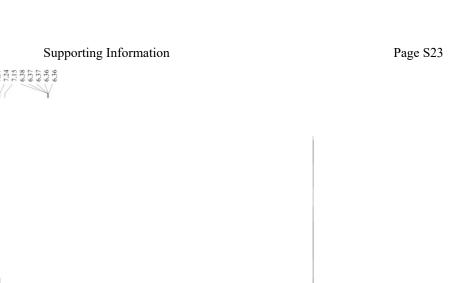
¹³C NMR spectrum of compound 9 (100 MHz, CDCl₃)



¹³C NMR spectrum of compound **17** (100 MHz, DMSO-*d*₆)



¹³C NMR spectrum of compound **15** (100 MHz, DMSO-*d*₆)





S1

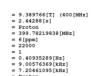
0.4

0.3

0.2

0.1





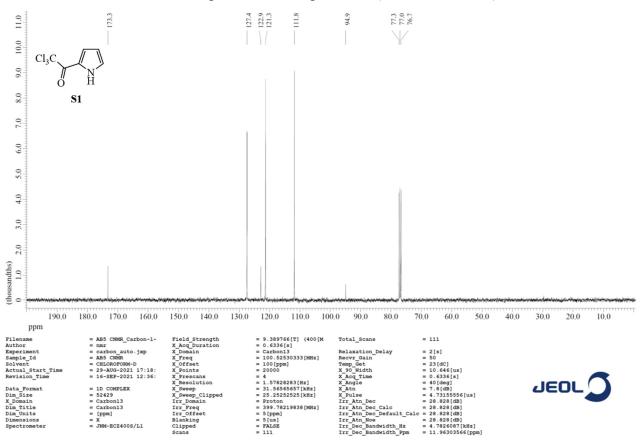
1.00



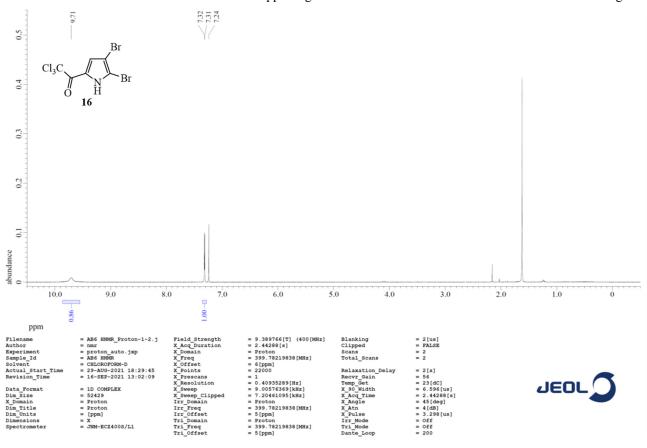


2.0

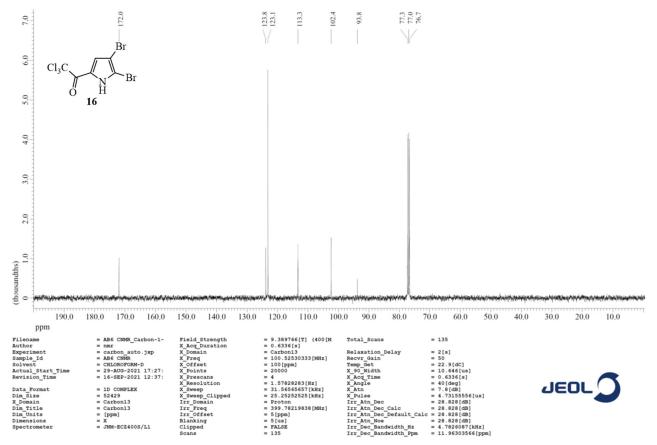
¹H NMR spectrum of compound **S1** (400 MHz, CDCl₃)



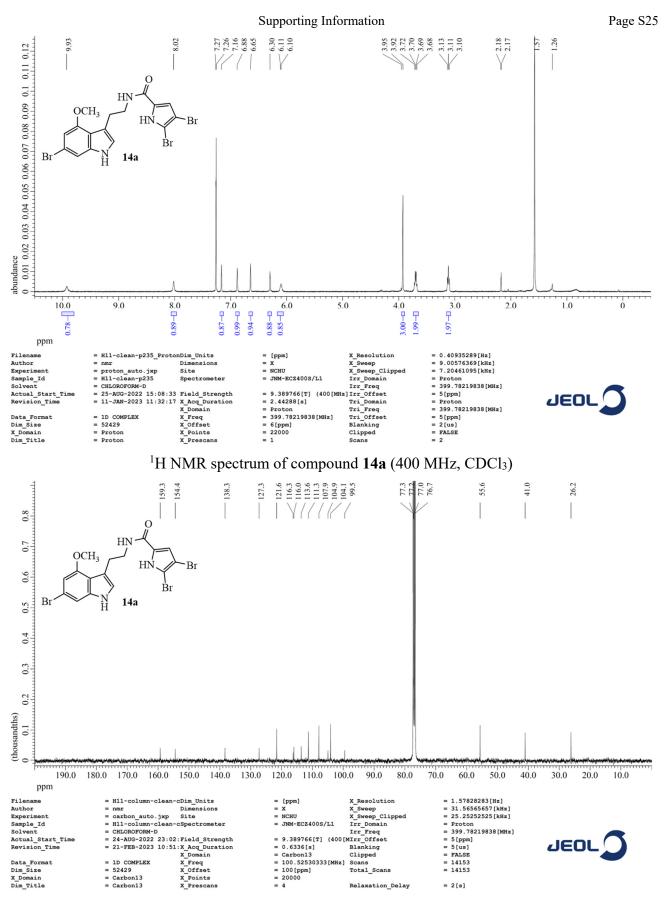
¹³C NMR spectrum of compound **S1** (100 MHz, CDCl₃)



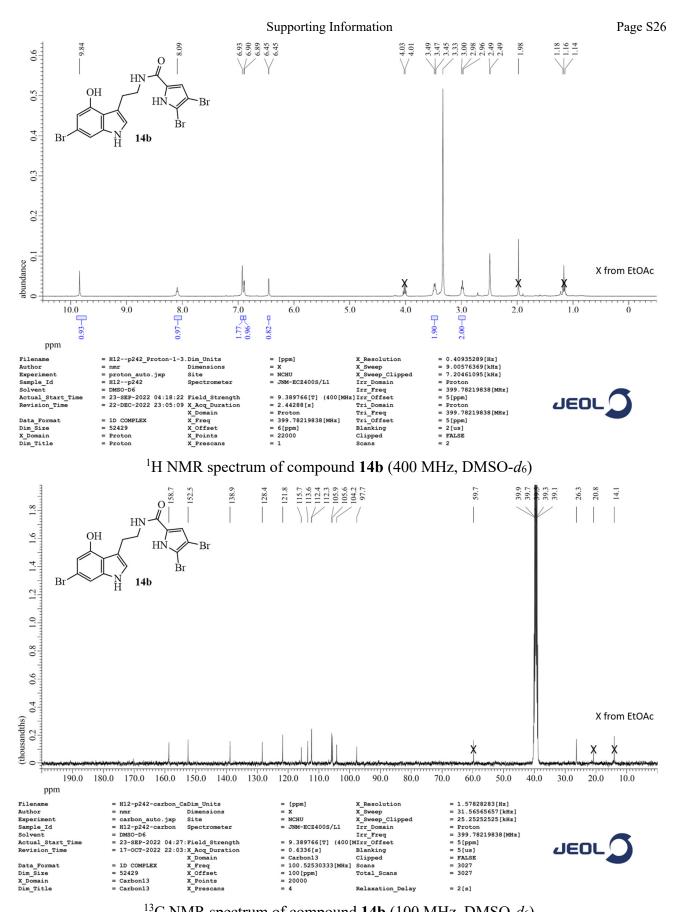
¹H NMR spectrum of compound **16** (400 MHz, CDCl₃)



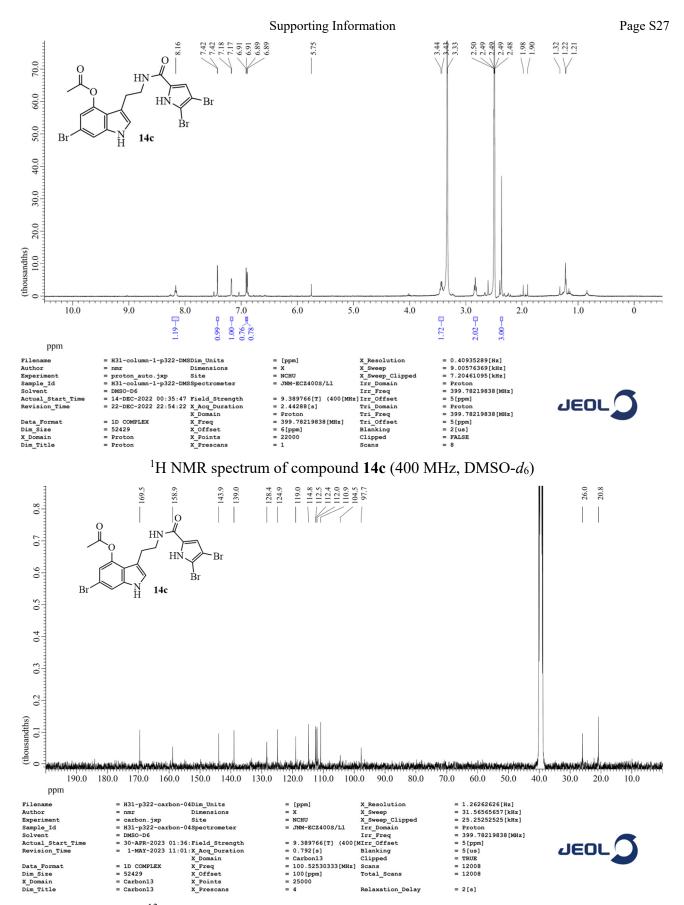
¹³C NMR spectrum of compound **16** (100 MHz, CDCl₃)



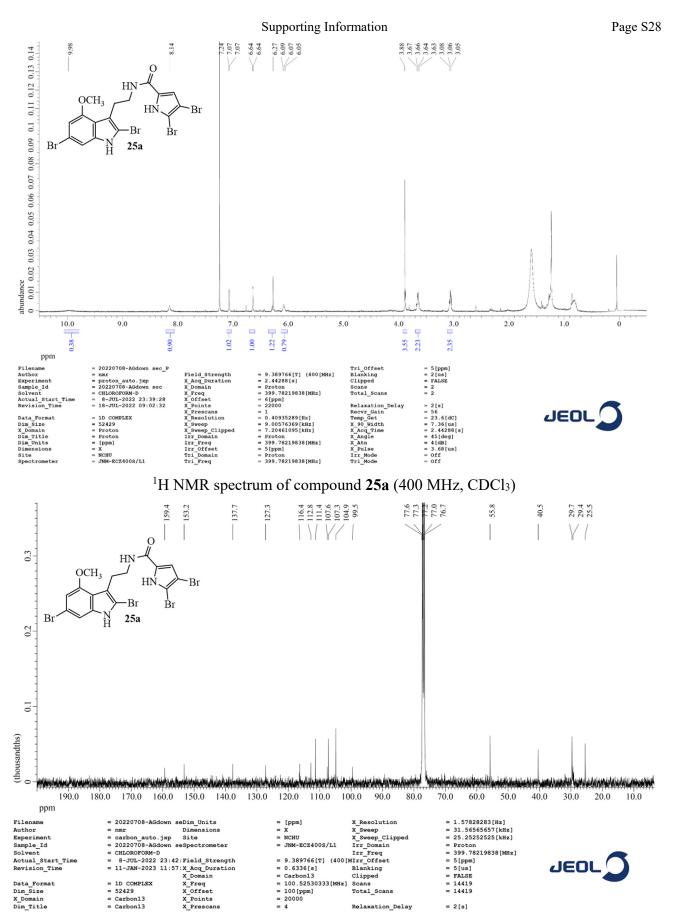
¹³C NMR spectrum of compound **14a** (100 MHz, CDCl₃)



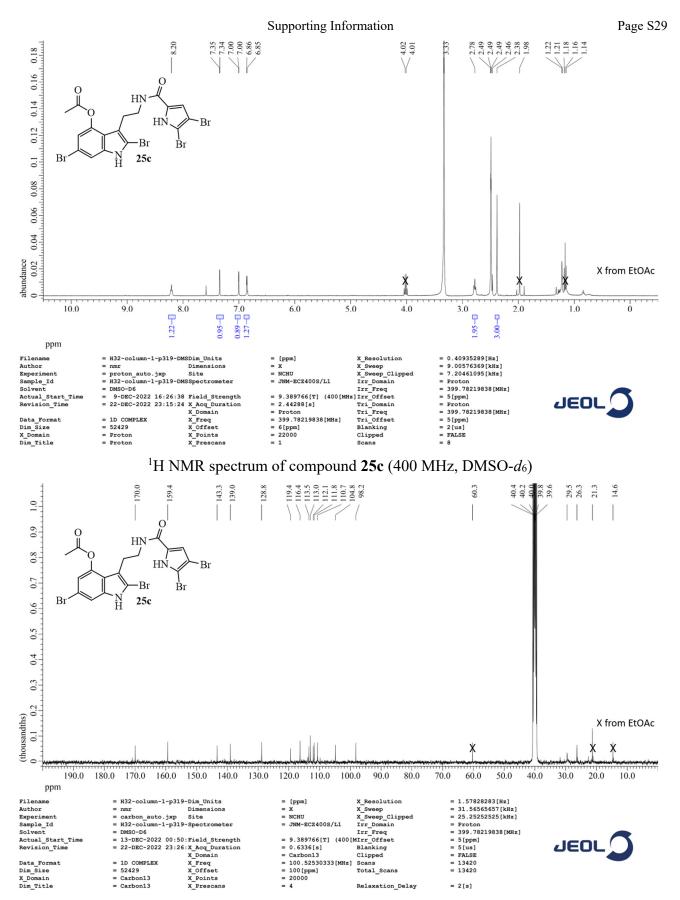
¹³C NMR spectrum of compound **14b** (100 MHz, DMSO-*d*₆)



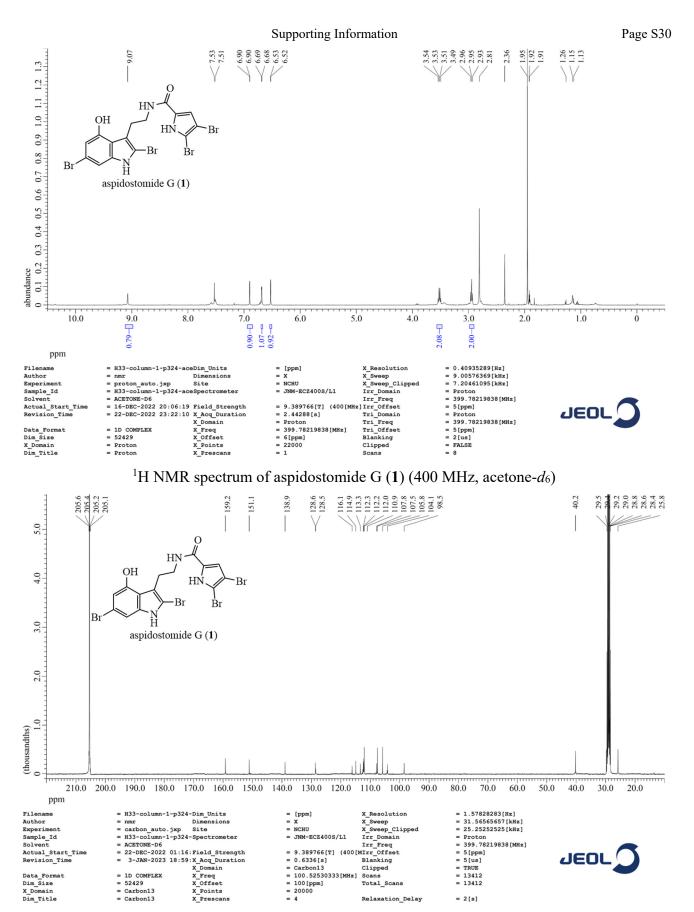
¹³C NMR spectrum of compound **14c** (100 MHz, DMSO-*d*₆)



¹³C NMR spectrum of compound **25a** (100 MHz, CDCl₃)



¹³C NMR spectrum of compound **25c** (100 MHz, DMSO-*d*₆)



¹³C NMR spectrum of aspidostomide G (1) (100 MHz, acetone-*d*₆)

References and Notes:

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