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## **Supplementary Information**

# Synthesis of 1,8-naphthalimides with inbuilt catechol and crown ether functionality

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#### S1. General Experimental and Synthesis of Compounds

All general reagents and solvents were obtained from commercial sources and used as supplied. All NMR spectra (<sup>1</sup>H and <sup>13</sup>C) were obtained using either a Bruker Avance 400SB or Bruker Avance 500SB spectrometer as specified. Samples were dissolved in DMSO- $d_6$  or MeOD with the residual solvent signal used as the internal reference: 2.50 ppm (<sup>1</sup>H) and 39.52 ppm (<sup>13</sup>C). <sup>1</sup>H spectra are reported as: chemical shift  $\delta$  (ppm), multiplicity (s = singlet, d = doublet, t = triplet, m = multiplet, dd = doublet of doublets), coupling constant *J* (Hz), integral, assignment. <sup>13</sup>C spectra are reported as: chemical shift  $\delta$  (ppm), multiplicity, coupling constant *J* (Hz).

High Resolution Mass Spectra (HRMS) analysis was conducted and recorded on a Shimadzu LCMS-9030 Q-TOF mass spectrometer in a 95% MeOH in H<sub>2</sub>O solvent system containing 0.1% formic acid. Analyte solutions were prepared in HPLC grade MeOH (conc. ~ 1  $\mu$ g 1 mL<sup>-1</sup>).

UV-Visible absorption spectra were collected on a Cary 60 Bio UV-Vis spectrophotometer and emission spectra were collected on a Cary Eclipse fluorescence spectrometer using a 1 cm quartz cuvette.

#### 5-Hydroxy-2-(2-methoxyethyl)-1*H*-benzo[*de*]isoquinoline-1,3(2*H*)-dione (5)



To a solution of 3-hydroxy-1,8-naphthalic anhydride (811 mg, 3.79 mmol) in EtOH (15 mL) was added 2-methoxyethylamine (370  $\mu$ L, 4.26 mmol, 1.1 equiv.) and the resultant mixture heated using microwave irradiation at 100 °C for 60 mins. After cooling to ambient the solution was diluted with H<sub>2</sub>O (10 mL) and cooled on ice, after which a precipitate formed that was collected by vacuum filtration to afford **5** (967 mg, 94%) as a pale yellow feathery crystals.

<sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  10.55 (br. s, 1H), 8.26–8.23 (m, 2H), 8.02 (d, *J* = 2.2, 1H), 7.76–7.72 (m, 1H), 7.65 (d, *J* = 2.2, 1H), 4.23 (t, *J* = 6.2, 2H), 3.57 (t, *J* = 6.2, 2H), 3.26 (s, 3H). <sup>13</sup>C NMR (101 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  163.5, 163.2, 156.2, 133.3, 132.6, 127.4, 127.3, 123.3, 122.1, 121.9, 121.8, 115.8, 68.7, 58.0, 38.6. HRMS (ESI, *m*/*z*): calculated for C<sub>15</sub>H<sub>13</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 272.0918, found 272.0925. 5-(Benzyloxy)-2-(2-methoxyethyl)-1H-benzo[de]isoquinoline-1,3(2H)-dione (6)



To a solution of **5** (2.33 g, 8.60 mmol) in MeCN (50 mL), was added  $K_2CO_3$  (2.38 g, 17.2 mmol, 2.0 equiv.) and benzyl bromide (1.63 mL, 12.9 mmol, 1.5 equiv.), and the resultant mixture stirred at 40 °C for 24 h. After cooling to ambient temperature, the mixture was diluted with H<sub>2</sub>O (30 mL) and a precipitate formed. Vacuum filtration afforded **6** (2.98 g, 96%) as a fluffy pale yellow powder.

<sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  8.32–8.30 (m, 2H), 8.10 (d, *J* = 2.5, 1H), 8.02 (d, *J* = 2.5, 1H), 7.82–7.79 (m, 1H), 7.55–7.54 (m, 2H), 7.45–7.41 (m, 2H), 7.38–7.34 (m, 1H), 5.35 (s, 2H), 4.23 (t, *J* = 6.2, 2H), 3.58 (t, *J* = 6.2, 2H), 3.26 (s, 3H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  162.5, 163.1, 156.8, 136.4, 133.15, 133.11, 128.6 (2xC)^ 128.4, 128.2, 127.9 (2xC)^, 127.8, 123.6, 123.0, 122.1, 121.9, 114.7, 70.0, 68.7, 58.0, 38.7. HRMS (ESI, m/z): calculated for C<sub>22</sub>H<sub>19</sub>NO<sub>4</sub> [M+H]<sup>+</sup> 362.1387; found 362.1384. ^coincident signals confirmed using inverse gated <sup>13</sup>C experiment.

5-(Benzyloxy)-6-chloro-2-(methoxyethyl)-1*H*-benzo[*de*]isoquinoline-1,3(2*H*)-dione (7)



To a solution of **6** (2.09 g, 5.78 mmol) in MeCN (50 mL) was added *N*-chlorosuccinimide (1.54 g, 11.6 mmol, 2.0 equiv.) and the resultant mixture stirred at 40 °C for 22 h. After this time the mixture was cooled to ambient temperature and diluted with H<sub>2</sub>O (20 mL). The resultant solid was collected using vacuum filtration, affording **7** (1.96 g, 86%) as a pale yellow powder.

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  8.53 (dd, *J* = 1.0, 8.5, 1H), 8.43 (dd, *J* = 1.0, 7.3, 1H), 8.42 (s, 1H), 7.96 (dd, *J* = 7.3, 8.5, 1H), 7.57–7.55 (m, 2H), 7.46–7.42 (m, 2H), 7.38–7.34 (m, 1H), 5.56 (s, 2H), 4.24 (t, *J* = 6.1, 2H), 3.59 (t, *J* = 6.1, 2H), 3.26 (s, 3H). <sup>13</sup>C NMR (101 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  163.1, 162.7, 152.4, 136.2, 129.9, 129.4, 129.2, 129.0, 128.6 (2xC), 128.2, 127.6 (2xC), 123.3, 123.1, 122.4, 121.8, 118.5, 71.0, 68.6, 58.0, 29.6. HRMS (ESI, *m*/*z*): calculated for C<sub>22</sub>H<sub>18</sub>ClNO<sub>4</sub> [M+H]<sup>+</sup> 396.0997, found 396.0989. ^coincident signals confirmed using inverse gated <sup>13</sup>C experiment.

5-(Benzyloxy)-6-hydroxy-2-(2-methoxyethyl)-1*H*-benzo[*de*]isoquinoline-1,3(2*H*)-dione (8)



To a solution of **7** (1.95 g, 4.93 mmol) in DMSO (50 mL) was added *N*-hydroxysuccinimide (624 mg, 5.43 mmol, 1.1 equiv.), and  $K_2CO_3$  (2.25 mg, 16.3 mmol, 3.3 equiv.), and the resultant mixture heated at 80 °C for 24 h. After this time the mixture was cooled and diluted with H<sub>2</sub>O (100 mL), then acidified to pH = 1 using 2M HCl. At this point a yellow precipitate formed in solution and was collected using vacuum filtration, affording **8** (1.45 g, 75%) as a yellow powder.

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>): δ 11.09 (s, 1H), 8.50–8.48 (m, 1H), 8.33–8.32 (m, 1H), 8.22 (s, 1H), 7.72–7.68 (m, 1H), 7.58–7.56 (m, 2H), 7.42–7.40 (m, 2H), 7.34–7.30 (m, 1H), 5.23 (s, 2H), 4.19 (t, J = 6.1, 2H), 3.55 (t, J = 6.1, 2H), 3.24 (s, 3H). <sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>): δ 163.7, 162.9, 148.4, 142.4, 136.8, 129.6, 128.5 (2xC)^, 128.4, 128.0, 127.7 (2xC)^, 125.8, 124.3, 123.0, 121.6, 119.8, 112.3, 70.9, 68.8, 58.0, 38.4. HRMS (ESI, *m/z*): calculated for C<sub>22</sub>H<sub>29</sub>NO<sub>5</sub> [M+H]<sup>+</sup> 378.1336, found 378.1340. ^coincident signals confirmed using inverse gated <sup>13</sup>C experiment.

#### 5,6-Bis(benzyloxy)-2-(methoxyethyl)-1*H*-benzo[*de*]isoquinoline-1,3(2*H*)-dione (9)



To a solution of **8** (1.80 mg, 4.76 mmol) in MeCN (50 mL) was added benzyl bromide (903  $\mu$ L, 7.14 mmol, 1.5 equiv.) and K<sub>2</sub>CO<sub>3</sub> (1.32 mg, 9.52 mmol, 2.0 equiv.), and the resultant mixture heated at 40 °C for 24 h. After this time the mixture was cooled to ambient temperature and the solvent removed under reduced pressure. The residue was then diluted with H<sub>2</sub>O (40 mL) and the resultant solid collected using vacuum filtration, affording **9** (1.78 g, 80%) as a pale yellow powder.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  8.48–8.47 (m, 2H), 8.39 (dd, J = 8.5, 1.1, 1H), 7.63 (dd, J = 8.5, 7.2, 1H), 7.54–7.52 (m, 2H), 7.45–7.38 (m, 5H), 7.36–7.33 (m, 3H), 5.38 (s, 2H), 5.36 (s, 2H), 4.43 (t, J = 8.9, 2H), 3.72 (t, J = 8.9, 2H), 3.38 (s, 3H). <sup>13</sup>C NMR (101 MHz, DMSO- $d_6$ ):

δ 163.5, 162.8, 148.2, 147.1, 136.7, 136.6, 129.3, 128.7 (2xC)<sup>^</sup>, 128.66 (2xC)<sup>^</sup>, 128.5 (2xC)<sup>^</sup>, 128.4, 128.24, 128.20, 128.0 (2xC)<sup>^</sup>, 127.3, 127.2, 123.7, 121.8, 120.3, 117.6, 75.2, 71.2, 68.7, 58.0, 38.6. HRMS (ESI, *m*/*z*): calculated for C<sub>29</sub>H<sub>25</sub>NO<sub>5</sub> [M+H]<sup>+</sup> 468.1806, found 468.1803. <sup>^</sup>coincident signals confirmed using inverse gated <sup>13</sup>C experiment.

#### 5,6-Dihydroxy-2-(methoxyethyl)-1*H*-benzo[*de*]isoquinoline-1,3(2*H*)-dione (Nap-Cat)



To a flask flushed with N<sub>2</sub> was added Pd(OH)<sub>2</sub>/C 10% w/w (68 mg, 0.481 mmol, 0.5 equiv.) and MeOH (30 mL). The flask was then evacuated and backfilled with N<sub>2</sub> three times, then **9** (450 mg, 0.963 mmol) added to the flask. The flask was then evacuated and backfilled with N<sub>2</sub> twice, and then evacuated and backfilled with H<sub>2</sub> twice. The resultant mixture was allowed to stir at 21 °C for 8 h. After this time the flask was evacuated and backfilled with N<sub>2</sub>, then diluted with 50 mL MeOH. The reaction mixture was then filtered through a bed of celite, washing with MeOH. The filtrate was transferred to a round bottom flask and the solvent removed *in vacuo* to afford **Nap-Cat** (214 mg, 77%) as a yellow powder.

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  8.45–8.44 (m, 1H), 8.30–8.28 (m, 1H), 8.09 (s, 1H), 7.68–7.65 (m, 1H), 4.21 (t, J = 6.1, 2H), 3.55 (t, J = 6.1 Hz, 2H), 3.25 (s, 3H). <sup>13</sup>C NMR (101 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  164.0, 163.0, 147.7, 141.8, 128.7, 128.3, 125.2, 123.9, 123.4, 121.6, 121.4, 111.3, 68.8, 58.0, 38.4. HRMS (ESI, m/z): calculated for C<sub>15</sub>H<sub>13</sub>NO<sub>5</sub>¬ [M+Na]<sup>+</sup> 310.0686 found 310.0682.

#### 5-(2-Methoxyethyl)-9,10,12,13,15,16,18,19-octahydro-4H-

benzo[*de*][1,4,7,10,13]pentaoxacyclopentadecino[2,3-g]isoquinoline-4,6(5*H*)-dione (Nap-Crown)



To a stirring solution of tetraethylene glycol di(*p*-toluenesulfonate) (2.25 g, 4.47 mmol, 2.0 equiv.) in acetone (50 mL) was added KI (1.71 g, 10.6 mmol, 4.8 equiv.) and the resultant mixture heated at 40 °C for 24 h. After this time the mixture was cooled and solvent removed under reduced pressure. The residue was then diluted with H<sub>2</sub>O and extracted with 10% MeOH

in CH<sub>2</sub>Cl<sub>2</sub> ( $3 \times 15$  mL), the combined organics dried over MgSO<sub>4</sub>, and solvent removed under reduced pressure to give a yellow oil. The crude material was immediately diluted with MeCN (15 mL) and **Nap-Cat** (635 mg, 2.21 mmol) and Cs<sub>2</sub>CO<sub>3</sub> (3.18 g, 9.76 mmol, 4.4 equiv.) added, then the resultant mixture heated at 40 °C in a foil-wrapped flask for 48 h. The reaction mixture was then cooled, solvent removed under reduced pressure, then the residue diluted with H<sub>2</sub>O and extracted with 10% MeOH in CH<sub>2</sub>Cl<sub>2</sub> ( $3 \times 5$ mL). The combined organics were dried over MgSO<sub>4</sub> and the solvent removed under reduced pressure. The crude solid was purified by flash column chromatography, using silica gel 60 (70–230 mesh) and 1% MeOH in CH<sub>2</sub>Cl<sub>2</sub>. The column was neutralised (before loading the crude product) by flushing with a dilute solution of Et<sub>3</sub>N in CH<sub>2</sub>Cl<sub>2</sub>. Fractions containing the product were combined and the solvent removed to afford **Nap-Crown** as a yellow powder (5 mg, 1%).

<sup>1</sup>H NMR (400 MHz, MeOD):  $\delta$  8.34–8.32 (m, 1H), 8.28 (dd, *J* = 7.3, 0.9 Hz, 1H), 8.17 (s, 1H), 7.62 (dd, *J* = 8.4, 7.3 Hz, 1H), 4.49 (t, *J* = 5.6 Hz, 2H), 4.36-4.34 (m, 2H), 4.29 (t, *J* = 6.1 Hz, 2H), 4.04 (t, *J* = 5.6 Hz, 2H), 3.93–3.91 (m, 2H), 3.78–3.76 (m, 2H), 3.72-3.64 (m, 10H), 3.30 (s, 3H). <sup>13</sup>C NMR (101 MHz, MeOD):  $\delta$  165.9, 165.2, 149.6, 131.0, 129.7, 128.7, 128.2, 125.8, 123.6, 121.0, 118.9, 75.5, 73.4, 72.3, 72.1, 71.3, 71.0, 70.8, 70.4, 70.2, 59.2, 40.4. HRMS (ESI, *m/z*): calculated for C<sub>23H27</sub>NO<sub>8</sub>[M+Na]<sup>+</sup> 468.1629, found 468.1644.

## S2. <sup>1</sup>H NMR time study of Nap-Cat in DMSO-d<sub>6</sub> with excess H<sub>2</sub>O<sub>2</sub>





S3. <sup>1</sup>H and <sup>13</sup>C NMR Spectra of New Compounds in DMSO-d<sub>6</sub>





#### **S4. Electrochemical Measurements**

Cyclic voltammetry (CV) measurements were performed on a Bio-Logic Science Instrument VSP potentiostat with a PC controlled electrochemical workstation. Solutions for cyclic voltammetry were prepared in HPLC grade MeCN, which was dried using an in-house solvent purification system. A minimum volume of solvent (ca. 2-3 mL) was used. Pt wires were used as the pseudoreference electrode and the counter electrode, while a Pt disk electrode was used for the working electrode. Both before and after the experiments, the Pt disk electrode was cleaned by polishing the surface in an ethanol slurry of abrasive powder (gamma alumina powder, 0.05 micron, CH Instruments), before sonicating in ethanol, washing with acetone, water, and finally ethanol, and then drying. The electrochemical cell consisted of a glass vessel (ca. 40 mL) covered with a plastic lid and drilled with four holes: one for each electrode, and one for nitrogen purging. The vessel was sealed with parafilm to ensure minimal oxygen contamination. Tetra(n-butyl)ammoniumhexafluorophosphate (TBAPF<sub>6</sub>; *ca*. 0.1 M in MeCN) was used as the supporting electrolyte, with the solution degassed by vigorous bubbling for about 10 min prior to scanning the blank sample to ensure a clean background. Small quantities of sample were then added in portions, and then the mixture degassed and scanned. This process was repeated until an adequate signal-to-background ratio became apparent. The measurements in the appropriate electrochemical windows were then carried out, at scan rates of 0.1 and 1 V s<sup>-1</sup>. At the end of each experiment, ferrocene was added as an internal reference, and then the scans were repeated, again at 0.1 and 1 V s<sup>-1</sup>. Estimated error:  $\pm 30$  mV. The redox potentials are reported relative to the ferrocenium/ferrocene (Fc+ /Fc) redox couple.



Cyclic voltammogram of the propyl analogue of compound **9** in MeCN.