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Supplementary Information

Nucleophilic Sulfur Controlled Efficient Ketothioamide Synthesis from Tribromomethyl Carbinols

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1. Control experiments

OH
$$CBr_{3} + S_{8} \xrightarrow{THF} \text{ no reaction}$$

$$CI \qquad 1a \qquad OH \qquad H \qquad THF$$

$$CBr_{3} + OH \qquad CBr_{3} + OH \qquad CBr_{4} +$$

Experiment no. 1: The tribromomethyl carbinol **1a** (0.5 mmol, 1.0 equiv) dissolved in THF (1.0 mL) was added to elemental sulfur (2.0 mmol, 4.0 equiv) in the round bottom flask equipped with a reflux condenser. The resulting reaction mixture was stirred at a temp of 65 °C.

Inference: No reaction.

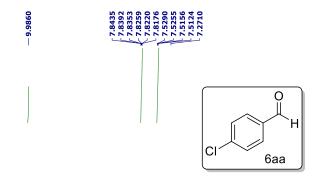
Experiment no. 2: The tribromomethyl carbinol **1a** (0.5 mmol, 1.0 equiv) dissolved in THF (1.0 mL) was added to amine **2a** (40 wt% in water) (10.0 mmol, 20.0 equiv) in the round bottom flask equipped with a reflux condenser. The resulting reaction mixture was stirred at a temp of 65 °C.

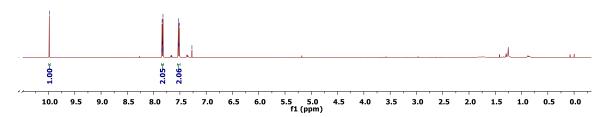
Inference: Formation of aldehyde and was separated as an intermediate **6aa**.

Experiment no. 3: The tribromomethyl carbinol **1a** (0.5 mmol, 1.0 equiv) dissolved in THF (1.0 mL) was added to amine **2a** (40 wt% in water) (10.0 mmol, 20.0 equiv) in the round bottom flask equipped with a reflux condenser. The resulting reaction mixture was stirred at a temp of 65 °C. After one hour of reaction elemental sulfur (2.0 mmol) was added.

Inference: Formation of thioamide 4aa.

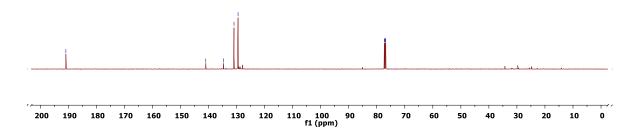
¹H NMR for compound 6aa (500 MHz, CDCl₃)

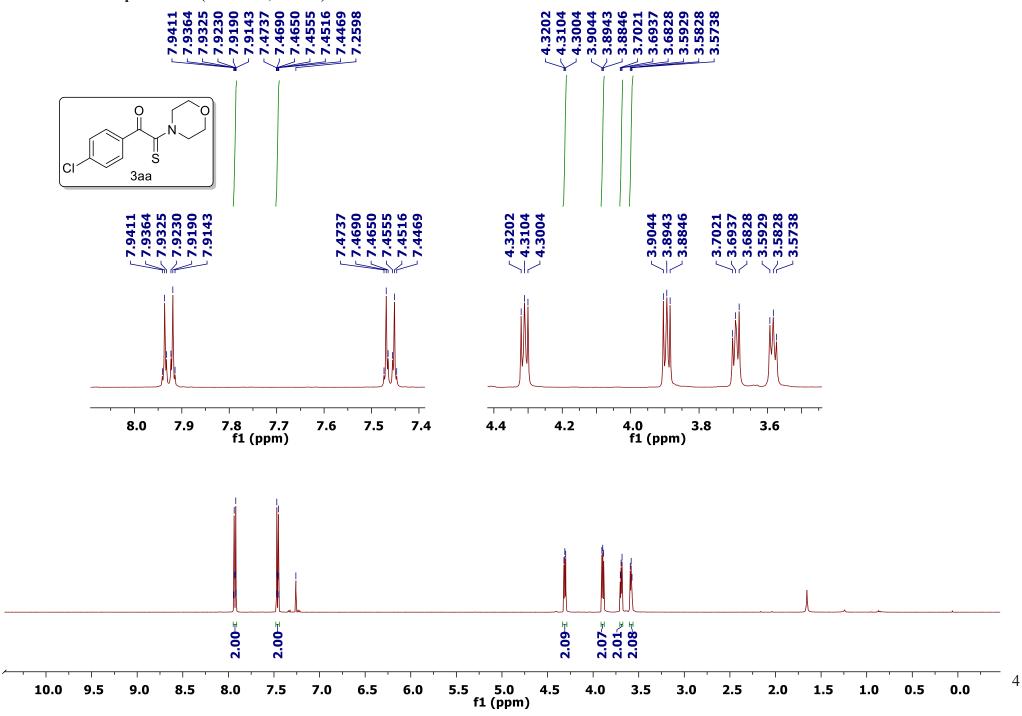


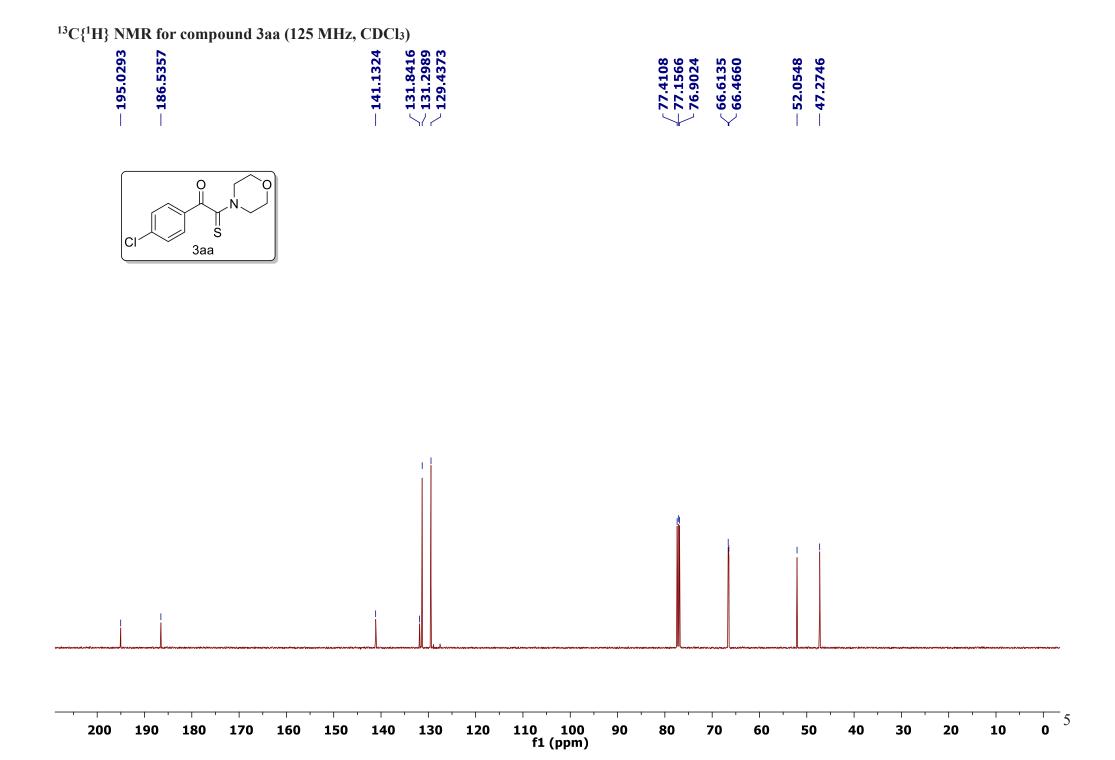


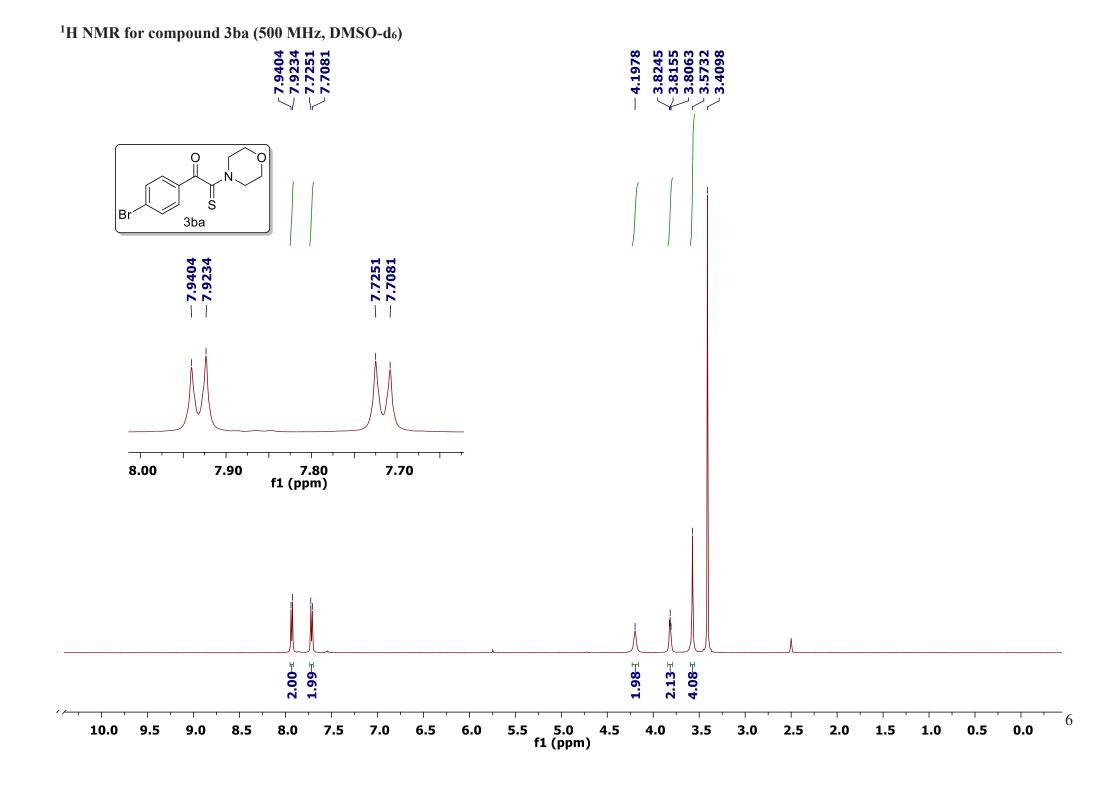
$^{13}\text{C}\{^1\text{H}\}$ NMR for compound 6aa (125 MHz, CDCl $_3)$

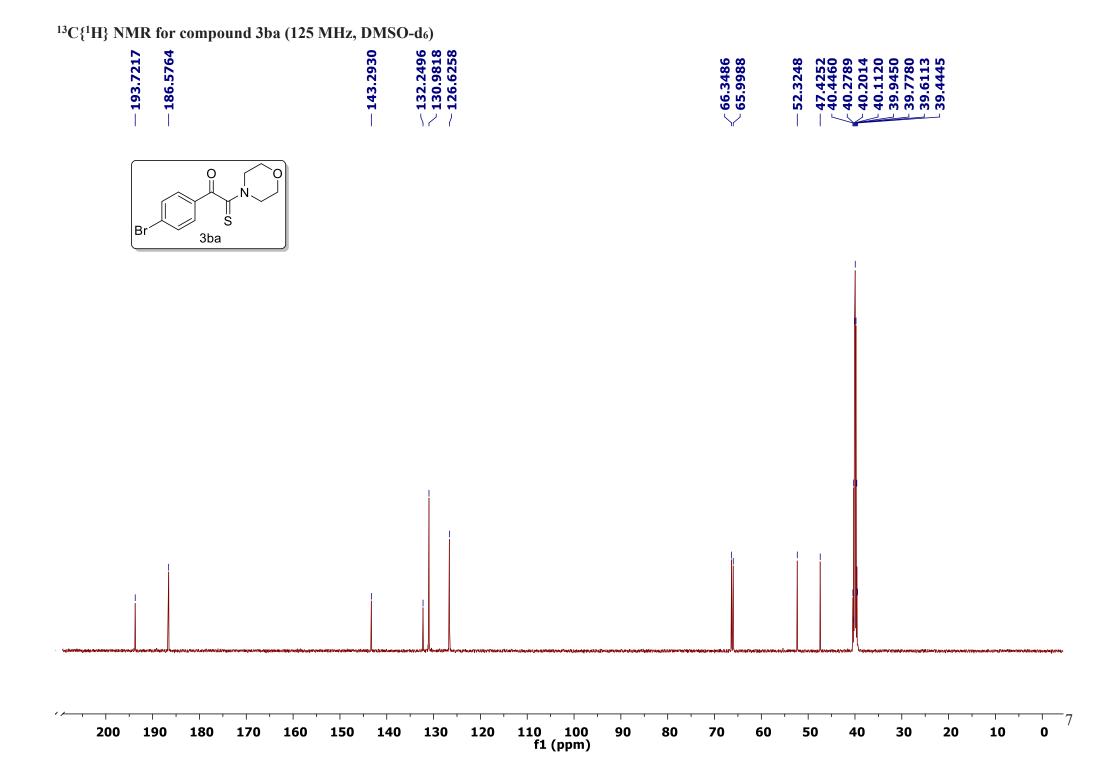
- 190.8911 - 140.9771 / 134,7181 / 129,4735 / 77.3017

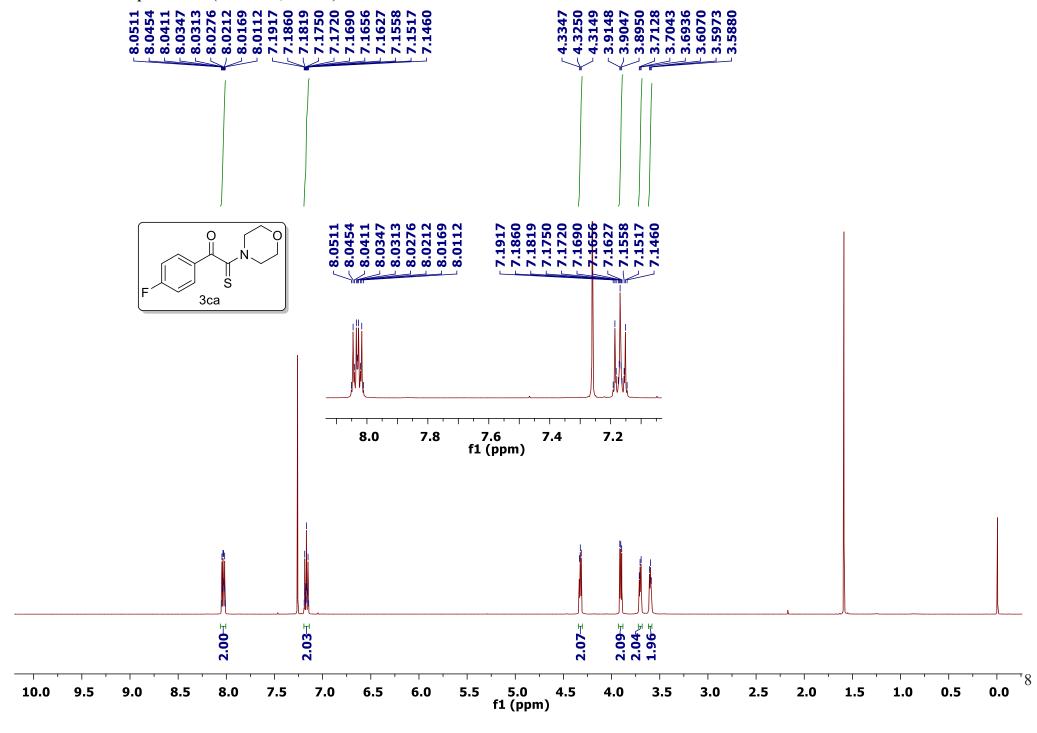


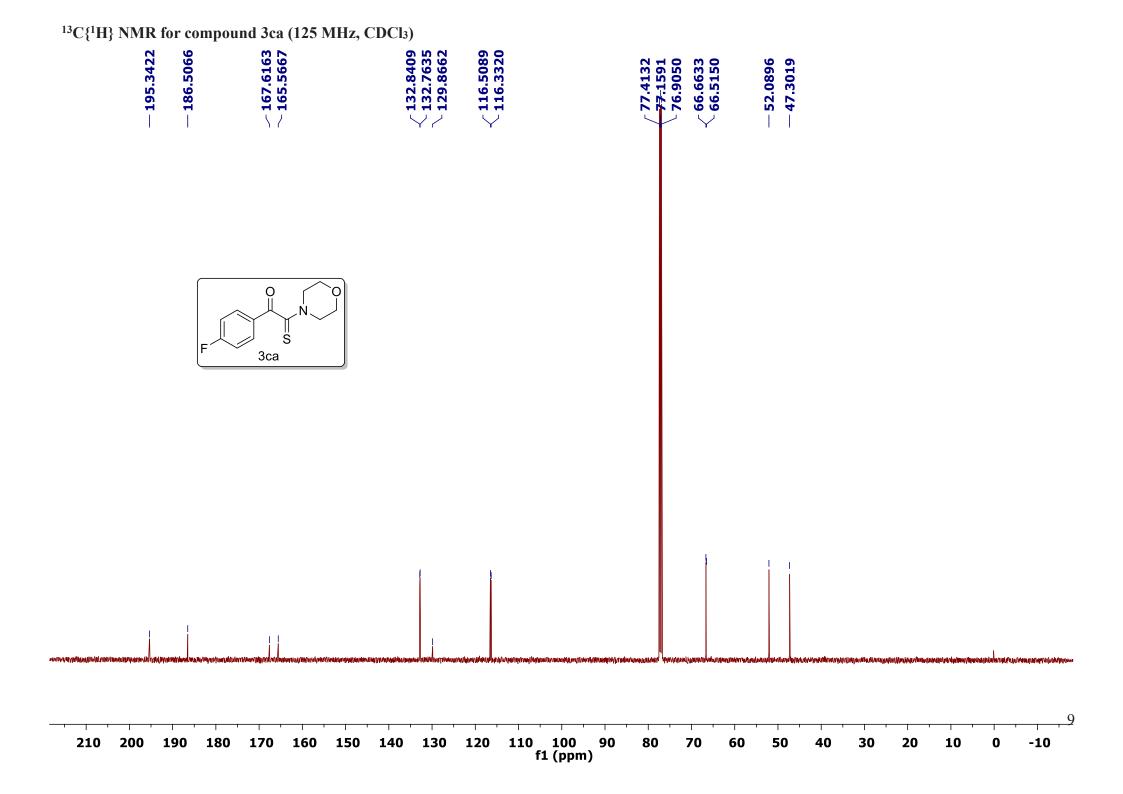


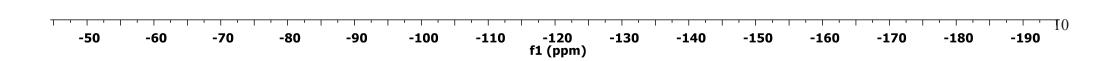


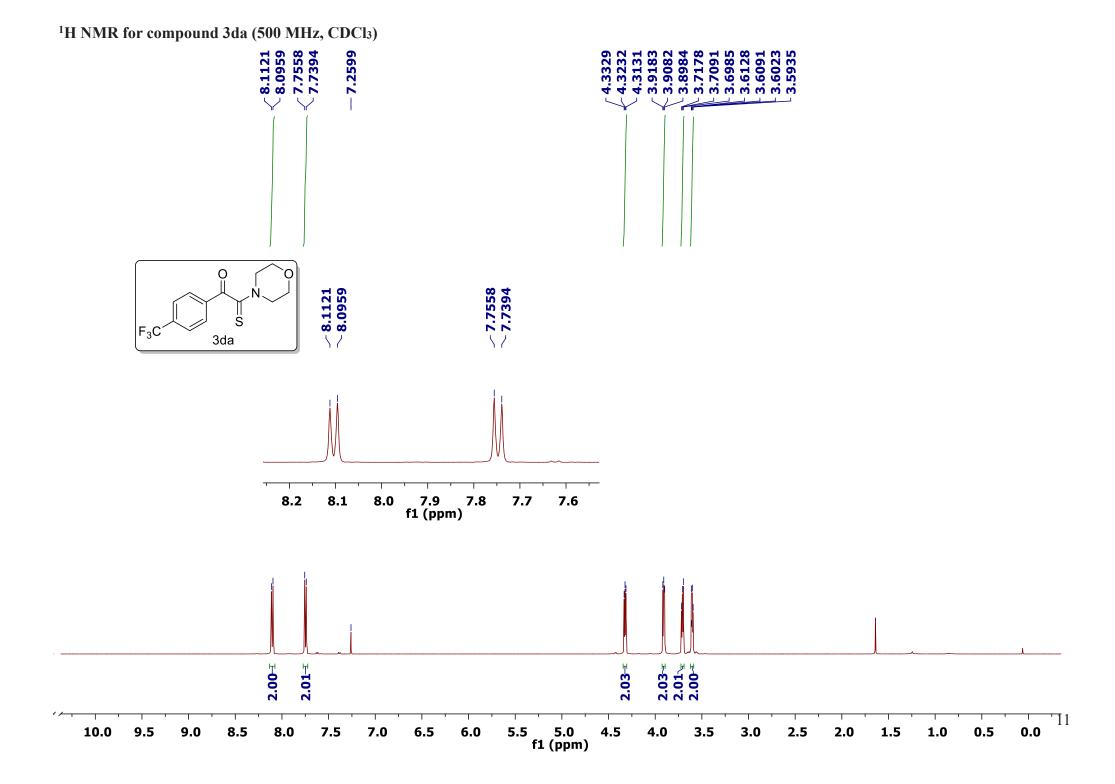






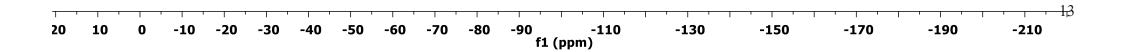


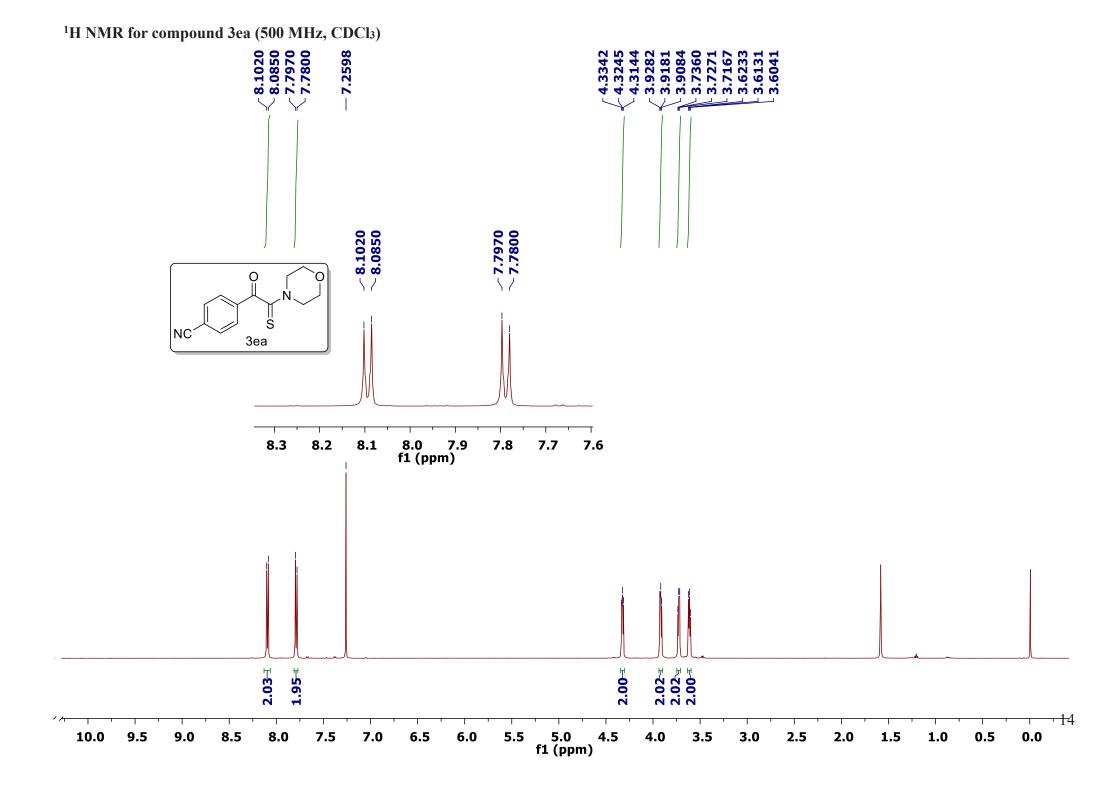


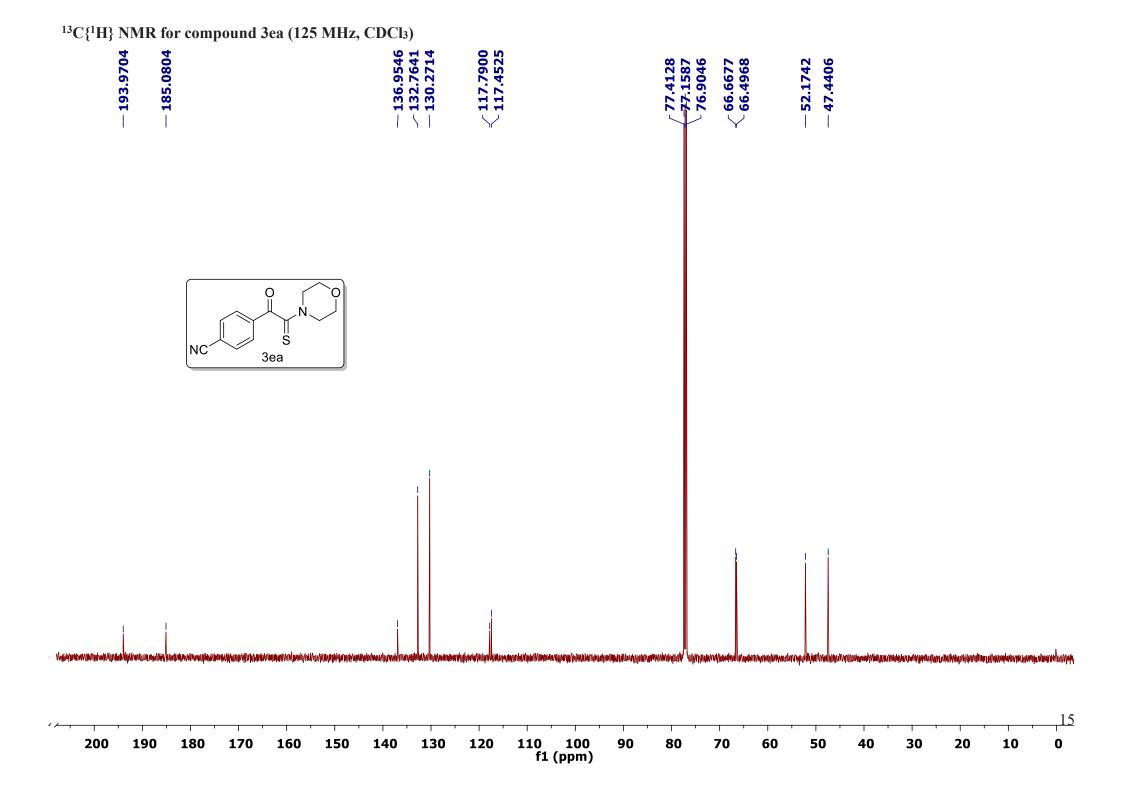


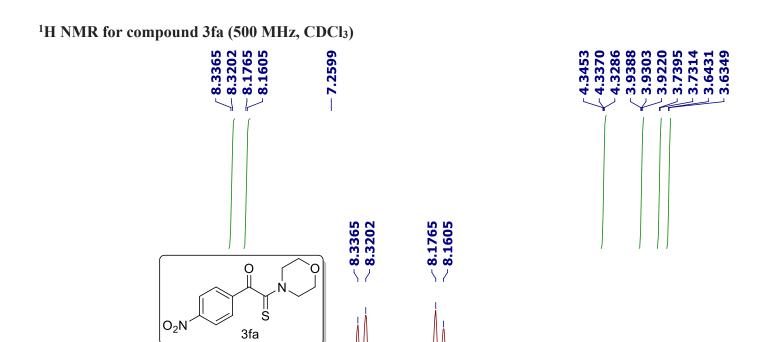
¹³C{¹H} NMR for compound 3da (125 MHz, CDCl₃) 136.4146 135.8992 135.6378 135.3770 135.1155 130.2612 126.7289 126.2661 126.0893 126.0893 126.0893 126.0893 126.0893 126.0893 126.0893 126.0893 185.9335 77.4141 77.1600 76.9057 66.6319 52.1172 135.8992 135.6378 135.1155 0 Ë 3da 135.8 f1 (ppm) 126.0 f1 (ppm) 136.4 135.2 126.2 .0 100 f1 (ppm) 10 200 190 180 170 130 120 110 90 60 **50** 160 150 140 80 70 40 30 20 10 0

--63.2986









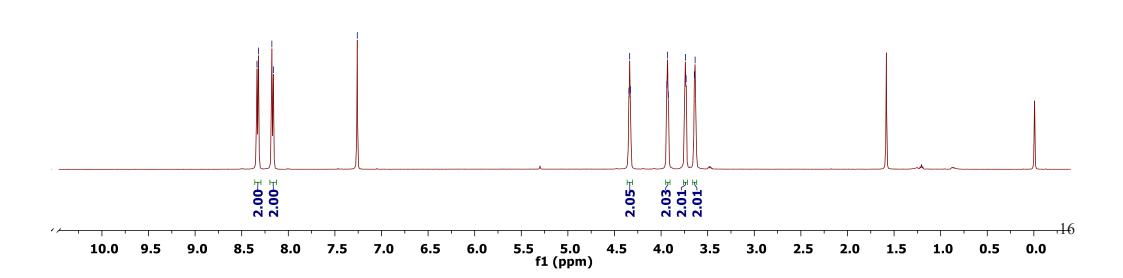
8.2 f1 (ppm)

8.1

8.3

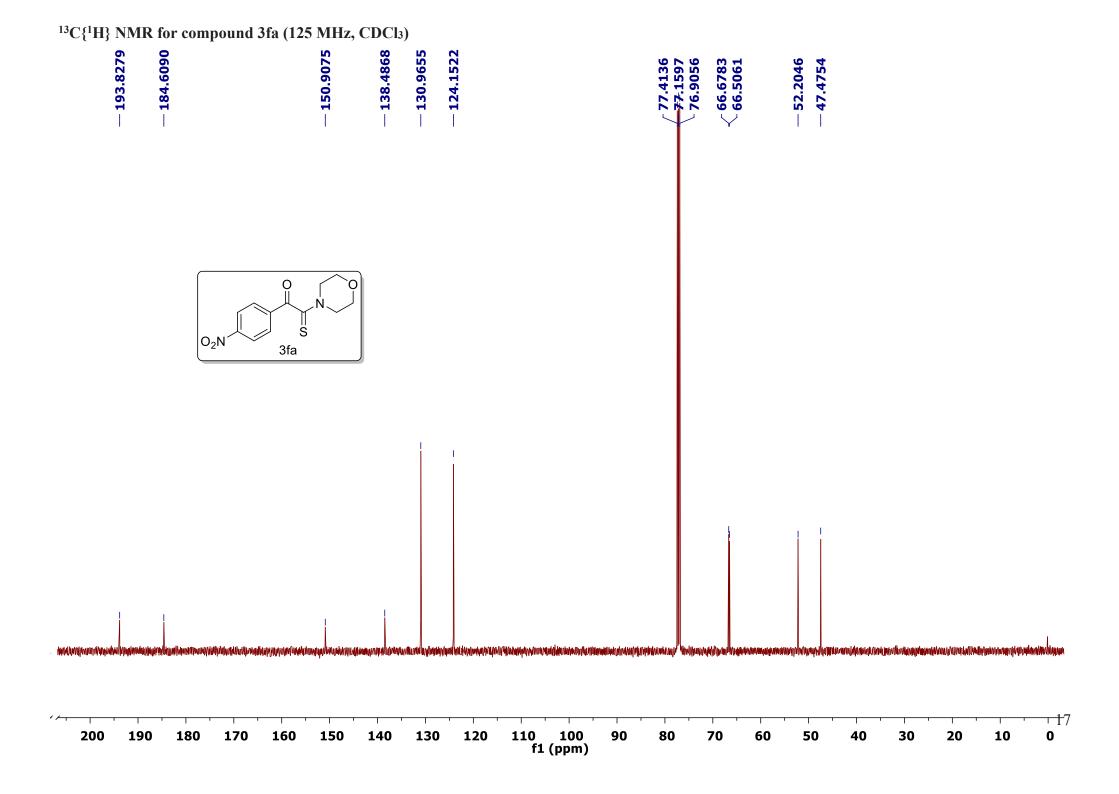
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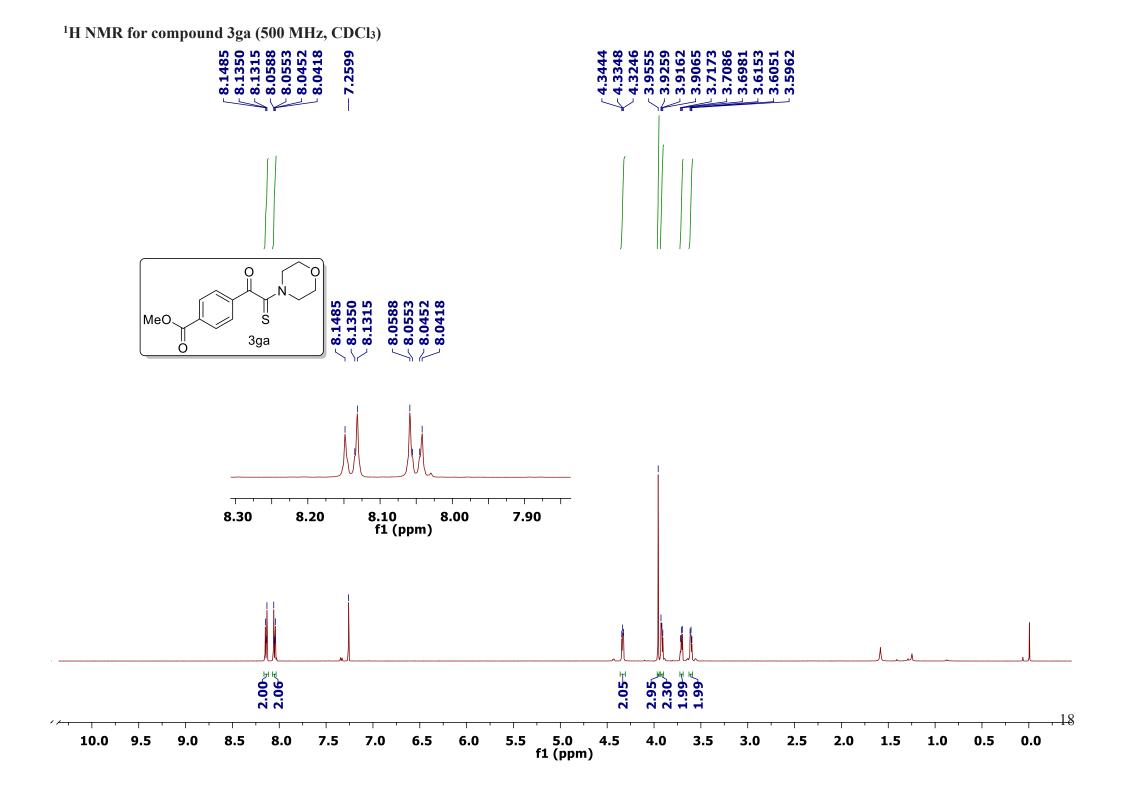
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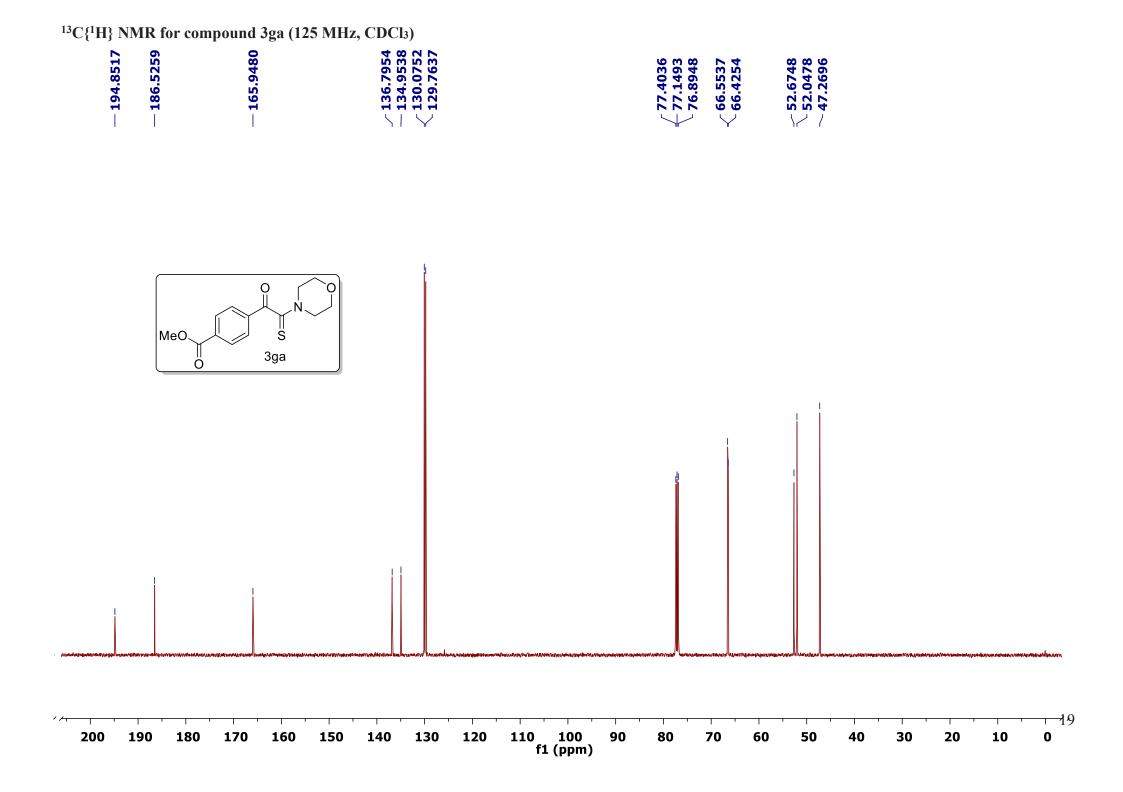


7.9

8.0







5.0 f1 (ppm)

4.5

4.0

3.5

3.0

5.5

2.0

2.5

1.5

0.5

0.0

1.0

10.0

9.5

9.0

8.5

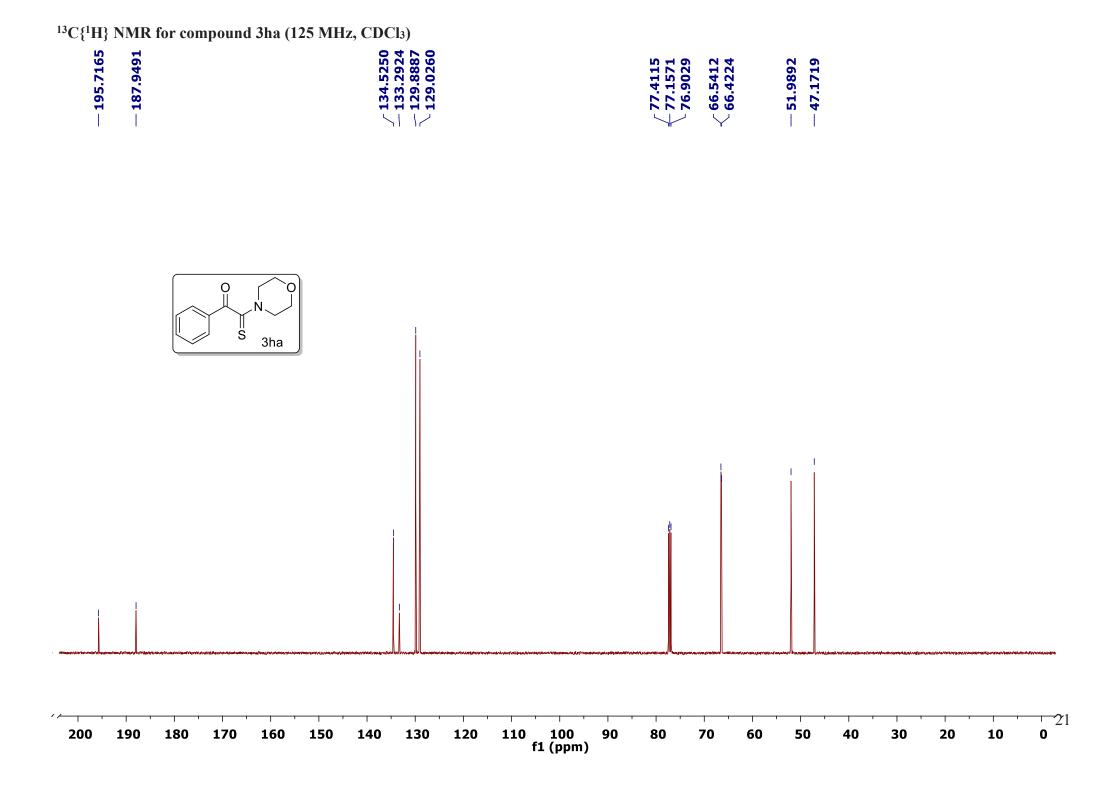
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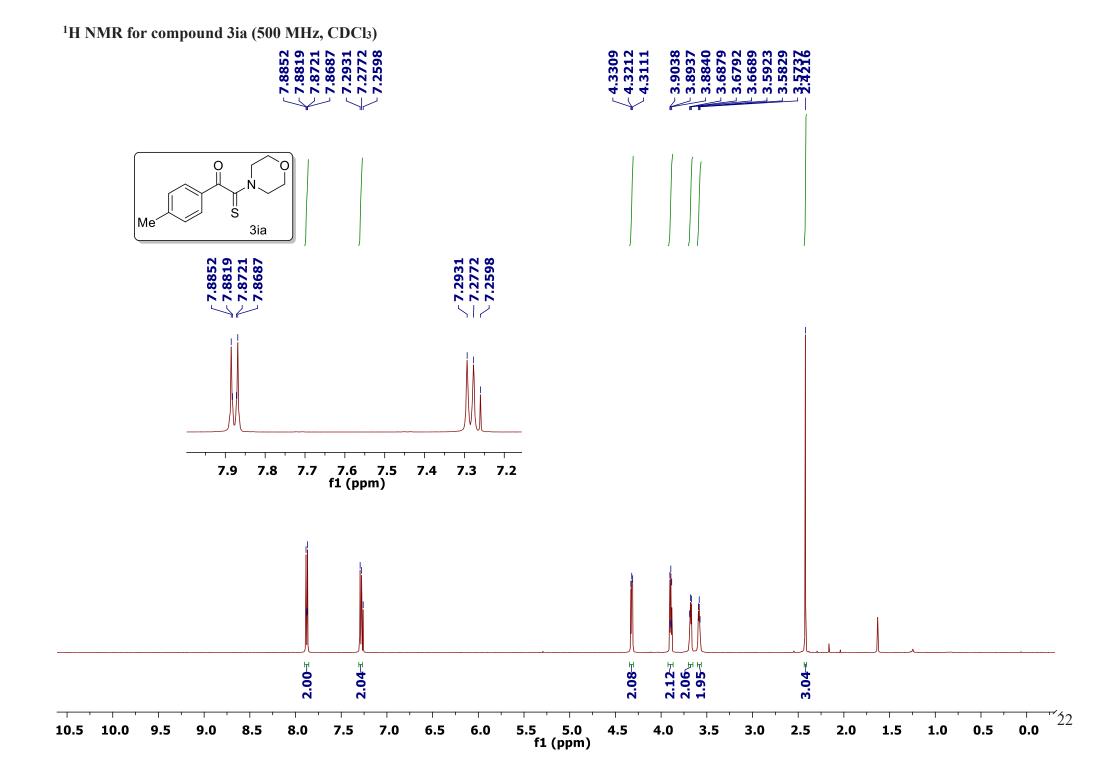
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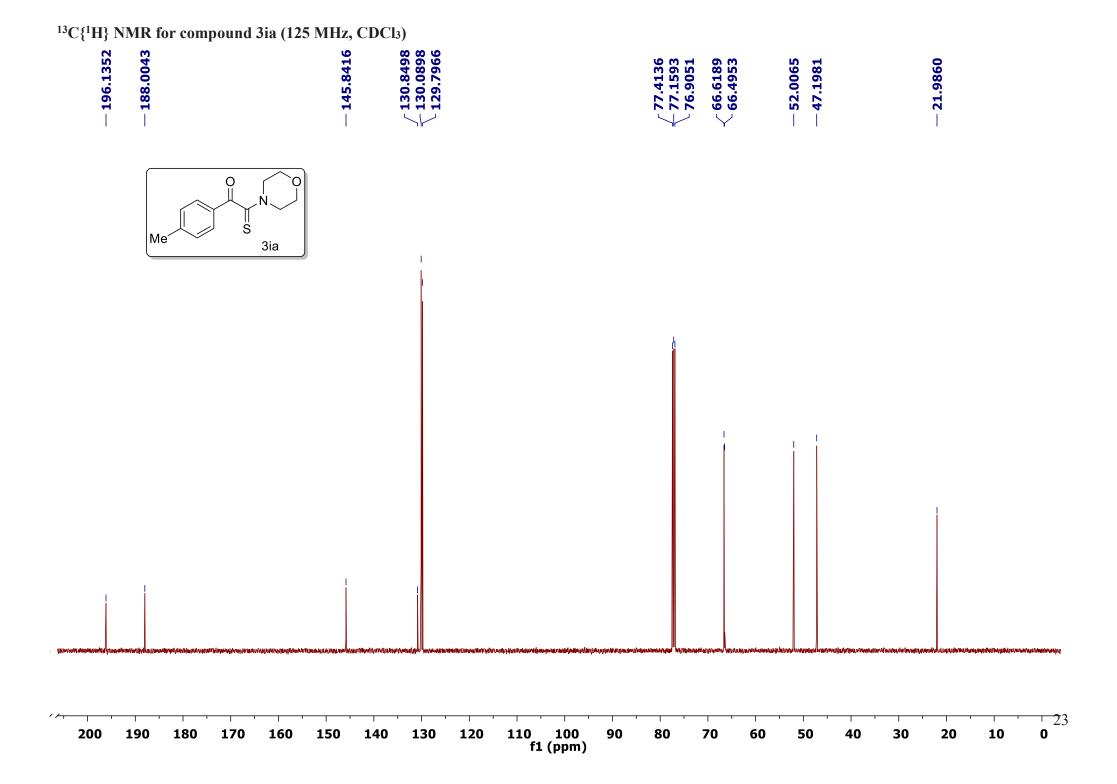
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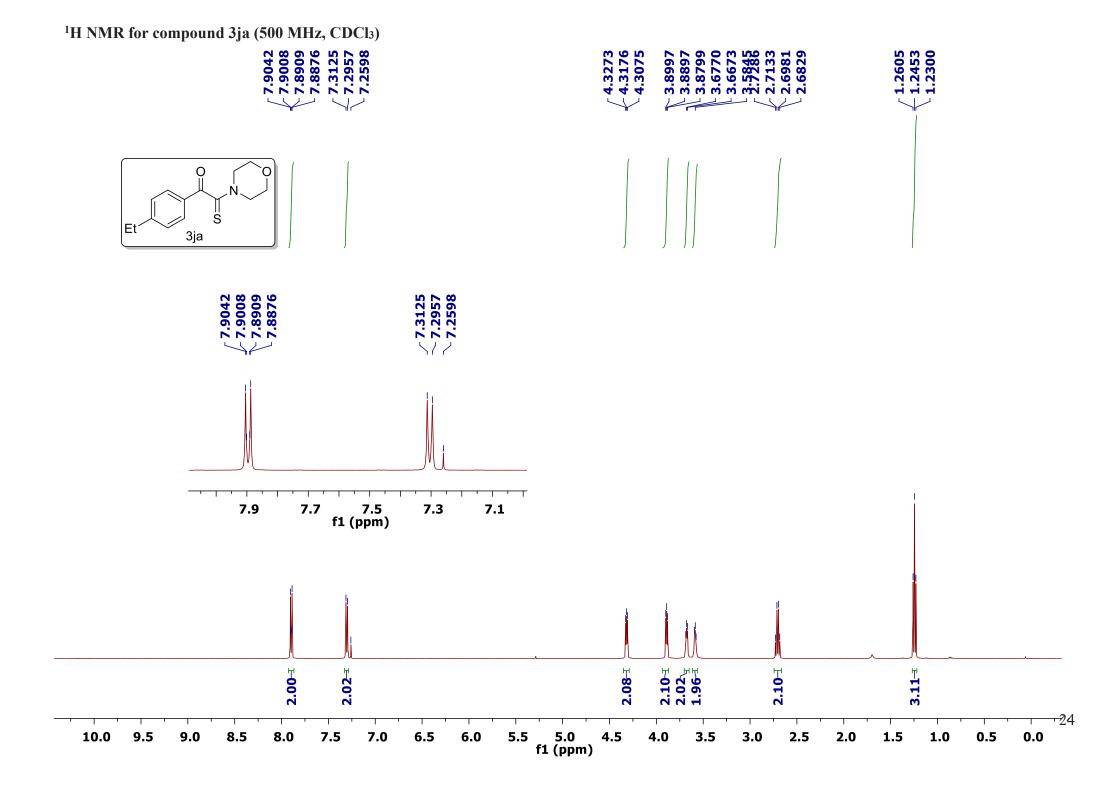
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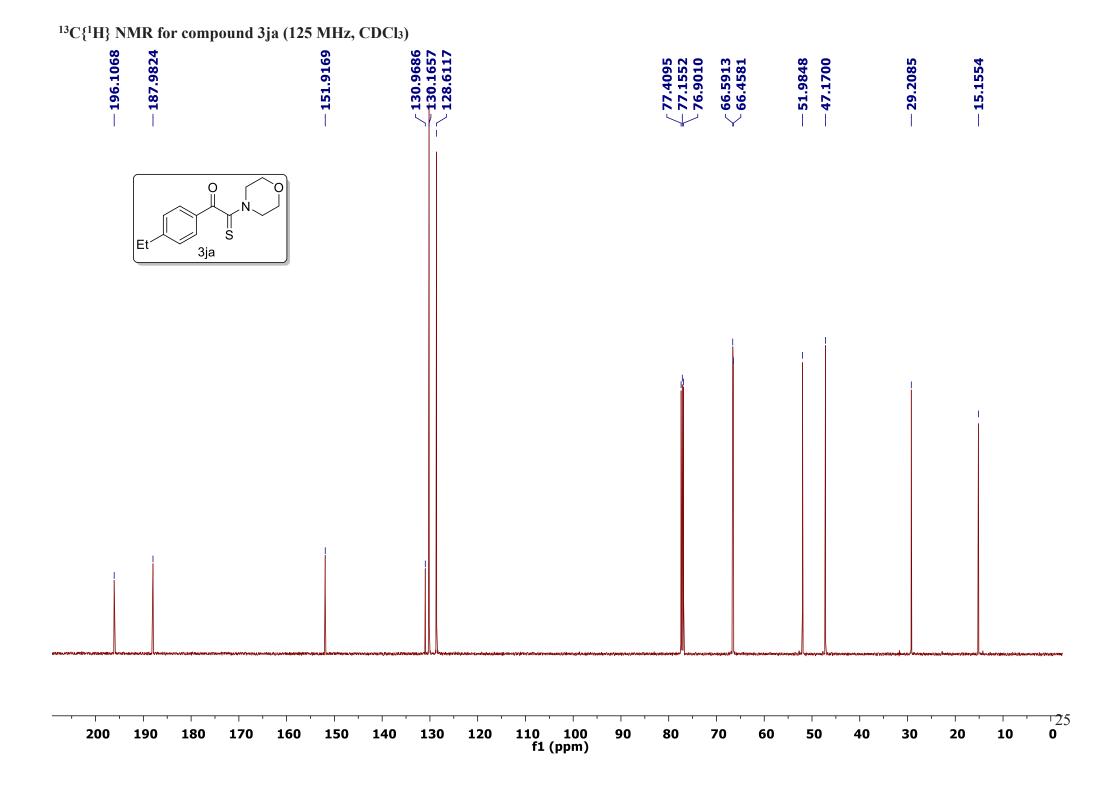
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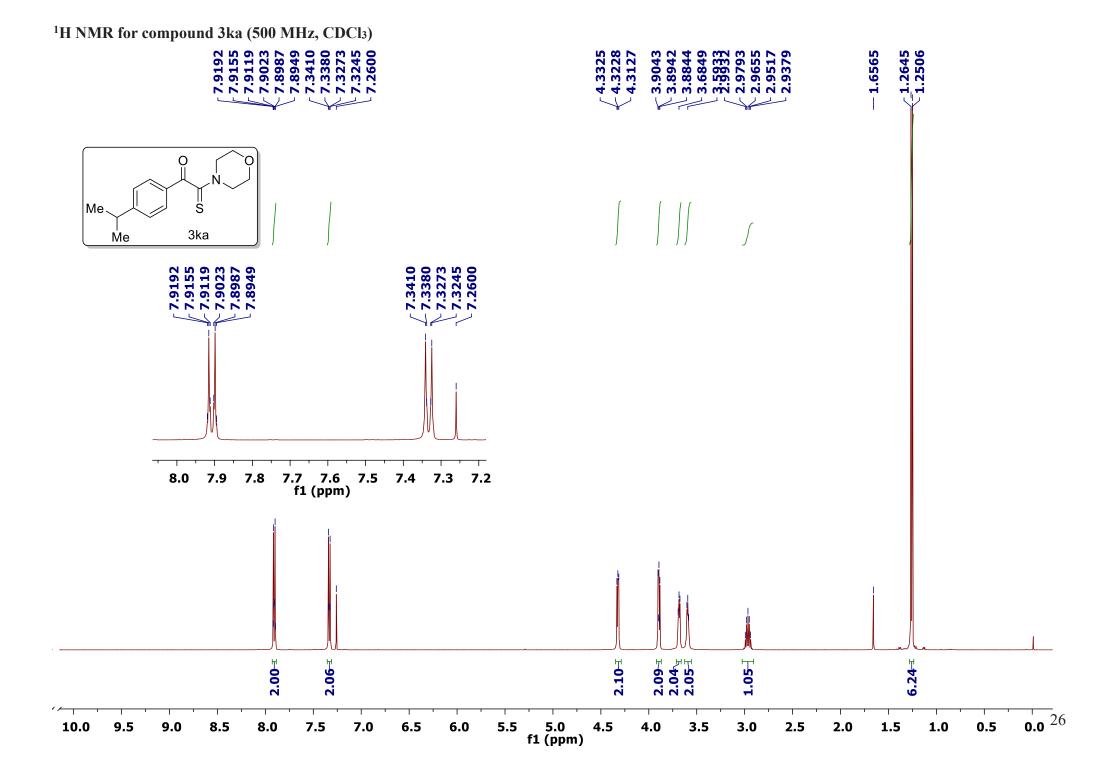


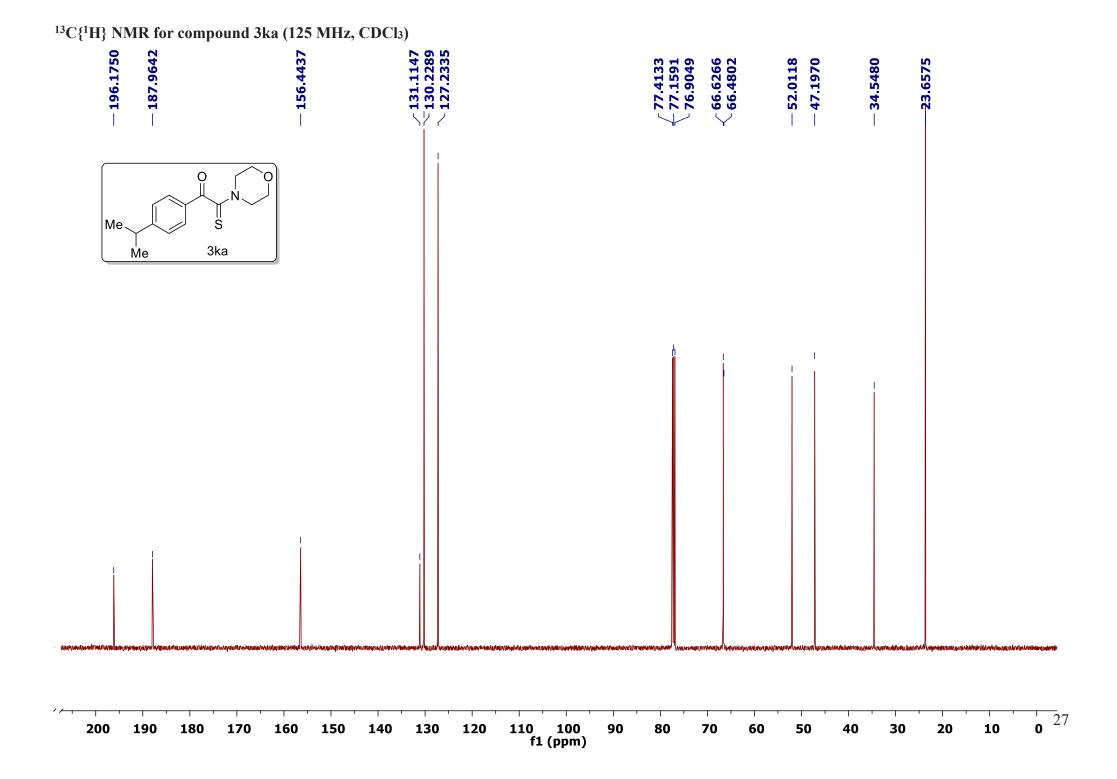


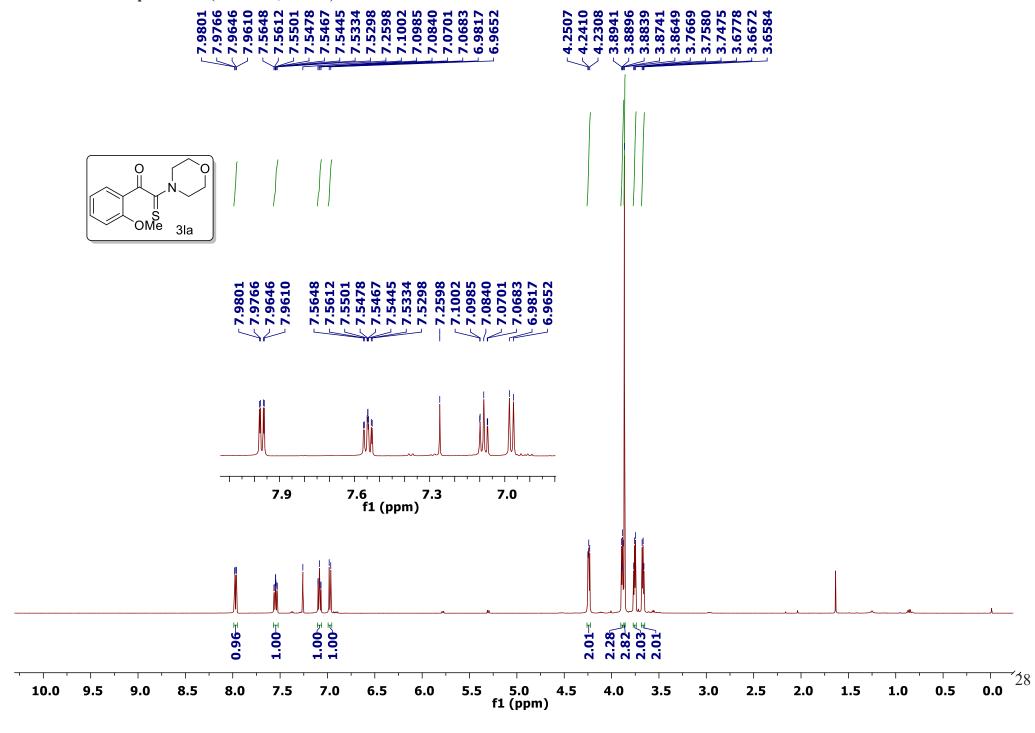


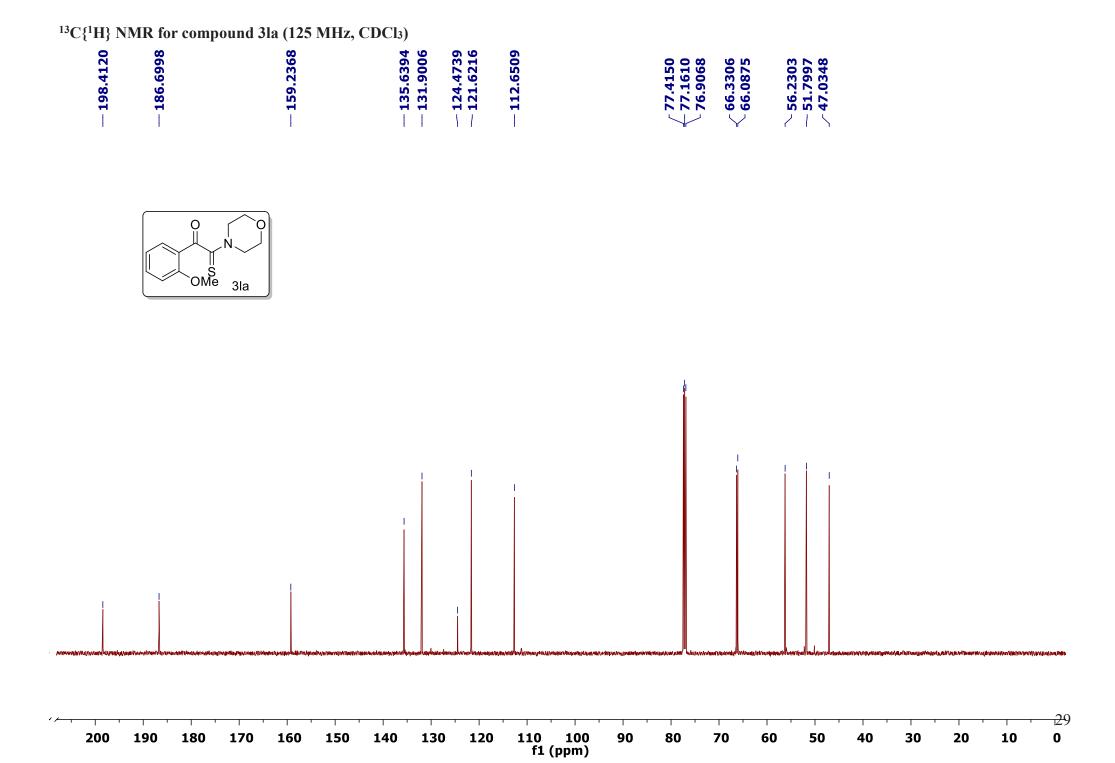


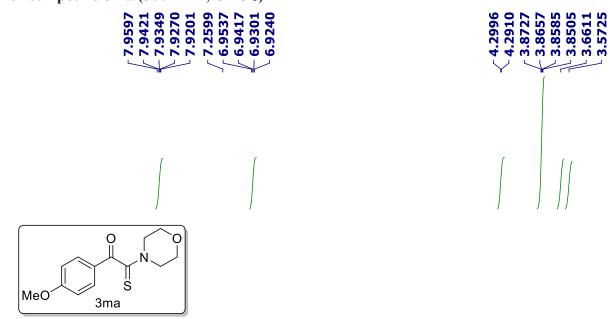


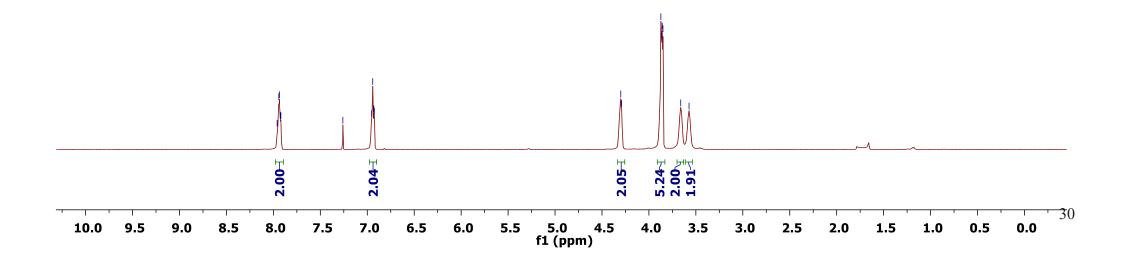


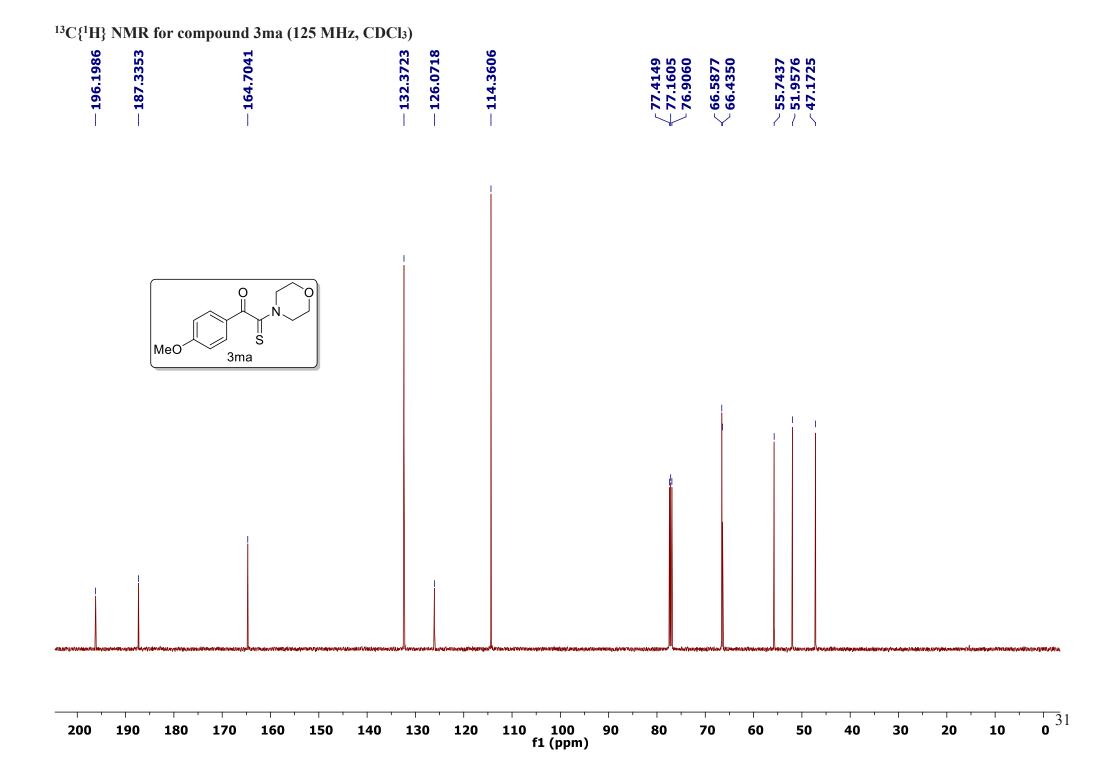


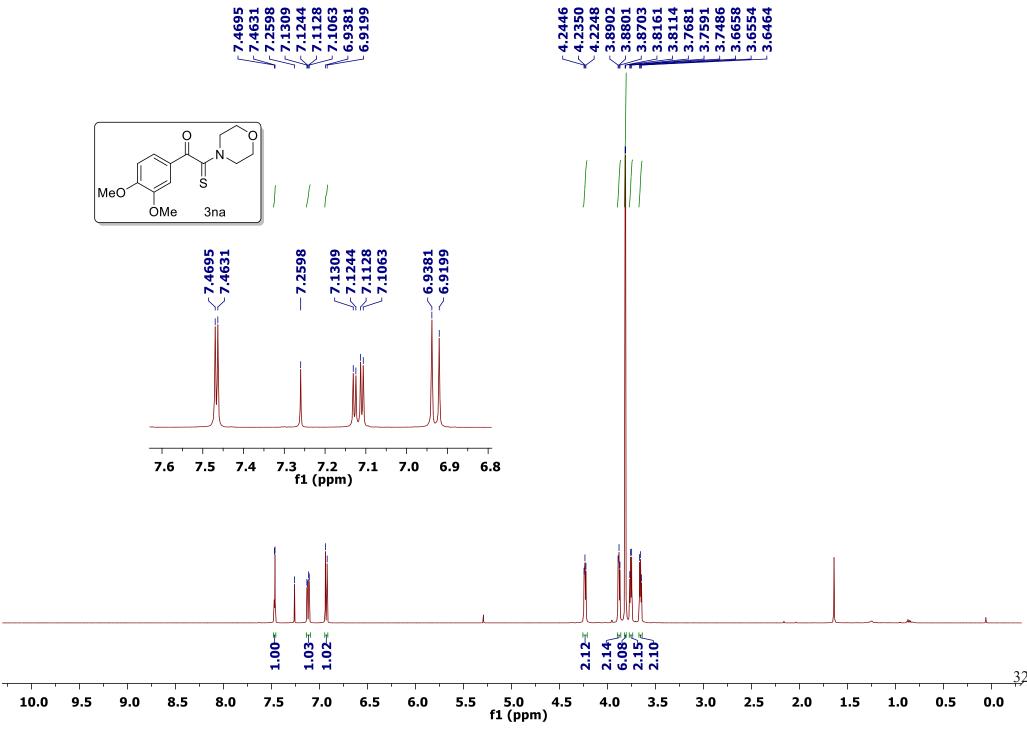


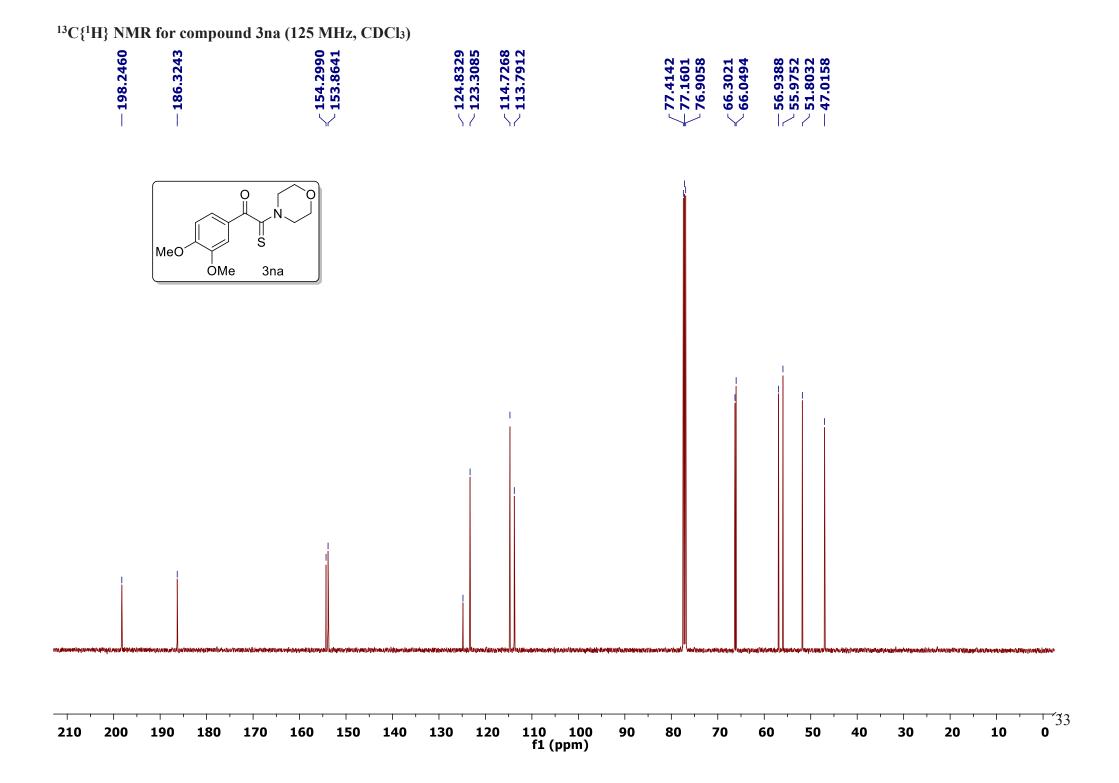












5.0

f1 (ppm)

4.5

4.0

3.5

3.0

2.5

2.0

1.5

1.0

0.5

0.0

5.5

10.0

9.5

9.0

8.5

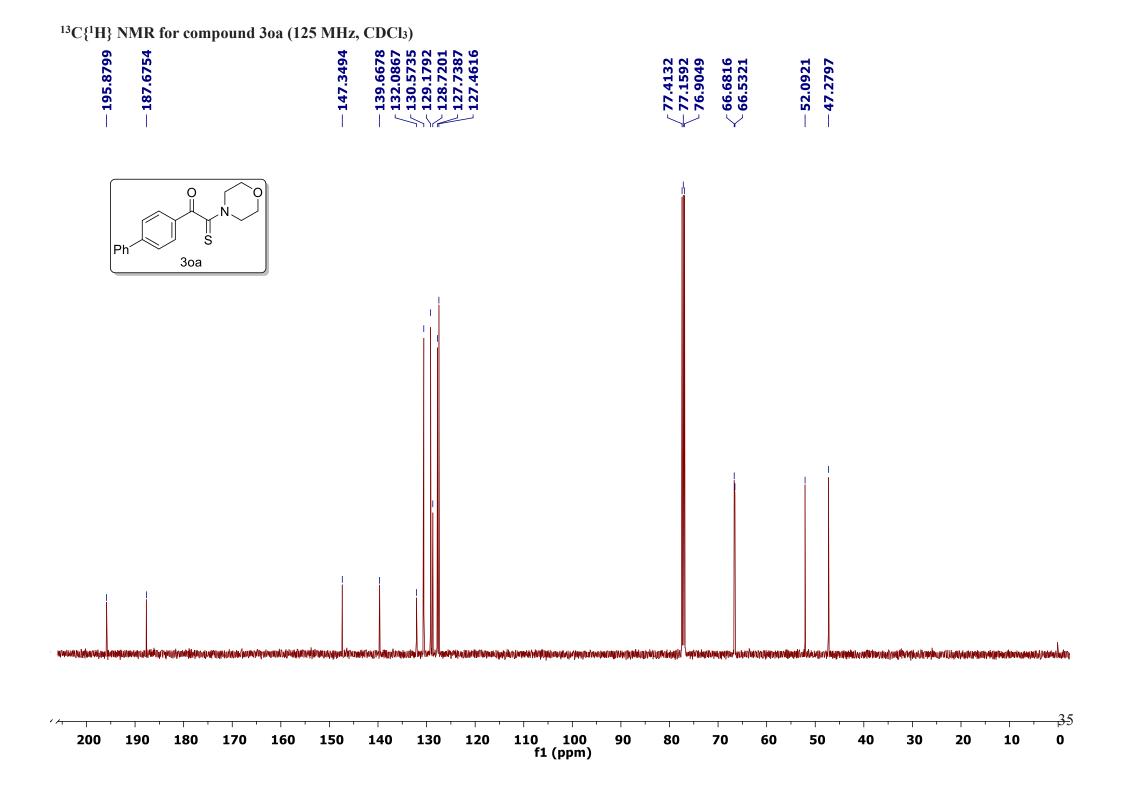
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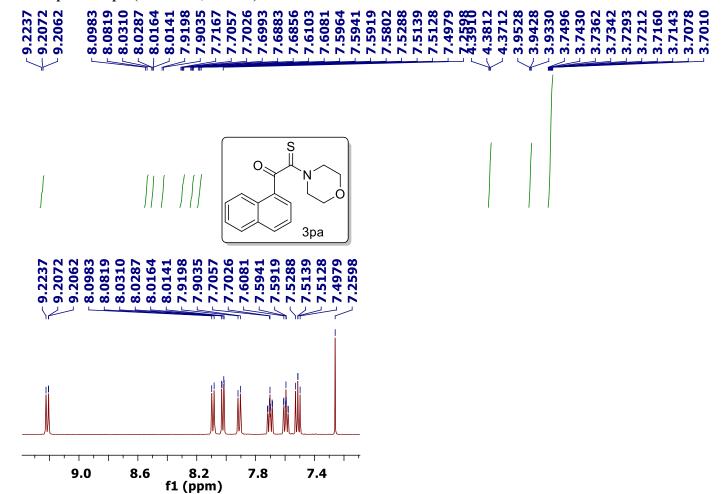
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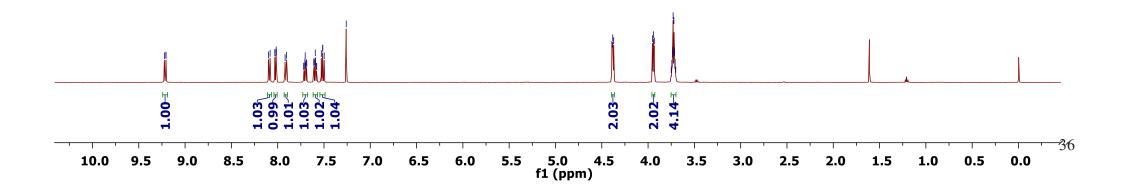
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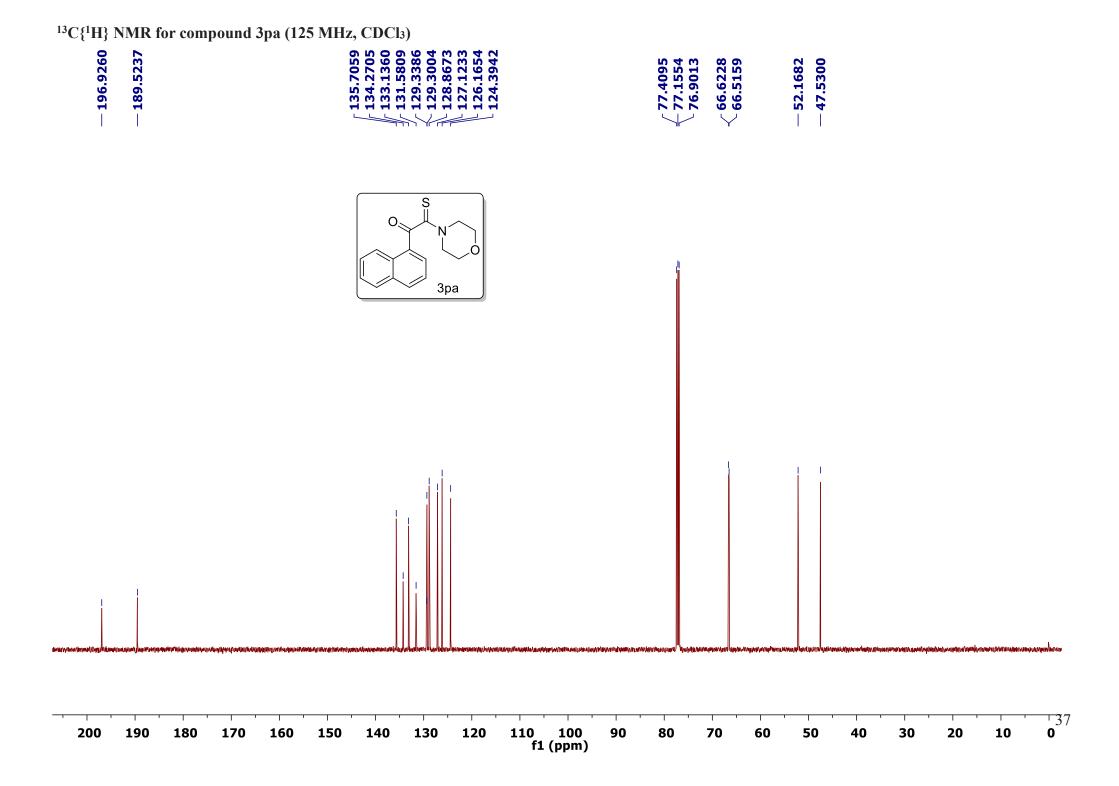
6.5

6.0









5.0 f1 (ppm)

4.0

3.5

3.0

2.5

2.0

1.5

1.0

0.5

10.0

9.5

9.0

8.5

8.0

7.5

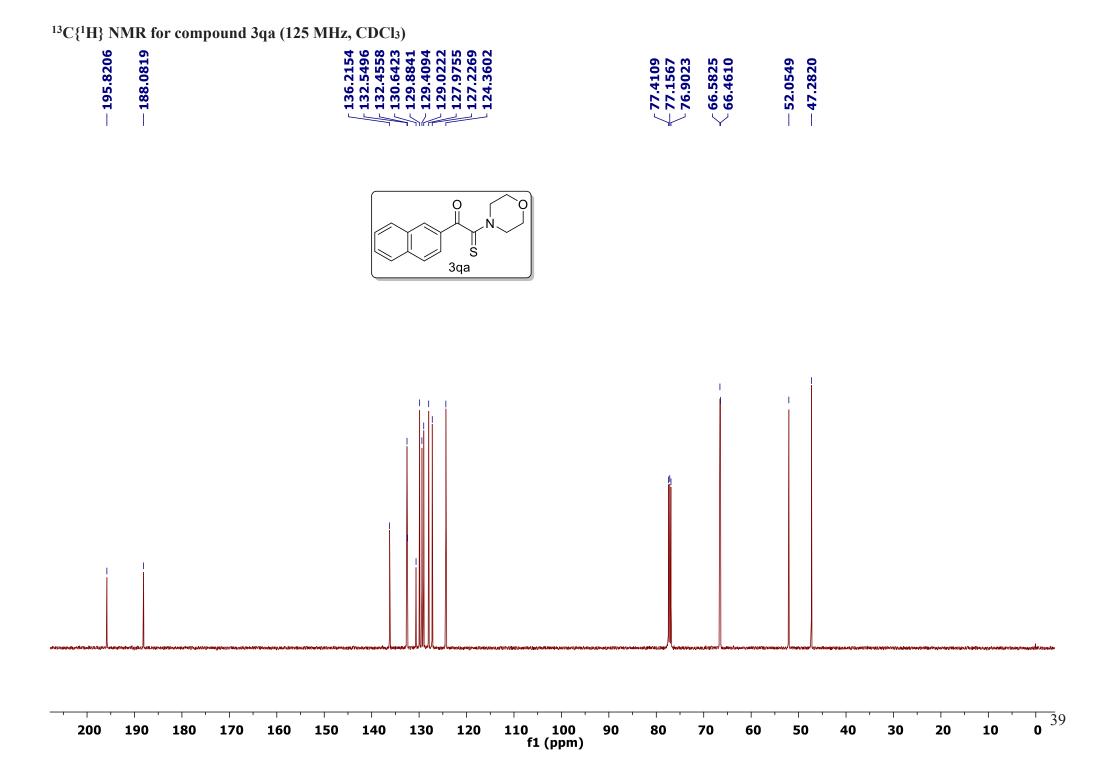
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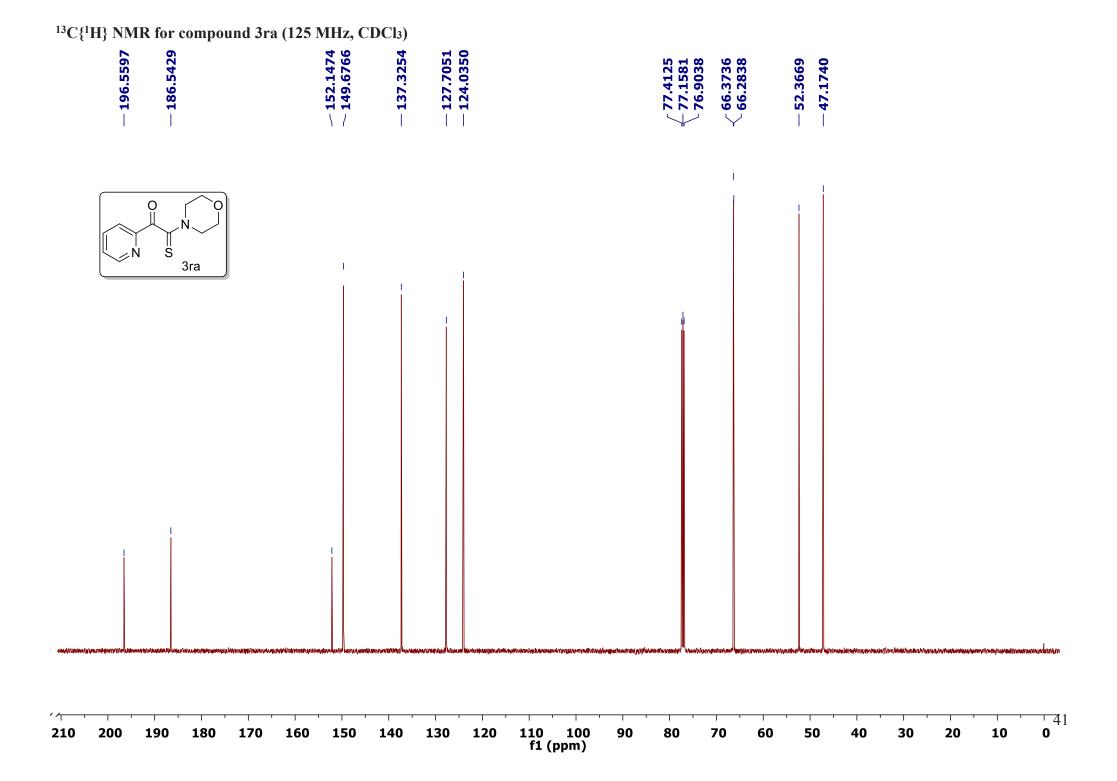
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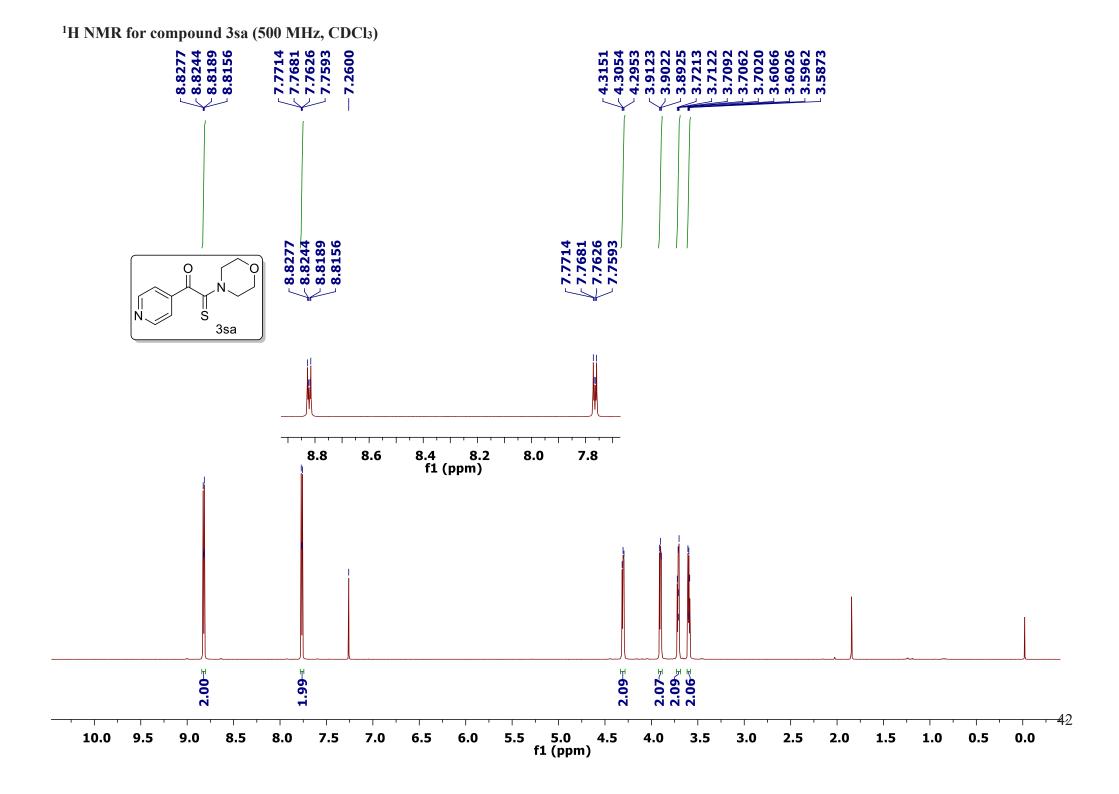
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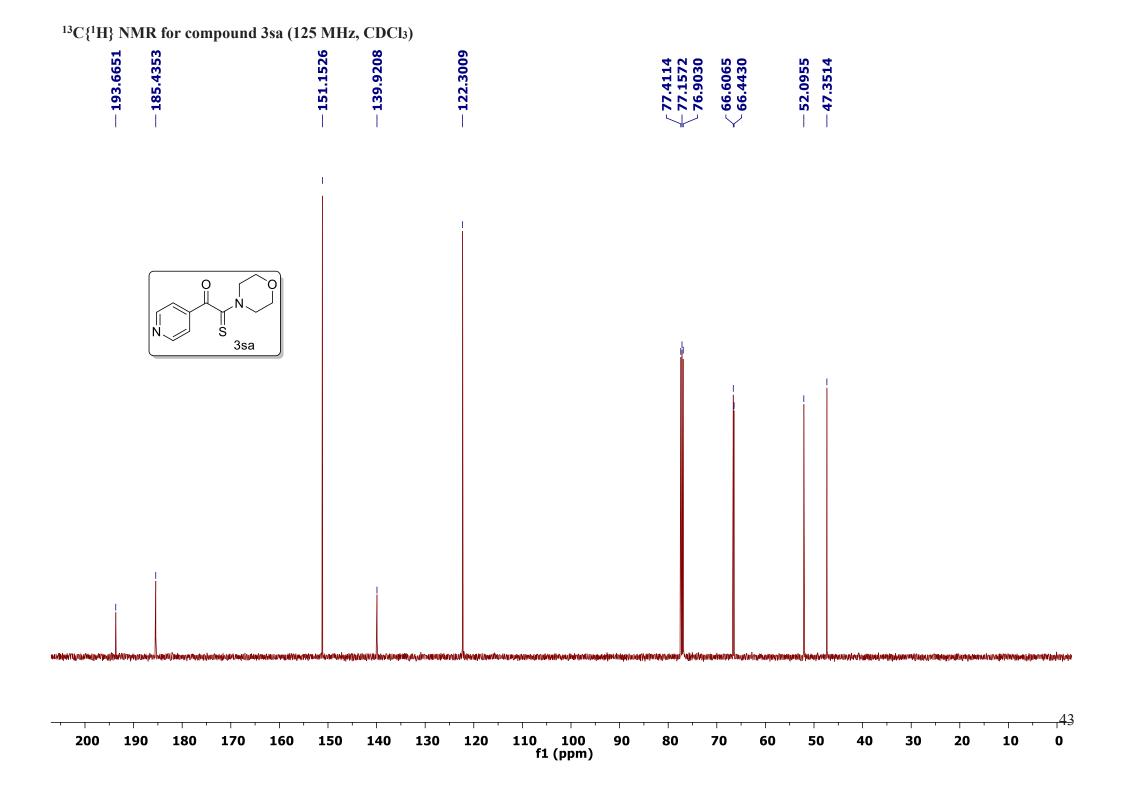
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0.0









5.5

10.0

9.5

9.0

8.5

8.0

7.5

7.0

6.5

6.0

5.0

f1 (ppm)

4.5

4.0

3.5

3.0

2.5

2.0

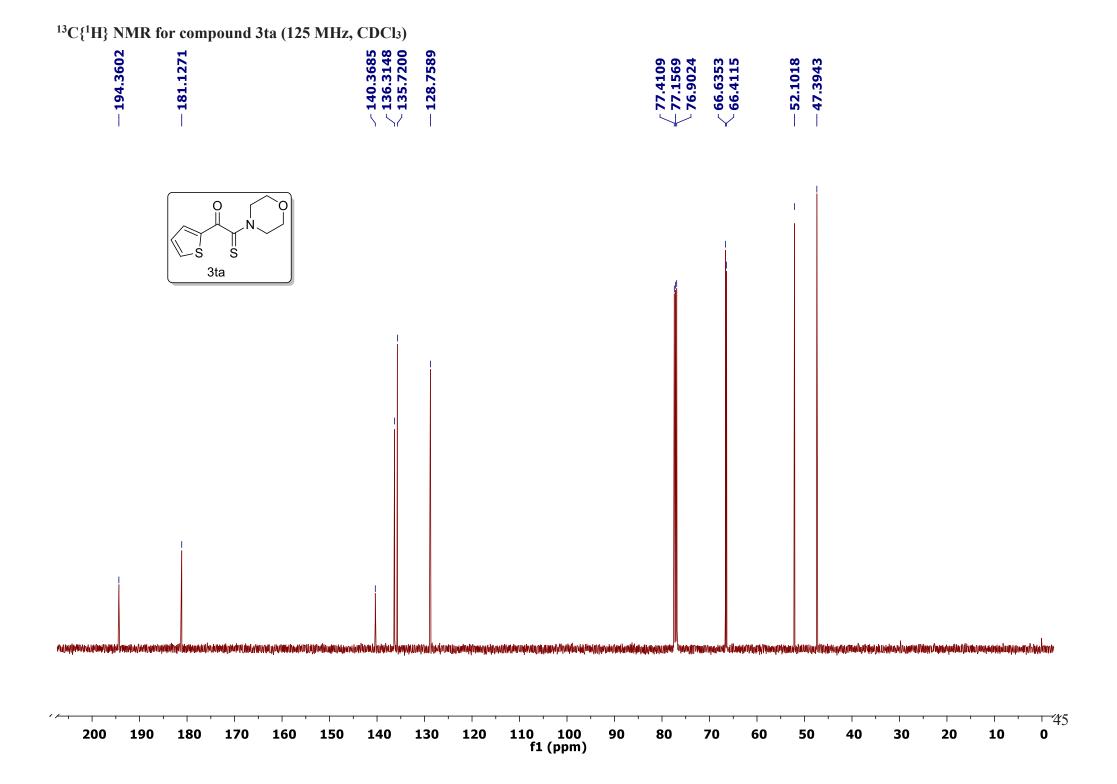
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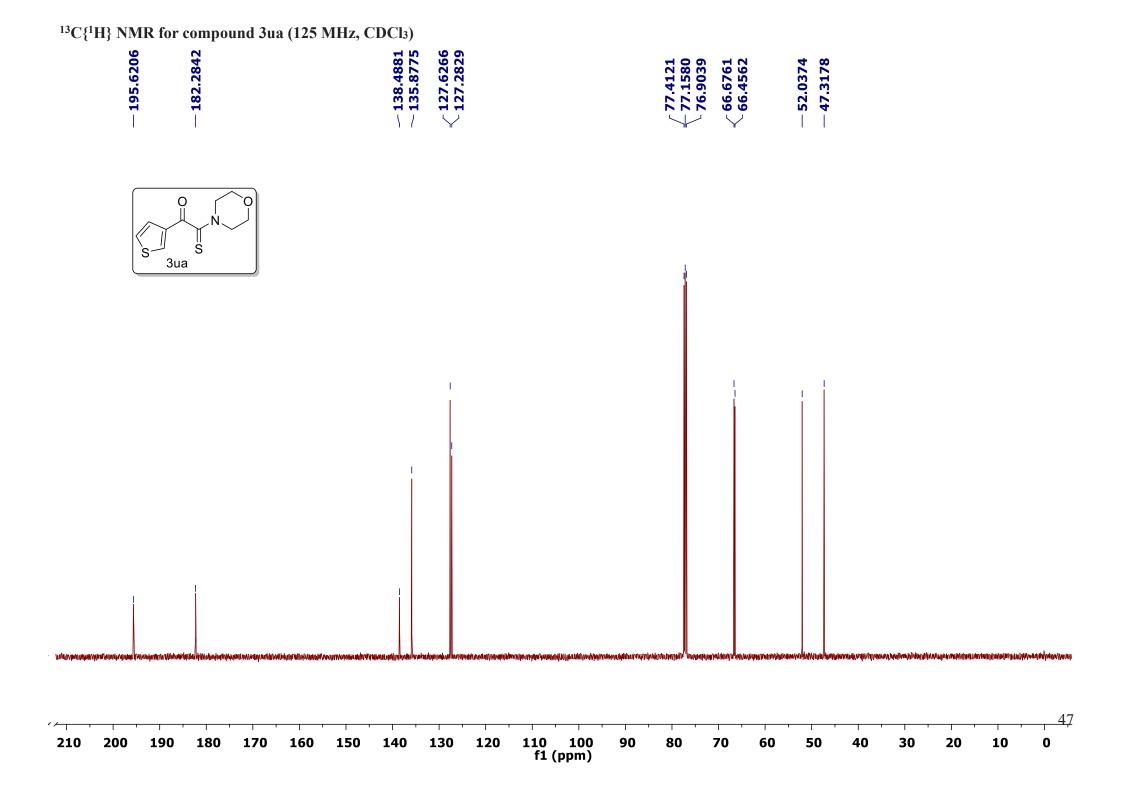
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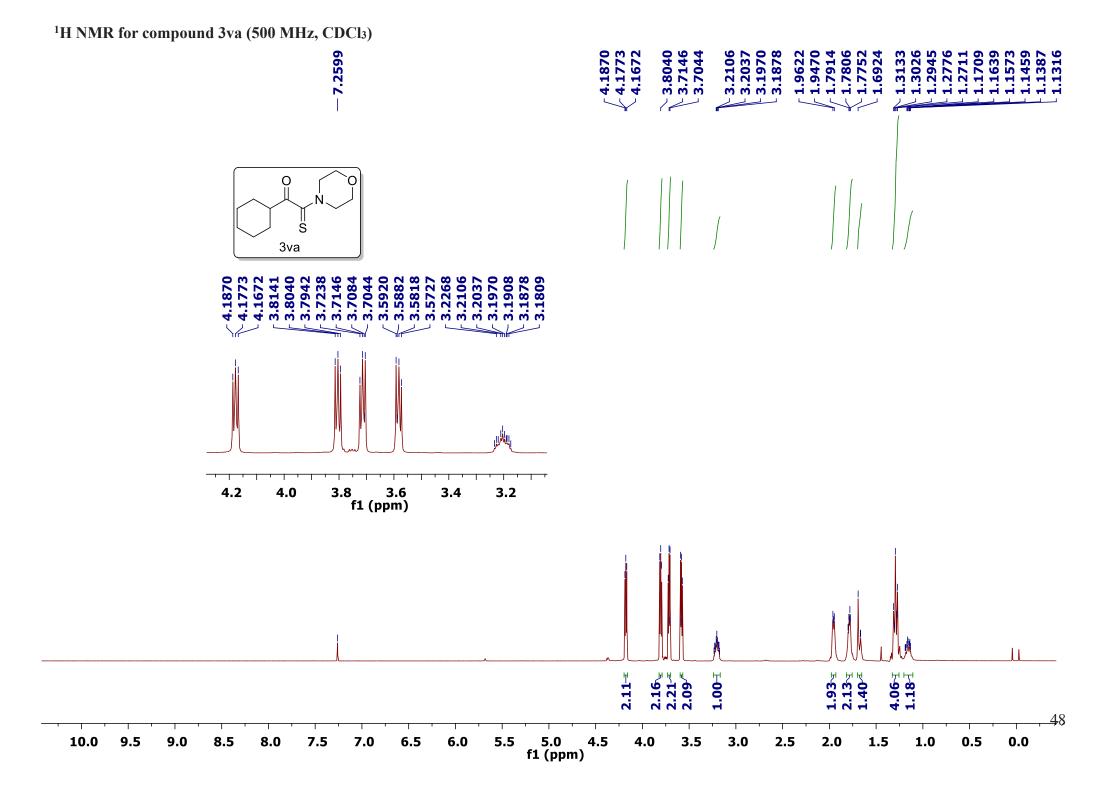
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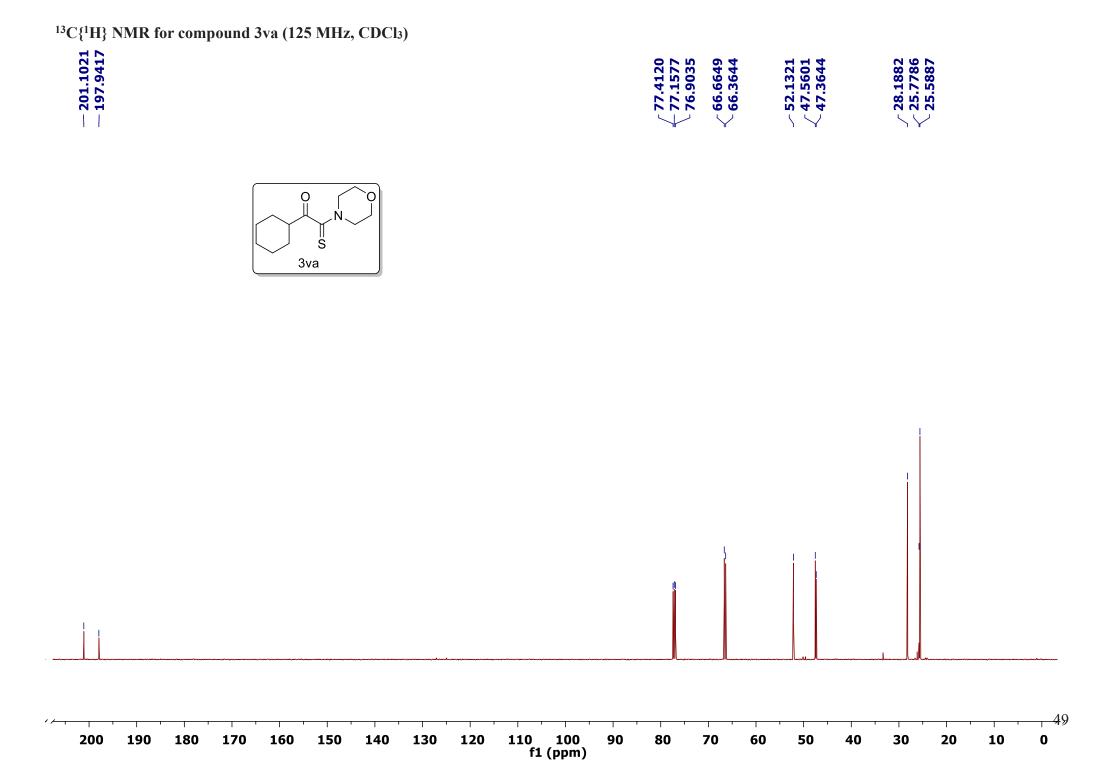
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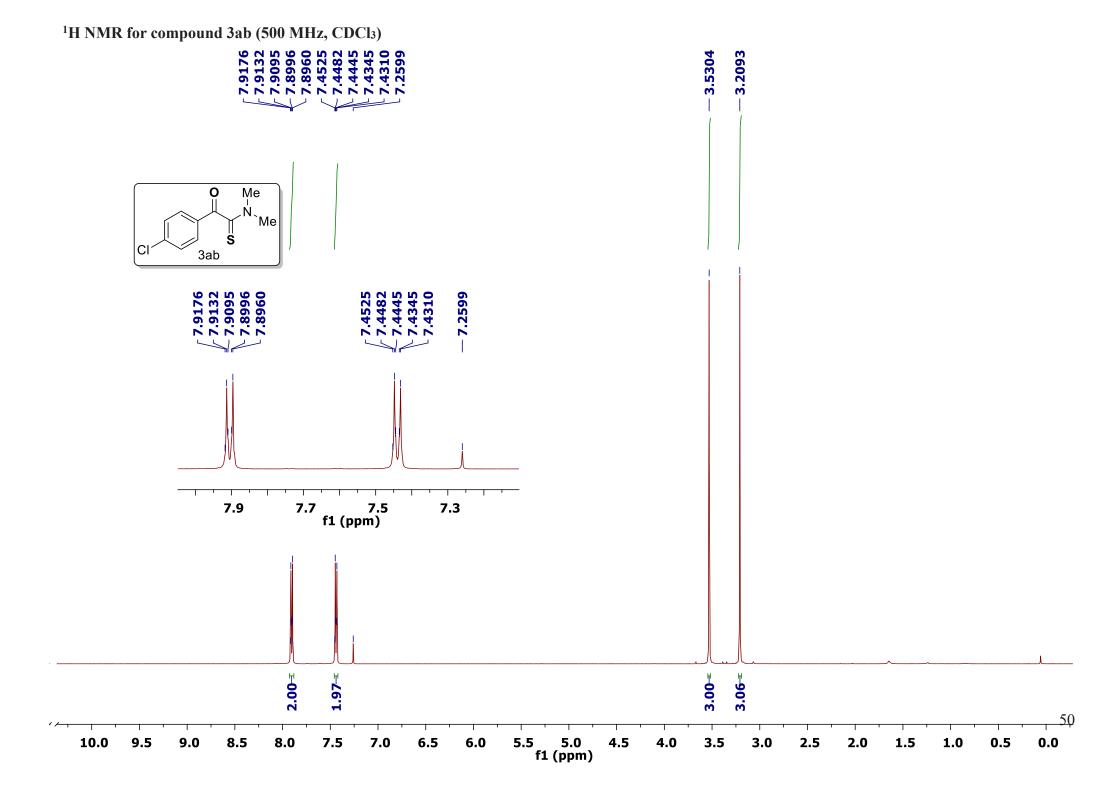
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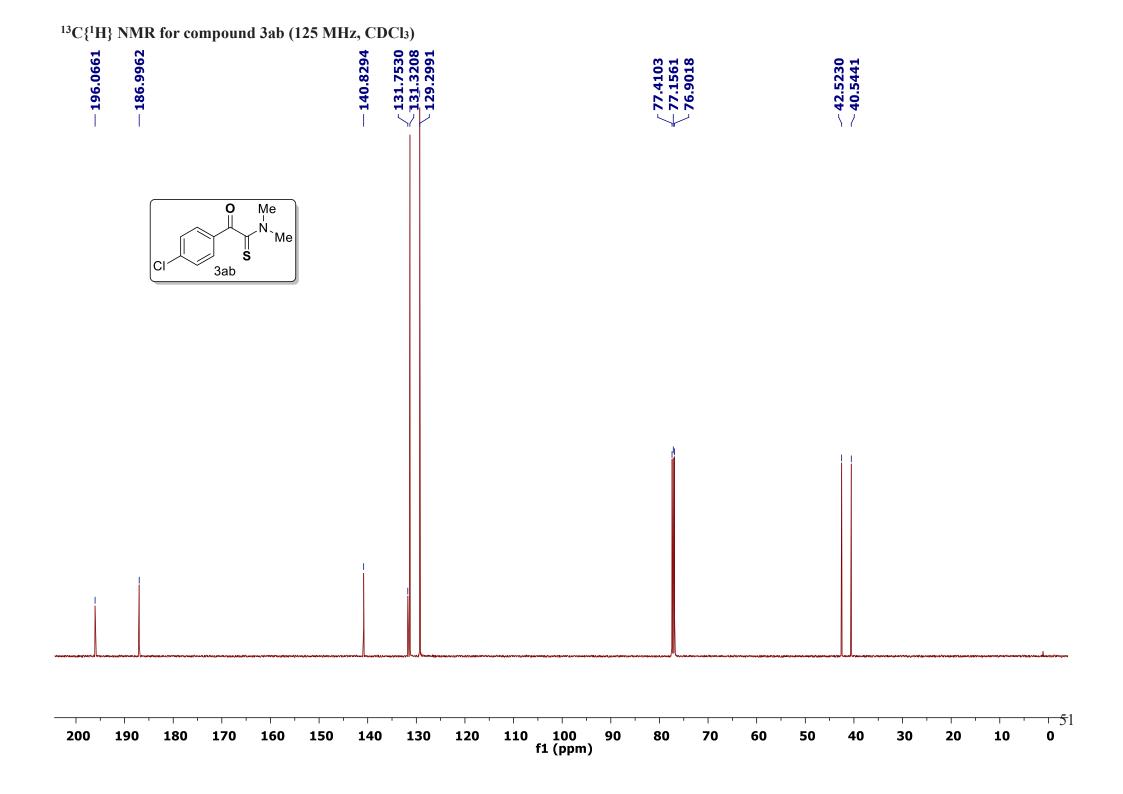


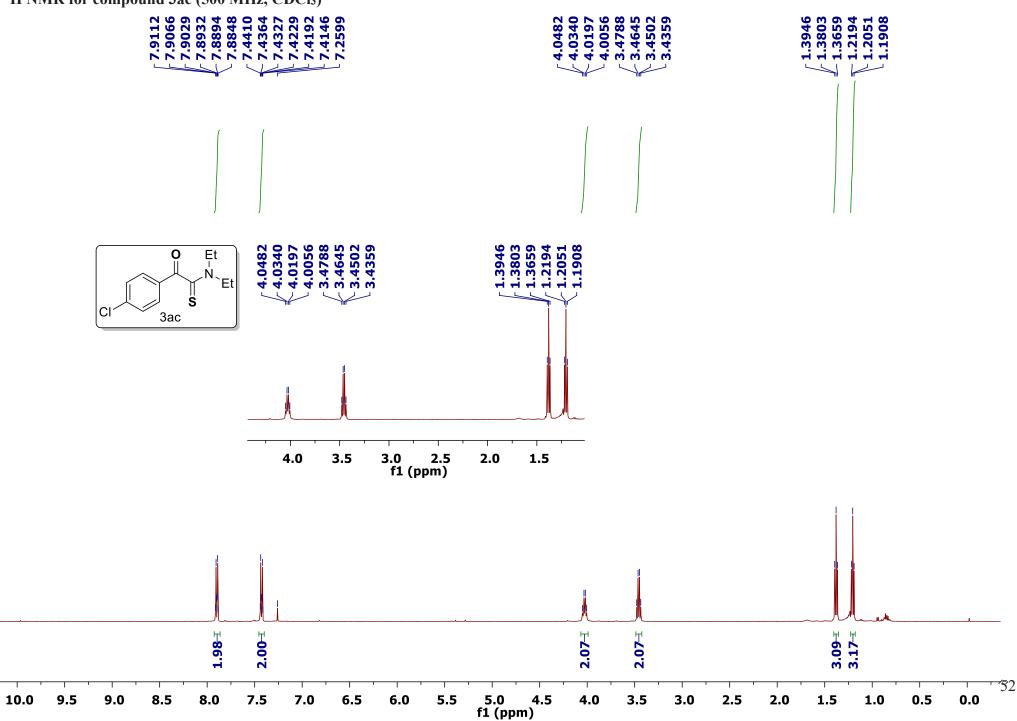


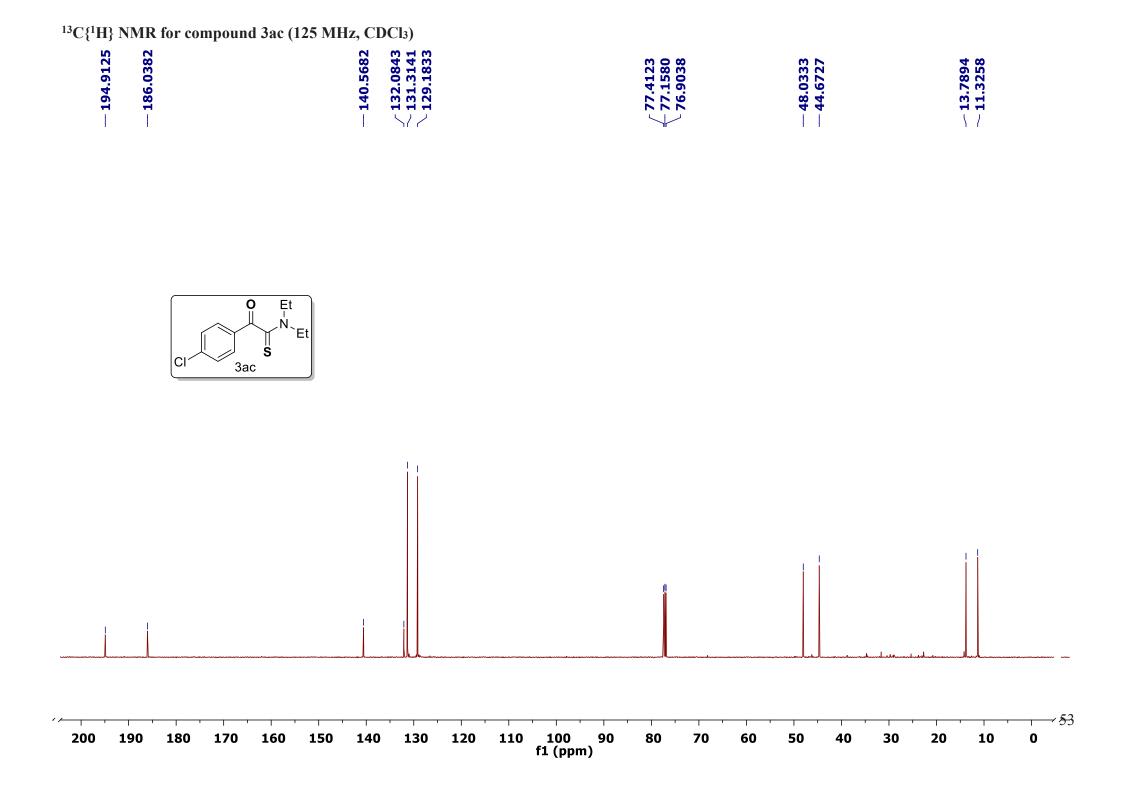


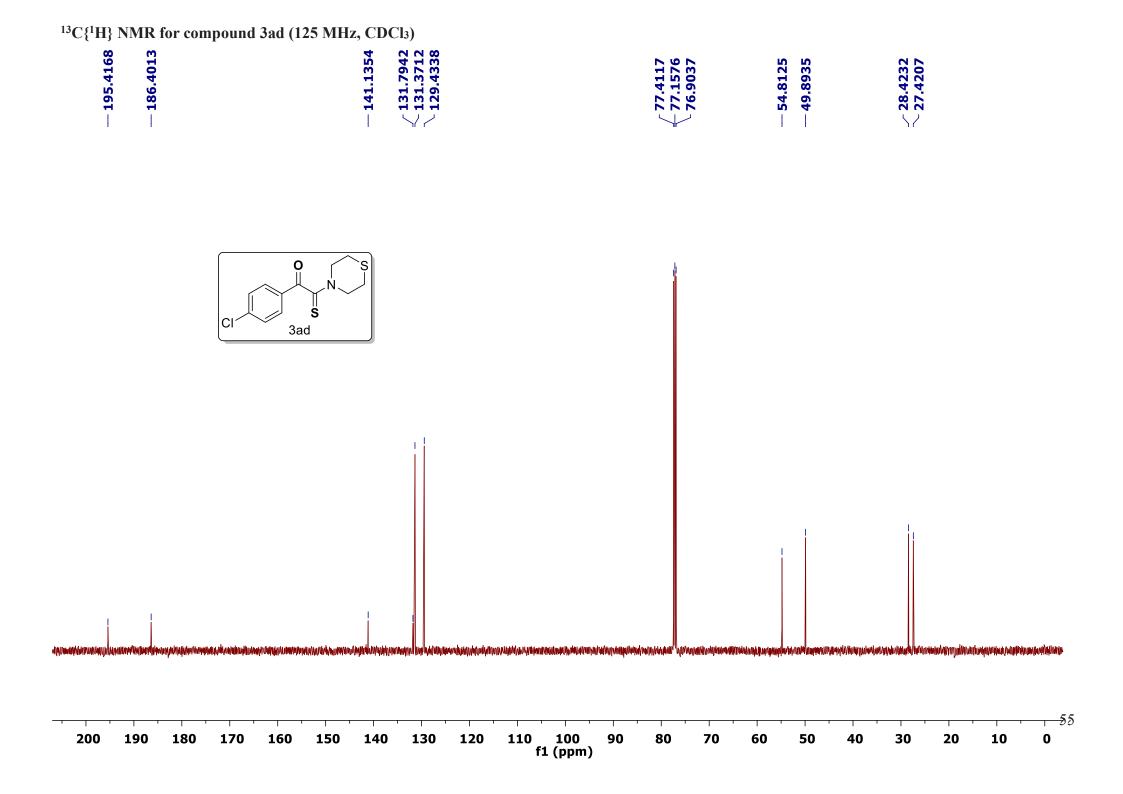




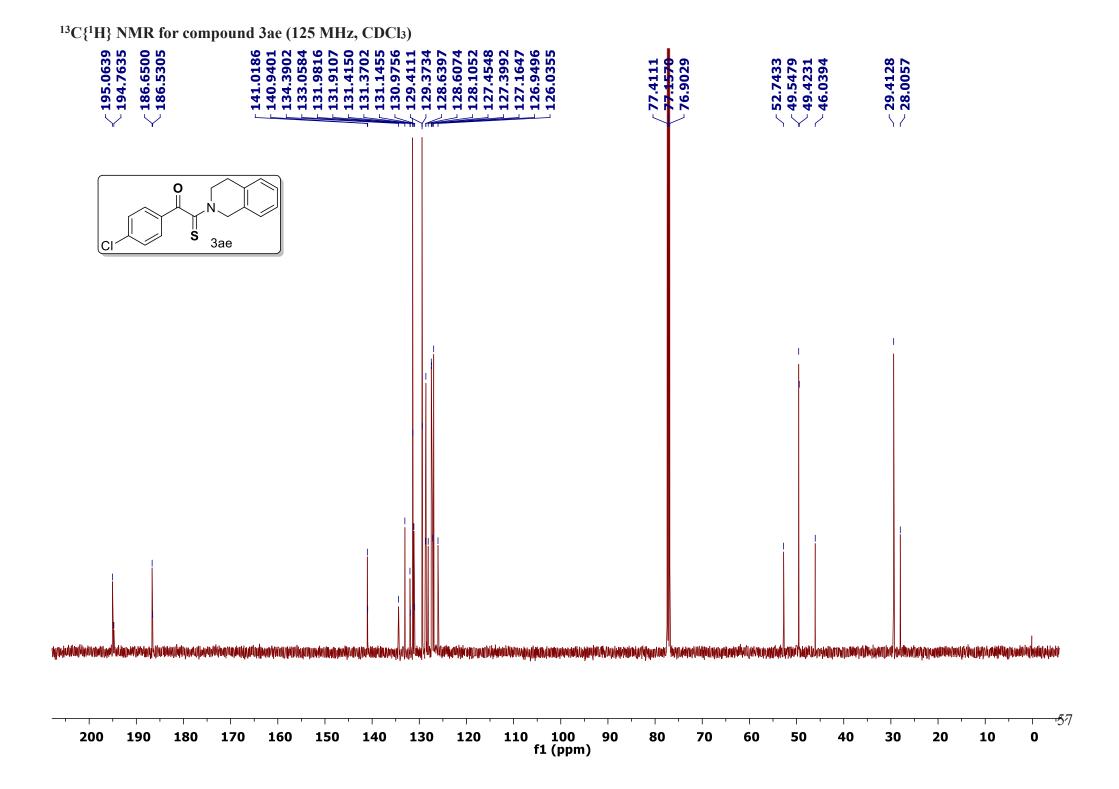


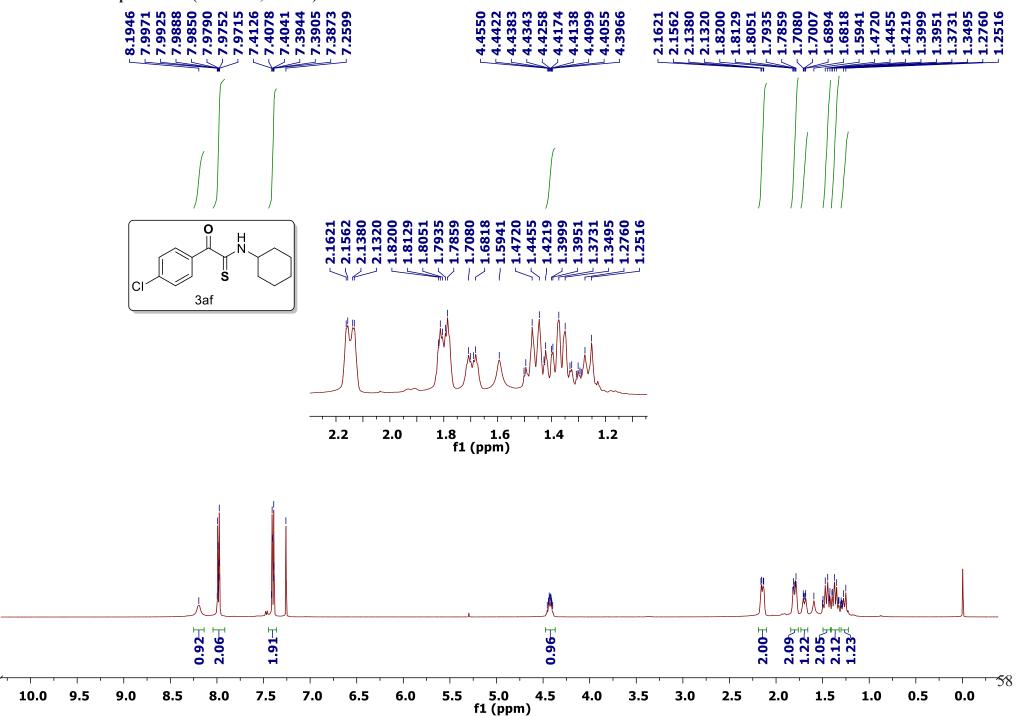


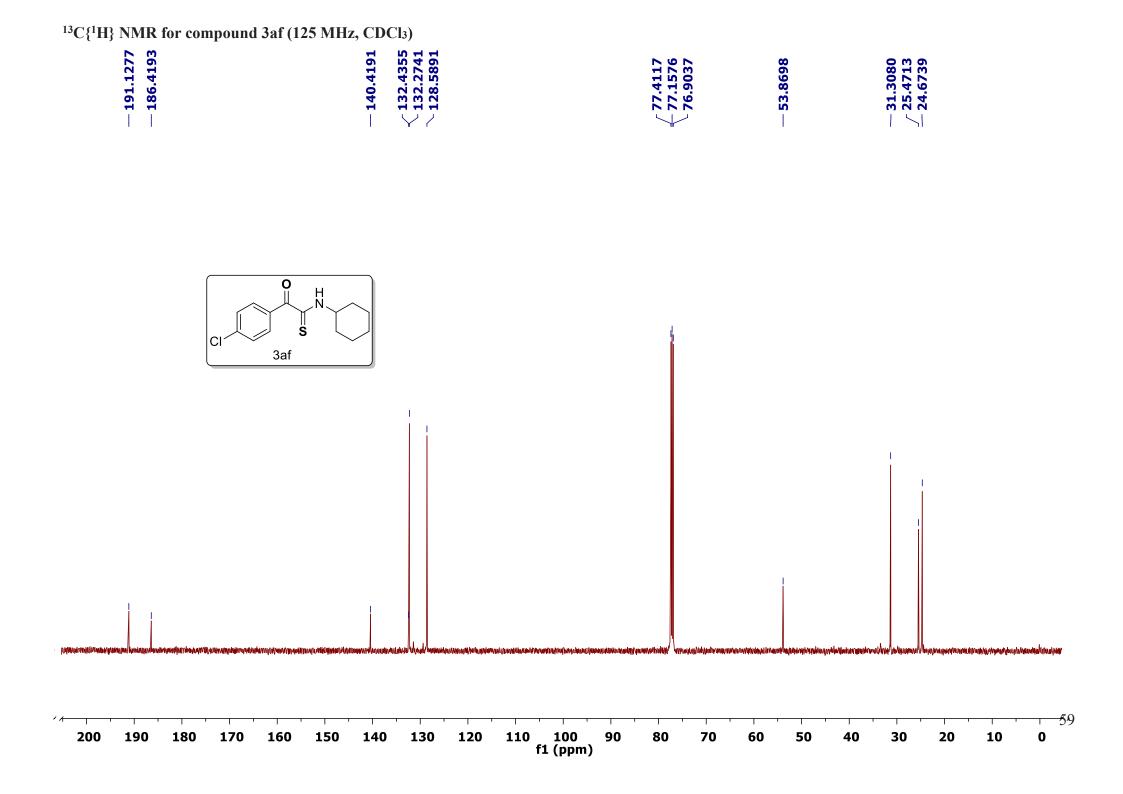




f1 (ppm)







5.0 f1 (ppm)

4.5

4.0

3.5

3.0

2.5

2.0

1.5

1.0

0.5

10.0

9.5

9.0

8.5

8.0

7.5

7.0

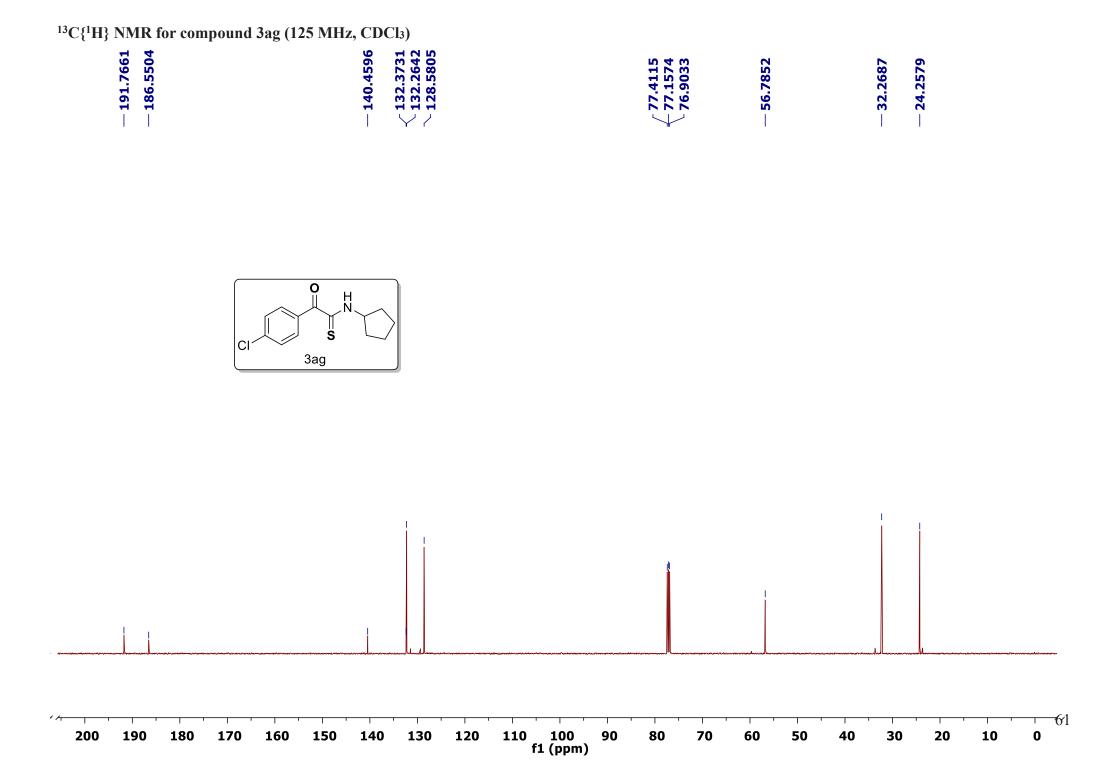
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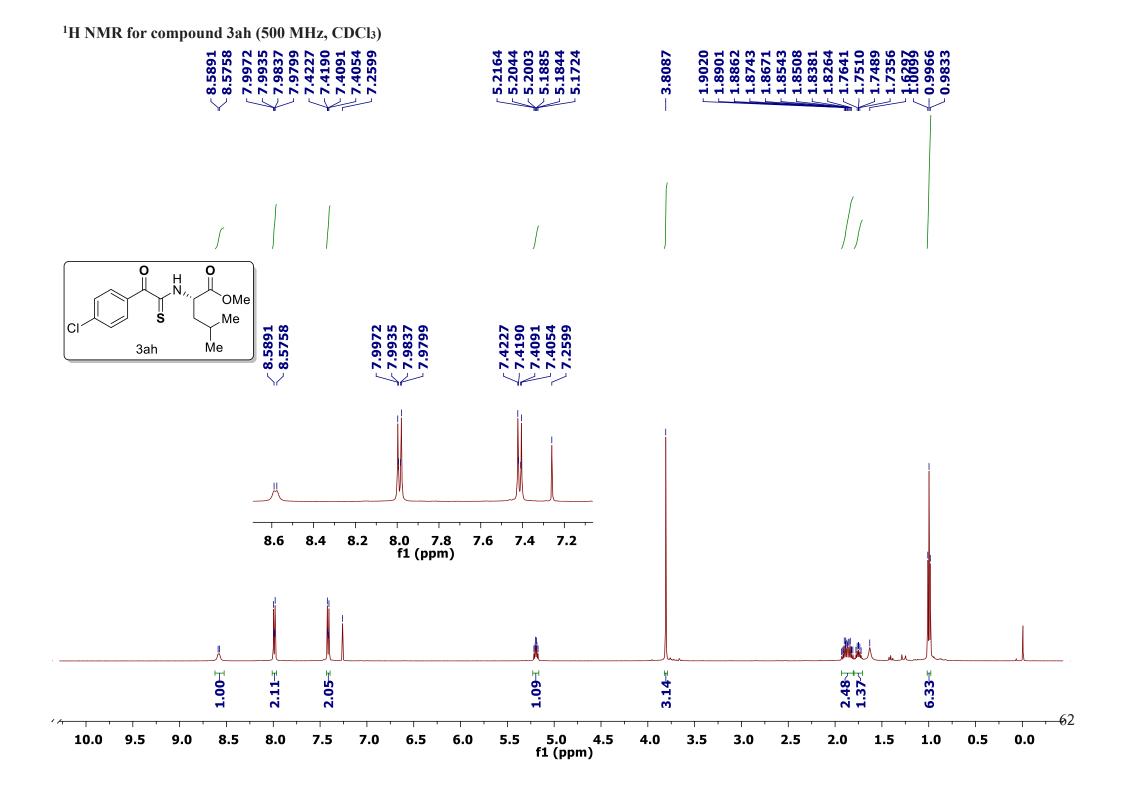
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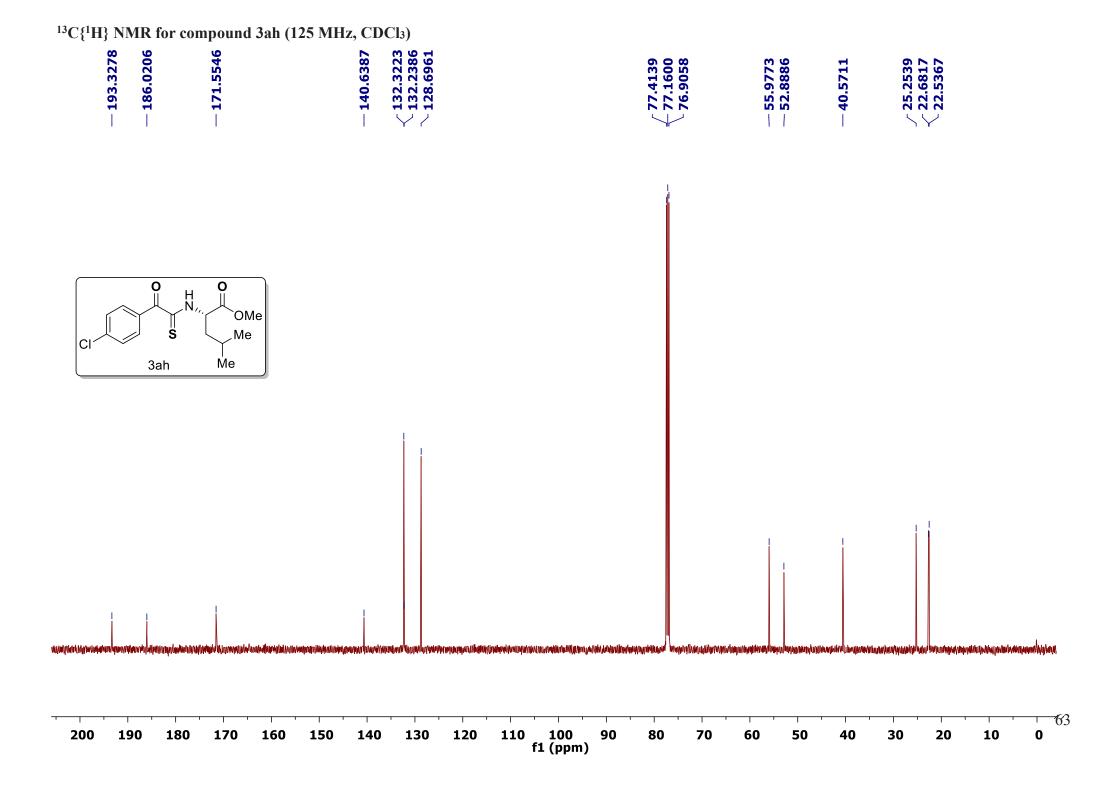
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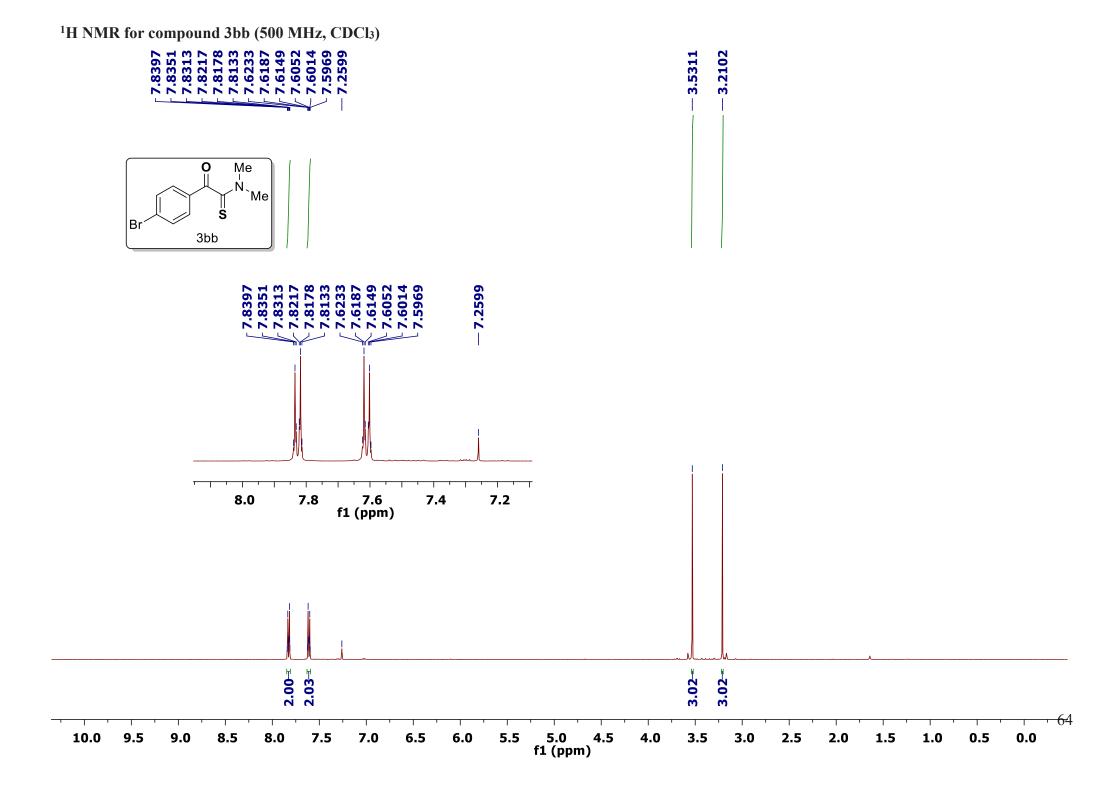
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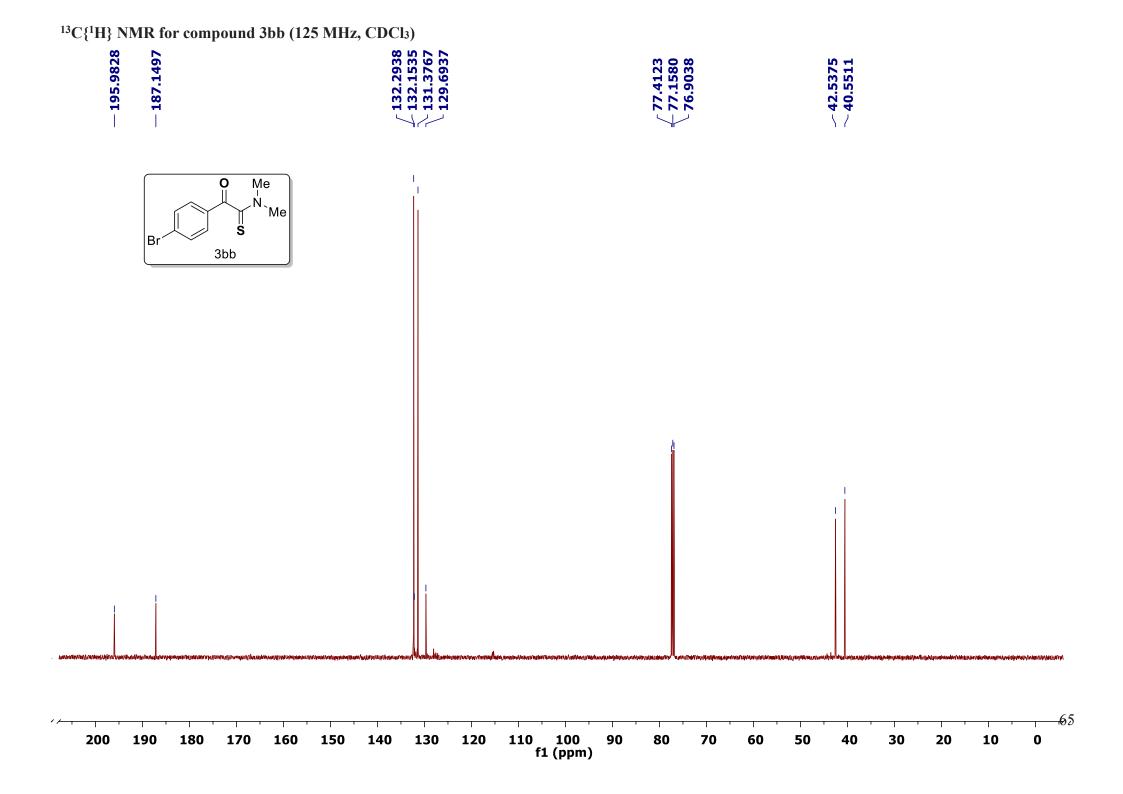
0.0











10.0

9.5

9.0

8.5

8.0

7.5

7.0

6.5

5.5

6.0

5.0

f1 (ppm)

4.5

3.5

4.0

3.0

2.5

2.0

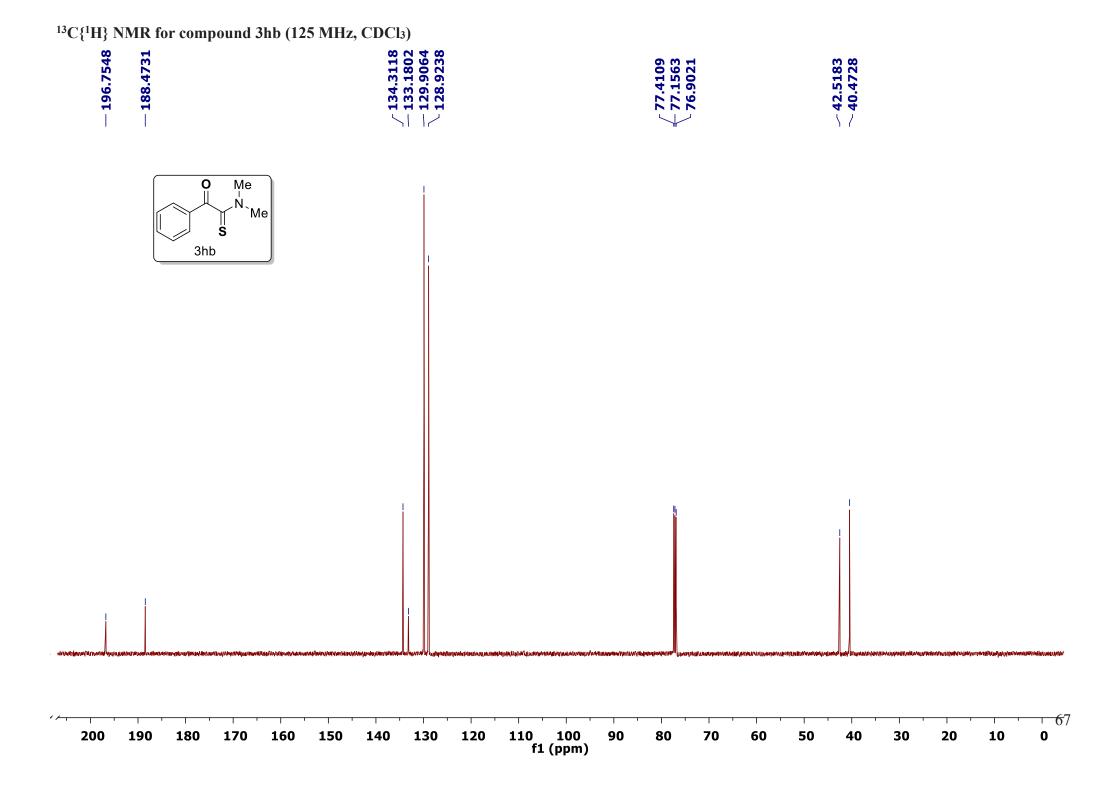
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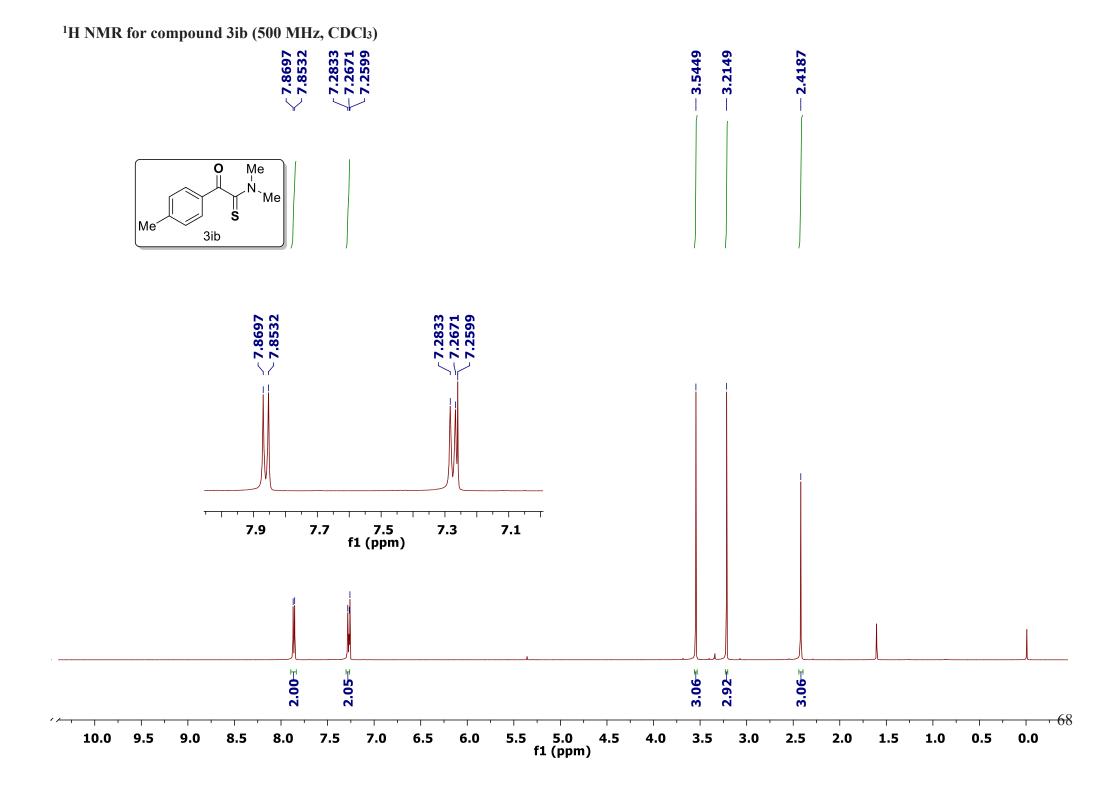
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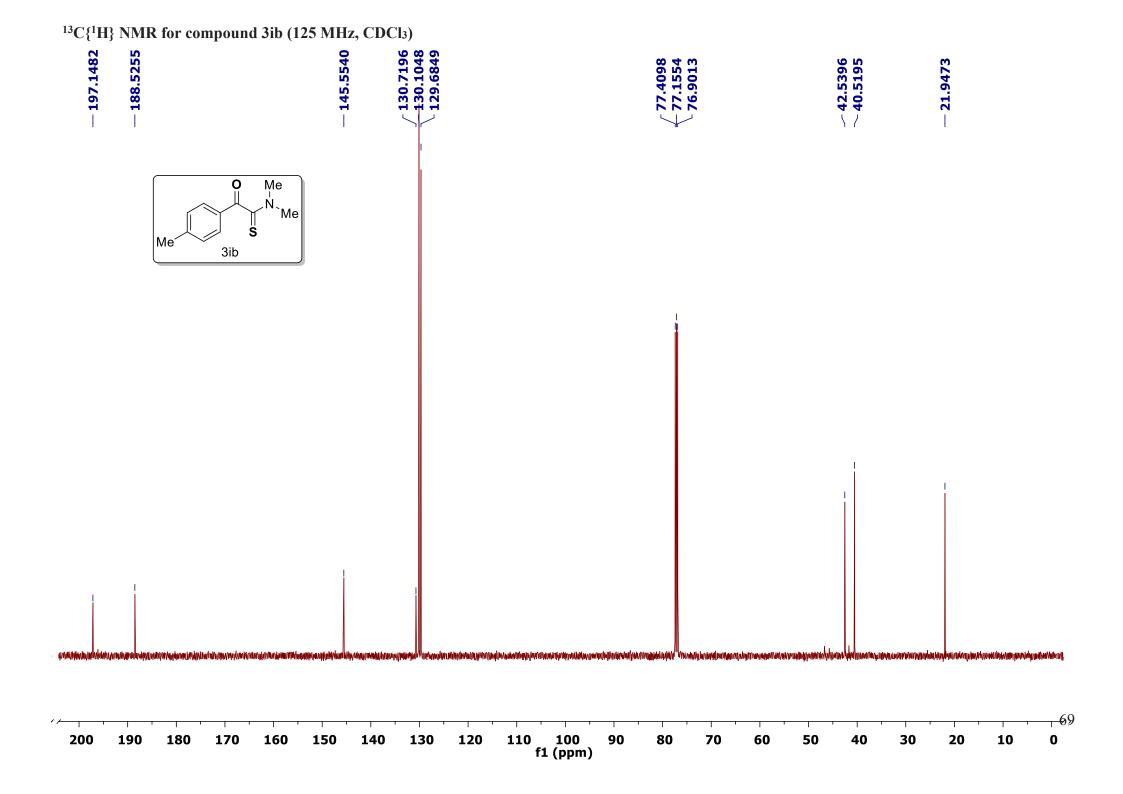
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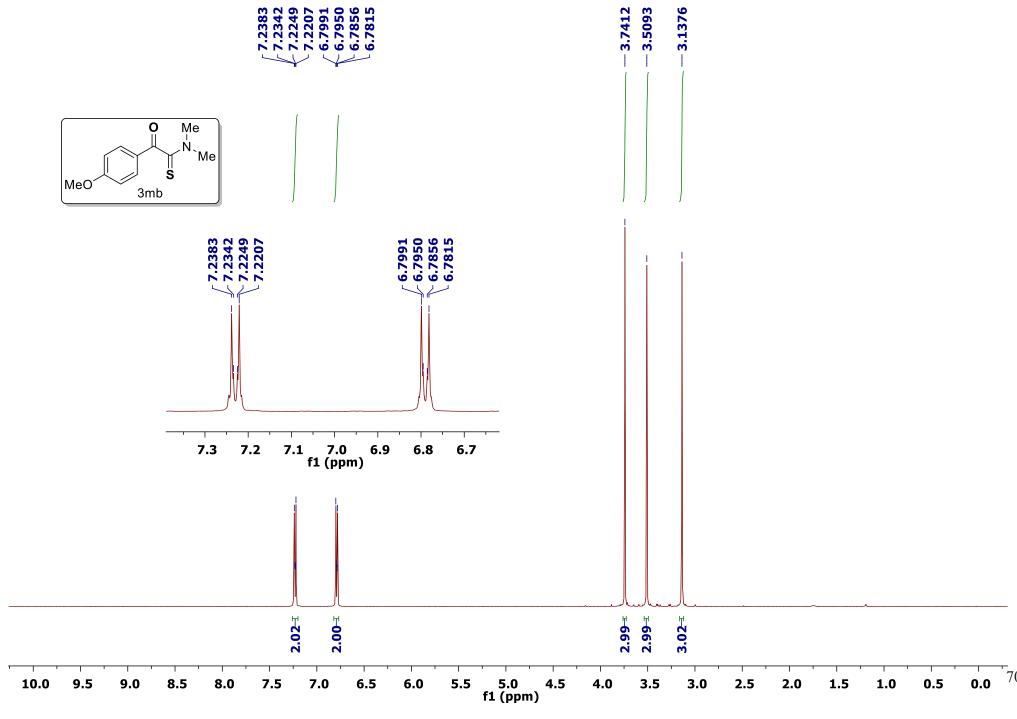
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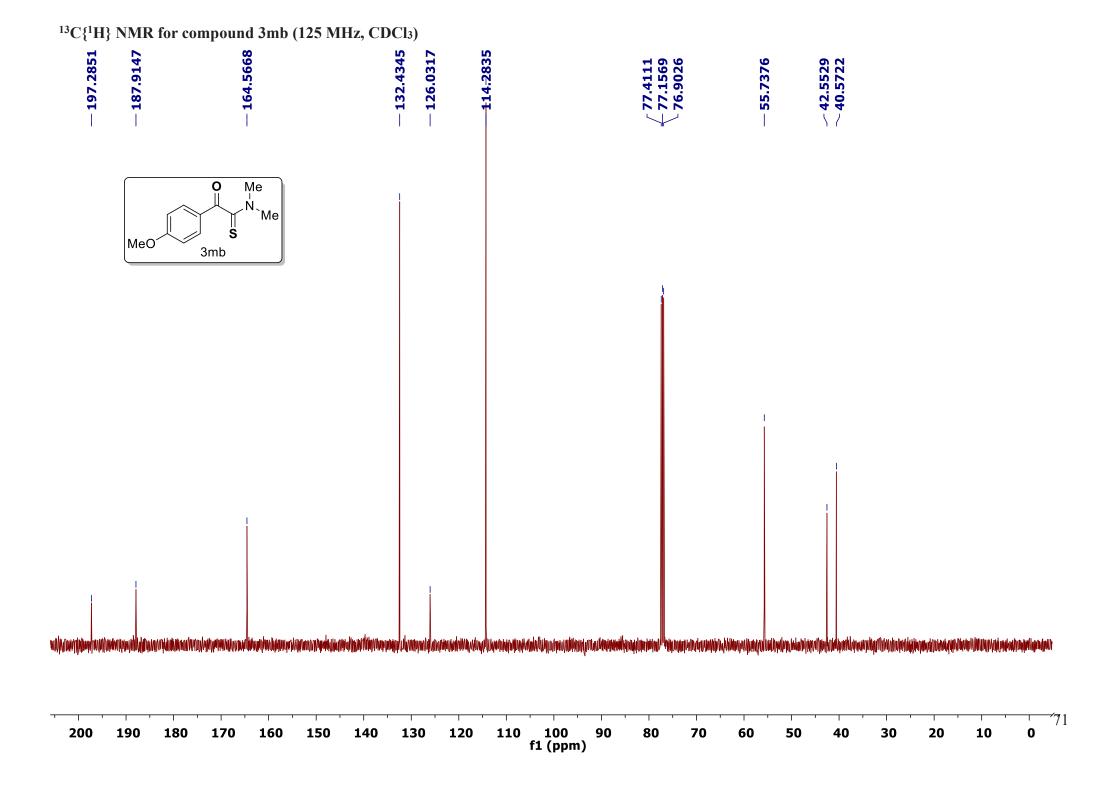
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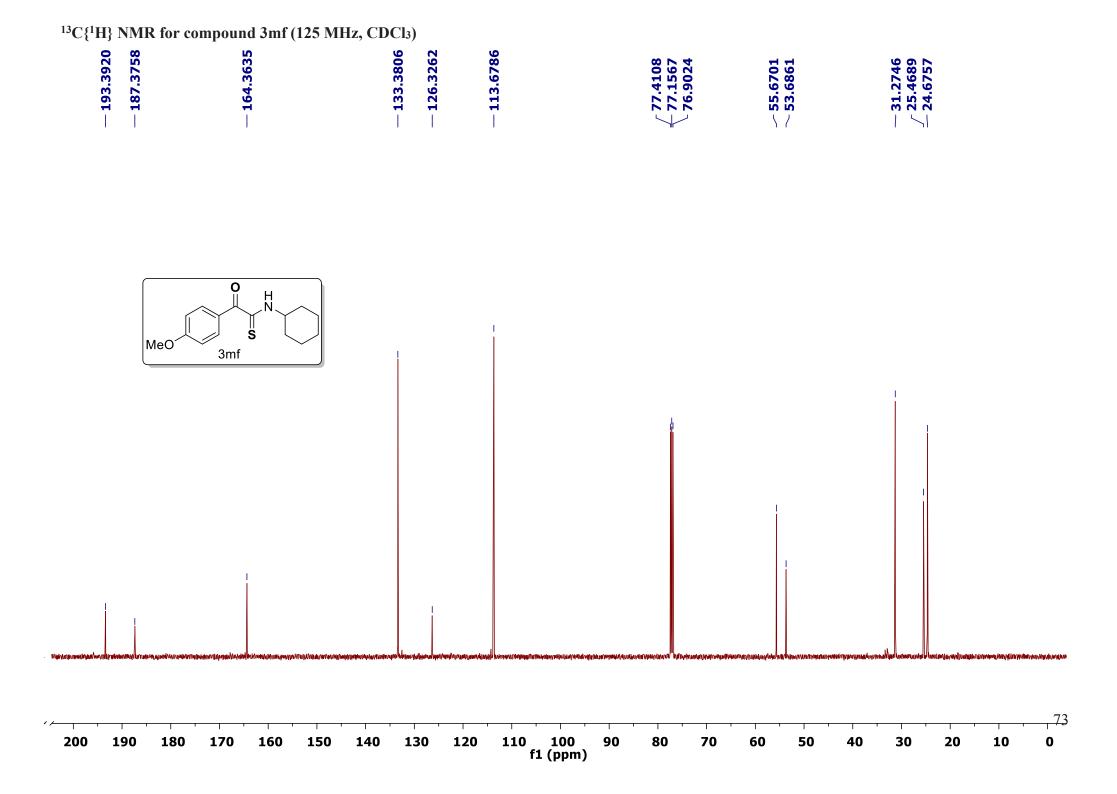


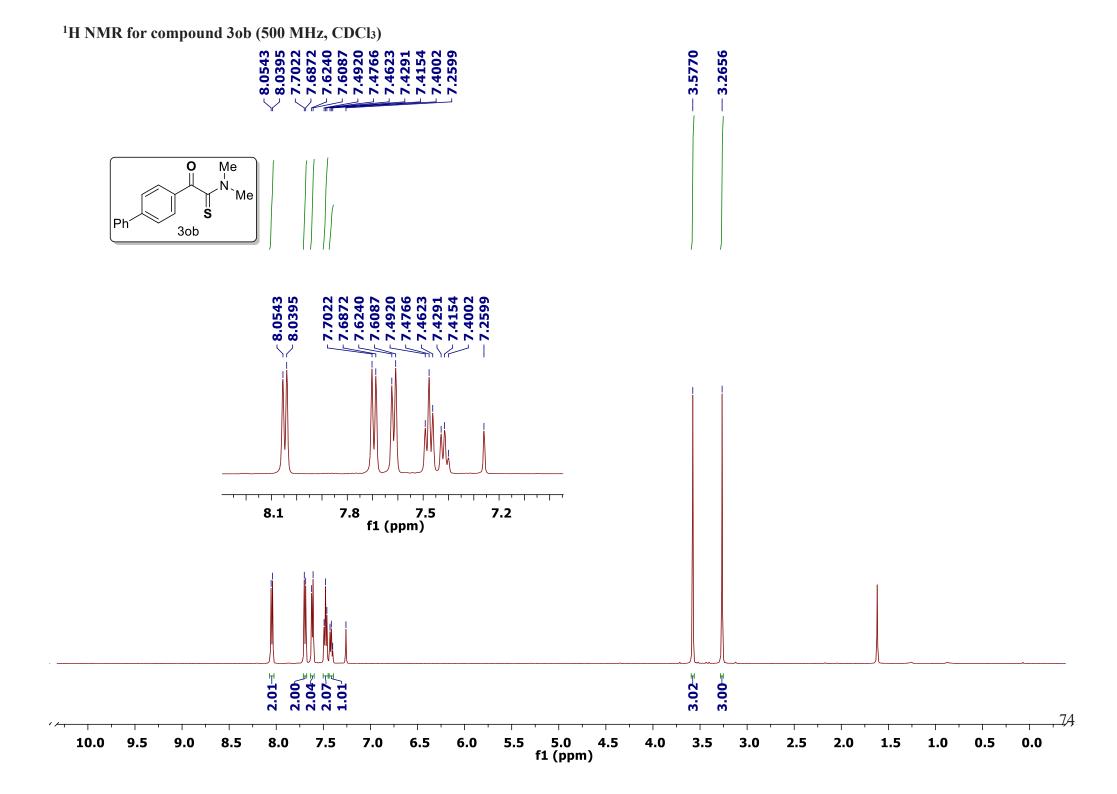


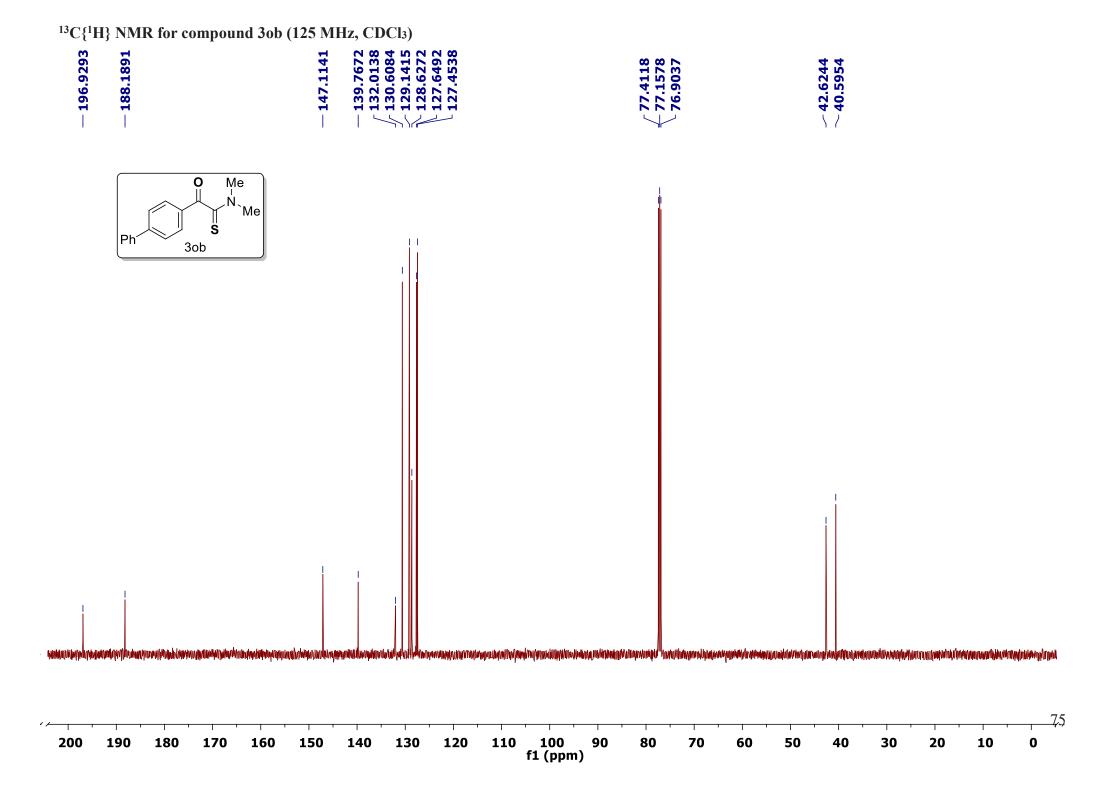


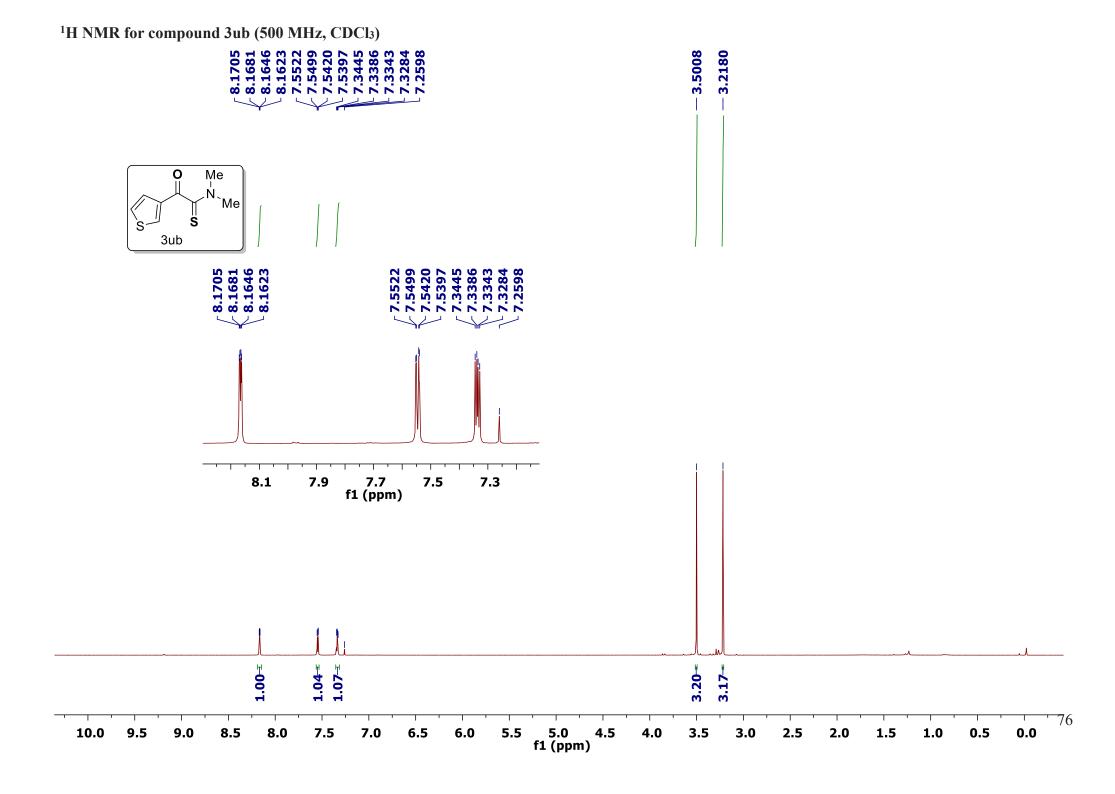


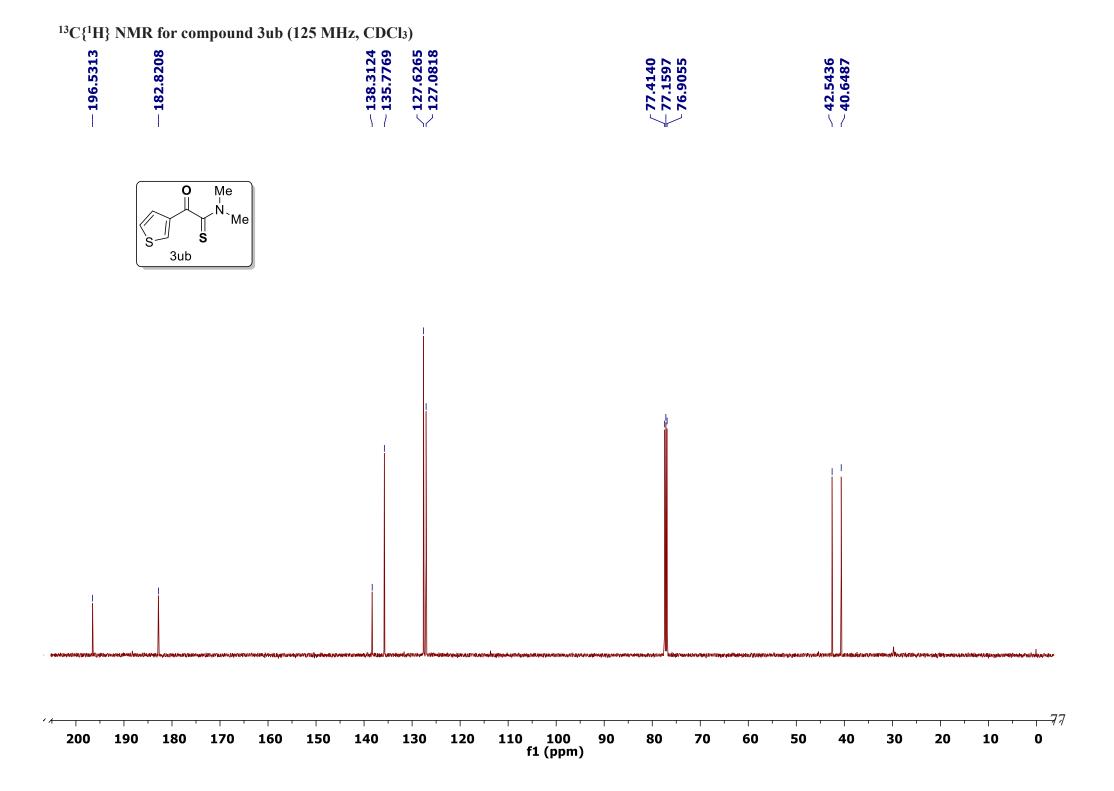


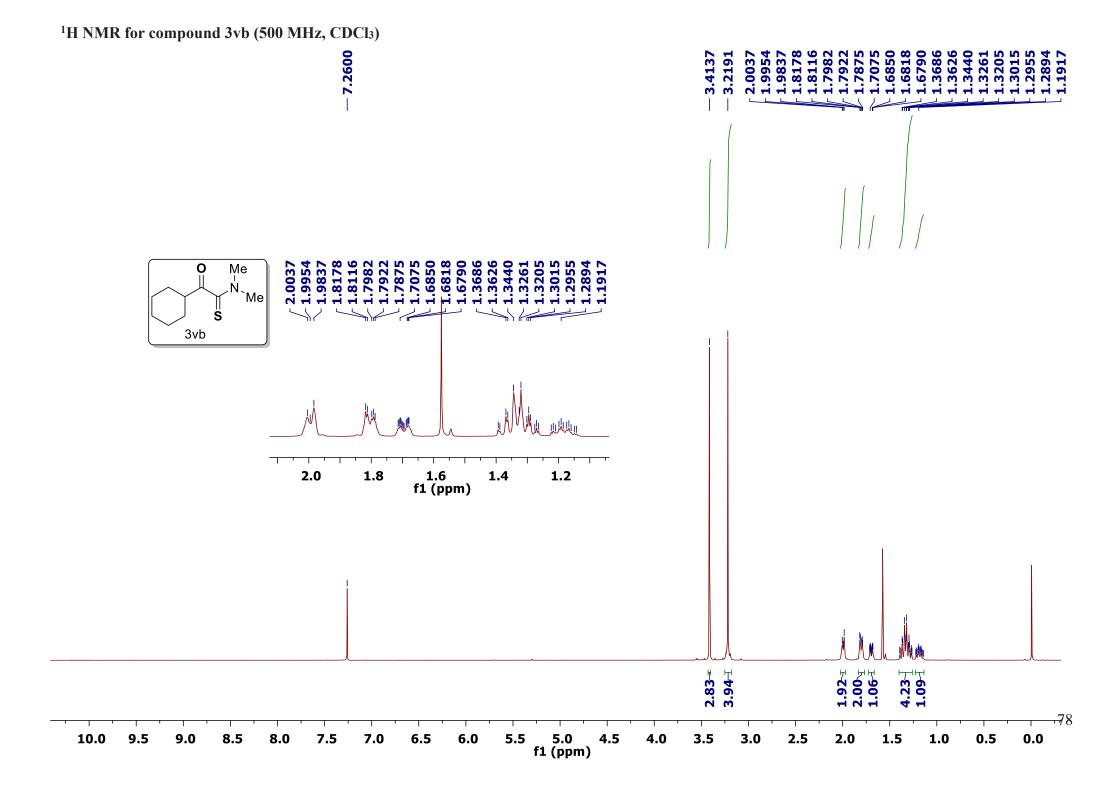


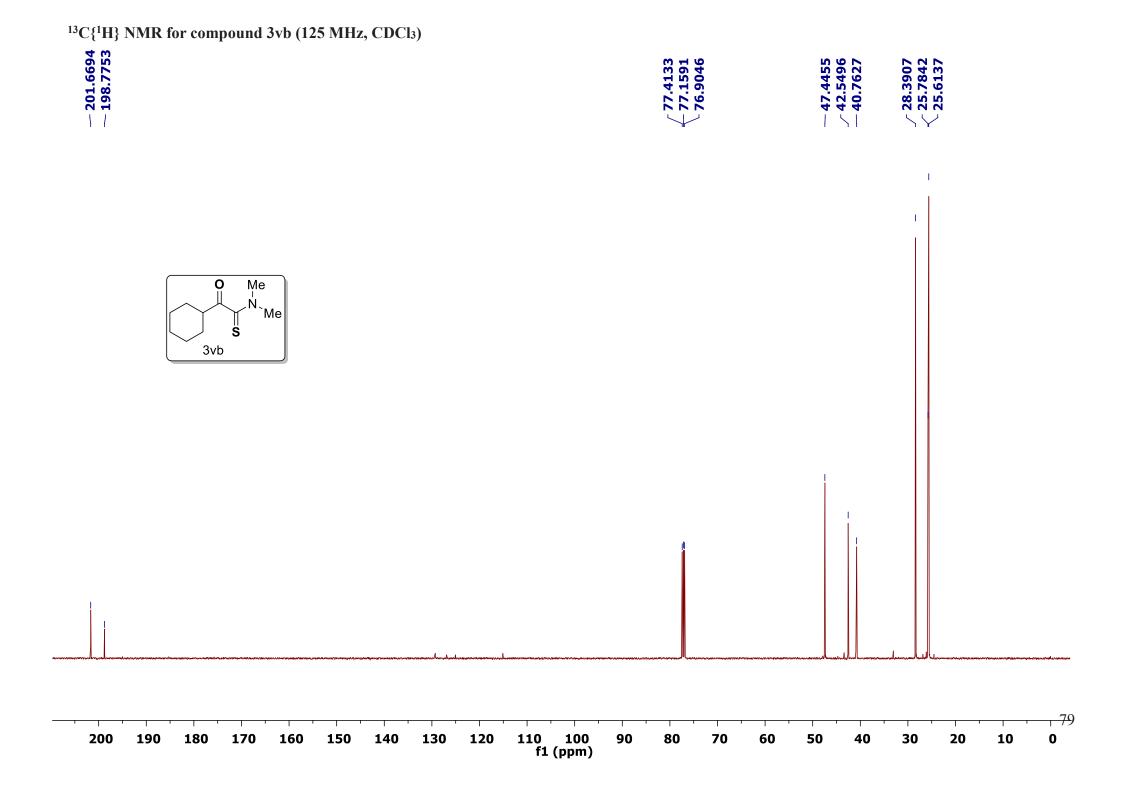




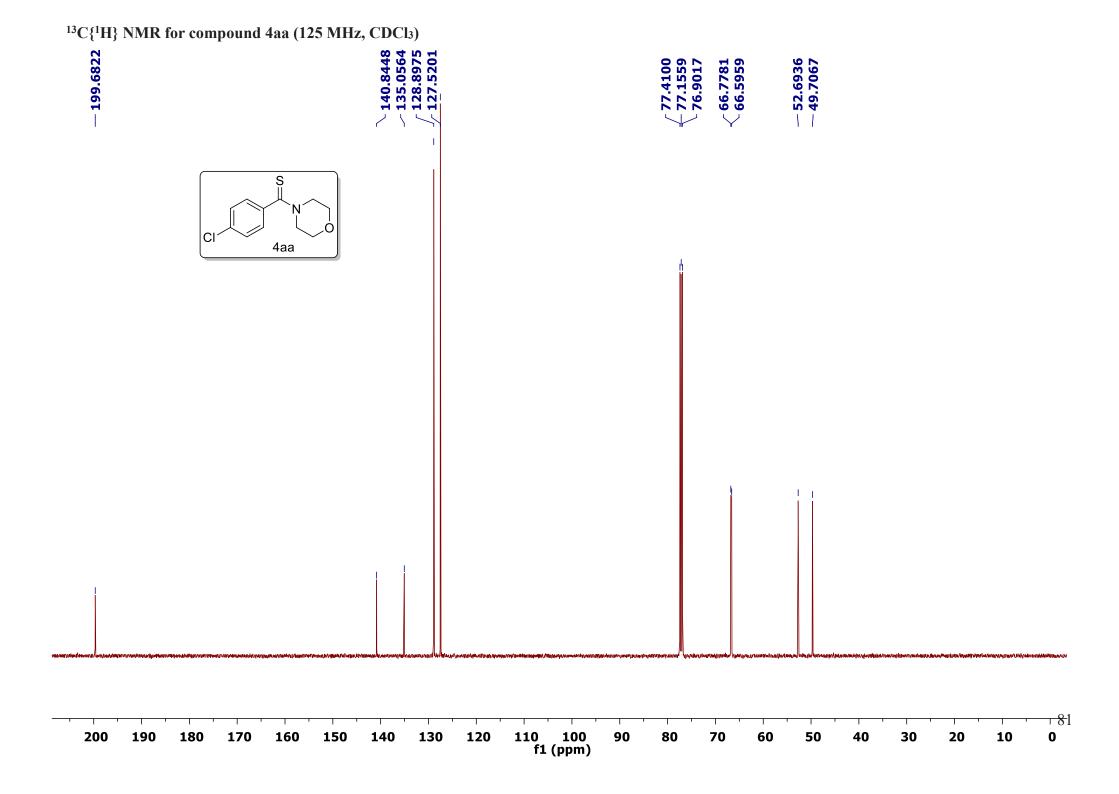


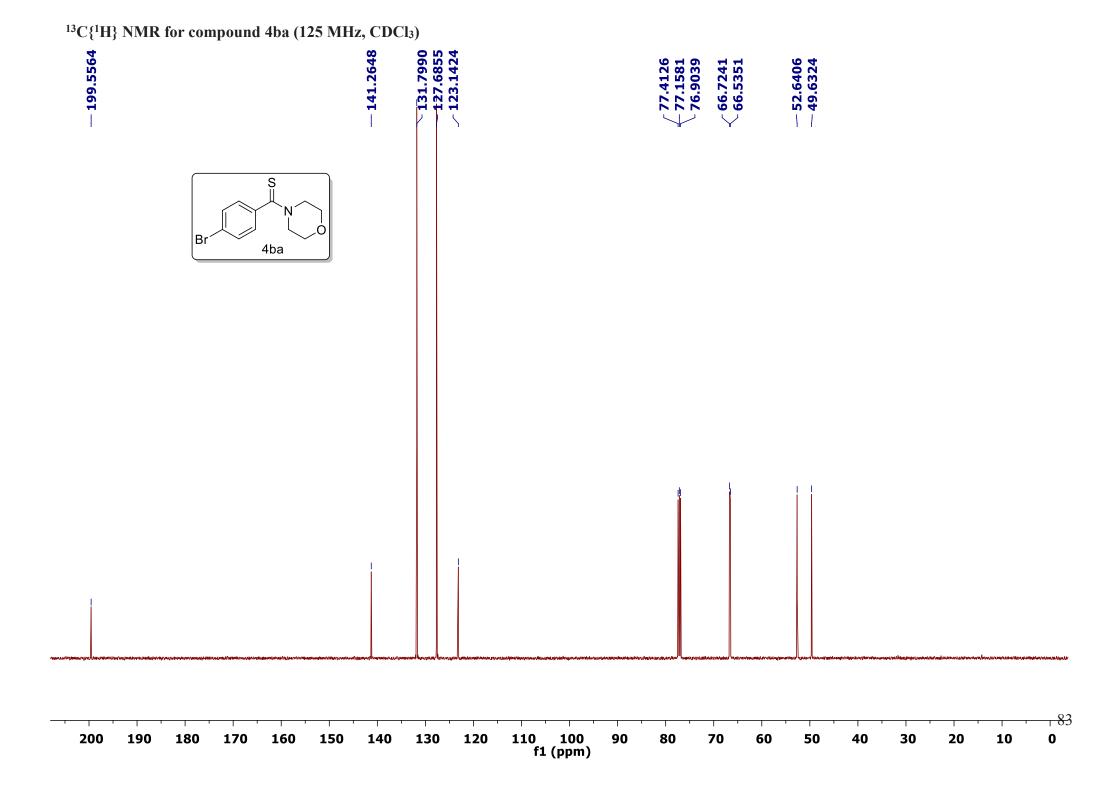




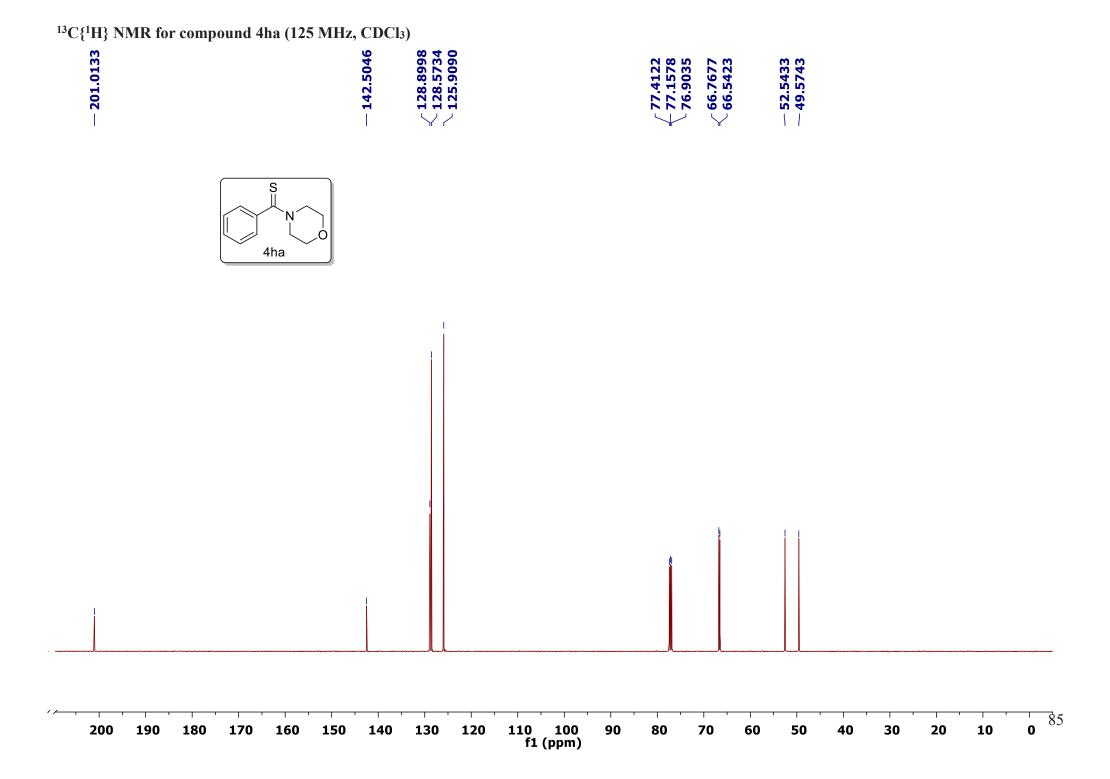


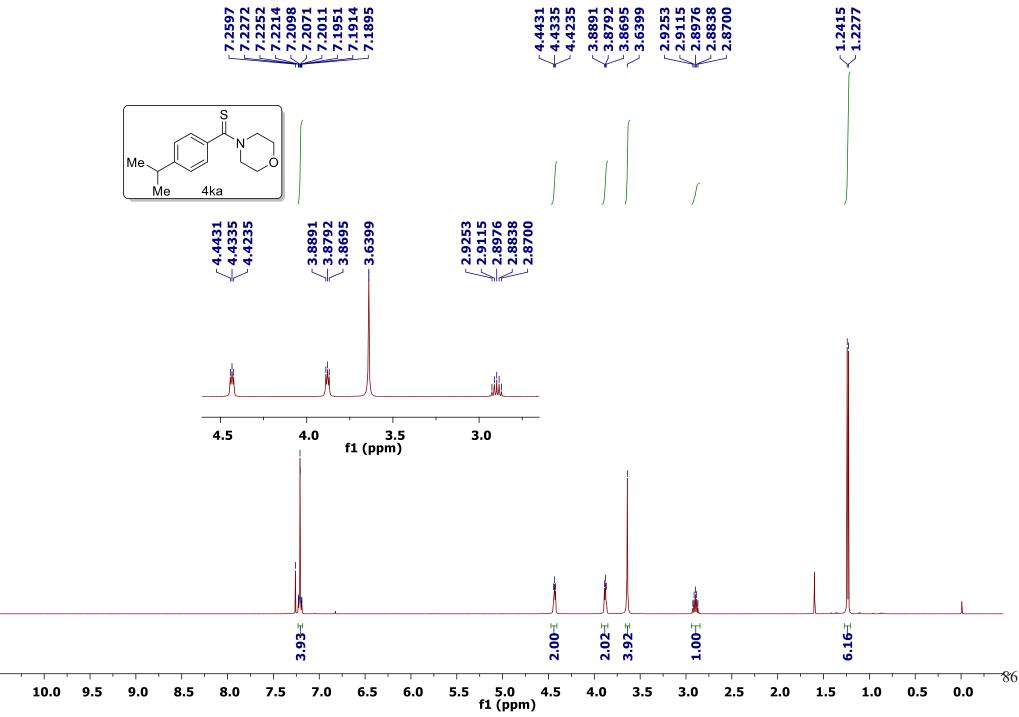
f1 (ppm)

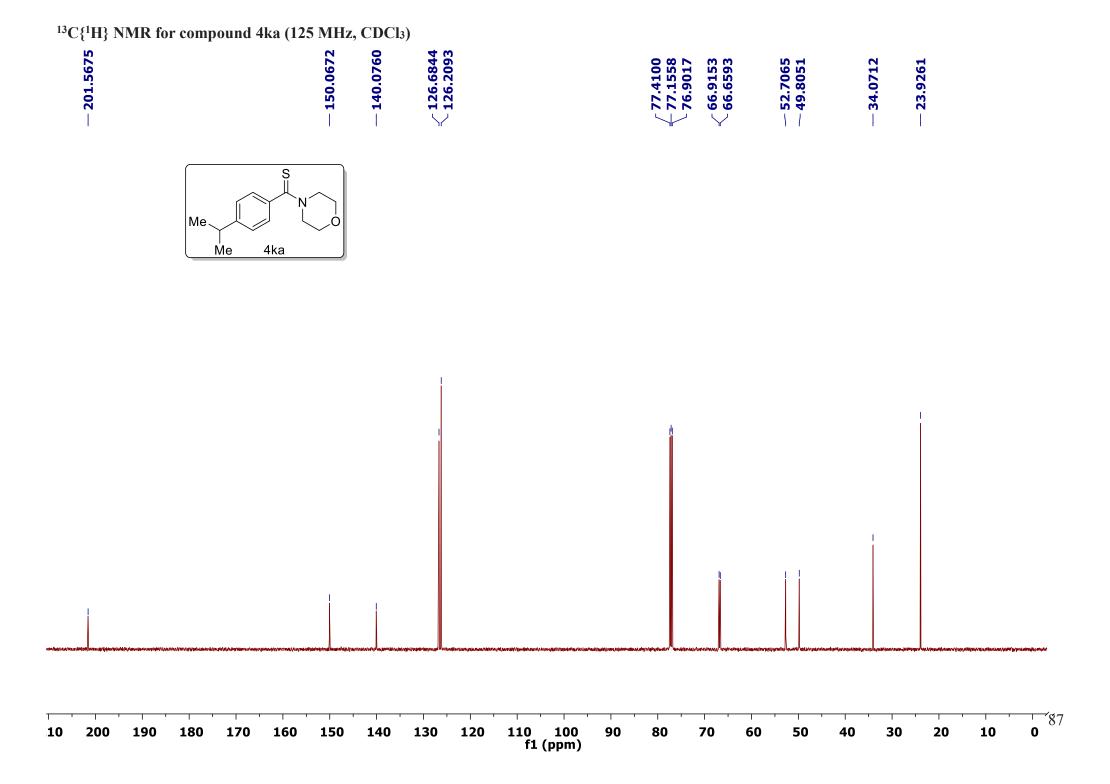


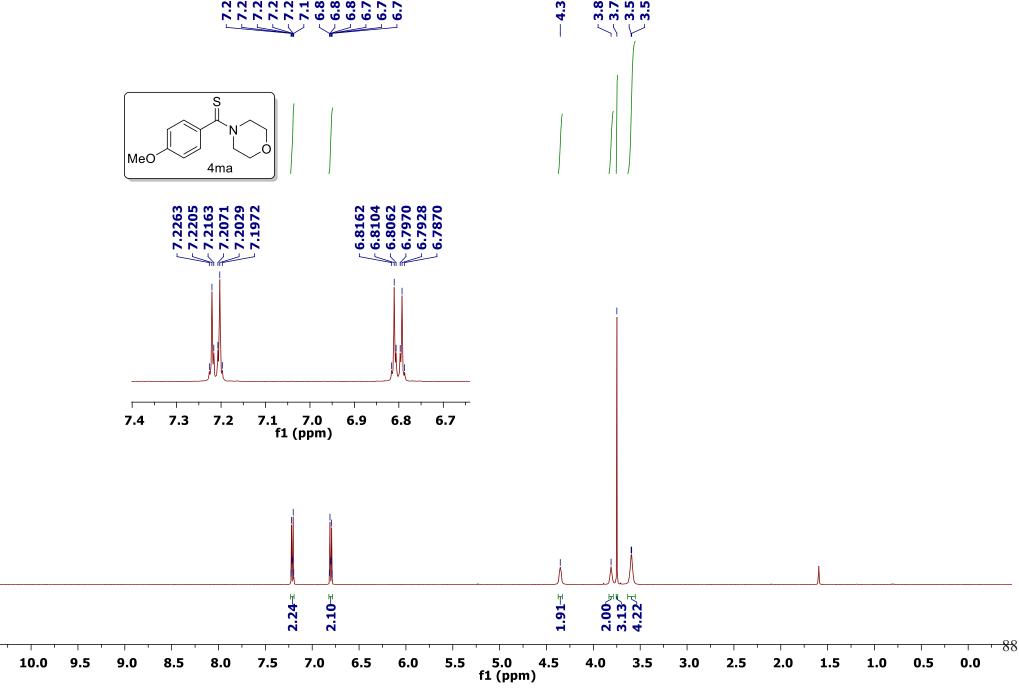


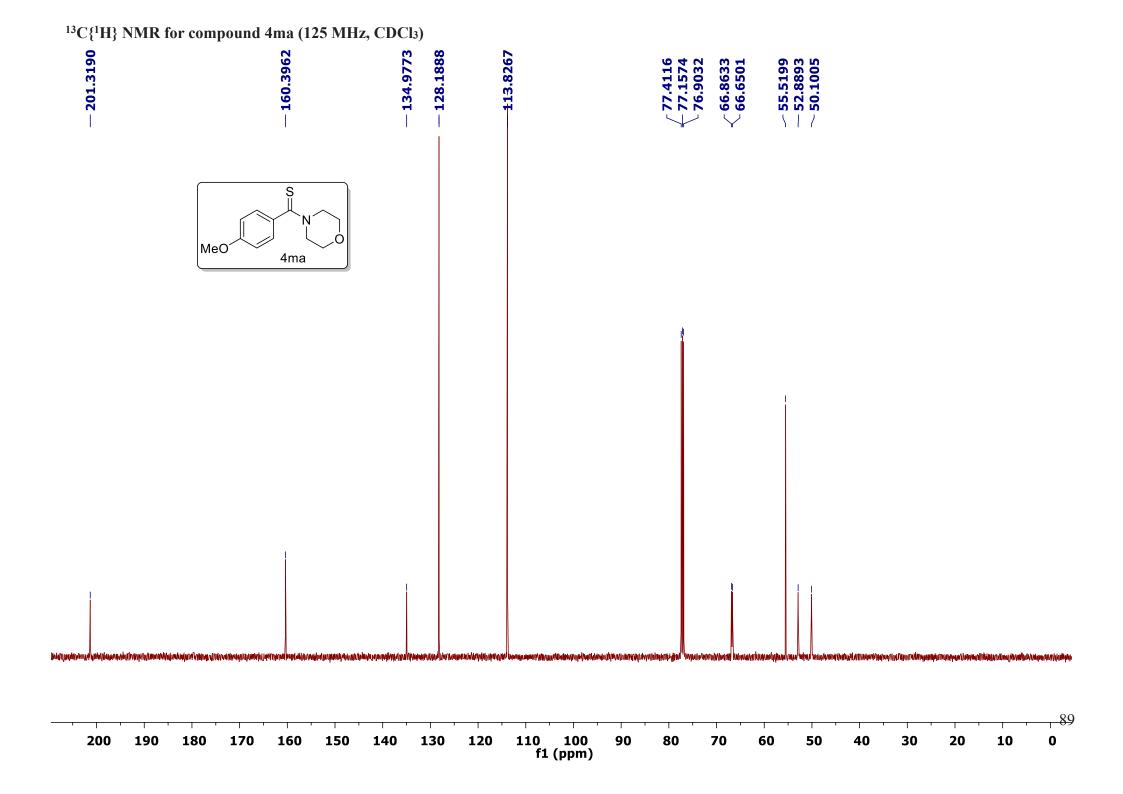
f1 (ppm)











5.5

5.0

f1 (ppm)

4.5

3.5

4.0

3.0

2.5

2.0

1.5

1.0

0.5

0.0

7.0

6.5

6.0

7.5

10.0

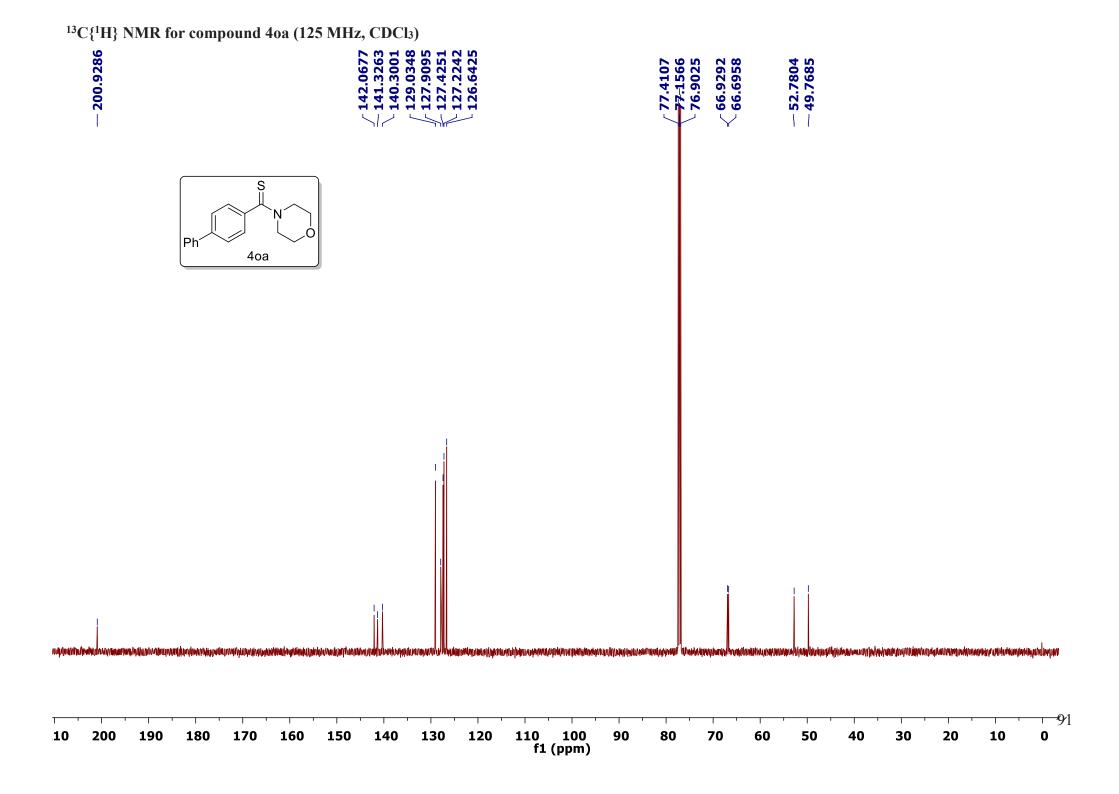
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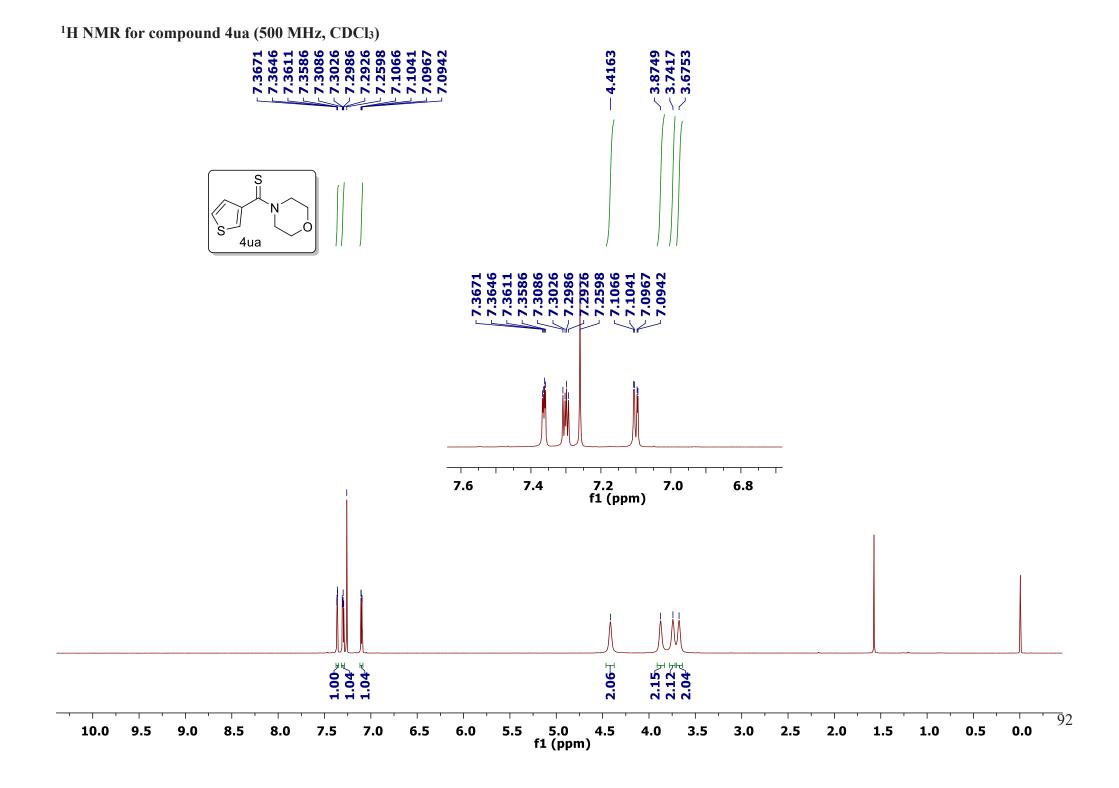
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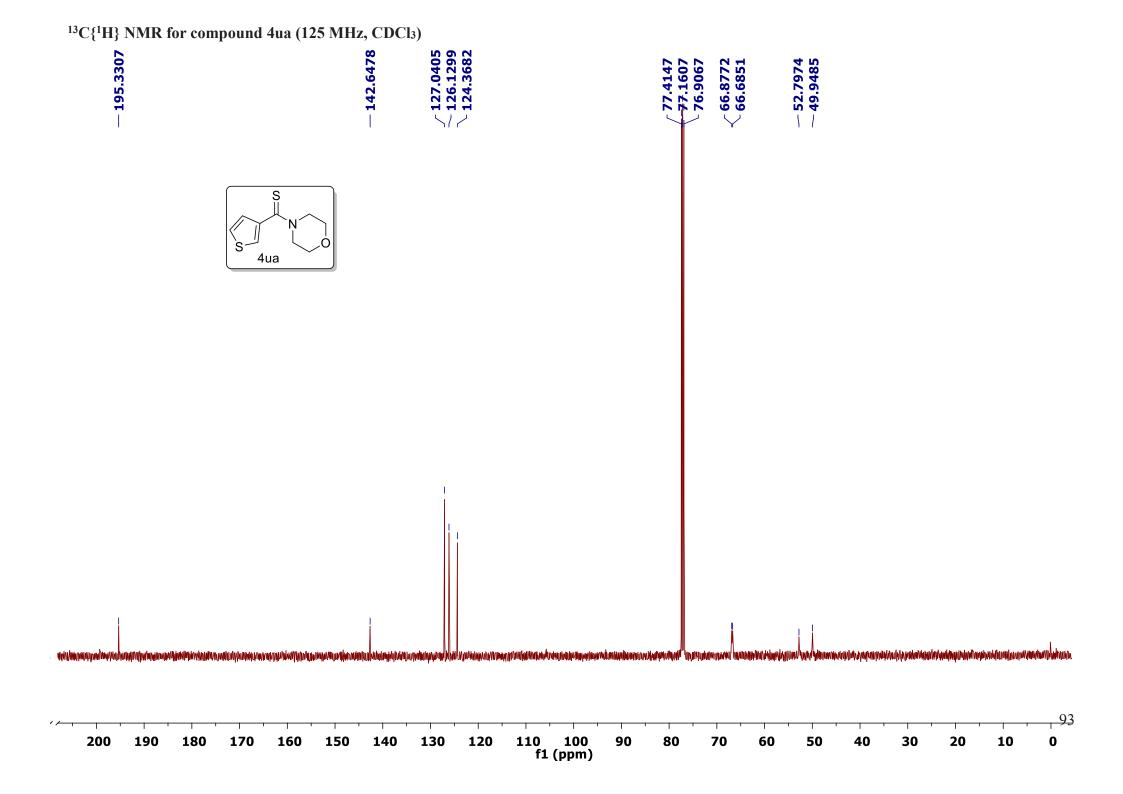
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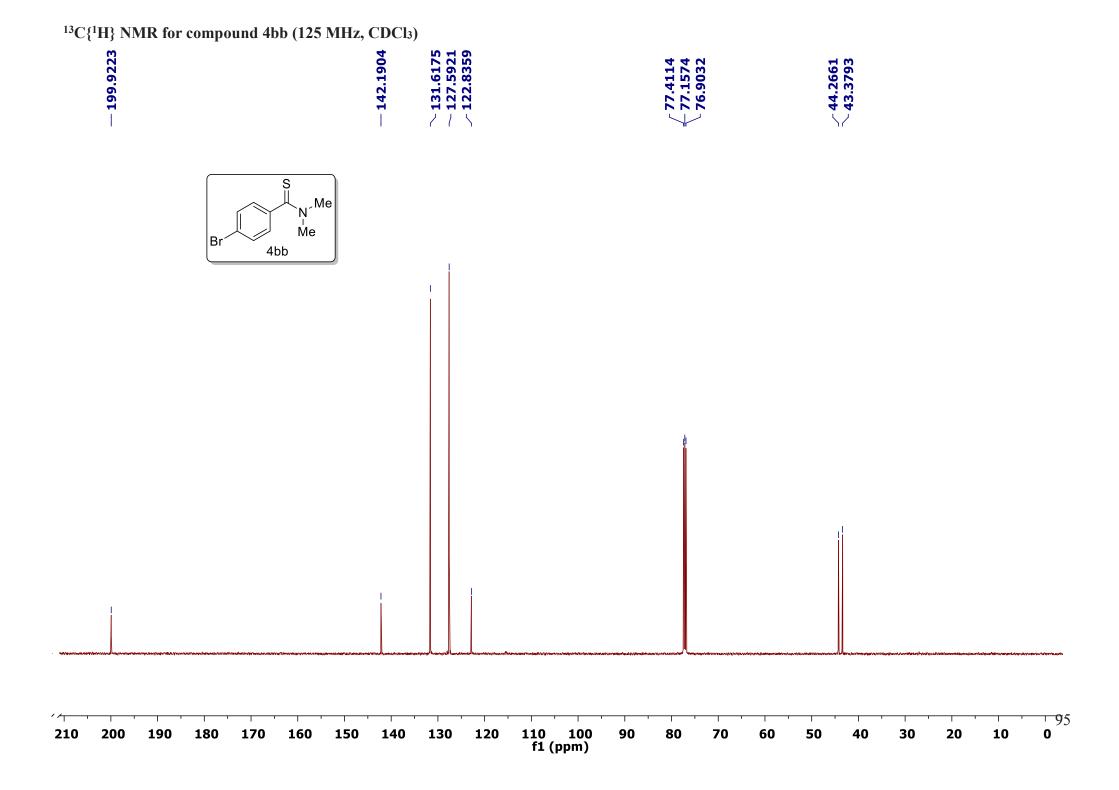
8.0

~ ~90









5.5

6.0

5.0

f1 (ppm)

4.5

3.5

4.0

3.0

2.5

2.0

1.5

1.0

0.5

0.0

10.0

9.5

8.5

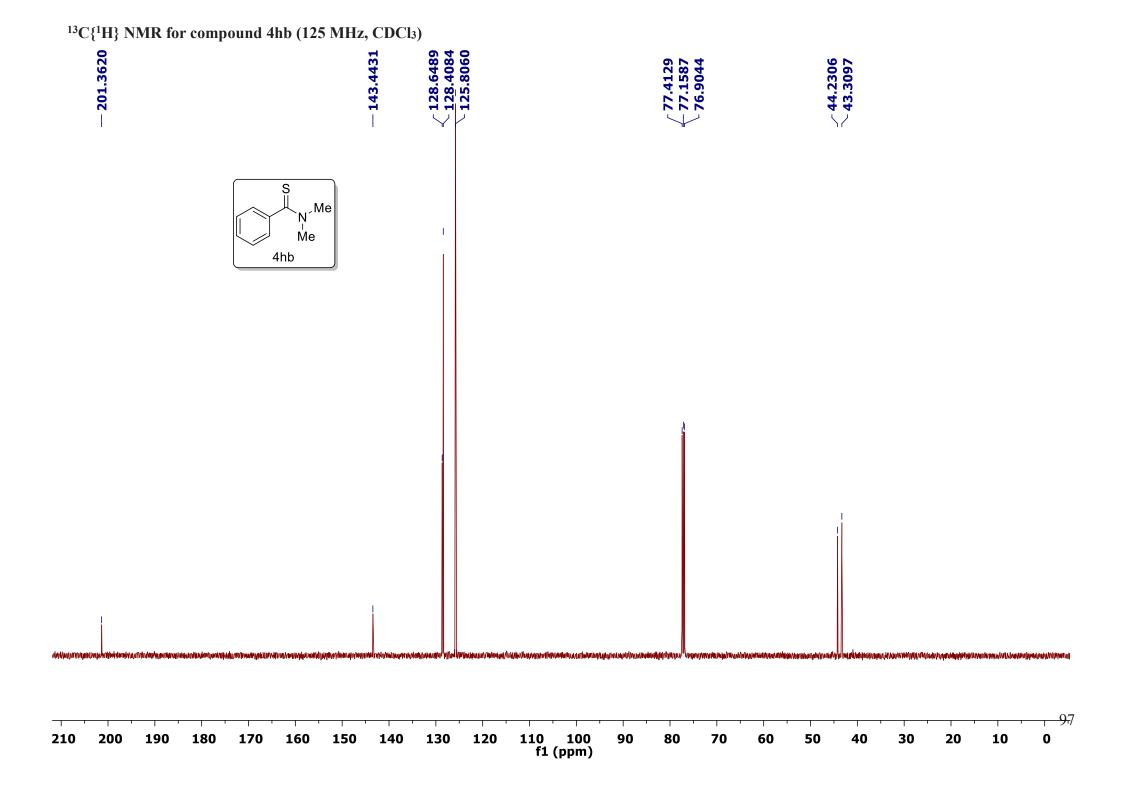
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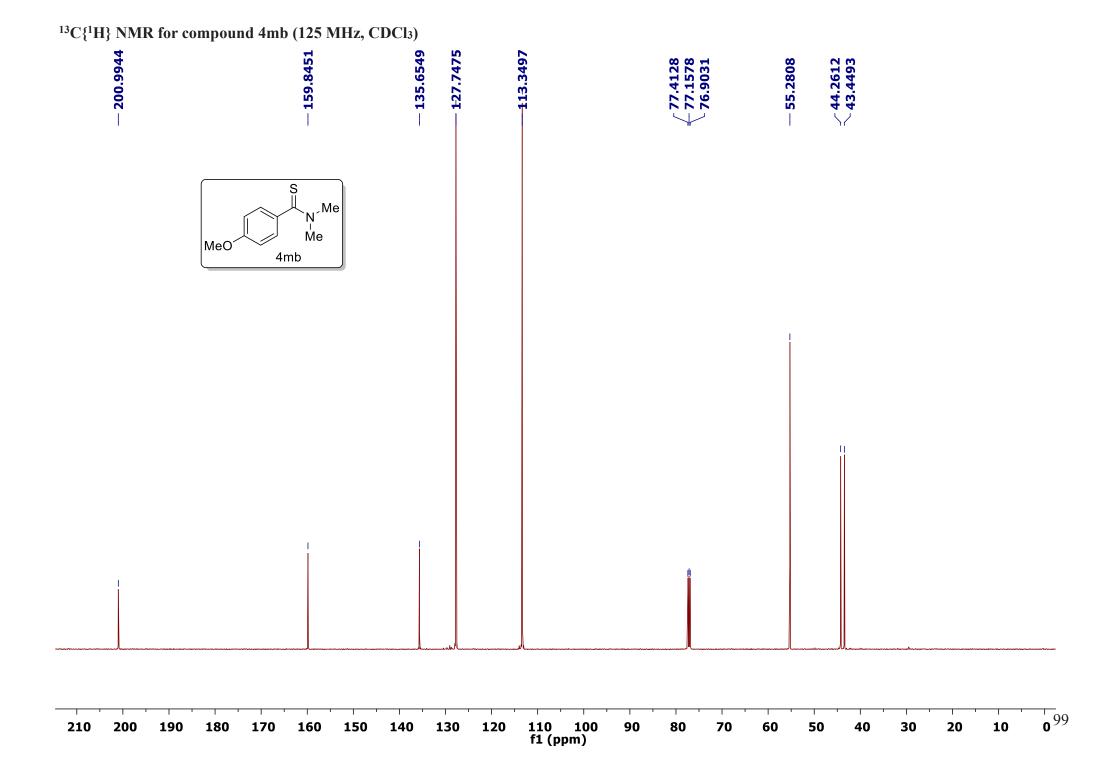
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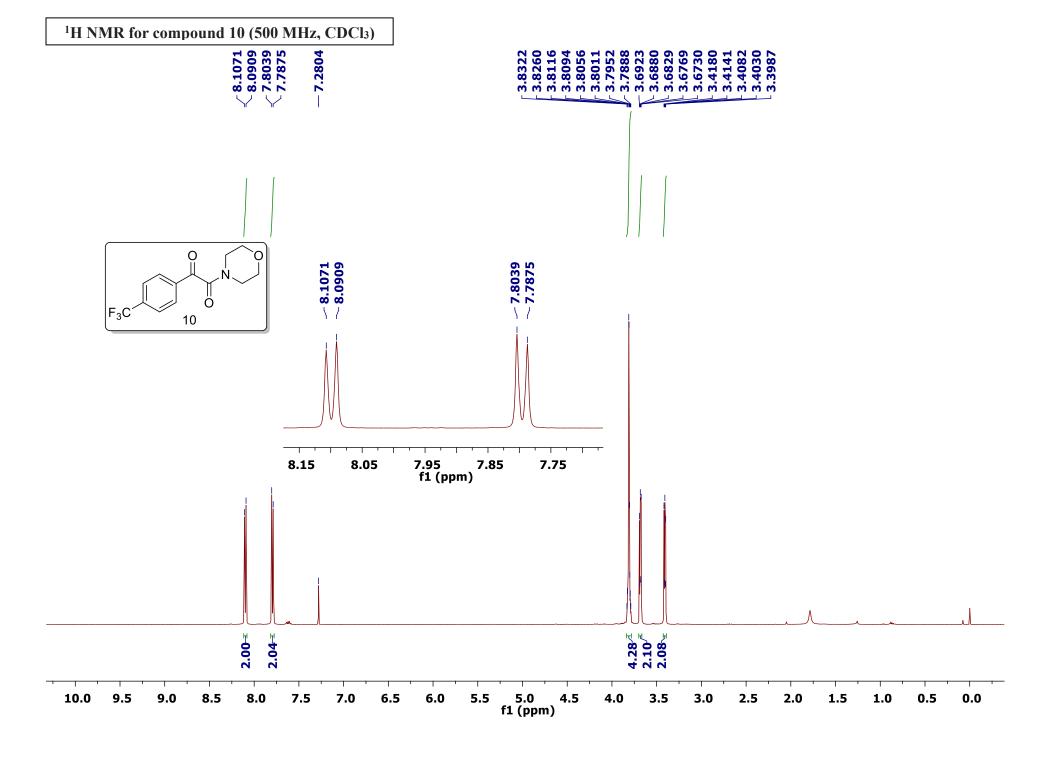
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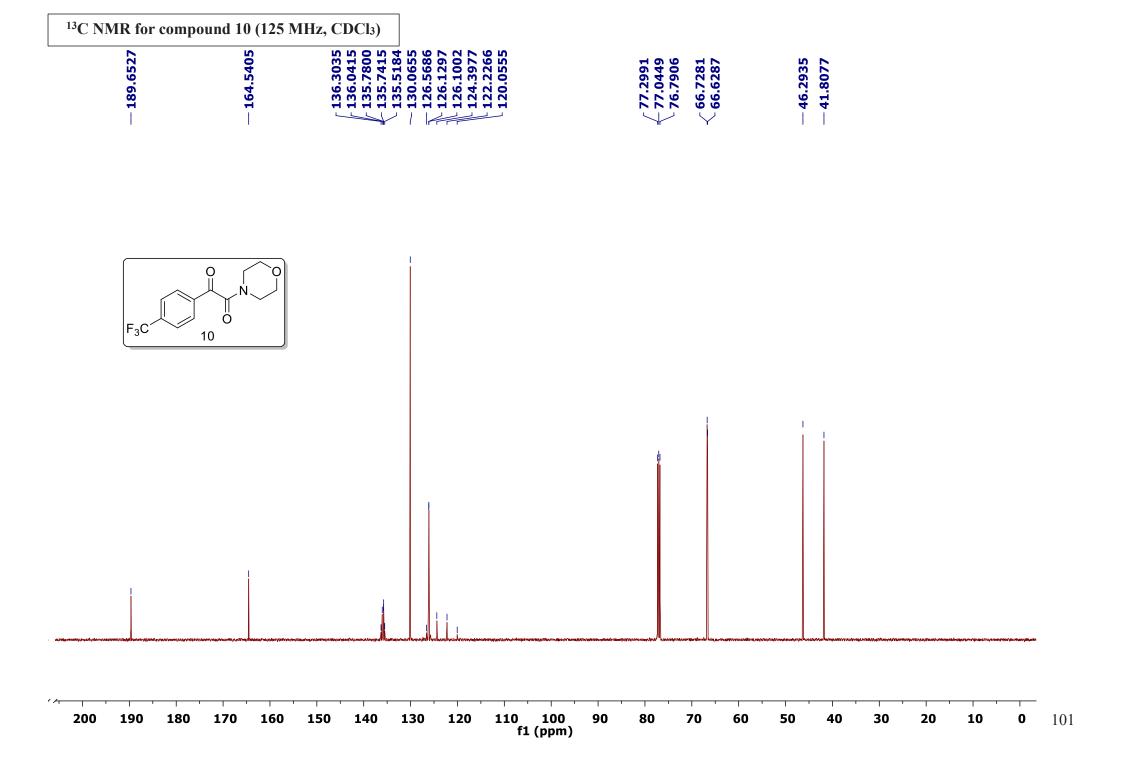
6.5

9.0

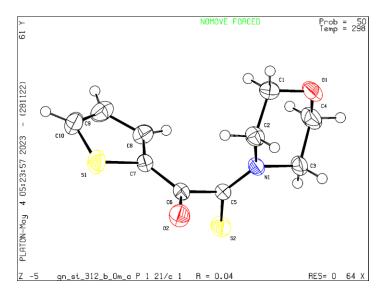








3. Crystallographic data of 3ta (CCDC No. 2259851)



Bond precision: C-C = 0.0017 A Wavelength=0.71073

Cell: a=10.8649(7) b=8.9005(5) c=11.3860(7) alpha=90 beta=91.427(4) gamma=90

Temperature: 298 K

	Calculated	Reported
	Calculated	Reported
Volume Space	1100.72(12)	1100.72(12)
groupHall	P 21/c	P 1 21/c 1
group	-P 2ybc	-P 2ybc
Moiety formula	C10 H11 N O2 S2	C10 H11 N O2 S2
Sum formula	C10 H11 N O2 S2	C10 H11 N O2 S2
Mr	241.32	241.32
Dx,g cm-3	1.456	1.456
Z	4	4
Mu (mm-1)	0.462	0.462
F000	504.0	504.0
F000'	505.18	
h,k,lmax	18,14,19	18,14,19
Nref	5383	5360
Tmin,Tmax	0.797,0.827	0.655,0.747
Tmin'	0.797	

 $Correction\ method=\#\ Reported\ T\ Limits:\ Tmin=0.655\ Tmax=0.747 AbsCorr=MULTI-SCAN$

Data completeness= 0.996 Theta(max)= 36.451

R(reflections) = 0.0419(4038) wR2(reflection) = 0.1162(5360)

S = 1.062 Npar= 136