# **Supplementary Material for:**

Oudemansin and 9-methoxystrobilurin derivatives with antimalarial activity from cultures of the basidiomycete *Favolaschia minutissima*: assignments of the absolute configurations of the isoprene-derived units

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**Fig. S1** Structures of the compounds isolated from cultures of *Favolaschia minutissima* TBRC-BCC 19434.











**Fig. S9** HRESIMS of oudemansin E (1) (positive ion mode)







Dept135



Fig. S13 COSY spectrum of oudemansin M (2) (CDCl<sub>3</sub>, 400 MHz)





















HRESIMS of oudemansin P (3) (positive ion mode)











Fig. S33 HRESIMS of oudemansin Q (4) (positive ion mode)







5.0

4.5

6.5 6.0 5.5 4.0 3.5 3.0 2.5 2.0 1.5 1.0 0.5

8.5 9.0

8.0 7.5 7.0 7.0 7.5 8.0 8.5 9.0

ppm

DEPT-135 spectrum of 9-methoxystrobilurin I (5) (CDCl<sub>3</sub>, 125 MHz)





Fig. S41 HRESIMS of 9-methoxystrobilurin I (5) (positive ion mode





















## Fig. S50 <sup>1</sup>H NMR spectrum of 12 (CDCl<sub>3</sub>, 400 MHz)



**Fig. S51** 

HRESIMS of **12** (positive ion mode)





**Fig. S52** <sup>1</sup>H NMR spectrum of 9-methoxystrobilurin P (**13**) (CDCl<sub>3</sub>, 500 MHz)



Dept135

















CD[mdeg] CD[mdeg] -20 -30 200 -20 Wavelength [nm] Wavelength [nm] Oudemansin E (1) (25  $\mu$ g/ml in MeCN) Oudemansin M (2) (50 µg/ml in MeCN) CD[mdeg] CD[mdeg] -20 -20 -40 -30 Wavelength [nm] Wavelength [nm]  $\gamma$ -Lactone 6 (10 µg/ml in MeCN) Oudemansin A (7) (30  $\mu$ g/ml in MeCN) CD[mdeg] -20 -30 Wavelength [nm]

**Fig. S62** Experimental ECD spectra of oudemansin E (1), oudemansin M (2),  $\gamma$ -lactone 6, oudemansin A (7), and 9-methoxystrobilurin E (9)

9-Methoxystrobilurin E (9) (50 µg/ml in MeCN)

#### **Computational Details for ECD Calculation of 9-Methoxystrobilurin E (9)**

The conformational analysis of each enantiomer for compound 9 was perform on Spartan' 18 using MMFF<sup>1</sup> as the molecular mechanics force field. Thirty-seven lowest energy conformers were found contributed to about 90% of the MMFF population. These conformers of 9 were further calculated for geometry optimization in gas phase using B3LYP <sup>2-4</sup> functional and 6-31G(d) <sup>5-7</sup> basis set in Gaussian 16<sup>8</sup>. All calculations of time-dependent density-functional theory (TDDFT) were performed on the gas-phase optimized structures to obtain ECD using CAM-B3LYP<sup>9</sup> functional and 6-311++G(2d,2p) <sup>5-7</sup> basis set with conductor-like polarizable continuum model (CPCM)<sup>10-11</sup> in acetonitrile. Twelve conformers of (2'S, 6'R)-9 and eleven conformers of (2'R, 6'S)-9 were found with the relative free energies within ~2 kcal/mol from their most stable conformers (Fig. S63 and Fig. S65). The ECD curve of each enantiomer for 9 was demonstrated by summing of ECD spectra of these conformers based on Boltzmann weighting factors at 298.15 K with a half bandwidth of 0.25 eV using SpecDis v.1.71<sup>12</sup>. Data visualization of the calculated ECD spectra of each enantiomer and experimental CD spectra for 9-methoxystrobilurin E (9) were performed using python (version 3.8.13) package<sup>13</sup>. Raw data was imported with pandas (version 1.4.3) library and plotted with matplotlib (version 3.5.1) library.

#### (1) Computational details for (2'S,6'R)-9

**Table S1.** Estimated thermodynamic parameters at 298.15 K and conformational analysis of (2'S,6'R)-9 at the CAM-B3LYP/6-311++G(2d,2p) level in acetonitrile solvent, assuming Boltzmann statistics at T = 298.15 K.

Conformers	B3LYP/6- 31G(d)	CAM-B3LYP/6-311++G(2d, 2p)							
	E (a.u.)	E* (a.u.)	E <sub>ZPE</sub> *(a.u.)	H <sub>298K</sub> *(a.u.)	G <sub>298K</sub> *(a.u.)	ΔG <sub>298K</sub> * (kcal/mol)	$\mathbf{P_G}^*$		
(2'S,6'R)-9_conf 1	-1651.053460	-1650.798558	-1650.226372	-1649.933117	-1650.301598	0.00	17.58		
(2'S,6'R)-9_conf 2	-1651.053402	-1650.798531	-1650.226337	-1649.933121	-1650.301375	0.14	13.87		
(2'S,6'R)-9_conf 3	-1651.053309	-1650.798432	-1650.226248	-1649.933010	-1650.301350	0.16	13.51		
(2'S,6'R)-9_conf 4	-1651.053296	-1650.798436	-1650.226246	-1649.933025	-1650.301292	0.19	12.71		
(2'S,6'R)-9_conf 5	-1651.053797	-1650.798497	-1650.226264	-1649.932634	-1650.301231	0.23	11.92		
(2'S,6'R)-9_conf 6	-1651.053986	-1650.798501	-1650.226267	-1649.932450	-1650.301161	0.27	11.06		
(2'S,6'R)-9_conf 7	-1651.053871	-1650.798391	-1650.226147	-1649.932330	-1650.300942	0.41	8.77		
(2'S,6'R)-9_conf 8	-1651.053812	-1650.798395	-1650.226153	-1649.932400	-1650.300870	0.46	8.13		
(2'S,6'R)-9_conf 9	-1651.052668	-1650.795962	-1650.223574	-1649.928596	-1650.298511	1.94	0.67		
(2'S,6'R)-9_conf 10	-1651.052677	-1650.795935	-1650.223555	-1649.928545	-1650.298454	1.97	0.63		
(2'S,6'R)-9_conf 11	-1651.052932	-1650.796067	-1650.223639	-1649.928541	-1650.298375	2.02	0.58		
(2'S,6'R)-9 conf 12	-1651.052850	-1650.796075	-1650.223656	-1649.928639	-1650.298350	2.04	0.56		

 $E = gas phase electronic energy; ZPE = gas phase zero-point energy; E_{ZPE} = E+ZPE; H_{298K} = enthalpy; G_{298K} = gas phase Gibbs free energy at the B3LYP/6-31G(d) level in gas phase$ 

 $E^*$  = electronic energy in acetonitrile;  $E_{ZPE}^* = E^* + ZPE$ ;  $H_{298K}^* =$  estimated enthalpy in acetonitrile,  $H_{298K}^* = E^* + (H_{298K}-E)$ ;  $G_{298K}^* =$  estimated Gibbs free energy in acetonitrile,  $G_{298K}^* = E^* + (G_{298K}-E)$ ;  $\Delta G_{298K}^* =$  estimated relative Gibbs free energy at the CAM-B3LYP/6-311++G(2d,2p) level in acetonitrile solvent;  $P_G^* =$  conformational distribution calculated from relative Gibbs free energy.

**Fig. S63.** Optimized conformers of (2'S,6'R)-9 at the B3LYP/6-31G(d) level in gas phase with the populations calculated from their estimated Gibbs free energies in acetonitrile solvent at the CAM-B3LYP/6-311++G(2d,2p) level





(2′*S*,6′*R*)**-9**\_Conf 2 (13.87%)



(2′*S*,6′*R*)**-9**\_Conf 4 (12.71%)



(2'S,6'R)-9\_Conf 6 (11.06%)



(2'*S*,6'*R*)**-9**\_Conf 8 (8.13%)



(2'S,6'R)-9\_Conf 10 (0.63%)



(2'S,6'R)-9\_Conf 12 (0.56%)

**Fig. S64.** Comparison of experimental CD spectrum of 9-methoxystrobilurin E (black) and calculated ECD spectrum of (2'S, 6'R)-9 (blue)



#### (2) Computational details for (2'R,6'S)-9

**Table S2.** Estimated thermodynamic parameters at 298.15 K and conformational analysis of (2'R,6'S)-9 at the CAM-B3LYP/6-311++G(2d,2p) level in acetonitrile solvent, assuming Boltzmann statistics at T = 298.15 K

Conformers	B3LYP/6-31G(d)	CAM-B3LYP/6-311++G(2d, 2p)							
	E (a.u.)	E* (a.u.)	E <sub>ZPE</sub> *(a.u.)	H <sub>298K</sub> *(a.u.)	G <sub>298K</sub> *(a.u.)	∆G <sub>298K</sub> * (kcal/mol)	$\mathbf{P_{G}}^{*}$		
(2'R,6'S)-9_conf 1	-1651.053460	-1650.798558	-1650.226371	-1649.933115	-1650.301599	0.00	19.28		
(2' <i>R</i> ,6' <i>S</i> )-9_conf 2	-1651.053402	-1650.798531	-1650.226337	-1649.933121	-1650.301375	0.14	15.21		
(2' <i>R</i> ,6' <i>S</i> )-9_conf 3	-1651.053309	-1650.798432	-1650.226248	-1649.933010	-1650.301350	0.16	14.81		
(2'R,6'S)-9_conf 4	-1651.053296	-1650.798436	-1650.226246	-1649.933025	-1650.301292	0.19	13.92		
(2'R,6'S)-9_conf 5	-1651.053797	-1650.798497	-1650.226264	-1649.932634	-1650.301231	0.23	13.07		
(2'R,6'S)-9_conf 6	-1651.053986	-1650.798501	-1650.226267	-1649.932450	-1650.301161	0.27	12.13		
(2'R,6'S)-9_conf 7	-1651.053812	-1650.798395	-1650.226153	-1649.932400	-1650.300869	0.46	8.91		
(2' <i>R</i> ,6' <i>S</i> )- <b>9</b> _conf 8	-1651.052668	-1650.795960	-1650.223571	-1649.928593	-1650.298511	1.94	0.73		
(2' <i>R</i> ,6' <i>S</i> )- <b>9</b> _conf 9	-1651.052677	-1650.795937	-1650.223557	-1649.928549	-1650.298456	1.97	0.69		
(2'R,6'S)-9_conf 10	-1651.052932	-1650.796070	-1650.223642	-1649.928546	-1650.298375	2.02	0.63		
(2' <i>R</i> ,6' <i>S</i> )- <b>9</b> _conf 11	-1651.052850	-1650.796074	-1650.223655	-1649.928637	-1650.298354	2.04	0.62		

 $E = gas phase electronic energy; ZPE = gas phase zero-point energy; E_{ZPE} = E+ZPE; H_{298K} = enthalpy; G_{298K} = gas phase Gibbs free energy at the B3LYP/6-31G(d) level in gas phase$ 

 $E^*$  = electronic energy in acetonitrile;  $E_{ZPE}^* = E^* + ZPE$ ;  $H_{298K}^* =$  estimated enthalpy in acetonitrile,  $H_{298K}^* = E^* + (H_{298K-E})$ ;  $G_{298K}^* =$  estimated Gibbs free energy in acetonitrile,  $G_{298K}^* = E^* + (G_{298K-E})$ ;  $\Delta G_{298K}^* =$  estimated relative Gibbs free energy at the CAM-B3LYP/6-311++G(2d,2p) level in acetonitrile solvent;  $P_G^* =$  conformational distribution calculated from relative Gibbs free energy.

**Fig. S65.** Optimized conformers of (2'R,6'S)-9 at the B3LYP/6-31G(d) level in gas phase with the populations calculated from their estimated Gibbs free energies in acetonitrile solvent at the CAM-B3LYP/6-311++G(2d,2p)





(2'*R*,6'*S*)**-9**\_Conf 2 (15.21%)



(2'*R*,6'*S*)**-9**\_Conf 4 (13.92%)



(2'*R*,6'*S*)**-9**\_Conf 6 (12.13%)



(2'*R*,6'*S*)**-9**\_Conf 8 (0.73%)



(2'R, 6'S)-9\_Conf 10 (0.63%)

(2'*R*,6'*S*)**-9**\_Conf 11 (0.62%)

**Fig. S66.** Comparison of experimental CD spectrum of 9-methoxystrobilurin E (black) and calculated ECD spectra for (2'R, 6'S)-9 (red)



#### (3) Conclusion

9-Metoxystrobilurin E (9) was identified to be the (2'R, 6'S)-isomer.

**Fig. 4.** (main text). Calculated ECD spectra of (2'S,6'R)-9 (blue) and (2'R,6'S)-9 (red) and experimental CD spectrum of 9-methoxystrobilurin E (black)



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