# **Electronic Supplementary Information**

# Alcohol synthesis based on the S<sub>N</sub>2 reactions of alkyl halides with squarate dianion

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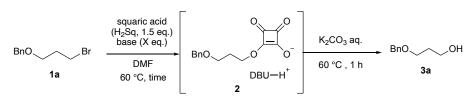
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# ESI-1. Optimization of base and solvent for the transformation of 1a into 3a



entry	base (X eq.)	time (h)	yield of alcohol <b>3a</b> (%)	recovery of <b>1a</b> (%)
1	Cs <sub>2</sub> CO <sub>3</sub> (1.5)	5	26	73
2	pyridine (3.0)	5	20	60
3	Et₃N (3.0)	1	14 <sup>a</sup>	20 <sup>a</sup>
4	K <sub>2</sub> CO <sub>3</sub> (1.5)	0.5	6	86
5	NaHCO3 (3.0)	0.5	4	87
6	<i>i</i> -Pr₂NEt (3.0)	0.5	27	20
7	TMEDA (1.5)	0.5	7	43
8	KOH (3.0)	0.5	6	86
9	DBU (3.0)	0.5	89	_
10	DBU (3.6)	0.5	96	-

<sup>a</sup> NMR yield

# 2. Screening of solvent

BnO、Br	H <sub>2</sub> Sq (1.5 eq.) DBU (3.6 eq.)	K <sub>2</sub> CO <sub>3</sub> aq.	BnO
1a	solvent 60 °C, 0.5 h	60 °C, 1 h	3a

entry	solvent	yield of alcohol <b>3a</b> (%)	recovery of 1a (%)
1	THF	43	48
2	EtOH	54	38
3	acetone	20	_
4	DMSO	80	_
4	MeCN	82	_
5	DMF	96	_

# 1. Screening of base

#### **ESI-2.** General information

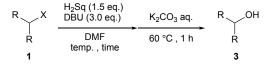
All reactions were carried out in flame- or oven-dried glassware under an argon atmosphere with dry solvents unless otherwise stated. The reactions heating to high temperatures were carried out using an oil bath unless otherwise stated. THF was distilled from sodium and benzophenone. Bis-tetra-*n*-propylammonium salt of squaric acid,<sup>1</sup> di-sodium salt of squaric acid,<sup>2</sup> compounds 1e,<sup>3</sup> 1m,<sup>4</sup> 1n,<sup>5</sup> 1r,<sup>6</sup> 1s<sup>7</sup>, 5,<sup>8</sup> and 8<sup>9</sup> were prepared by the known procedures. A salt, as shown entry 4 in Table 1, was prepared by squaric acid and diazabicyclo[5.4.0]undecene (DBU) as a 1:2 mol ratio. All other reagents and solvents were used as received from commercial sources without further purification.

All reaction were monitored by thin-layer chromatography which was performed by using Merck silica gel 60 F254 pre-coated plates (0.25 mm) and visualized with UV (254 nm) and stained with a solution of 2% anisaldehyde, 5% H<sub>2</sub>SO<sub>4</sub> in ethanol or a solution of 10% phosphomolybdic acid in ethanol under heating at ca. 200 °C. Flash column chromatography was performed by using Kanto Silica Gel 60 N. IR spectra were recorded on JASCO FT/IR-4100 and the major absorbance bands are all reported in wavenumbers (cm<sup>-1</sup>). Preparative TLC (PTLC) was performed by using Merck silica gel 60 F254 pre-coated plates (0.5 mm). HRMS were recorded on a JEOL JMS-T100GCV at Research Faculty of Agriculture, Hokkaido University for FD/FI method. NMR spectra were recorded on JEOL JNM-ECX-500 (500 MHz for <sup>1</sup>H and 126 MHz for <sup>13</sup>C) with residual CHCl<sub>3</sub> of chloroform-*d* or benzene of benzene-*d*<sub>6</sub> as internal reference (<sup>1</sup>H NMR: 7.26 ppm for chloroform-*d*, 7.15 ppm for benzene-*d*<sub>6</sub>, <sup>13</sup>C NMR: 77.16 ppm for chloroform-*d*, 128.06 ppm for benzene-*d*<sub>6</sub>). Chemical shift was reported in part per million (ppm) and signals of <sup>1</sup>H NMR spectra were expressed as follows: singlet (s), doublet (d), triplet (t), quartet (q), quintet (quint) and multiplet (m).

- 1. M. G. Topuzova, S. V. Kotov and T. M. Kolev, Appl. Catal., A, 2005, 281, 157.
- 2. Q. Zhao, J. Wang, Y. Lu, Y. Li, G. Liang and J. Chen, Angew. Chem. Int. Ed., 2016, 55, 12528.
- 3. A. Itoh, T. Saito, K. Oshima and H. Nozaki, Bull. Chem. Soc. Jpn., 1981, 54, 1456.
- 4. S. G. Hegde, D. Beckwith, R. Doti and J. Wolinsky, J. Org. Chem., 1985, 50, 894.
- 5. M. Xuan, I. Paterson and S. M. Dalby, Org. Lett., 2012, 14, 5492.
- 6. H. Zhao, A. J. McMillan, T. Constantin, R. C. Mykura, F. Juliá, and D. Leonori, J. Am. Chem. Soc., 2021, 143, 14806.
- 7. I. Ryu, H. Matsubara, S. Yasuda, H. Nakamura, and D. P. Curran, J. Am. Chem. Soc., 2002, 124, 12946.
- 8. K. Zhao and R. R. Knowles, J. Am. Chem. Soc., 2022, 144, 137.
- 9. N. Saygili, R. J. Brown, P. Day, R. Hoelzl, P. Kathirgamanathan, E. R. Mageean, T. Ozturk, M. Pilkington, M. M. B. Qayyum, S. S. Turner, L. Vorwerg and J. D. Wallis, *Tetrahedron*, 2001, **57**, 5015.

#### **ESI-3.** Experimental section

# General synthetic procedure



To a solution of 1 (1.0 eq.) and squaric acid (1.5 eq.) in DMF (ca. 0.5 M) was added DBU (3.0 eq.) at rt, and the reaction mixture was stirred at the optimized reaction temperature among 60–120 °C. After stirring for the reaction time corresponding to the substrate, 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) was added. After stirring for 1 h at 60 °C, brine was added to the reaction mixture, and the aqueous mixture was extracted with EtOAc for three times. The combined organic layers were dried over MgSO4 and filtered. The filtrate was concentrated under reduced pressure to give a crude product. Purification of the crude product by silica gel column chromatography (hexane/EtOAc system) gave **3**. When the reaction protocol or the reaction conditions differed from the general procedure, the detailed information was mentioned in the procedures of each of synthesized compounds.

Synthesis of 3a from bromide 1a

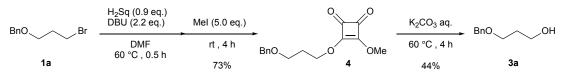
According to the general procedure, the reaction of **1a** (45.9 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 300  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3a**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3a** (29.6 mg, 178  $\mu$ mol, 89%) as a colorless oil. The spectral data of **3a** was in good agreement with the literature data.<sup>10</sup>

#### Synthesis of 3a from chloride 1b

According to the general procedure, the reaction of **1b** (37.6 mg, 204 µmol) and H<sub>2</sub>Sq (34.7 mg, 304 µmol) in DMF (410 µL) with DBU (91.4 µL, 611 µmol) at 80 °C for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3a**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3a** (30.6 mg, 184 µmol, 91%) as a colorless oil. The spectral data of **3a** was in good agreement with the literature data.<sup>10</sup>

<sup>10.</sup> L. V. Heumann and G. E. Keck, Org. Lett., 2007, 9, 1951.

#### Synthesis of 3a from dialkyl squarate 4

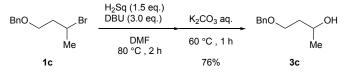


To a solution of **1a** (100 mg, 436  $\mu$ mol) and H<sub>2</sub>Sq (44.1 mg, 387  $\mu$ mol) in DMF (870  $\mu$ L) was added DBU (141  $\mu$ L, 944  $\mu$ mol). After stirring for 0.5 h at 60 °C, MeI (136  $\mu$ L, 2.18 mmol) was added at 0 °C. After stirring for 4 h at rt, H<sub>2</sub>O (1 mL) was added to the reaction mixture, and the aqueous mixture was extracted with EtOAc (5 mL × 3). The organic layer was dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc = 4/1) to give **4** (78.7 mg, 285  $\mu$ mol, 73%) as a yellow oil.

A mixture of 4 (28.1 mg, 123  $\mu$ mol) in 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) was stirred for 4 h at 60 °C. Then, the reaction mixture was extracted with EtOAc (1 mL × 3). The organic layer was dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc = 4/1) to give **3a** (8.9 mg, 53.5  $\mu$ mol, 44%) as a colorless oil. The spectral data of **3a** was in good agreement with the literature data.<sup>10</sup>

Data for 4: IR (ATR): 2961, 2866, 1812, 1732, 1601, 1476, 1454, 1395, 1361, 1340, 1100, 1026, 917, 823, 741, 700 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.37-7.23$  (m, 5H), 4.79 (t, J = 6.0 Hz, 2H), 4.49 (s, 2H), 4.30 (s, 3H), 3.60 (t, J = 6.0 Hz, 2H), 2.09 (quint, J = 6.0 Hz, 2H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 189.30, 189.25, 184.5, 184.3, 138.1, 128.5 (2C), 127.9 (2C), 127.8, 73.2, 71.9, 65.7, 61.0, 30.2. HRMS (FD): m/z [M]<sup>+</sup> calcd for C<sub>15</sub>H<sub>16</sub>O<sub>5</sub>: 276.0998; found: 276.0993.

Synthesis of 3c from bromide 1c



According to the general procedure, the reaction of **1c** (49.1 mg, 202 µmol) and H<sub>2</sub>Sq (34.5 mg, 302 µmol) in DMF (400 µL) with DBU (90.6 µL, 606 µmol) at 80 °C for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3c**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3c** (27.6 mg, 153 µmol, 76%) as a colorless oil. The spectral data of **3c** was in good agreement with the literature data.<sup>11</sup>

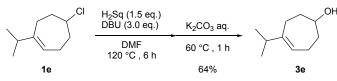
11. C. Cadot, P. I. Dalko, J. Cossy, C. Ollivier, R. Chuard, and P. Renaud, J. Org. Chem., 2002, 67, 7193.

## Synthesis of 3c from chloride 1d

BnO	H <sub>2</sub> Sq (1.5 eq.) DBU (3.0 eq.)	K <sub>2</sub> CO <sub>3</sub> aq.	BnO
 Me	DMF 90 °C , 2 h	60 °C , 1 h	 Me
1d	,	70%	3c

According to the general procedure, the reaction of **1d** (39.6 mg, 199  $\mu$ mol) and H<sub>2</sub>Sq (34.1 mg, 299  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.4  $\mu$ L, 598  $\mu$ mol) at 90 °C for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3c**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3c** (25.1 mg, 139  $\mu$ mol, 70%) as a colorless oil. The spectral data of **3c** was in good agreement with the literature data.<sup>11</sup>

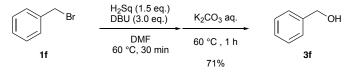
## Synthesis of 3e from chloride 1e



According to the general procedure, the reaction of **1e** (18.4 mg, 107  $\mu$ mol) and H<sub>2</sub>Sq (18.1 mg, 159  $\mu$ mol) in DMF (210  $\mu$ L) with DBU (47.9  $\mu$ L, 320  $\mu$ mol) at 120 °C for 6 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3e**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3e** (10.5 mg, 68.1  $\mu$ mol, 64%) as a colorless oil.

Data for **3e**: IR (ATR): 3342, 2957, 2927, 2853, 1463, 1361, 1293, 1035, 829 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 5.52$  (t, J = 6.6 Hz, 1H), 3.81–3.75 (m, 1H), 2.24–2.13 (m, 3H), 1.96–1.87 (m, 4H), 1.34–1.26 (m, 2H), 0.97 (d, J = 6.9, 3H), 0.96 (d, J = 6.9, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 150.1, 122.6, 74.8, 37.1, 35.9, 35.8, 24.5, 22.5, 21.4, 21.2. HRMS (FI): m/z [M]<sup>+</sup> calcd for C<sub>10</sub>H<sub>18</sub>O: 154.1358; found: 154.1365.

# Synthesis of 3f from bromide 1f



According to the general procedure, the reaction of **1f** (34.3 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 301  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3f**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3f** (15.4 mg, 142  $\mu$ mol, 71%) as a colorless oil. The spectral data of **3f** was in good agreement with the literature data.<sup>12</sup>

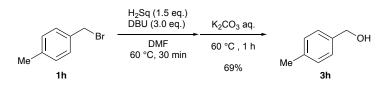
12. P. Bhattacharya, J. A. Krause, and H. Guan, Organometallics, 2011, 30, 4720.

Synthesis of 3g from chloride 1g

$$MeO \begin{array}{c} H_2Sq (1.5 eq.) \\ DBU (3.0 eq.) \\ \hline DBU (3.0 eq.) \\ \hline DMF \\ 60 \ ^\circ C, \ 30 \ min \\ 1g \\ 96\% \\ \hline 3g \\ \hline \end{array} \begin{array}{c} K_2CO_3 aq. \\ \hline MeO \\ \hline MeO \\ \hline 3g \\ \hline \end{array} \begin{array}{c} OH \\ MeO \\ \hline 3g \\ \hline \end{array}$$

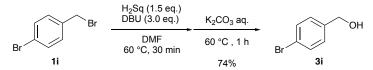
According to the general procedure, the reaction of **1g** (31.1 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 300  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3g**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3g** (26.6 mg, 193  $\mu$ mol, 96%) as a white solid. The spectral data of **3g** was in good agreement with the literature data.<sup>12</sup>

#### Synthesis of 3h from bromide 1h



According to the general procedure, the reaction of **1h** (36.5 mg, 197  $\mu$ mol) and H<sub>2</sub>Sq (34.4 mg, 301  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3h**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3h** (16.8 mg, 134  $\mu$ mol, 69%) as a white solid. The spectral data of **3h** was in good agreement with the literature data.<sup>12</sup>

#### Synthesis of 3i from bromide 1i



According to the general procedure, the reaction of **1i** (50.0 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.1 mg, 299  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3i**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **3i** (27.7 mg, 148  $\mu$ mol, 74%) as a white solid. The spectral data of **3i** was in good agreement with the literature data.<sup>13</sup>

13. A. Harinath, J. Bhattacharjee, T. K. Panda, Chem. Commun., 2019, 55, 1386.

Synthesis of 3j from bromide 1j

According to the general procedure, the reaction of **1j** (39.1 mg, 199  $\mu$ mol) and H<sub>2</sub>Sq (34.0 mg, 298  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3j**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 2/1) gave **3j** (23.2 mg, 174  $\mu$ mol, 87%) as a white solid. The spectral data of **3j** was in good agreement with the literature data.<sup>14</sup>

Synthesis of 3k from bromide 1k

$$\begin{array}{c|c} & H_2Sq \ (1.5 \ eq.) \\ \hline Br & Br & Br \\ O_2N & DMF \\ \hline 60 \ ^\circ C, \ 30 \ min \\ \hline 1k & 66\% & 3k \end{array} \xrightarrow{K_2CO_3 \ aq.} OH$$

According to the general procedure, the reaction of **1k** (43.1 mg, 200 µmol) and H<sub>2</sub>Sq (34.2 mg, 300 µmol) in DMF (400 µL) with DBU (89.7 µL, 600 µmol) at 60 °C for 0.5 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3k**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 2/1) gave **3k** (20.2 mg, 132 µmol, 66%) as a yellow solid. The spectral data of **3k** was in good agreement with the literature data.<sup>13</sup>

Synthesis of 31 from bromide 11

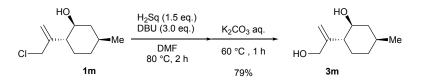
$$\begin{array}{c} \text{Ph} & \begin{array}{c} \text{H}_2\text{Sq} (1.5 \text{ eq.}) \\ \text{DBU} (3.0 \text{ eq.}) \\ \text{DBU} (3.0 \text{ eq.}) \\ \hline \\ \text{DWF} \\ 60 \ ^\circ\text{C}, 1 \text{ h} \\ \hline \\ \text{S7\%} \end{array} \begin{array}{c} \text{Ph} \\ \text{OH} \\ \begin{array}{c} \text{OH} \\ \text{S7\%} \end{array} \end{array}$$

According to the general procedure, the reaction of **11** (38.4 mg, 195  $\mu$ mol) and H<sub>2</sub>Sq (33.6 mg, 295  $\mu$ mol) in DMF (390  $\mu$ L) with DBU (87.5  $\mu$ L, 585  $\mu$ L) at 60 °C for 40 min, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **31**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **31** (14.8 mg, 110  $\mu$ mol, 57%) as a white solid. The spectral data of **31** was in good agreement with the literature data.<sup>15</sup>

<sup>14.</sup> M. Zhao, W. Xie, and C. Cui, Chem. Eur. J., 2014, 20, 9259.

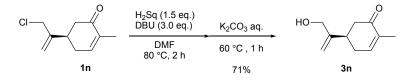
<sup>15.</sup> K. Zhu, M. P. Shaver, and S. P. Thomas, Eur. J. Org. Chem., 2015, 2119.

#### Synthesis of 3m from chloride 1m



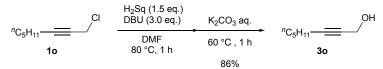
According to the general procedure, the reaction of 1m (37.3 mg, 198 µmol) and H<sub>2</sub>Sq (33.9 mg, 297 µmol) in DMF (400 µL) with DBU (88.7 µL, 593 µmol) at 80 °C for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3m**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 1/2) gave **3m** (26.5 mg, 156 µmol, 79%) as a white solid. The spectral data of **3m** was in good agreement with the literature data.<sup>16</sup>

#### Synthesis of 3n from chloride 1n



According to the general procedure, the reaction of **1n** (39.0 mg, 211 µmol) and H<sub>2</sub>Sq (35.9 mg, 315 µmol) in DMF (420 µL) with DBU (94.8 µL, 634 µmol) at 80 °C for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3n**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 1/1) gave **3n** (24.8 mg, 149 µmol, 71%) as a yellow oil. The spectral data of **3n** was in good agreement with the literature data.<sup>5</sup>

#### Synthesis of 30 from chloride 10



According to the general procedure, the reaction of **10** (28.9 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 300  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at 80 °C for 1 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **30**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 3/1) gave **30** (21.7 mg, 172  $\mu$ mol, 86%) as a yellow oil. The spectral data of **30** was in good agreement with the literature data.<sup>17</sup>

16. N. A. Clanton, N. A. Wilson, E. Ortiz, S. T. Blumberg, and D. E. Frantz, Org. Lett., 2023, 25, 277.

<sup>17.</sup> A. Gansäuer, C.-A. Fan, F. Keller, and J. Keil, J. Am. Chem. Soc., 2007, 129, 3484.

#### Synthesis of 3p from iodochloride 1p

$$I \xrightarrow{H_2Sq} (1.5 \text{ eq.})$$

$$DBU (3.0 \text{ eq.}) \xrightarrow{K_2CO_3 \text{ aq.}} HO \xrightarrow{H_4} CI$$

$$DMF \xrightarrow{f_1, 2h} 60 \text{ °C}, 1h \xrightarrow{g_2\%} 3p$$

According to the general procedure, the reaction of **1p** (31.0 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.5 mg, 302  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at rt for 2 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3p**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave **3p** (23.9 mg, 175  $\mu$ mol, 87%) as a pale yellow oil. The spectral data of **3p** was in good agreement with the literature data.<sup>18</sup> Synthesis of **3q** from bromochloride **1q** 

$$Br \underbrace{ \begin{array}{c} \begin{array}{c} H_2Sq (1.5 eq.) \\ DBU (3.0 eq.) \end{array}}_{Iq} HO \underbrace{ \begin{array}{c} H_2Sq (1.5 eq.) \\ DBU (3.0 eq.) \end{array}}_{K_2CO_3 aq.} HO \underbrace{ \begin{array}{c} \begin{array}{c} \end{array}}_{f_5} CI \\ HO \underbrace{ \begin{array}{c} \end{array}}_{f_5} CI \\ Iq \end{array} \\ 1q \\ 73\% \end{array}$$

According to the general procedure, the reaction of **1q** (44.1 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 300  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.7  $\mu$ L, 600  $\mu$ mol) at rt for 6 h, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **3q**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave **3q** (21.9 mg, 145  $\mu$ mol, 73%) as a pale yellow oil. The spectral data of **3q** was in good agreement with the literature data.<sup>19</sup> Synthesis of **3r** from bromotosylate **1r** 

$$Br \underbrace{()_{5}}_{1r} OTs = \underbrace{\begin{array}{c} H_{2}Sq (3.0 \text{ eq.}) \\ DBU (6.0 \text{ eq.}) \\ DMF \\ 100 \text{ °C}, 2.5 \text{ h} \end{array}}_{229/} HO \underbrace{()_{5}}_{929/} OH \\ 3r \\ \end{array}$$

According to the general procedure, a mixture of **1r** (74.0 mg, 212 µmol) and H<sub>2</sub>Sq (72.6 mg, 637 µmol) in DMF (420 µL) was reacted with DBU (190 µL, 1.27 mmol) at 60 °C for 2 h, and the reaction mixture was treated with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 100 °C for 2.5 h. Due to the high polarity of **3r**, the aqueous mixture was extracted with EtOAc (1 mL × 4) and THF (1 mL × 3). Purification of an obtained crude product including **3r** by silica gel column chromatography (hexane/EtOAc = 1/2) gave **3r** (23.2 mg, 175 µmol, 83%) as a colorless oil. The spectral data of **3r** was in good agreement with the literature data.<sup>20</sup>

18. K. Matsuoka, N. Komami, M. Kojima, T. Mita, K. Suzuki, S. Maeda, T. Yoshino, and S. Matsunaga, J. Am. Chem. Soc. 2021, **143**, 103.

19. R. I. Khusnutdinov, N. A. Shchadneva, R. Yu. Burangulova, Z. S. Muslimov, and U. M. Dzhemilev, *Russ. J. Org. Chem.*, 2006, **42**, 1615.

20. L. Wu, I. Fleischer, R. Jackstell, I. Profir, R. Franke, and M. Beller, J. Am. Chem. Soc., 2013, 135, 14306.

## Synthesis of 3r from bromochloride 1q

$$Br \xrightarrow{H_2 \circ q} (3.0 \text{ eq.}) \xrightarrow{K_2 CO_3 \text{ aq.}} HO \xrightarrow{H_5 \circ C} H$$

$$1q \qquad Br \xrightarrow{DMF} 100 \text{ °C}, 2.5 \text{ h} \xrightarrow{SFO}$$

According to the general procedure, a mixture of 1q (44.1 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (68.6 mg, 601  $\mu$ mol) in DMF (400  $\mu$ L) was reacted with DBU (170  $\mu$ L, 1.27 mmol) at 80 °C for 2 h, and the reaction mixture was treated with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 100 °C for 2.5 h. Due to the high polarity of 3r, the aqueous mixture was extracted with THF (1 mL × 5) Purification of an obtained crude product including 3r by silica gel column chromatography (hexane/EtOAc = 1/2) gave 3r (22.5 mg, 170  $\mu$ mol, 85%) as a colorless oil. The spectral data of 3r was in good agreement with the literature data.<sup>20</sup>

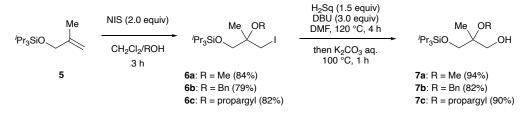
Synthesis of 3s from dibromide 1s

Note: When this reaction was conducted according to the general procedure, an E2 reaction triggered by DBU occurred preferentially; thus, the following protocol was performed.

To a solution of H<sub>2</sub>Sq (57.5 mg, 504  $\mu$ mol) in DMF (210  $\mu$ L) was added DBU (75.5  $\mu$ L, 505  $\mu$ mol) and the mixture was stirred for 5 min at 120 °C. Then, a solution of **1s** (57.2 mg, 210  $\mu$ mol) in DMF (210  $\mu$ L) was added to the reaction mixture. After stirring for 2 h at 120 °C, to the reaction mixture was added 10%(w/w) K<sub>2</sub>CO<sub>3</sub> aq (1 mL). After stirring for 1 h at 100 °C, the reaction was quenched with brine (1 mL), and the mixture was extracted with EtOAc (1 mL × 3). The organic layer was dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product including **3s**, which was purified by silica gel column chromatography (hexane/EtOAc = 1/1) to give **3s** (18.0 mg, 123  $\mu$ mol, 59%) as a pale yellow oil. The spectral data of **3s** was in good agreement with the literature data.<sup>21</sup>

21. Q. Yao, Org. Lett. 2002, 4, 2197.

#### Synthesis of alcohol 7 from alkene 5



#### General procedure of synthesis of ether 6 and 7

To a solution of 5 in a mixture of CH<sub>2</sub>Cl<sub>2</sub> and ROH (ca. 1:1. 0.24 M) was added NIS (2.0 eq.) at rt. After stirring for 3 h, the reaction was quenched with 10% Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> aq. (1 mL). The mixture was extracted with diethyl ether for three times, and the combined organic layers were dried over MgSO4 and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography to give 6.

To a solution of 6 (1.0 eq.) and H<sub>2</sub>Sq (1.5 eq.) in DMF (0.5 M) was added DBU (3.0 eq.) at rt, and the reaction mixture was stirred at 120 °C. After stirring for 4 h, the reaction mixture was added 10% K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 100 °C, and the mixture was further stirred for 1 h at the same temperature. The reaction was quenched with brine (1 mL), and the aqueous mixture was extracted with EtOAc (1 mL × 3). The combined organic layers were dried over MgSO4 and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography to give 7.

# Synthesis of ether 6a

<sup>/</sup>Pr₂SiO

According to the general procedure, the reaction of 5 (61.0 mg, 267  $\mu$ mol) in MeOH (530 OMe Me  $\mu L,\,13.1$  mmol) and CH\_2Cl\_2 (530  $\mu L)$  with NIS (122 mg, 542  $\mu mol)$  at rt for 3 h afforded a crude product including 6a. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 20/1) gave **6a** (86.1 mg, 223 µmol, 84%) as a colorless oil.

Data for **6a**: IR (ATR): 2941, 2892, 2865, 1462, 1201, 1119, 882, 809, 682 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 3.76$  (d, J = 9.7 Hz, 1H), 3.54 (d, J = 9.7 Hz, 1H), 3.51 (d, J = 10.3 Hz, 1H), 3.33 (d, J = 10.3Hz, 1H), 3.27 (s, 3H), 1.36 (s, 3H), 1.19–1.06 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): δ = 75.6, 67.7, 50.0, 19.2, 18.2 (6C), 13.3, 12.0 (3C). HRMS (FI): m/z [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>32</sub>IO<sub>2</sub>Si: 387.1216; found: 387.1203.

## Synthesis of ether 6b

According to the general procedure, the reaction of 5 (47.8 mg, 209  $\mu$ mol) in BnOH (420 OBn Me <sup>/</sup>Pr<sub>3</sub>SiO  $\mu$ L, 4.04 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (530  $\mu$ L) with NIS (94.3 mg, 419  $\mu$ mol) at rt for 3 h afforded 6h a crude product including **6b**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 100/1) gave **6b** (76.0 mg, 164  $\mu$ mol, 79%) as a colorless oil.

Data for **6b**: IR (ATR): 2942, 2890, 2865, 1462, 1383, 1183, 1115, 1070, 882, 814, 736, 683 cm<sup>-1</sup>. <sup>1</sup>H NMR  $(CDCl_3, 500 \text{ MHz}): \delta = 7.41-7.26 \text{ (m, 5H)}, 4.54 \text{ (d, } J = 10.9 \text{ Hz}, 1\text{ H}), 4.48 \text{ (d, } J = 10.9 \text{ Hz}, 1\text{ H}), 3.85 \text{ (d, } J = 10.9 \text{ Hz}, 1\text{ H})$ = 9.8 Hz, 1H), 3.63 (d, J = 9.8 Hz, 1H), 3.62 (d, J = 10.9 Hz, 1H), 3.45 (d, J = 10.9 Hz, 1H), 1.47 (s, 3H),

1.25–1.07 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta$  = 138.7, 128.4 (2C), 127.9 (2C), 127.6, 76.0, 68.2, 64.5, 20.2, 18.2 (6C), 13.7, 12.0 (3C). HRMS (FI): m/z [M + H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>36</sub>IO<sub>2</sub>Si: 463.1529; found: 463.1548.

#### Synthesis of ether 6c

According to the general procedure, the reaction of **5** (113 mg, 494 µmol) in propargyl alcohol (990 µL, 17.1 mmol) and CH<sub>2</sub>Cl<sub>2</sub> (990 µL) with NIS (222 mg, 987 µmol) at rt for 3 h afforded a crude product including **6c**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 30/1) gave **6c** (166.9 mg, 407 µmol, 82%) as a colorless oil. Data for **6c**: IR (ATR): 3310, 2942, 2892, 2866, 2364, 1462, 1382, 1180, 1114, 1070, 996, 882, 798, 683 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 4.24$  (d, J = 14.9 Hz, 1H), 4.19 (d, J = 14.9 Hz, 1H), 3.81 (d, J = 9.7 Hz, 1H), 3.64 (d, J = 9.7 Hz, 1H), 3.50 (d, J = 10.9 Hz, 1H), 3.35 (d, J = 10.9 Hz, 1H), 2.41 (s, 1H), 1.43 (s, 3H), 1.29–0.94 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 80.8, 77.0, 73.8, 68.0, 51.3, 20.2, 18.1 (6C), 12.8, 11.2 (3C). HRMS (FD): <math>m/z$  [M]<sup>+</sup> calcd for C<sub>16</sub>H<sub>31</sub>IO<sub>2</sub>Si: 411.1216; found: 411.1216.

# Synthesis of alcohol 7a

<sup>i</sup>Pr<sub>3</sub>SiO  $\xrightarrow{\text{OMe}}$  OH According to the general procedure, **6a** (77.6 mg, 201 µmol) and H<sub>2</sub>Sq (34.3 mg, 301 µmol) in DMF (400 µL) with DBU (90.2 µL, 603 µmol) at 120 °C for 4 h, followed by the treatment of the mixture with 10% K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 100 °C for 1 h afforded a crude product including **7a**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave **7a** (51.9 mg, 188 µmol, 94%) as a colorless oil.

Data for **7a:** IR (ATR): 3459, 2942, 2866, 1464, 1109, 1069, 882, 805, 681 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 3.77$  (d, J = 9.7 Hz, 1H), 3.65–3.55 (m, 3H), 3.30 (s, 3H), 2.27 (t, J = 6.0 Hz, 1H), 1.15 (s, 3H), 1.14–0.86 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 77.2$ , 67.1, 65.6, 50.1, 18.1 (6C), 16.9, 12.0 (3C). HRMS (FD): m/z [M + H]<sup>+</sup> calcd for C<sub>14</sub>H<sub>33</sub>O<sub>3</sub>Si: 277.2199; found: 277.2212.

# Synthesis of alcohol 7b

 $\begin{array}{c} \stackrel{\text{Me}}{}_{\text{Pr}_{3}\text{SiO}} \stackrel{\text{OBn}}{}_{\text{7b}} \quad \text{According to the general procedure, 6b (103 mg, 223 \mu mol) and H_2Sq (38.3 mg, 336 \mu mol) in DMF (450 \mu L) with DBU (100 \mu L, 669 \mu mol) at 120 °C for 4 h, followed by the treatment of the mixture with 10% K_2CO_3 aq. (1 mL) at 100 °C for 1 h afforded a crude product including 7b. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave 7b (64.7 mg, 184 \mu mol, 82%) as a colorless oil.$ 

Data for **7b**: IR (ATR): 3473, 2942, 2866, 1463, 1385, 1109, 1065, 882, 804, 736, 682 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.34-7.23$  (m, 5H), 4.57 (s, 2H), 3.85 (d, J = 9.7 Hz, 1H), 3.70 (d, J = 9.7 Hz, 1H), 3.70 (d, J = 11.5 Hz, 1H), 3.65 (d, J = 11.5 Hz, 1H), 1.27 (s, 3H), 1.17–0.99 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 139.2$ , 128.5 (2C), 127.6 (2C), 127.5, 77.9, 67.6, 66.3, 64.7, 18.1 (6C), 17.7, 12.0 (3C). HRMS (FD): m/z [M + H]<sup>+</sup> calcd for C<sub>20</sub>H<sub>37</sub>O<sub>3</sub>Si: 353.2512; found: 353.2527.

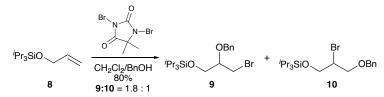
## Synthesis of alcohol 7c

According to the general procedure, **6c** (81.8 mg, 200  $\mu$ mol) and H<sub>2</sub>Sq (34.2 mg, 300  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (89.6  $\mu$ L, 599  $\mu$ mol) at 120 °C for 4 h, followed by the treatment of the mixture with 10% K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 100 °C for 1 h afforded a

crude product including 7c. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave 7c (53.8 mg, 180 µmol, 90%) as a colorless oil.

Data for **7c**: IR (ATR): 3447, 3311, 2941, 2866, 2359, 1463, 1385, 1108, 1063, 881, 804, 681 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 4.28$  (d, J = 15.8 Hz, 1H), 4.24 (d, J = 15.8 Hz, 1H), 3.80 (d, J = 10.3 Hz, 1H), 3.67–3.58 (m, 3H), 2.42 (t, J = 2.3 Hz, 1H), 2.25 (t, J = 6.3 Hz, 1H), 1.23 (s, 3H), 1.14–0.86 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 81.4$ , 79.0, 73.7, 67.5, 65.9, 51.3, 18.1 (6C), 17.4, 11.9 (3C). HRMS (FI): m/z [M]<sup>+</sup> calcd for C<sub>16</sub>H<sub>32</sub>O<sub>3</sub>Si: 300.2121; found: 300.2112.

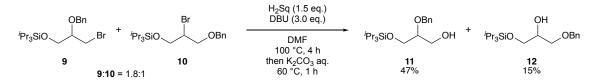
Synthesis of a 1.8:1 mixture of bromide 9 and 10



To a solution of **8** (187 mg, 871 µmol) in (CH<sub>2</sub>Cl)<sub>2</sub> (870 µL) and BnOH (870 µL) was added 1,3-dibromo-5,5-dimethylhydantoin (150 mg, 523 µmol). After stirring for 4 h at rt, the reaction was quenched with saturated NH<sub>4</sub>Cl aq. (2 mL), and the mixture was extracted with diethyl ether (1 mL × 3). The organic layer was dried (MgSO<sub>4</sub>) and filtered. The filtrate was concentrated under reduced pressure to give crude product. To remove excess BnOH, the crude product was roughly purified by silica gel column chromatography (hexane/EtOAc = 20/1) to give an impure product including **9** and **10**. The mixture was further purified by silica gel column chromatography (hexane/EtOAc = 100/1) to give an inseparable mixture of **9** and **10** (139 mg) as a pale yellow oil and a mixture of two desired products and benzaldehyde dibenzylacetal (157 mg). The impure product was purified by PTLC (0.05 mm, hexane/EtOAc = 10/1) to give an inseparable mixture of **9** and **10** (140 mg). The total amount of the obtained mixture of **9** and **10** were 279 mg (696 µmol, 80%, **9:10** = 1.8:1).

NMR data of a 1.8:1 mixture of **9** and **10**: <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz): δ = 7.39–7.26 (m, 5H), 4.71 (d, *J* = 11.5 Hz, 0.64H), 4.64 (d, *J* = 11.5 Hz, 1H, 0.64H), 4.59 (s, 0.72H), 4.12 (dddd, *J* = 6.5, 5.8, 5.3, 4.5 Hz, 0.36H), 4.03–3.94 (m, 0.72H), 3.88 (dd, *J* = 10.3, 5.3 Hz, 0.36H), 3.84–3.74 (m, 1.28 + 0.36H), 3.70–3.61 (m, 1.28H), 3.54 (dd, *J* = 10.4 , 4.8 Hz, 0.64H), 1.15–0.96 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 138.2, 138.0, 128.5 (4C), 128.0 (2C), 127.9, 127.8 (3C), 78.9, 73.4, 72.4 70.9, 64.6, 63.7, 52.4, 33.1, 18.1 (12C), 12.1 (3C), 12.0 (3C).

#### Synthesis of alcohol 11 and 12

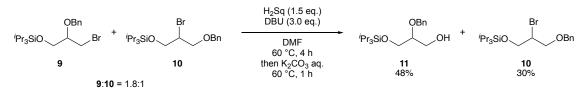


To a solution of H<sub>2</sub>Sq (32.5 mg, 285 µmol) in DMF (180 µL) was added DBU (85.1 µL, 569 µmol), and the mixture was stirred at 100 °C for 5 minutes. Then, a 1.8:1 mixture of **9** and **10** (76.2 mg, 190 µmol) in DMF (180 µL) was added to the reaction mixture. After stirring for 4 h at 100 °C, 10% K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) was added to the reaction mixture, and the mixture was further stirred for 1 h at 60 °C. The reaction was quenched with brine (1 mL), and the mixture was filtered though a Celite pad to remove insoluble solids. The filtrate was extracted with Et<sub>2</sub>O (1 mL × 3). The organic layer was dried (MgSO<sub>4</sub>) and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc = 15/1) to give **11** (30.0 mg, 88.7 µmol, 47%) as a pale yellow oil and **12** (9.6 mg, 28.4 µmol, 15%) as a pale yellow oil.

Data for **11**: IR (ATR): 3442, 2942, 2866, 1463, 1105, 1068, 882, 683 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.39-7.26$  (m, 5H), 4.71 (d, J = 11.5 Hz, 1H), 4.63 (d, J = 11.5 Hz, 1H), 3.86 (dd, J = 10.5, 5.0 Hz, 1H), 3.82-3.74 (m, 2H), 3.69 (dd, J = 11.5, 5.0 Hz, 1H), 3.61 (dd, J = 10.5, 5.0 Hz, 1H), 1.15–0.98 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 138.5, 128.6 (2C), 127.9 (3C), 79.6, 72.4, 63.9, 63.2, 18.1 (6C), 12.0 (3C). HRMS (FI): m/z [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>35</sub>O<sub>3</sub>Si: 339.2356; found: 339.2351.

Data for **12**: IR (ATR): 2943, 2866, 1456, 1104 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.37-7.26$  (m, 5H), 4.56 (s, 2H), 3.90–3.84 (m, 1H), 3.79–3.69 (m, 2H), 3.57 (dd, J = 9.5, 5.0 Hz, 1H), 3.53 (d, J = 9.5, 6.0 Hz, 1H), 1.15–0.98 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 138.2, 128.5 (2C), 127.9 (2C), 127.8, 73.6, 71.1, 71.0, 64.4, 18.1 (6C), 12.0 (3C). HRMS (FI): m/z [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>35</sub>O<sub>3</sub>Si: 339.2356; found: 339.2355.

Synthesis of alcohol 11 and recovery of bromide 10



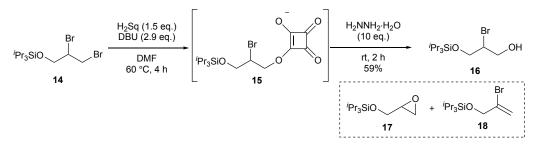
To a solution of H<sub>2</sub>Sq (20.4 mg, 179 µmol) in DMF (120 µL) was added DBU (53.5 µL, 358 µmol), and the mixture was stirred at 100 °C for 5 minutes. Then, a solution of a 1.8:1 mixture of **9** and **10** (111 mg, 0.276 mmol) in DMF (120 µL) was added to the reaction mixture. After stirring for 2 h at 60 °C, 10% K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) was added to the reaction mixture, and the mixture was further stirred for 1 h at 60 °C. The reaction was quenched with brine (1 mL), and the mixture was filtered through a Celite pad to remove insoluble solids. The filtrate was extracted with Et<sub>2</sub>O (1 mL × 3). The organic layer was dried (MgSO<sub>4</sub>) and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc = 30/1 to 10/1) to give **11** (19.2 mg, 56.7 µmol, 48%) as a pale yellow oil and **10** (14.4 mg, 35.9 µmol, 30%) as a pale yellow oil.

Data for **10**: IR (ATR): 3446, 2942, 2866, 1457, 1100, 882, 773, 682 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.39-7.26$  (m, 5H), 4.59 (s, 2H), 4.12 (dddd, J = 6.5, 5.8, 5.3, 4.5 Hz, 1H), 4.03–3.94 (m, 2H), 3.88 (dd, J = 10.3, 5.3 Hz, 1H), 3.77 (dd, J = 10.3, 5.8 Hz, 1H), 1.15–0.96 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 138.0, 128.5 (2C), 127.8 (3C), 73.4, 70.9, 64.6, 52.4, 18.1 (6C), 12.1 (3C). HRMS (FD): m/z [M + H]<sup>+</sup> calcd for C<sub>19</sub>H<sub>34</sub>BrO<sub>2</sub>Si: 401.1511; found: 401.1499.

#### Synthesis of dibromide 14

To a solution of **8** (627 mg, 2.92 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (8.9 mL) was added pyridinium tribromide (1.12 g, 3.51 mmol) at 0 °C, and the reaction mixture was stirred for 2.5 h at rt. After strring for 1 h at 0 °C, the reaction was quenched with 10 % NaHSO<sub>3</sub> aq. (5 mL). The mixtue was diluted with hexane (10 mL), and the organic layer was washed with water (10 mL × 3). The organic layer was dried over MgSO<sub>4</sub>, and filtered. The filtrate was concentrated under reduced pressure to give **14** (1.07 g, 2.85 mmol, 98%) as a colorless oil. Data for **14**: IR (ATR): 2943, 2866, 1459, 1142, 1068, 995, 930,882, 774, 684 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 4.24-4.12$  (m, 2H), 4.01 (dd, J = 10.5, 4.8 Hz, 1H), 3.90 (dd, J = 10.0, 8.3 Hz, 1H), 3.79 (dd, J = 10.5, 4.5 Hz, 1H), 1.19–0.98 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 64.7, 52.1, 33.0, 18.1 (6C), 12.1 (3C). HRMS (FD): m/z [M + H]<sup>+</sup> calcd for C<sub>12</sub>H<sub>27</sub>Br<sub>2</sub>OSi: 373.0198; found: 373.0207.

#### Synthesis of bromohydrin 16



To a solution of H<sub>2</sub>Sq (30.8 mg, 277 µmol) in DMF (180 µL) was added DBU (79.3 µL, 531 µmol). After stirring for 5 min at 60 °C, a solution of **14** (69.0 mg, 184 µmol) in DMF (180 µL) was added to the reaction mixture. After stirring for 4 h at 60 °C, H<sub>2</sub>NNH<sub>2</sub>·H<sub>2</sub>O (89.4 µL, 1.84 mmol) was added to the reaction mixture at 0 °C. After stirring for 2 h at rt, the reaction was quenched with brine (1 mL). After the mixture was filtered through a Celite pad, the filtrate was extracted with EtOAc (1 mL × 3). The combined organic layers were dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/ethyl acetate = 30/1 to 10/1) to give **16** (33.8 mg, 109 µmol, 59%) as a colorless oil. Byproducts **17** (2.0 mg, 8.68 µmol, 5%) and **18** (10.3 mg, 35.1 µmol, 19%) was also isolated through this purification.

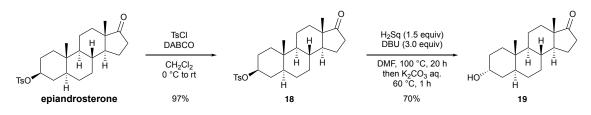
Data for **16**: IR (ATR): 3404, 2943, 2866, 1463, 1102, 881, 791, 684 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  = 4.17–4.11 (m, 1H), 4.05 (dd, *J* = 10.5, 4.5 Hz, 1H), 4.02–3.90 (m, 3H), 2.36 (t, *J* = 6.6 Hz, 1H), 1.17–0.96 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 66.0, 65.4, 54.7, 18.0 (6C), 12.0 (3C). HRMS (FI): *m/z* [M + H]<sup>+</sup> calcd for C<sub>12</sub>H<sub>28</sub>BrO<sub>2</sub>Si: 311.1042; found: 311.1041.

#### Synthesis of azidoalcohol 13

$$\stackrel{\text{Br}}{\stackrel{\text{iPr}_{3}\text{SiO}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}\\{\stackrel{\text{OH}}}\stackrel{\text{OH}}{\stackrel{\text{OH}}}\stackrel{\text{OH}}\\{\stackrel{\text{OH}}}\stackrel{\text{OH}}\\{\stackrel{\text{OH}}}\stackrel{\text{OH}}\\{\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text{OH}}}\stackrel{\text$$

To a solution of **16** (29.8 mg, 95.7 µmol) in DMF (480 µL) was added NaN<sub>3</sub> (62.2 mg, 957 µmol). After stirring for 2 d at 60 °C, the reaction was quenched with brine (1 mL), and the mixture was extracted with EtOAc (1 mL × 3). The combined organic layers were dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/ethyl acetate = 10/1) to give **13** (22.7 mg, 83.0 µmol, 87%) as a colorless oil. Data for **13**: IR (ATR): 3392, 2943, 2892, 2867, 2095, 1463, 1270, 1123, 1028, 882, 774, 684 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta$  = 3.89 (d, *J* = 5.7 Hz, 2H), 3.78 (ddd, *J* = 11.2, 6.0, 4.7 Hz, 1H), 3.69 (dt, *J* = 11.2, 6.0 Hz, 1H), 3.58 (ddd, *J* = 6.0, 5.7, 4.7 Hz, 1H), 1.96 (t, *J* = 6.0 Hz, 1H), 1.19–0.99 (m, 21H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz): 64.5, 64.4, 62.9, 18.0 (6C), 12.0 (3C). HRMS (FI): *m/z* [M + H]<sup>+</sup> calcd for C<sub>12</sub>H<sub>28</sub>N<sub>3</sub>O<sub>2</sub>Si: 274.1951; found: 274.1942.

#### Synthesis of tosylate (18) and androsterone (19)



#### Synthesis of tosylate (18)

To a solution of epiandrosterone (600 mg, 2.07 mmol) and DABCO (600 mg, 5.35 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (6 mL) was added TsCl (600 mg, 3.15 mmol) at 0 °C. After stirring for 1 h at rt, the reaction was quenched with sat. NaHCO<sub>3</sub> aq. (6 mL), and the aqueous mixture was extracted with CH<sub>2</sub>Cl<sub>2</sub> for three times. The combined organic layers were dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc = 5/1) to give tosylate **18** (581 mg, 2.00 mmol, 97%) as a white solid.

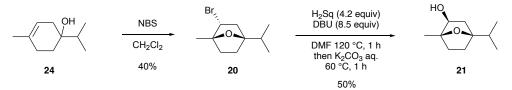
Data for **18**: mp 166–167 °C (recrystallized from Et<sub>2</sub>O, colorless needle). IR (ATR): 2937, 2860, 1738, 1360, 1175, 930 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 7.79$  (d, J = 8.0 Hz, 2H), 7.33 (d, J = 8.0 Hz, 2H), 4.44–4.38 (m, 1H), 2.47–2.40 (m, 4H), 2.09–2.02 (m, J = 9.5 Hz, 1H), 1.93–1.88 (m, 1H), 1.78–1.75 (m, 3H), 1.69 (dt, J = 13.4, 3.6 Hz, 1H), 1.65–1.44 (m, 6H), 1.32–1.17 (m, 5H), 1.12–1.07 (m, 1H), 0.97–0.88 (m, 2H), 0.85 (d, J = 11.5 Hz, 3H), 0.81 (s, 3H), 0.64 (td, J = 11.3, 3.6 Hz, 1H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 221.2$ , 144.4, 134.6, 129.7 (2C), 127.6 (2C), 82.1, 54.1, 51.2, 47.7, 44.7, 36.7, 35.8, 35.3, 34.9, 34.8, 31.4, 30.7, 28.2, 28.0, 21.7, 21.6, 20.4, 13.8, 12.1. HRMS (FD): m/z [M]<sup>+</sup> calcd for C<sub>26</sub>H<sub>36</sub>O<sub>4</sub>S: 444.2334; found: 444.2319.

# Synthesis of androsterone (19)

According to the general procedure for the synthesis of **7** from **6**, the reaction of **18** (89.4 mg, 202 µmol) with a mixture of H<sub>2</sub>Sq (34.2 mg, 300 µmol) and DBU (89.6 µL, 600 µmol)) in DMF (1 mL) at 100 °C for 20 h, followed by the treatment of the mixture with 10% K<sub>2</sub>CO<sub>3</sub> aq. (2 mL) at 60 °C for 1 h afforded a crude product including **19**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc = 5/1) gave **19** (40.6 mg, 140 µmol, 70%) as a white solid.

Data for **19**: mp 185–188 °C (recrystallized from hexane, colorless needle). IR (ATR): 3433, 2854, 1735, 1450, 1370, 1059, 1027, 1000, 731 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 4.04$  (t, J = 2.6 Hz, 1H), 2.45–2.39 (m, 1H), 2.10–2.02 (m, 1H), 1.95–1.90 (m, 1H), 1.78 (qd, J = 6.4, 3.4 Hz, 2H), 1.71–1.17 (m, 16H), 1.01 (qd, J = 12.3, 4.9 Hz, 1H), 0.85 (s, 3H), 0.80–0.77 (m, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 221.5$ , 66.4, 54.4, 51.5, 47.8, 39.1, 36.2, 35.8, 35.7, 35.0, 32.1, 31.5, 30.8, 29.0, 28.2, 21.7, 20.0, 13.8, 11.2. HRMS (FD): m/z [M]<sup>+</sup> calcd for C<sub>19</sub>H<sub>30</sub>O<sub>2</sub>: 290.2246; found: 290.2232.

## Synthesis of bromide (20) and alcohol (21)



## Synthesis of bromide (20)

To a solution of NBS (801 mg, 4.50 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10.0 mL) was added  $24^{22}$  (500 µL, 3.05 mmol). After stirring for 1.5 h at rt, the reaction was quenched with sat. NaHCO<sub>3</sub> aq., and the aqueous mixture was extracted with hexane for three times. The combined organic layers were dried over MgSO<sub>4</sub> and filtered. The filtrate was concentrated under reduced pressure to give a crude product, which was purified by silica gel column chromatography (hexane/EtOAc= 35/1) to give 20 (281 mg, 1.21 mmol, 40%) as a colorless oil.

Data for **20**: IR (ATR): 2960, 2874, 1467, 1384, 831, 671 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 3.86$  (qd, J = 5.3, 2.3 Hz, 1H), 2.56–2.51 (m, 1H), 2.35–2.30 (m, 1H), 2.03–1.98 (m, 1H), 1.73 (dd, J = 12.9, 4.9 Hz, 1H), 1.67–1.63 (m, 2H), 1.56–1.52 (m, 1H), 1.44 (s, 3H), 0.96 (d, J = 6.9 Hz), 0.96 (d, J = 6.9 Hz), 0.94 (d, J = 6.9 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 126 MHz):  $\delta = 90.4, 85.9, 53.6, 43.7, 33.1, 32.7, 32.0, 18.8, 17.8, 17.6.$  HRMS (FD): m/z [M]<sup>+</sup> calcd for C<sub>10</sub>H<sub>17</sub>BrO: 232.0463; found: 232.0467.

#### Synthesis of alcohol (21)

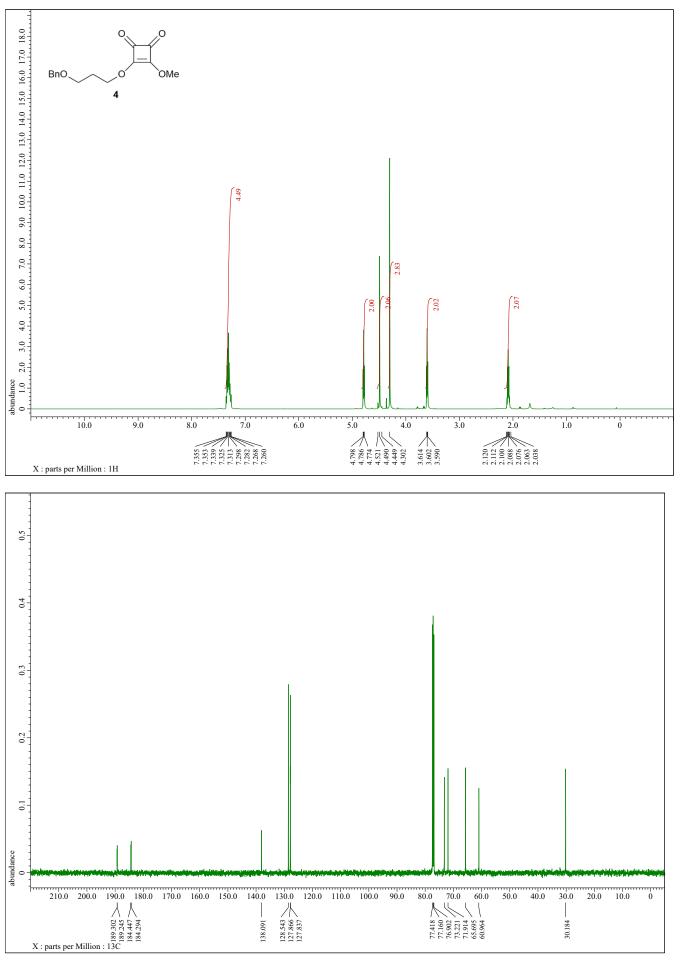
According to the general procedure for the synthesis of **3** from **1**, the reaction of **20** (44.1 mg, 189  $\mu$ mol) and H<sub>2</sub>Sq (91.2 mg, 800  $\mu$ mol) in DMF (400  $\mu$ L) with DBU (240  $\mu$ L, 1.71 mmol) at 120 °C for 1 d, followed by the treatment of the mixture with 10% (w/w) K<sub>2</sub>CO<sub>3</sub> aq. (1 mL) at 60 °C for 1 h afforded a crude product including **21**. Purification of the crude product by silica gel column chromatography (hexane/EtOAc= 4/1) gave **21** (16.1 mg, 94.5  $\mu$ mol, 50%) as a white solid.

Data for **21**: mp. 61–64 °C (recrystallized from CDCl<sub>3</sub>, colorless needle). IR (ATR): 3260, 2964, 1460, 1115, 1053 cm<sup>-1</sup>. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 500 MHz):  $\delta = 3.74$  (dd, J = 7.7, 7.7 Hz, 1H), 2.09–2.04 (m, 2H), 1.69–1.36 (m, 9H), 0.97 (d, J = 6.9 Hz, 3H), 0.95 (d, J = 6.9 Hz, 3 H). <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 126 MHz): 88.1, 85.5, 76.8, 45.8, 33.2, 33.0, 32.5, 18.3, 18.2, 16.6. HRMS (FD): m/z [M]<sup>+</sup> calcd for C<sub>10</sub>H<sub>18</sub>O<sub>2</sub>: 170.1307; found: 170.1313.

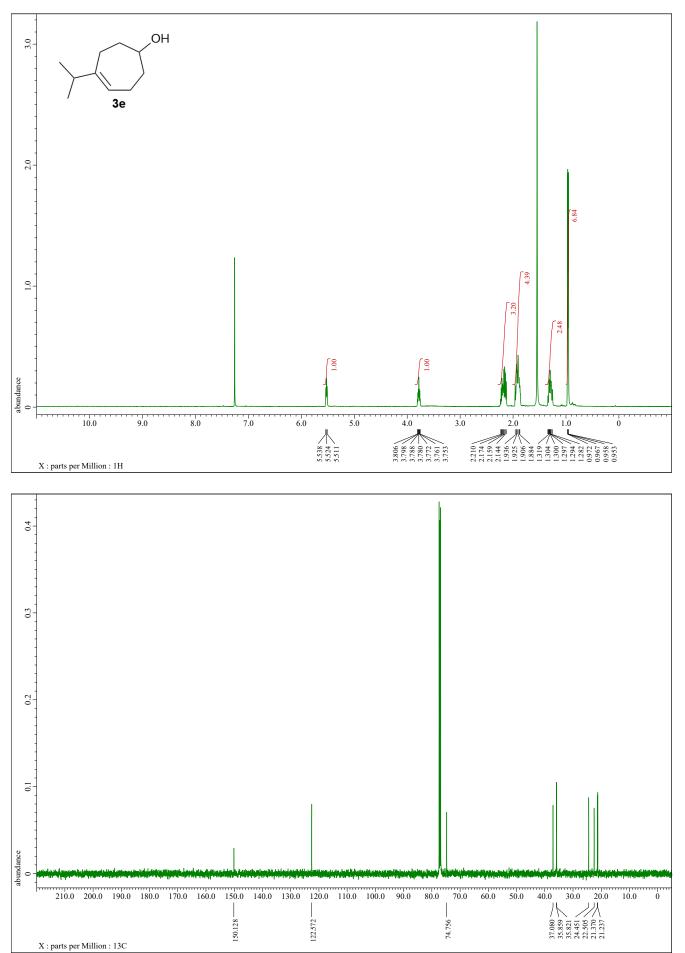
22. R. M. Carman, and B. N. Venzke, Aust. J. Chem., 1973, 26, 2235.

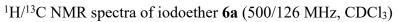
ESI-4 NMR spectra of novel compounds

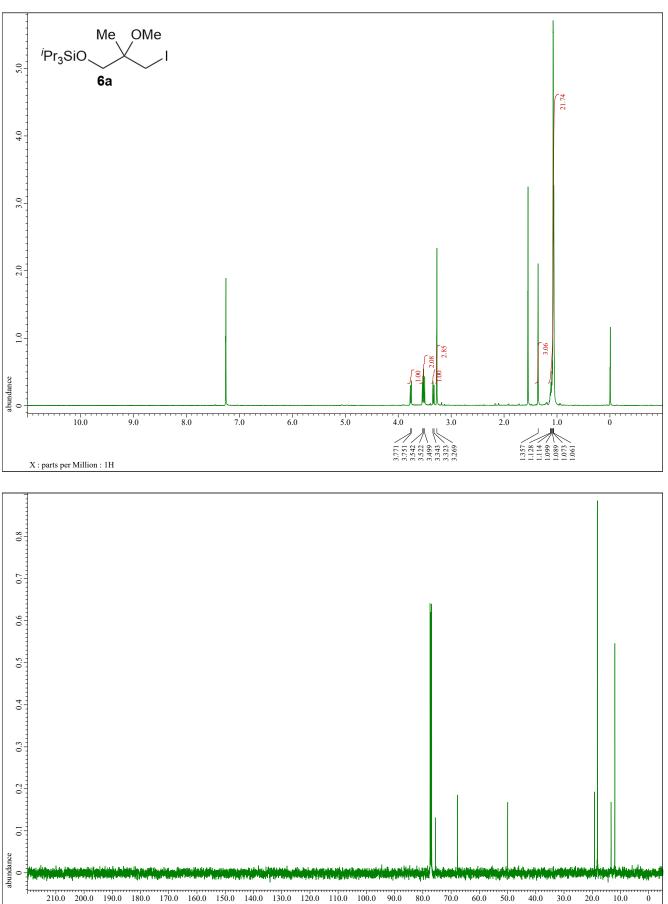




<sup>1</sup>H/<sup>13</sup>C NMR spectra of alcohol **3e** (500/126 MHz, CDCl<sub>3</sub>)







X : parts per Million : 13C

60.0

49.966

70.0

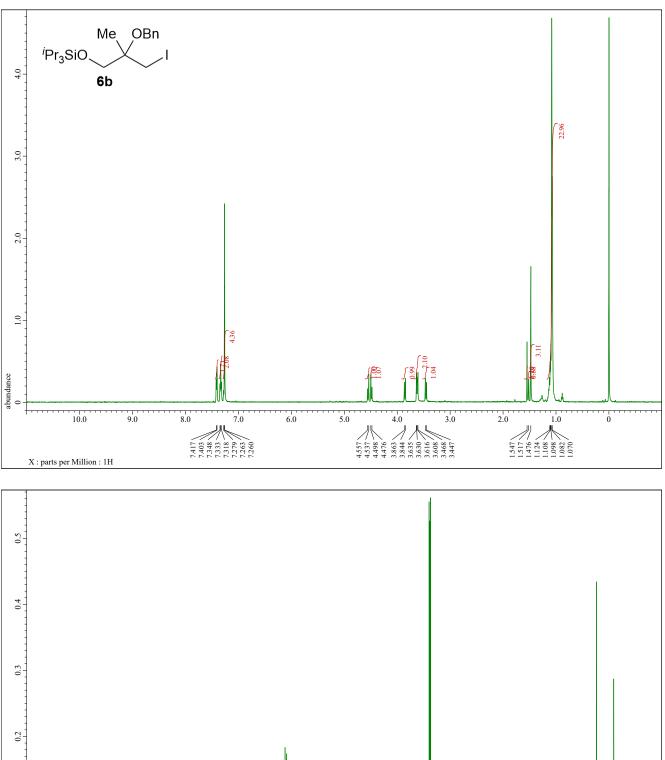
80.0

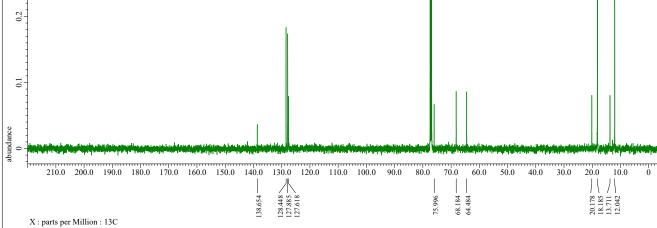
75.567 67.717 30.0

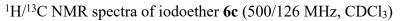
19.205 18.146 13.310 12.023

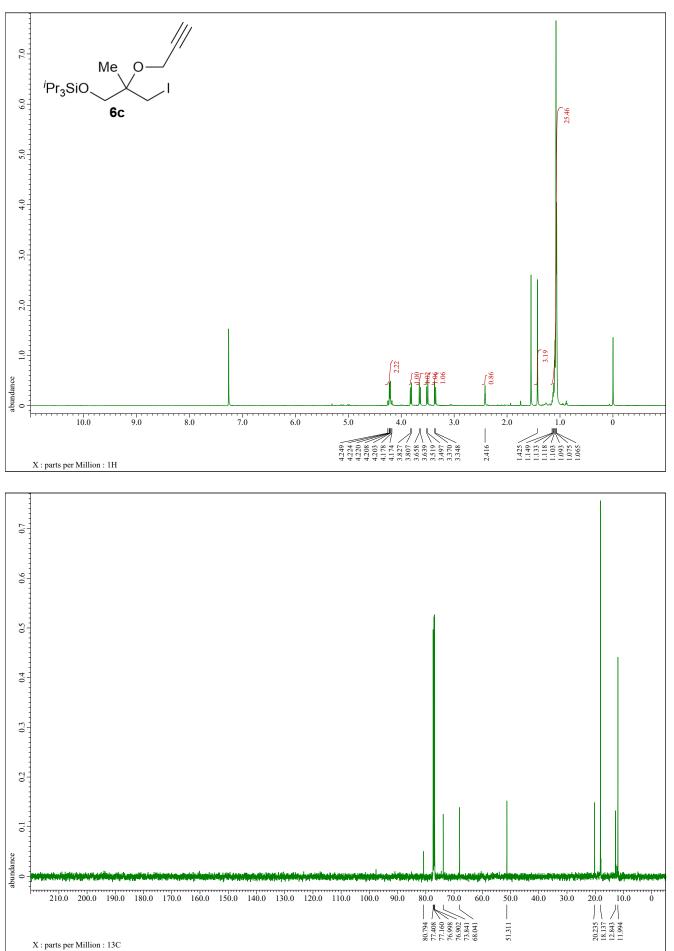
0

210.0 200.0 190.0 180.0 170.0 160.0 150.0 140.0 130.0 120.0 110.0 100.0 90.0

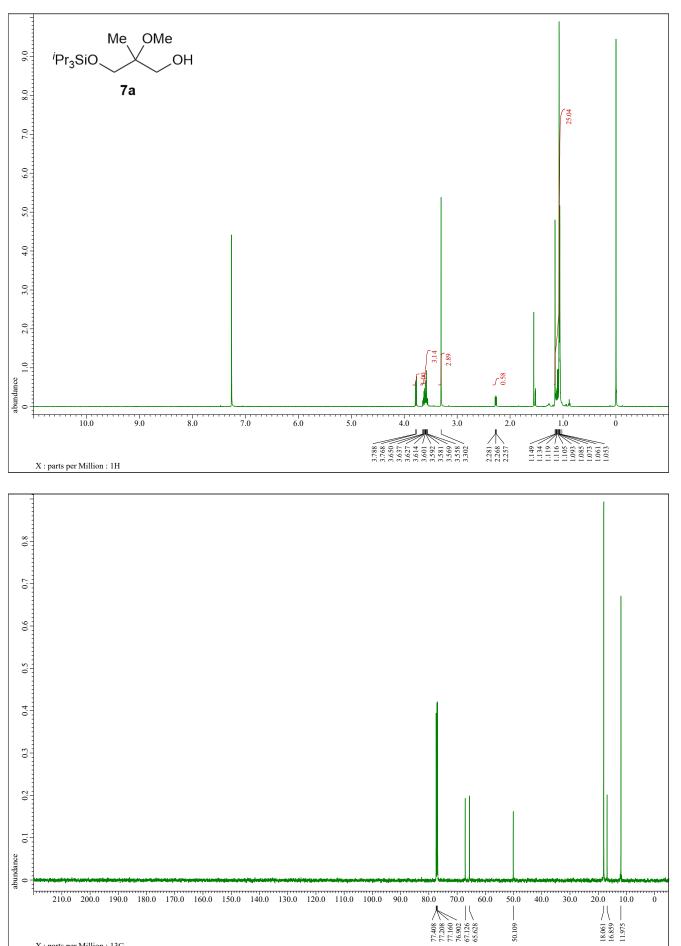




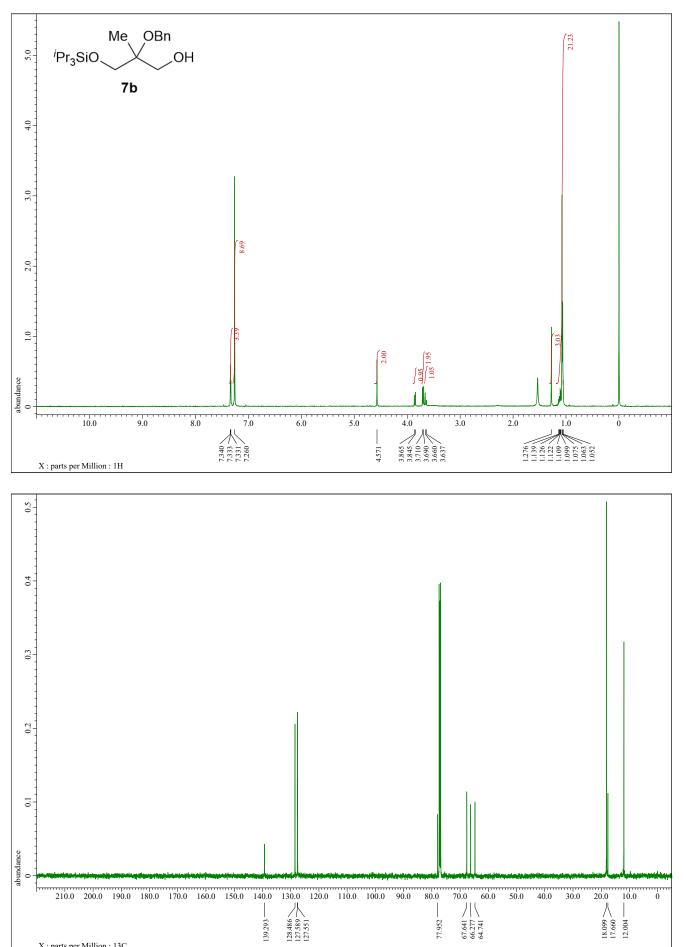




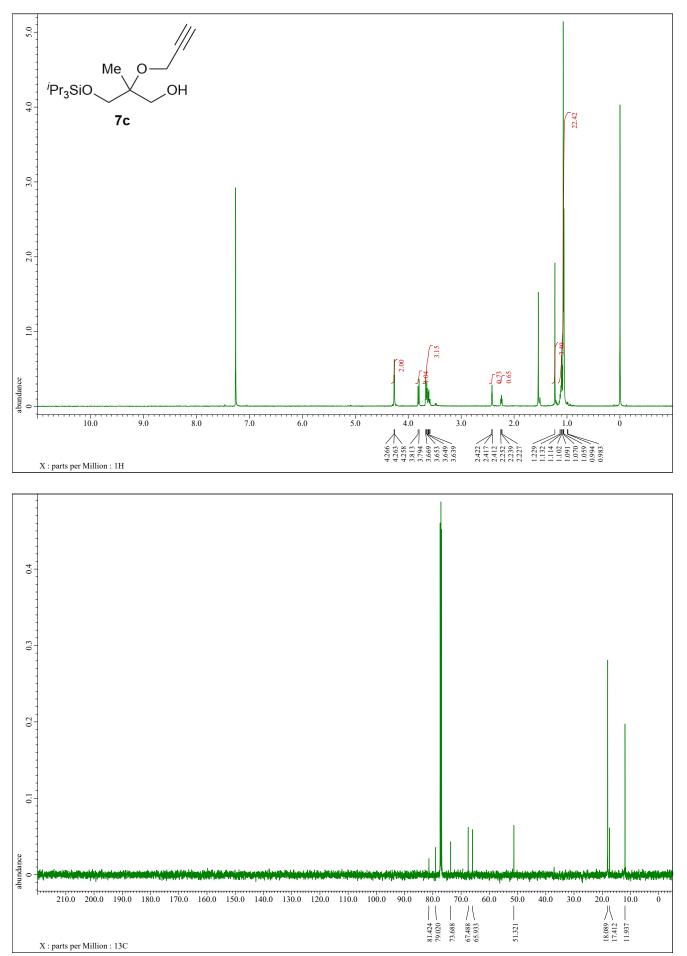
S5

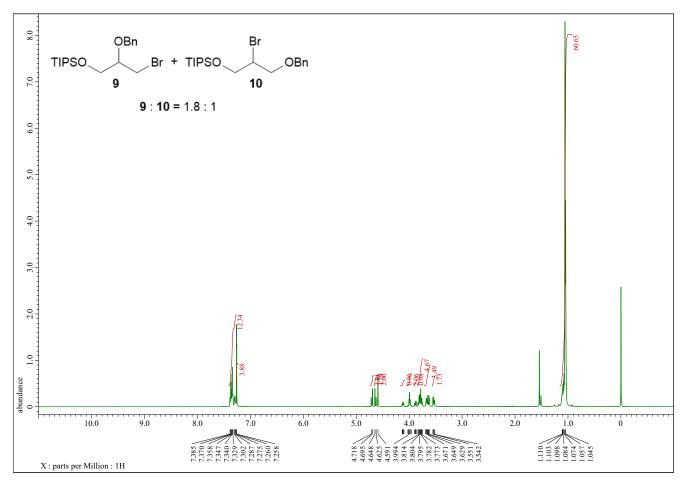


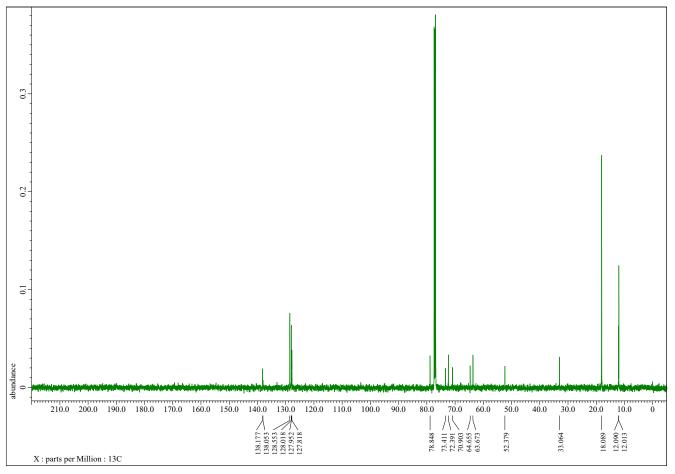
X : parts per Million : 13C

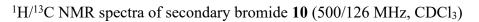


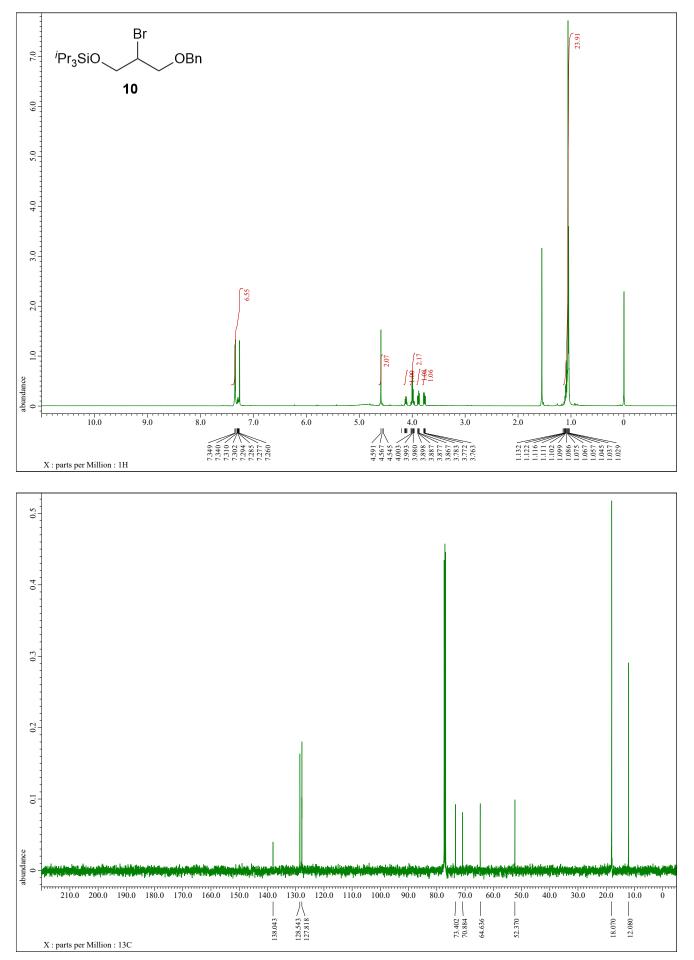
X : parts per Million : 13C

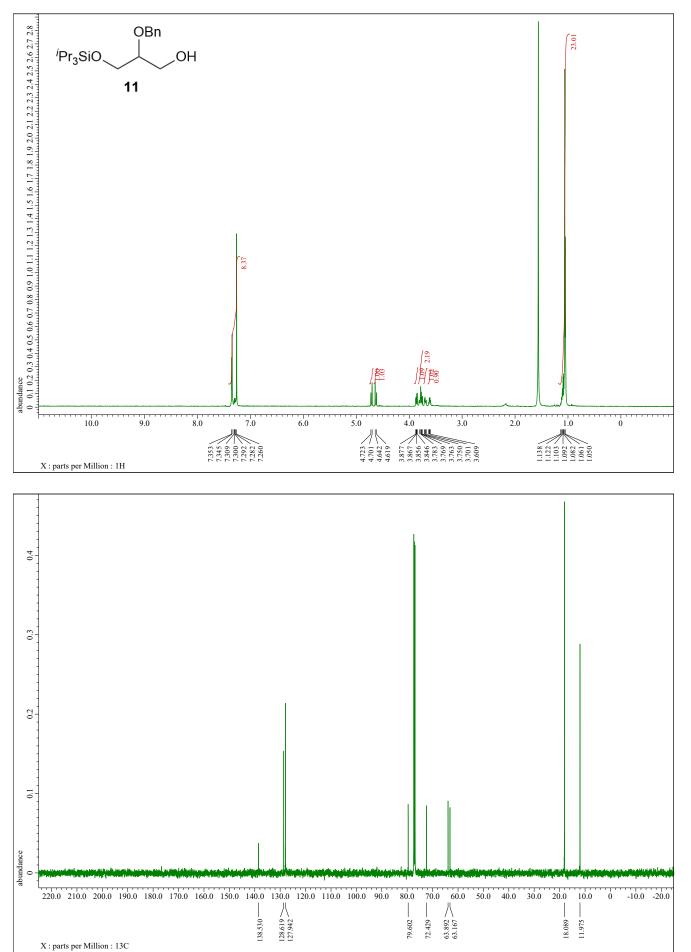




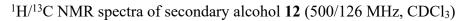


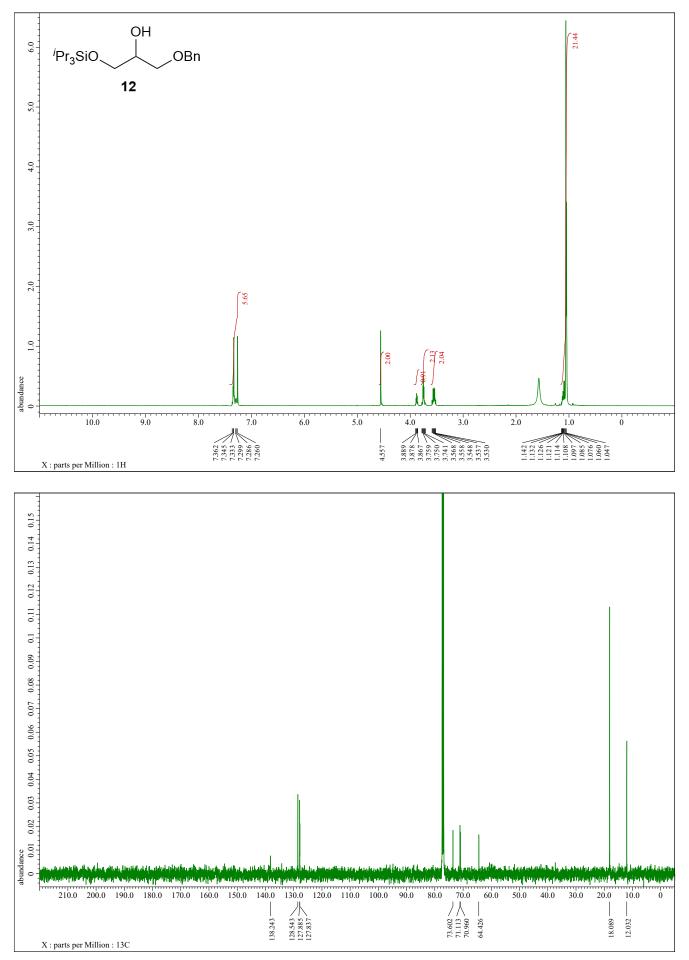




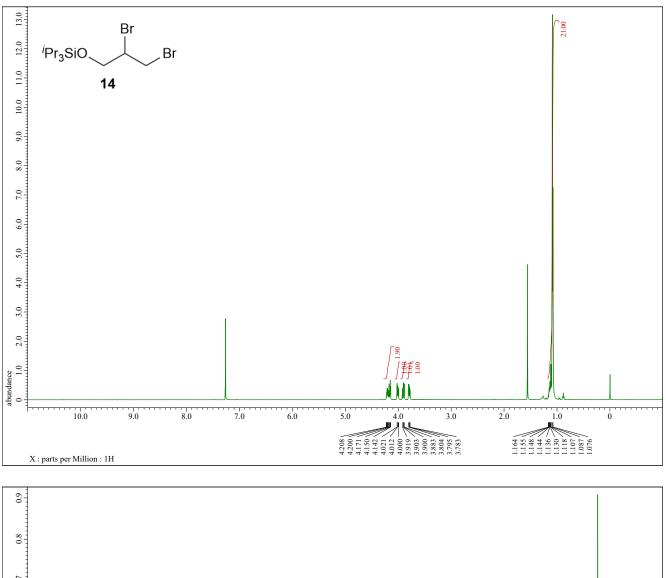


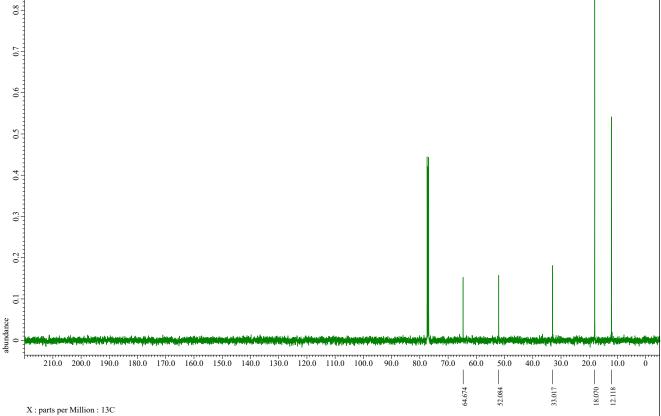
S11

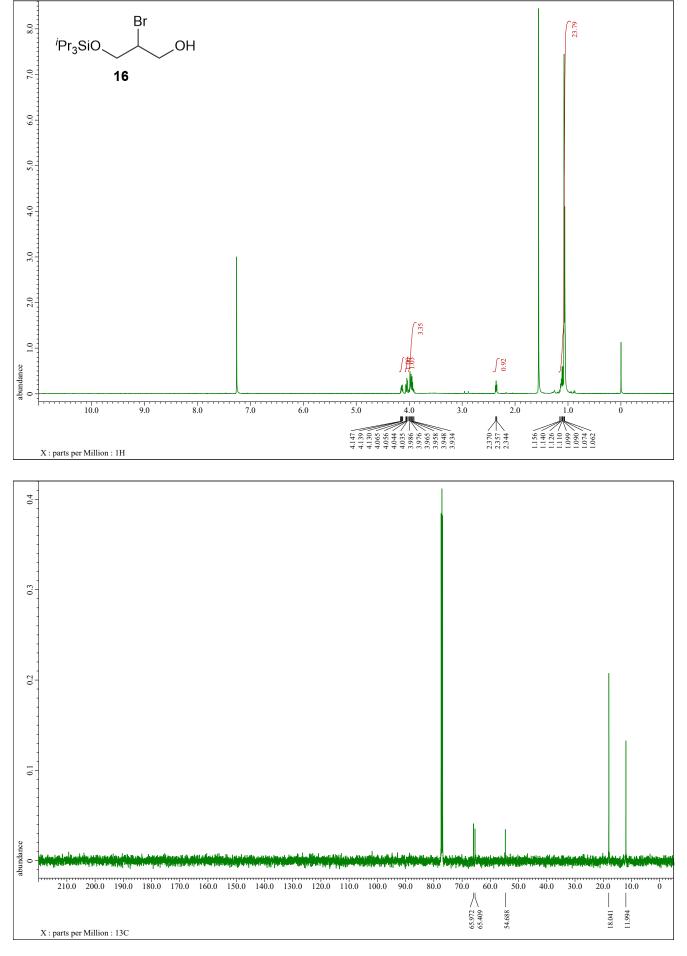


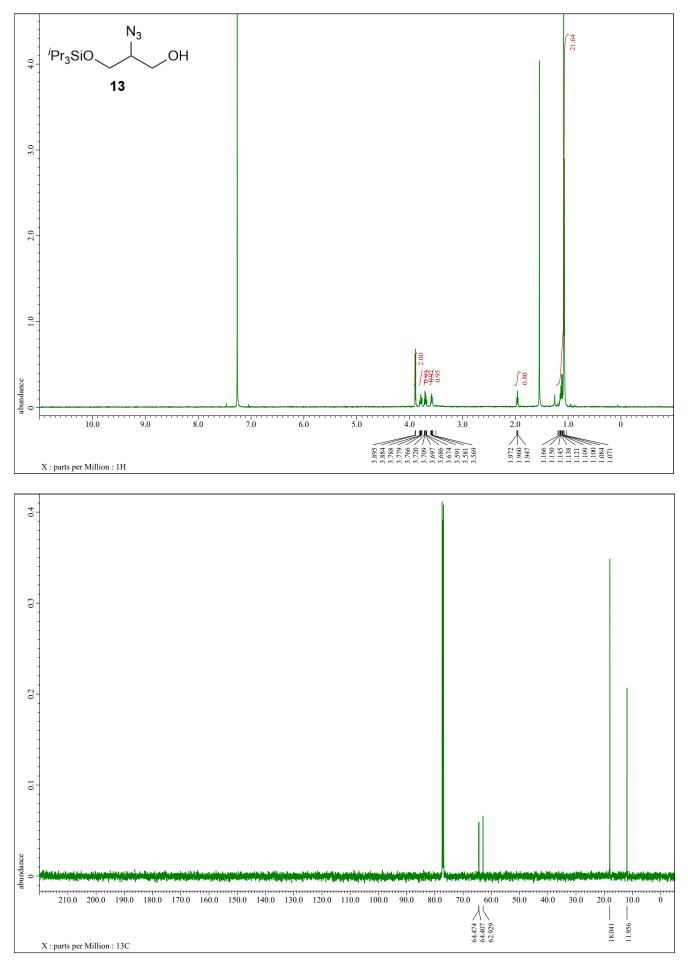


<sup>1</sup>H/<sup>13</sup>C NMR spectra of dibromide 14 (500/126 MHz, CDCl<sub>3</sub>)

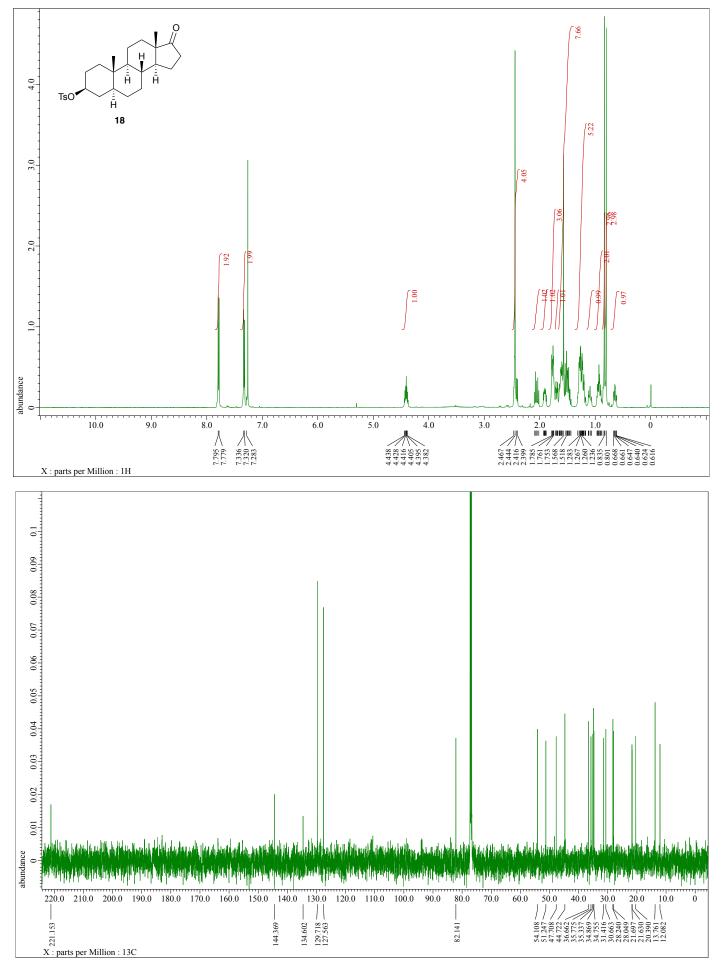








# <sup>1</sup>H/<sup>13</sup>C NMR spectra of tosylate **18** (500/126 MHz, CDCl<sub>3</sub>)



<sup>1</sup>H/<sup>13</sup>C NMR spectra of androsterone **19** (500/126 MHz, CDCl<sub>3</sub>)

