

## Supporting Information

### Synthesis of diazatricycloundecane scaffold via gold(I)-catalysed Conia-ene-type 5-*exo-dig* cyclization and stepwise substituent assembly for construction of sp<sup>3</sup>-rich compound library

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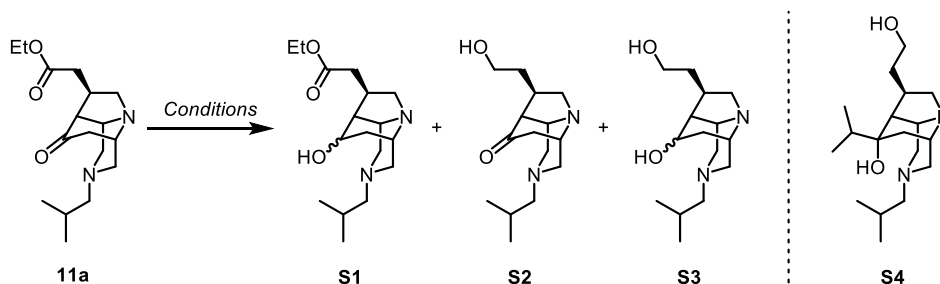
- 1) School of Life Science and Technology, Tokyo Institute of Technology, 4259 Nagatsuta-cho, Midori-ku, Yokohama 226-8503, Japan.
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## 1. Investigation on reduction of ketones

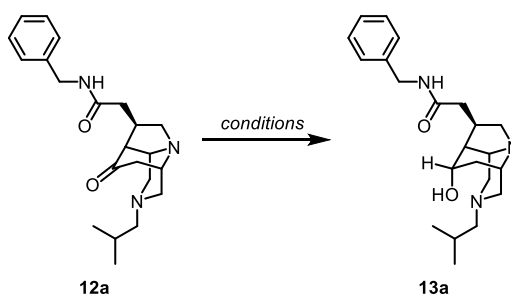
**Table S1.** Investigation of reduction of ketone **11a**



Entry	Reagent	Solvent	Temp.	Time	Yield (%) of S1/S2/S3/S4
1	NaBH <sub>4</sub> (5.7 eq.)	EtOH	rt to 50 °C	28 h	0/31/0/-
2	Red-Al® (2.6 eq.)	Et <sub>2</sub> O	0 °C	6.5 h	0/58/0/-
3	<i>i</i> -PrMgCl (4 eq.)	THF	0 °C to rt	7 h	0/-/-/0
4	LiAlH <sub>4</sub> (7 eq.)	Et <sub>2</sub> O	0 °C to rt	2 h	0/trace/< 93 <sup>a</sup> /-

<sup>a</sup> diol **S3** was obtained as a mixture of unidentified byproducts.

**Table S2.** Ketone-selective reduction of **12a** with various reductants



Entry	Reagent	Solvent	Temp.	Time (h)	Yield (%)
1	LiAlH <sub>4</sub> (7 eq.)	Et <sub>2</sub> O	0 °C	2	8
2	NaBH <sub>4</sub> (5 eq.)	MeOH	0 °C to 40 °C	5	8
	CeCl <sub>3</sub> · 7H <sub>2</sub> O (5 eq.)				
3	LiBH <sub>4</sub> (4 eq.)	Et <sub>2</sub> O	0 °C	1	Not detected
4	NaBH <sub>4</sub> (15 eq.)	EtOH	0 °C to reflux	Overnight	Not detected
5	L-selectride® (4 eq.)	THF	-78 °C to rt	12	No reaction
6	DIBAL-H (6 eq.)	THF	-78 °C to rt	Overnight	Complex mixture
7	Al( <i>Oi</i> -Pr) <sub>3</sub> (10 eq.)	<i>i</i> -PrOH	reflux	10	No reaction
8	SmI <sub>2</sub> (3 eq.)	THF	-78 °C	2	71
	H <sub>2</sub> O (30 eq.)				

## 2. X-ray structural analysis of 13a (cr171127)

### checkCIF/PLATON report

Structure factors have been supplied for datablock(s) cr171127\_auto

THIS REPORT IS FOR GUIDANCE ONLY. IF USED AS PART OF A REVIEW PROCEDURE FOR PUBLICATION, IT SHOULD NOT REPLACE THE EXPERTISE OF AN EXPERIENCED CRYSTALLOGRAPHIC REFEREE.

No syntax errors found.    CIF dictionary    Interpreting this report

### Datablock: cr171127\_auto

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Bond precision:    C-C = 0.0022 A                      Wavelength=1.54184  
Cell:                      a=16.4271 (3)              b=9.6289 (2)              c=12.6774 (2)  
                                    alpha=90                      beta=96.421 (2)              gamma=90  
Temperature:              90 K

	Calculated	Reported
Volume	1992.67 (6)	1992.67 (6)
Space group	P 21/c	P 1 21/c 1
Hall group	-P 2ybc	-P 2ybc
Moiety formula	C22 H33 N3 O2	C22 H33 N3 O2
Sum formula	C22 H33 N3 O2	C22 H33 N3 O2
Mr	371.51	371.51
Dx, g cm <sup>-3</sup>	1.238	1.238
Z	4	4
Mu (mm <sup>-1</sup> )	0.628	0.628
F000	808.0	808.0
F000'	810.21	
h, k, lmax	20, 12, 15	20, 11, 15
Nref	4176	4044
Tmin, Tmax	0.919, 0.992	0.807, 1.000
Tmin'	0.919	

Correction method= # Reported T Limits: Tmin=0.807 Tmax=1.000

AbsCorr = MULTI-SCAN

Data completeness= 0.968                      Theta(max) = 76.633

R(reflections)= 0.0497 ( 3610)

wR2(reflections)=  
0.1213 ( 4044)

S = 1.069

Npar= 247

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The following ALERTS were generated. Each ALERT has the format

**test-name\_ALERT\_alert-type\_alert-level.**

Click on the hyperlinks for more details of the test.

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● **Alert level C**

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The relevant atom site should be identified.  
PLAT094\_ALERT\_2\_C Ratio of Maximum / Minimum Residual Density .... 3.01 Report  
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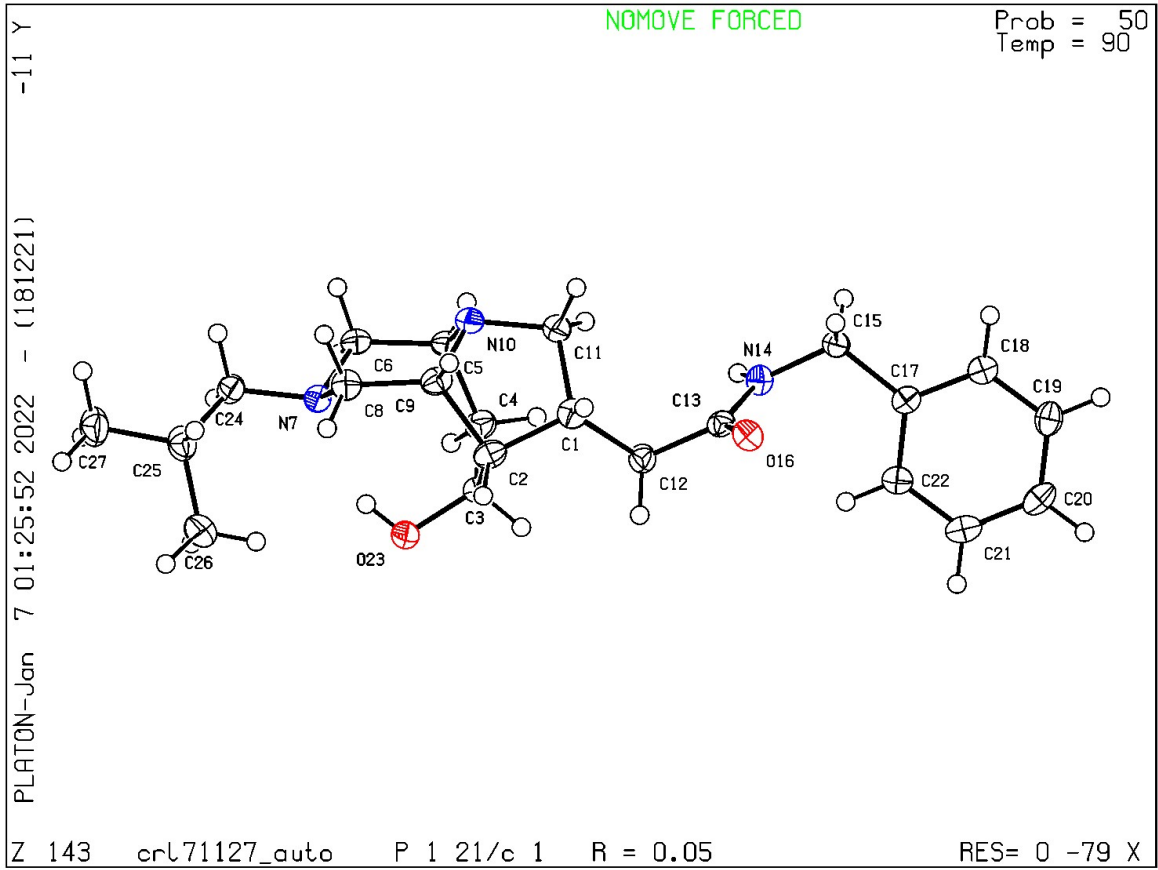
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PLAT978\_ALERT\_2\_G Number C-C Bonds with Positive Residual Density. 12 Info

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0 **ALERT level A** = Most likely a serious problem - resolve or explain  
0 **ALERT level B** = A potentially serious problem, consider carefully  
4 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight  
8 **ALERT level G** = General information/check it is not something unexpected

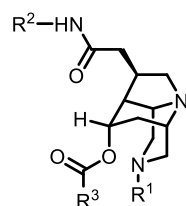
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3 ALERT type 2 Indicator that the structure model may be wrong or deficient  
1 ALERT type 3 Indicator that the structure quality may be low  
6 ALERT type 4 Improvement, methodology, query or suggestion  
1 ALERT type 5 Informative message, check

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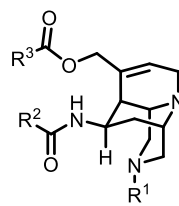
### 3. PMI analysis of diazatricyclic compounds

#### (a) structure of tricyclic compounds



**6a-h**

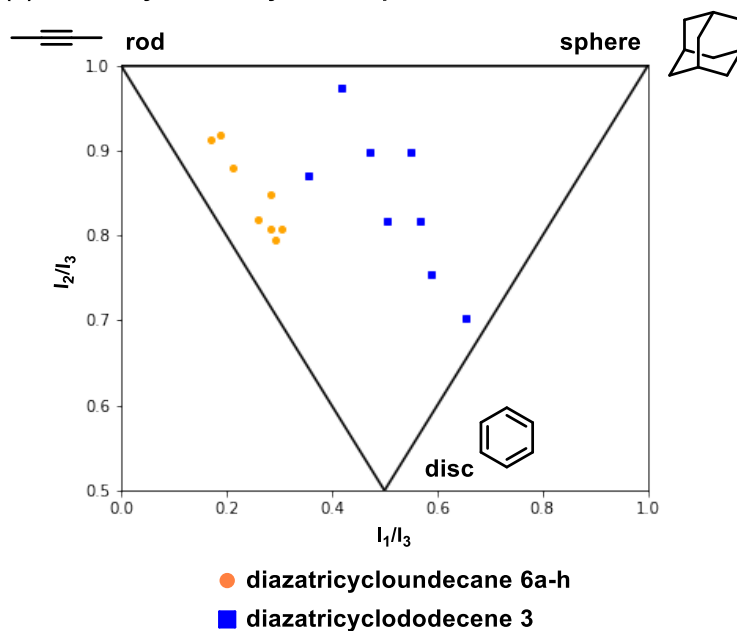
diazatricycloundecane  
 $R^1, R^2, R^3 = \text{Bn or } i\text{-Bu}$   
(This work)



**3**

diazatricyclododecene  
 $R^1, R^2, R^3 = \text{Bn or } i\text{-Bu}$   
(Previous work)

#### (b) PMI analysis of tricyclic compounds



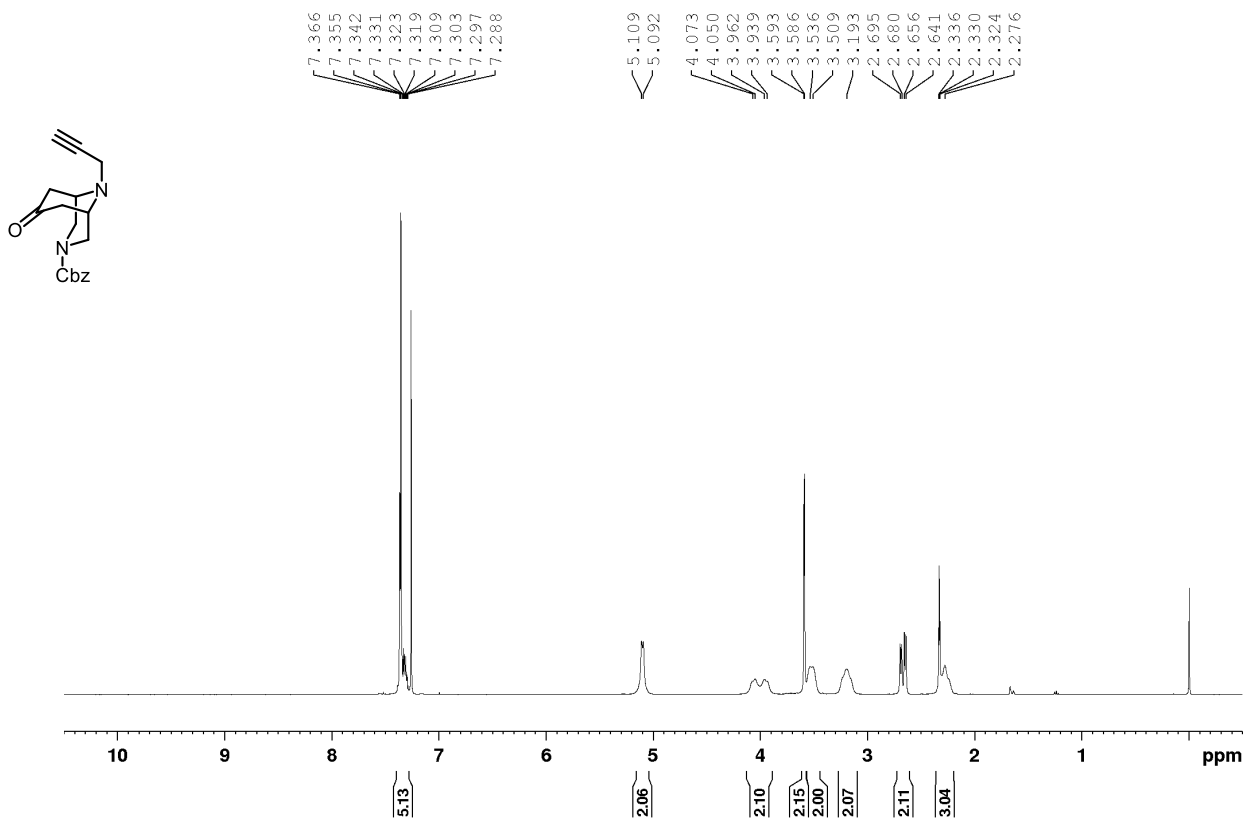
**Figure S1.** PMI analysis of compounds with diazatricyclic scaffolds. (a) chemical structures used for PMI analysis. (b) PMI plot<sup>[1]</sup> of diazatricycloundecanes **6a-h** (orange circle) and diazatricyclododecenes **3** with three isobutyl or benzyl groups (blue square). Conformers with the lowest energy generated by iCon<sup>[2]</sup> implemented in LigandScout 4.4.5 were used for the analysis. PMI = principal moment of inertia.

#### 4. References

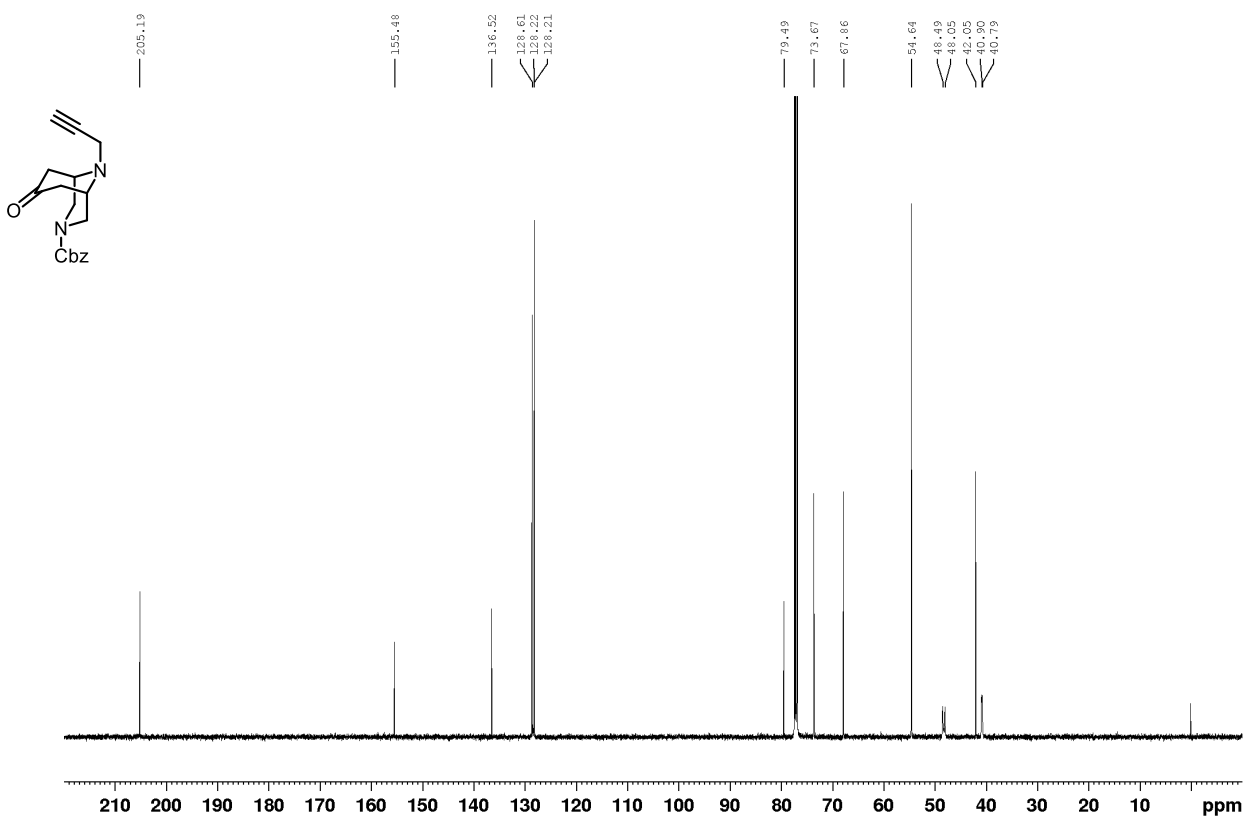
1. W. H. B. Sauer and M. K. Schwarz, *J. Chem. Inf. Comput. Sci.*, 2003, **43**, 987–1003.
2. G. Poli, T. Seidel and T. Langer, *Front Chem.*, 2018, **6**, 229.

## 5. NMR spectra

### Ketone 8 ( $^1\text{H}$ NMR, $\text{CDCl}_3$ , 400 MHz)

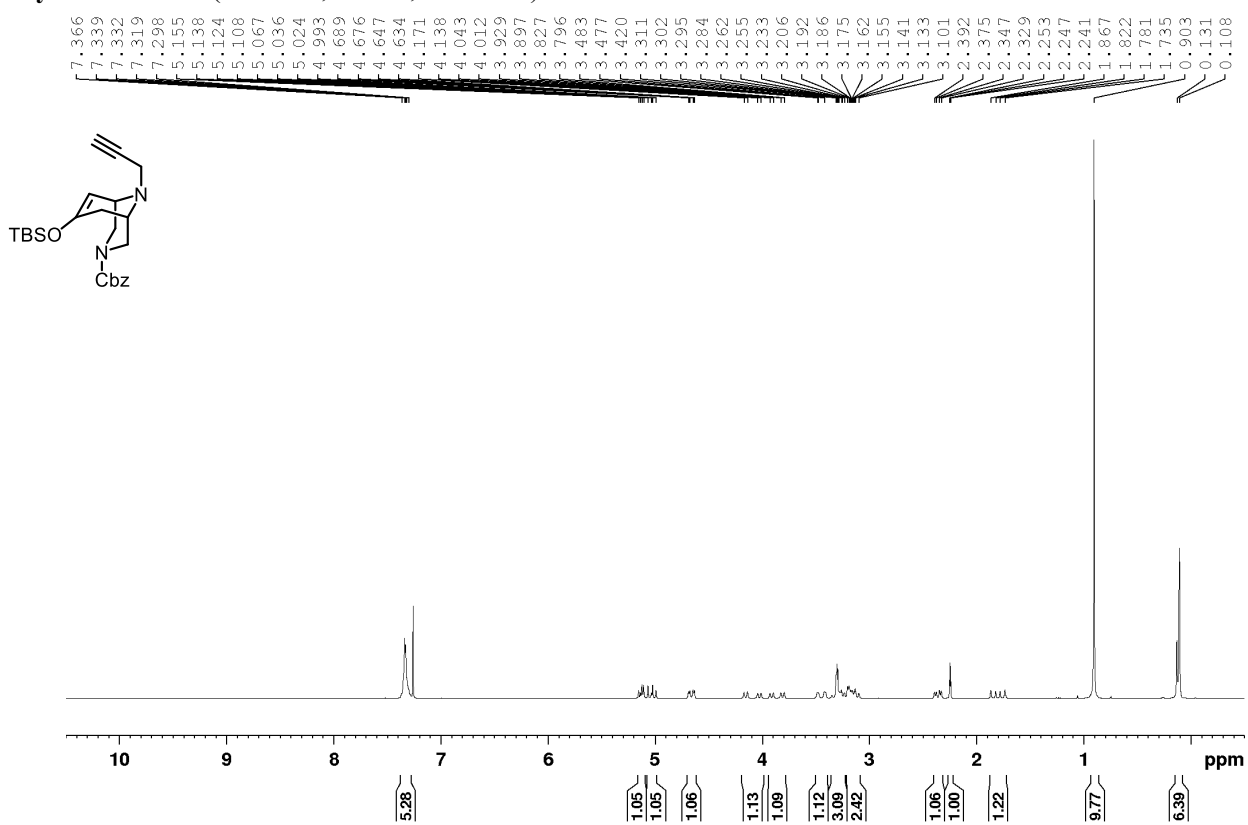


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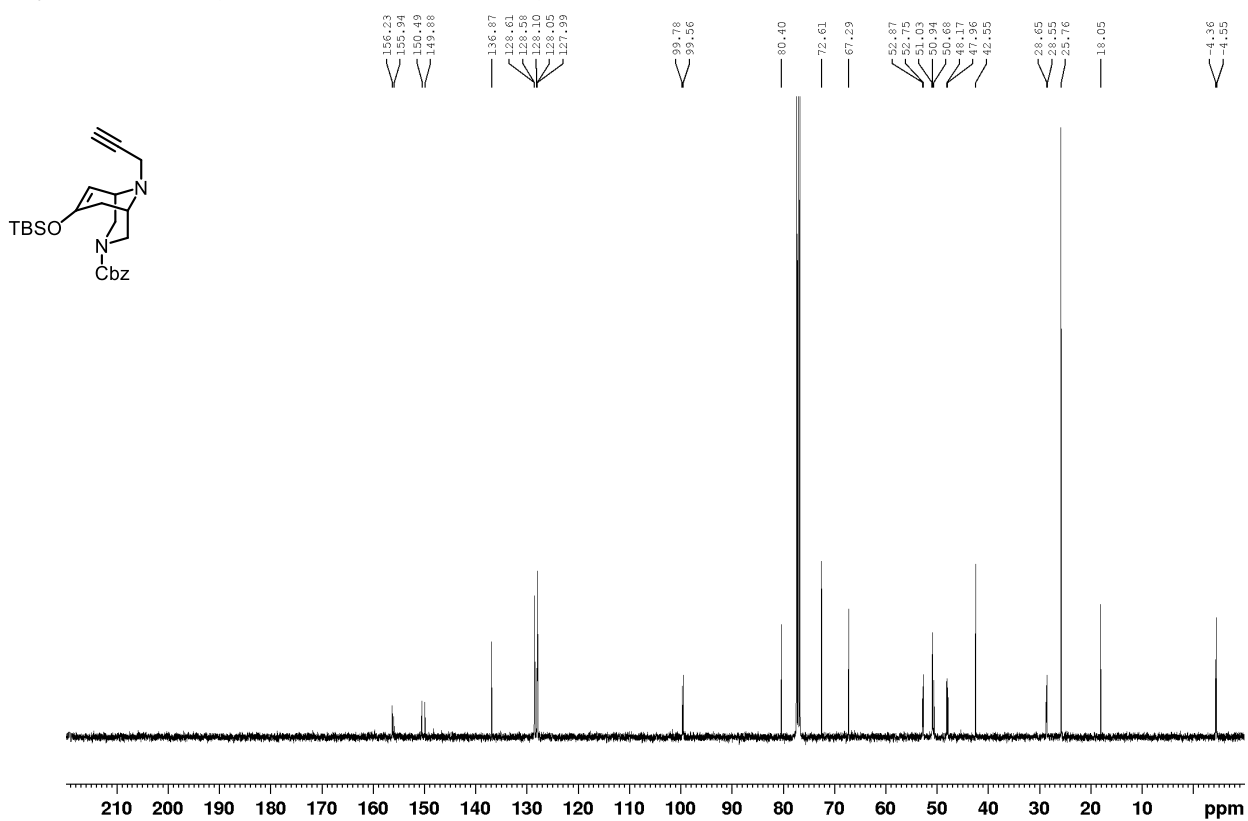




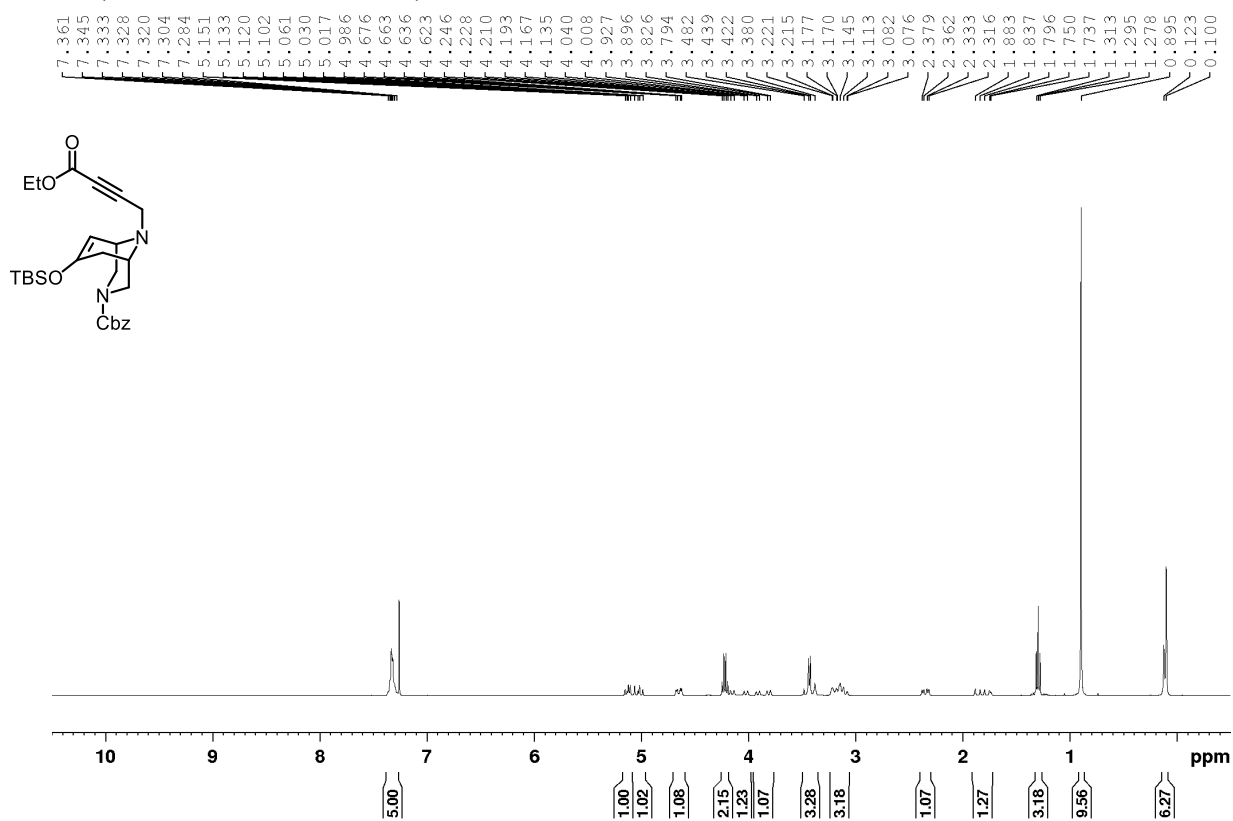
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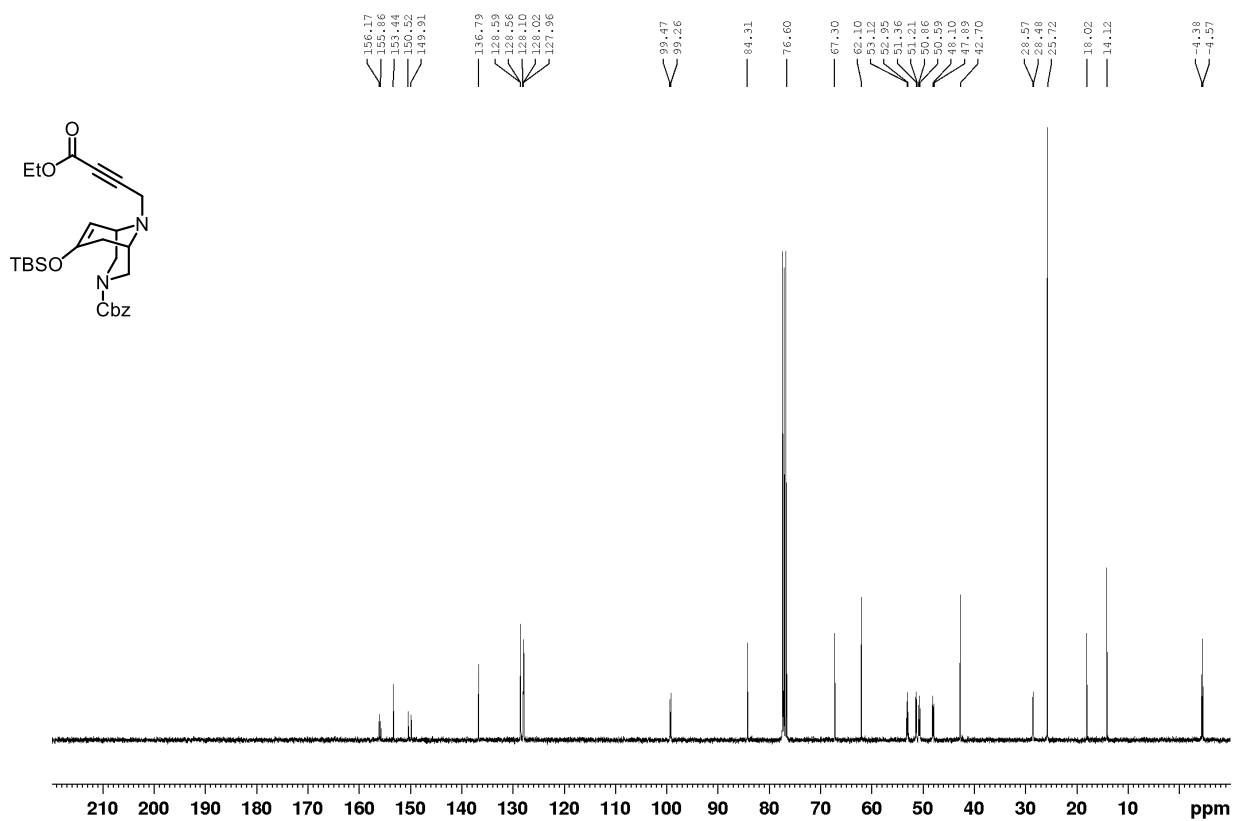
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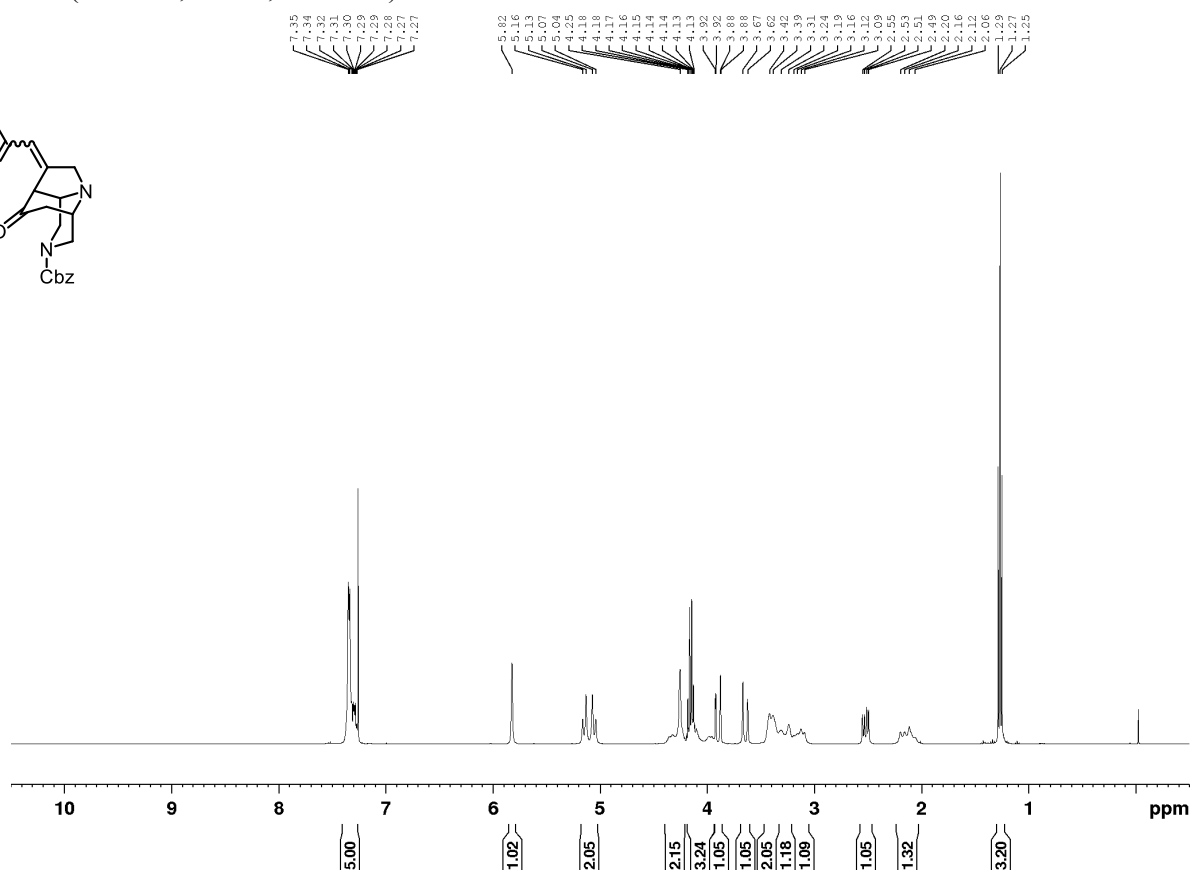
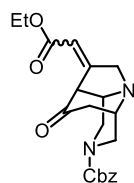
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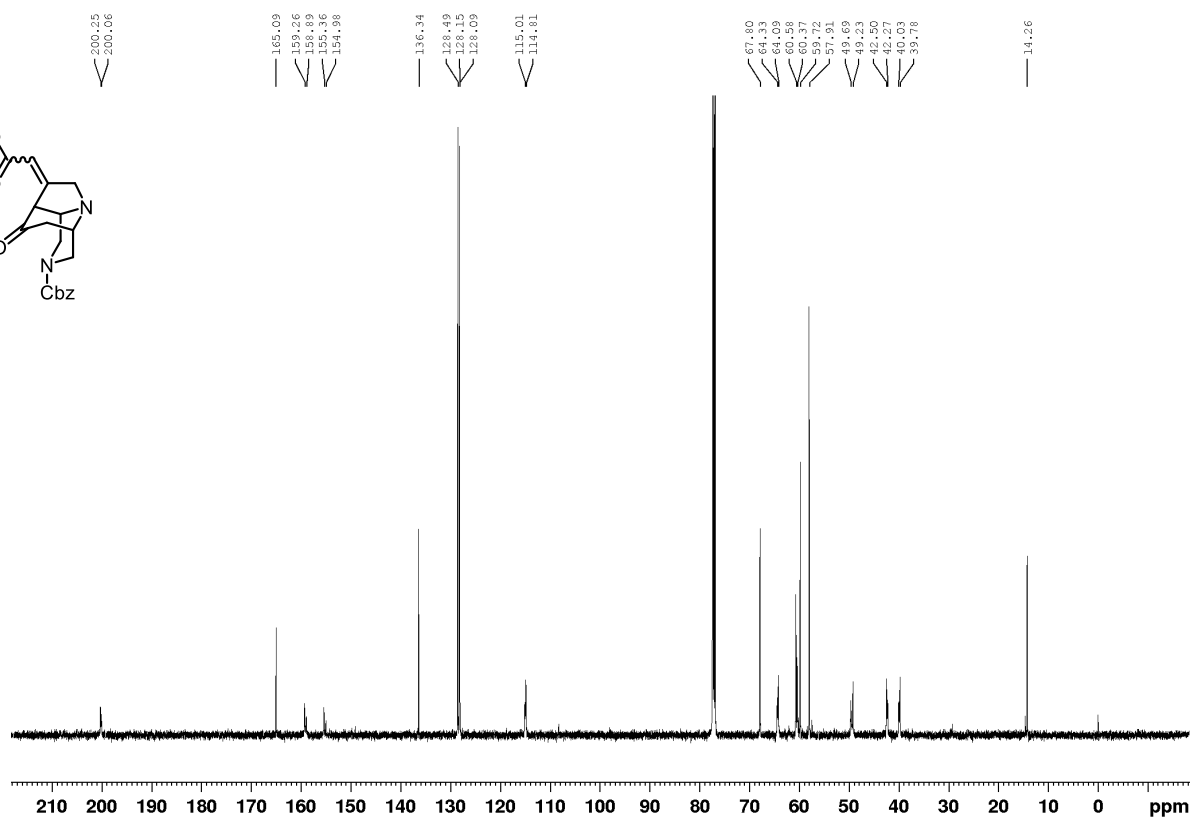
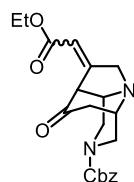
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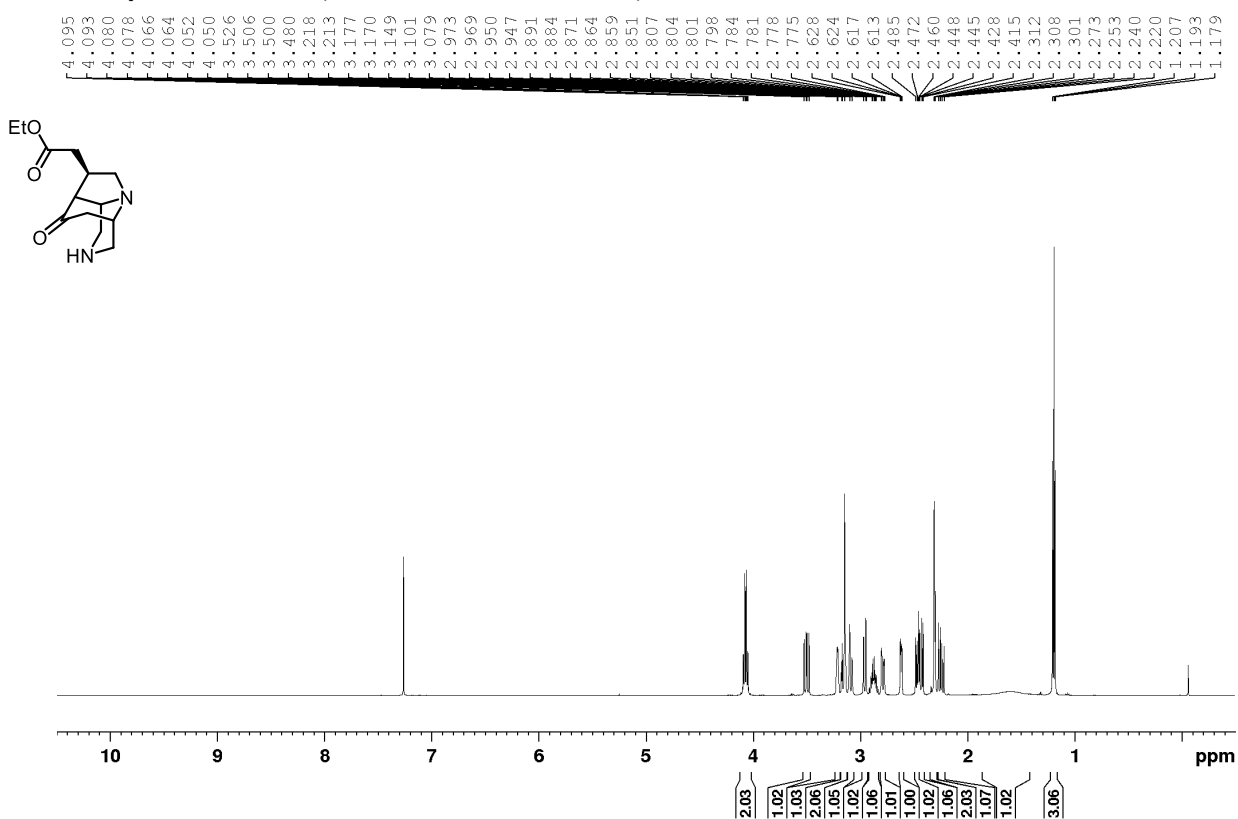
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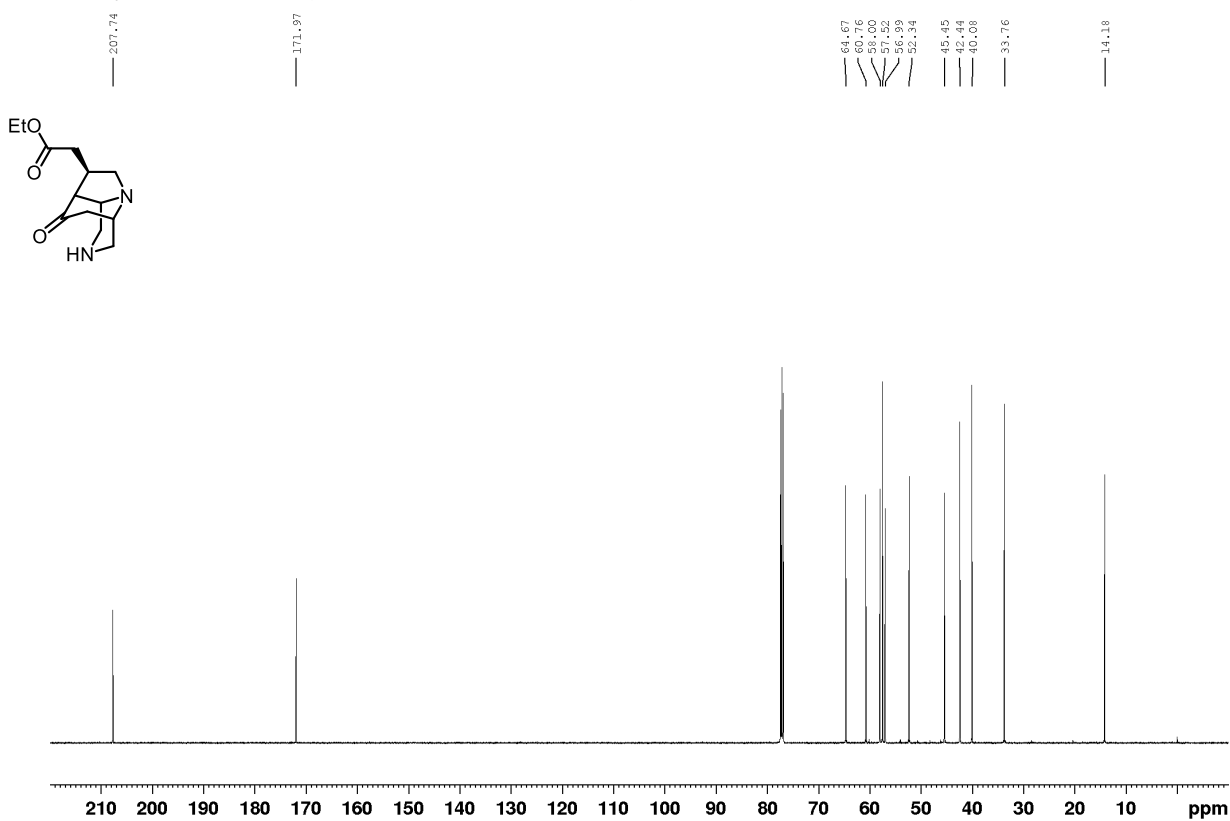
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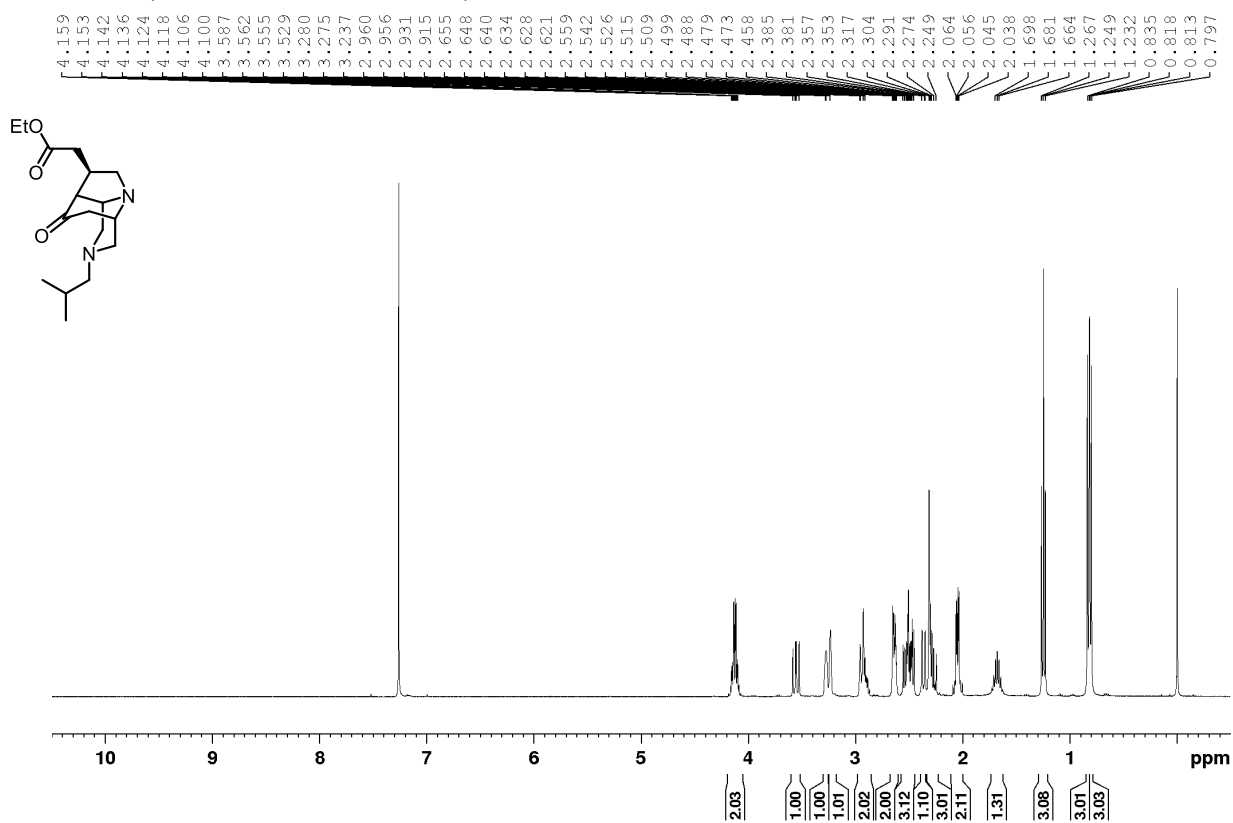
Diazatricycloundecane 5 (<sup>1</sup>H NMR, CDCl<sub>3</sub>, 500 MHz)



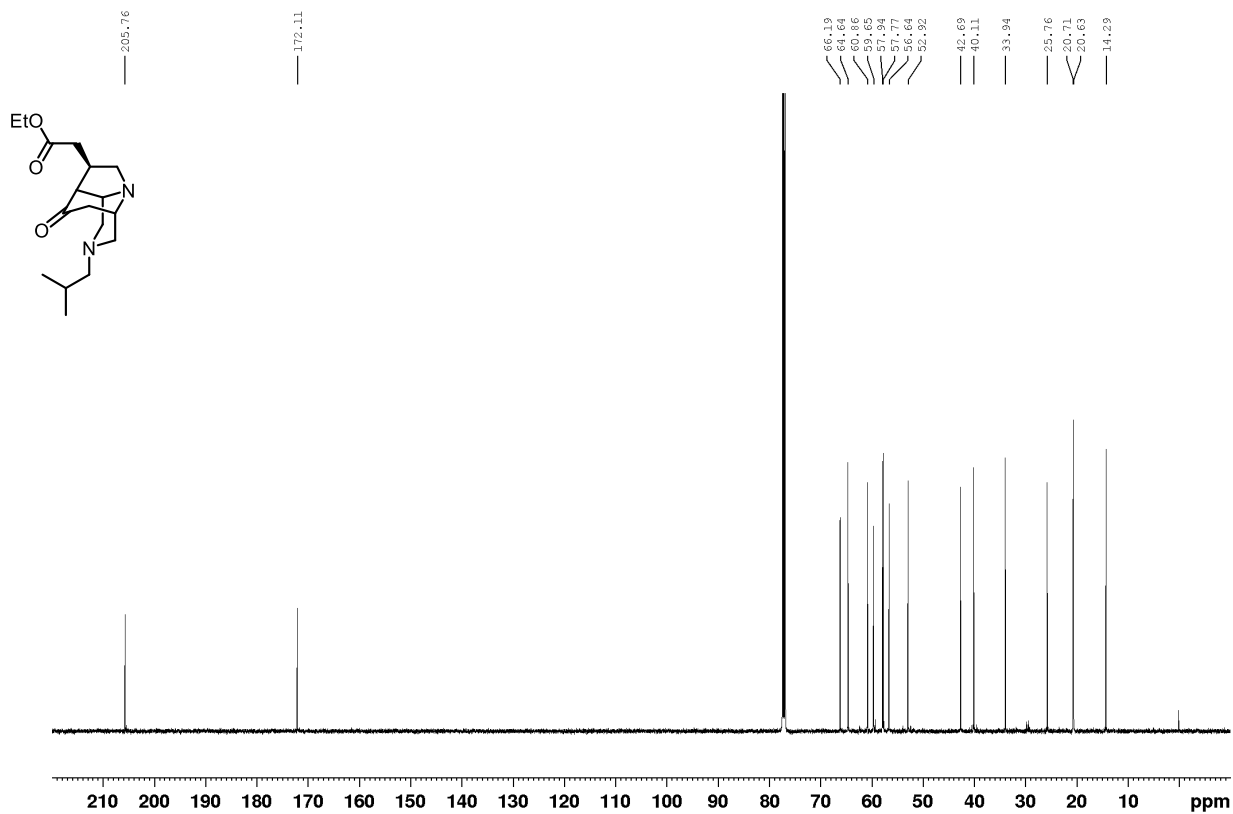
Diazatricycloundecane 5 (<sup>13</sup>C NMR, CDCl<sub>3</sub>, 125 MHz)



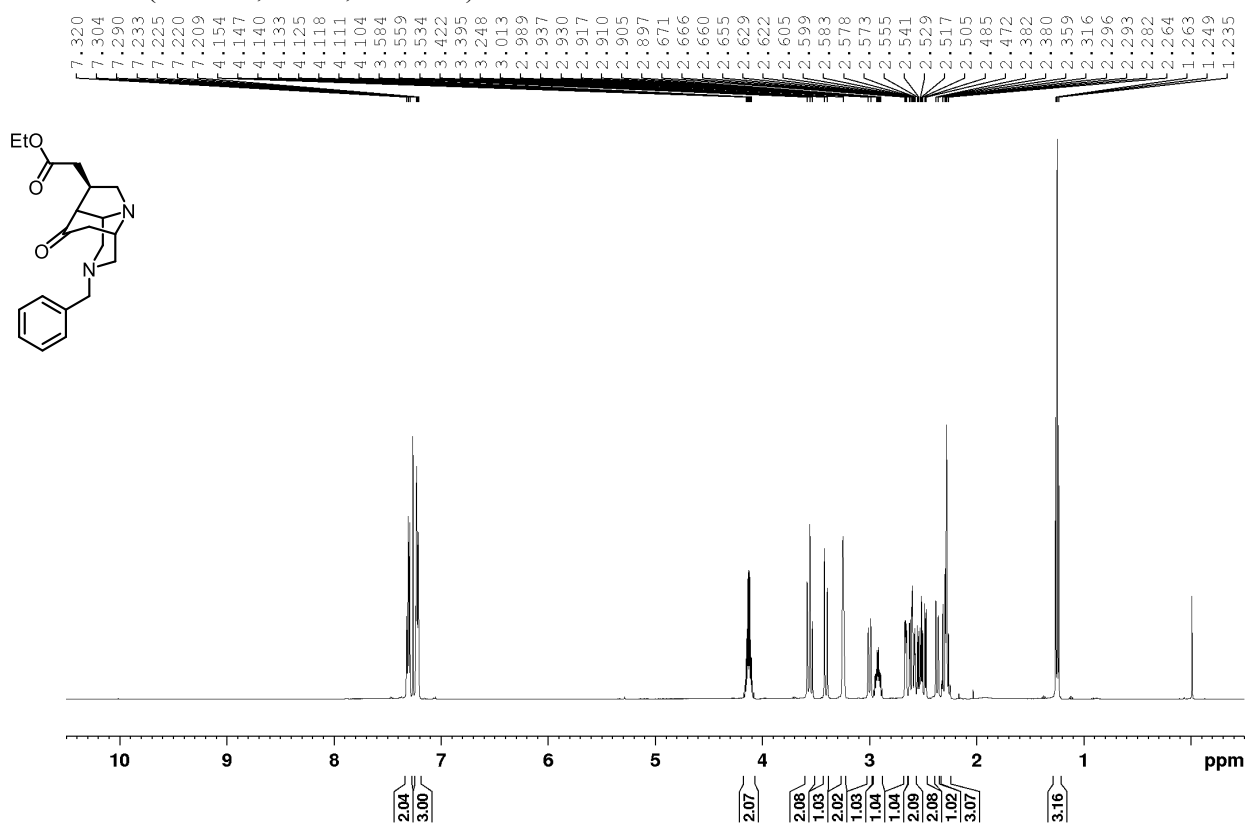
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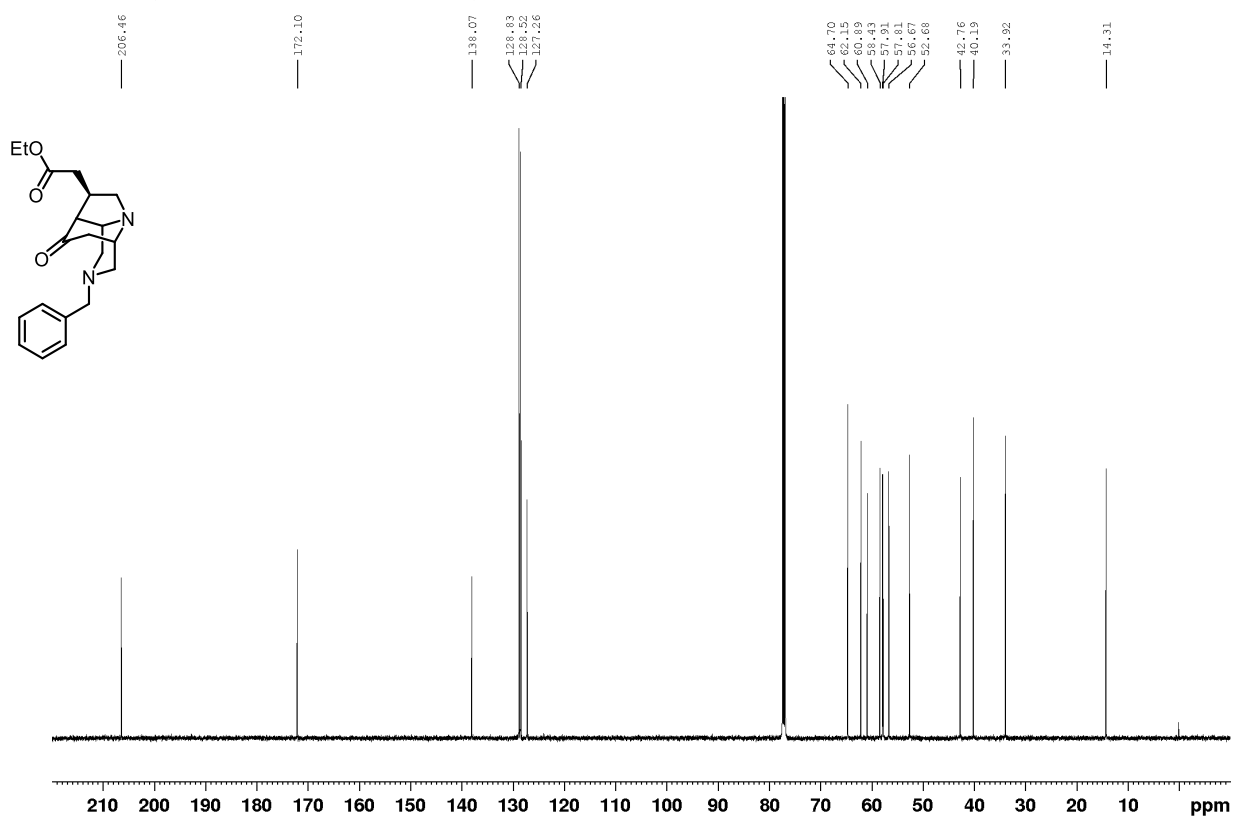
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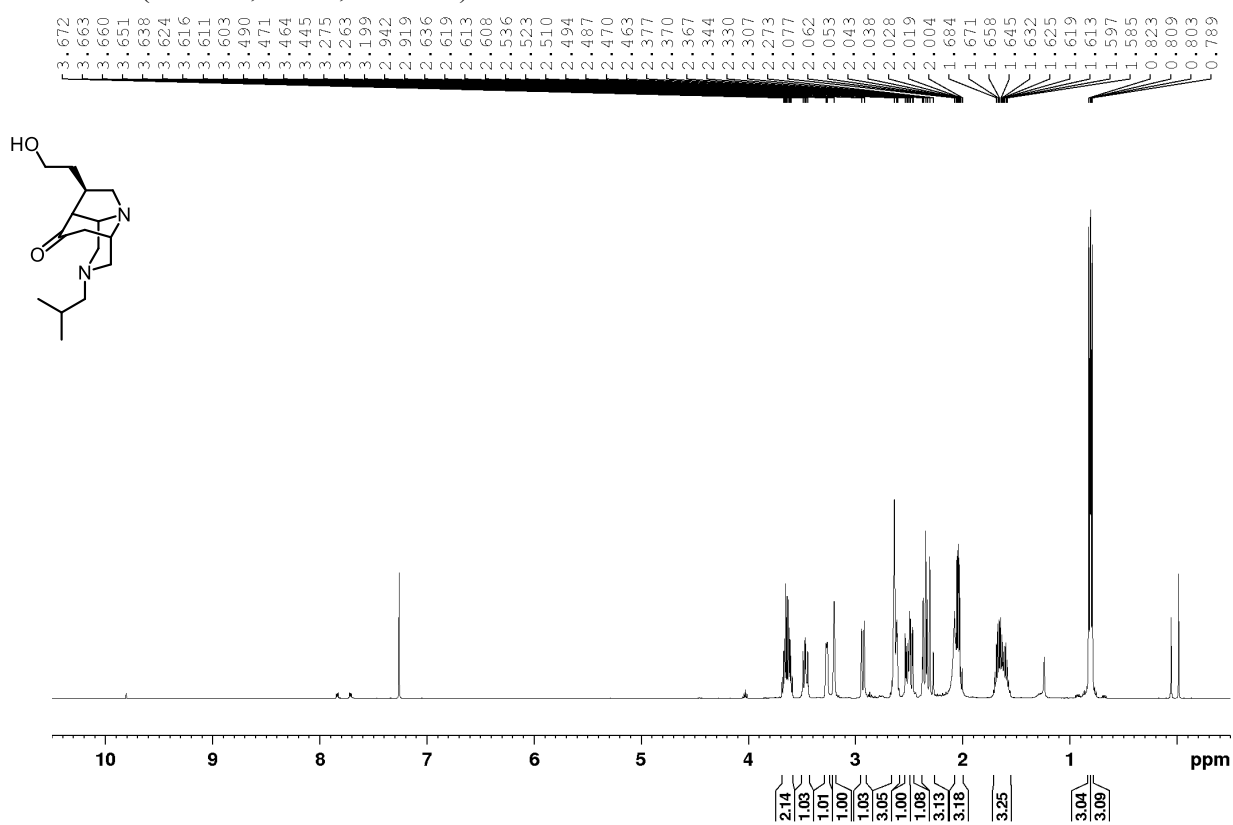
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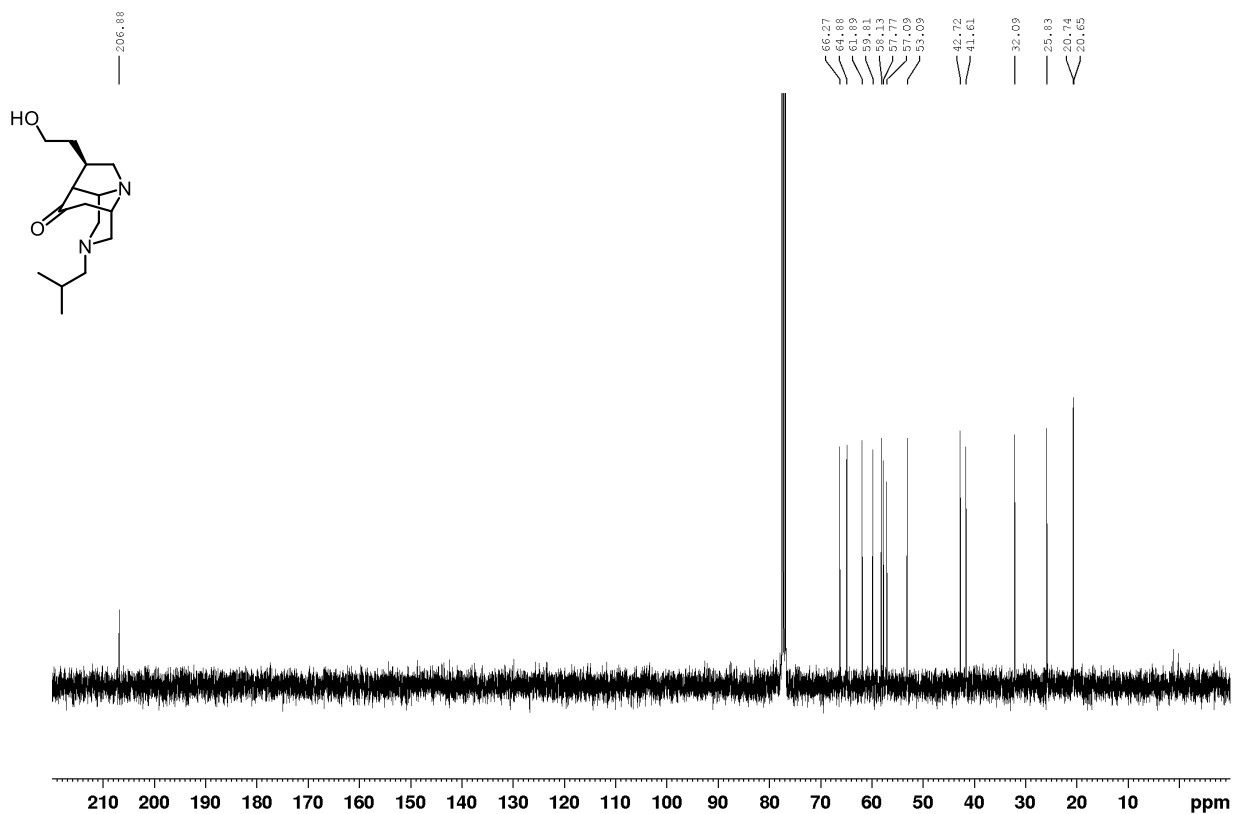
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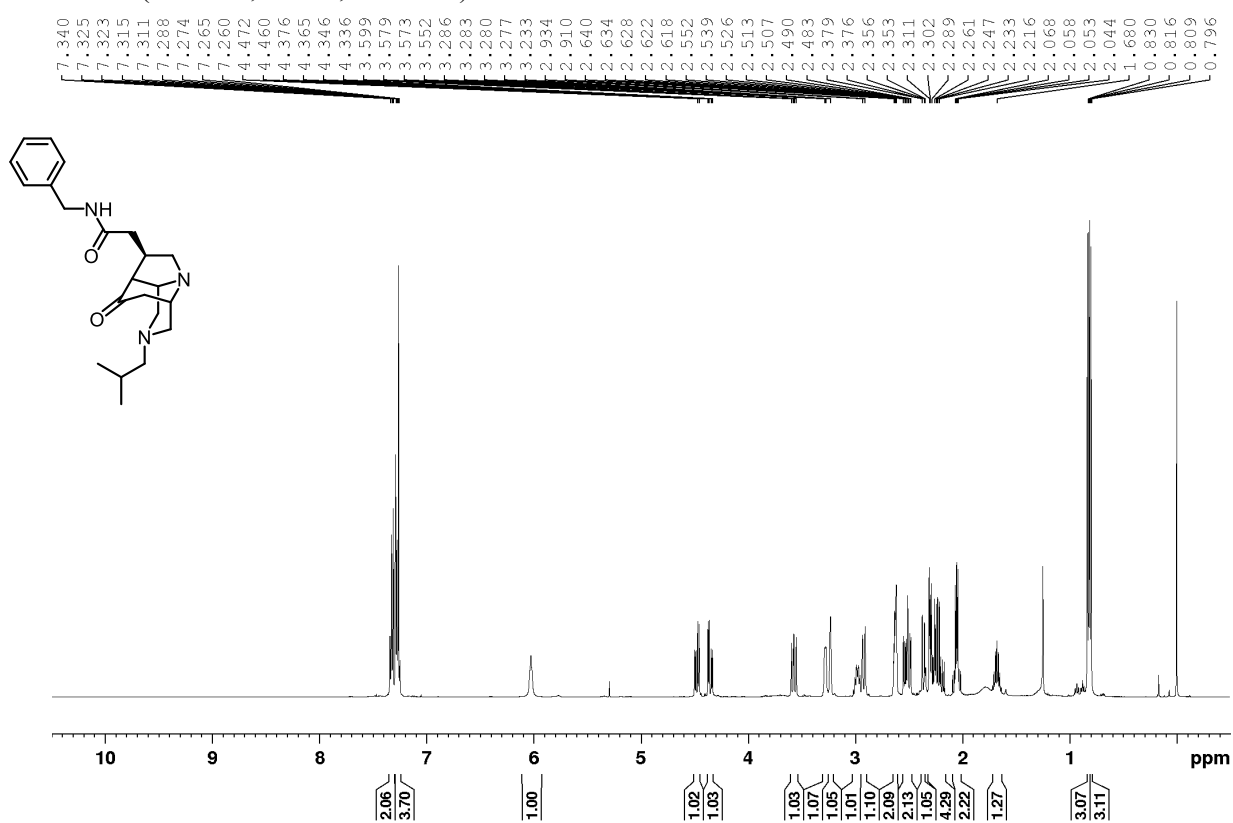
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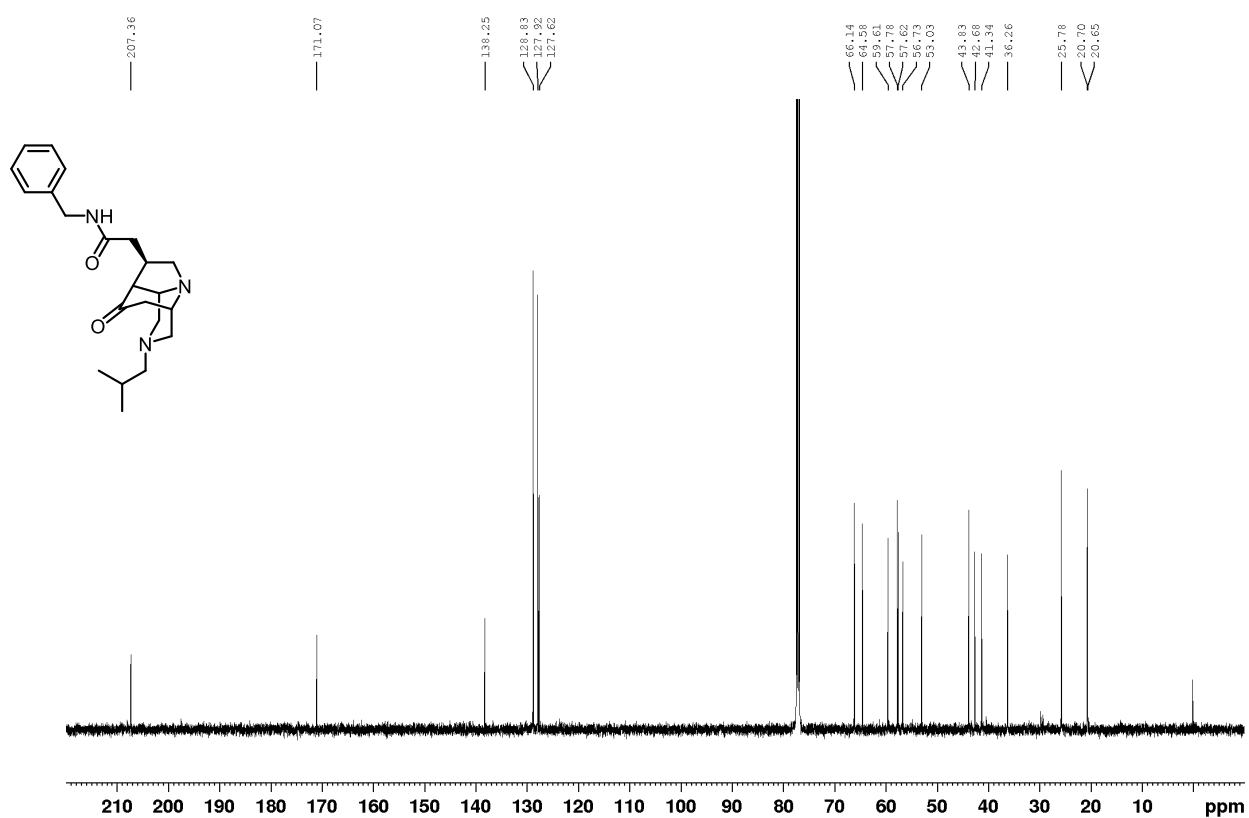
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Amide 12a (<sup>1</sup>H NMR, CDCl<sub>3</sub>, 500 MHz)

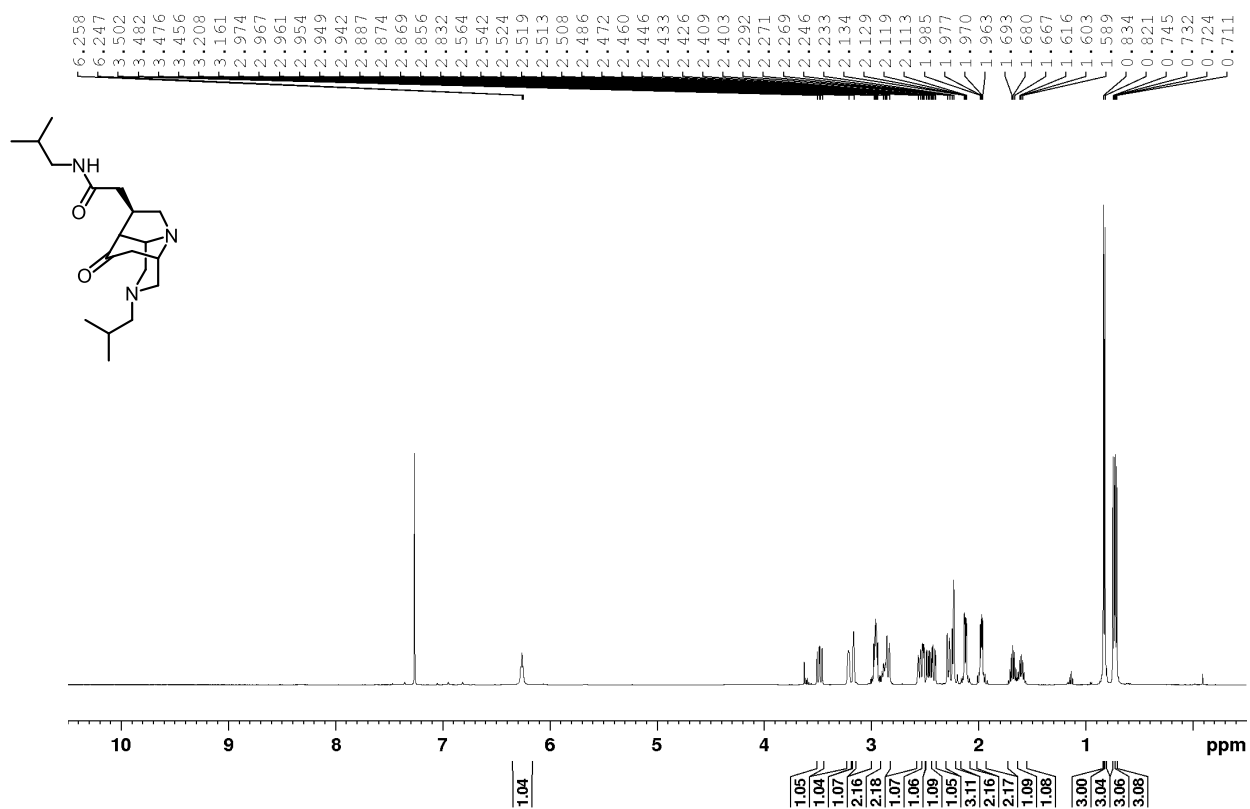


Amide 12a (<sup>13</sup>C NMR, CDCl<sub>3</sub>, 125 MHz)

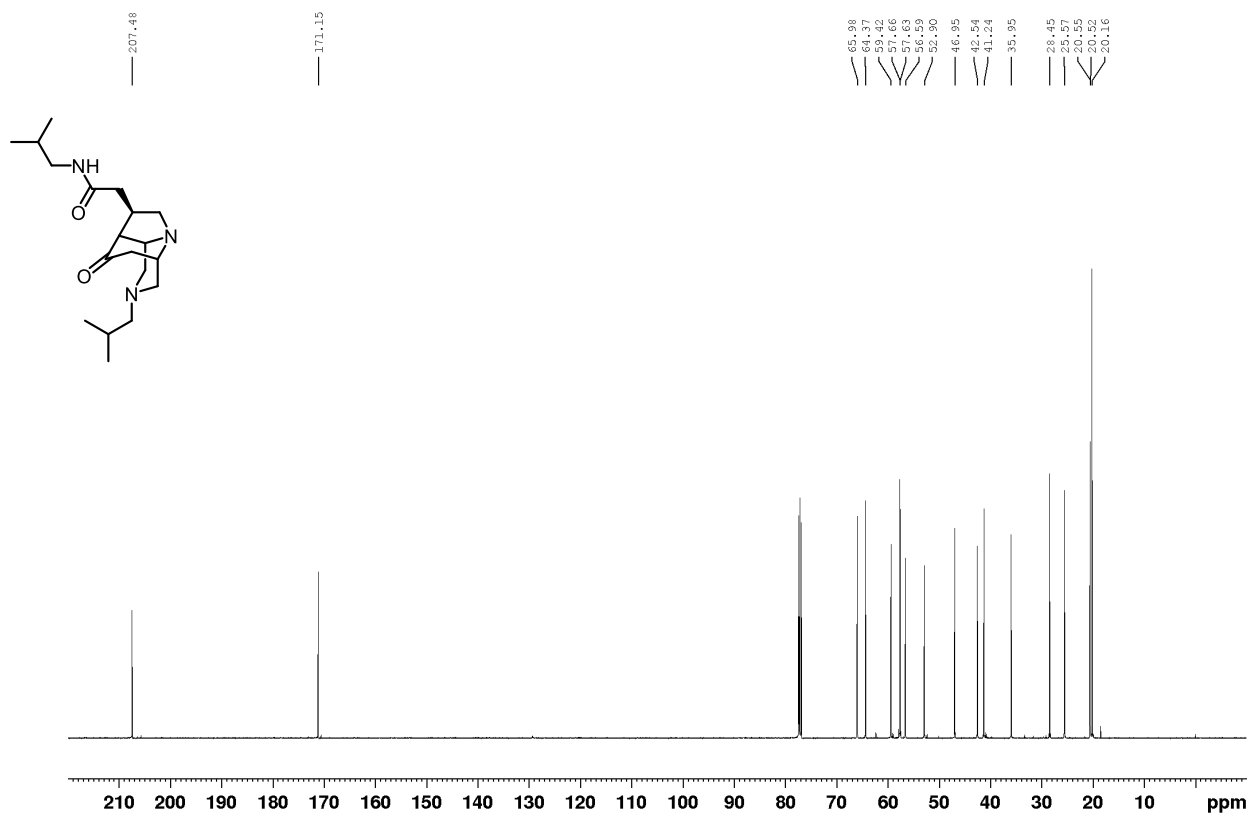




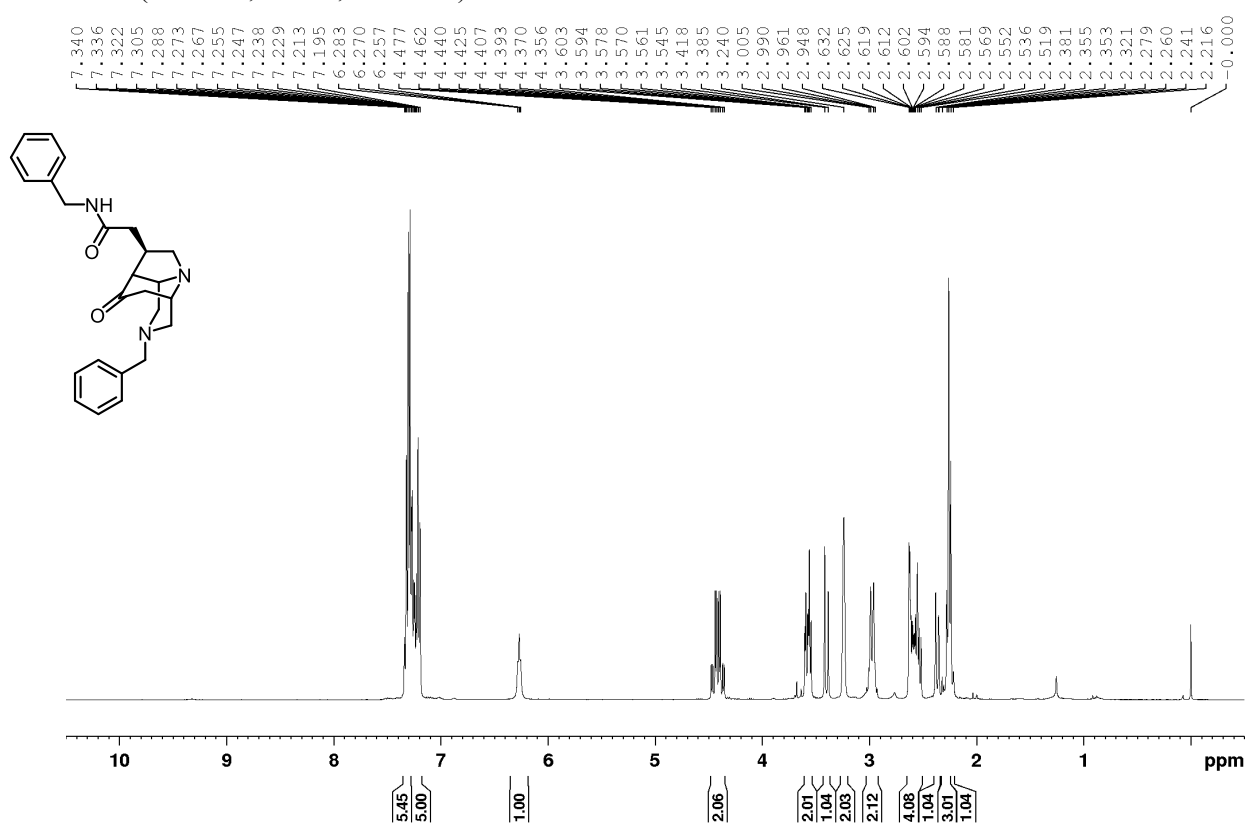
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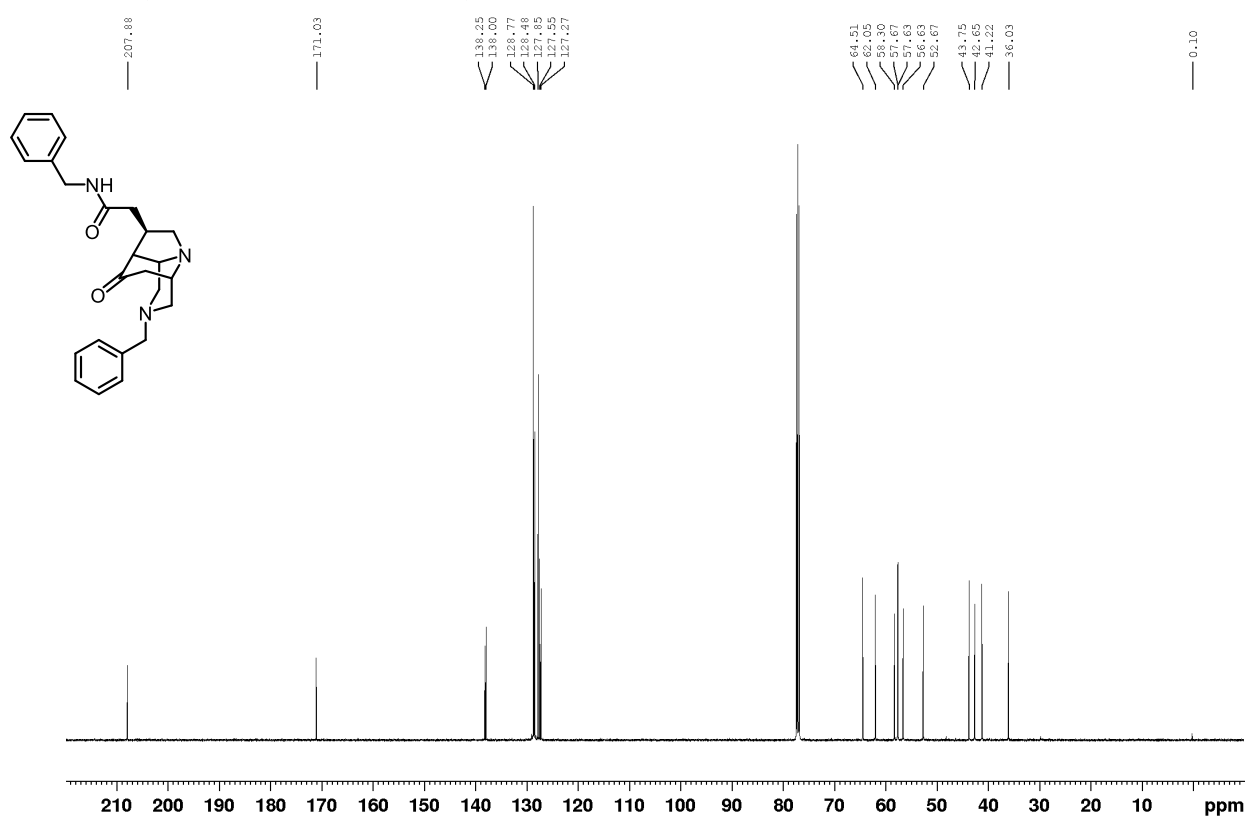
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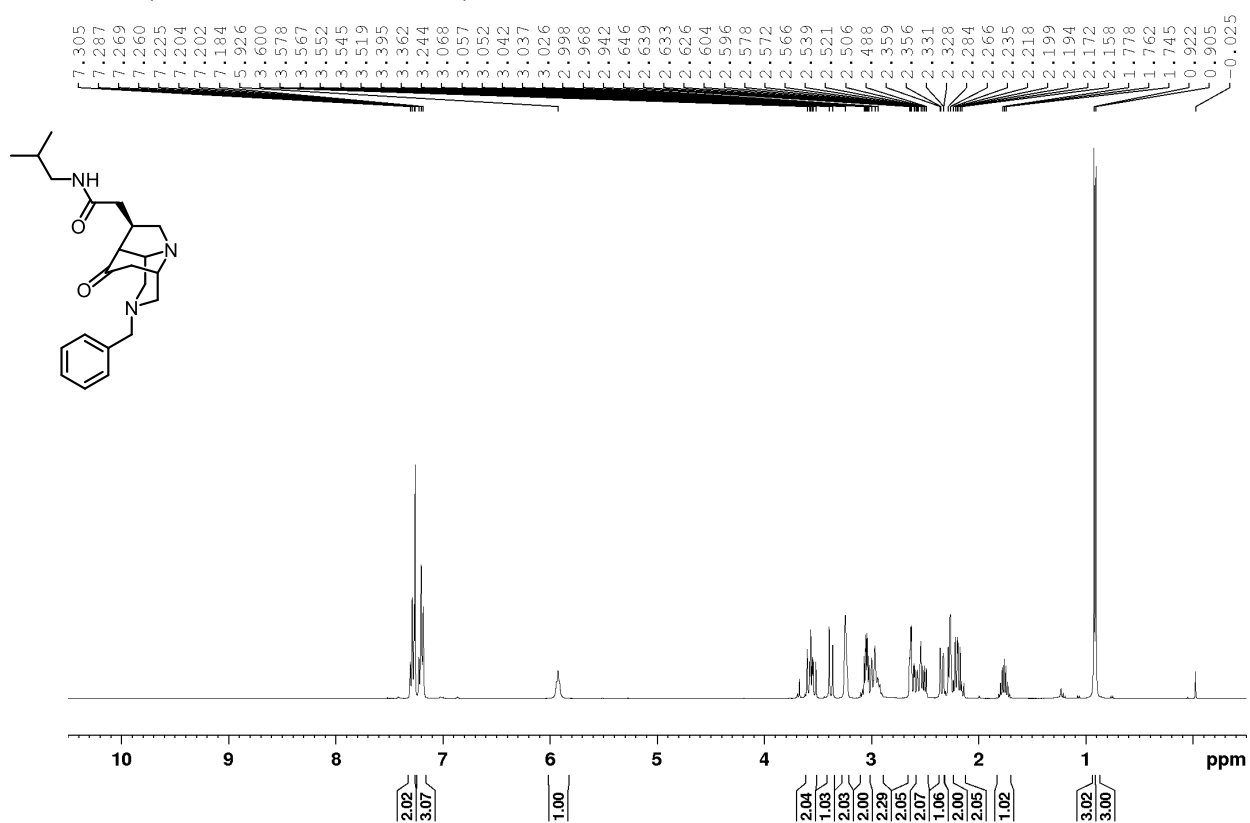
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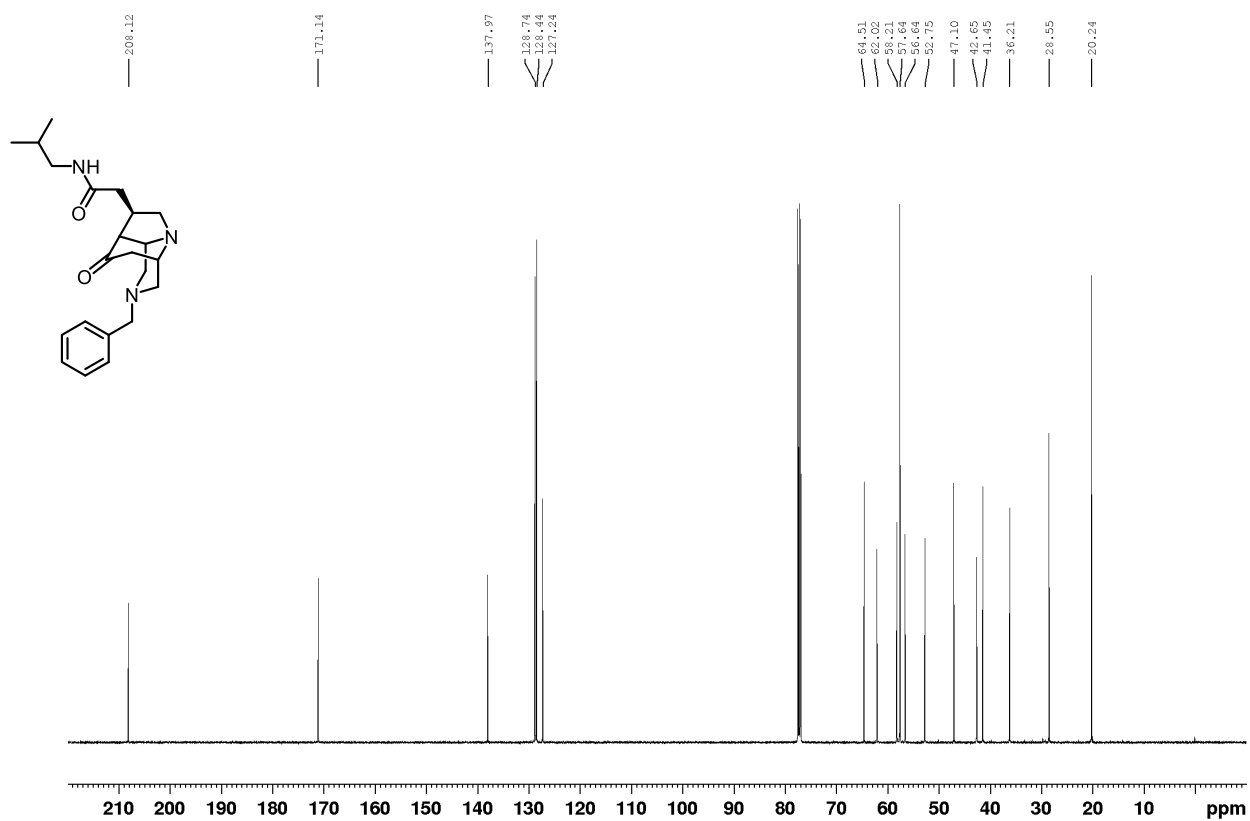
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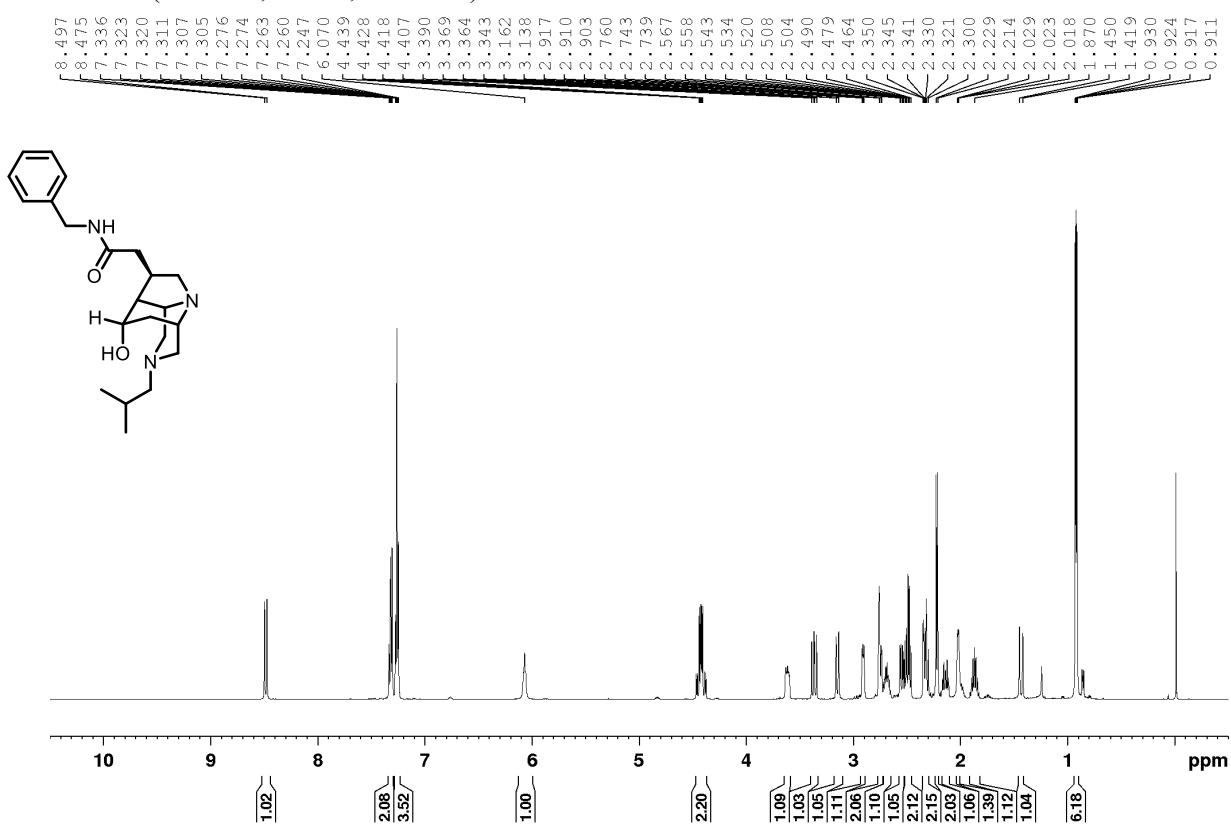
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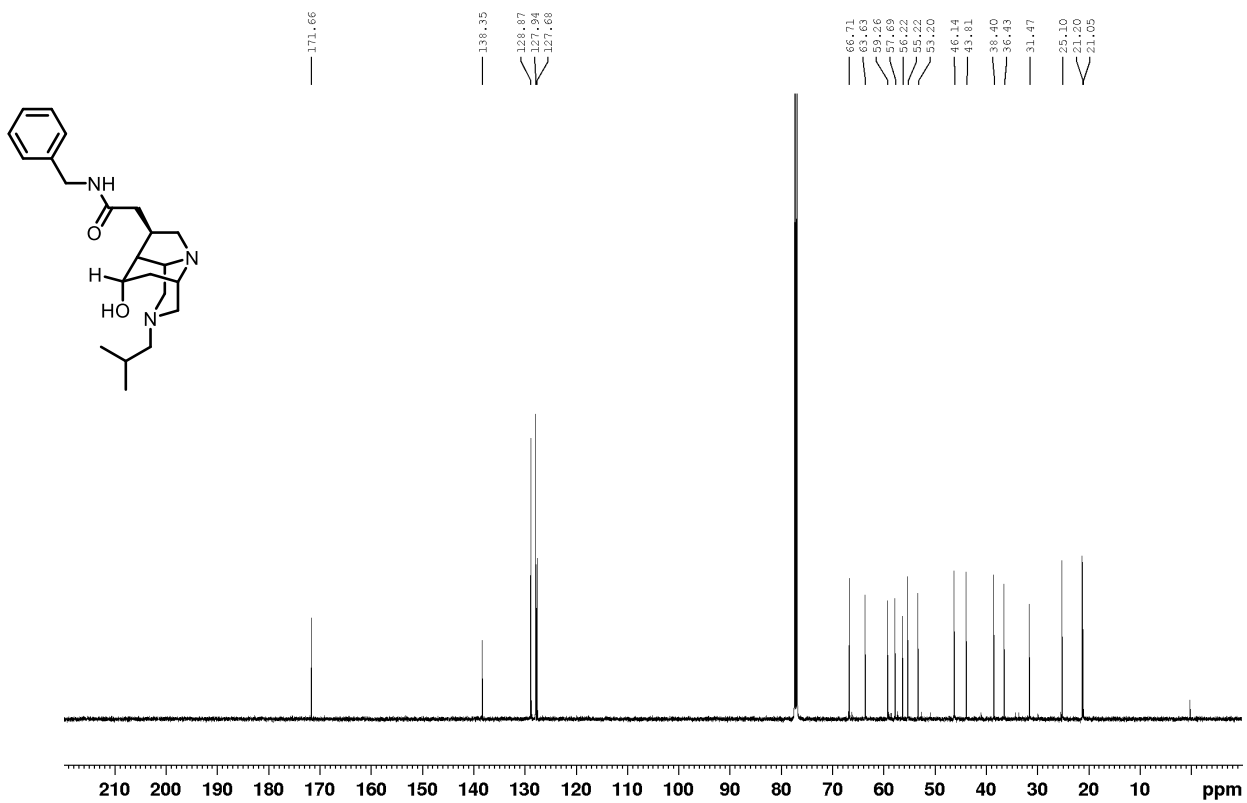
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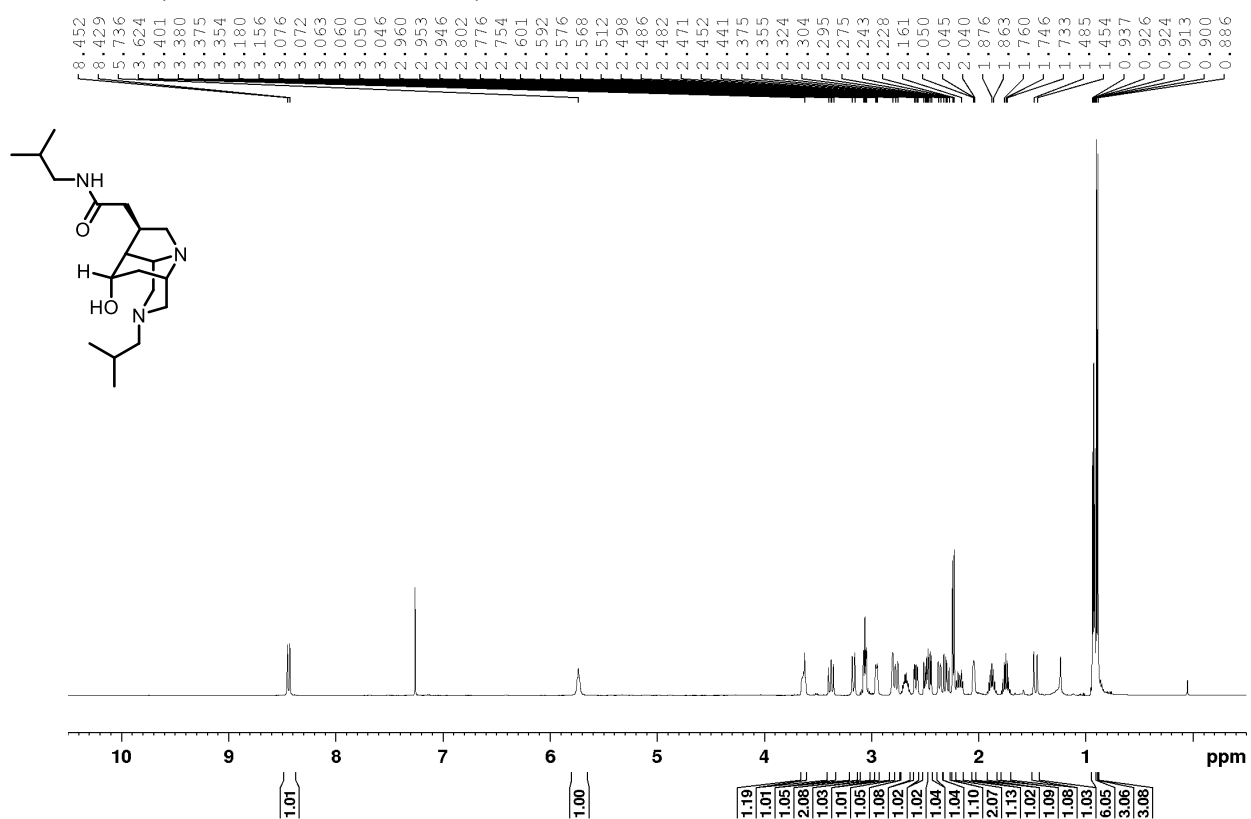
**Alcohol 13a** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 500 MHz)



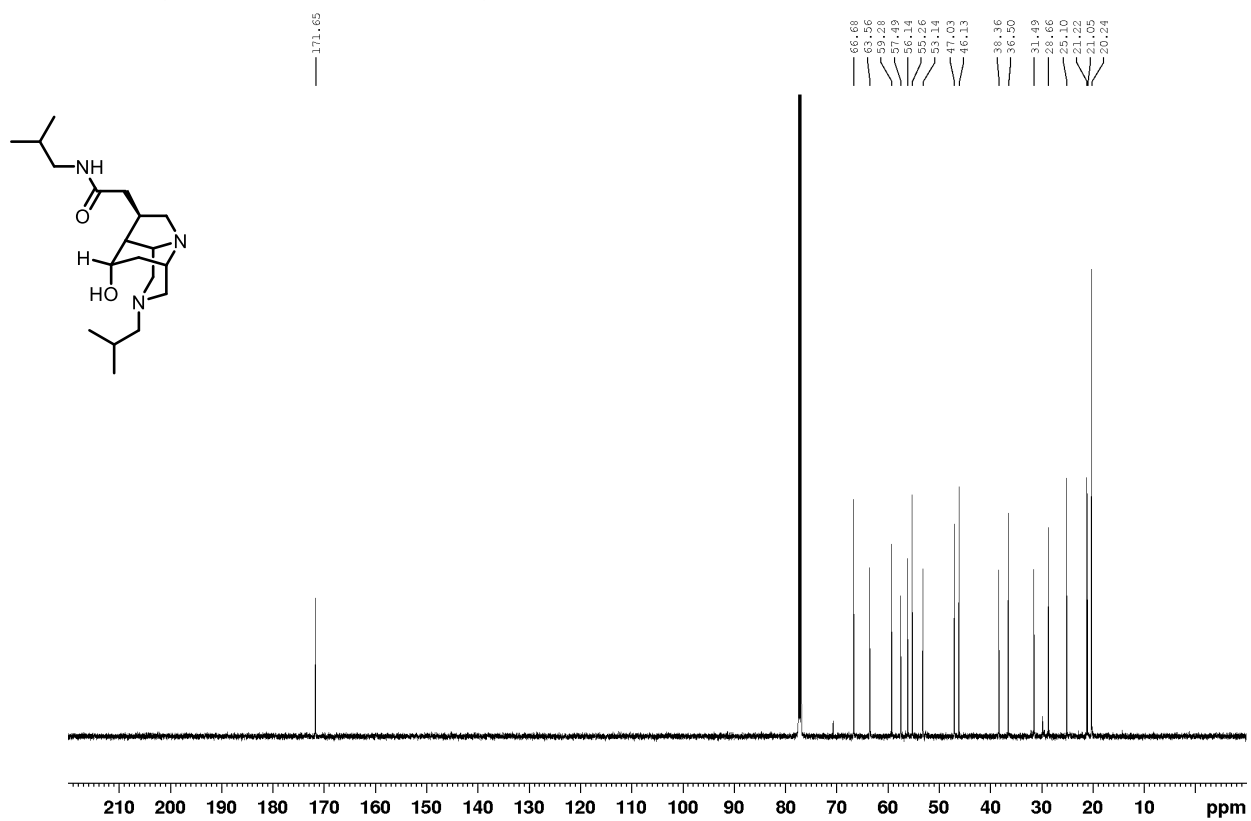
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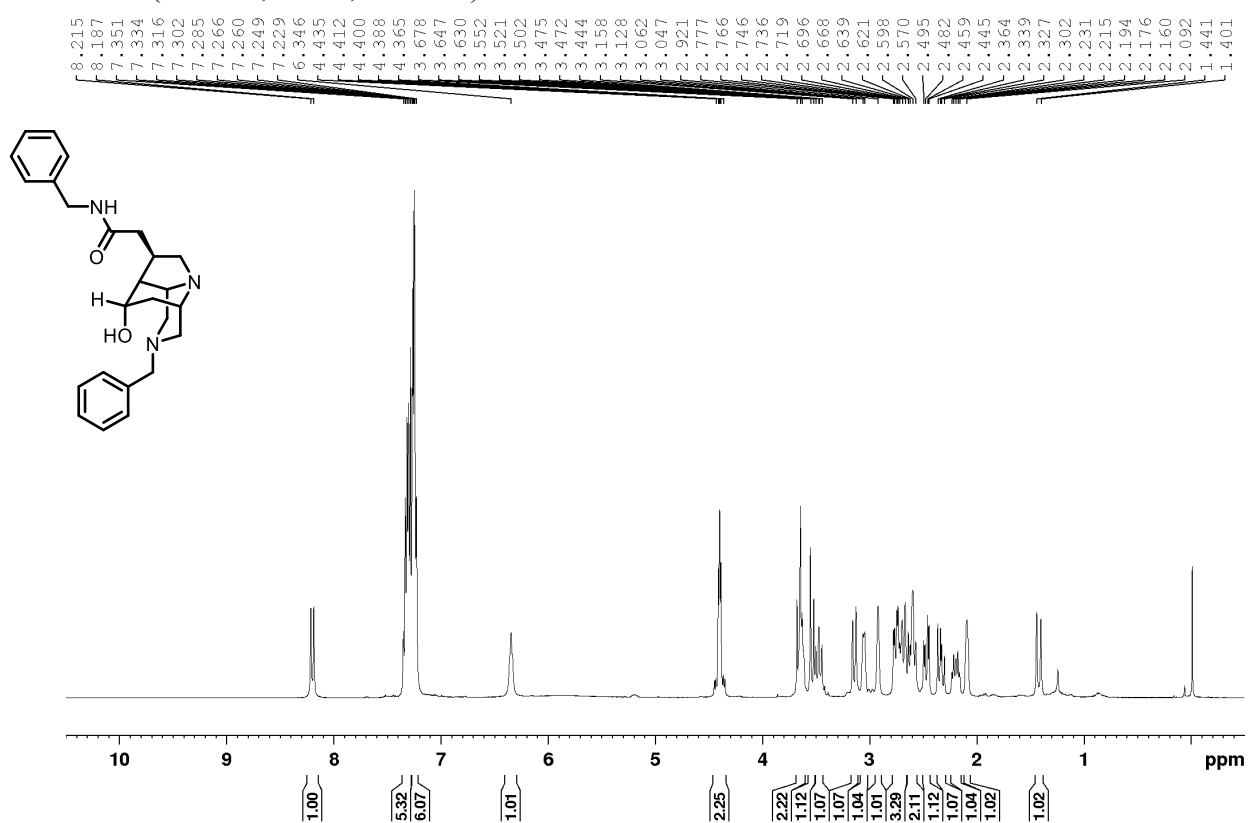
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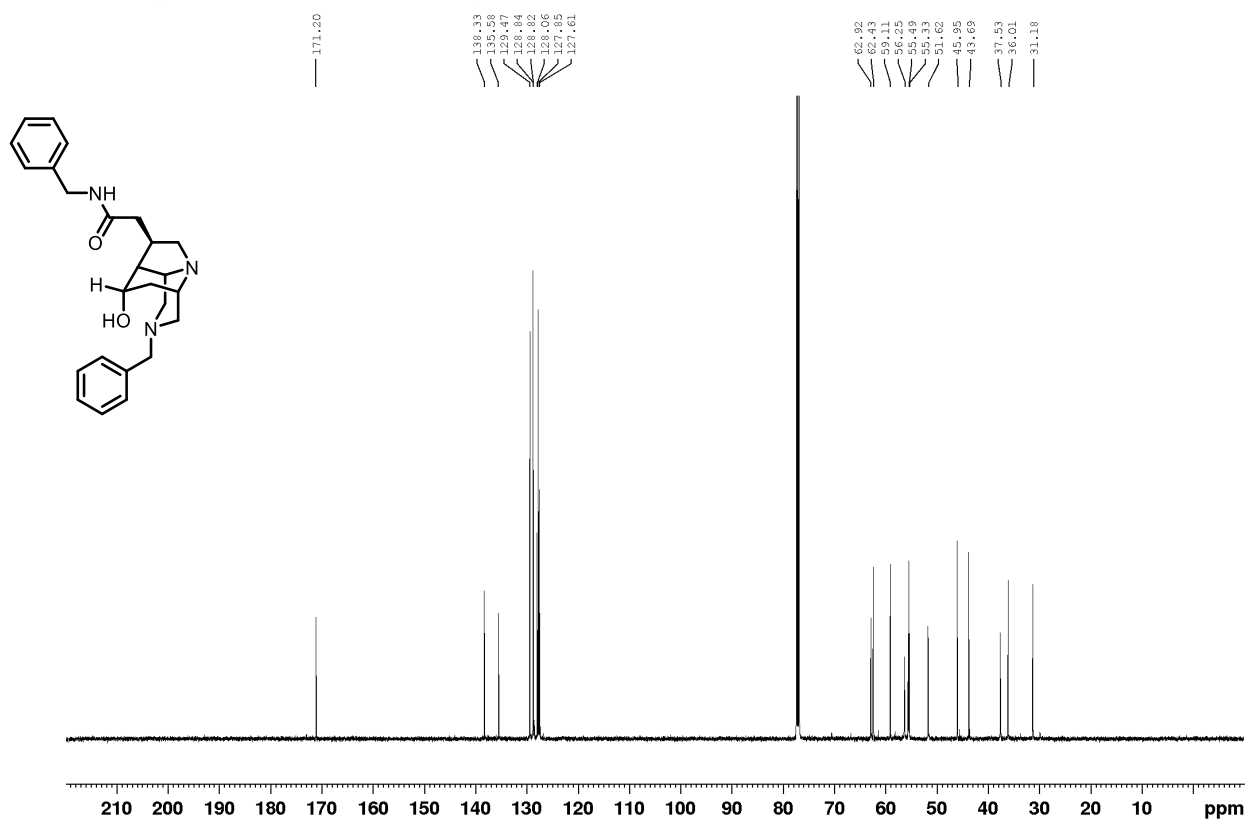
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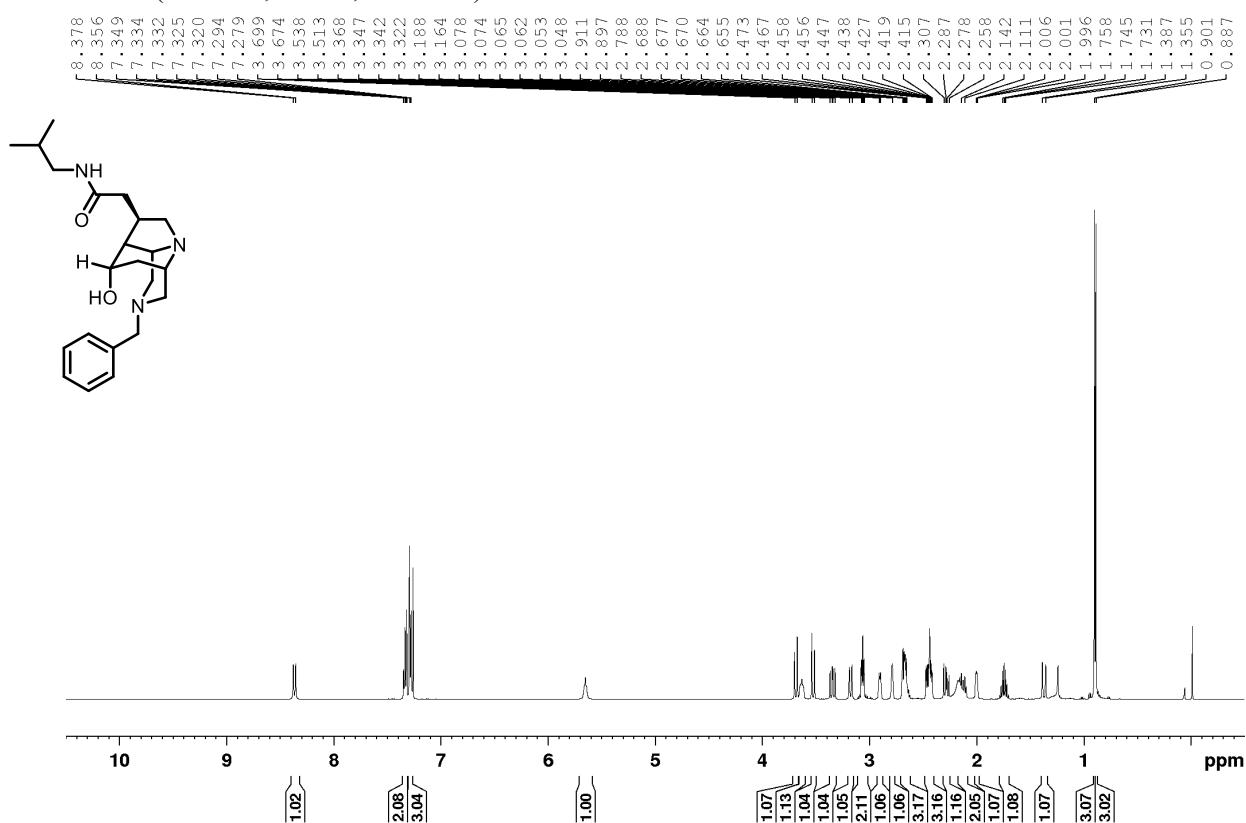
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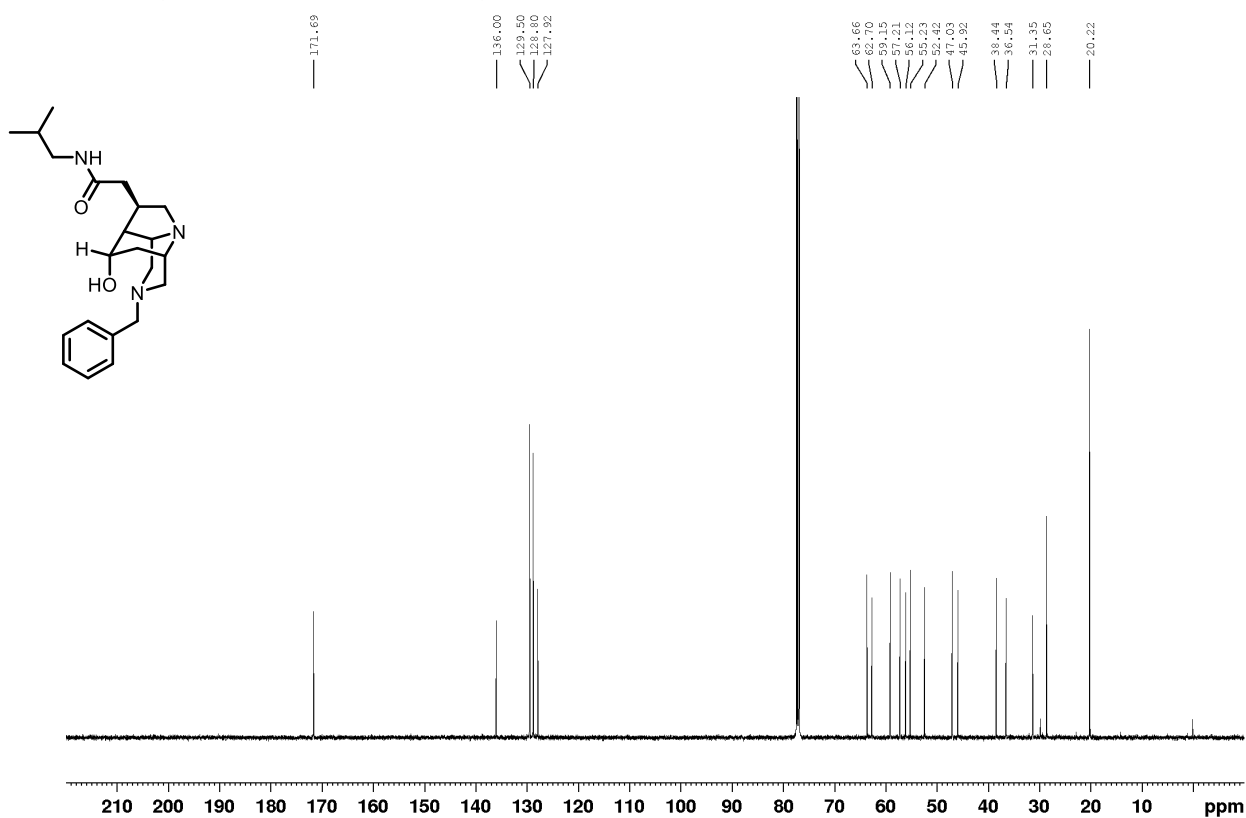
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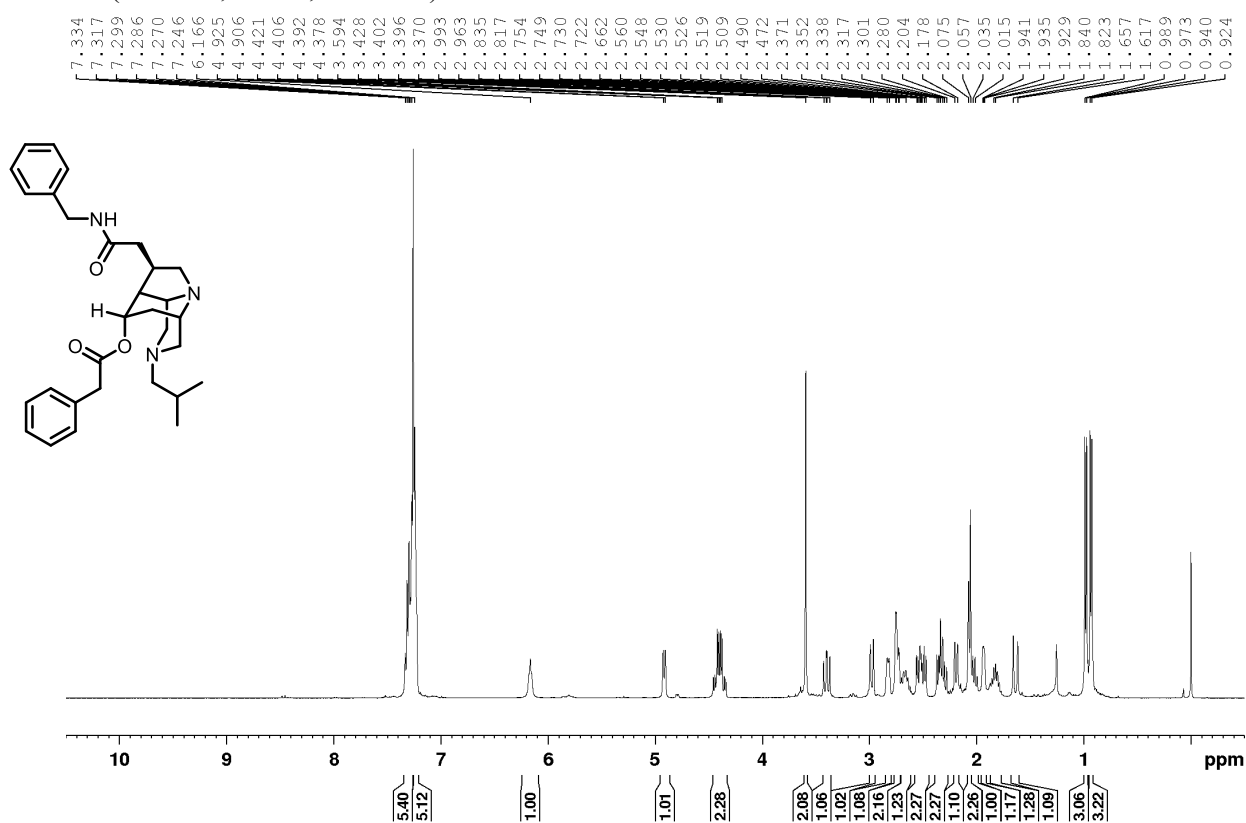
**Alcohol 13d** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 500 MHz)



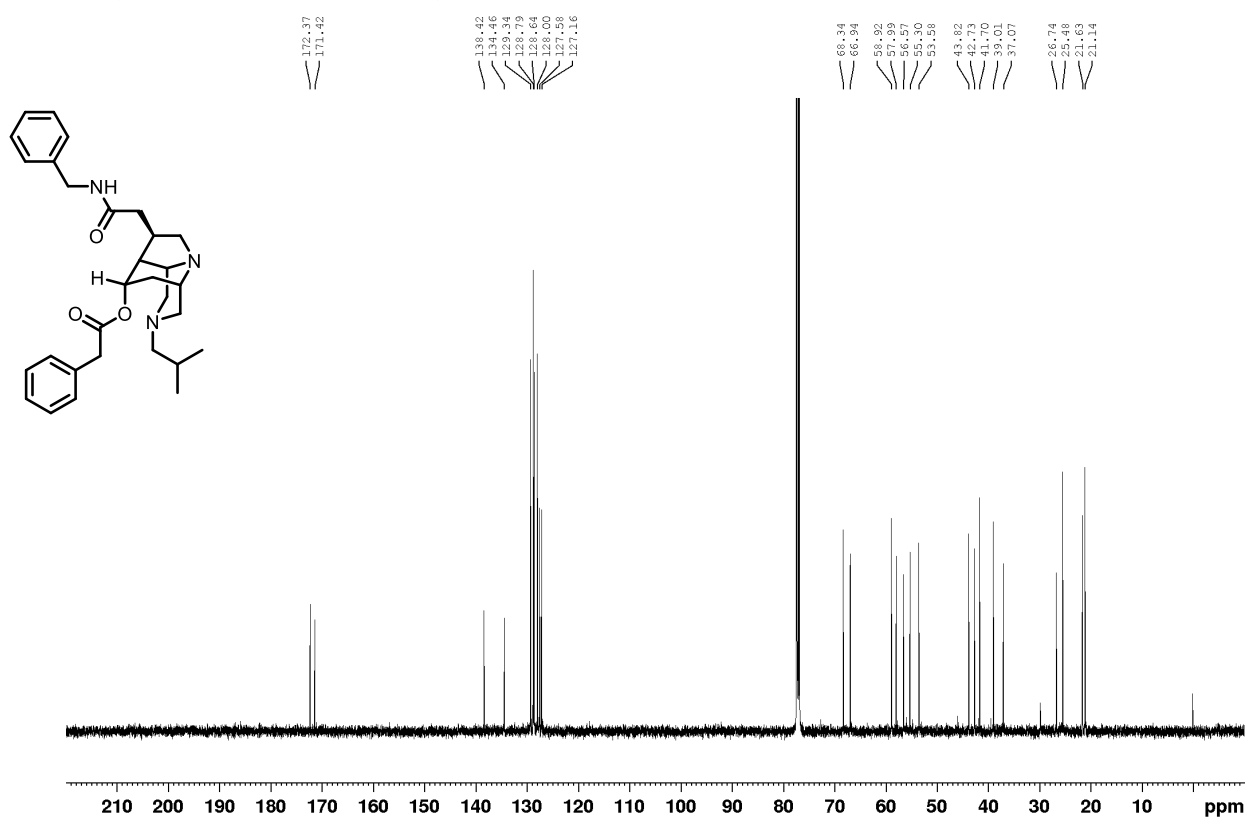
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**Ester 6a** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 400 MHz)

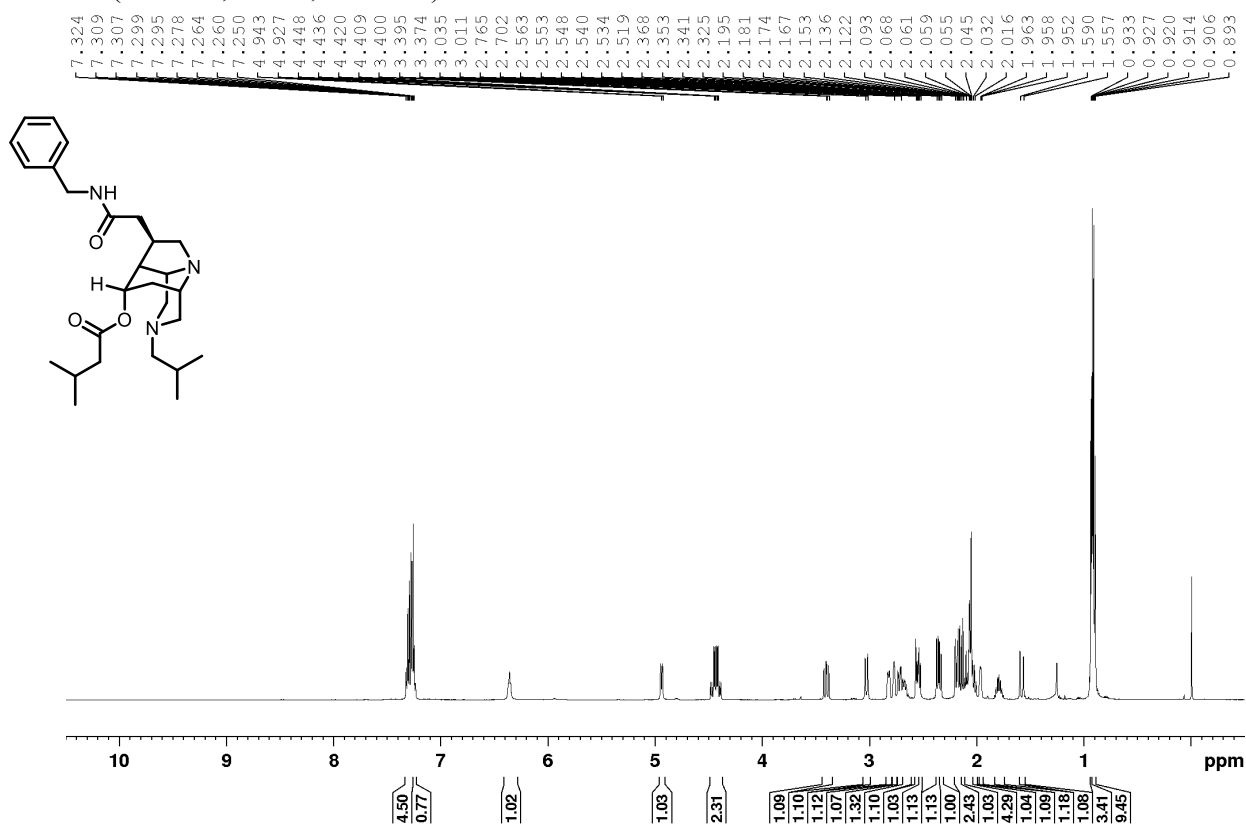


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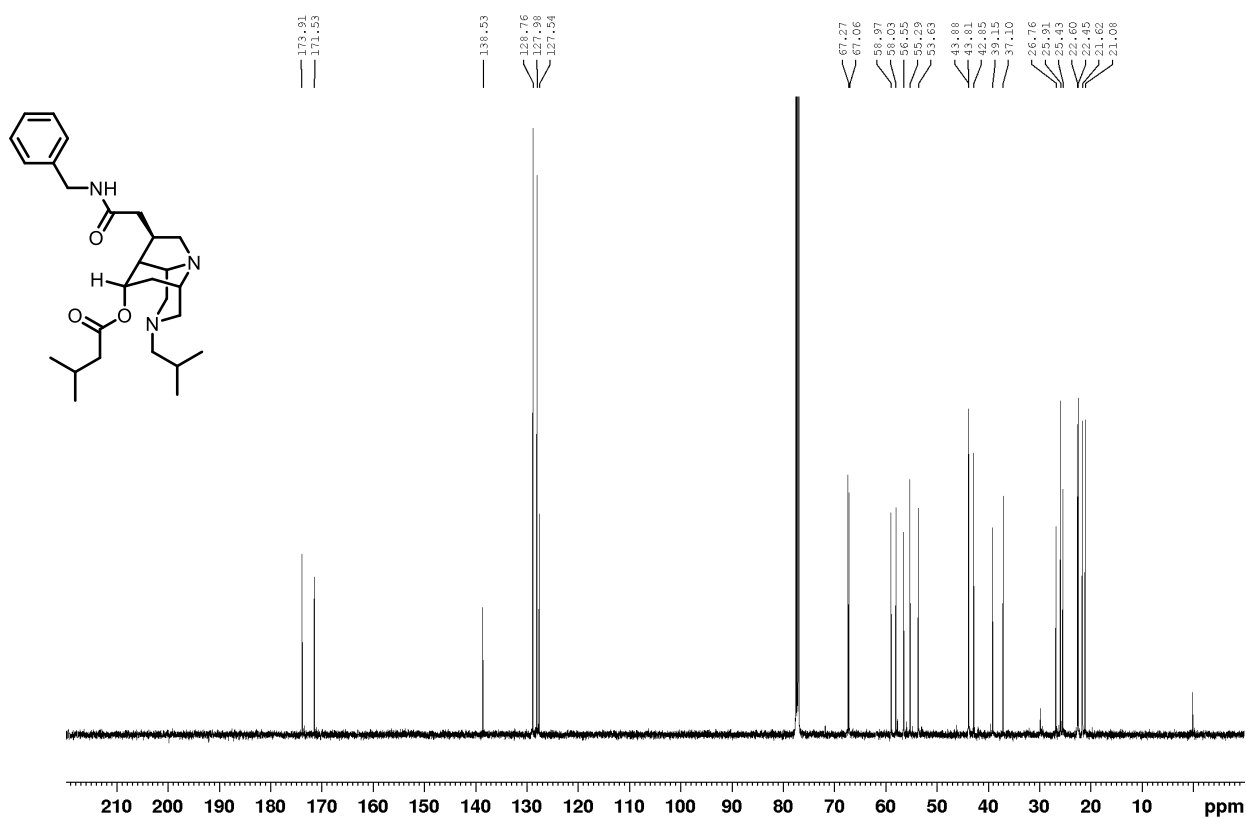




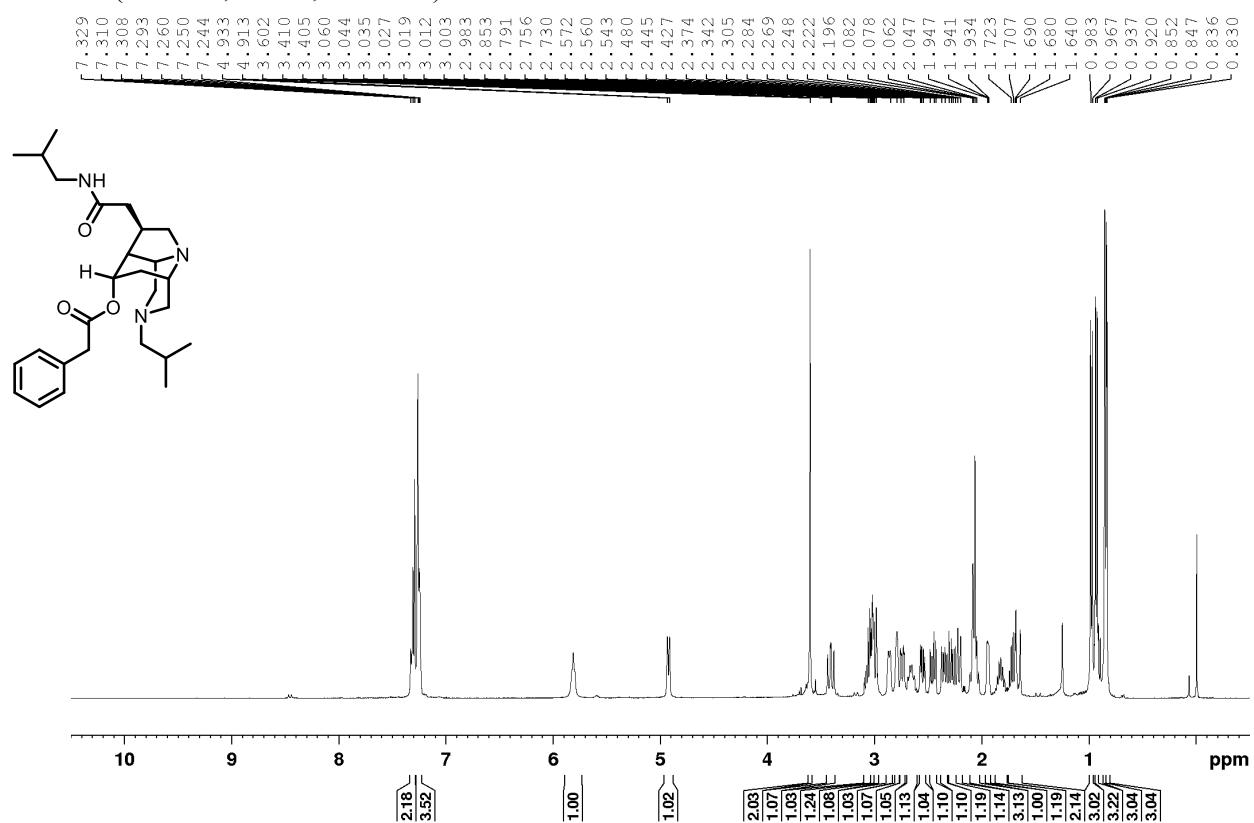
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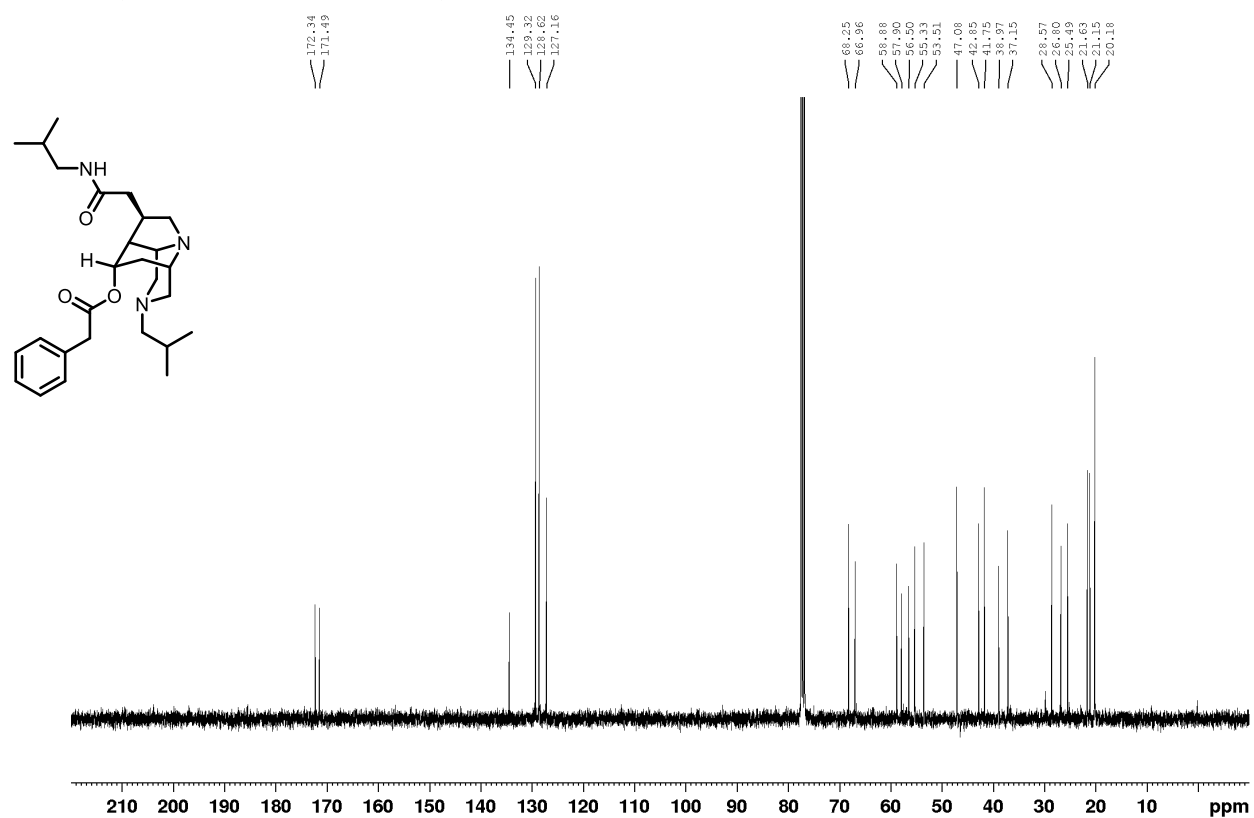
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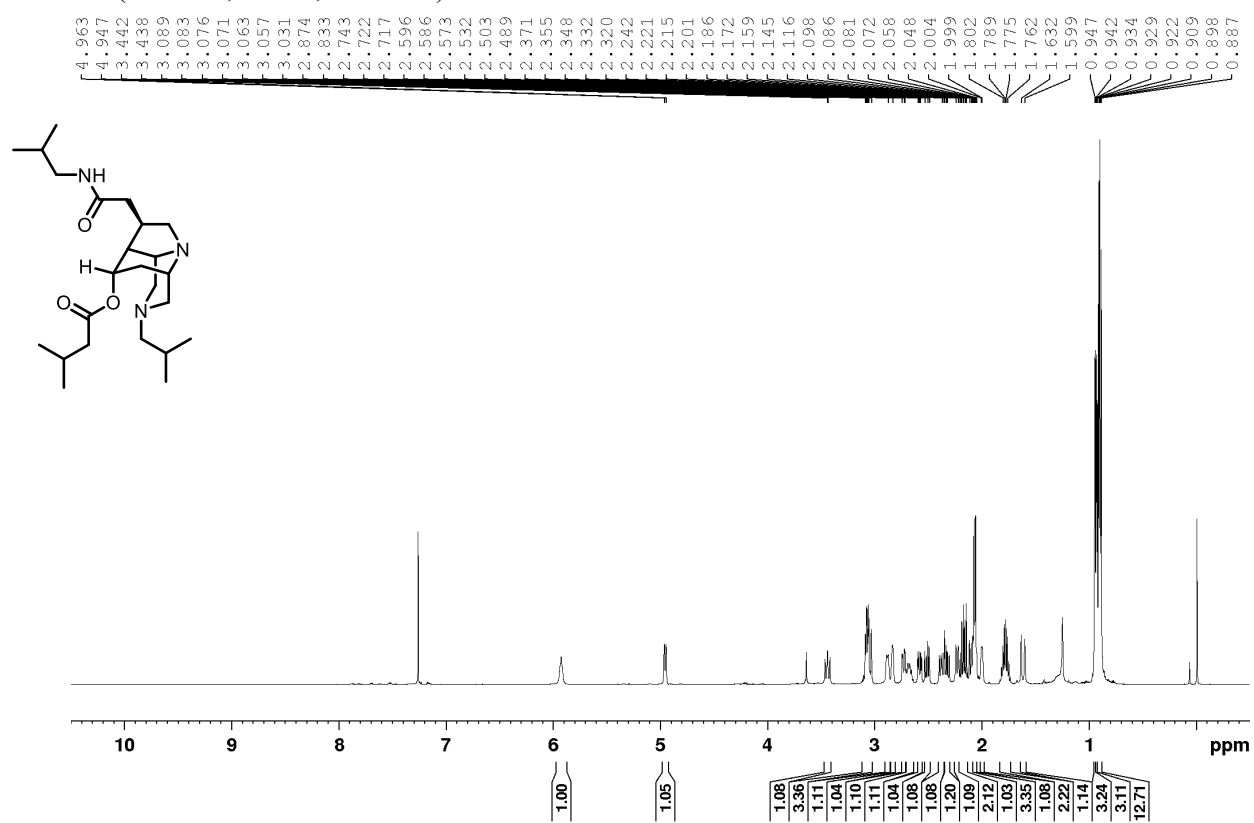
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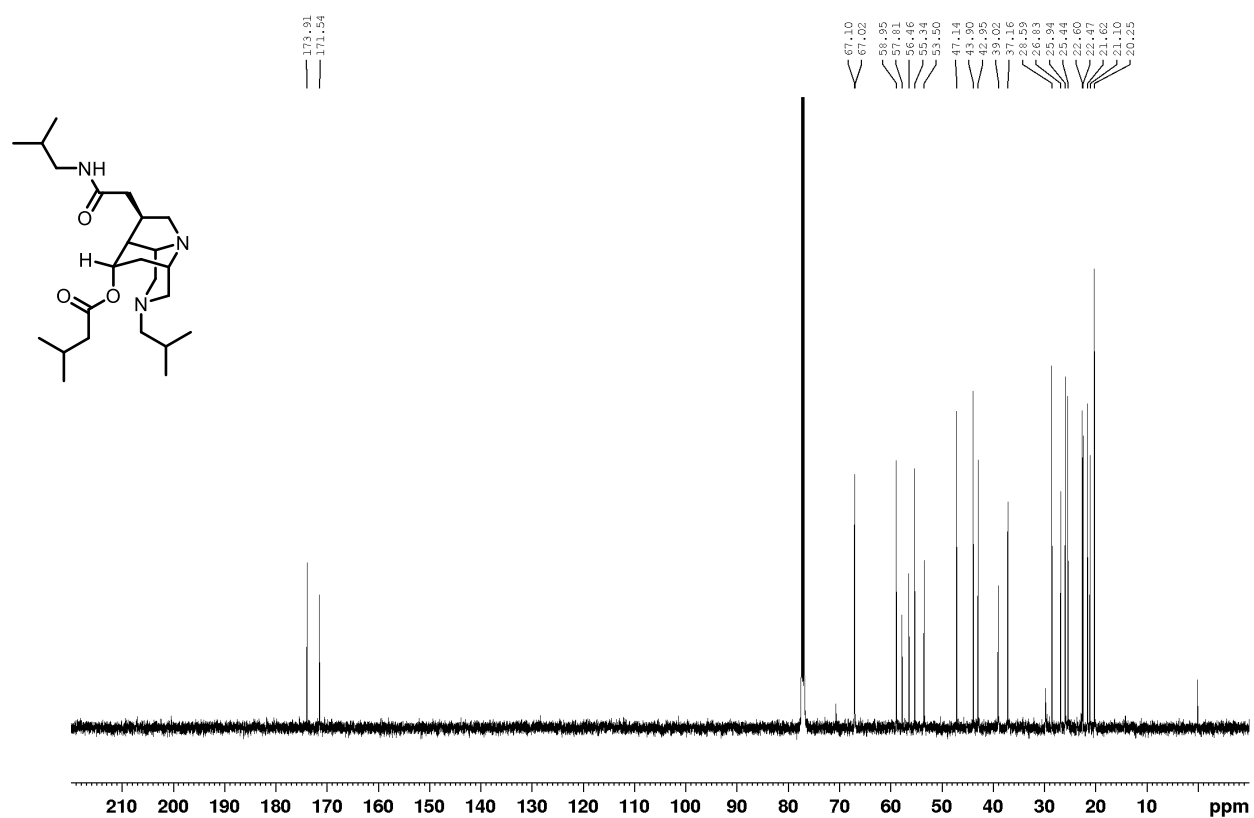
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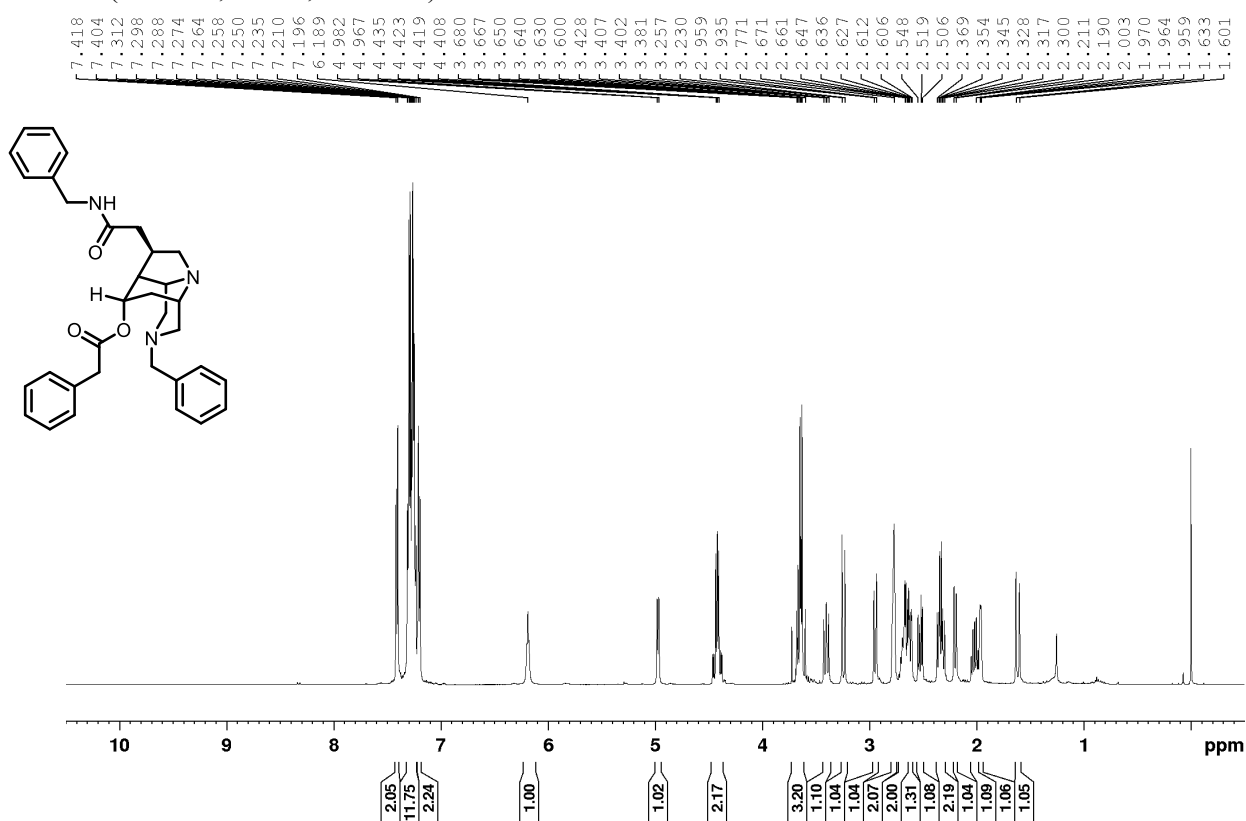
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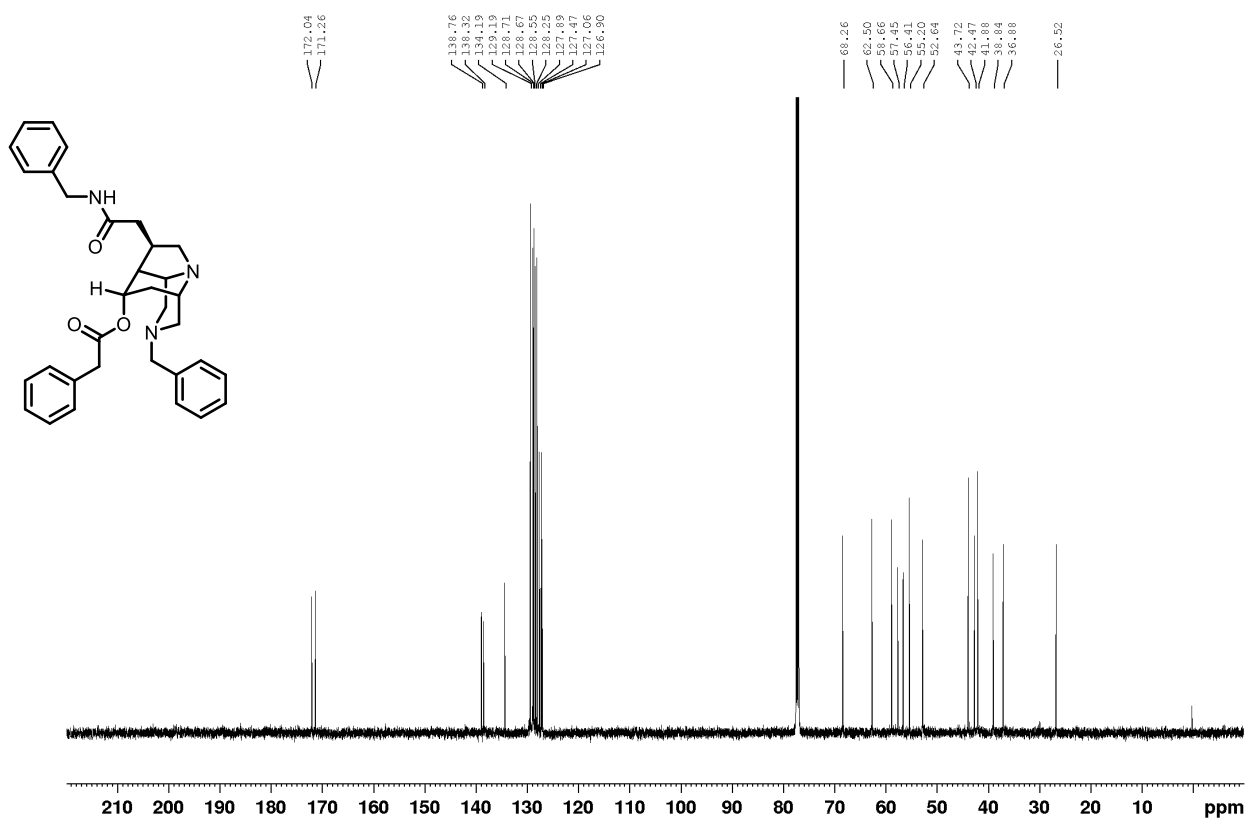
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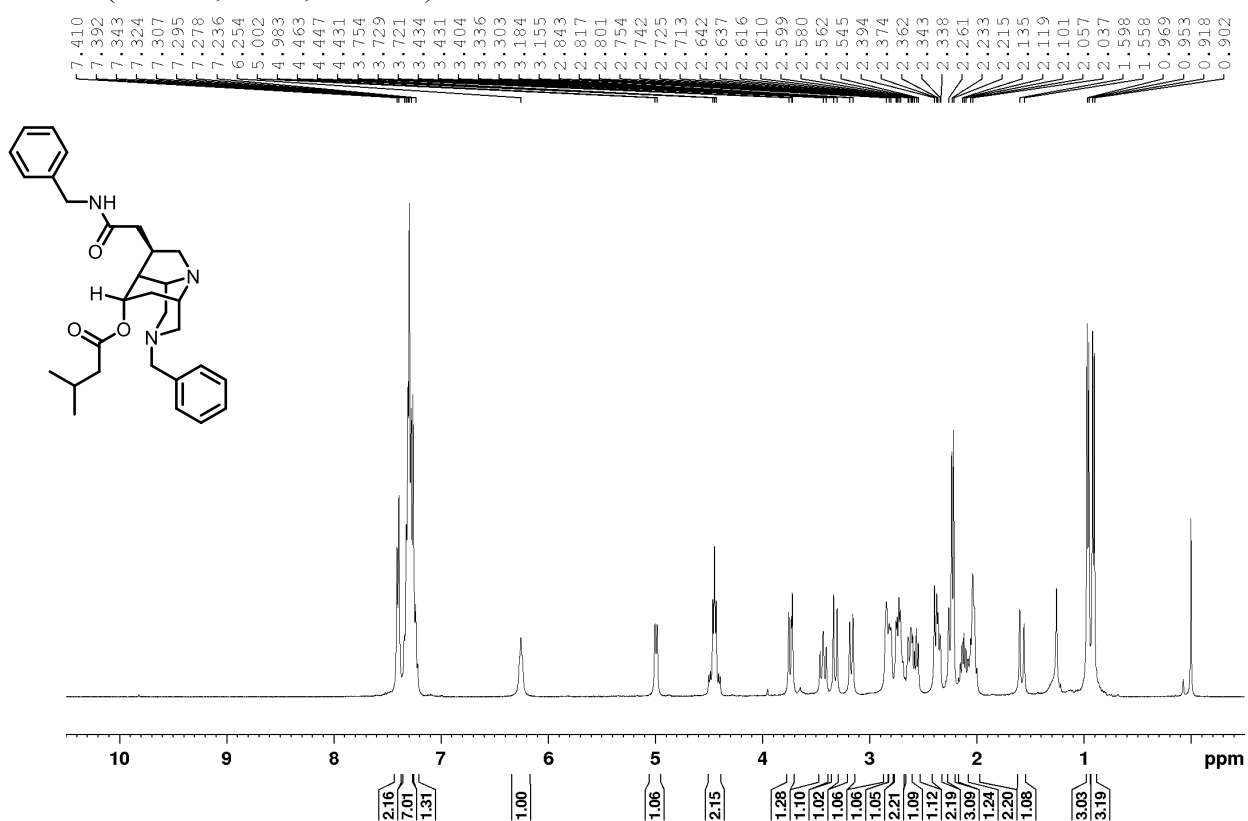
**Ester 6e** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 500 MHz)



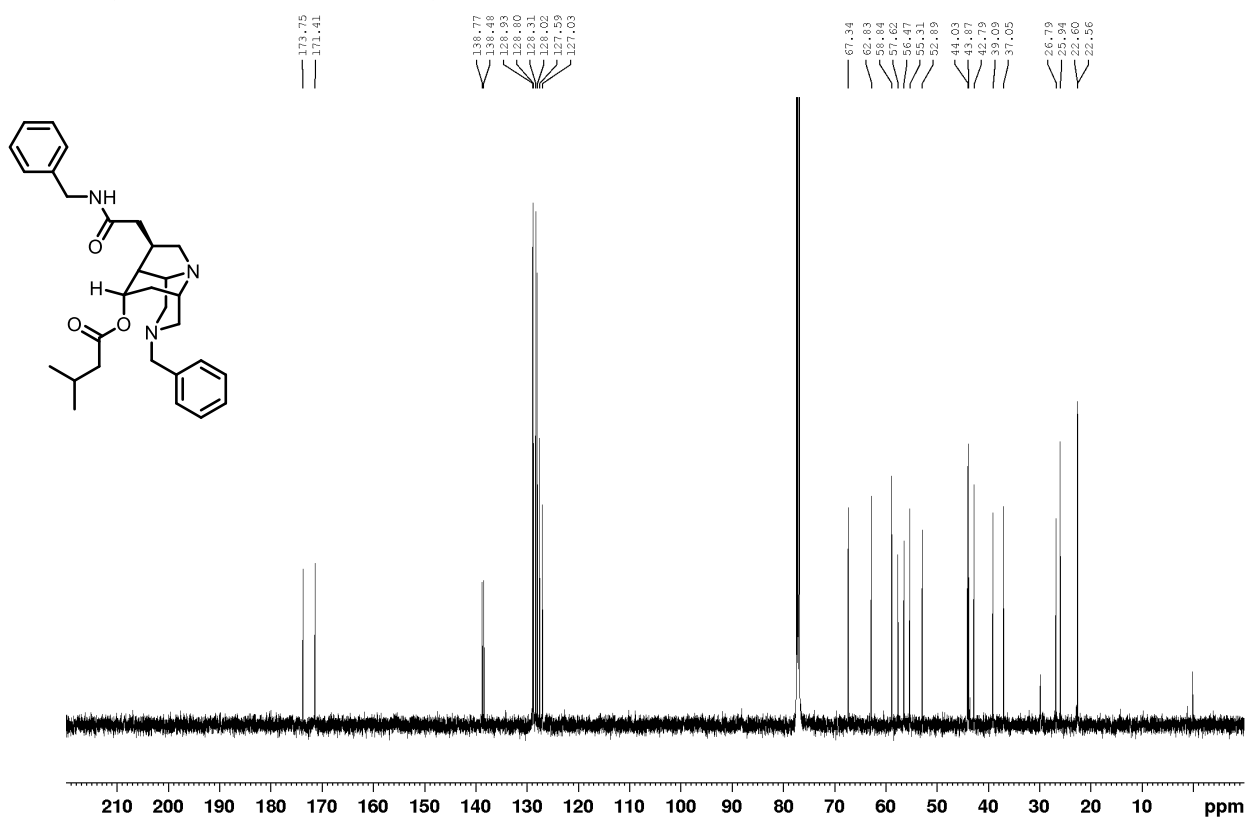
**Ester 6e** ( $^{13}\text{C}$  NMR,  $\text{CDCl}_3$ , 125 MHz)



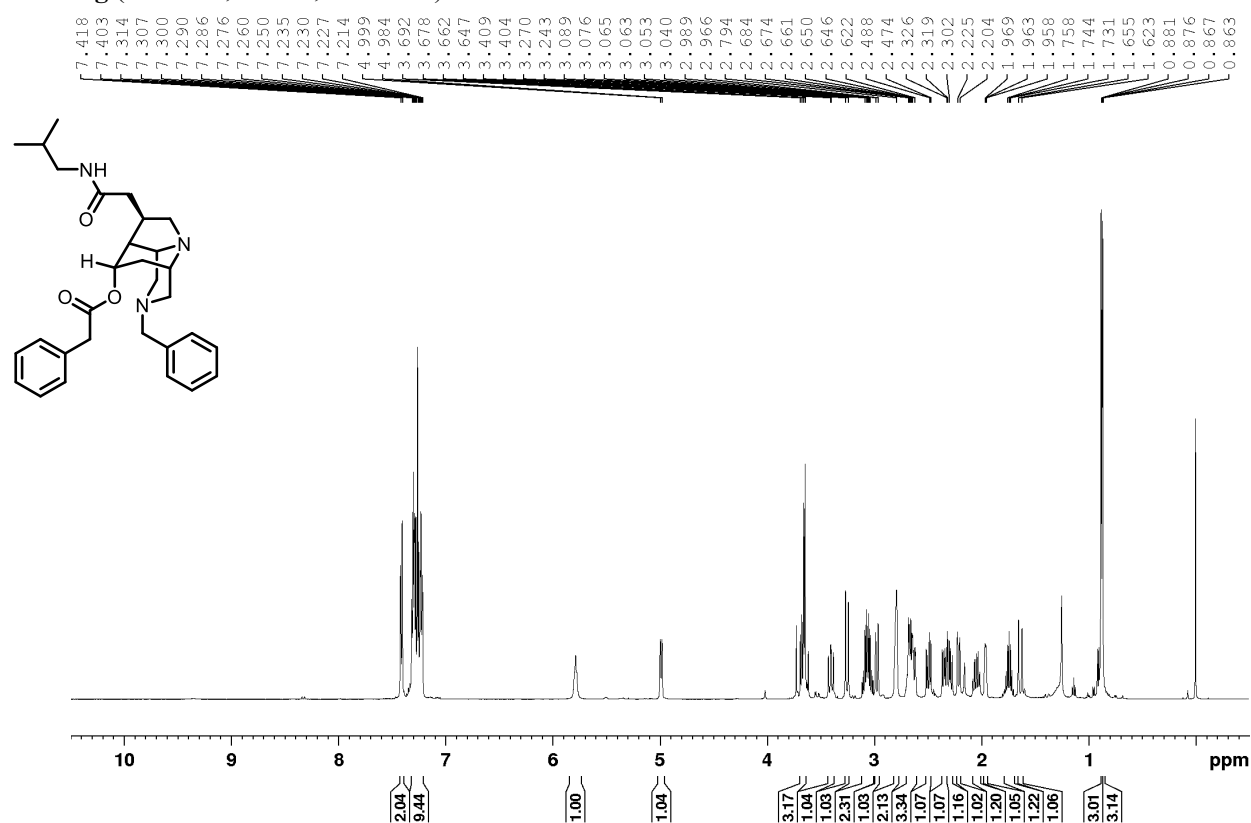
**Ester 6f** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 400 MHz)



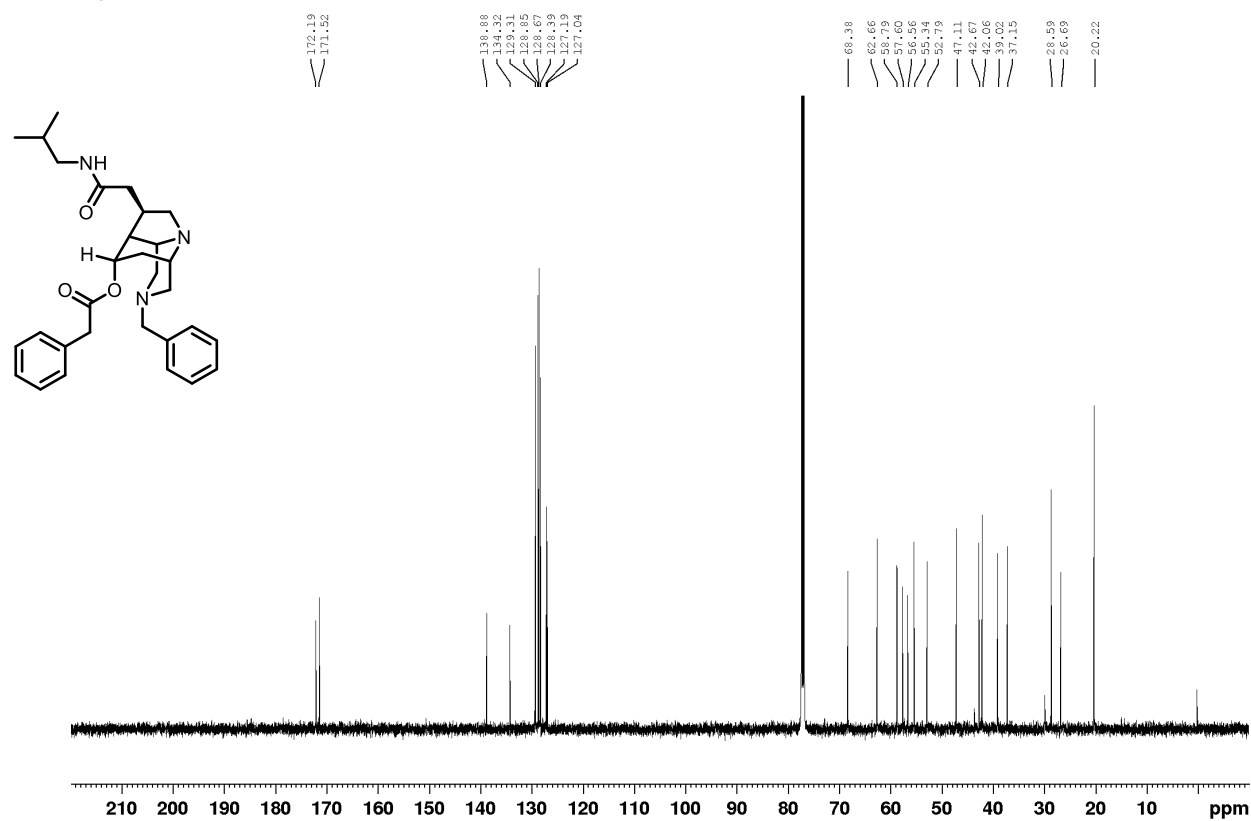
**Ester 6f** ( $^{13}\text{C}$  NMR,  $\text{CDCl}_3$ , 125 MHz)



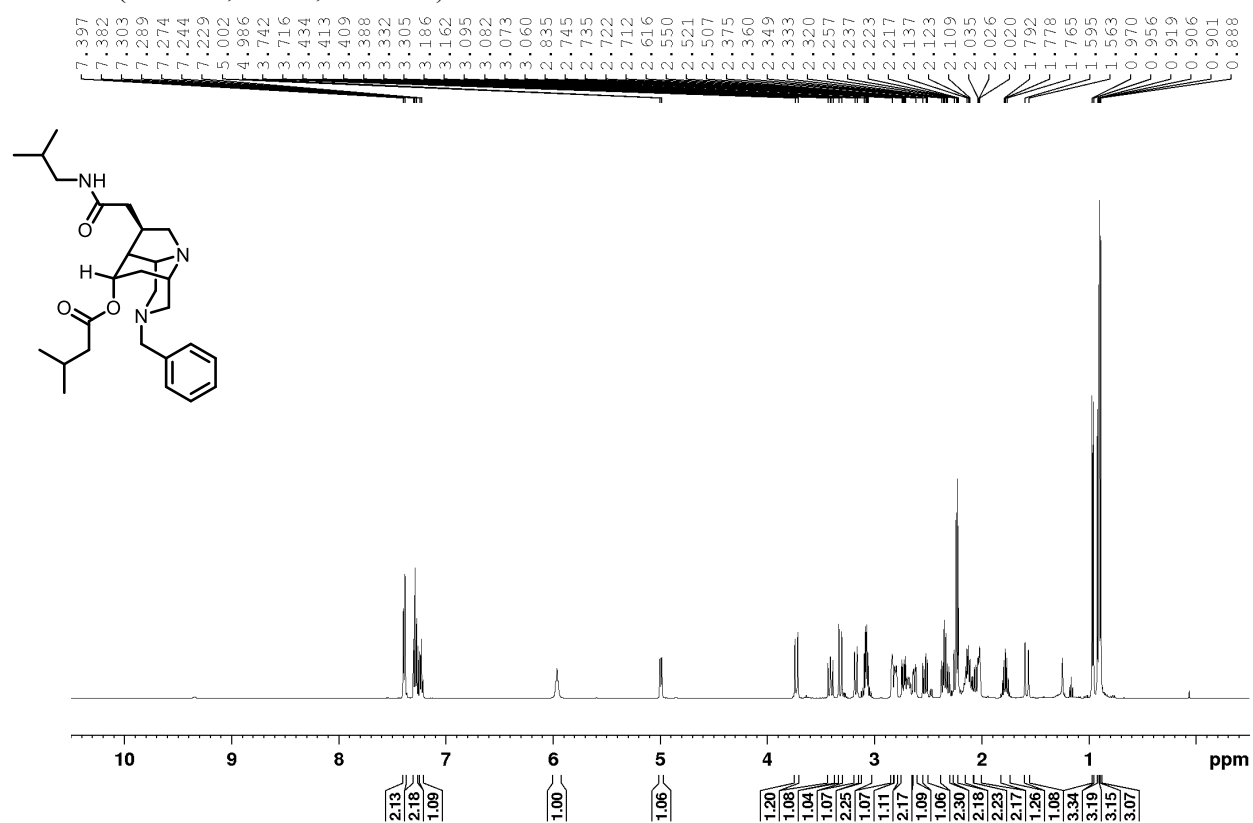
**Ester 6g** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 500 MHz)



**Ester 6g** ( $^{13}\text{C}$  NMR,  $\text{CDCl}_3$ , 125 MHz)



**Ester 6h** ( $^1\text{H}$  NMR,  $\text{CDCl}_3$ , 500 MHz)



**Ester 6h** ( $^{13}\text{C}$  NMR,  $\text{CDCl}_3$ , 125 MHz)

