

### Supporting information

#### **Visible light mediated synthesis of 1,3-diarylated imidazo[1,5a] pyridines via oxidative amination of C-H catalyzed by graphitic carbon nitride**

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## Chemicals, reagents and instrumentation

All reagents and the chemicals used for the preparation of catalyst were purchased from commercial sources (Merck and Spectrochem). The crystalline or amorphous nature of graphitic carbon nitride (g-C<sub>3</sub>N<sub>4</sub>) was analysed by using powder X-ray diffractometer (PXRD) (Shimadzu, Maxima 7000 S) using CuK $\alpha$  ( $\lambda=1.5418$  Å at 40 kV and 40 mA.  $2\theta$  range was from 5 to 80 °C (scanning speed= 5° min<sup>-1</sup>). Fourier-transform infrared spectroscopy (FTIR) studies were done for functional group analysis using Perkin Elmer, Frontier equipment. Morphology studies were analyzed using FESEM with OXFORD EDS, Zeiss, Sigma.. Thermogravimetric analysis was conducted from 25 °C to 700 °C (heating rate, 10°C/min) under nitrogen atmosphere using STAR system. <sup>1</sup>H NMR (600 MHz) and <sup>13</sup>C NMR (151MHz) were recorded on Bruker NMR spectrometer using TMS as internal standard in CDCl<sub>3</sub>.

## Preparation of g-C<sub>3</sub>N<sub>4</sub>

Synthesis of g-C<sub>3</sub>N<sub>4</sub> was accomplished according to the method reported in our earlier report <sup>1</sup>. The graphitic carbon nitride support (g-C<sub>3</sub>N<sub>4</sub>) was synthesized by calcinations of urea at 550 °C for 3h in Muffle furnace in an inert atmosphere.

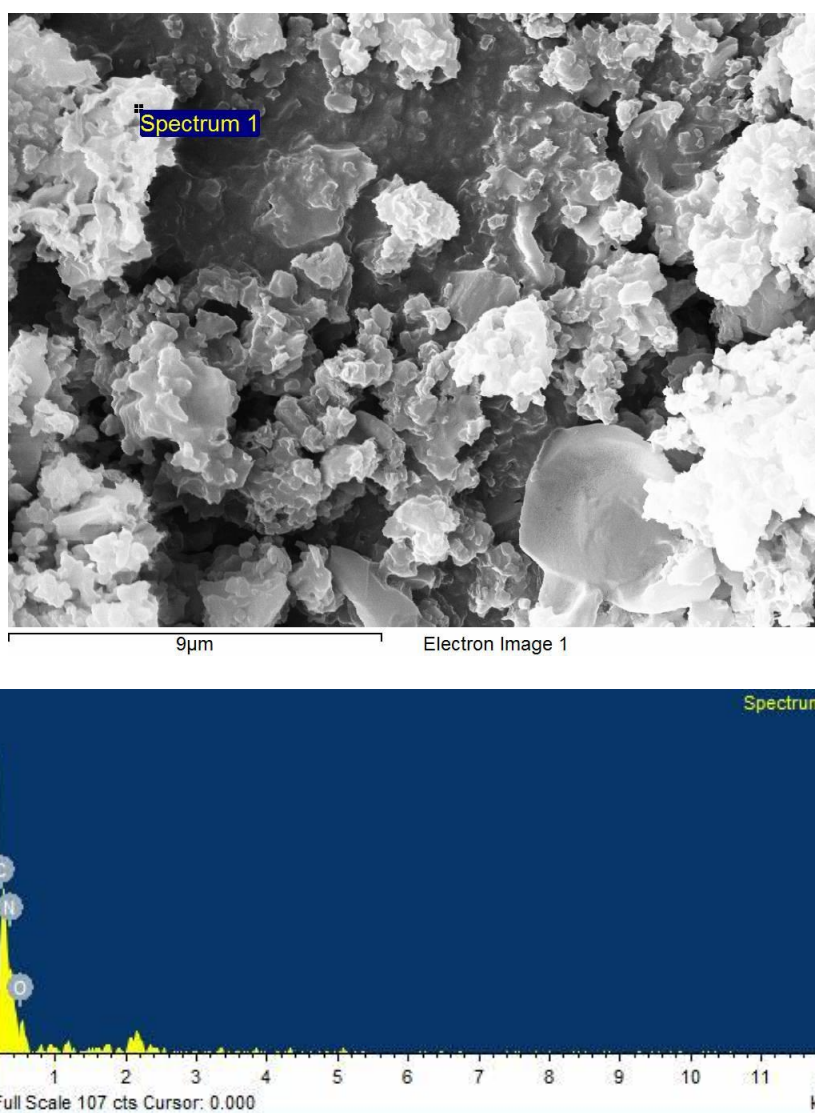


Figure S1. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub>(C = 28.24%, N = 53.14% and O = 18.62%).

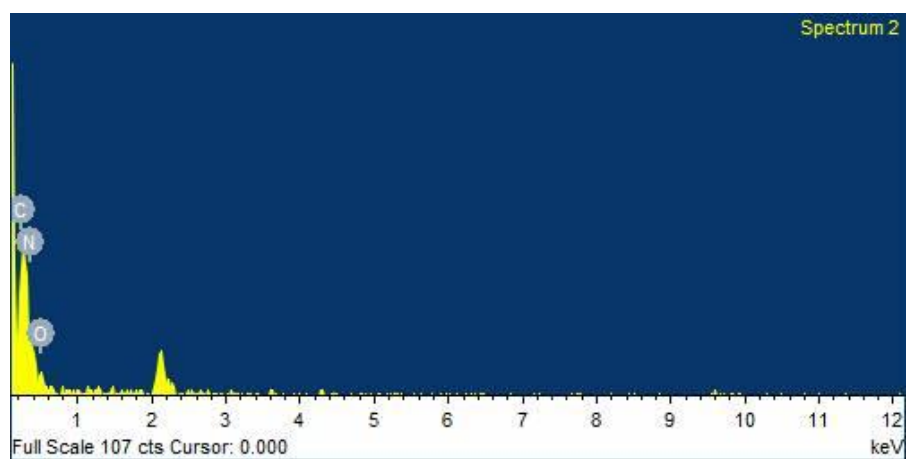
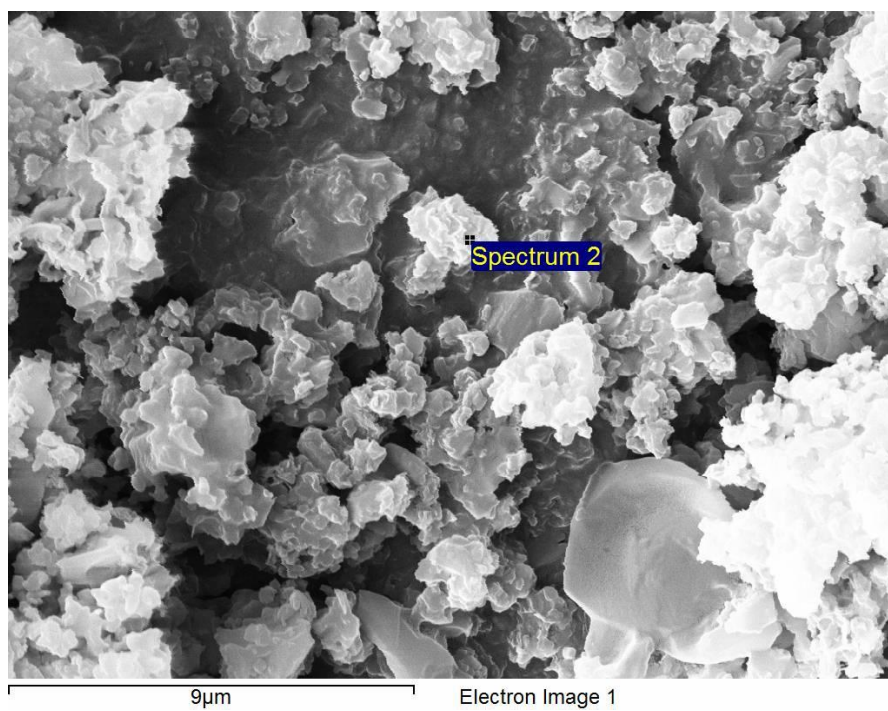


Figure S2. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub>(C = 31.97%, N = 53.44% and O = 14.59%).

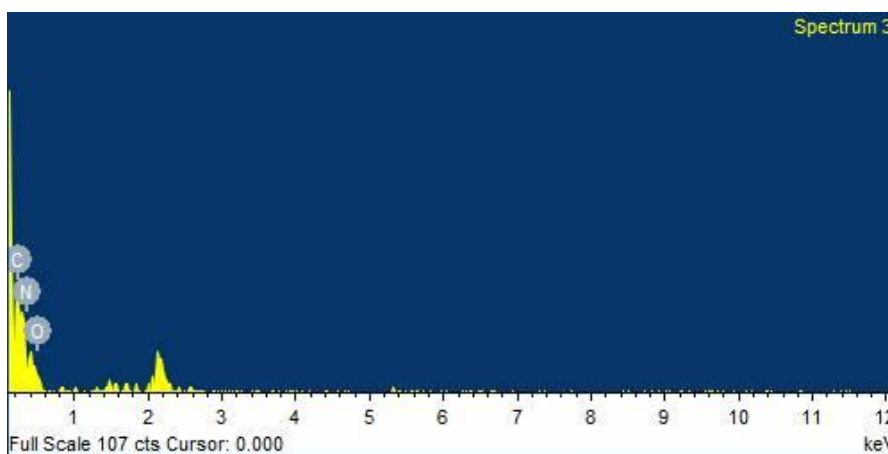
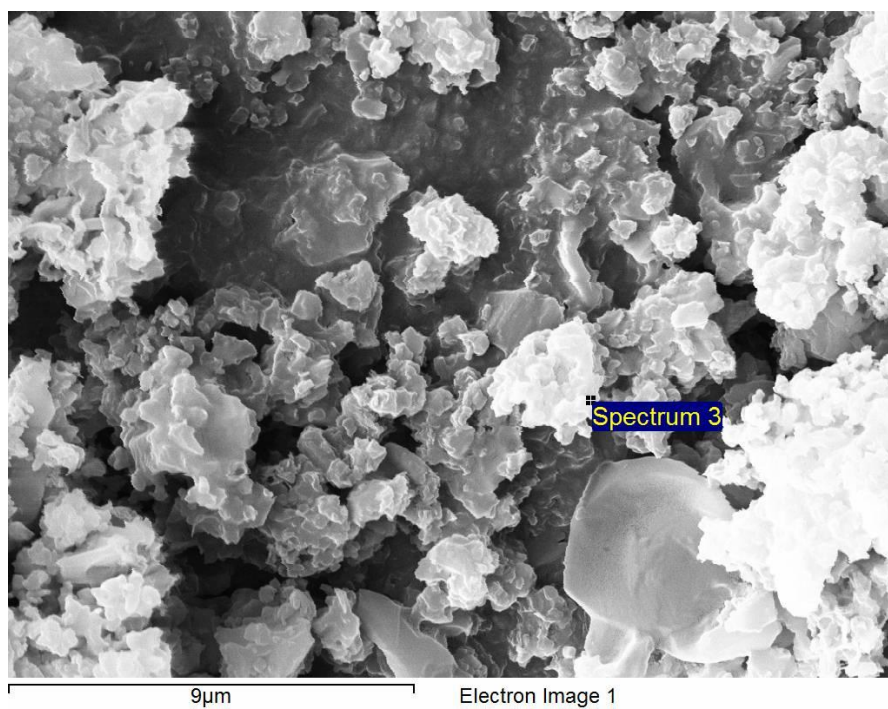


Figure S3. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 34.57%, N = 47.17% and O = 18.26%).

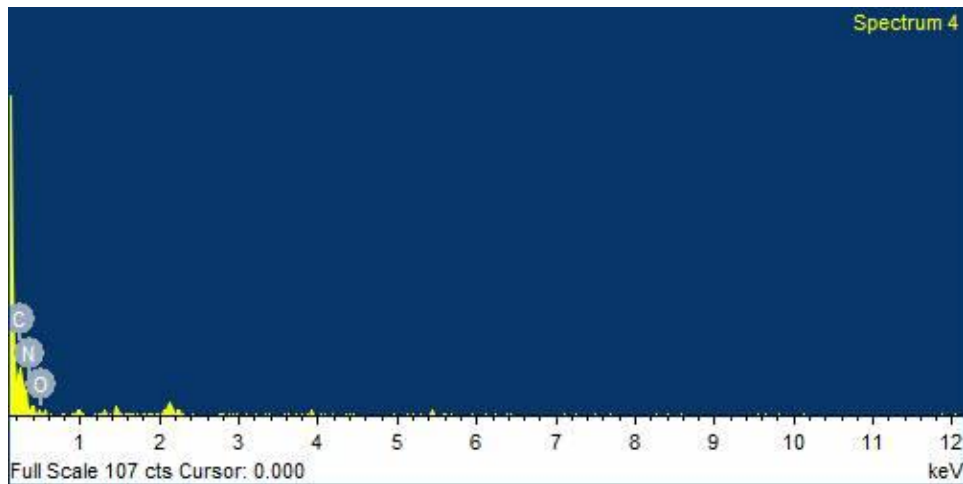
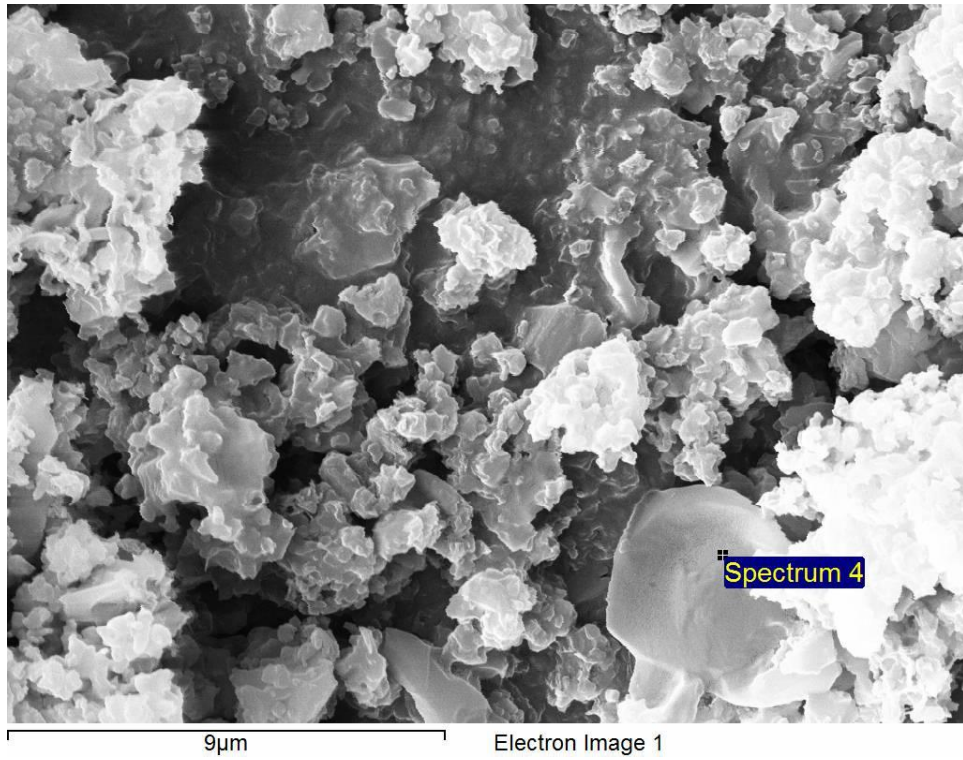


Figure S4. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 34.27%, N = 47.43% and O = 18.30%).

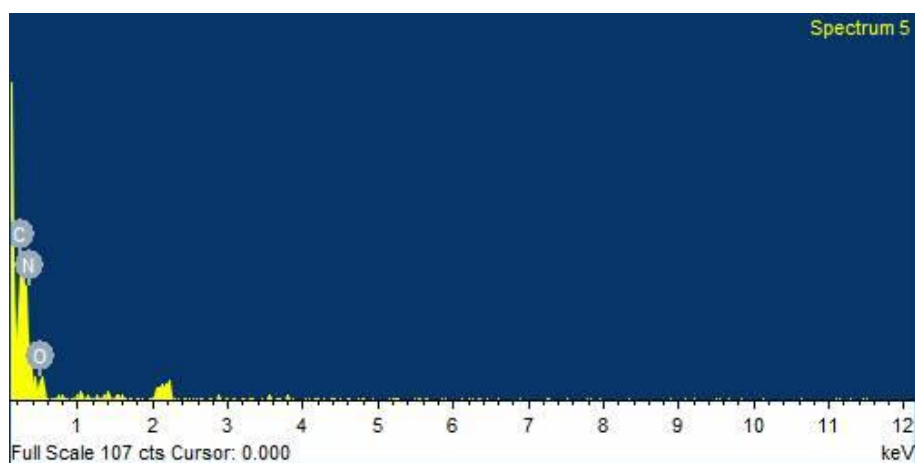
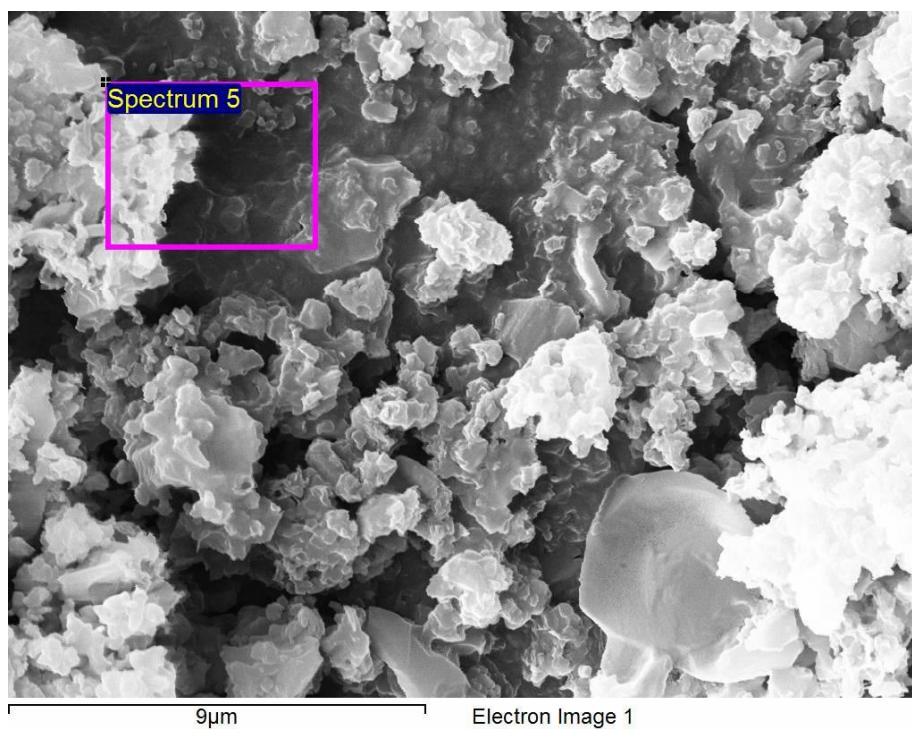
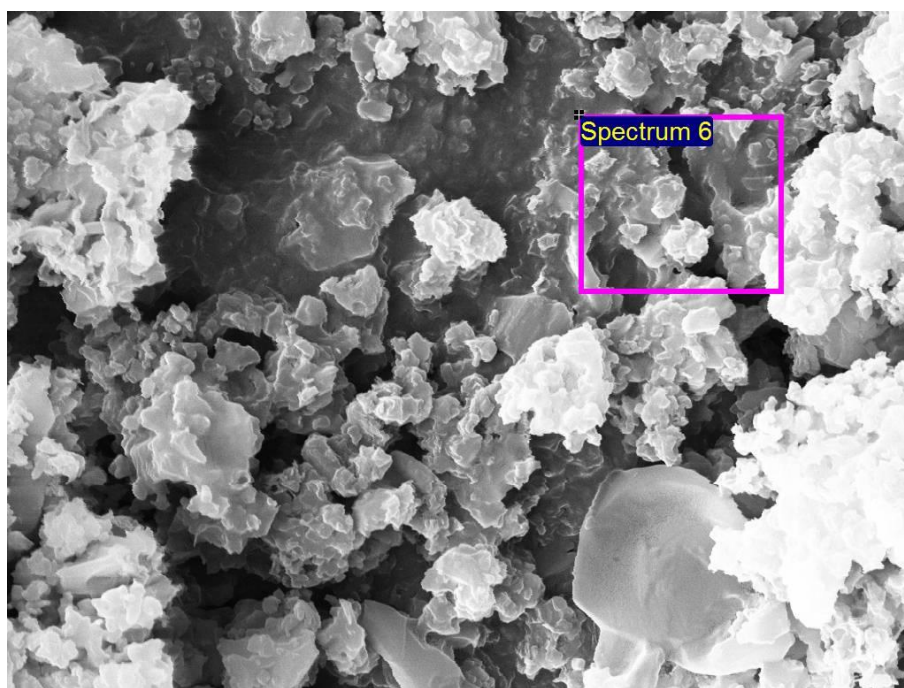


Figure S5. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 30.27%, N = 48.43% and O = 21.30%).





9µm

Electron Image 1

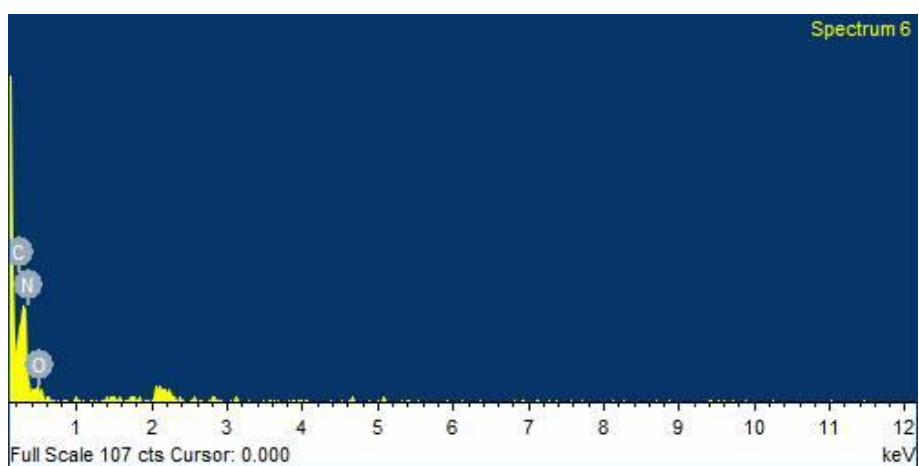


Figure S6. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 32.44%, N = 47.16% and O = 20.41%).

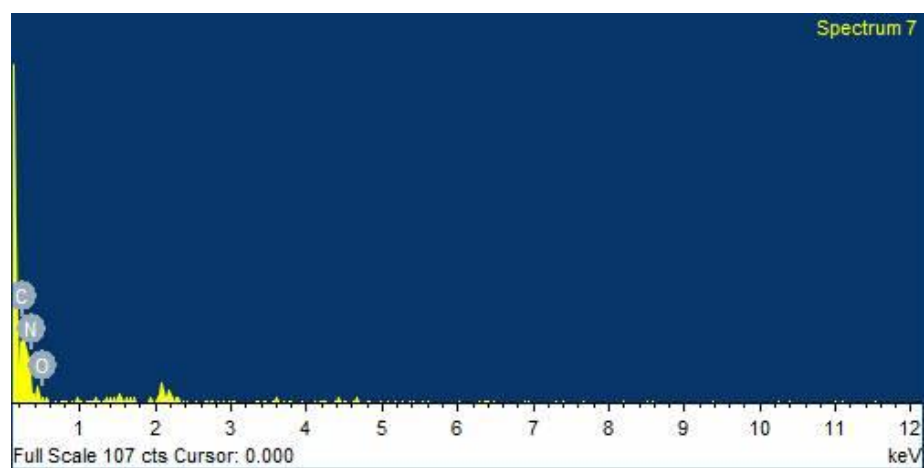
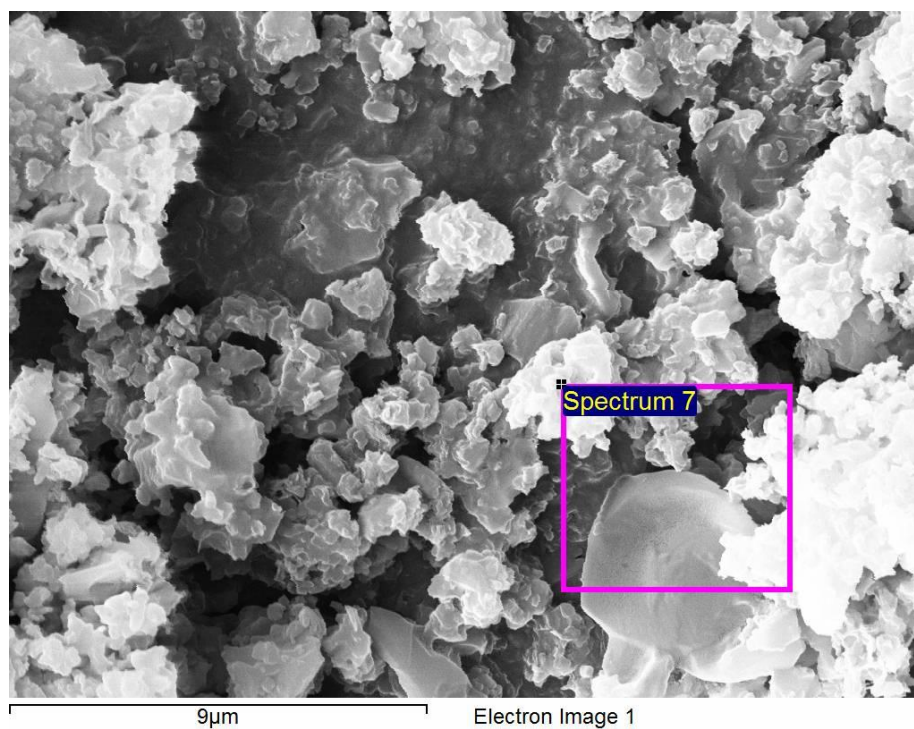


Figure S7. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 68.45%, N = 15.74% and O = 15.82%).

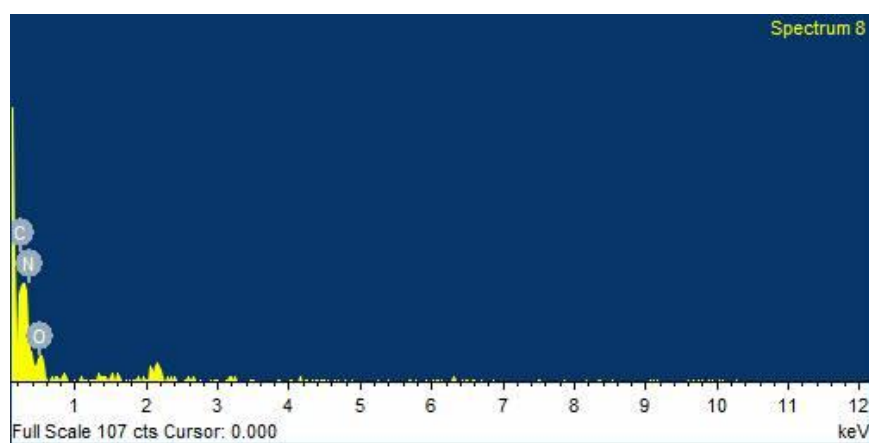
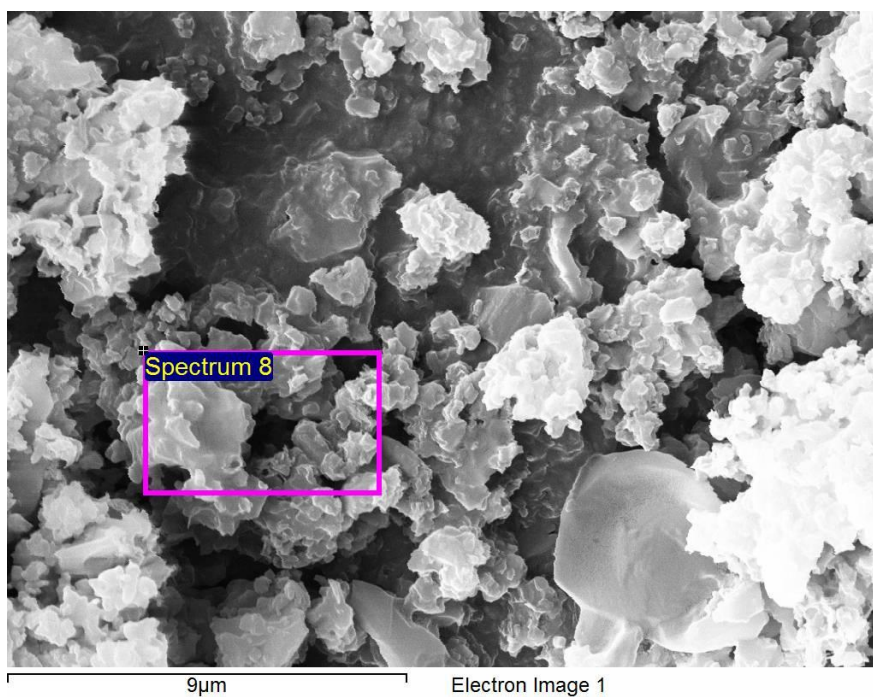


Figure S8. SEM-EDX images of g-C<sub>3</sub>N<sub>4</sub> (C = 29.62%, N = 44.81% and O = 25.57%).

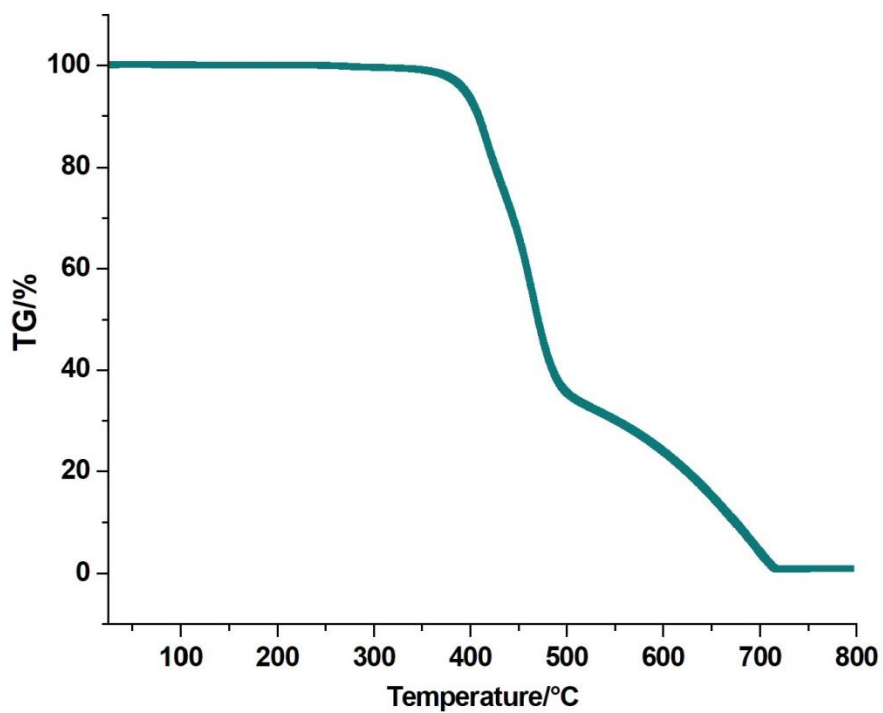
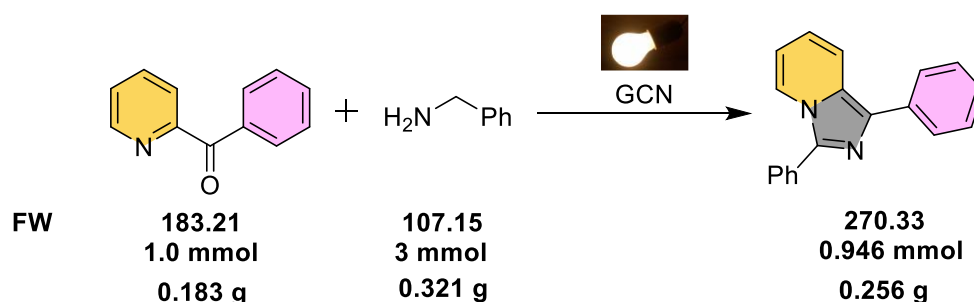


Figure S9. TGA of g-C<sub>3</sub>N<sub>4</sub>.

Table S1. Comparison table for the GCM (*green chemistry metrics*) for the synthesis of Compound 3a.

Green Metrics	Catalyst			
	Cu(OAc) <sub>2</sub> .H <sub>2</sub> O	I <sub>2</sub>	CuBr	g-C <sub>3</sub> N <sub>4</sub>
Environmental impact factor ( <i>Ef</i> )	38.73	47.33	135	0.97
Process mass intensity ( <i>PMI</i> )	39.73	48.33	136	1.96
Reaction mass efficiency ( <i>RME</i> )	49.53	89.03	40.68	51.0
Atom economy ( <i>AE</i> )	48	52.36	59	93.10
Carbon efficiency ( <i>CE</i> )	53.23	95.48	43.6	54.46
Chemical yield ( <i>CY</i> )	92.48	99.15	43.72	95.0
Mass intensity ( <i>MI</i> )	2.13	1.69	2.67	1.96
Mass productivity ( <i>MP</i> )	45.0	59.17	37.45	51.02
Optimum efficiency ( <i>OE</i> )	103.18	170.03	69	55.0
References	2	3	4	This Work

Calculation of Green chemistry metrics (GCM) for compound (3a)<sup>5-6</sup>.



**Environmental impact factor (*Ef*):**

*E-factor* = [Total mass of raw materials minus the total mass of product] / Mass of product

$$E\text{-factor} = [0.183 \text{ g} + 0.321 \text{ g} - 0.256 \text{ g}] / [0.256 \text{ g}]$$

$$E\text{-Factor} = 0.97$$

**Process mass intensity (*PMI*):**

$$PMI = \sum(\text{Mass of stoichiometric reactants}) / [\text{Mass of product}]$$

$$PMI = \sum[(0.183 \text{ g} + 0.321 \text{ g}) / [0.256 \text{ g}]]$$

$$PMI = 1.96$$

**Reaction mass efficiency (*RME*):**

$$RME = \text{Mass of product} / \sum(\text{Mass of stoichiometric reactants}) \times 100$$

$$RME = [0.256] / \sum[0.183 \text{ g} + 0.321 \text{ g}] \times 100$$

$$\text{RME} = 51.0 \%$$

**Atom economy (AE):**

$$AE = [MW \text{ of product}] \% \sum (MW \text{ of stoichiometric reactants}) \times 100$$

$$AE = [270.33] \% \sum [183.21 + 107.15] \times 100$$

$$AE = 93.10\%$$

**Carbon efficiency (CE):**

$$CE = [\text{Amount of carbon in product}] / [\text{Total carbon present in reactants}] \times 100$$

$$CE = [\text{no. of moles of product} \times \text{no. of carbons in product}] \times 100 / [\text{no. of moles} \times \text{no. of carbon} + (\text{no. of moles} \times \text{no. of carbon atoms})]$$

$$CE = [0.946 \times 19] \times 100 / [1 \times 12 + 3 \times 7]$$

$$CE = 54.46\%$$

**Chemical yield (CY):**

$$CY = [\text{Weight of product} \times \text{MW of starting material}] \times 100 / [\text{Weight of Starting material} \times \text{MW of Product}]$$

$$CY = [0.256 \text{ g} \times 183.21] \times 100 / [0.183 \times 270.33]$$

$$CY = 95.0 \%$$

**Mass intensity (MI):**

$$MI = \sum \text{Weight (Total materials input)} / [\text{Weight of Product}]$$

$$MI = \sum [(0.183 \text{ g} + 0.321 \text{ g}) / [0.256 \text{ g}]]$$

$$MI = 1.96$$

**Mass productivity (MP):**

$$MP = 100/MI$$

$$MP = 51.02 \%$$

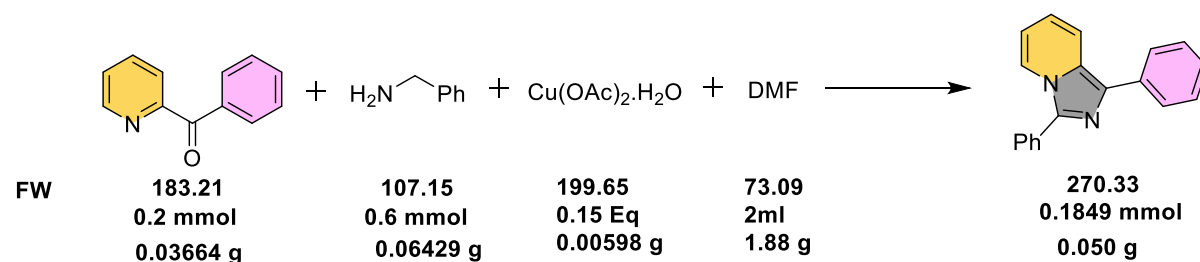
**Optimum efficiency (OE):**

$$OE = [\text{RME}] \times 100/AE$$

$$OE = 55.0 \%$$

**Note:- The Reaction was performed with g-C<sub>3</sub>N<sub>4</sub> catalyst that was recovered and recycled. Hence the mass of the catalyst is excluded.**

**Calculation of Green chemistry metrics (GCM) for compound (3a) by Cu(OAc)<sub>2</sub>.H<sub>2</sub>O (reported ones).**



***E-factor*** = [Total mass of raw materials minus the total mass of product] / mass of product

***E-factor*** = [0.03664 g + 0.06429 g + 0.00598 g (Catalyst) + 1.88 g (DMF) - 0.050 g] / [0.050 g]

**E-Factor = 38.73**

***PMI*** =  $\sum$ (mass of stoichiometric reactants) / [Mass of product]

***PMI*** =  $\sum$ [(0.03664 g + 0.06429 g + 0.00598 g (Catalyst) + 1.88 g (DMF)] / [0.050 g]

**PMI = 39.73**

***RME*** = mass of product /  $\sum$ (mass of stoichiometric reactants)  $\times$  100

***RME*** = [0.050] /  $\sum$ [0.03664 g + 0.06429 g]  $\times$  100

**RME = 49.53%**

***AE*** = [MW of product] %  $\sum$ (MW of stoichiometric reactants)  $\times$  100

***AE*** = [270.33] %  $\sum$ [183.21 + 107.15 + 199.65(Catalyst) + 73.09 (DMF)]  $\times$  100

**AE = 48%**

***CE*** = [Amount of carbon in product] / [Total carbon present in reactants]  $\times$  100

***CE*** = [no. of moles of product  $\times$  no. of carbons in product]  $\times$  100 / [no. of moles  $\times$  no. of carbon + (no. of moles  $\times$  no. of carbon atoms + (no. of moles  $\times$  no. of carbon atoms)]

***CE*** = [0.1849  $\times$  19]  $\times$  100 / [0.2  $\times$  12 + 0.6  $\times$  7]

**CE = 53.23**

***CY*** = [Weight of product  $\times$  MW of starting material]  $\times$  100 / [Weight of Starting material  $\times$  MW of Product]

***CY*** = [0.050 g  $\times$  183.21]  $\times$  100 / [0.03664  $\times$  270.33]

**CY = 92.48%**

$$MI = \sum Weight \text{ (Total materials input)} / [\text{Weight of Product}]$$

$$MI = \sum [(0.03664 \text{ g} + 0.06429 \text{ g} + 0.00598 \text{ g}) / [0.050 \text{ g}]]$$

$$MI = 2.13$$

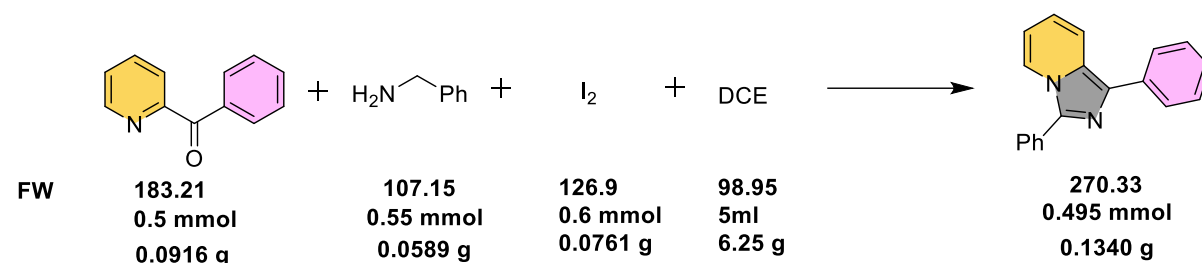
$$MP = 100/MI$$

$$MP = 45.0$$

$$OE = [RME] \times 100/AE$$

$$OE = 103.18$$

Calculation of Green chemistry metrics (GCM) for compound (3a) by I<sub>2</sub> (reported ones).



$$E\text{-factor} = [\text{Total mass of raw materials minus the total mass of product}] / \text{mass of product}$$

$$E\text{-factor} = [0.0916 \text{ g} + 0.0589 \text{ g} + 0.0761 \text{ g (I}_2\text{)} + 6.25 \text{ g (DCE)} - 0.1340 \text{ g}] / [0.1340 \text{ g}]$$

$$E\text{-Factor} = 47.33$$

$$PMI = \sum (\text{mass of stoichiometric reactants}) / [\text{Mass of product}]$$

$$PMI = \sum [(0.0916 \text{ g} + 0.0589 \text{ g} + 0.0761 \text{ g (I}_2\text{)} + 6.25 \text{ g (DCE)})] / [0.1340 \text{ g}]$$

$$PMI = 48.33$$

$$RME = \text{mass of product} / \sum (\text{mass of stoichiometric reactants}) \times 100$$

$$RME = [0.1340] / \sum [0.0916 \text{ g} + 0.0589 \text{ g}] \times 100$$

$$RME = 89.03 \%$$

$$AE = [MW \text{ of product}] \% \sum (MW \text{ of stoichiometric reactants}) \times 100$$

$$AE = [270.33] \% \sum [183.21 + 107.15 + 126.9(\text{I}_2) + 98.95 (\text{DCE})] \times 100$$

$$AE = 52.36 \%$$

$$CE = [\text{Amount of carbon in product}] / [\text{Total carbon present in reactants}] \times 100$$



$CE = [\text{no. of moles of product} \times \text{no. of carbons in product}] \times 100 / [\text{no. of moles} \times \text{no. of carbon} + (\text{no. of moles} \times \text{no. of carbon atoms})]$

$$CE = [0.495 \times 19] \times 100 / [0.5 \times 12 + 0.55 \times 7]$$

**CE = 95.48 %**

**Chemical yield (CY):**

$CY = [\text{Weight of product} \times \text{MW of starting material}] \times 100 / [\text{Weight of Starting material} \times \text{MW of Product}]$

$$CY = [0.1340 \text{ g} \times 183.21] \times 100 / [0.0916 \times 270.33]$$

**CY = 99.15 %**

**Mass intensity (MI):**

$MI = \sum \text{Weight (Total materials input)} / [\text{Weight of Product}]$

$$MI = \sum [(0.0916 \text{ g} + 0.0589 \text{ g} + 0.0761 \text{ g}) / [0.1340 \text{ g}]]$$

**MI = 1.69**

**Mass productivity (MP):**

$MP = 100 / MI$

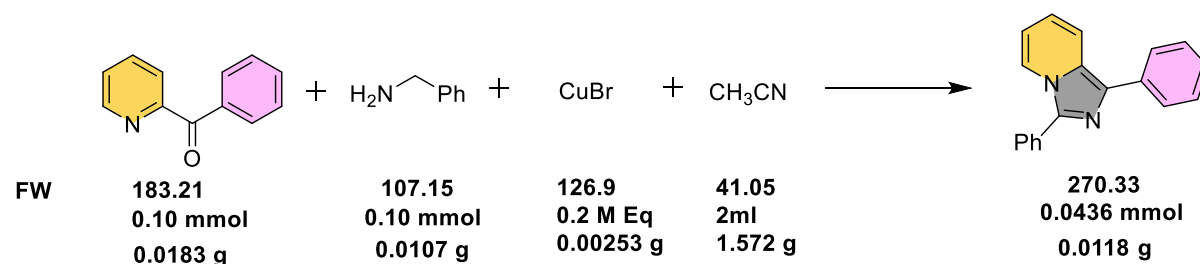
**MP = 59.17**

**Optimum efficiency (OE):**

$OE = [\text{RME}] \times 100 / \text{AE}$

**OE = 170.03 %**

**Calculation of Green chemistry metrics (GCM) for compound (3a) by CuBr (reported ones).**



$E\text{-factor} = [\text{Total mass of raw materials minus the total mass of product}] / \text{mass of product}$

$$E\text{-factor} = [0.0183 \text{ g} + 0.0107 \text{ g} + 0.00253 \text{ g (CuBr)} + 1.572 \text{ g (CH}_3\text{CN)} - 0.0118 \text{ g}] / [0.0118 \text{ g}]$$

**E-Factor = 135**

$$PMI = \sum(\text{mass of stoichiometric reactants}) / [\text{Mass of product}]$$

$$PMI = \sum[(0.0183 \text{ g} + 0.0107 \text{ g} + 0.00253 \text{ g (CuBr)} + 1.572 \text{ g (CH}_3\text{CN)}] / [0.0118 \text{ g}]$$

**PMI = 136**

$$RME = \text{mass of product} / \sum(\text{mass of stoichiometric reactants}) \times 100$$

$$RME = [0.0118 \text{ g}] / \sum[0.0183 \text{ g} + 0.0107 \text{ g}] \times 100$$

**RME = 40.68 %**

$$AE = [\text{MW of product}] \% \sum(\text{MW of stoichiometric reactants}) \times 100$$

$$AE = [270.33 \text{ g/mol}] \% \sum[183.21 \text{ g/mol} + 107.15 \text{ g/mol} + 126.9 \text{ g/mol (CuBr)} + 41.05 \text{ g/mol (CH}_3\text{CN)}] \times 100$$

**AE = 59 %**

$$CE = [\text{Amount of carbon in product}] / [\text{Total carbon present in reactants}] \times 100$$

$$CE = [\text{no. of moles of product} \times \text{no. of carbons in product}] \times 100 / [\text{no. of moles} \times \text{no. of carbon} + (\text{no. of moles} \times \text{no. of carbon atoms})]$$

$$CE = [0.0436 \times 19] \times 100 / [0.1 \times 12 + 0.10 \times 7]$$

**CE = 43.6 %**

$$CY = [\text{Weight of product} \times \text{MW of starting material}] \times 100 / [\text{Weight of Starting material} \times \text{MW of Product}]$$

$$CY = [0.0118 \text{ g} \times 183.21 \text{ g/mol}] \times 100 / [0.0183 \text{ g} \times 270.33 \text{ g/mol}]$$

**CY = 43.72 %**

$$MI = \sum \text{Weight (Total materials input)} / [\text{Weight of Product}]$$

$$MI = \sum[(0.0183 \text{ g} + 0.0107 \text{ g} + 0.00253 \text{ g} + 1.572 \text{ g})] / [0.0118 \text{ g}]$$

**MI = 2.67**

$$MP = 100 / MI$$

**MP = 37.45 %**

$$OE = [RME] \times 100 / AE$$

**OE = 69 %**

Table S2. Comparison table for the synthesis of 1,3-diarylated Imidazo[1,5a] pyridines *compound 3a* by deploying other reported catalysts.

Catalyst	T (°C)	Solvent	Time	Catalyst Quantity	Base	Yield	Ref.
Cu(OAc) <sub>2</sub> .H <sub>2</sub> O	110	DMF	8 h	0.15 Eq	-	93	2
I <sub>2</sub>	Reflux	DCE	6 h	0.0006 Eq	NaOAc (0.0015 Eq)	99	3
CuBr	80	CH <sub>3</sub> CN	24 h	20 mol %	-	42	4
g-C <sub>3</sub> N <sub>4</sub>	RT	-	10 h	g-C <sub>3</sub> N <sub>4</sub> (50 mg)	-	95	Present work

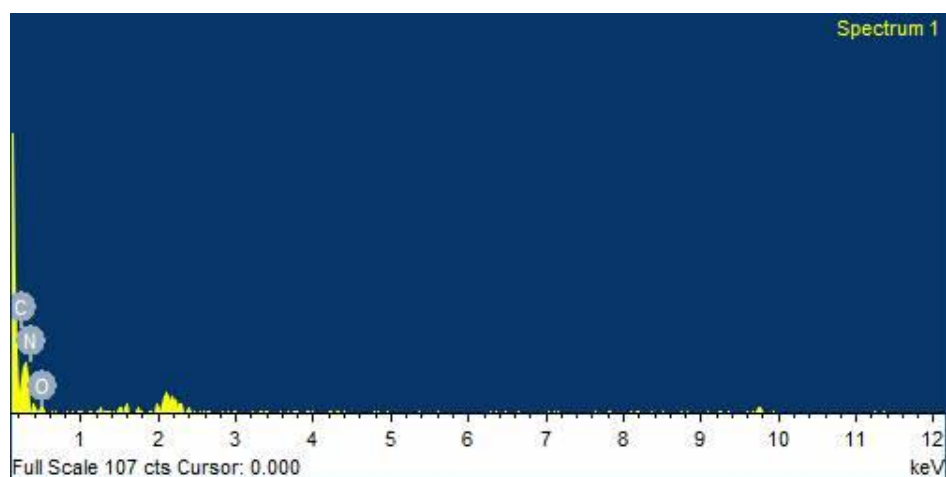
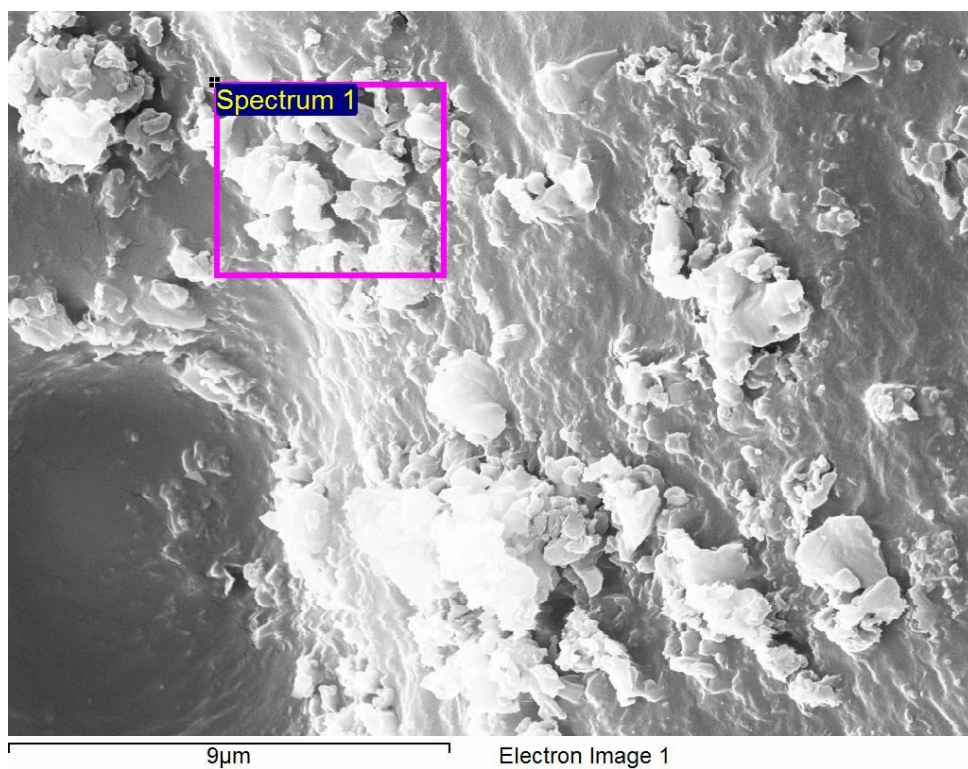


Figure S10. SEM-EDX spectrum of reused  $g\text{-C}_3\text{N}_4$  (C = 31.72 %, N = 48.06 %, O = 20.22 %).

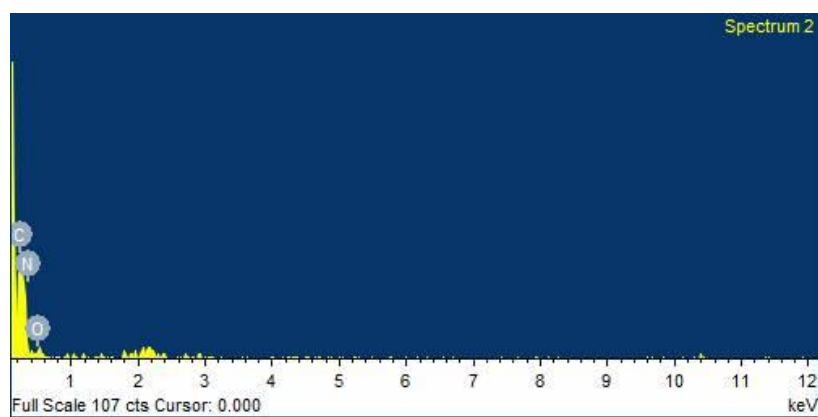
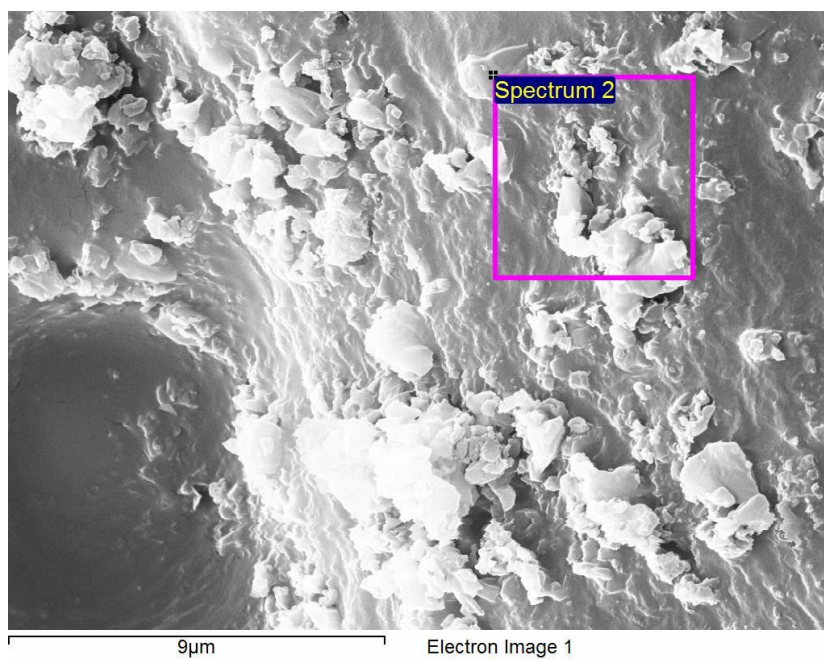


Figure S11. SEM-EDX spectrum of reused g-C<sub>3</sub>N<sub>4</sub> (C = 56.16 %, N = 28.45 %, O = 15.39 %).

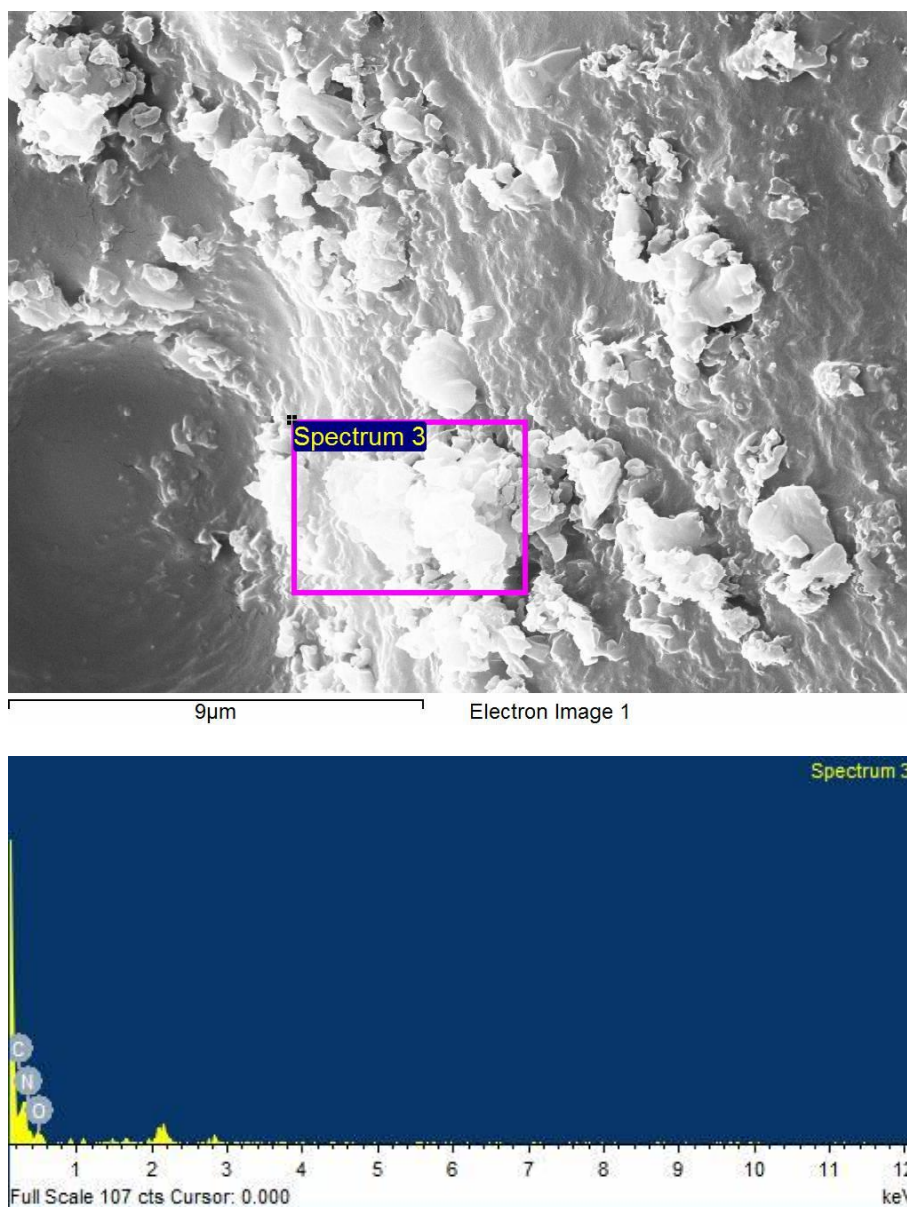


Figure S12. SEM-EDX spectrum of reused g-C<sub>3</sub>N<sub>4</sub> (C = 27.04 %, N = 44.68 %, O = 28.28 %).

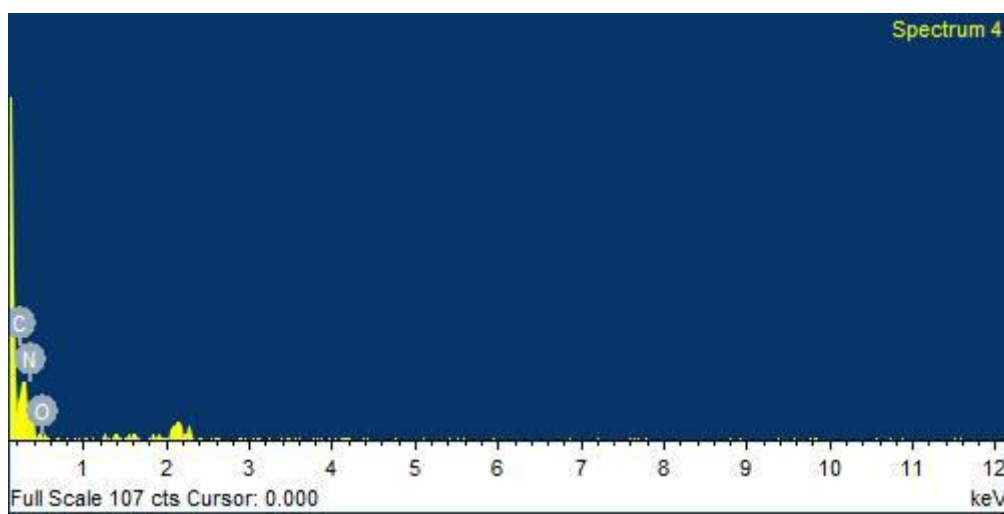
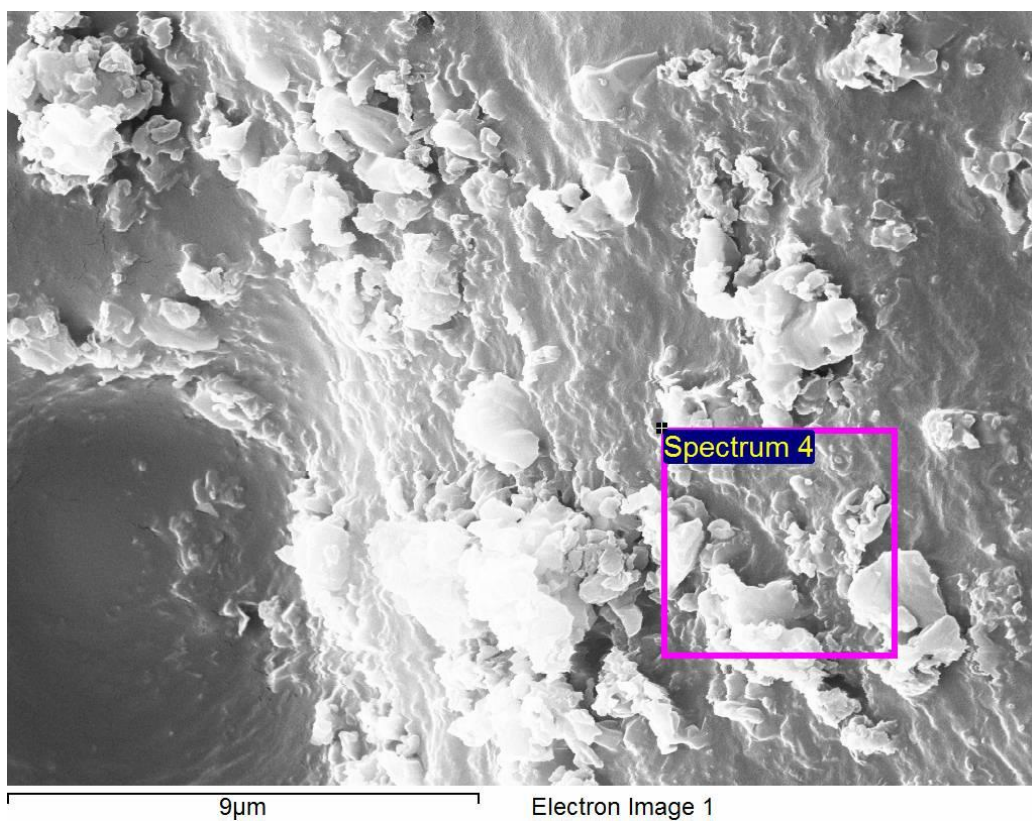
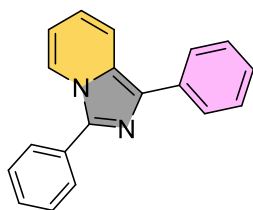


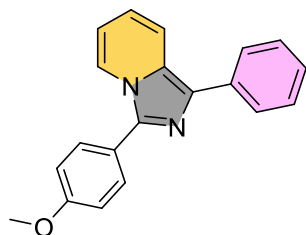
Figure S13. SEM-EDX spectrum of reused g-C<sub>3</sub>N<sub>4</sub> (C = 30.13 %, N = 51.71 %, O = 18.15 %).

### Synthesis and Structural characterization of Compounds 3a to Compound 3v



In a 20 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *benzyl amine* (0.321, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3 × 10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3a**).

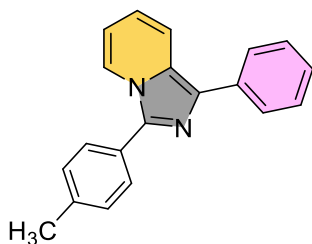
**1,3-Diphenylimidazo[1,5-a]pyridine (Compound 3a)<sup>2</sup>**: LCMS 97.98%(254 nm) (ES<sup>+</sup>) calcd for C<sub>19</sub>H<sub>14</sub>N<sub>2</sub><sup>+</sup>, 270.11; found, 271.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.21 (d, *J* = 7.2 Hz, 1H), 7.94 (d, *J* = 7.8 Hz, 2H), 7.87 (d, *J* = 9.6 Hz, 1H), 7.81 (d, *J* = 8.4 Hz, 2H), 7.53 (d, *J* = 8.4 Hz, 2H), 7.49 (t, *J* = 7.2 Hz, 2H), 7.33 (t, *J* = 7.2 Hz, 1H), 6.82-6.84 (m, 1H), 6.63 (t, *J* = 6.6 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 138.14, 134.97, 132.02, 130.18, 129.03, 128.83, 128.73, 128.34, 126.85, 126.55, 121.78, 119.71, 119.18, 113.24.



In a 20 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *4-methoxy benzyl amine* (0.411 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3 × 10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3b**).

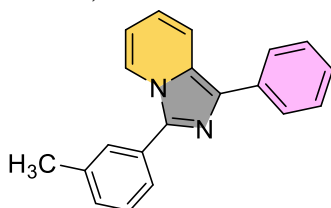
**3-(4-Methoxyphenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3b)<sup>2</sup>**: LCMS 93.66%(220 nm) (ES<sup>+</sup>) calcd for C<sub>20</sub>H<sub>16</sub>N<sub>2</sub>O<sup>+</sup>, 300.12; found, 301.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.18 (d, *J* = 7.2 Hz, 1H), 7.95 (d, *J* = 7.2 Hz, 2H), 7.84 (d, *J* = 9 Hz, 1H), 7.77 (d, *J* = 9.0 Hz, 2H), 7.48 (t, *J* = 7.8 Hz, 2H), 7.31 (t, *J* = 7.2 Hz, 1H), 7.08 (d, *J* = 8.4 Hz, 2H), 6.79-6.77 (m, 1H), 6.57 (t, *J* = 6.6 Hz, 1H), 3.90 (s, 3H). <sup>13</sup>C NMR(151MHz,CDCl<sub>3</sub>)δ160.05,138.15,135.05,131.65,129.81,128.71,127.36,126.77,126.45,122.62,121.76,119.44,119.13,114.47,113 03, 55.43.





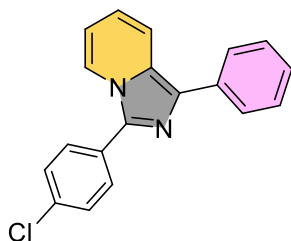
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *4-methylbenzyl amine* (0.363 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3c**).

**1-Phenyl-3-p-tolylimidazo[1,5-a]pyridine (Compound 3c)<sup>2</sup>**. LCMS 97.14%(254 nm) (ES+) calcd for C<sub>20</sub>H<sub>16</sub>N<sub>2</sub><sup>+</sup>, 284.13; found, 285.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.22 (*d*, *J* = 7.2 Hz, 1H), 7.96 (*d*, *J* = 7.8 Hz, 2H), 7.85 (*d*, *J* = 9 Hz, 1H), 7.75 (*d*, *J* = 7.8 Hz, 2H), 7.49 (*t*, *J* = 7.8 Hz, 2H), 7.36 (*d*, *J* = 7.8 Hz, 2H), 7.32 (*t*, *J* = 7.2 Hz, 1H), 6.77-6.80 (*m*, 1H), 6.57 (*t*, *J* = 6.6 Hz, 1H), 2.46 (*s*, 3H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 138.84, 138.32, 135.05, 131.82, 129.70, 128.72, 128.25, 127.55, 127.28, 126.81, 126.48, 121.85, 119.56, 119.12, 113.09, 21.46.



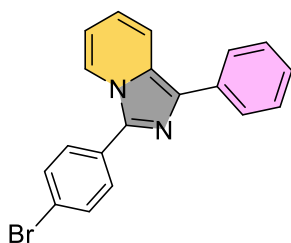
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *3-methylbenzyl amine* (0.363 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3d**).

**1-Phenyl-3-m-tolylimidazo[1,5-a]pyridine (Compound 3d)<sup>2</sup>**. Mass LCMS 100%(210 nm) (ES+) calcd for C<sub>20</sub>H<sub>16</sub>N<sub>2</sub><sup>+</sup>, 284.13; found, 285.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.25 (*d*, *J* = 7.2 Hz, 1H), 7.97 (*d*, *J* = 7.8 Hz, 2H), 7.85 (*d*, *J* = 9 Hz, 1H), 7.70(*s*, 1H), 7.64 (*d*, *J* = 7.2 Hz, 1H), 7.49 (*t*, *J* = 7.2 Hz, 2H), 7.44 (*t*, *J* = 7.8 Hz, 1H), 7.33 (*t*, *J* = 7.2 Hz, 1H), 7.28-7.30 (*m*, 1H), 6.78-6.80 (*m*, 1H), 6.58 (*t*, *J* = 6.6 Hz, 1H), 2.48 (*s*, 3H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 138.92, 138.32, 135.02, 131.92, 130.05, 129.65, 129.24, 128.8, 127.62, 126.83, 126.51, 125.11, 121.89, 119.64, 119.12, 113.14, 21.52.



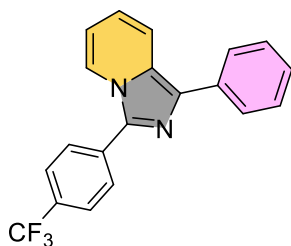
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183g, 1 mmol), *4-chloro benzyl amine* (0.424g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3e**).

**3-(4-Chlorophenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3e)**<sup>2</sup>. LCMS 100% (210 nm) (ES<sup>+</sup>) calcd for C<sub>19</sub>H<sub>13</sub>ClN<sub>2</sub><sup>+</sup>, 304.07; found, 305.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.21 (*d*, *J* = 7.2 Hz, 1H), 7.94 (*d*, *J* = 7.8 Hz, 2H), 7.87 (*d*, *J* = 9.6 Hz, 1H), 7.81 (*d*, *J* = 8.4 Hz, 2H), 7.53 (*d*, *J* = 8.4 Hz, 2H), 7.49 (*t*, *J* = 7.2 Hz, 2H), 7.33 (*t*, *J* = 7.2 Hz, 1H), 6.82-6.84 (*m*, 1H), 6.63 (*t*, *J* = 6.6 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 136.93, 134.76, 132.33, 129.4, 129.18, 128.72, 128.47, 127.90, 126.80, 121.45, 119.81, 119.28, 113.61.



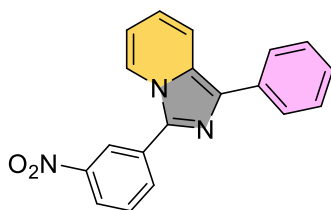
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *4-bromo benzyl amine* (0.558 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3f**).

**3-(4-Bromophenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3f)**<sup>7</sup>. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.21 (*d*, *J* = 7.2 Hz, 1H), 7.94 (*d*, *J* = 7.8 Hz, 2H), 7.87 (*d*, *J* = 9 Hz, 1H), 7.75 (*d*, *J* = 8.4 Hz, 2H), 7.69 (*d*, *J* = 9.0 Hz, 2H), 7.49 (*t*, *J* = 7.8 Hz, 2H), 7.33 (*t*, *J* = 7.2 Hz, 1H), 6.82-6.84 (*m*, 1H), 6.63 (*t*, *J* = 7.2 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 136.93, 134.74, 132.38, 132.23, 129.69, 129.38, 129.11, 128.78, 128.39, 127.9, 126.8, 122.83, 121.53, 119.89, 119.29, 113.65.



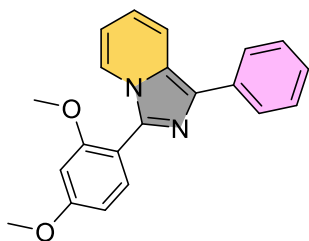
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183g, 1 mmol), *4-trifluoromethyl* (0.525 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3g**).

**1-Phenyl-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine (Compound 3g)<sup>2</sup>.** <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.28 (d, J = 7.8 Hz, 1H), 8.02 (d, J = 8.4 Hz, 2H), 7.95 (d, J = 7.2 Hz, 2H), 7.90 (d, J = 9 Hz, 1H), 7.81 (d, J = 7.8 Hz, 2H), 7.50 (t, J = 7.8 Hz, 2H), 7.35 (t, J = 7.8 Hz, 1H), 6.86 (dd, J = 9.6, 6.6 Hz, 1H), 6.67 (t, J = 7.8 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 136.48, 134.64, 133.71, 132.86, 130.37 (q, J = 32.6 Hz), 128.82, 128.32, 128.25, 126.87, 126.00 (q, J = 3.7 Hz), 124.02 (d, J = 272.1 Hz), 121.49, 120.21, 119.37, 122.15, 119.65, 113.95.



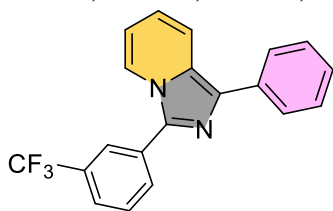
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *3-nitrobenzyl amine* (0.456 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow orange solid*). (**Compound 3h**)<sup>7</sup>.

**3-(3-nitrophenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3h).** <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.75 (s, 1H), 8.26-8.31 (m, 3H), 7.91-7.95 (m, 3H), 7.74 (t, J = 7.8 Hz, 1H), 7.51 (t, J = 6.6 Hz, 2H), 7.36 (t, J = 6.6 Hz, 1H), 6.91 (t, J = 7.2 Hz, 1H), 6.74 (s, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 148.66, 133.96, 131.93, 130.21, 128.87, 127.03, 126.87, 123.17, 122.39, 121.25, 120.51, 119.51, 114.47.



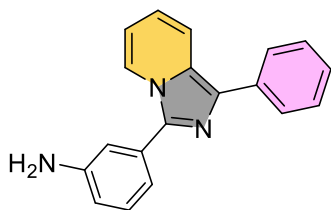
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), 2, 4 dimethoxy benzyl amine (0.501 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Pale yellow solid*). (**Compound 3i**).

**3-(3-nitrophenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3i).** <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.97 (d, J = 8.4 Hz, 2H), 7.86 (d, J = 9 Hz, 1H), 7.60 (t, J = 7.2 Hz, 2H), 7.47 (t, J = 7.8 Hz, 2H), 7.29 (t, J = 7.2 Hz, 1H), 6.79-6.83 (m, 1H), 6.68 (dd, J = 1.8 Hz, 1H), 6.62 (d, J = 2.4 Hz, 1H), 6.53 (t, J = 7.2 Hz, 1H), 3.90 (s, 1H), 3.81 (s, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 162.18, 158.64, 136.19, 135.27, 133.57, 131.16, 128.63, 127.24, 126.72, 126.19, 123.46, 119.41, 118.61, 111.8, 105.30, 98.79, 55.58.



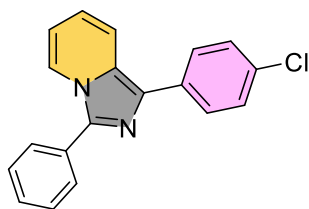
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), 3-trifluoromethyl benzyl amine (0.525, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3j**).

**1-Phenyl-3-(3-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine (Compound 3j)**<sup>3</sup>. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.24 (d, J = 7.2 Hz, 1H), 8.16 (s, 1H), 8.06 (d, J = 7.2 Hz, 1H), 7.95 (d, J = 7.8 Hz, 2H), 7.89 (d, J = 9 Hz, 1H), 7.72 (d, J = 7.8 Hz, 1H), 7.68 (t, J = 7.8 Hz, 1H), 7.51 (t, J = 7.8 Hz, 2H), 7.35 (t, J = 7.8 Hz, 1H), 6.85-6.87 (m, 1H), 6.67 (t, J = 6.6 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 136.45, 134.64, 132.66, 131.68 (t, J = 32.7 Hz), 131.24, 131.07, 129.59, 128.83, 128.14, 126.87, 126.84, 125.32 (q, J = 3.9 Hz), 125.07 (q, J = 3.7 Hz), 123.93 (d, J = 272.8 Hz), 121.33, 121.23, 120.15, 119.36, 113.97.



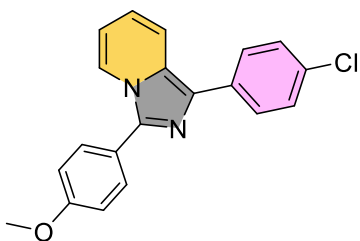
In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *3-amino benzyl amine* (0.366 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3k**).

**3-(1-phenylimidazo[1,5-a]pyridin-3-yl)aniline (Compound 3k)**. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.27 (d, J = 7.2 Hz, 1H), 7.95 (d, J = 7.2 Hz, 2H), 7.84 (d, J = 9.6 Hz, 1H), 7.48 (t, J = 7.8 Hz, 2H), 7.31 (t, J = 7.8 Hz, 2H), 7.18 (t, J = 7.2 Hz, 2H), 6.76-6.80 (m, 2H), 6.56 (t, J = 6.6 Hz, 1H), 3.85 (s, 2H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 147.20, 138.36, 135.00, 131.75, 131.01, 129.81, 128.72, 127.61, 126.82, 126.49, 122.16, 119.65, 119.05, 117.98, 115.62, 115.13, 113.04.



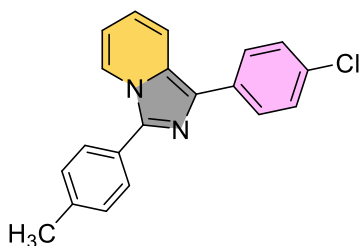
In a 30 mL glass vial, *(4-chlorophenyl)(pyridin-2-yl)methanone* (0.217 g, 1 mmol), *benzyl amine* (0.321 g, 3 mmol), *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3l**).

**1-(4-Chlorophenyl)-3-phenylimidazo[1,5-a]pyridine (Compound 3l)**<sup>3</sup>. LCMS 98.26% (254 nm) (ES<sup>+</sup>) calcd for C<sub>19</sub>H<sub>13</sub>ClN<sub>2</sub><sup>+</sup>, 304.07; found, 305.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 8.26 (s, 1H), 7.89 (d, J = 5.4 Hz, 2H), 7.82-7.84 (m, 3H), 7.56 (s, 2H), 7.45-7.48 (m, 3H), 6.83 (s, 1H), 6.61 (s, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 138.32, 133.52, 132.10, 130.77, 129.99, 129.07, 128.97, 128.86, 128.33, 127.86, 127.75, 121.91, 120.14, 118.87, 113.31.



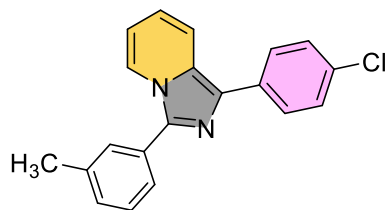
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 4-methoxy benzylamine (0.411 g, 3 mmol),  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3m**).

**1-(4-chlorophenyl)-3-(4-methoxyphenyl)imidazo[1,5-a]pyridine (Compound 3m)**. LCMS 97.25% (210 nm) ( $\text{ES}^+$ ) calcd for  $\text{C}_{20}\text{H}_{15}\text{ClN}_2\text{O}^+$ , 334.08; found, 335.2 ( $\text{MH}^+$ ).  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.18 (*d*,  $J = 7.2$  Hz, 1H), 7.89 (*d*,  $J = 8.4$  Hz, 2H), 7.75-7.79 (*m*, 3H), 7.44 (*d*,  $J = 8.4$  Hz, 2H), 7.08 (*d*,  $J = 8.4$  Hz, 2H), 6.79-6.82 (*m*, 1H), 6.58 (*t*,  $J = 7.2$  Hz, 1H), 3.90 (*s*, 3H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  160.17, 138.35, 133.63, 131.99, 130.42, 129.81, 128.84, 127.82, 127.46, 122.44, 121.90, 119.88, 118.84, 114.53, 113.10, 55.43.



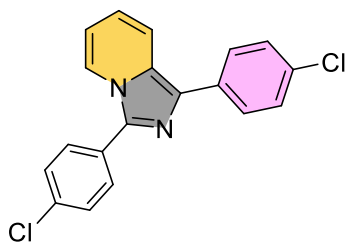
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 4-methyl benzylamine (0.363 g, 3 mmol),  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3n**).

**1-(4-chlorophenyl)-3-(p-tolyl)imidazo[1,5-a]pyridine (Compound 3n)**. LCMS 100% (210 nm) ( $\text{ES}^+$ ) calcd for  $\text{C}_{20}\text{H}_{15}\text{ClN}_2^+$ , 318.09; found, 319.2 ( $\text{MH}^+$ ).  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.22 (*d*,  $J = 7.2$  Hz, 1H), 7.89 (*d*,  $J = 8.4$  Hz, 2H), 7.78 (*d*,  $J = 9$  Hz, 1H), 7.72 (*d*,  $J = 7.8$  Hz, 2H), 7.44 (*d*,  $J = 8.4$  Hz, 2H), 7.36 (*d*,  $J = 7.8$  Hz, 2H), 6.79-6.82 (*m*, 1H), 6.58 (*t*,  $J = 7.2$  Hz, 1H), 2.46 (*s*, 3H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  139.01, 138.50, 133.61, 132.01, 130.57, 129.74, 128.84, 128.24, 127.85, 127.62, 127.10, 121.99, 119.99, 118.83, 113.15, 21.44.



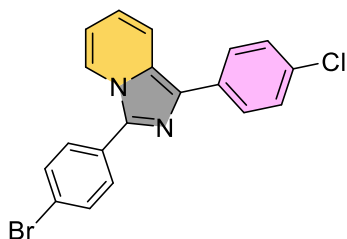
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 3-methyl benzyl amine (0.363 g, 3 mmol),  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3o**).

**1-(4-chlorophenyl)-3-(m-tolyl)imidazo[1,5-a]pyridine (Compound 3o)**. LCMS 100% (210 nm) (ES+) calcd for  $\text{C}_{20}\text{H}_{15}\text{ClN}_2^+$ , 318.09; found, 319.2 (MH<sup>+</sup>).  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.25 (d,  $J = 6.6$  Hz, 1H), 7.89 (d,  $J = 8.4$  Hz, 2H), 7.79 (d,  $J = 9.0$  Hz, 1H), 7.67 (s, 1H), 7.62 (d,  $J = 7.2$  Hz, 1H), 7.43-7.45 (m, 3H), 7.29 (t,  $J = 7.2$  Hz, 1H), 6.81-6.84 (m, 1H), 6.60 (t,  $J = 6.6$  Hz, 1H), 2.47 (s, 3H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  138.98, 138.51, 133.55, 132.49, 132.06, 130.66, 129.86, 129.21, 128.84, 127.87, 127.68, 125.12, 122.03, 120.07, 118.83, 113.20, 21.50.



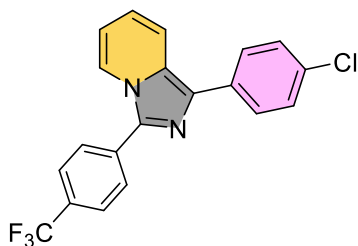
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 4-chloro benzyl amine (0.424 g, 3 mmol),  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3p**).

**1,3-bis(4-chlorophenyl)imidazo[1,5-a]pyridine (Compound 3p)**. LCMS 197.90% (254 nm) (ES+) calcd for  $\text{C}_{19}\text{H}_{12}\text{Cl}_2\text{N}_2^+$ , 338.03; found, 339.2 (MH<sup>+</sup>).  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.21 (d,  $J = 7.2$  Hz, 1H), 7.87 (d,  $J = 9.0$  Hz, 2H), 7.79-7.82 (m, 3H), 7.53 (d,  $J = 8.4$  Hz, 2H), 7.45 (d,  $J = 8.4$  Hz, 2H), 6.85-6.87 (m, 1H), 6.65 (t,  $J = 7.2$  Hz, 1H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  137.12, 134.85, 133.31, 132.31, 131.09, 129.49, 129.35, 128.92, 128.49, 127.97, 127.88, 121.69, 120.30, 119.00, 113.68.



In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 4-bromo benzyl amine (0.558 g, 3 mmol) and  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3q**).

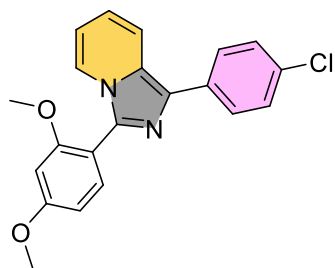
**3-(4-bromophenyl)-1-(4-chlorophenyl)imidazo[1,5-a]pyridine (Compound 3q).**  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.21 (d,  $J = 7.2$  Hz, 1H), 7.87 (d,  $J = 8.4$  Hz, 2H), 7.81 (d,  $J = 9.6$  Hz, 1H), 7.73 (d,  $J = 8.4$  Hz, 2H), 7.69 (d,  $J = 7.8$  Hz, 2H), 7.45 (d,  $J = 7.8$  Hz, 2H), 6.86 (t,  $J = 7.2$  Hz, 1H), 6.64 (t,  $J = 6.6$  Hz, 1H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  137.14, 133.29, 132.29, 131.14, 129.69, 128.92, 128.02, 127.88, 123.02, 121.68, 120.33, 119.01, 113.73.



In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 4-trifluoromethyl benzylamine (0.525 g, 3 mmol) and  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3r**).

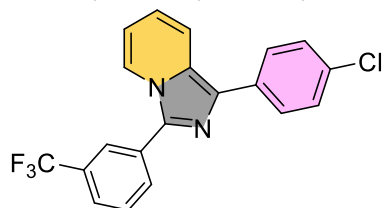
**1-(4-chlorophenyl)-3-(4-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine (Compound 3r).**  $^1\text{H NMR}$  (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.28 (d,  $J = 7.2$  Hz, 1H), 8.00 (d,  $J = 7.8$  Hz, 2H), 7.88 (d,  $J = 8.4$  Hz, 2H), 7.81-7.85 (m, 3H), 7.46 (d,  $J = 7.8$  Hz, 2H), 6.89 (t,  $J = 6.6$  Hz, 1H), 6.69 (t,  $J = 6.6$  Hz, 1H).  $^{13}\text{C NMR}$  (151 MHz,  $\text{CDCl}_3$ )  $\delta$  136.65, 133.51, 133.16, 132.47, 131.57, 130.55 (q,  $J = 33.2$  Hz), 128.95, 128.36, 128.28, 127.92, 126.05 (q,  $J = 3.9$  Hz), 123.97 (d,  $J = 272.1$  Hz), 121.62, 120.64, 119.08, 114.03.





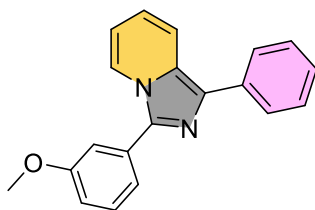
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 2,4-dimethoxy benzylamine (0.501 g, 3 mmol) and  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*) (**Compound 3s**).

**1-(4-chlorophenyl)-3-(2,4-dimethoxyphenyl)imidazo[1,5-a]pyridine (Compound 3s).**  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  7.90 (d,  $J = 5.4$  Hz, 2H), 7.80 (d,  $J = 7.8$  Hz, 1H), 7.59 (s, 2H), 7.43 (s, 2H), 6.83 (s, 1H), 6.68 (d,  $J = 5.4$  Hz, 1H), 6.62 (s, 1H), 6.55 (s, 1H), 3.90 (s, 3H), 3.81 (s, 3H).  $^{13}\text{C}$  NMR (151 MHz,  $\text{CDCl}_3$ )  $\delta$  162.28, 158.65, 136.42, 133.87, 133.48, 131.68, 129.97, 128.75, 127.75, 127.35, 123.58, 119.83, 118.30, 111.97, 105.36, 98.82, 55.58.



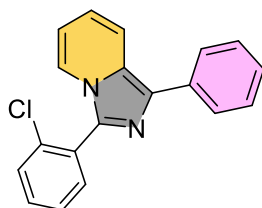
In a 30 mL glass vial, (4-chlorophenyl)(pyridin-2-yl)methanone (0.217 g, 1 mmol), 3-trifluoromethyl benzyl amine (0.525 g, 3 mmol) and  $g\text{-C}_3\text{N}_4$  (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc ( $3 \times 10$  mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*) (**Compound 3t**).

**1-(4-chlorophenyl)-3-(3-(trifluoromethyl)phenyl)imidazo[1,5-a]pyridine (Compound 3t).**  $^1\text{H}$  NMR (600 MHz,  $\text{CDCl}_3$ )  $\delta$  8.24 (d,  $J = 7.2$  Hz, 1H), 8.14 (s, 1H), 8.05 (d,  $J = 7.2$  Hz, 1H), 7.89 (d,  $J = 8.4$  Hz, 2H), 7.84 (d,  $J = 9$  Hz, 1H), 7.73 (d,  $J = 7.8$  Hz, 1H), 7.69 (t,  $J = 7.2$  Hz, 1H), 7.46 (d,  $J = 8.4$  Hz, 2H), 6.88-6.90 (m, 1H), 6.69 (t,  $J = 7.2$  Hz, 1H).  $^{13}\text{C}$  NMR (151 MHz,  $\text{CDCl}_3$ )  $\delta$  136.63, 133.19, 132.44, 131.54 (t,  $J = 32.6$  Hz), 131.24, 130.90, 129.63, 128.95, 128.20, 127.91, 126.60, 125.46 (q,  $J = 3.9$  Hz), 125.07 (q,  $J = 3.7$  Hz), 123.89 (d,  $J = 272.7$  Hz), 121.47, 121.18, 120.55, 119.07, 114.02.



In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *3-methoxybenzylamine* (0.411 g, 3 mmol) and *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3u**).

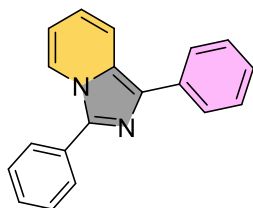
**3-(2-methoxyphenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3u)**<sup>7</sup>. LCMS 96.47% (254 nm) (ES<sup>+</sup>) calcd for C<sub>20</sub>H<sub>16</sub>N<sub>2</sub>O<sup>+</sup>, 300.12; found, 301.2 (MH<sup>+</sup>). <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ 8.49 (*d*, *J* = 7.2 Hz, 1H), 8.01 (*d*, *J* = 9.6 Hz, 1H), 7.95 (*d*, *J* = 7.6 Hz, 2H), 7.51-7.44 (*m*, 4H), 7.39 (*s*, 1H), 7.30 (*t*, *J* = 7.6 Hz, 1H), 7.09 (*d*, *J* = 7.2 Hz, 1H), 6.98 (*t*, *J* = 6.0 Hz, 1H), 6.79 (*d*, *J* = 6.8 Hz, 1H), 3.86 (*s*, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>) δ 160.14, 137.96, 134.88, 131.93, 131.33, 130.04, 128.76, 127.76, 126.87, 126.62, 121.97, 120.42, 119.81, 119.18, 114.98, 113.78, 113.32, 55.51.



In a 30 mL glass vial, *phenyl(pyridin-2-yl)methanone* (0.183 g, 1 mmol), *2-chlorobenzylamine* (0.424 g, 3 mmol) and *g-C<sub>3</sub>N<sub>4</sub>* (50 mg) were added at room temperature (RT). The reaction mixture (RM) was stirred under visible light irradiation (60 W) for 10 h. After the reaction completion, the resulting reaction mass was extracted with EtOAc (3×10 mL) and dried with sodium sulphate. The obtained filtrate was concentrated under reduced pressure. The obtained crude material was further purified by column chromatography (90-10% hexane:ethyl acetate) to afford the desired product (*Yellow solid*). (**Compound 3v**).

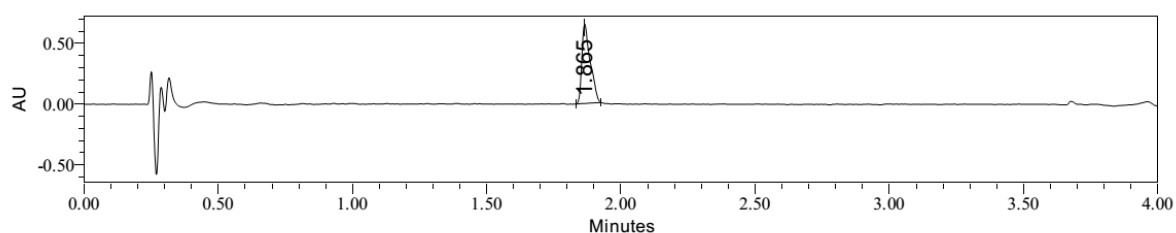
**3-(2-chlorophenyl)-1-phenylimidazo[1,5-a]pyridine (Compound 3v)**<sup>7</sup>. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>) δ 7.98 (*d*, *J* = 5.2 Hz, 2H), 7.91 (*d*, *J* = 9 Hz, 1H), 7.70 (*d*, *J* = 6.6 Hz, 1H), 7.62 (*d*, *J* = 7.2 Hz, 1H), 7.58 (*d*, *J* = 7.2 Hz, 1H), 7.44-7.50 (*m*, 4H), 7.32 (*t*, *J* = 7.2 Hz, 1H), 6.87 (*t*, *J* = 7.2 Hz, 1H), 6.63 (*t*, *J* = 6.6 Hz, 1H). <sup>13</sup>C NMR (151 MHz, CDCl<sub>3</sub>) δ 135.71, 134.92, 134.40, 133.40, 132.93, 131.65, 130.98, 130.86, 129.96, 129.35, 128.72, 128.17, 127.29, 127.27, 126.76, 126.53, 124.63, 122.62, 119.94, 118.89, 112.86.

**Copies of LC-MS, <sup>1</sup>H NMR and <sup>13</sup>C NMR**

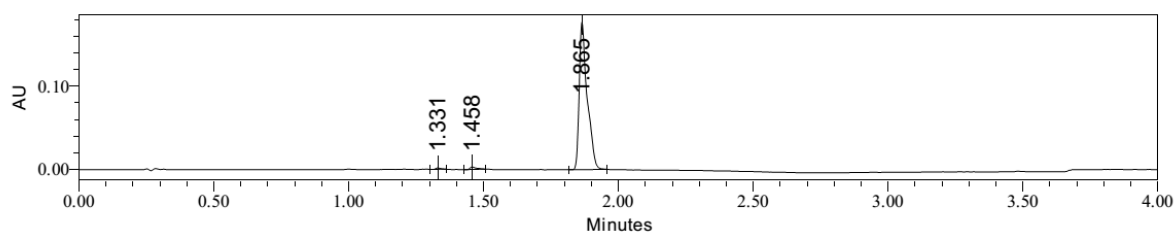


Compound (3a)

**LC-MS of Compound 3a**



Channel Name 210.0nm; Channel PDA Spectrum

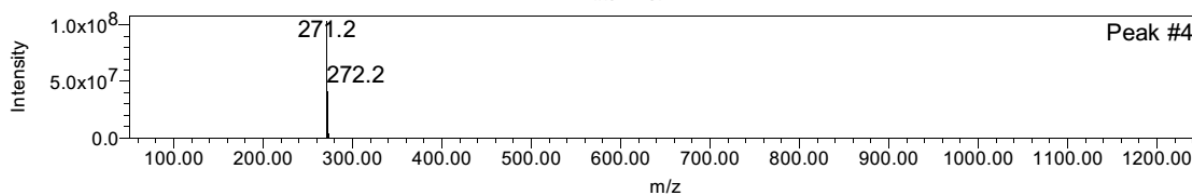


Channel Name 254.0nm; Channel PDA Spectrum

**Peak Results  
Channel: PDA Spectrum**

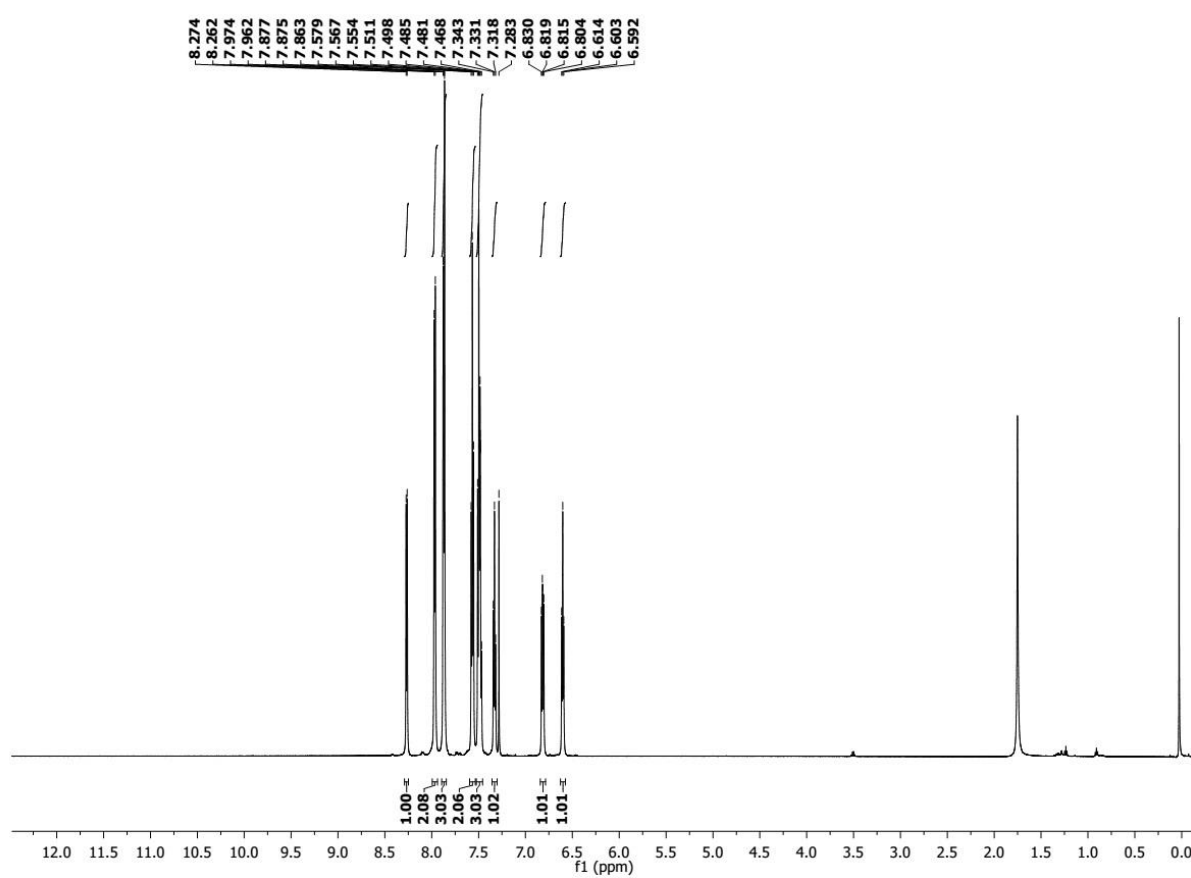
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.331		1624	2266	0.61	PDA Spectrum	254.0nm
2	1.458		3019	5248	1.41	PDA Spectrum	254.0nm
3	1.865		653321	1419185	100.00	PDA Spectrum	210.0nm
4	1.865		176653	363634	97.98	PDA Spectrum	254.0nm

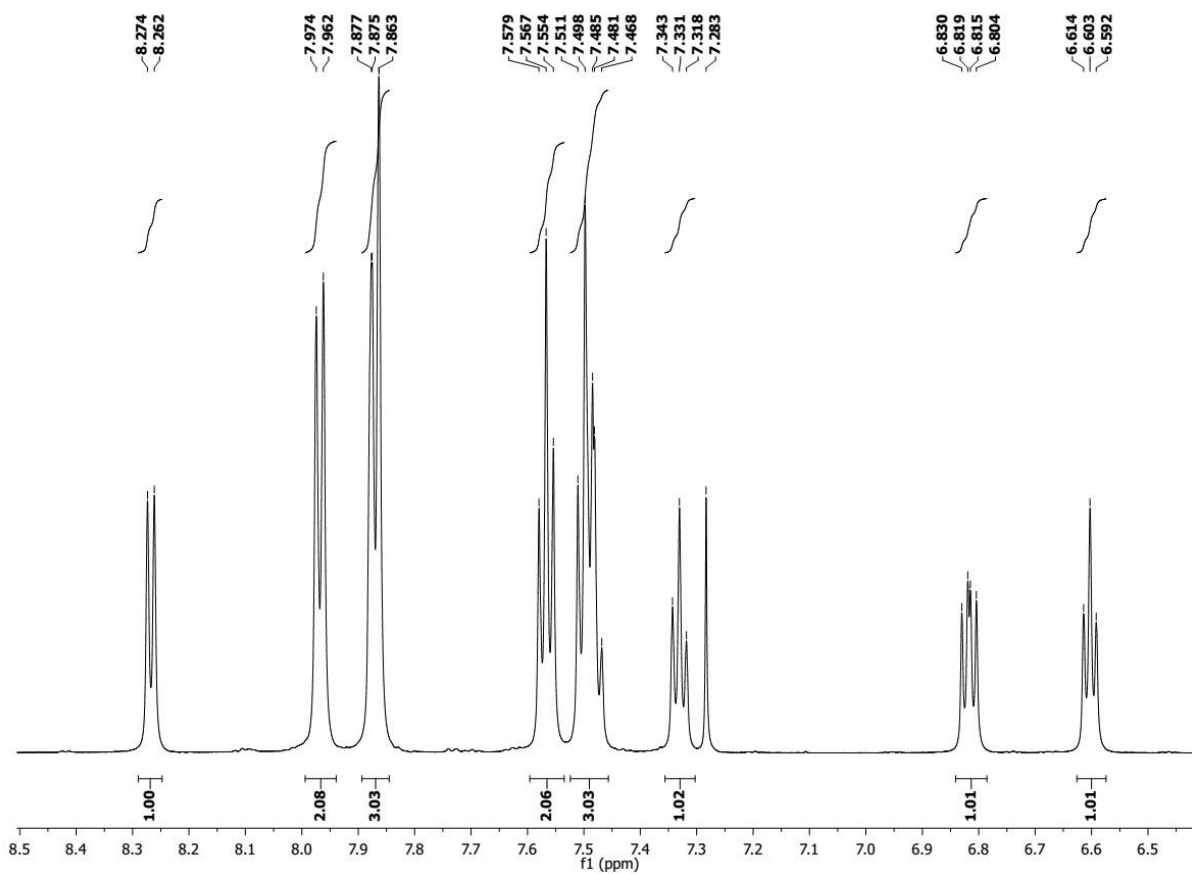
**Match Plot**



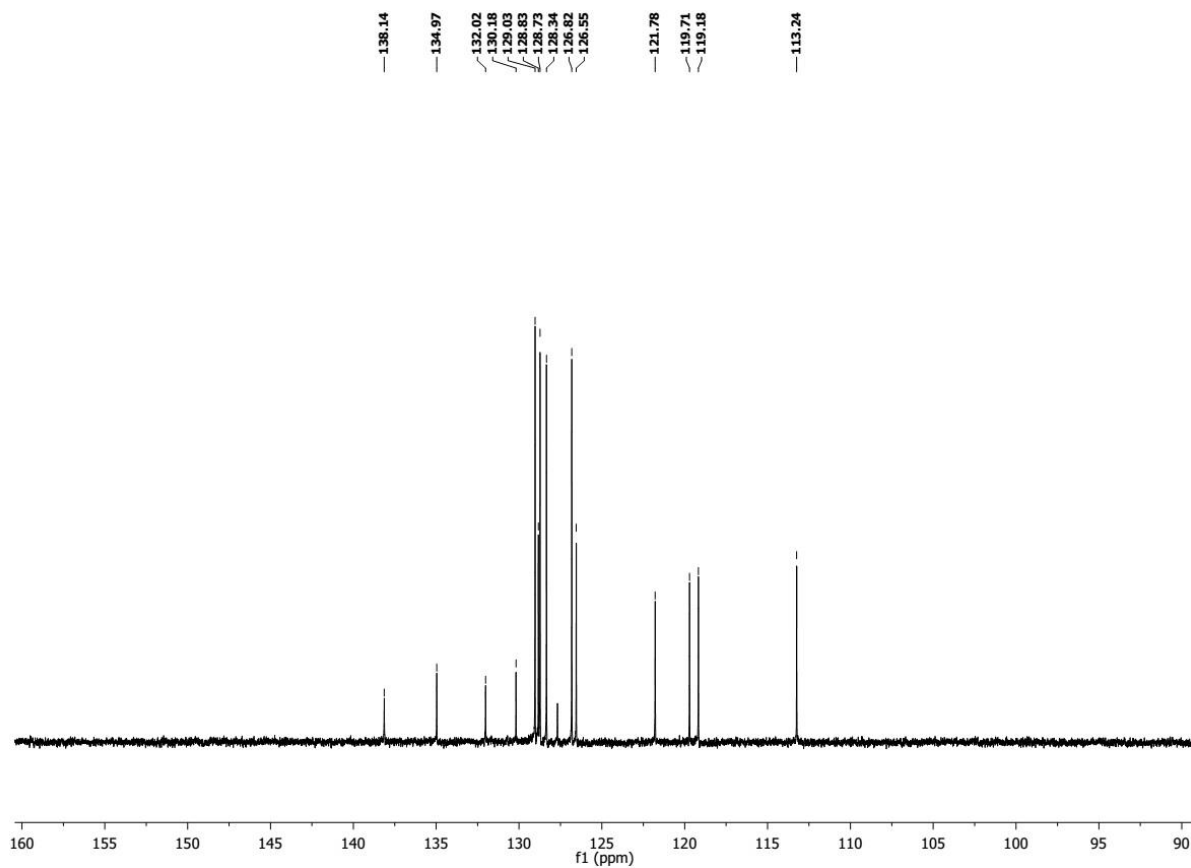
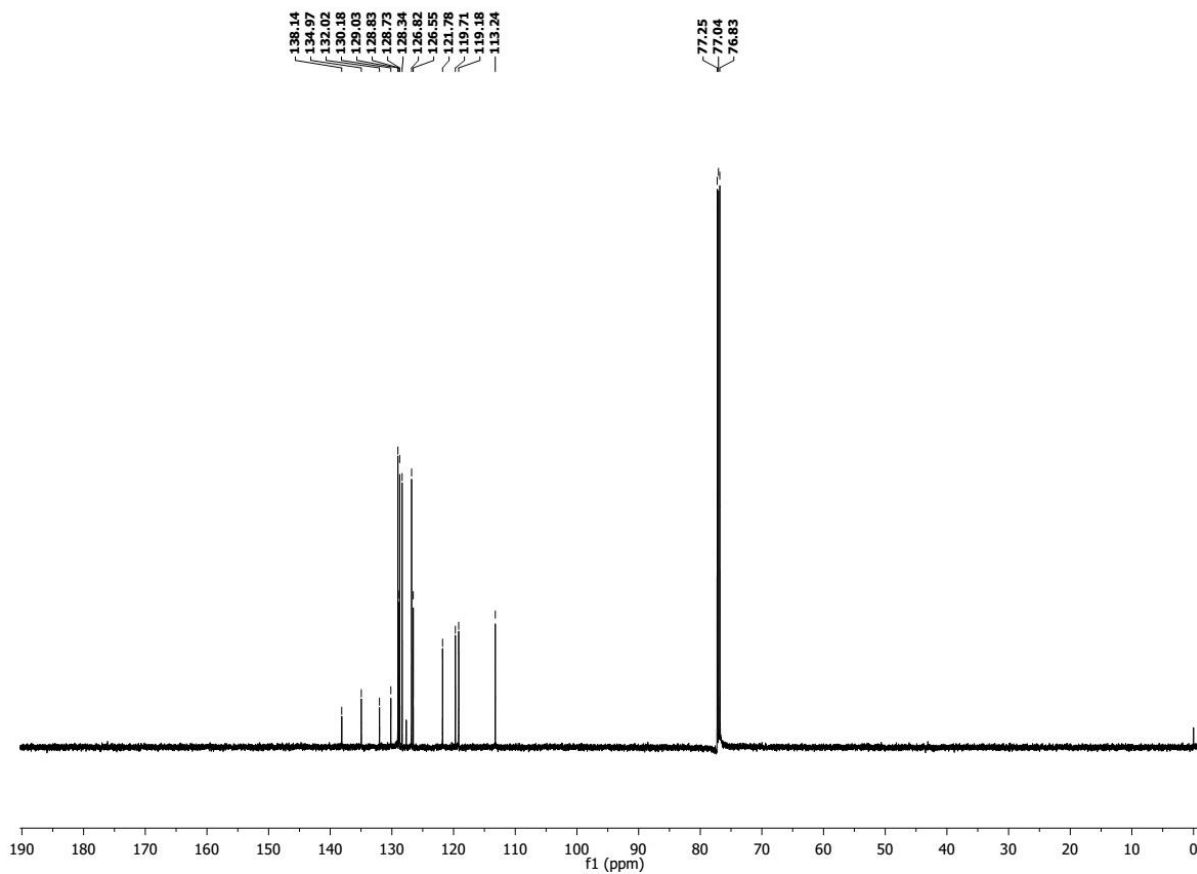
Base Peak 271.19 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (1.6:1.7;2.2:3.8;0.2:1.2) x 20.000 Th: 0.010 Retention Time 1.889

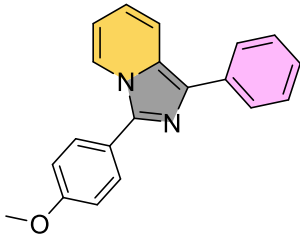
<sup>1</sup>H NMR of Compound 3a





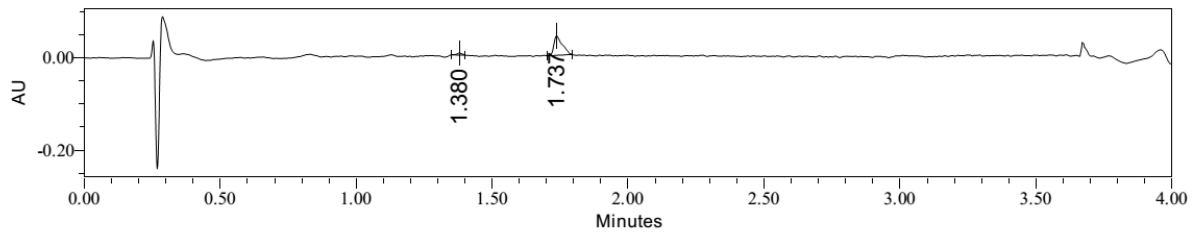
$^{13}\text{C}$  NMR of Compound 3a





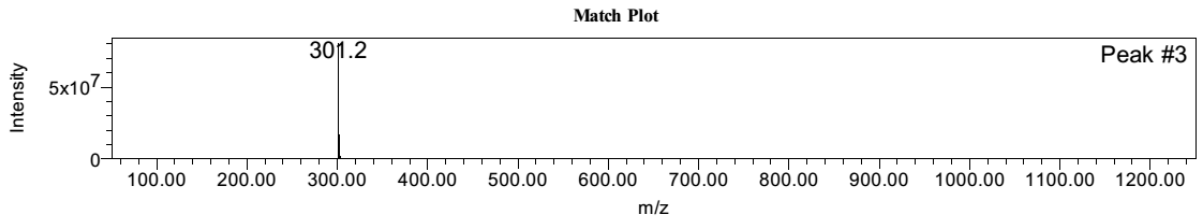
Compound (3b)

LC-MS analysis of Compound 3b



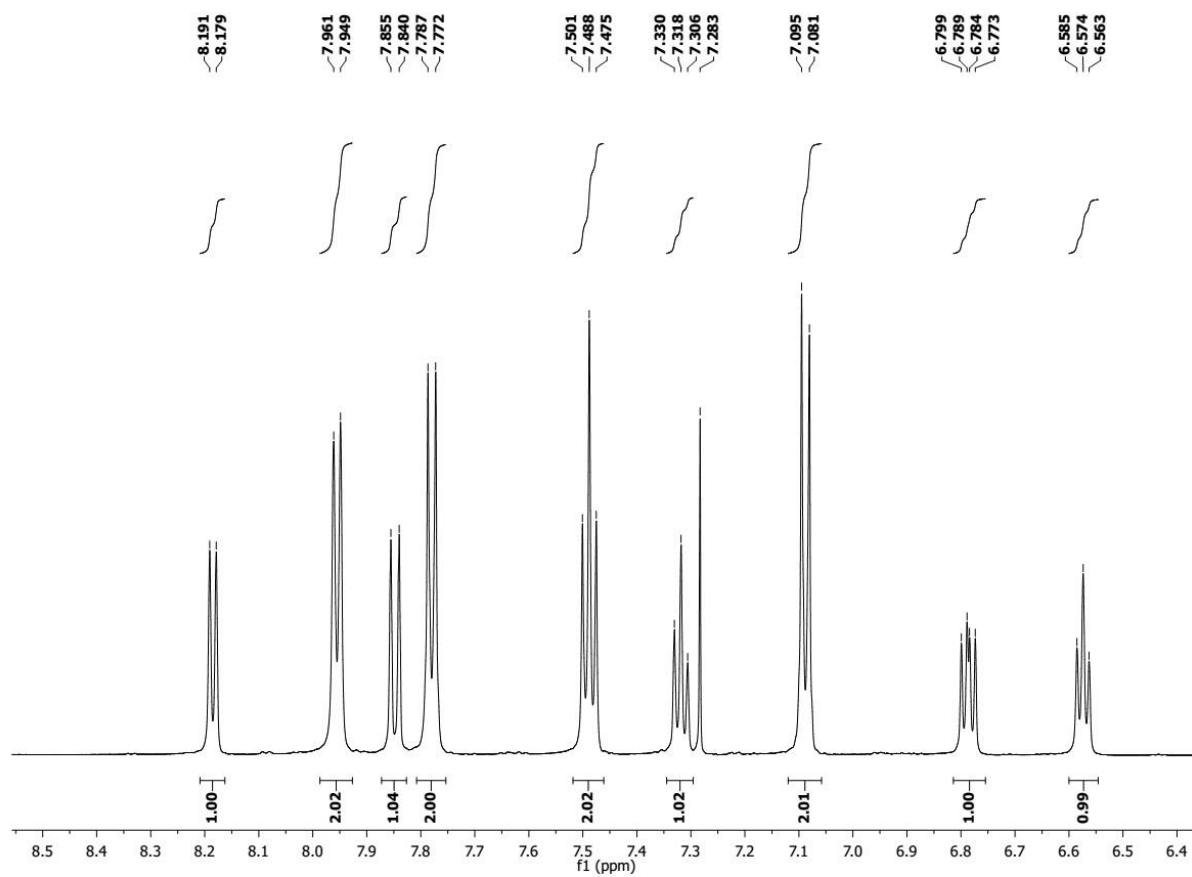
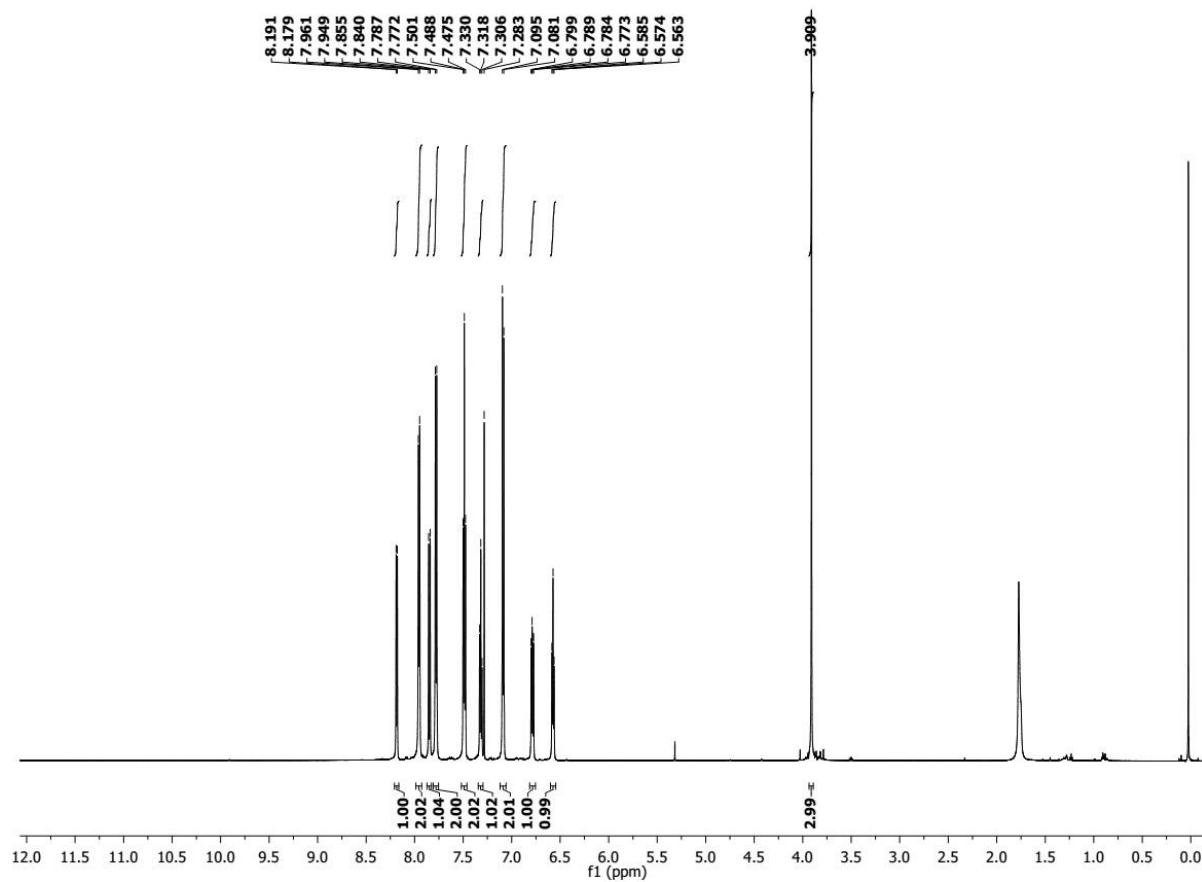
**Peak Results**  
Channel: PDA Spectrum

	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.380		4384	6113	6.34	PDA Spectrum	220.0nm
2	1.737		42212	90240	93.66	PDA Spectrum	220.0nm

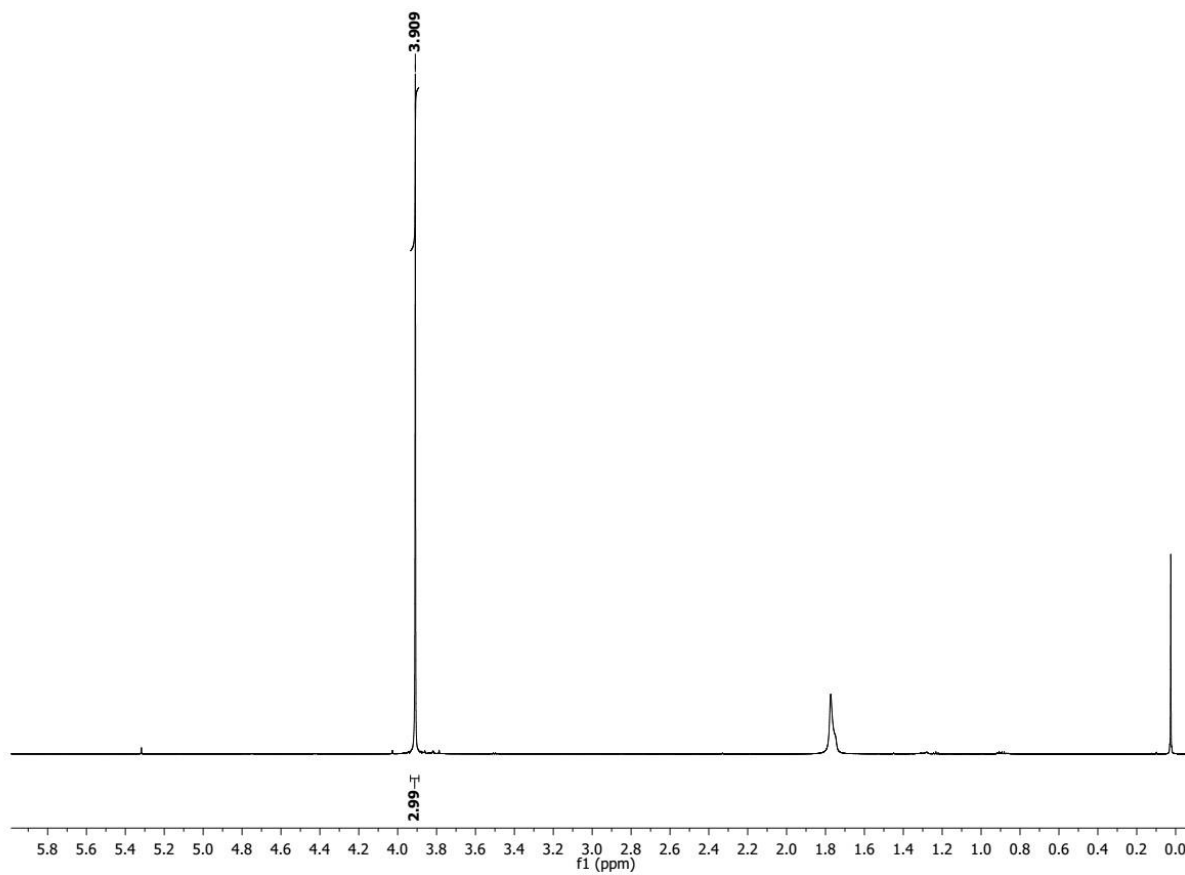


Base Peak 301.18 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (0.1:1.5;1.9:3.8) x 20.000 Th: 0.010 Retention Time 1.766

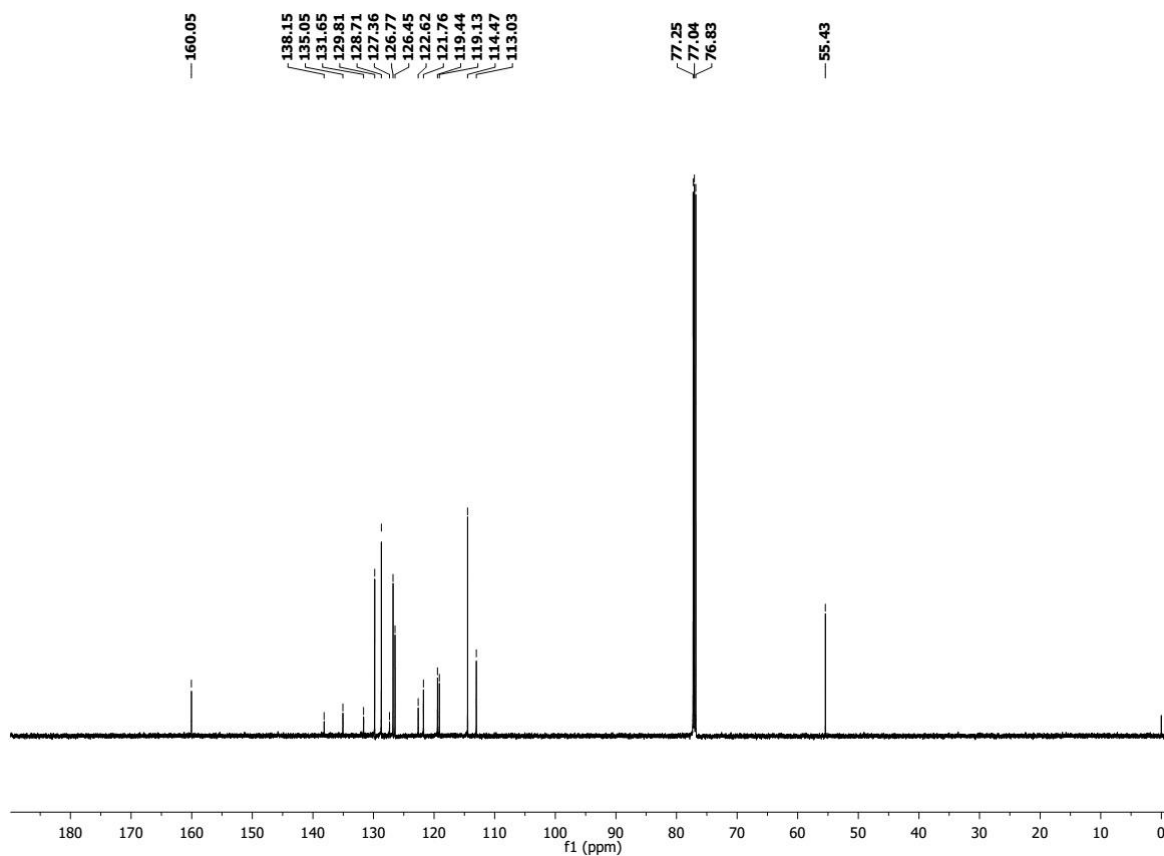
<sup>1</sup>H NMR of Compound 3b

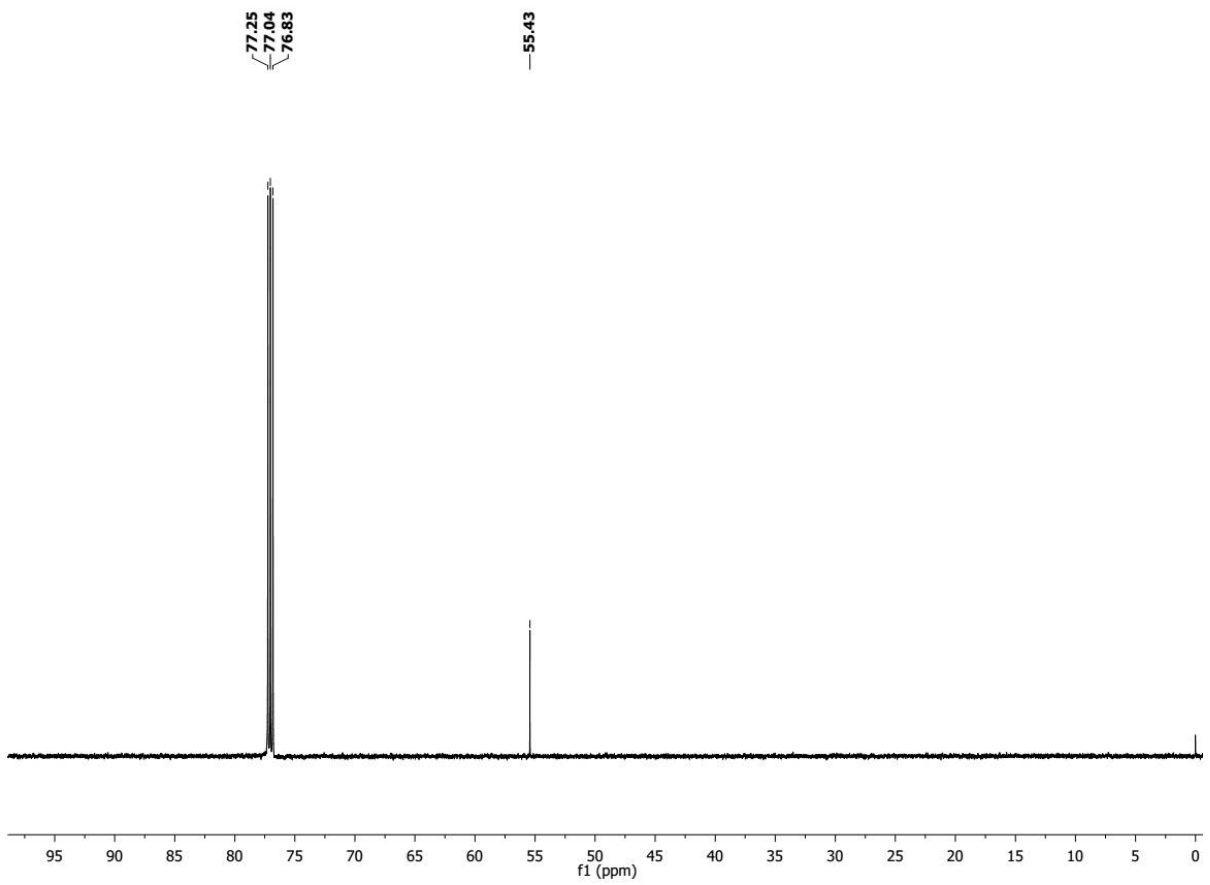
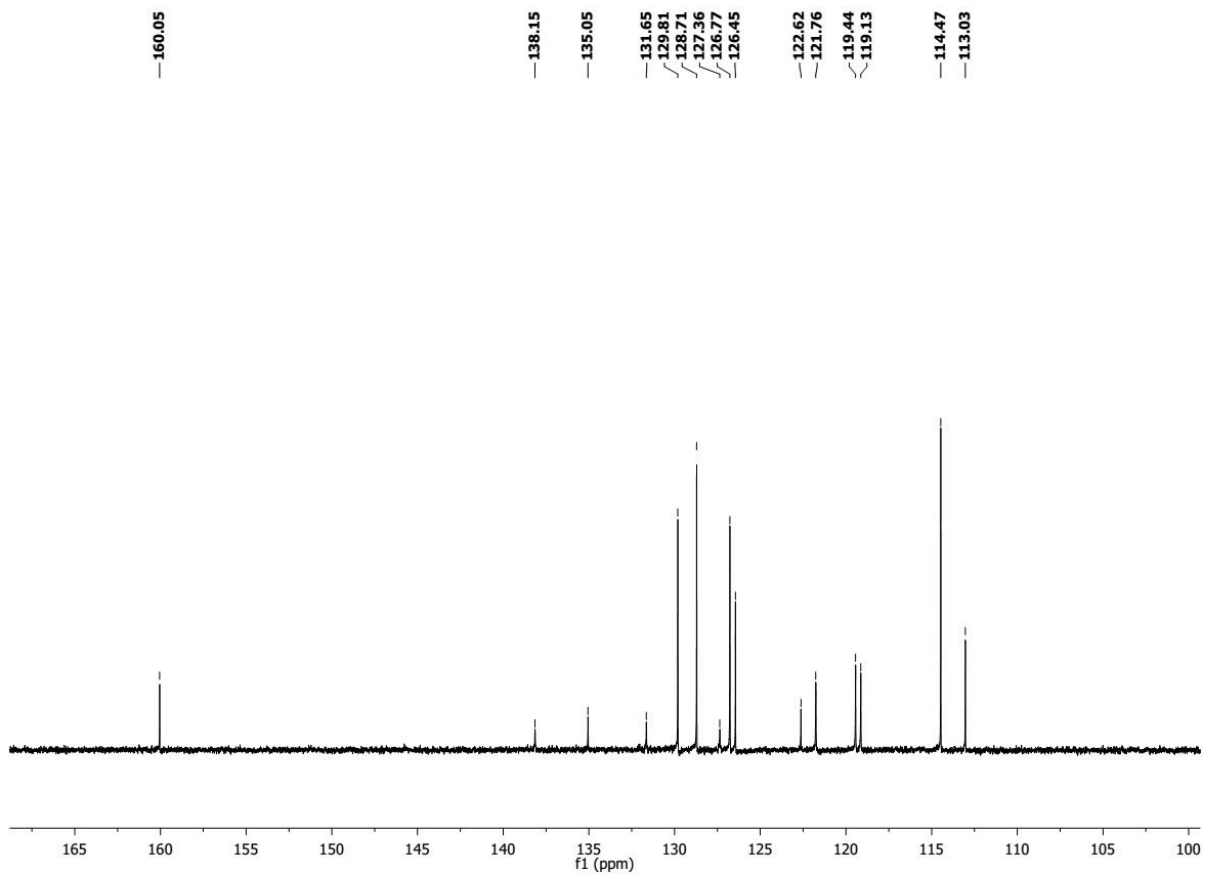


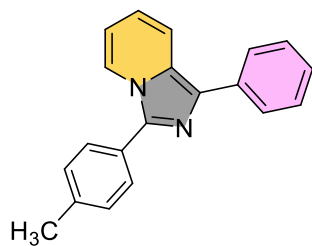




<sup>13</sup>C NMR of Compound 3b

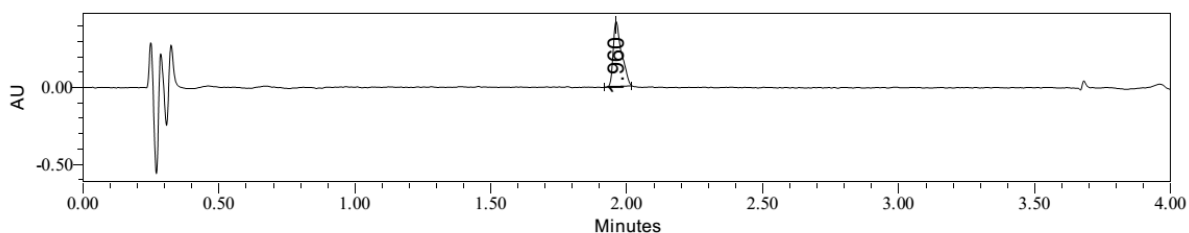




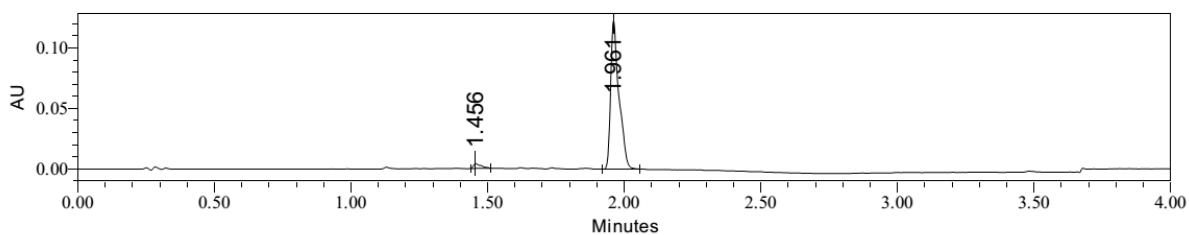


Compound (3c)

LC-MS analysis of Compound 3c



Channel Name 210.0nm; Channel PDA Spectrum

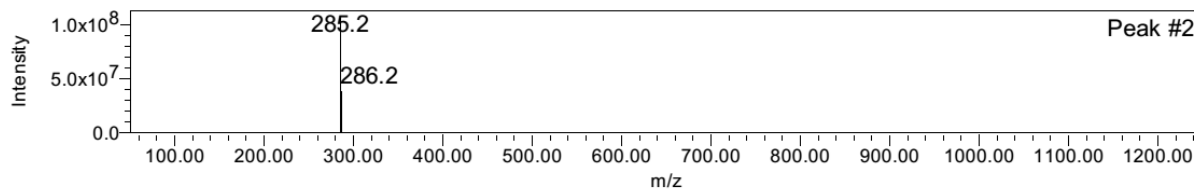


Channel Name 254.0nm; Channel PDA Spectrum

**Peak Results**  
**Channel: PDA Spectrum**

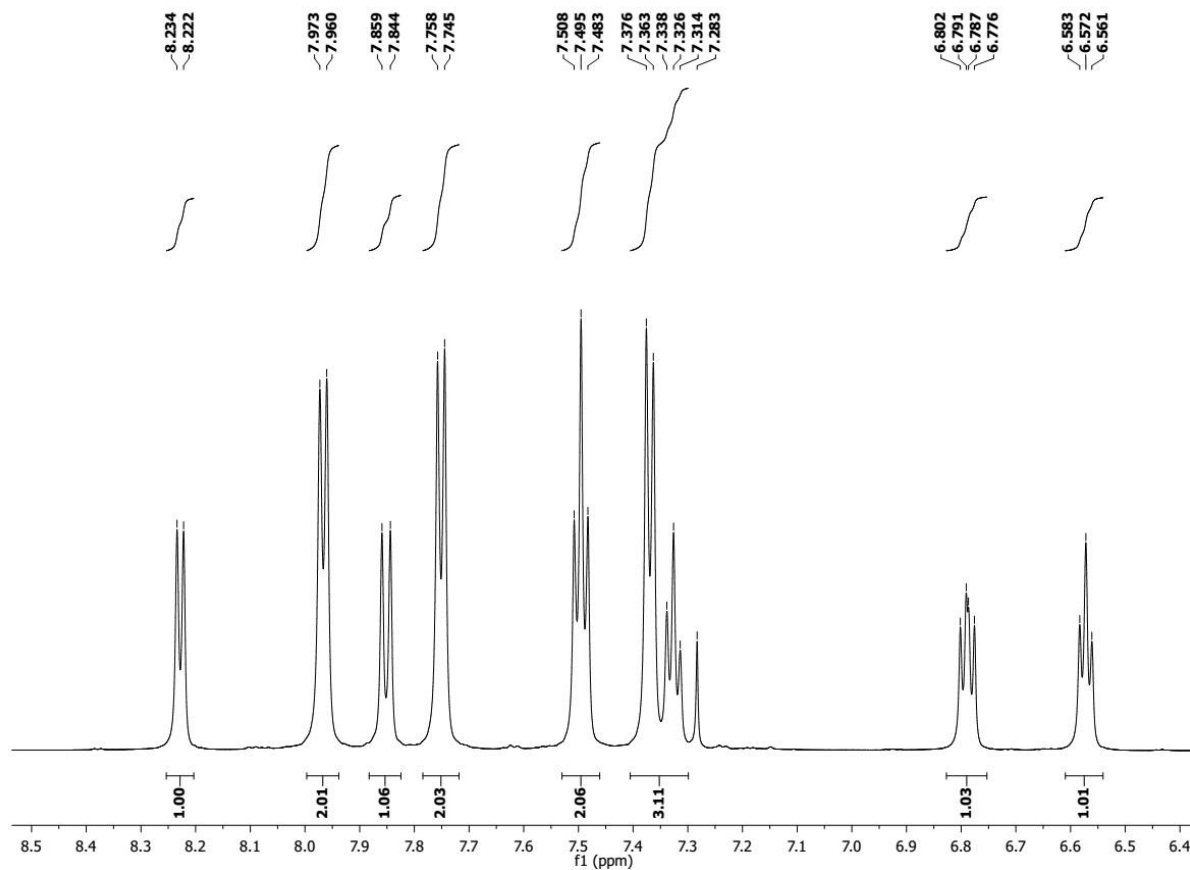
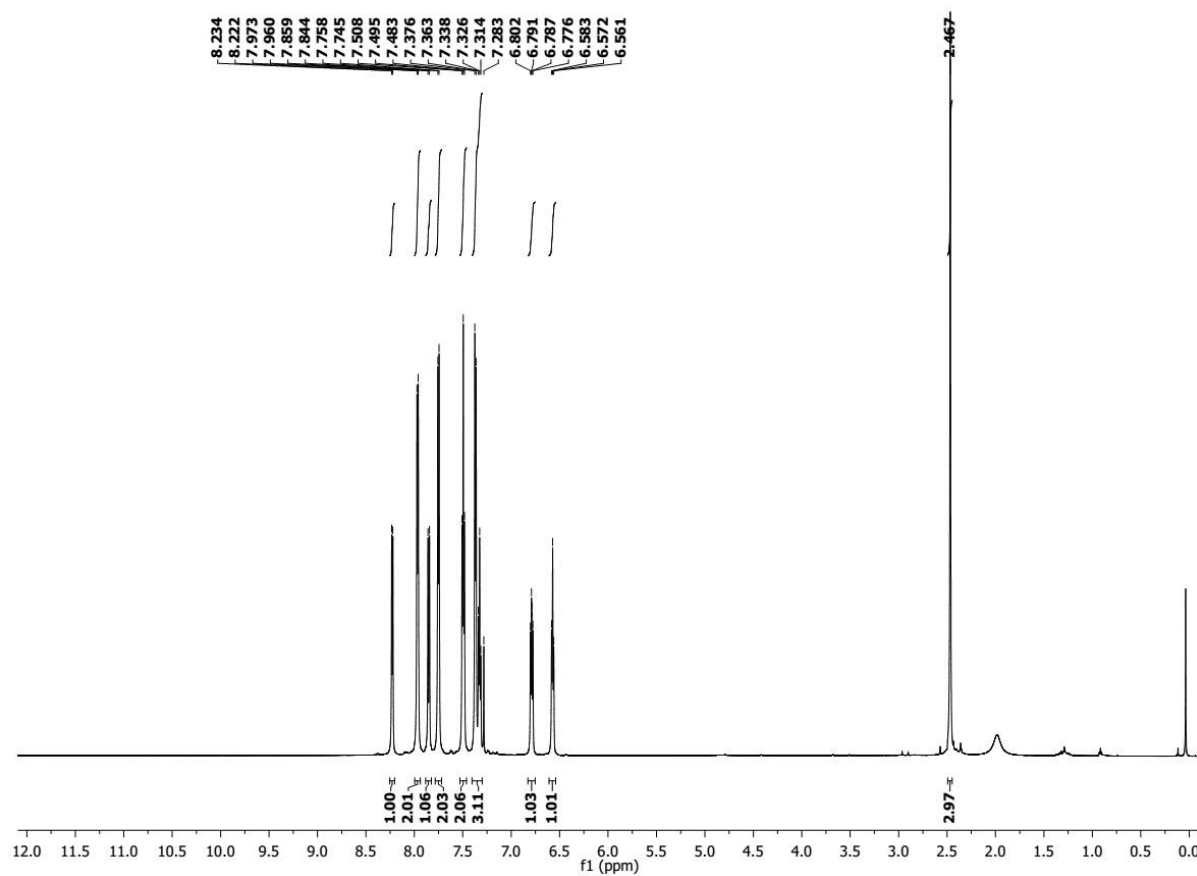
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.456		3931	7630	2.86	PDA Spectrum	254.0nm
2	1.960		424520	902743	100.00	PDA Spectrum	210.0nm
3	1.961		122448	259546	97.14	PDA Spectrum	254.0nm

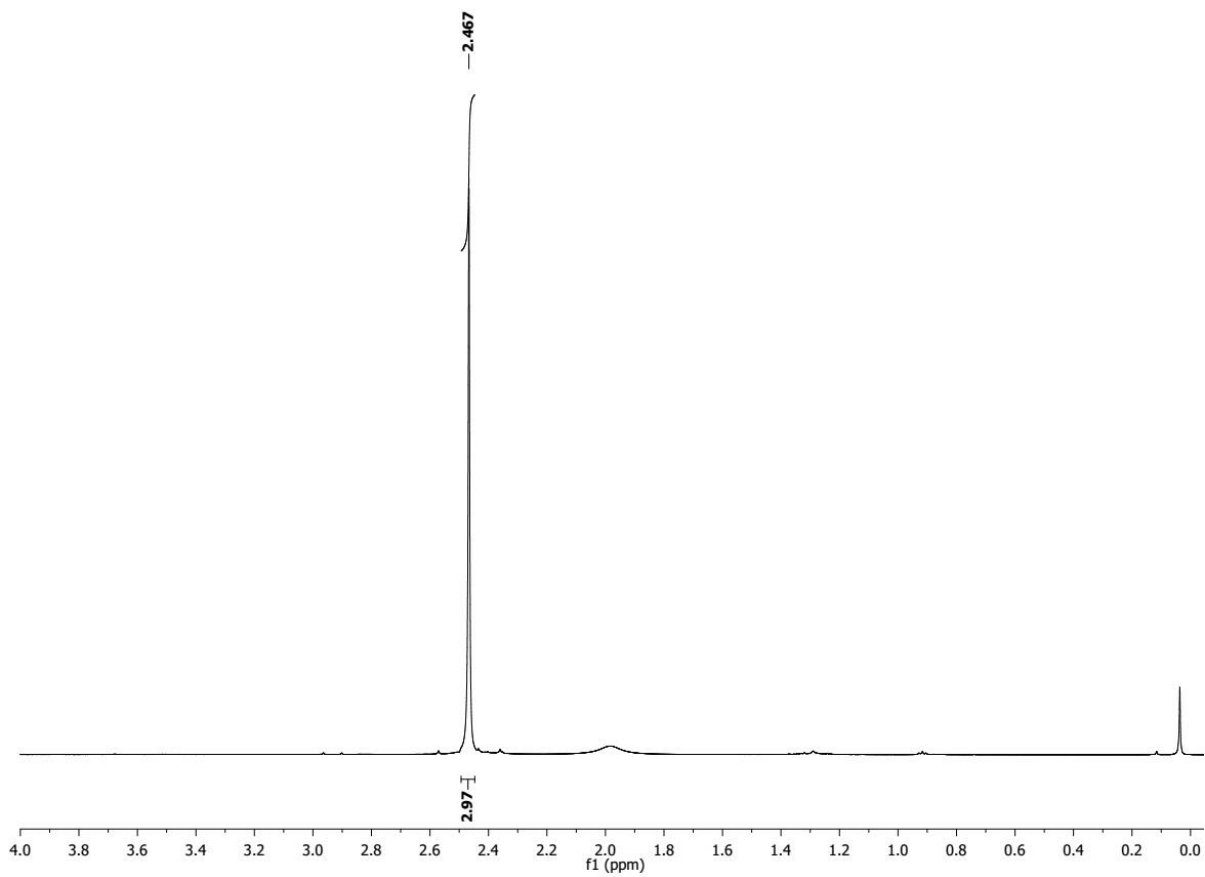
**Match Plot**



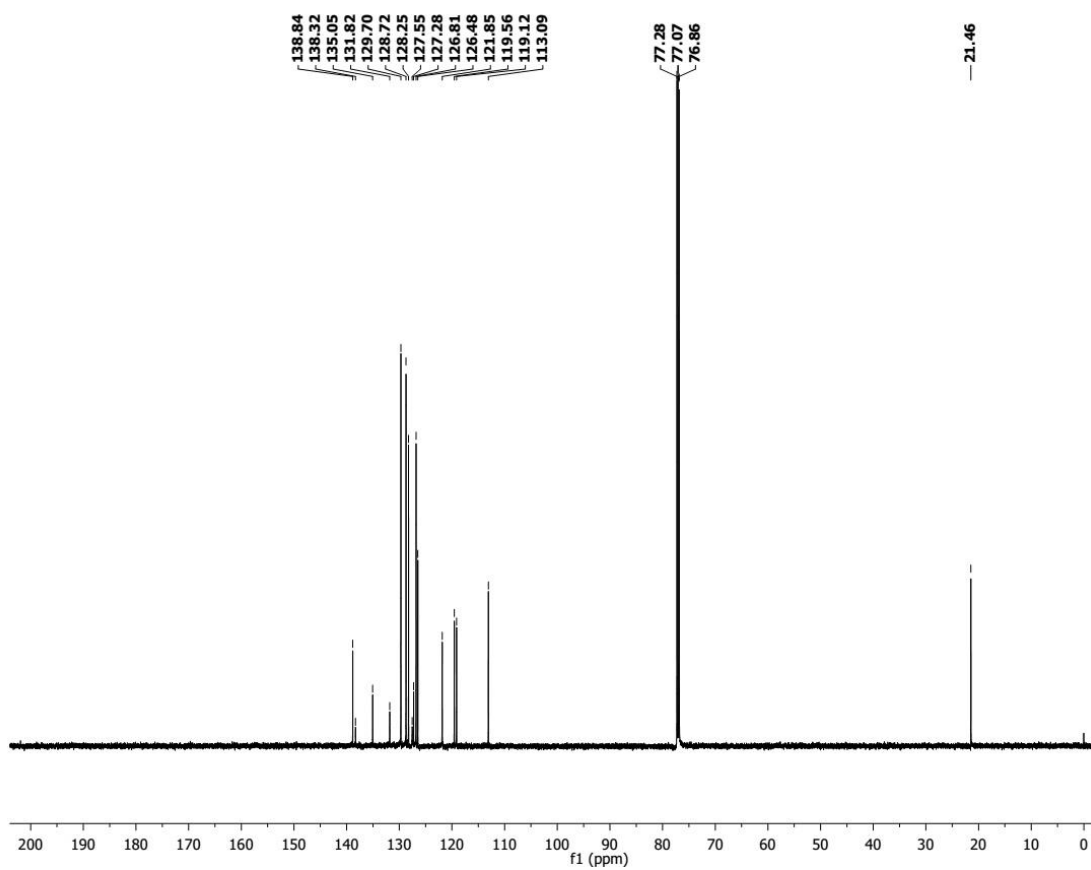
Base Peak 285.21 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (1.6:1.8;0.5:1.3;2.2:3.8) x 20.000 Th: 0.010 Retention Time 1.990

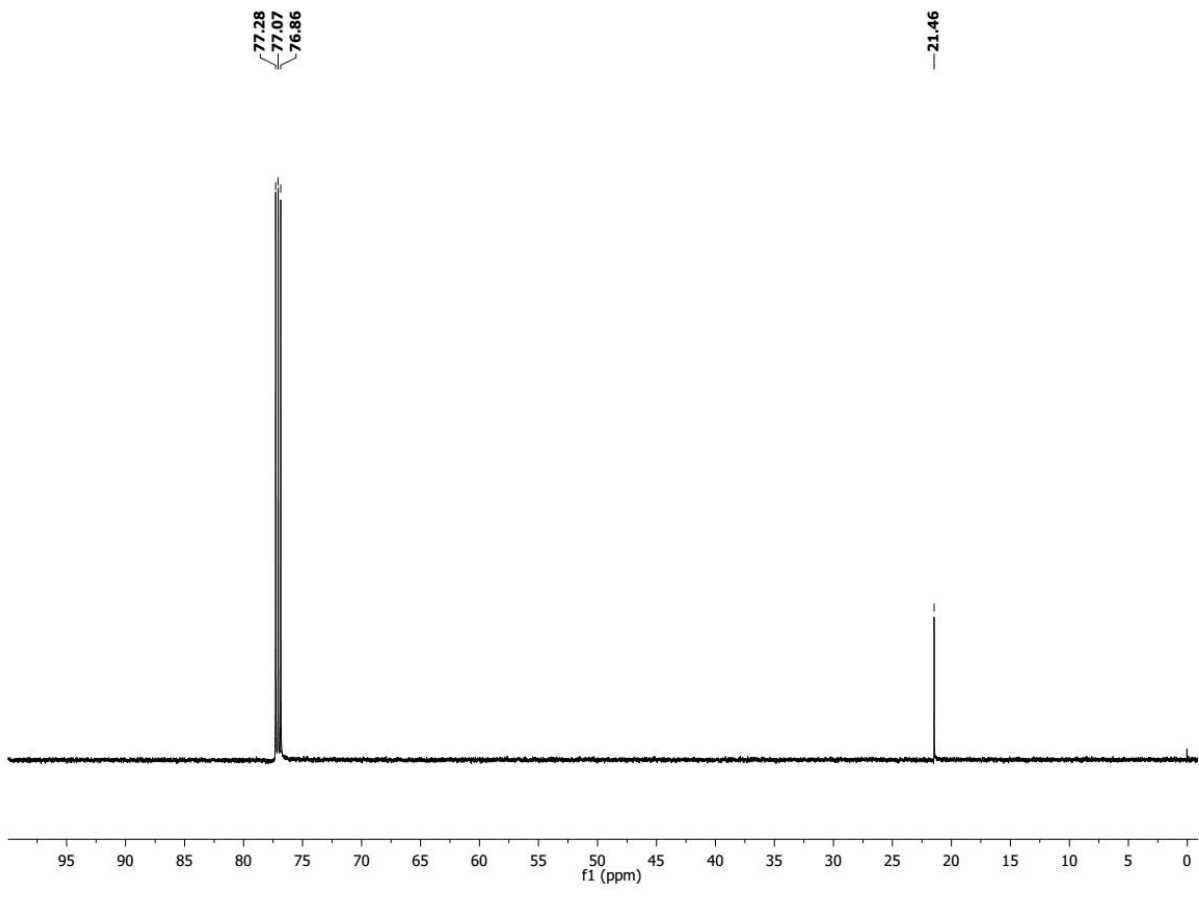
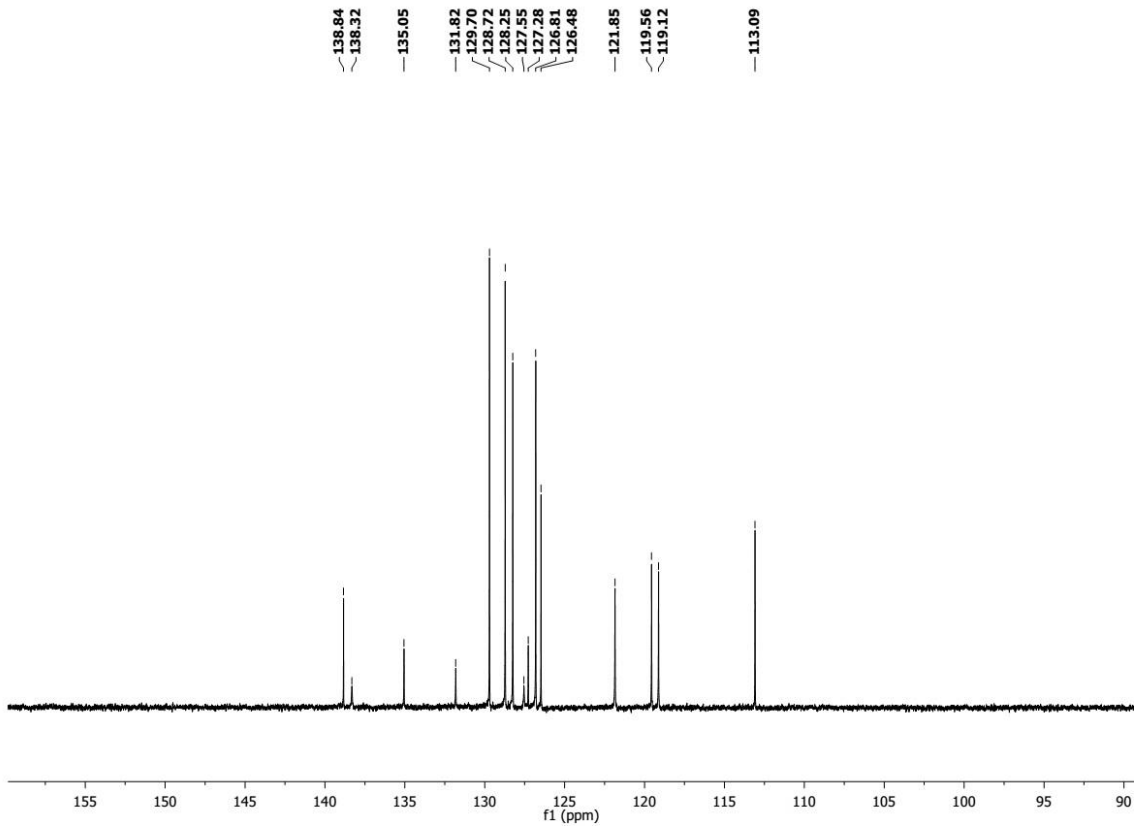
<sup>1</sup>H NMR of Compound 3c

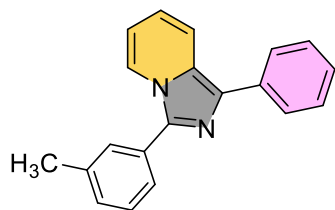




$^{13}\text{C}$  NMR of Compound 3c

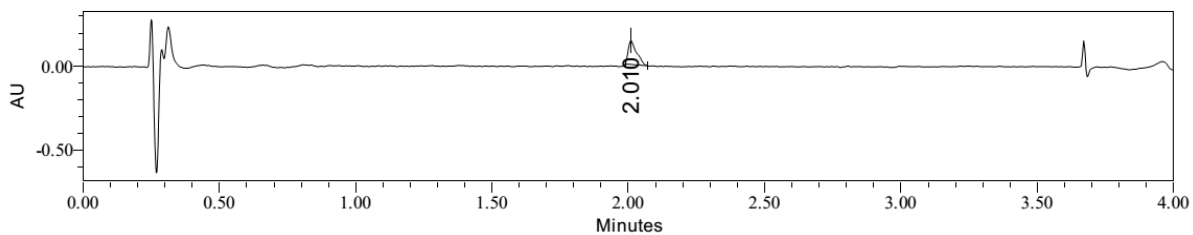






Compound (3d)

LC-MS analysis of Compound 3d

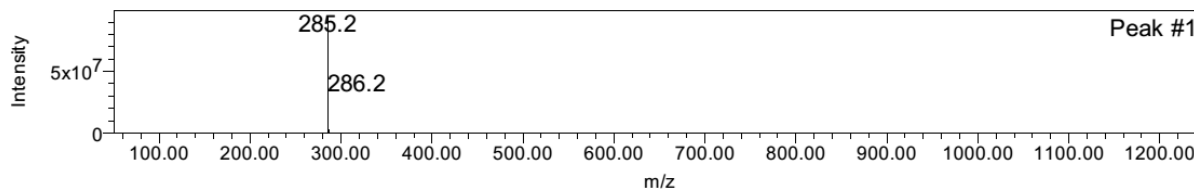


Channel Name 210.0nm; Channel PDA Spectrum

**Peak Results**  
Channel: PDA Spectrum

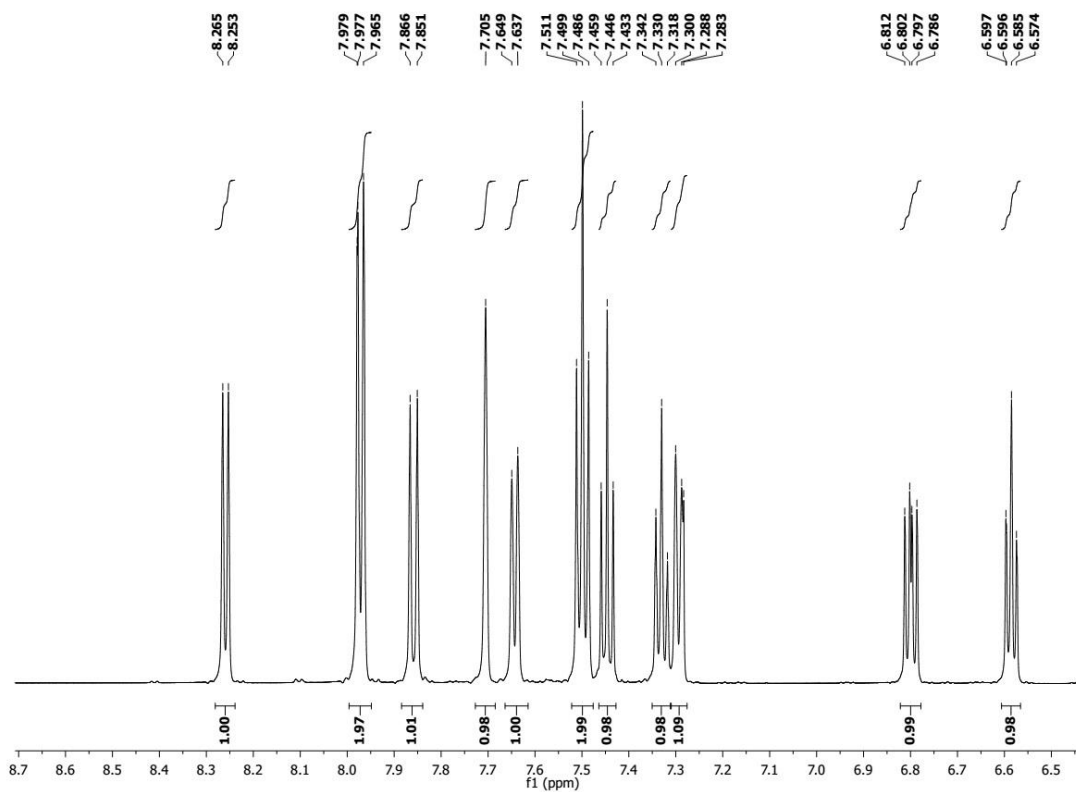
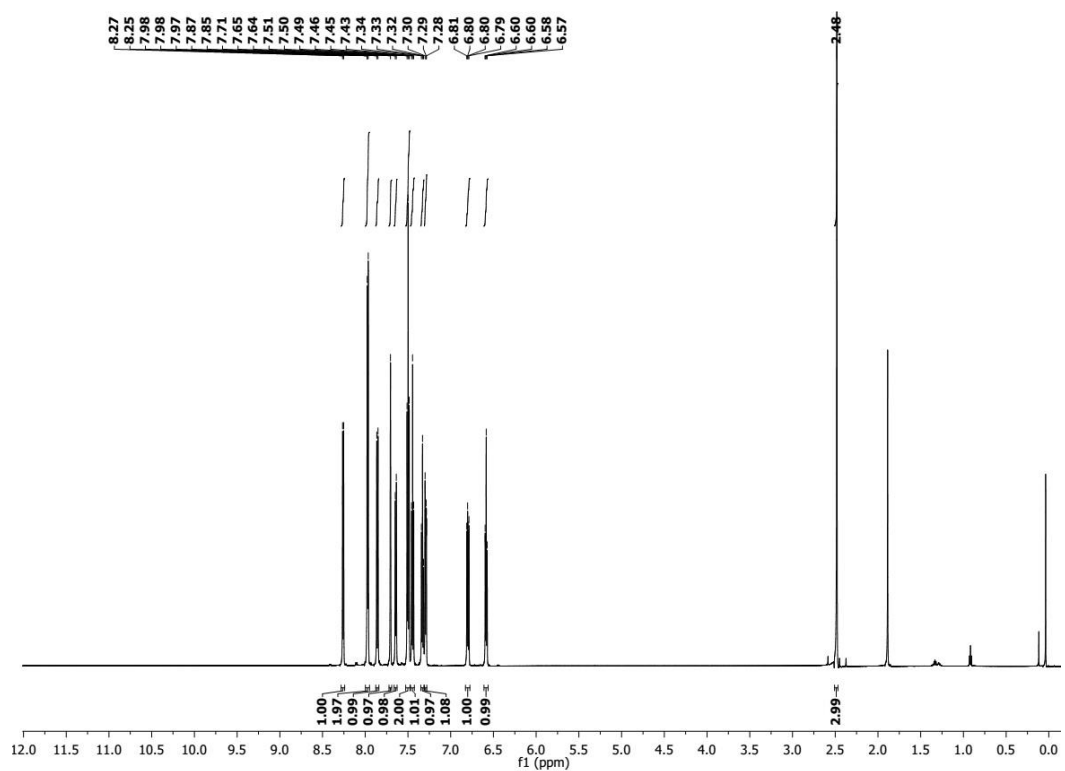
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	2.010		141643	302143	100.00	PDA Spectrum	210.0nm

**Match Plot**

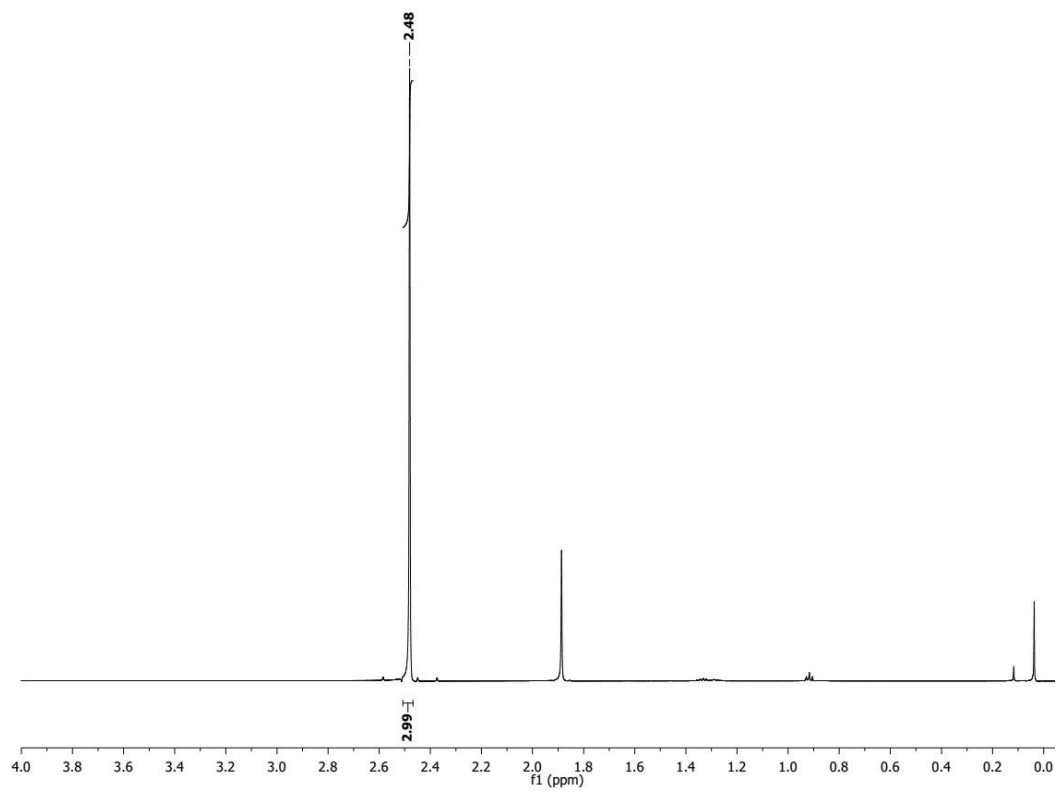


Base Peak 285.18 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (0.0:1.9;2.2:3.8) x 20.000 Th: 0.010 Retention Time 2.039

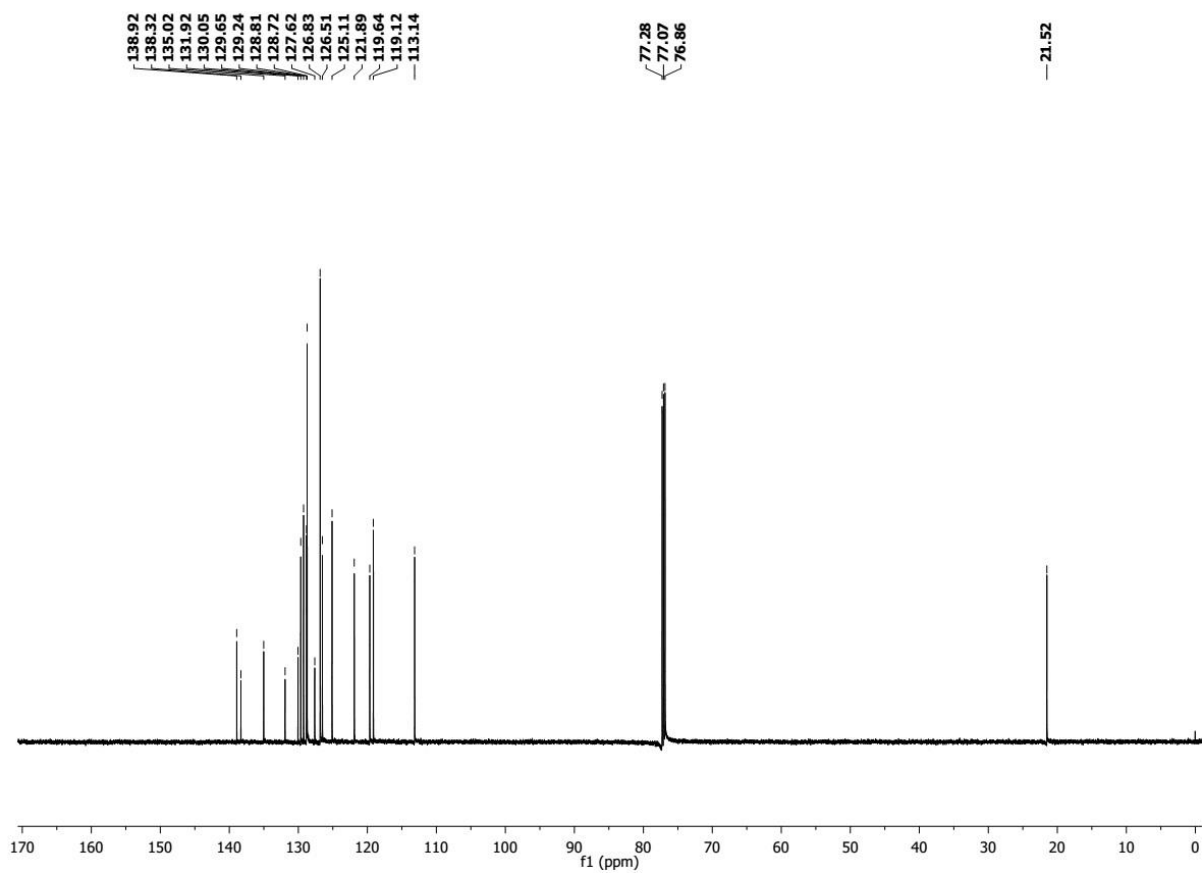
# <sup>1</sup>H NMR of Compound 3d

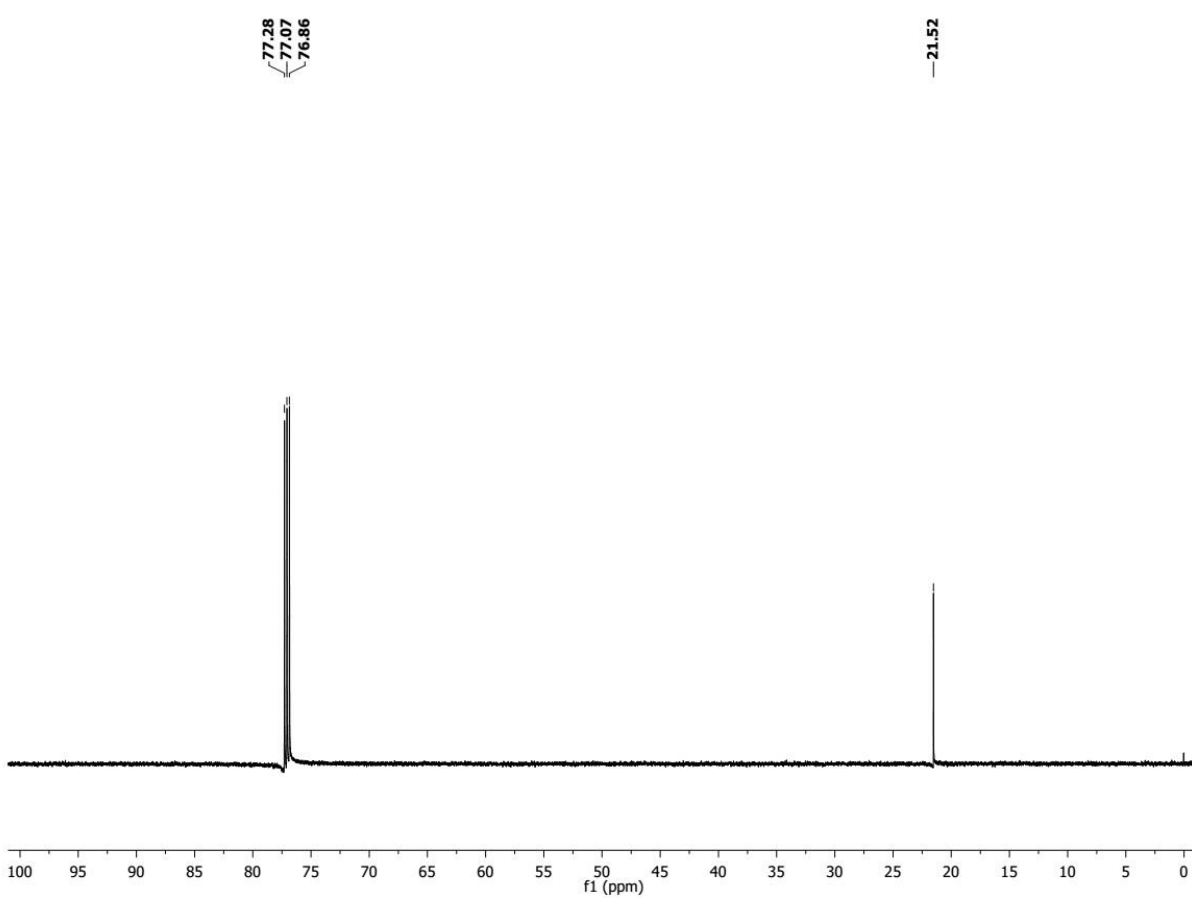
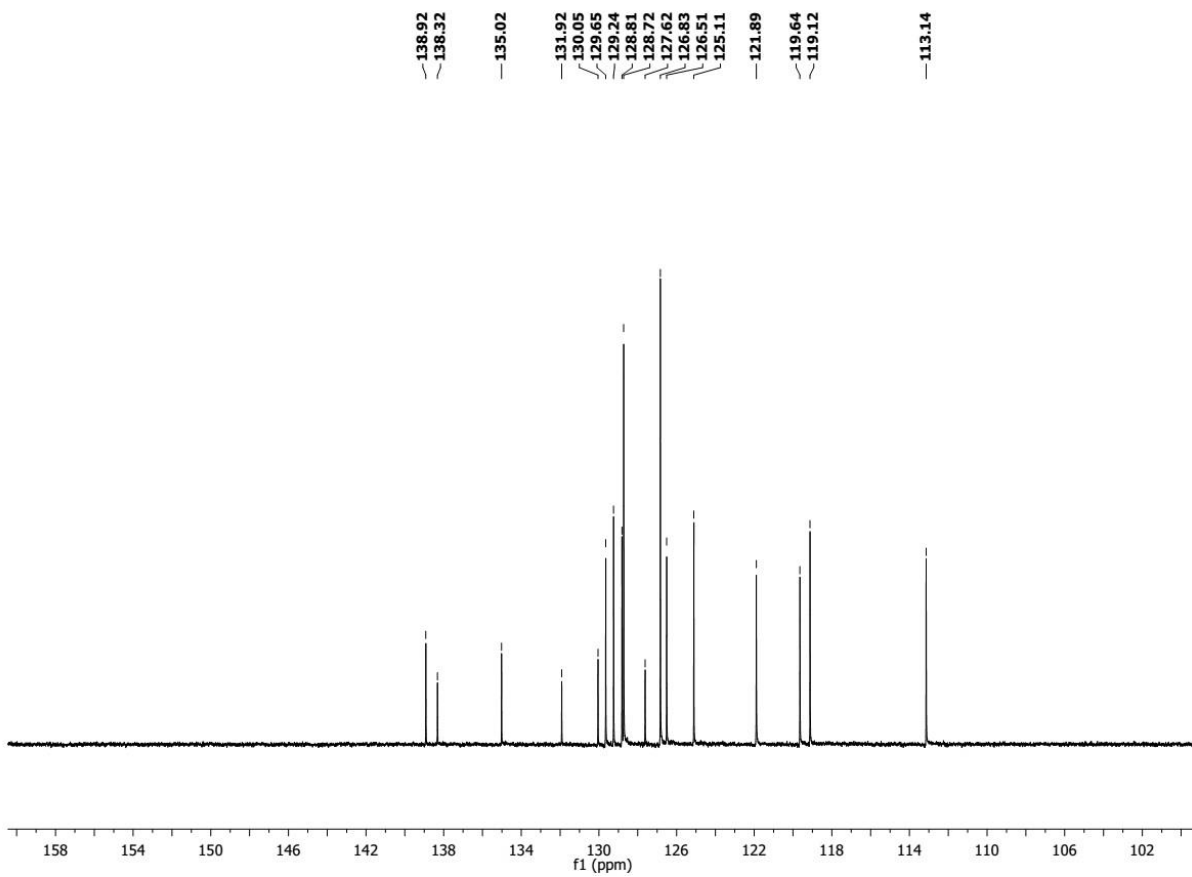


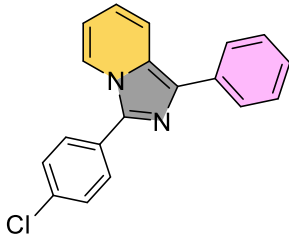




<sup>13</sup>C NMR of Compound 3d

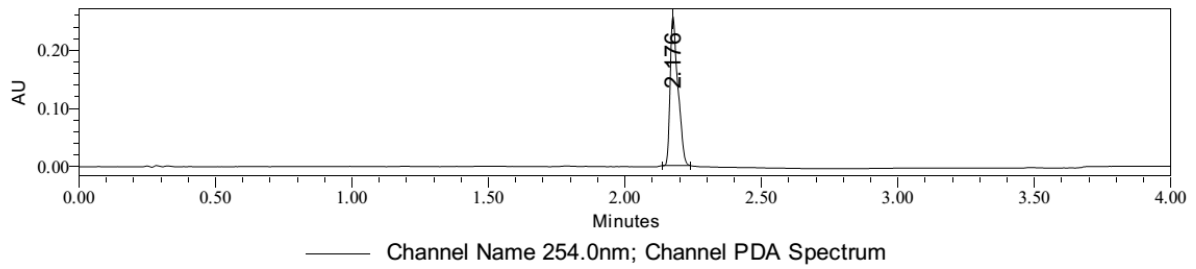
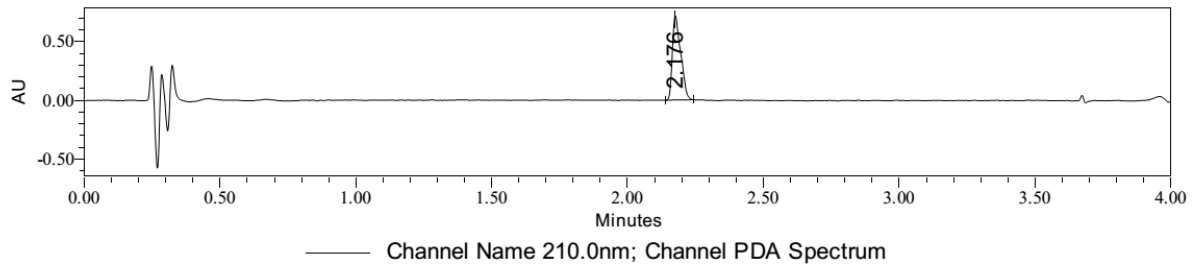






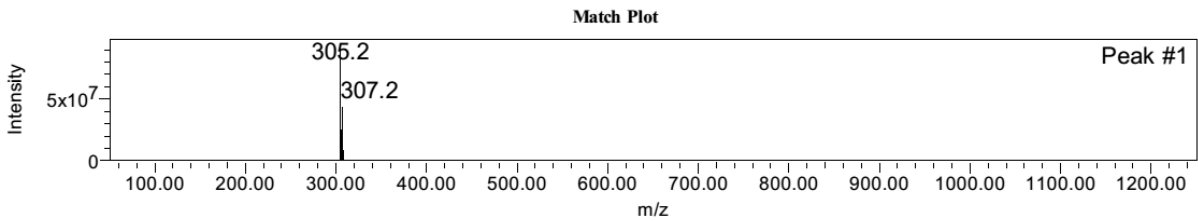
Compound (3e)

LC-MS analysis of Compound 3e



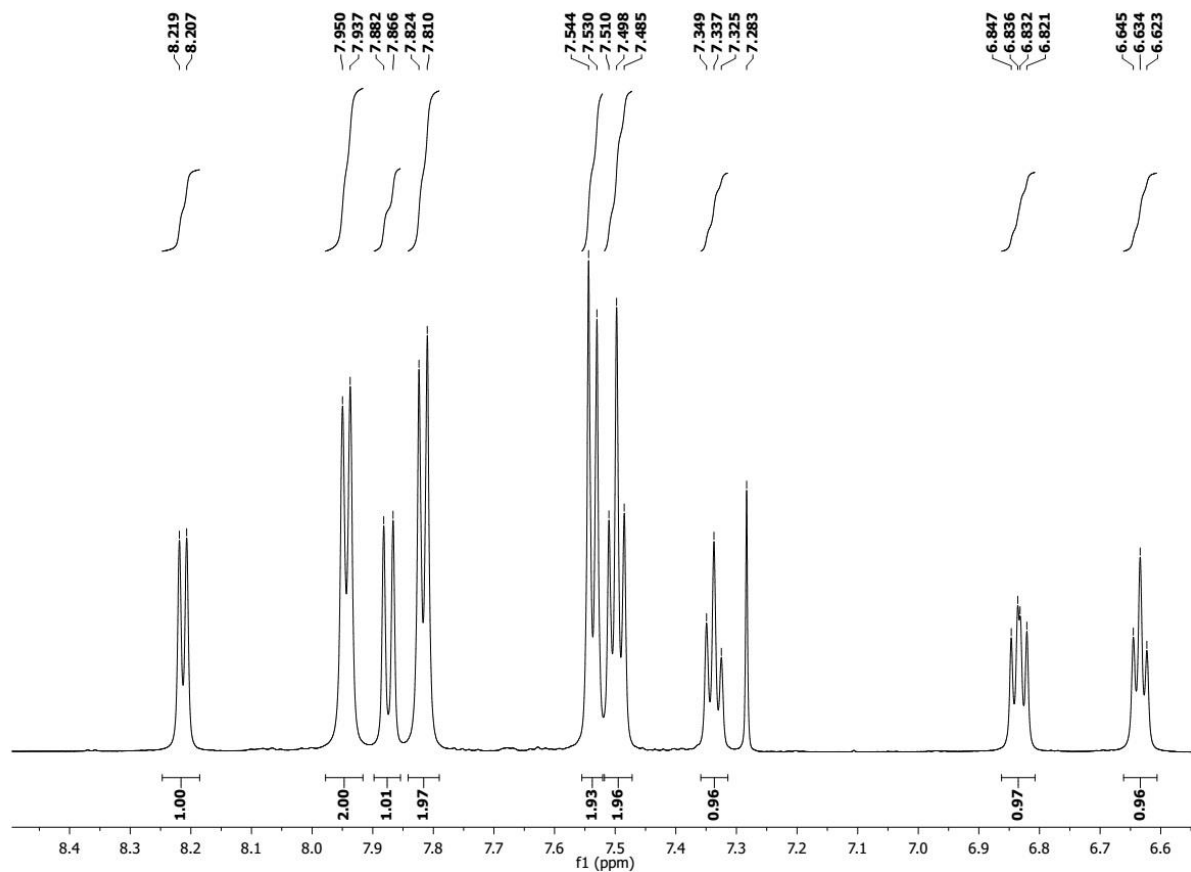
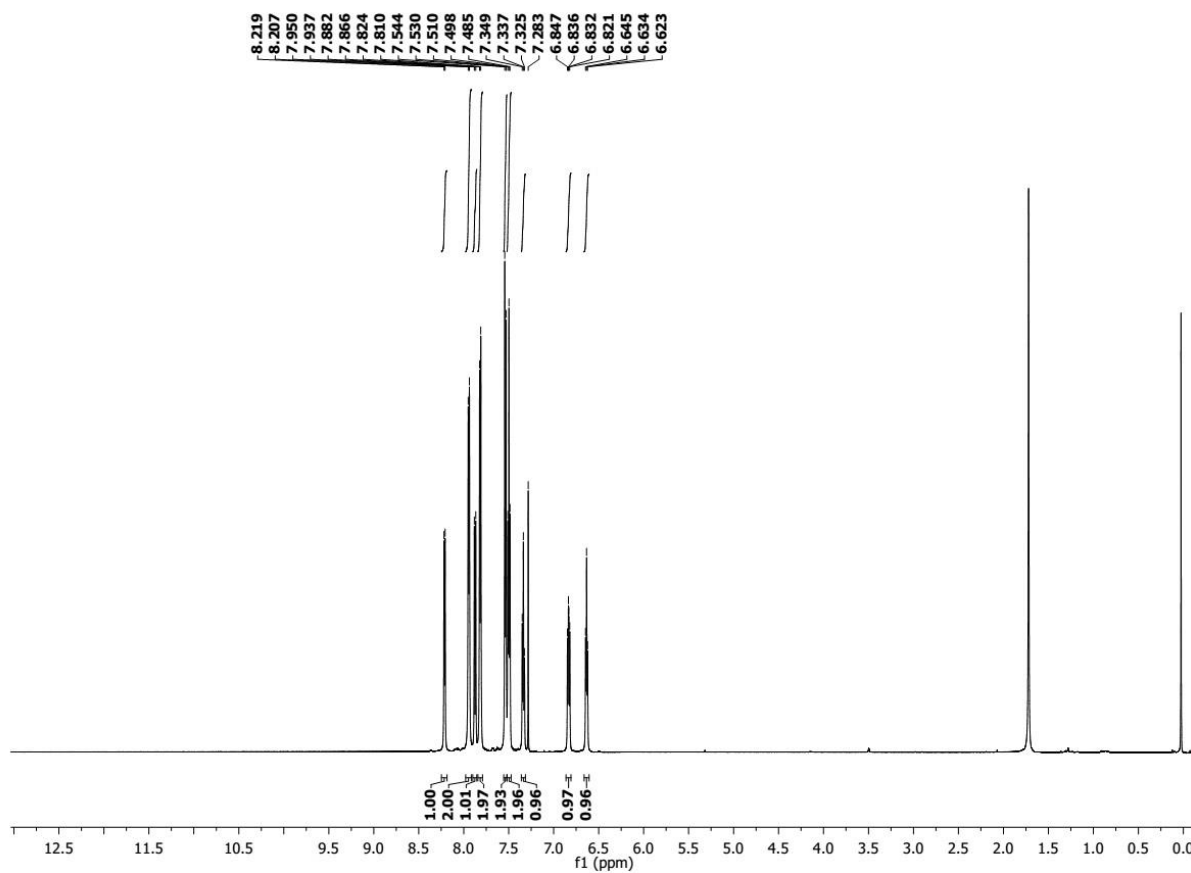
**Peak Results**  
Channel: PDA Spectrum

	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	2.176		257212	506509	100.00	PDA Spectrum	254.0nm
2	2.176		716419	1524302	100.00	PDA Spectrum	210.0nm

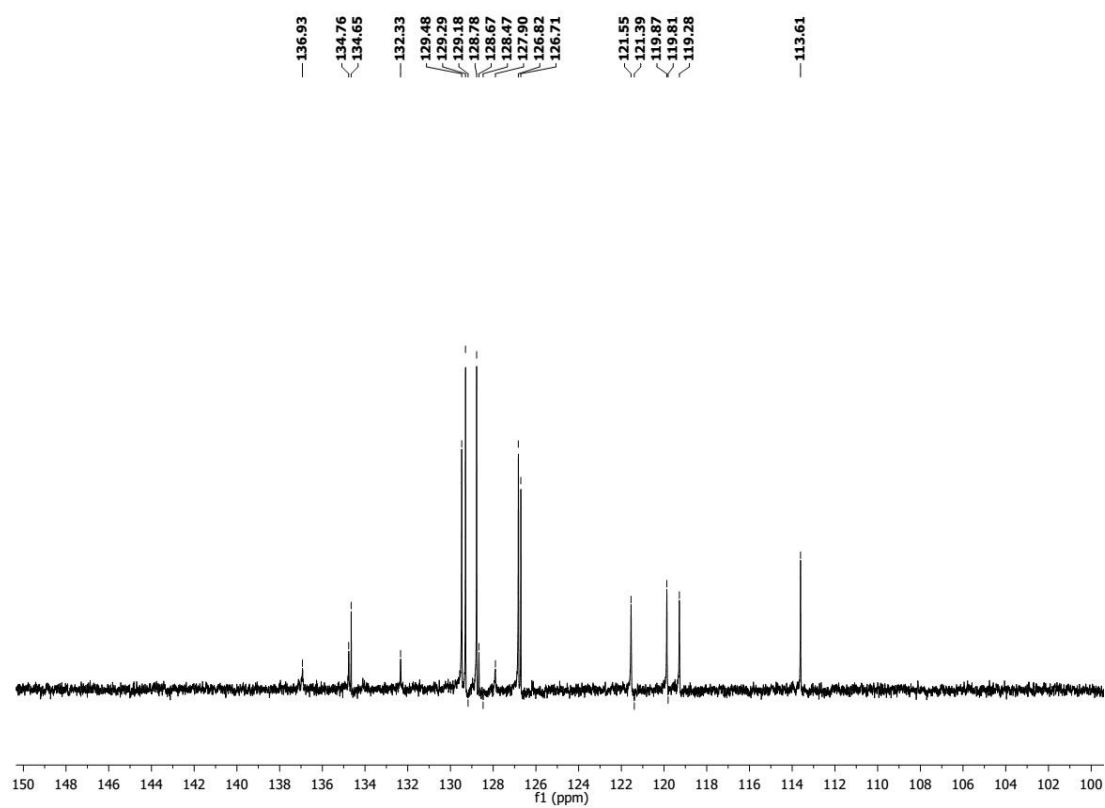
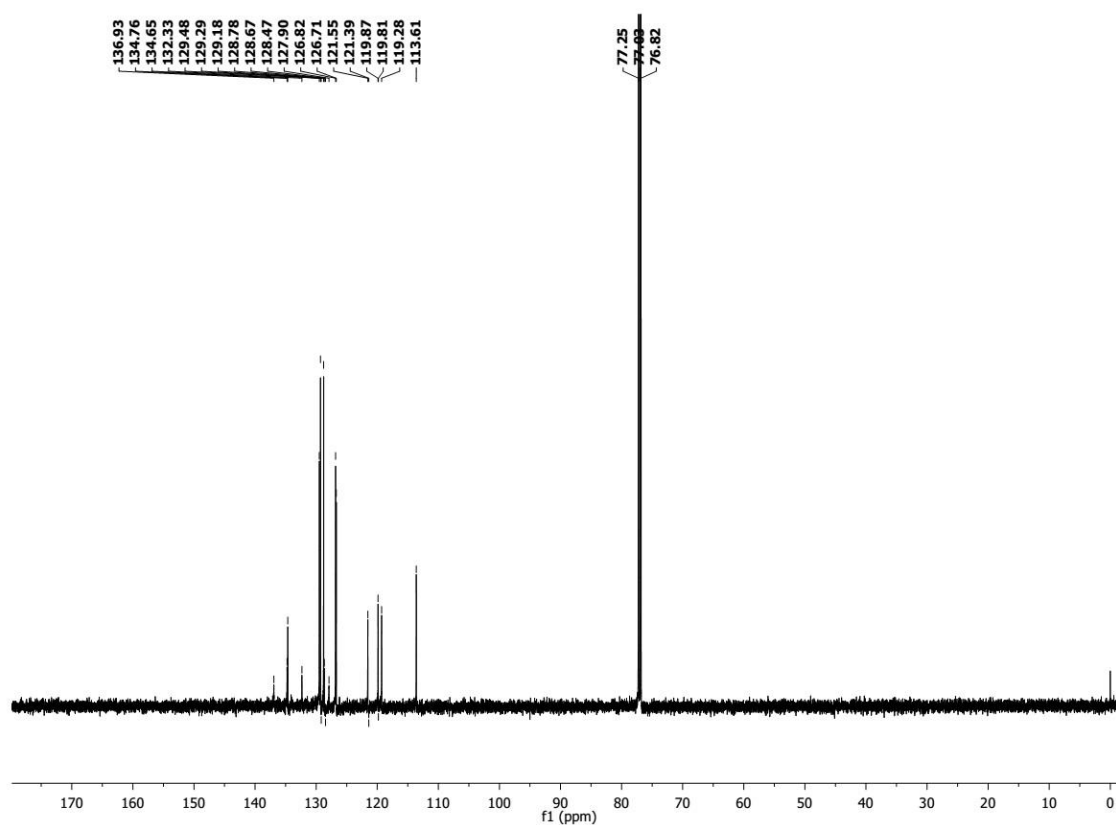


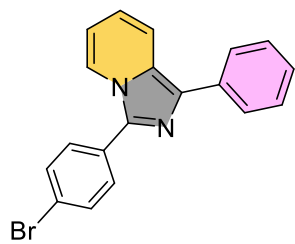
Base Peak 305.16 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (0.1:2.1;2.5:3.8) x 20.000 Th: 0.100 Retention Time 2.207

<sup>1</sup>H NMR of Compound 3e



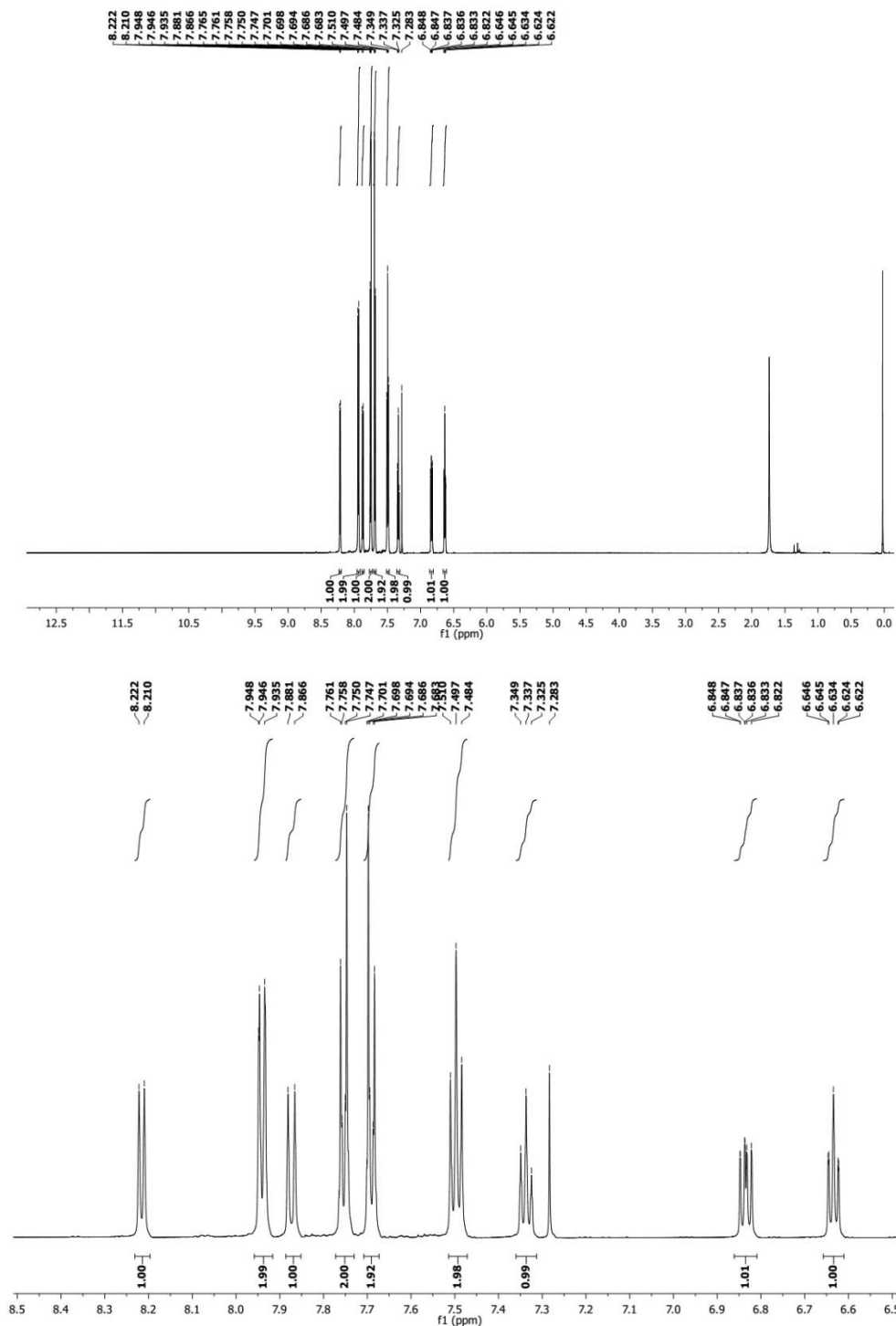
# <sup>13</sup>C NMR of Compound 3e



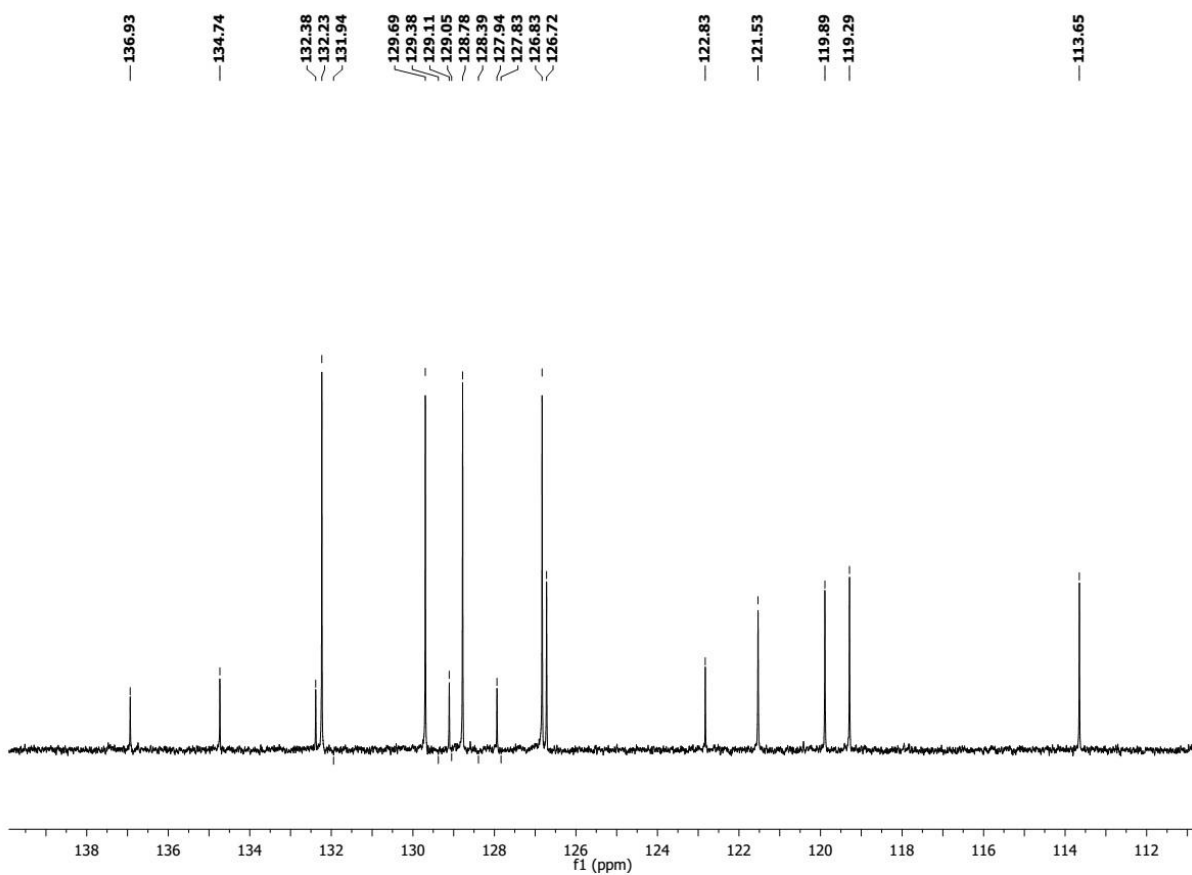
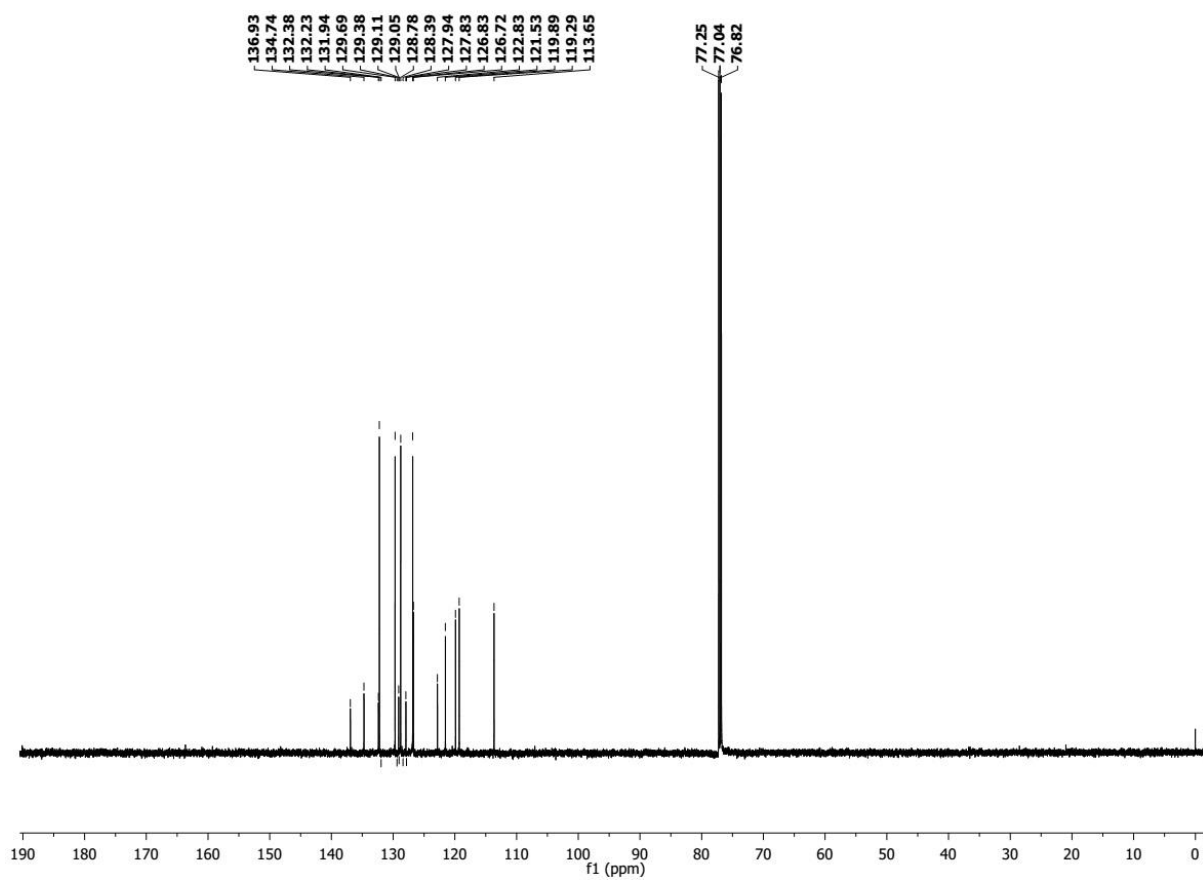


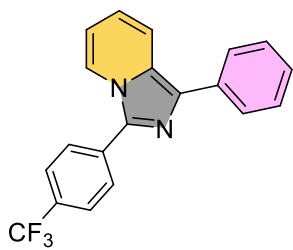
Compound (3f)

$^1\text{H}$  NMR of Compound 3f



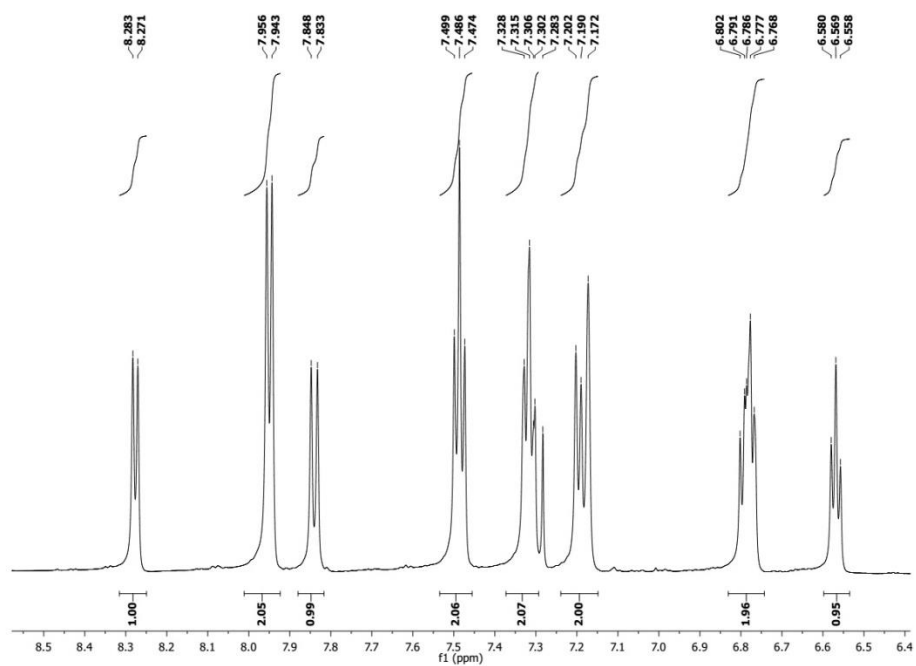
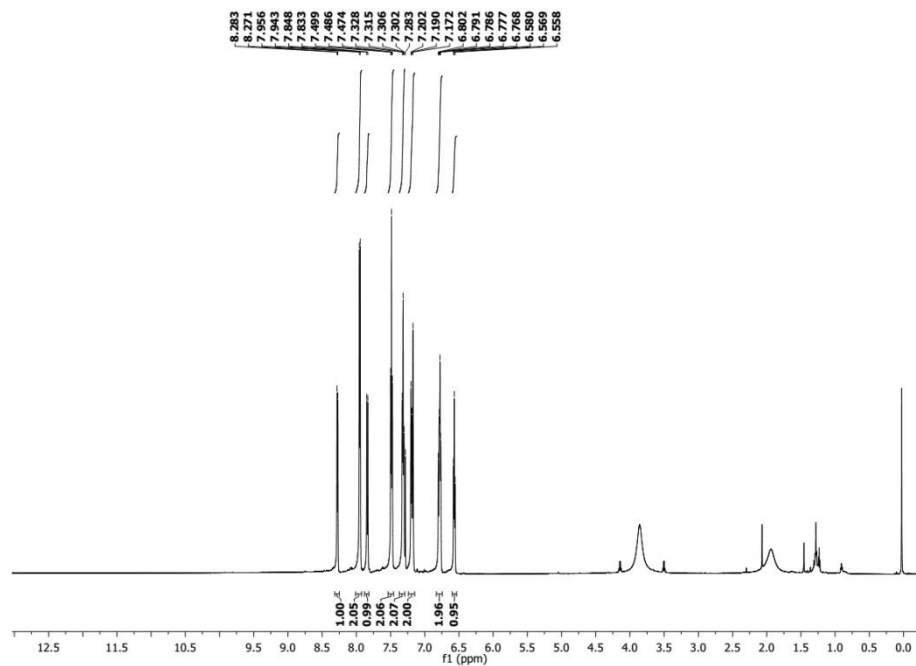
<sup>13</sup>C NMR of Compound 3f





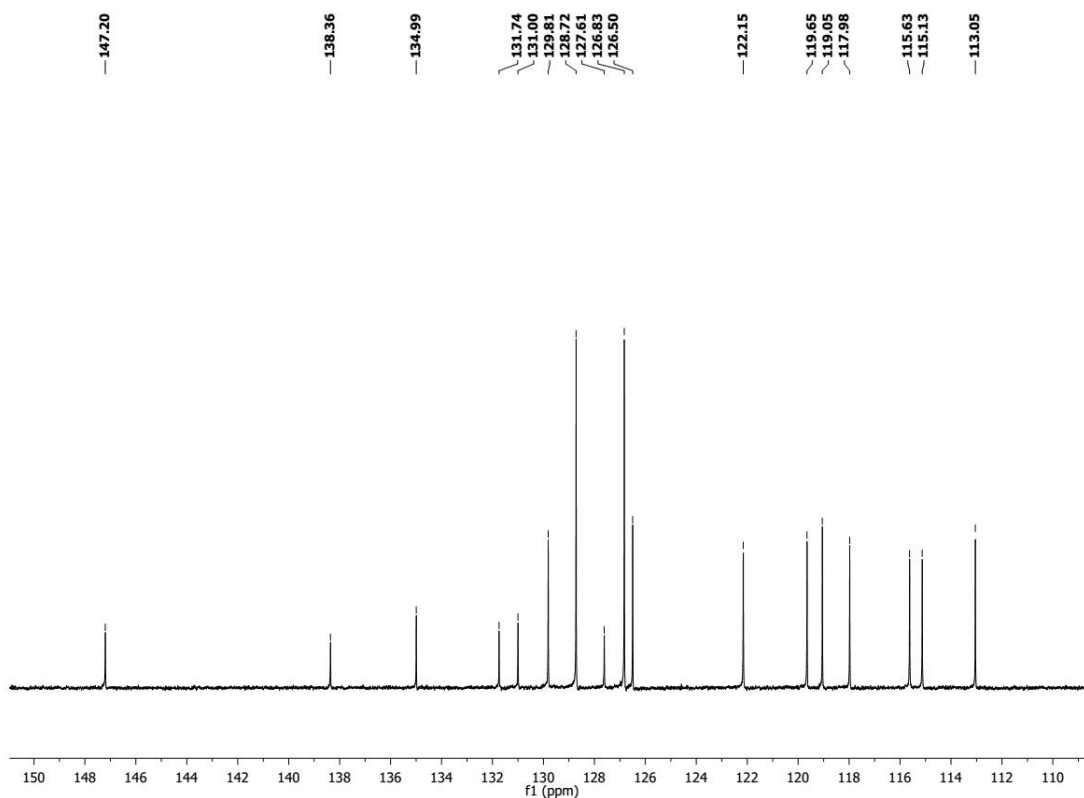
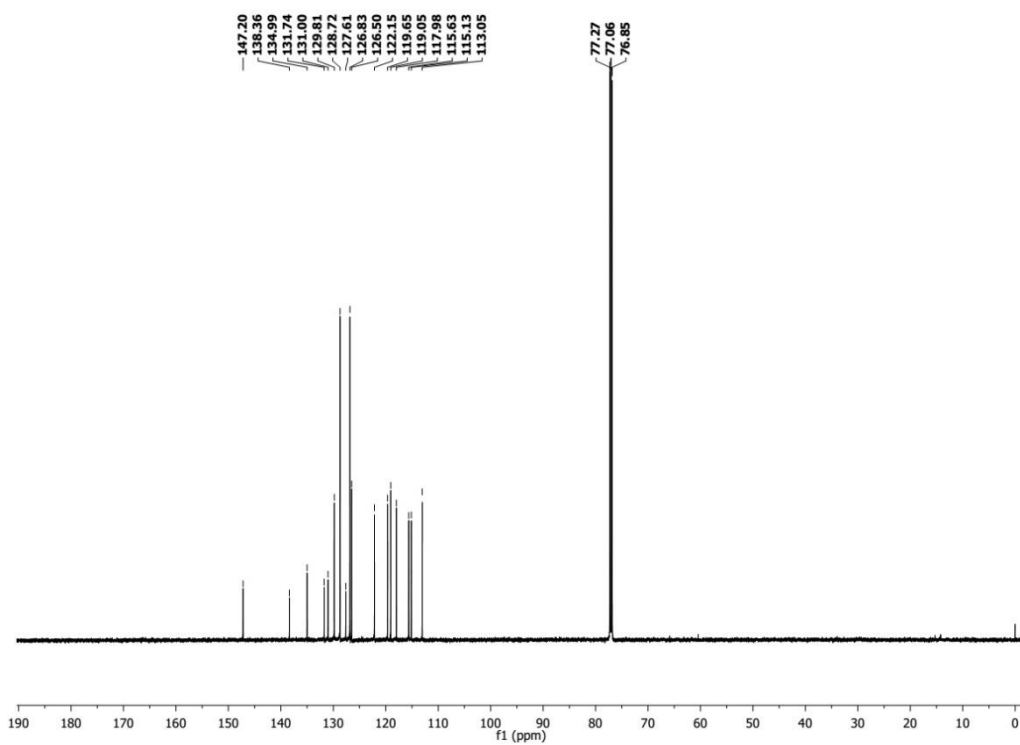
Compound (3g)

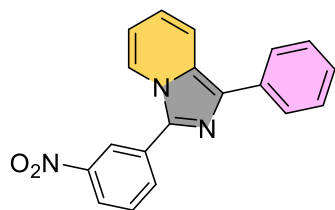
$^1\text{H}$  NMR of Compound 3g





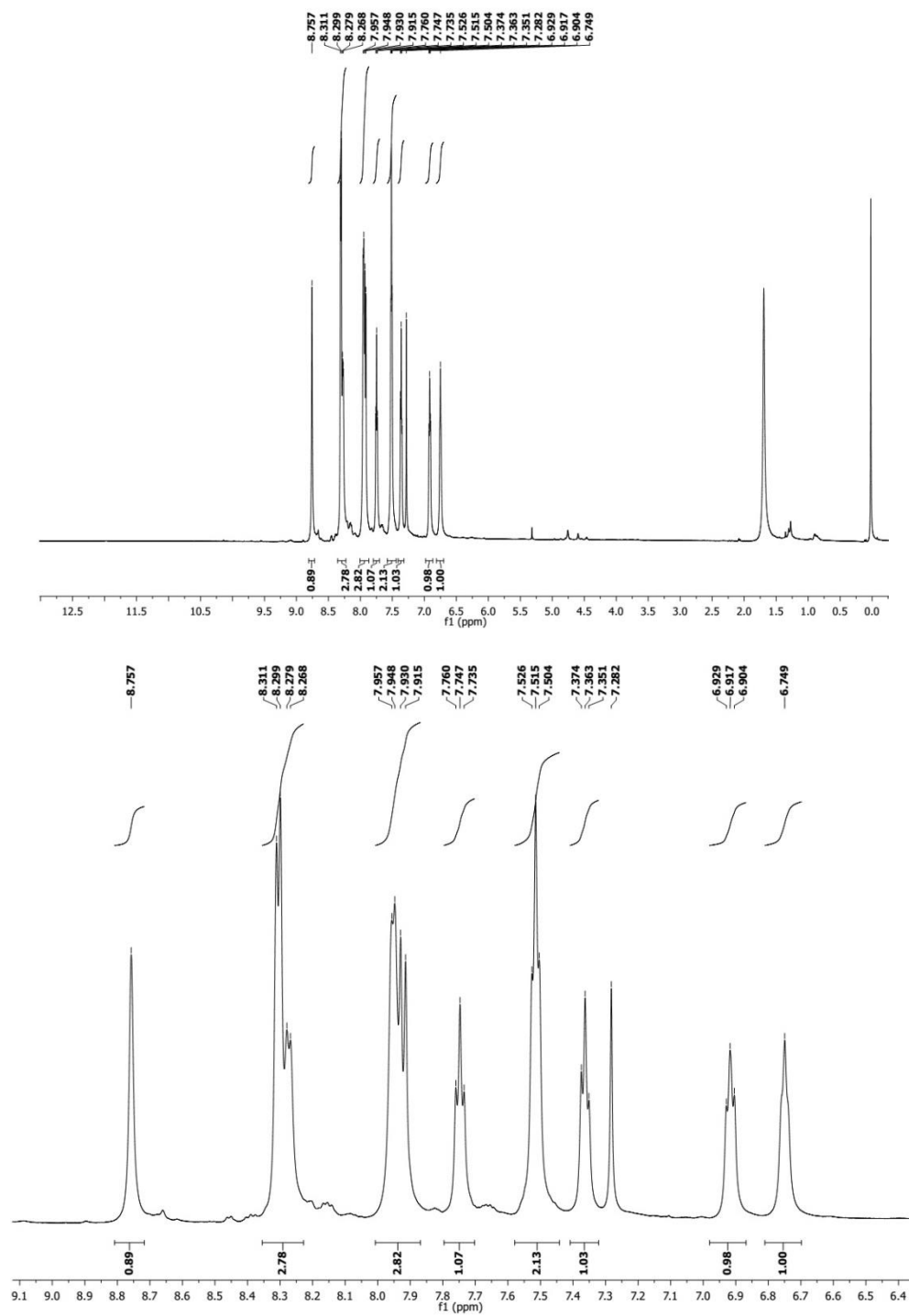
<sup>13</sup>C NMR of Compound 3g



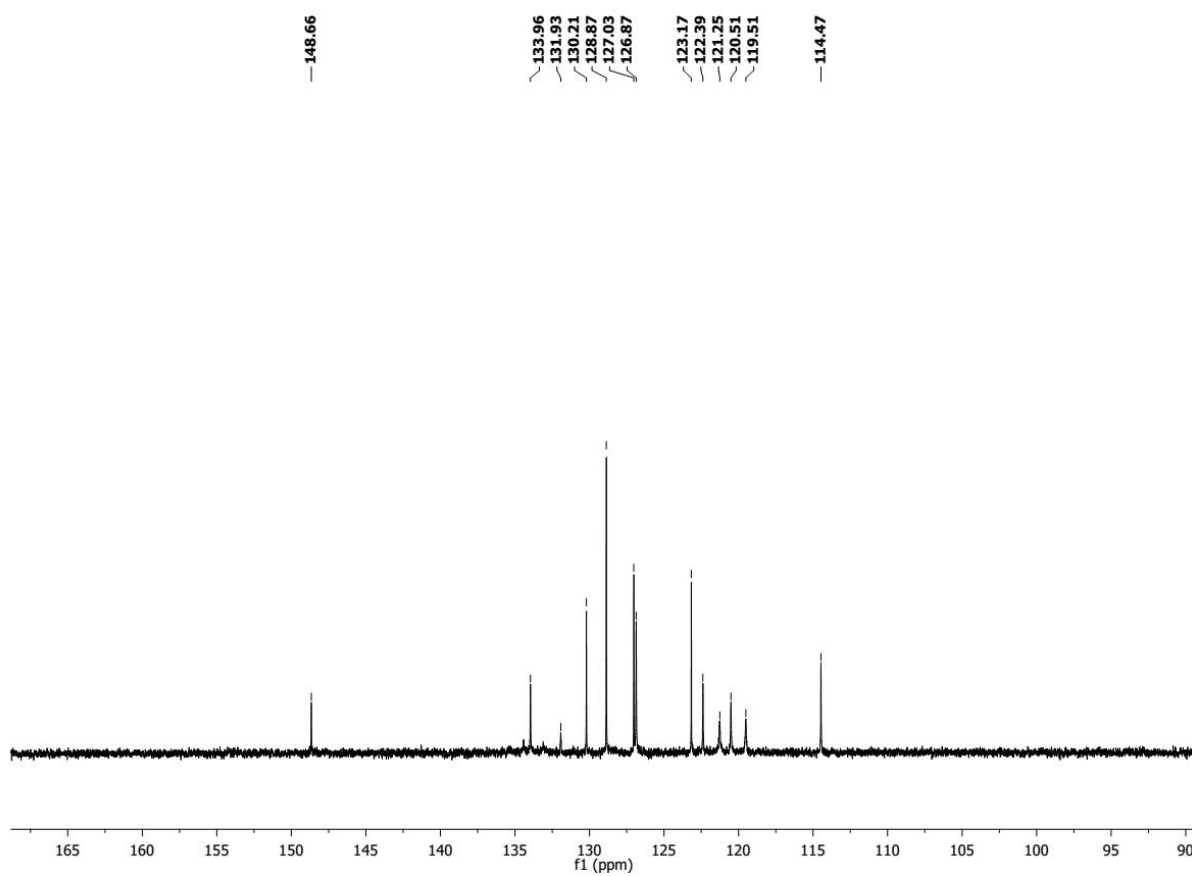
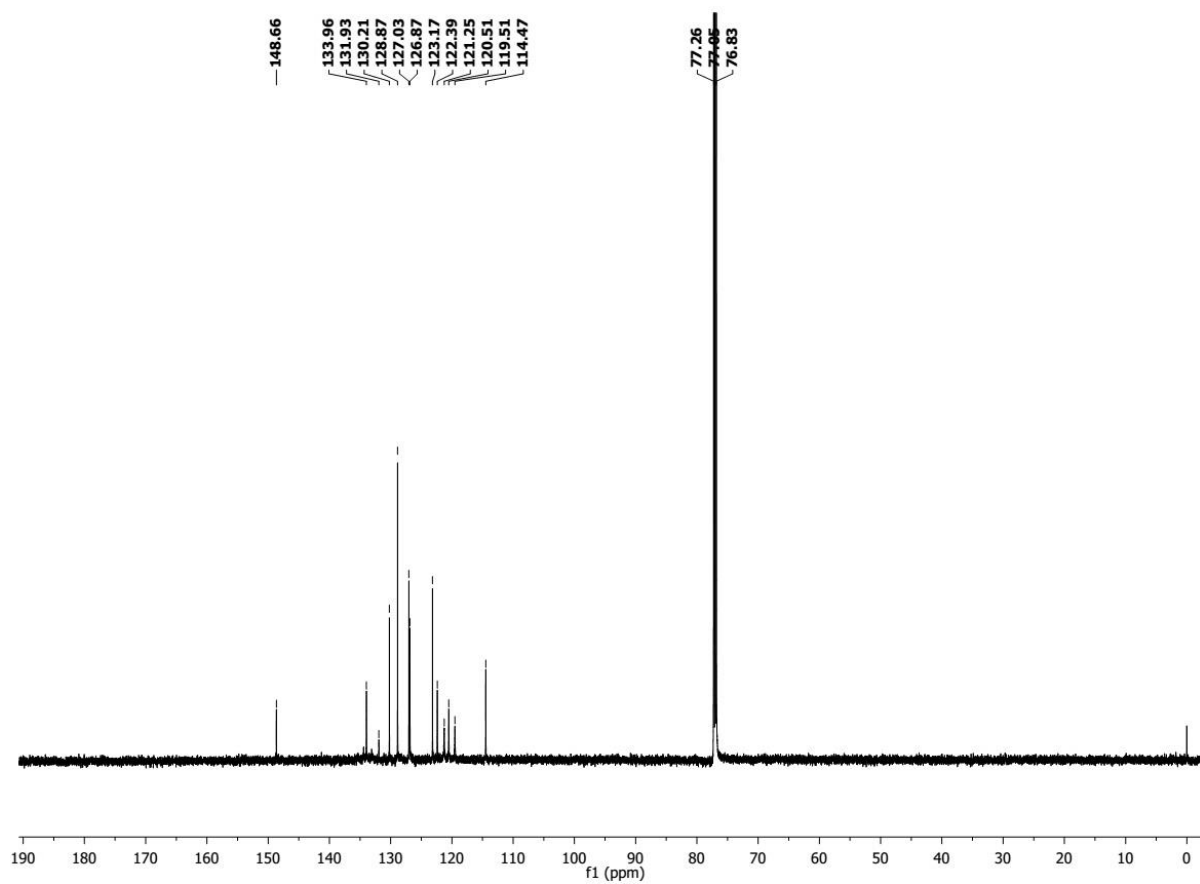


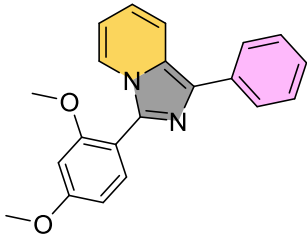
Compound (3h)

$^1\text{H}$  NMR of Compound 3h



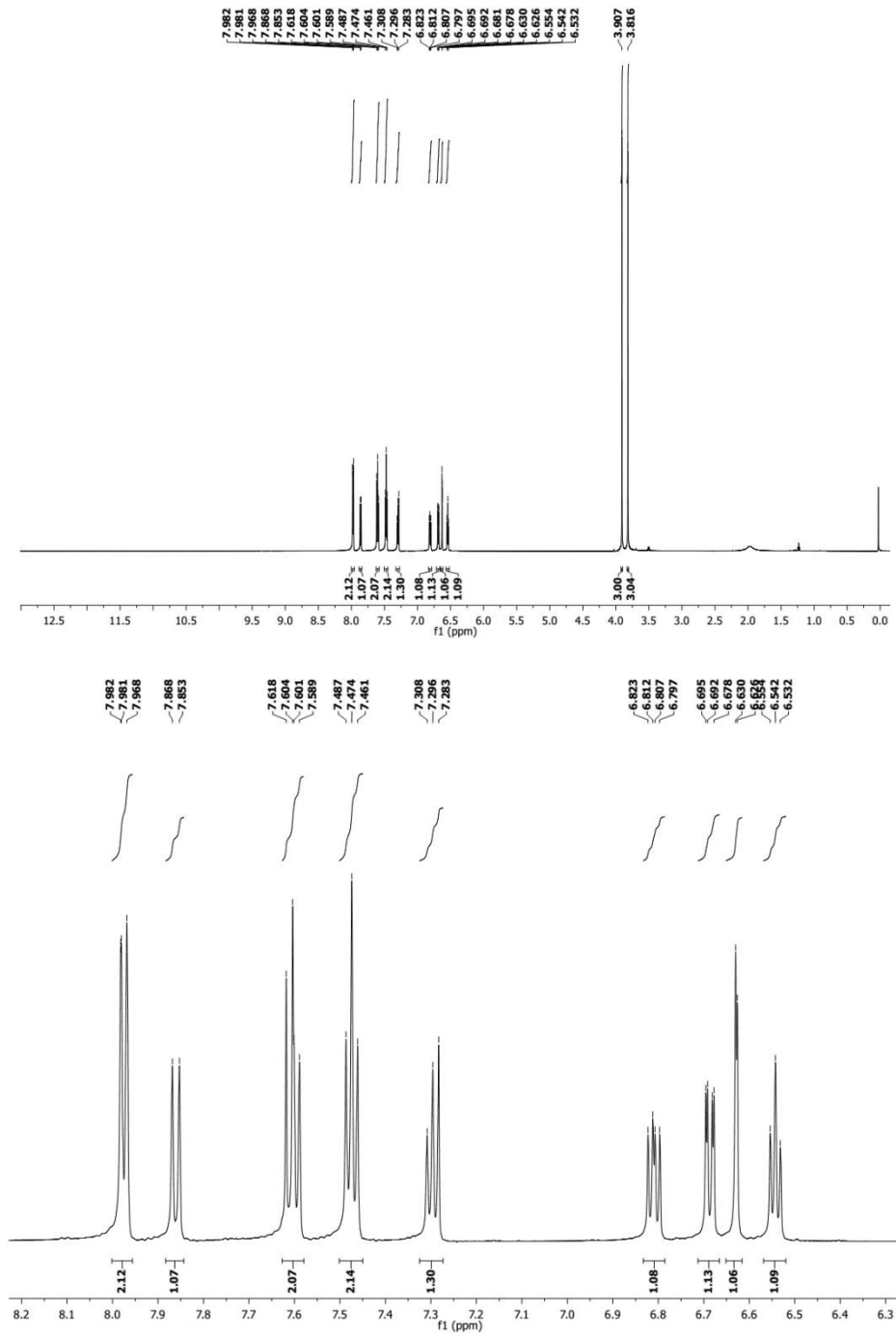
<sup>13</sup>C NMR of Compound 3h

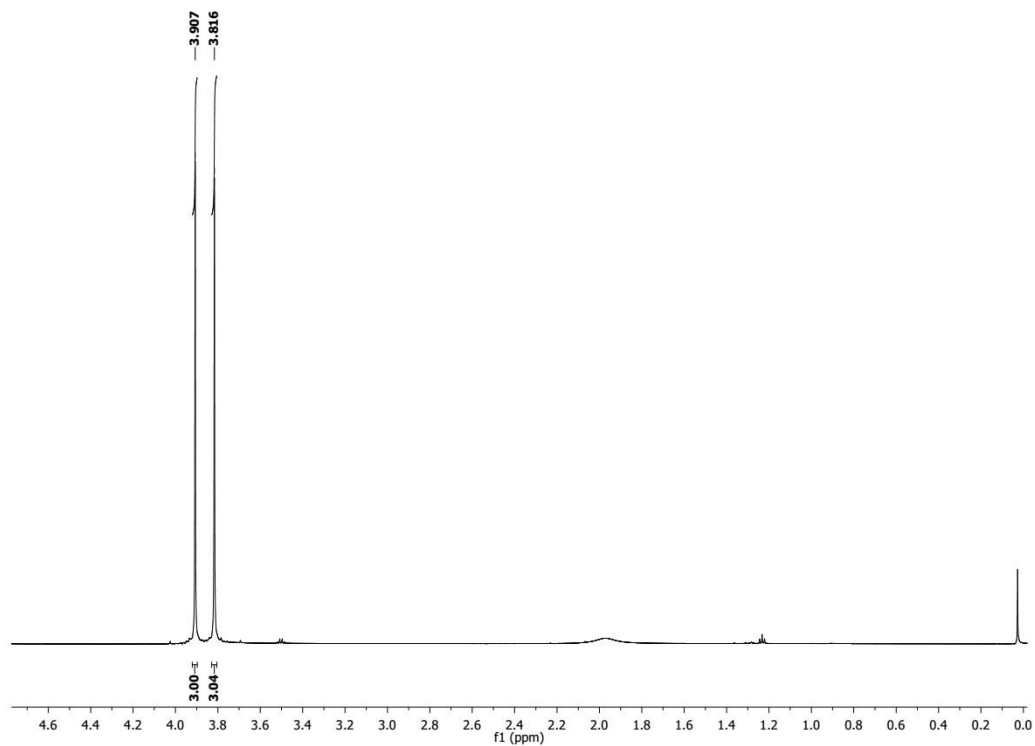




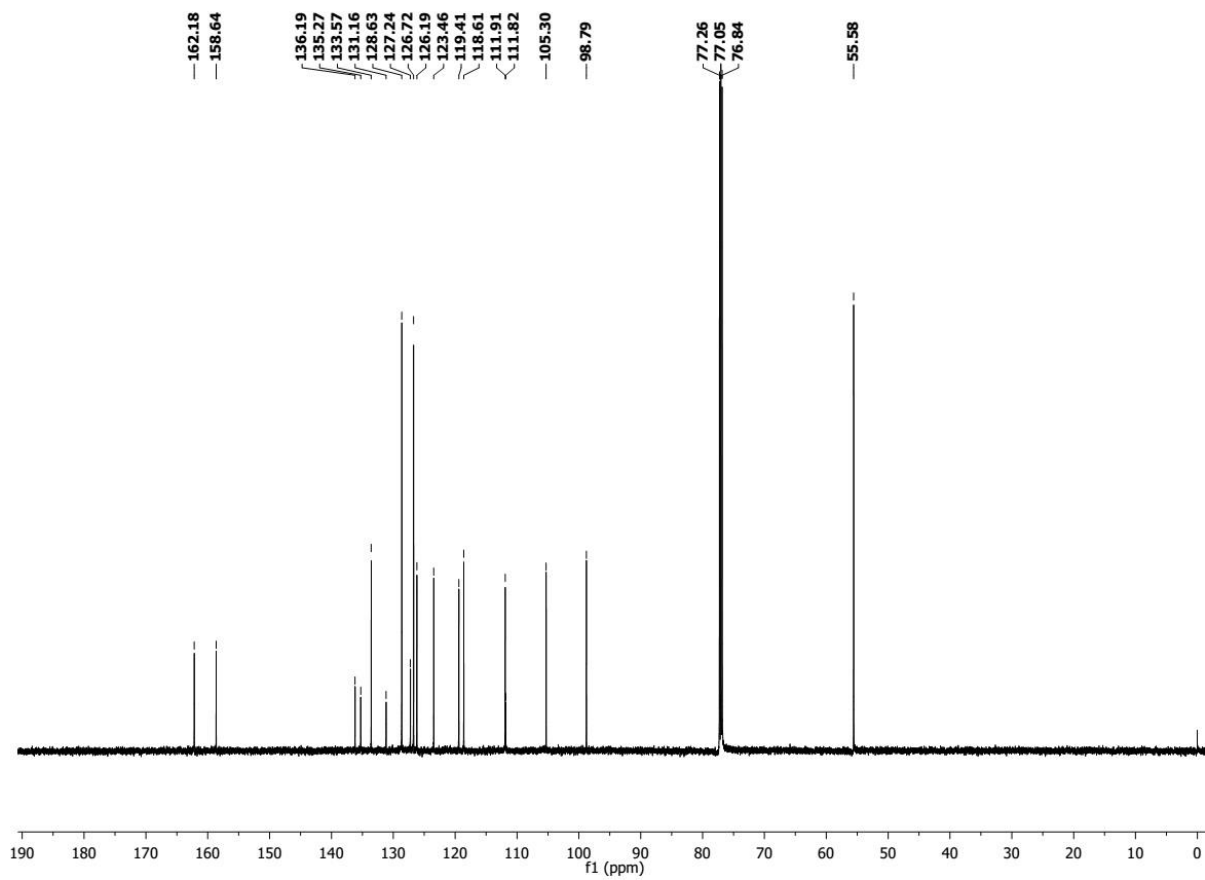
Compound (3i)

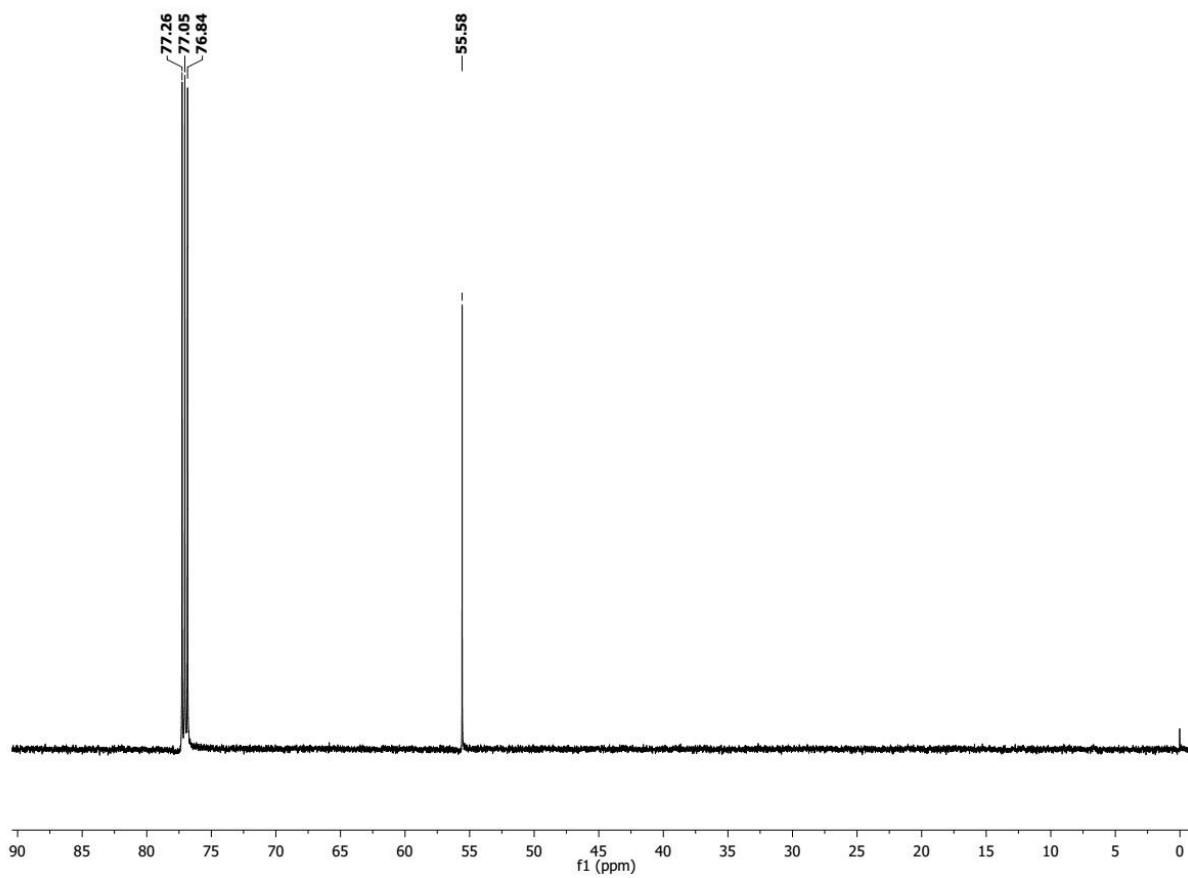
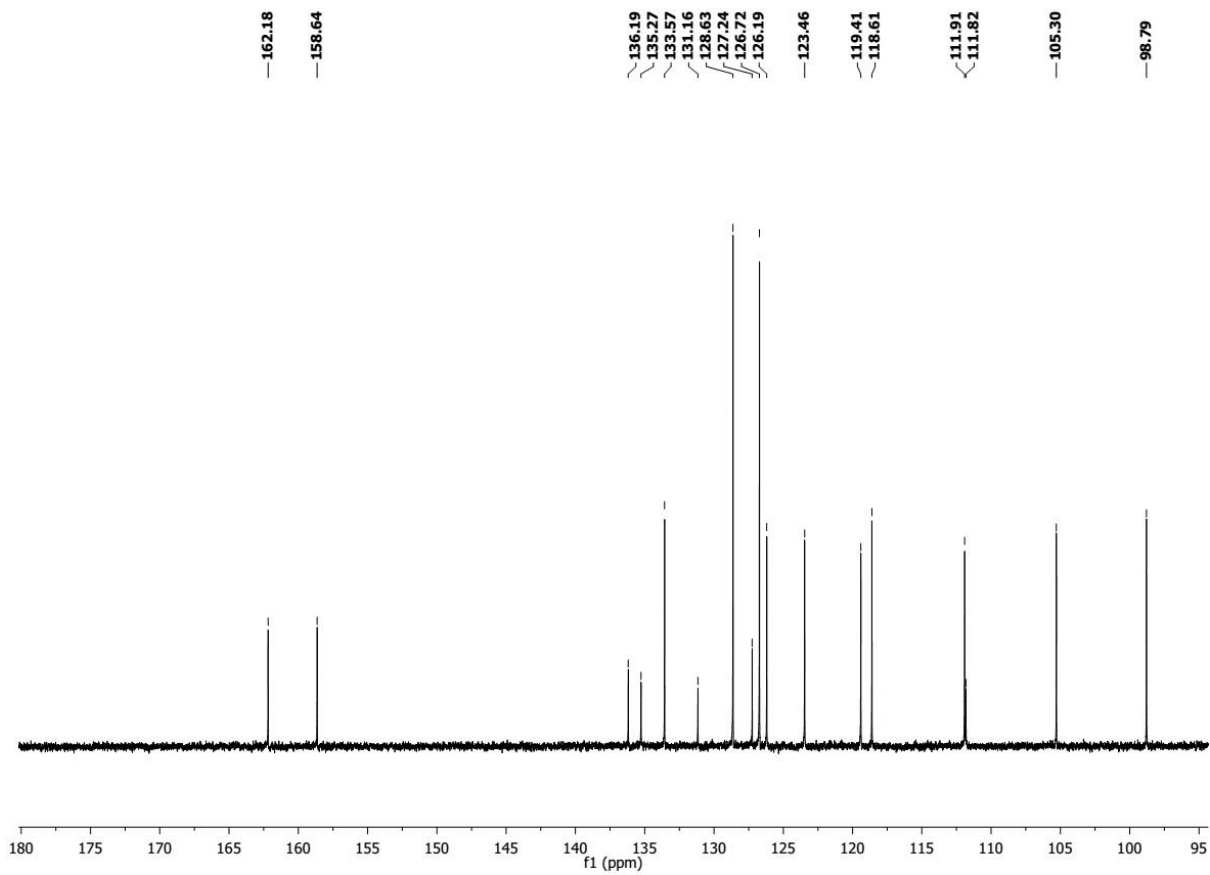
<sup>1</sup>H NMR of Compound 3i

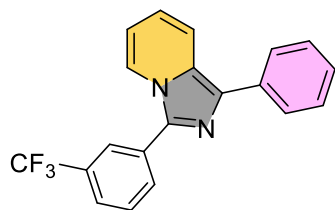




<sup>13</sup>C NMR of Compound 3i

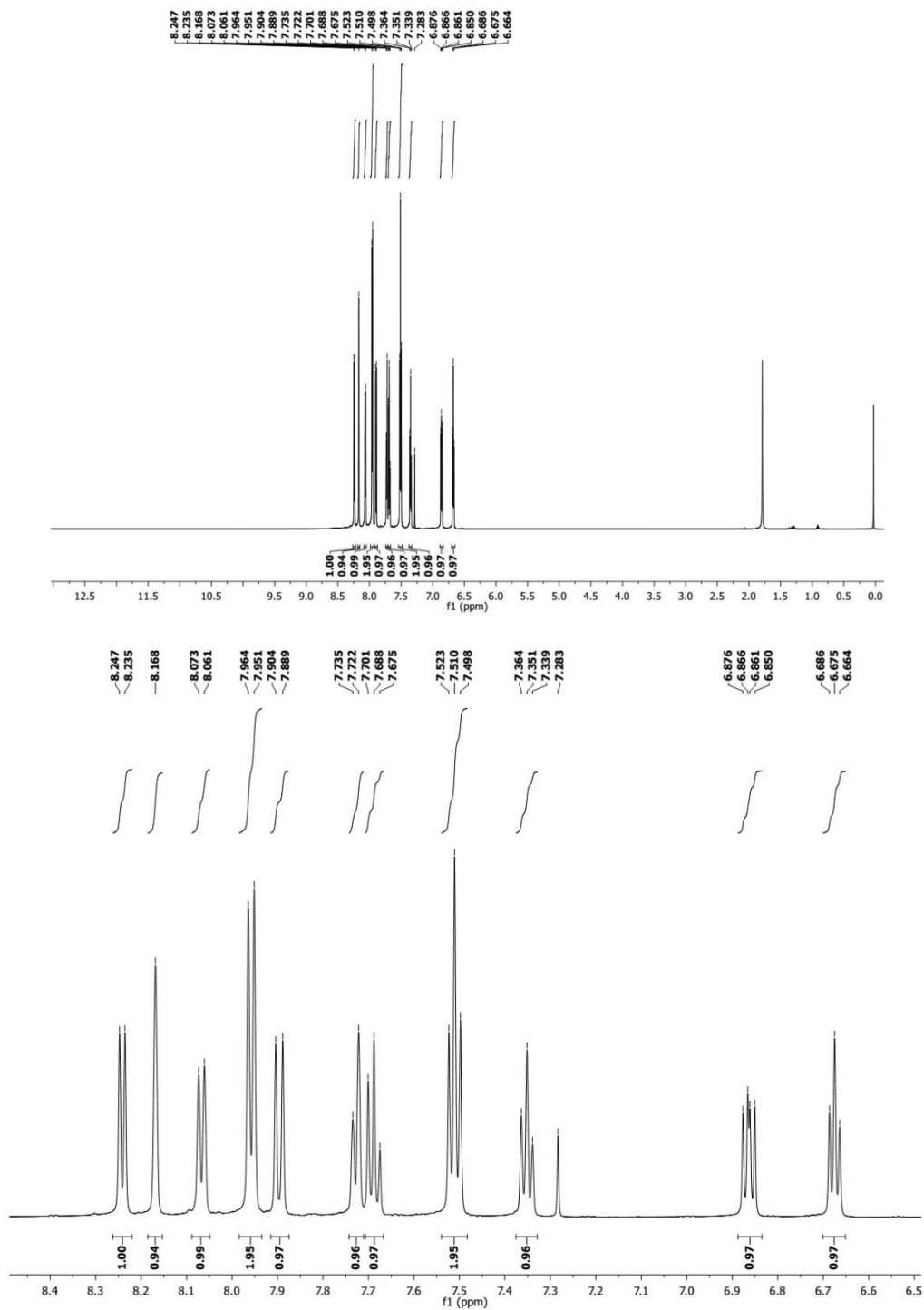




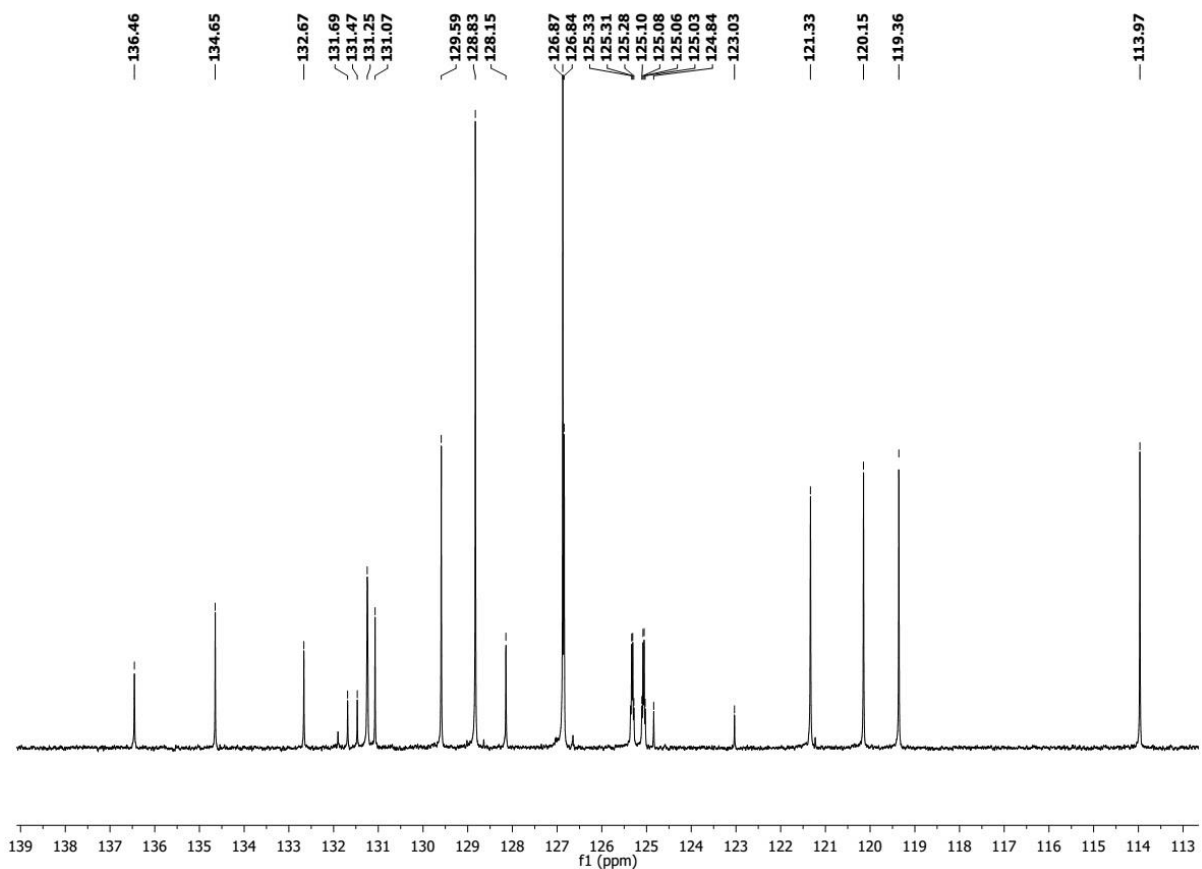
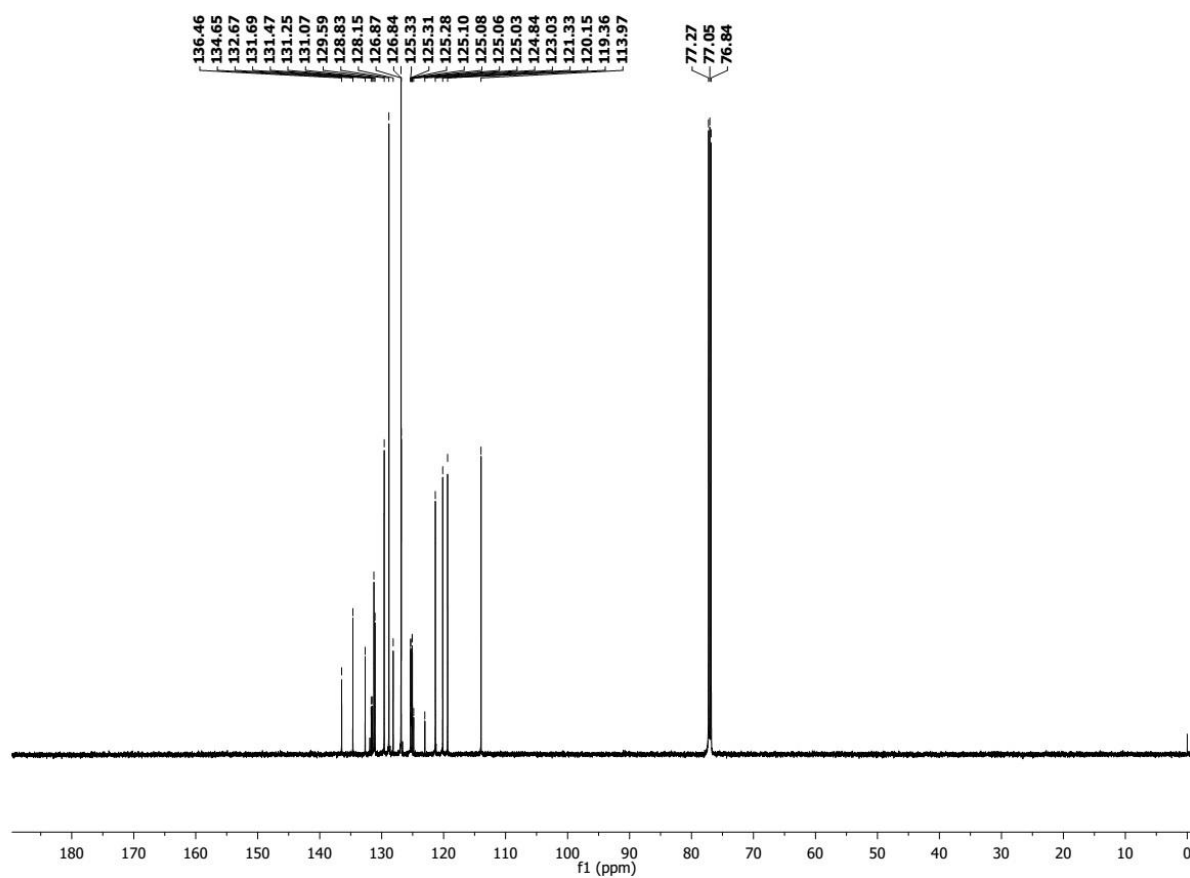


Compound (3j)

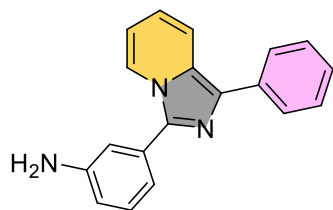
$^1\text{H}$  NMR of Compound 3j



<sup>13</sup>C NMR of Compound 3j

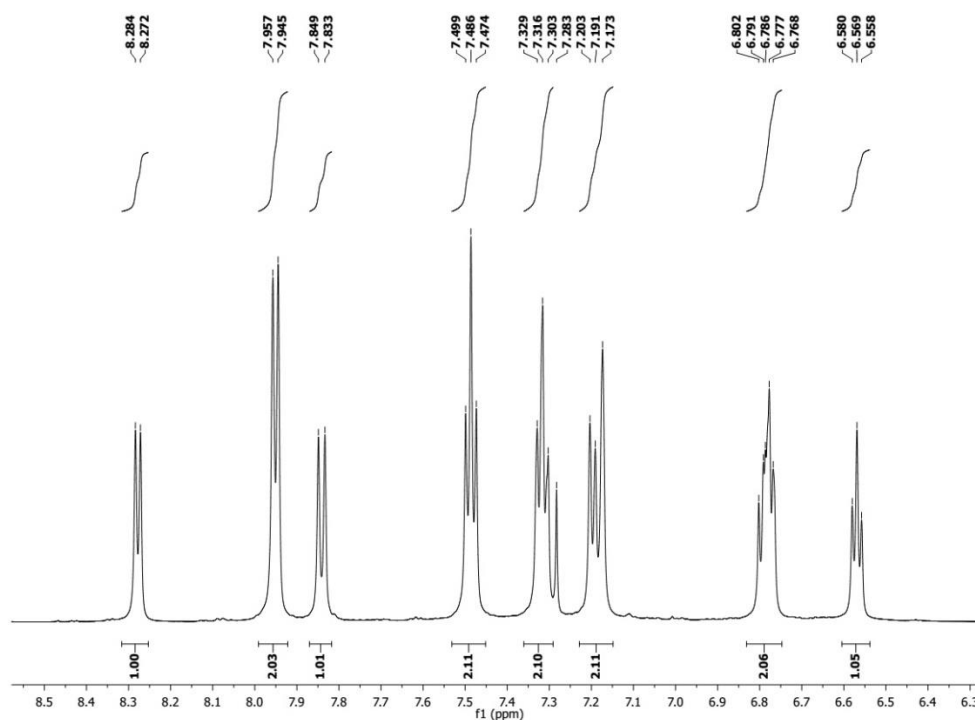
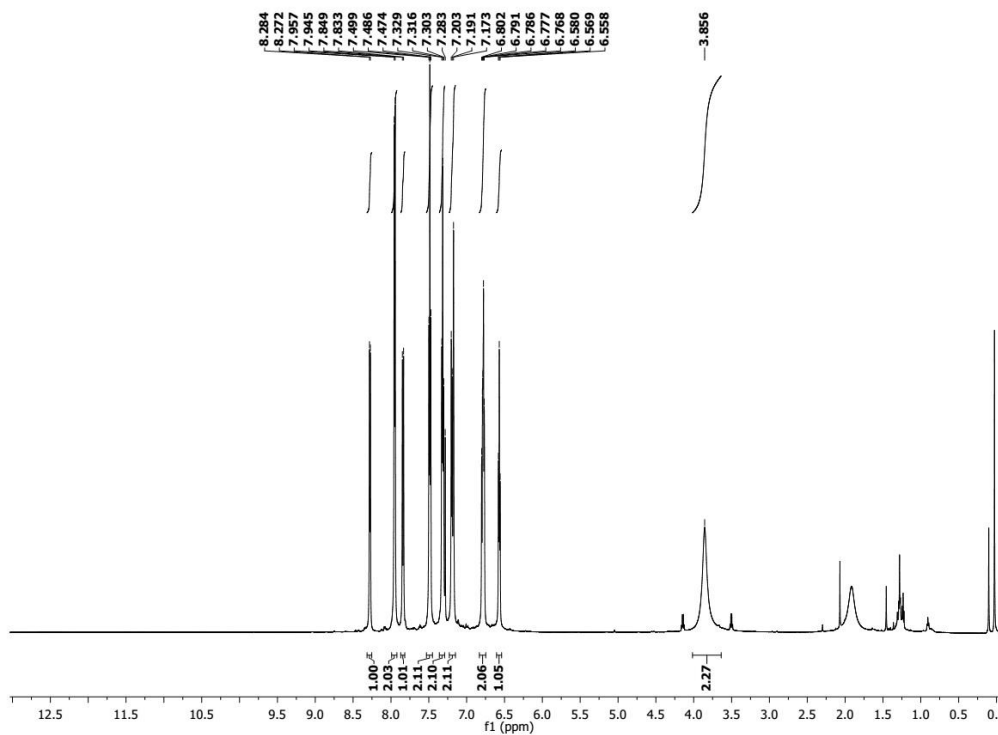




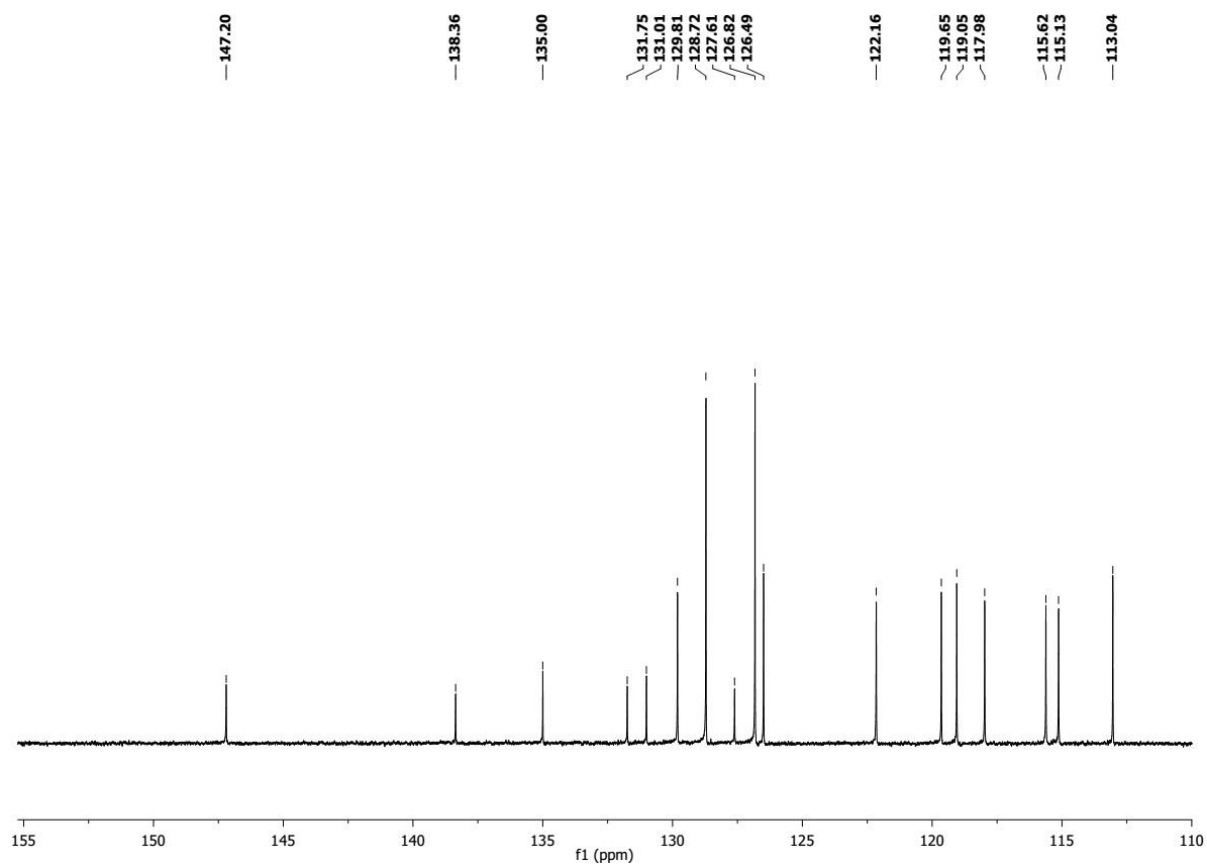
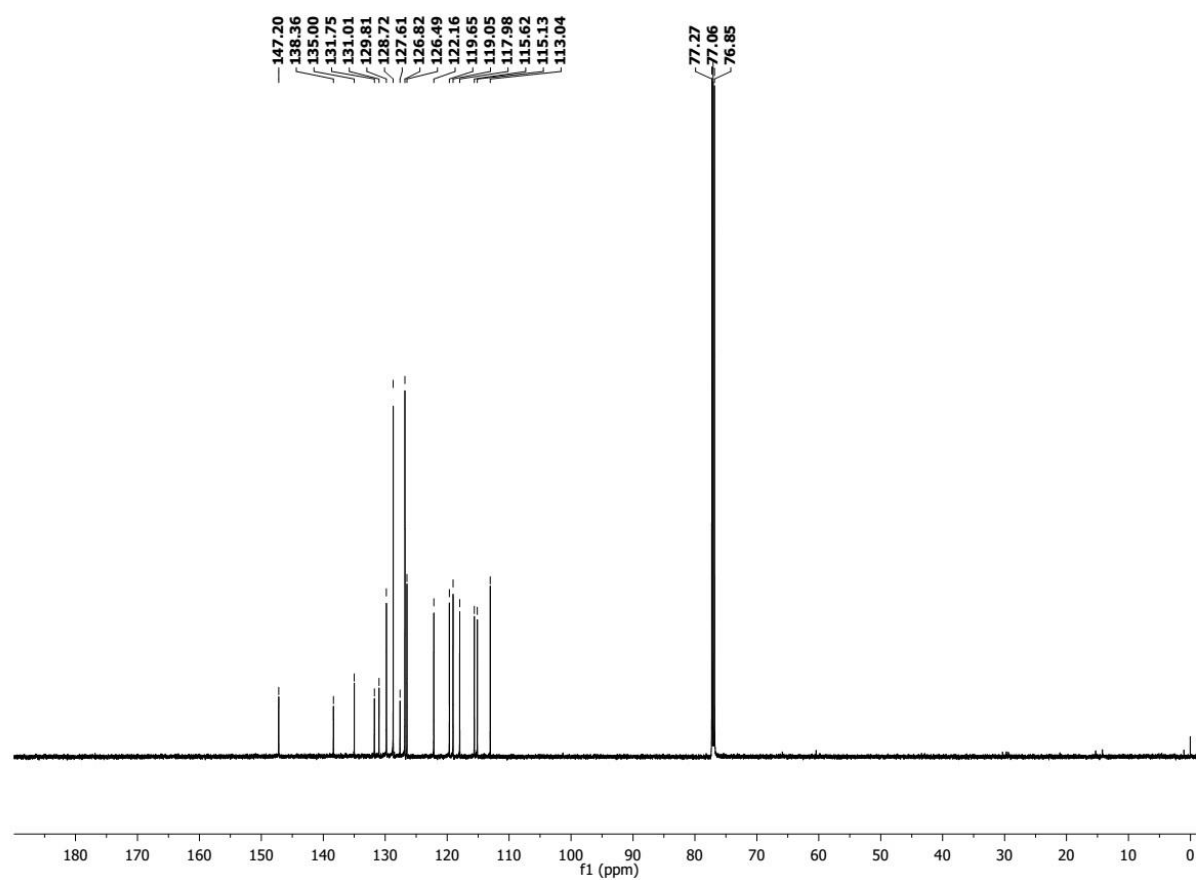


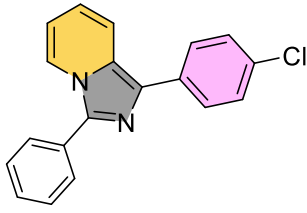
Compound (3k)

$^1\text{H}$  NMR of Compound 3k



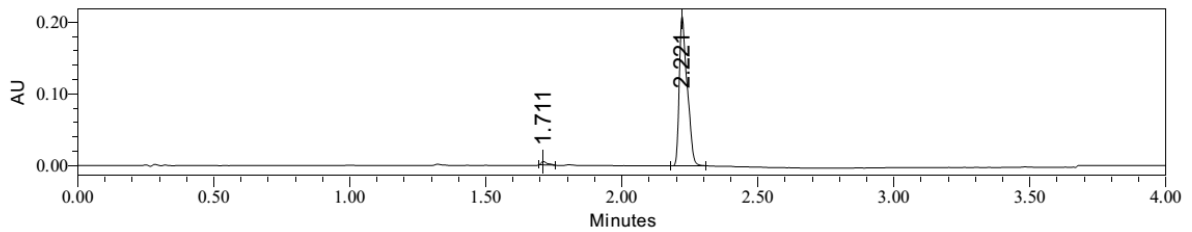
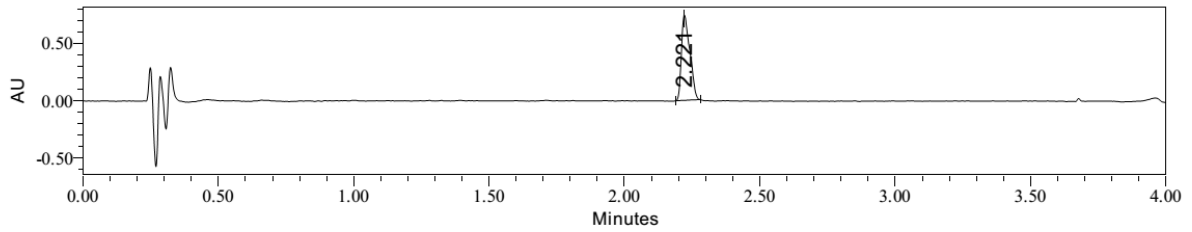
<sup>13</sup>C NMR of Compound 3k





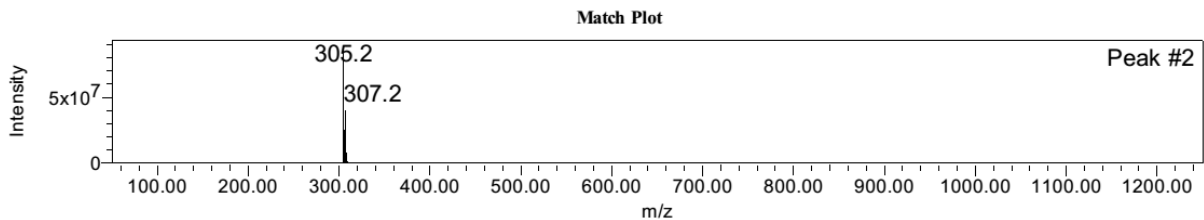
Compound (3l)

LC-MS analysis of Compound 3l



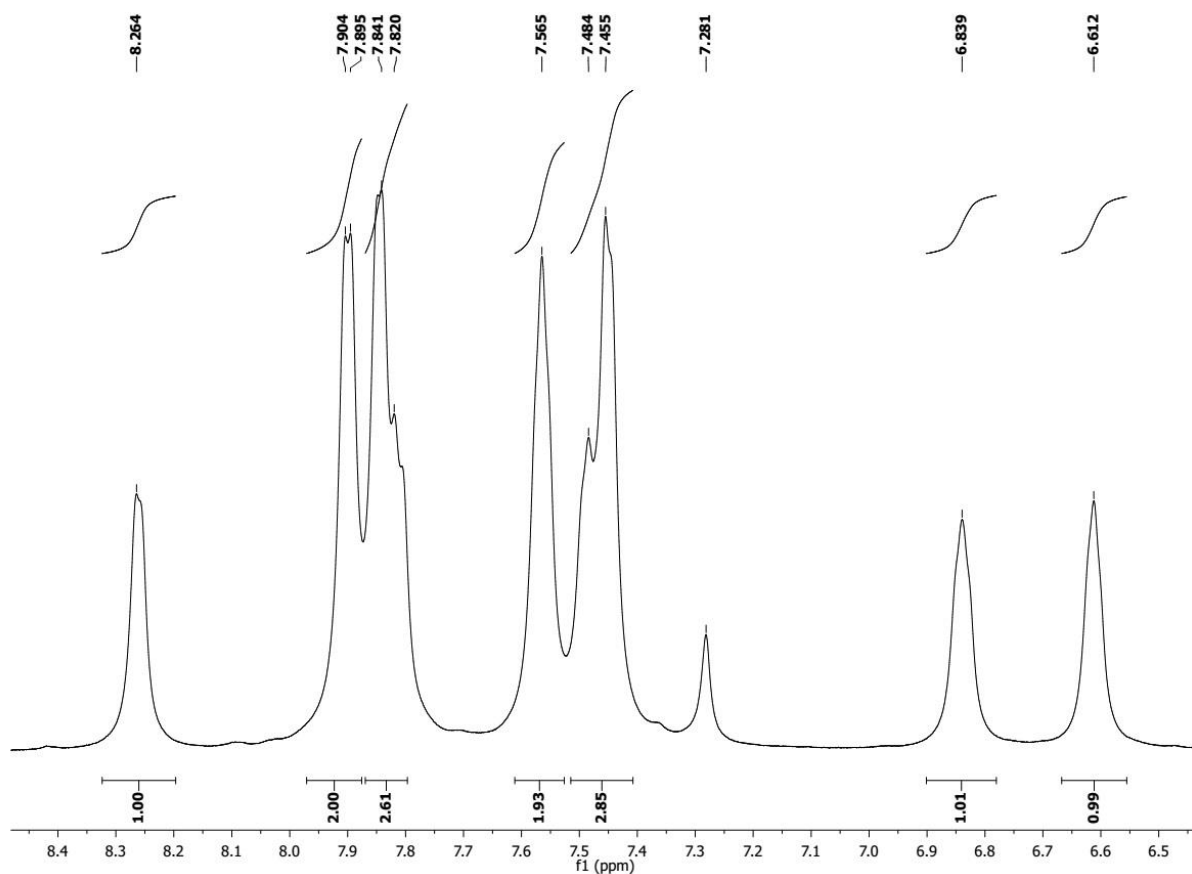
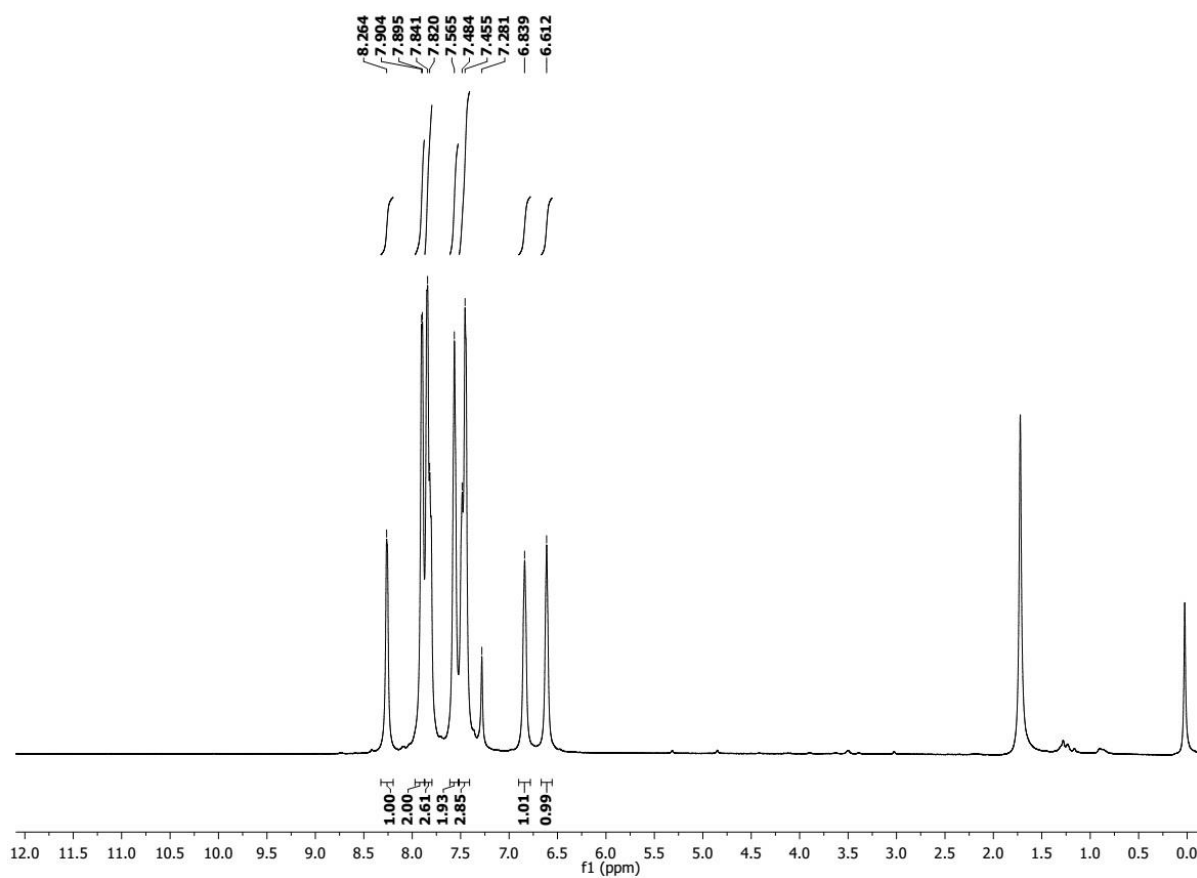
Peak Results  
Channel: PDA Spectrum

	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.711		4749	7451	1.74	PDA Spectrum	254.0nm
2	2.221		208743	421874	98.26	PDA Spectrum	254.0nm
3	2.221		743355	1605691	100.00	PDA Spectrum	210.0nm

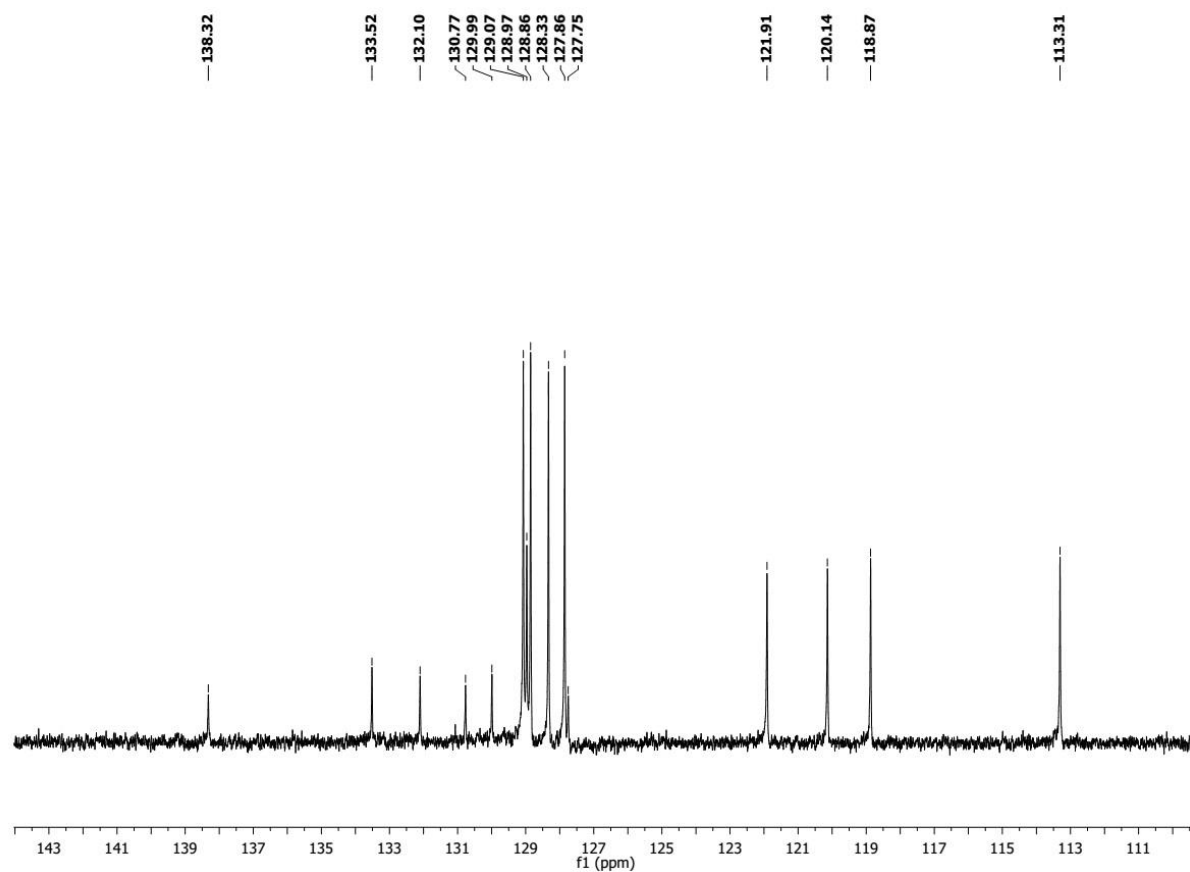
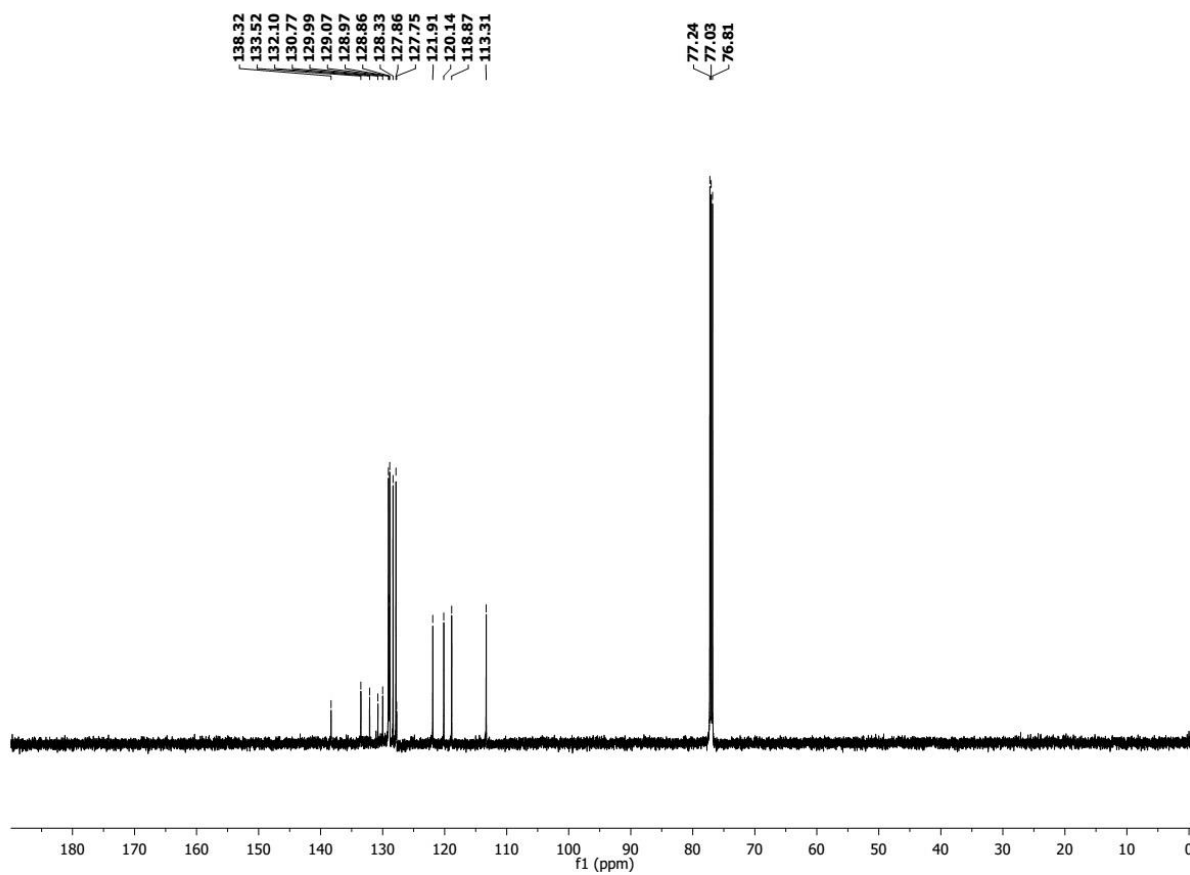


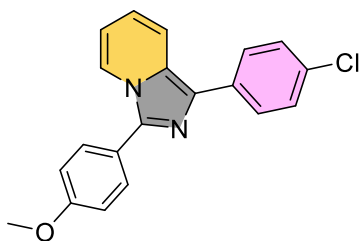
Base Peak 305.15 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (1.8:2.1;0.1:1.3;2.4:3.9) x 20.000 Th: 0.010 Retention Time 2.251

<sup>1</sup>H NMR of Compound 31



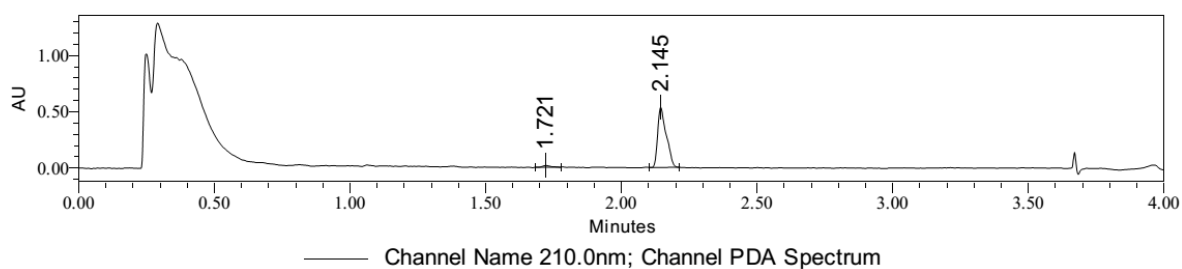
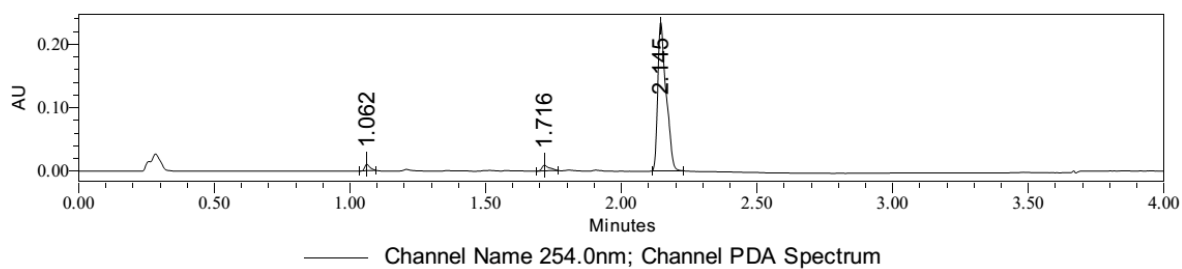
<sup>13</sup>C NMR of Compound 31





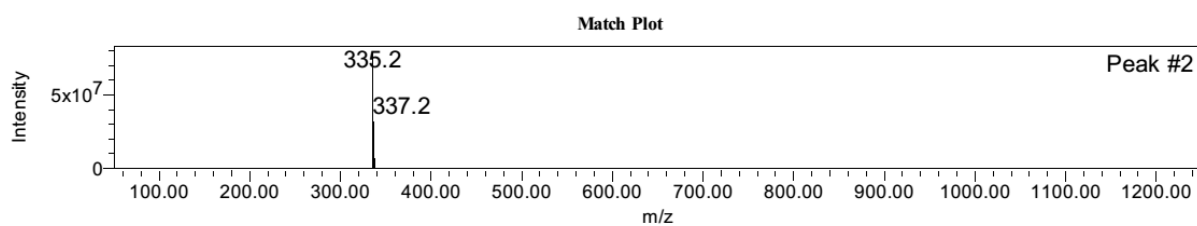
Compound (3m)

LC-MS analysis of Compound 3m



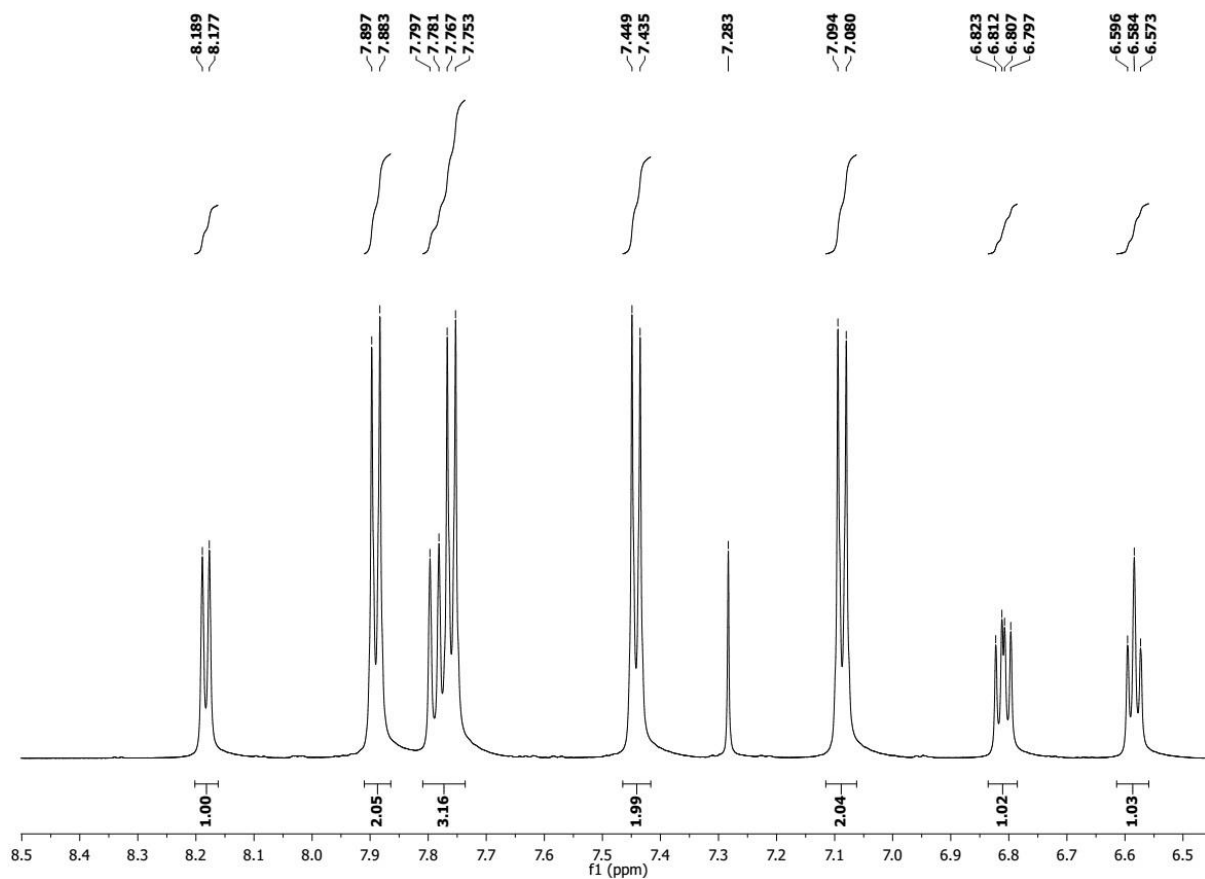
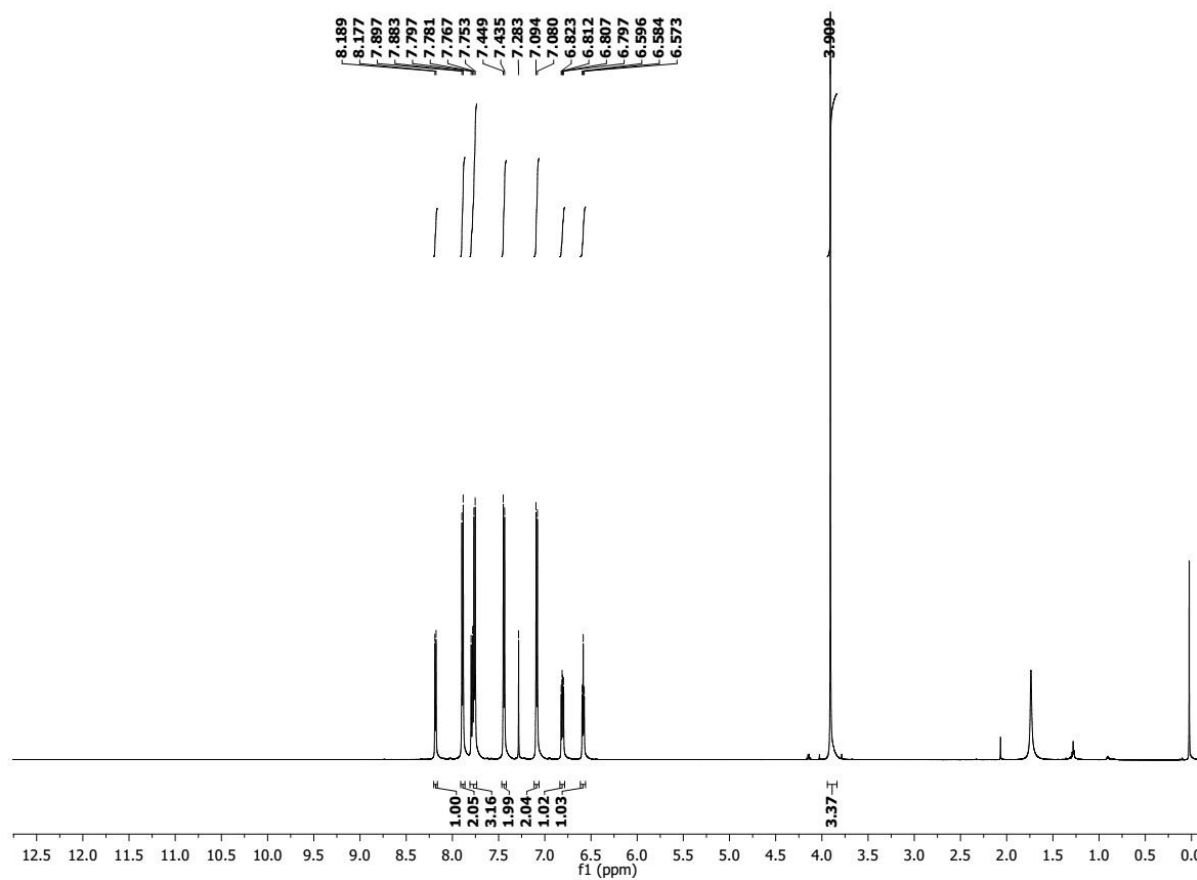
**Peak Results**  
Channel: PDA Spectrum

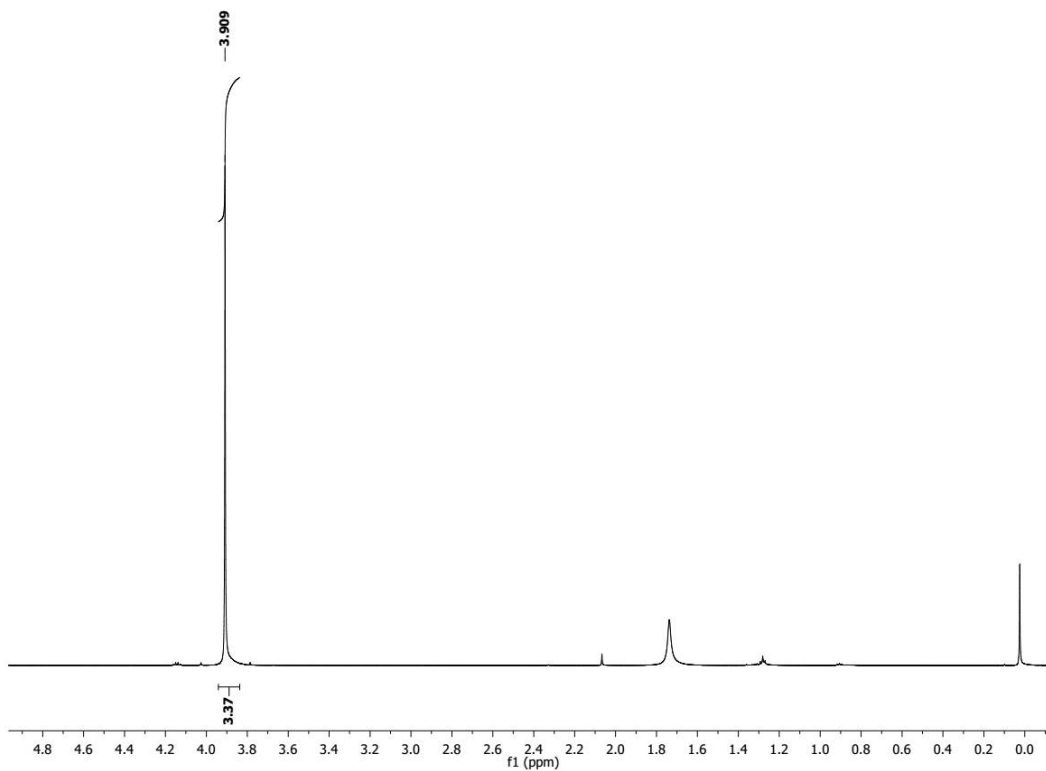
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.062		10211	13613	2.56	PDA Spectrum	254.0nm
2	1.716		8991	17321	3.26	PDA Spectrum	254.0nm
3	1.721		15545	33244	2.75	PDA Spectrum	210.0nm
4	2.145		533284	1177286	97.25	PDA Spectrum	210.0nm
5	2.145		234635	500175	94.18	PDA Spectrum	254.0nm



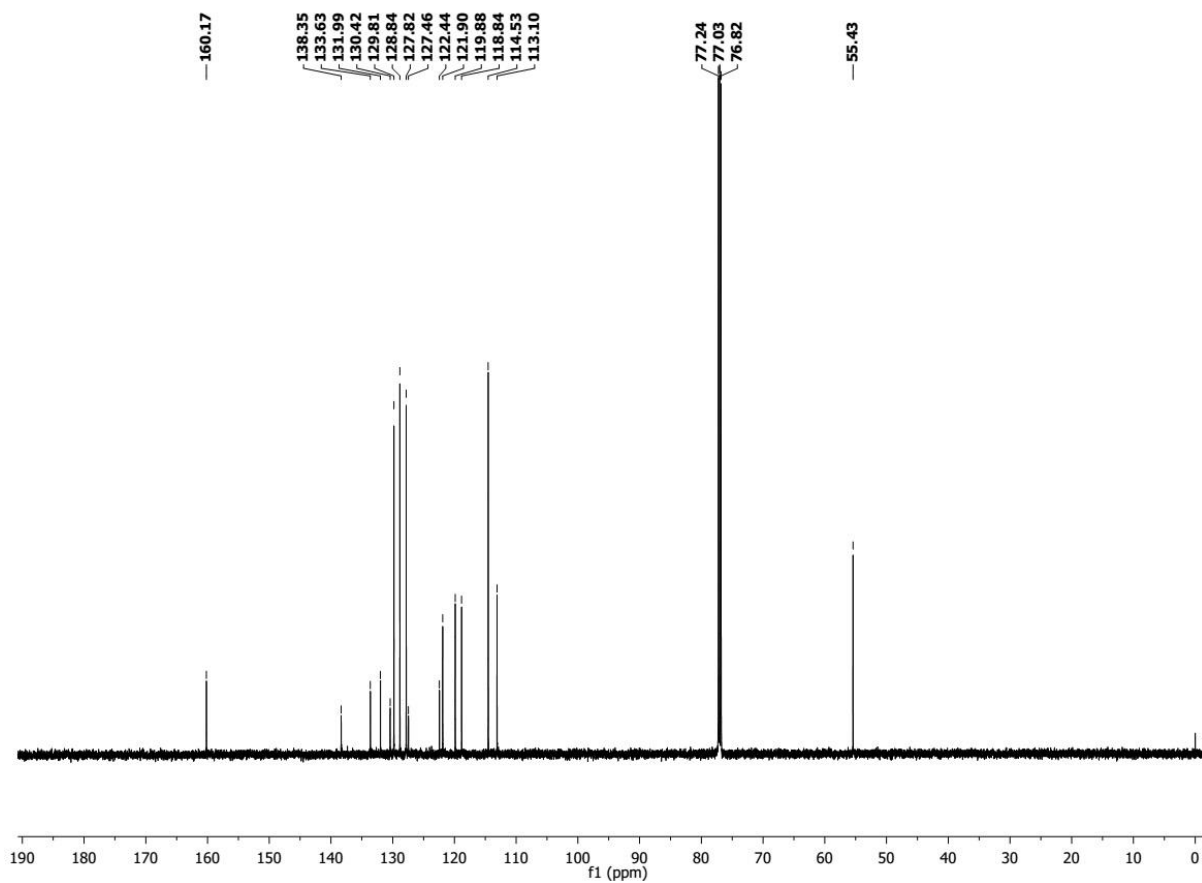
Base Peak 335.21 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (0.1:1.8;2.4:3.9) x 20.000 Th: 0.010 Retention Time 2.177

<sup>1</sup>H NMR of Compound 3m

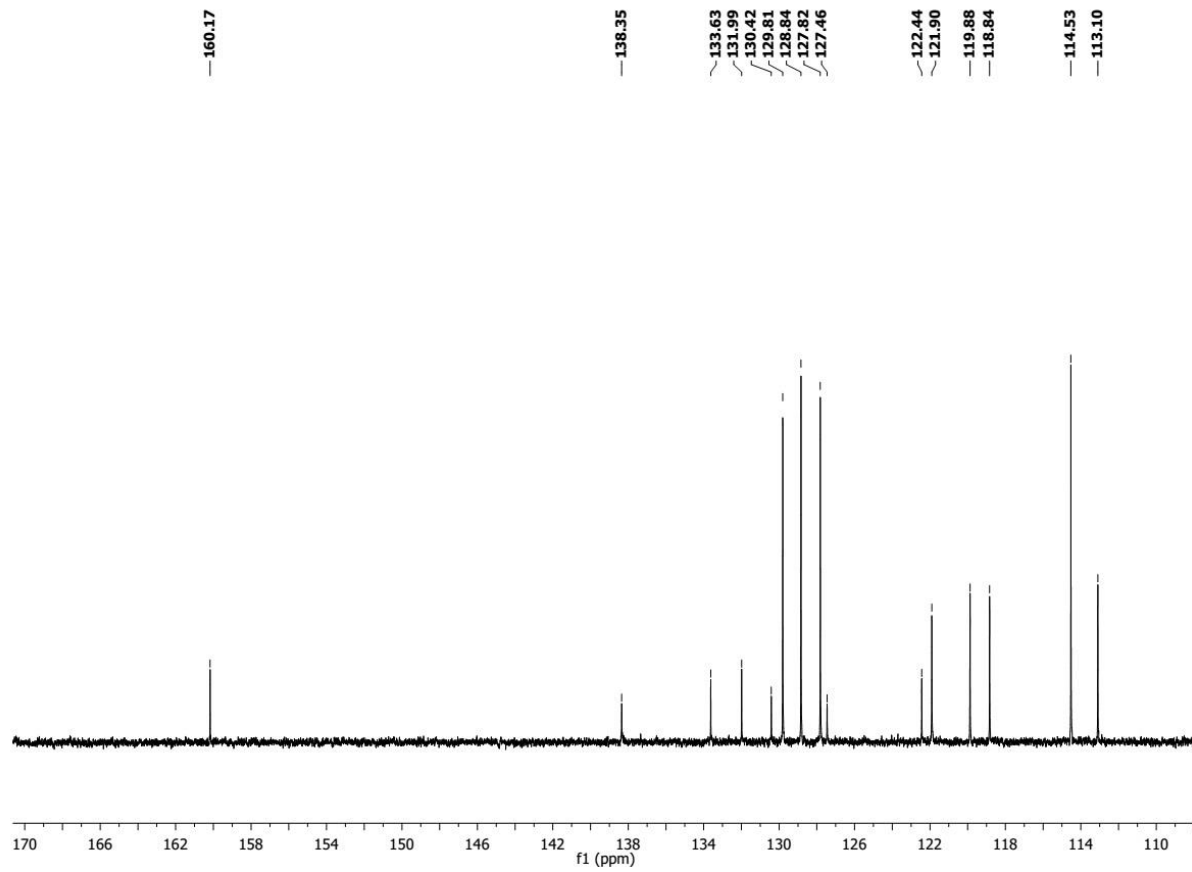
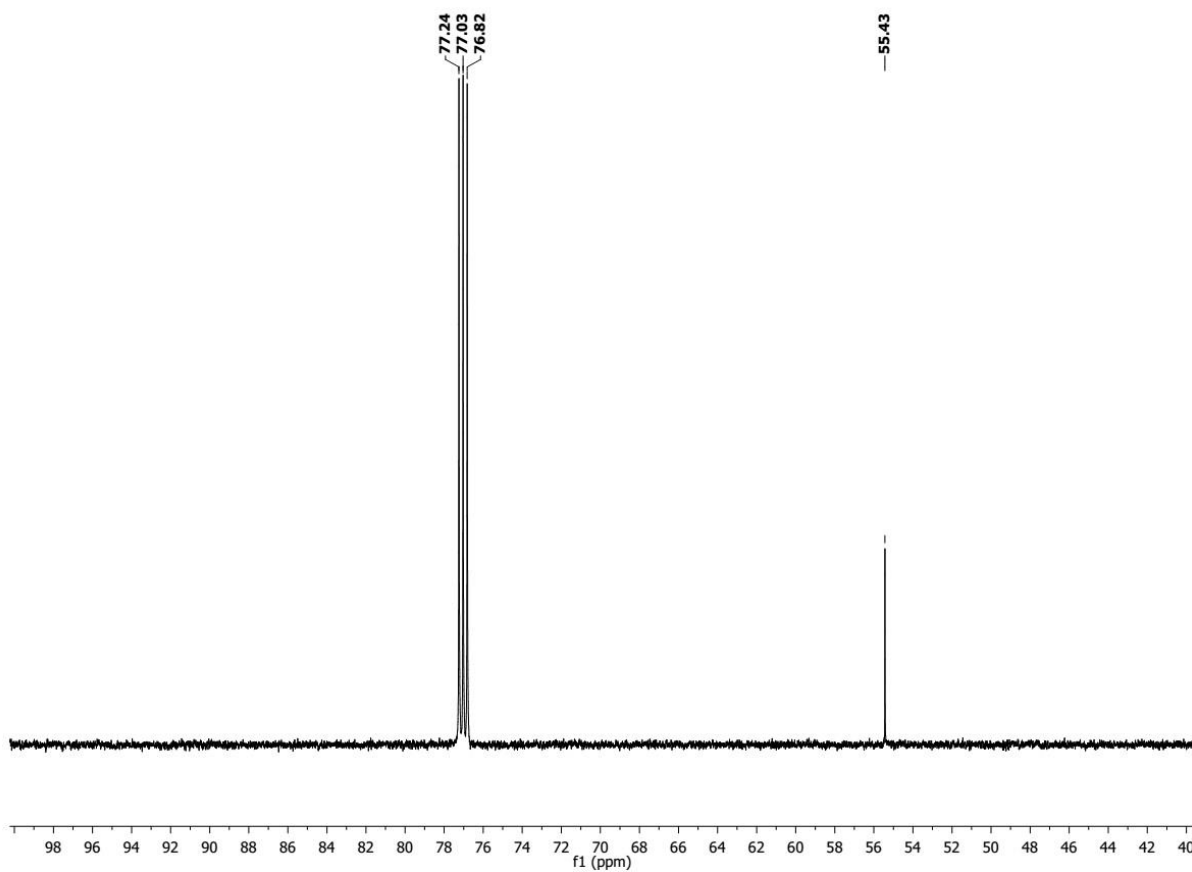


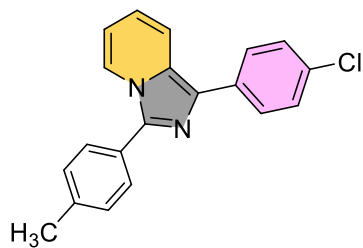


<sup>13</sup>C NMR of Compound 3m



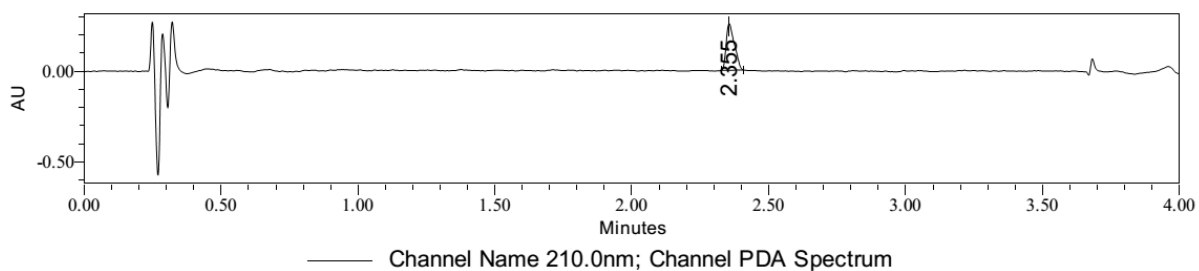






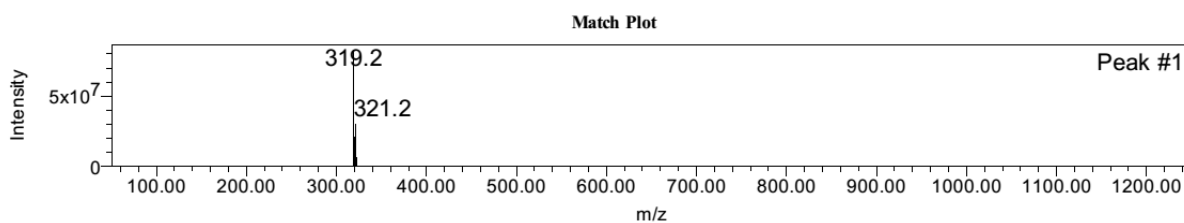
Compound (3n)

LC-MS analysis of Compound 3n



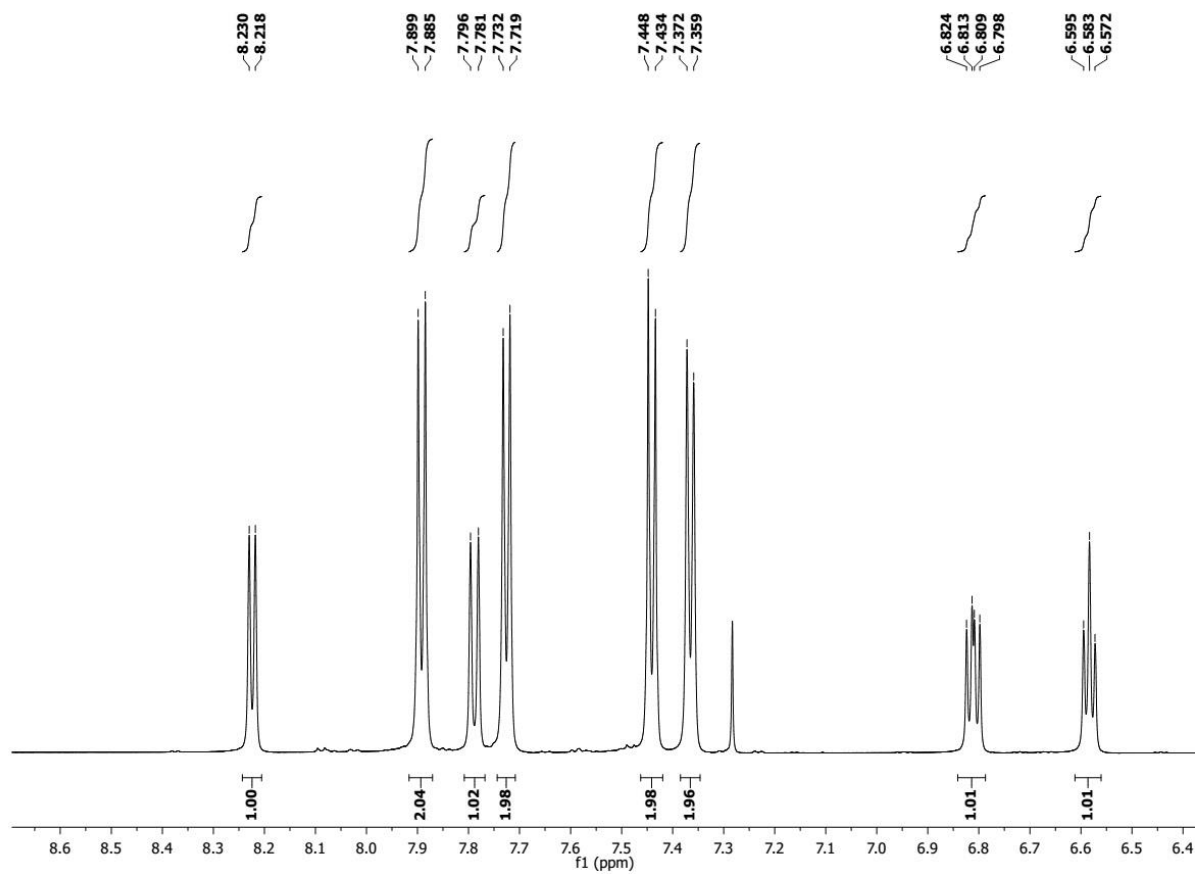
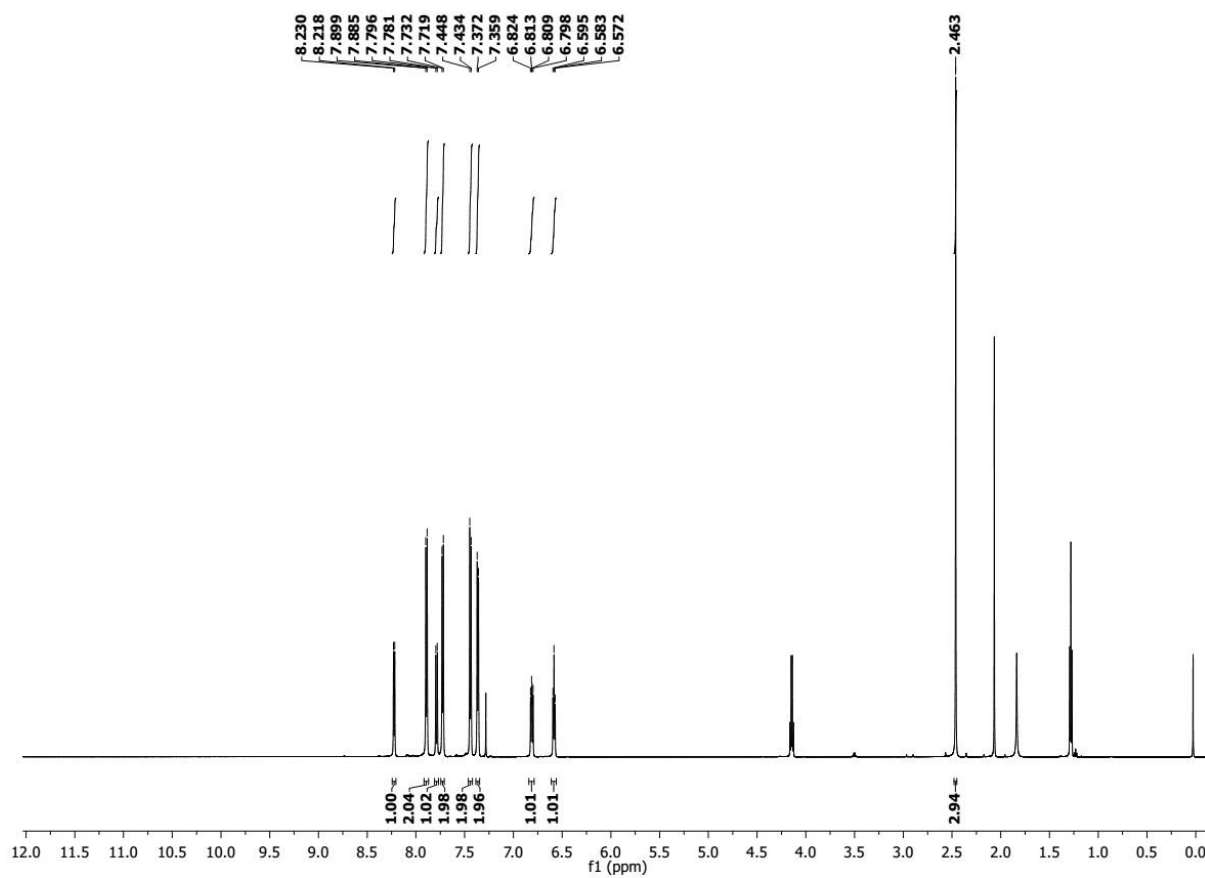
**Peak Results**  
**Channel: PDA Spectrum**

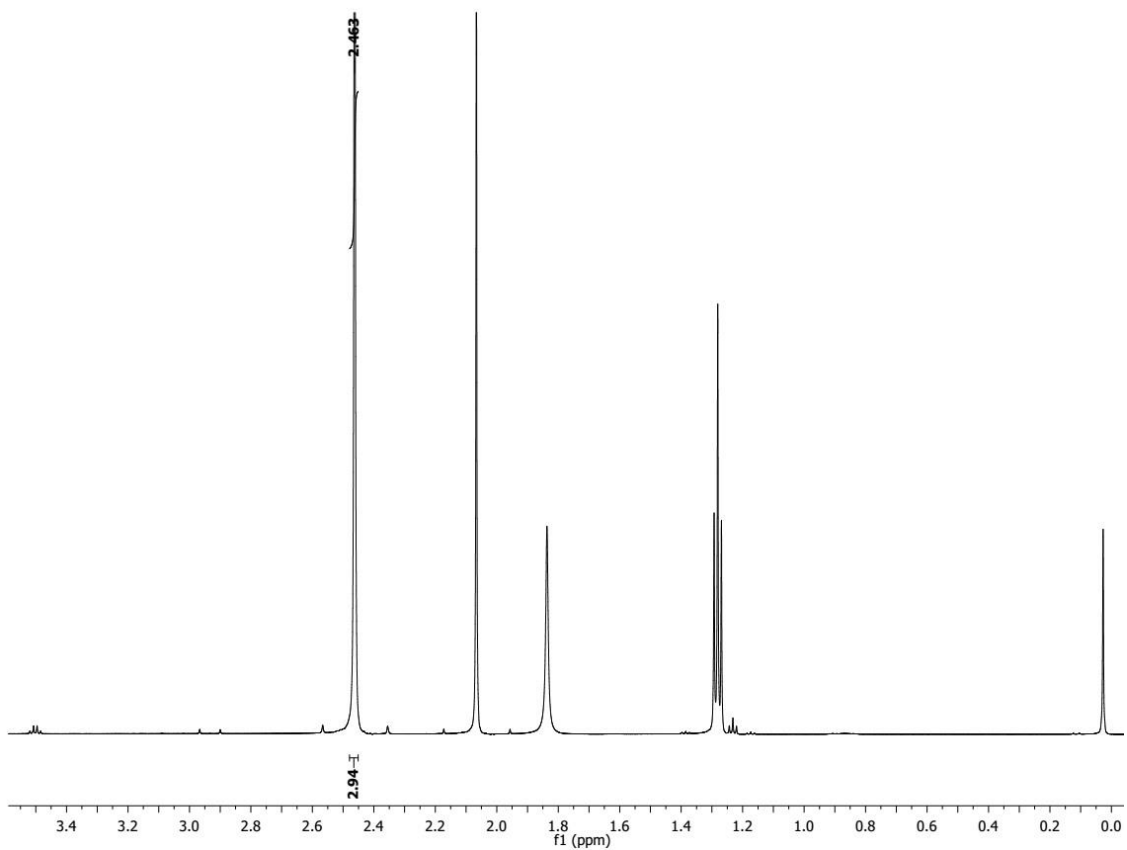
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	2.355		257651	531715	100.00	PDA Spectrum	210.0nm



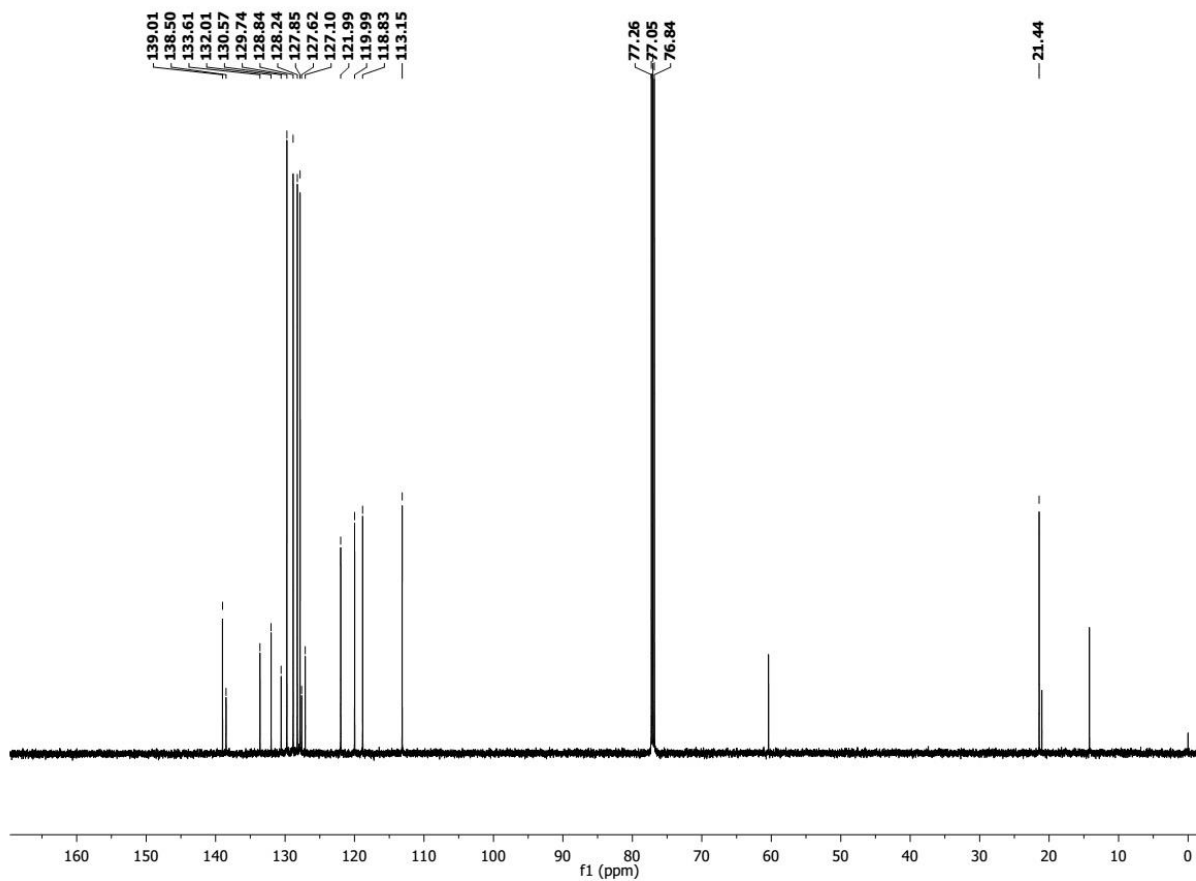
Base Peak 319.20 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (0.2:2.2;2.8:3.9) x 20.000 Th: 0.001 Retention Time 2.382

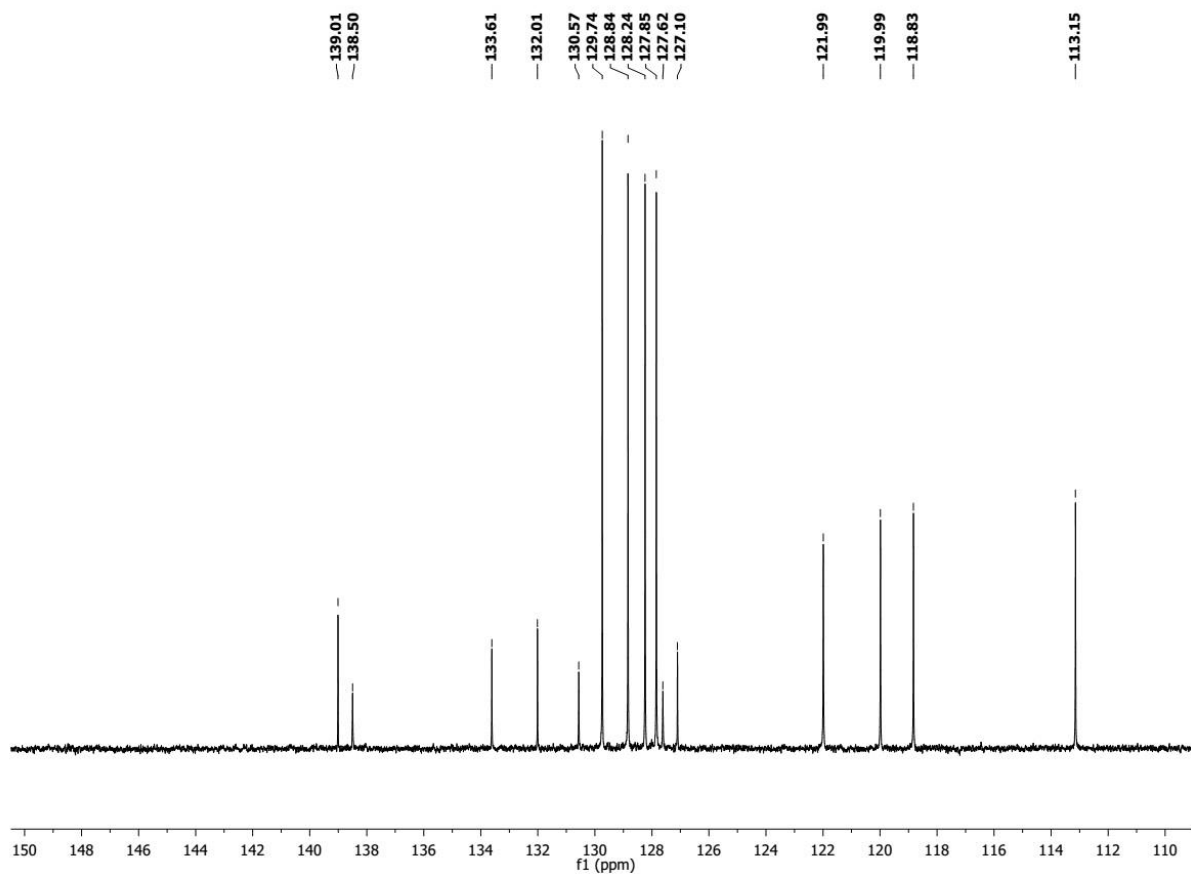
<sup>1</sup>H NMR of Compound 3n

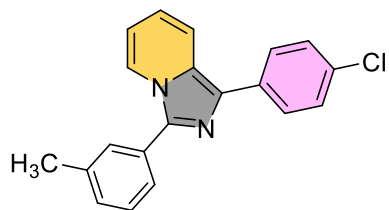




<sup>13</sup>C NMR of Compound 3n

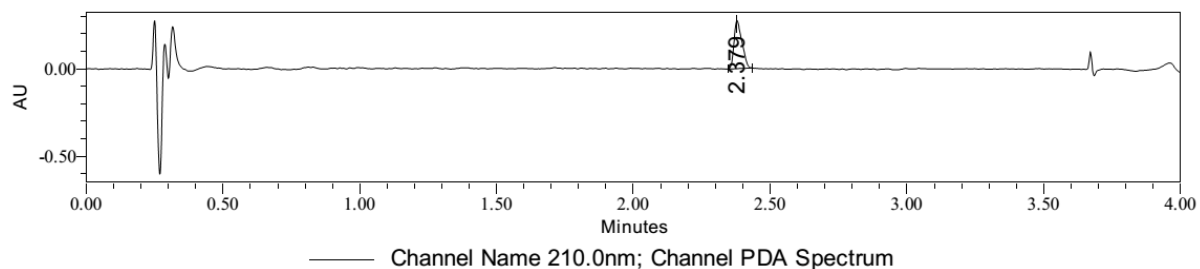






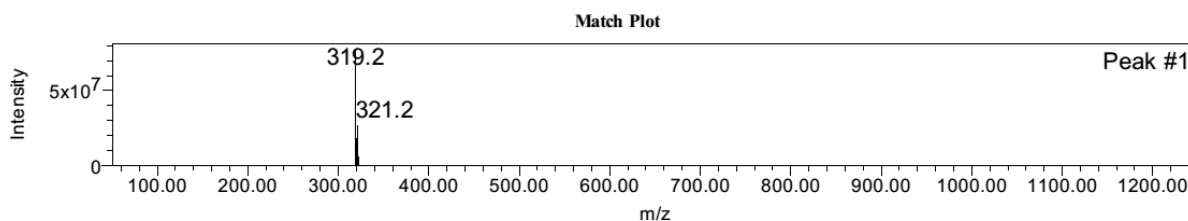
Compound (30)

LC-MS of Compound 30



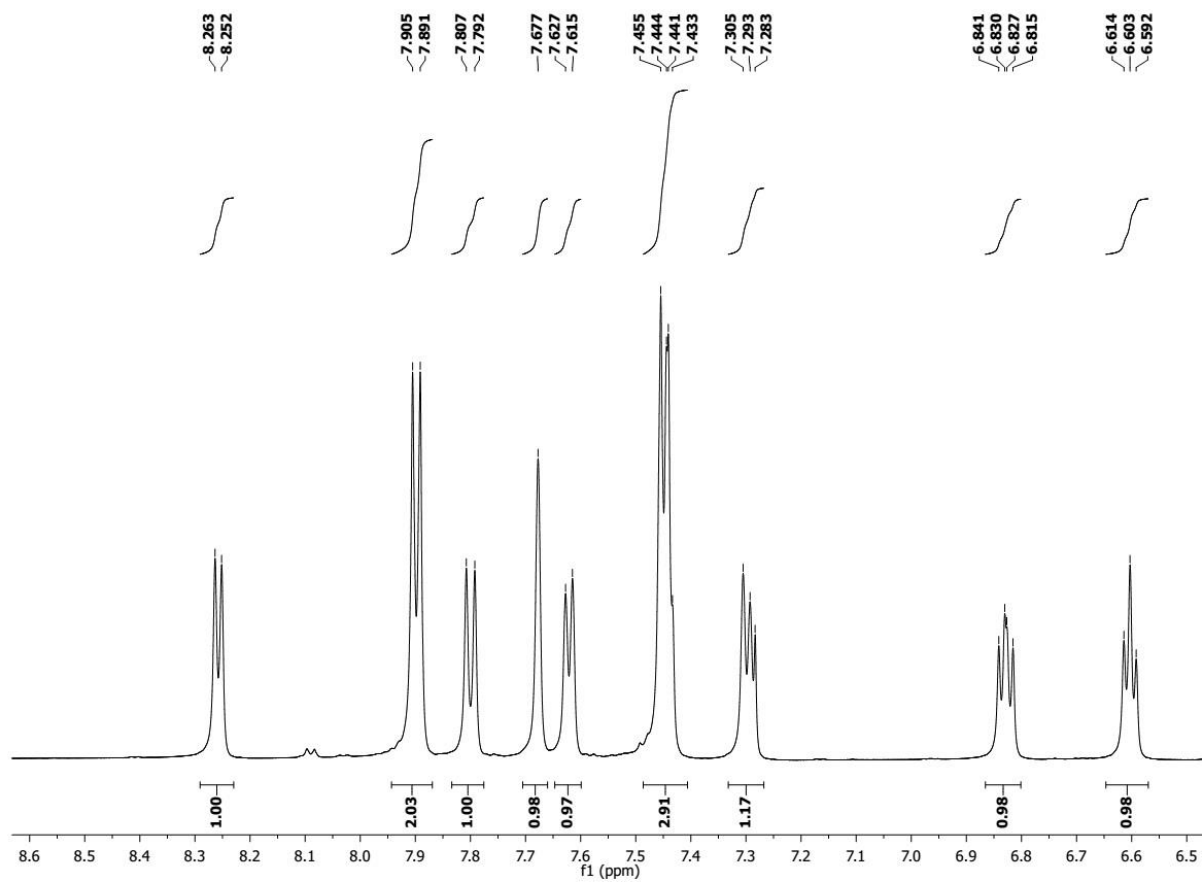
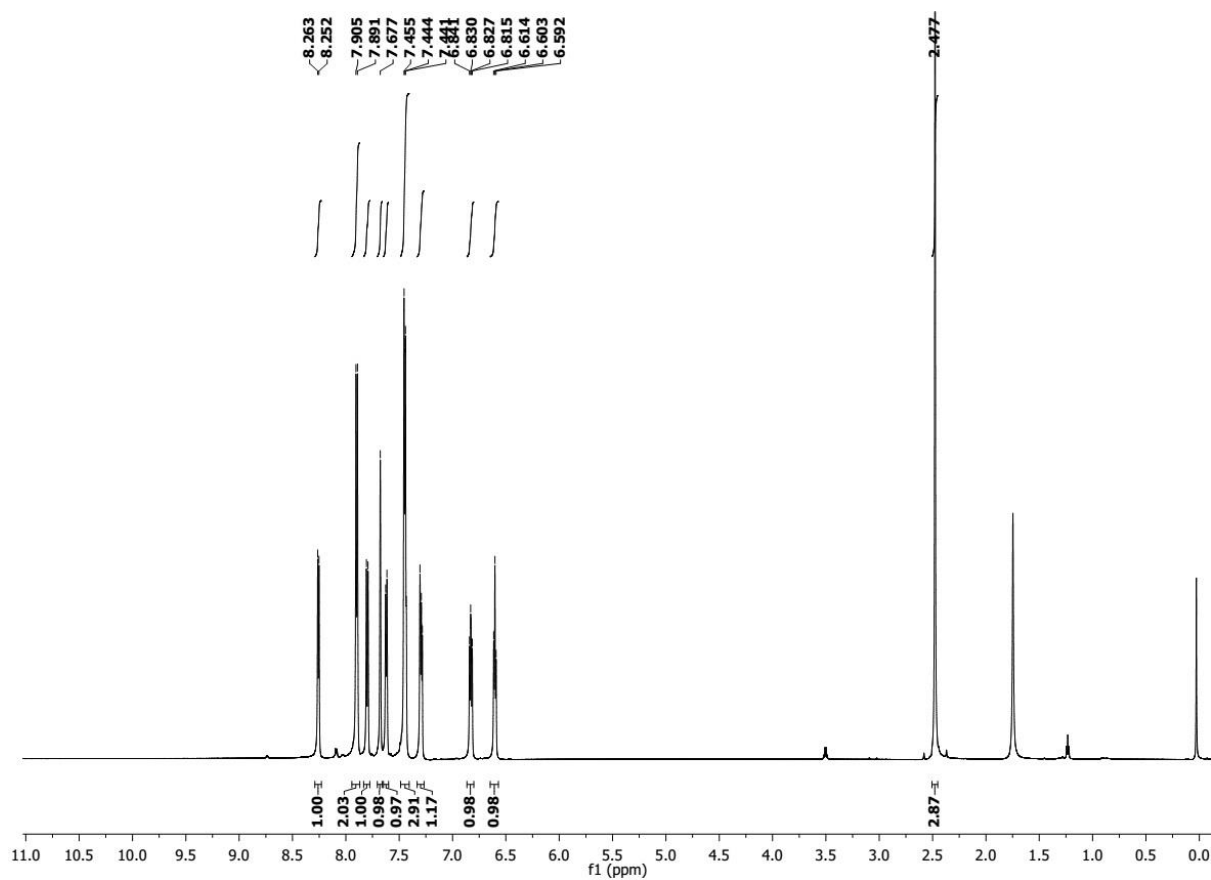
**Peak Results**  
**Channel: PDA Spectrum**

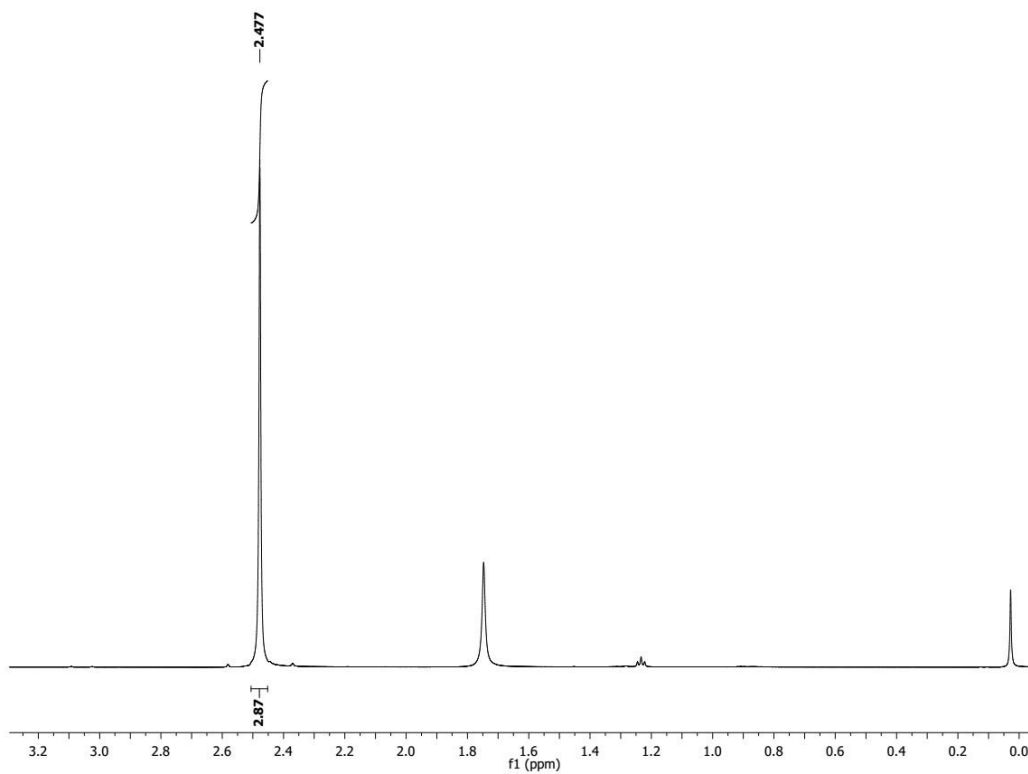
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	2.379		275987	550076	100.00	PDA Spectrum	210.0nm



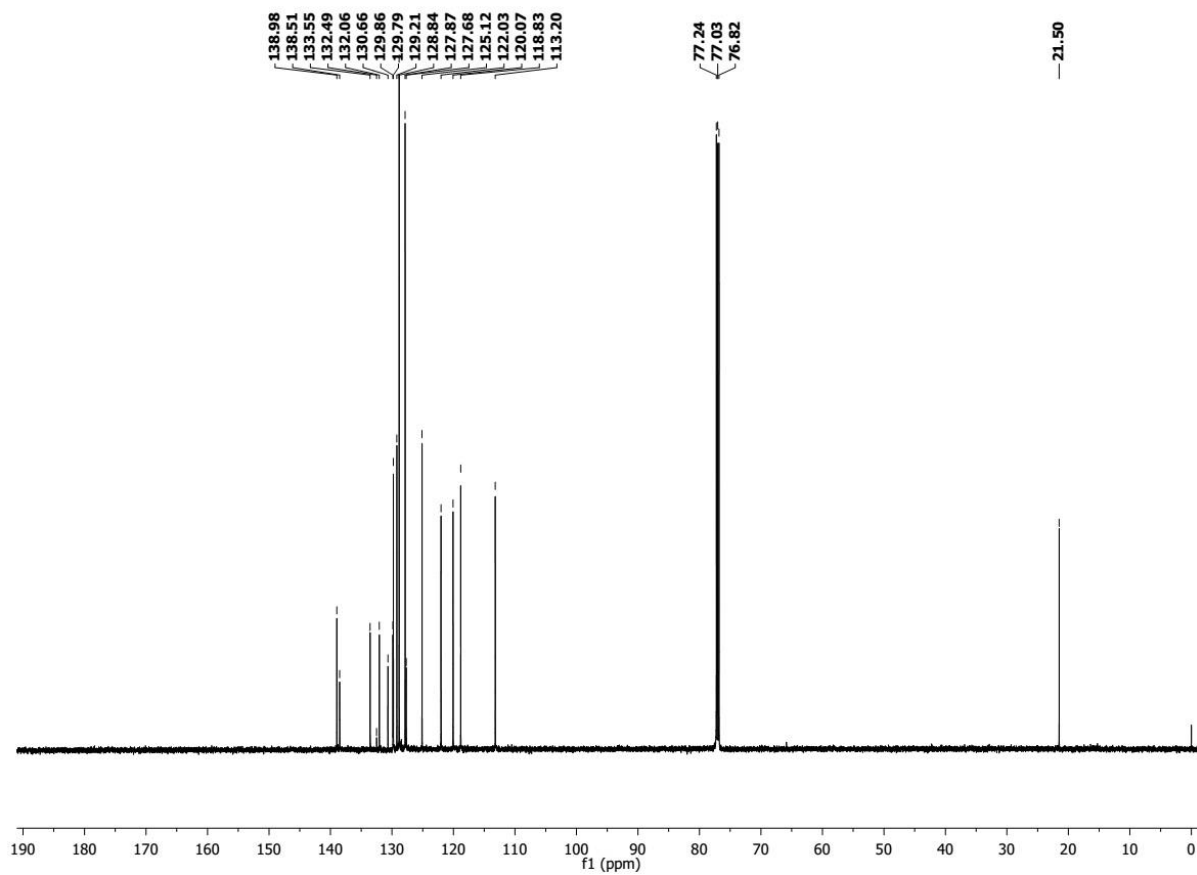
Base Peak 319.20 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (1.0:2.2;2.7:3.9;0.1:1.5) x 20.000 Th: 0.010 Retention Time 2.407

<sup>1</sup>H NMR of Compound 30

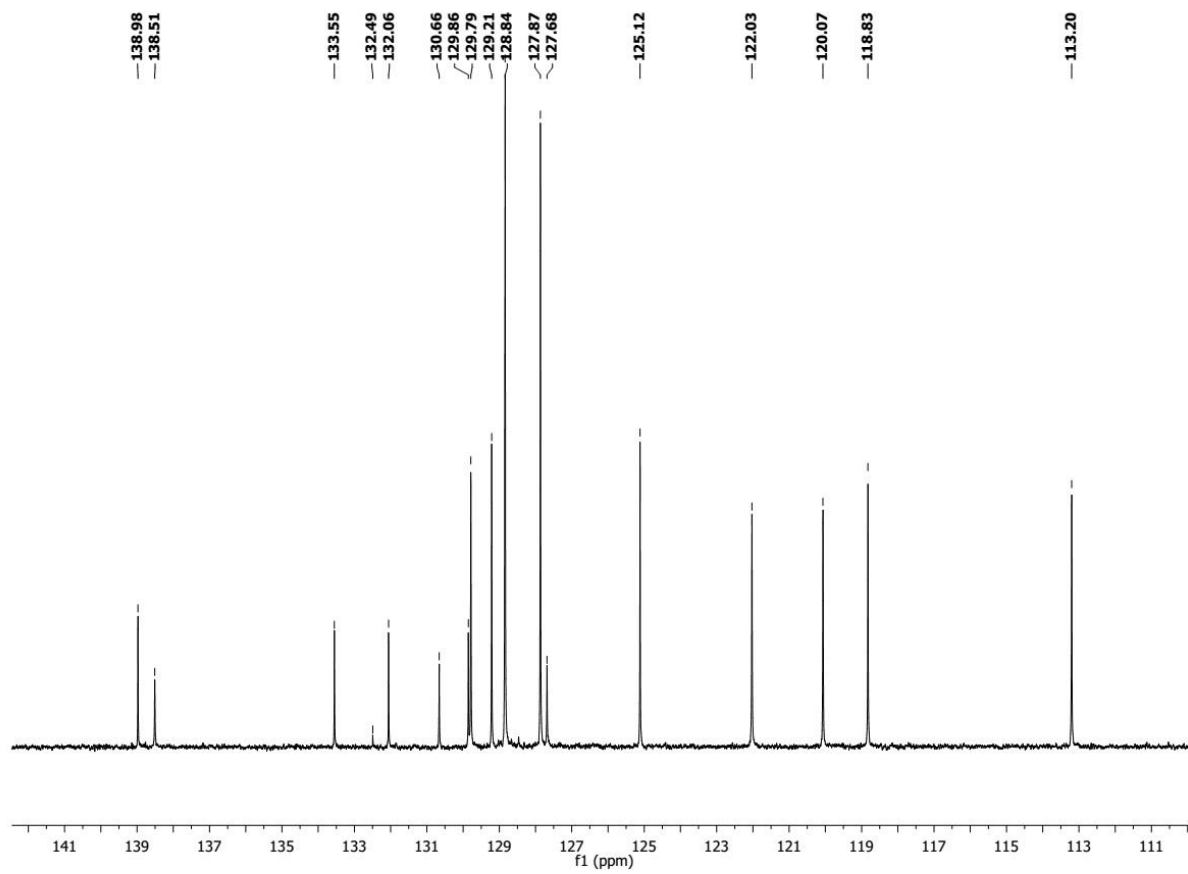


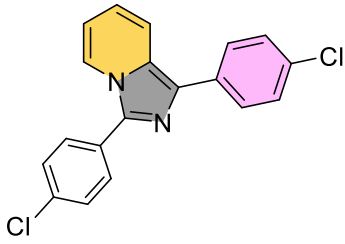


<sup>13</sup>C NMR of Compound 30



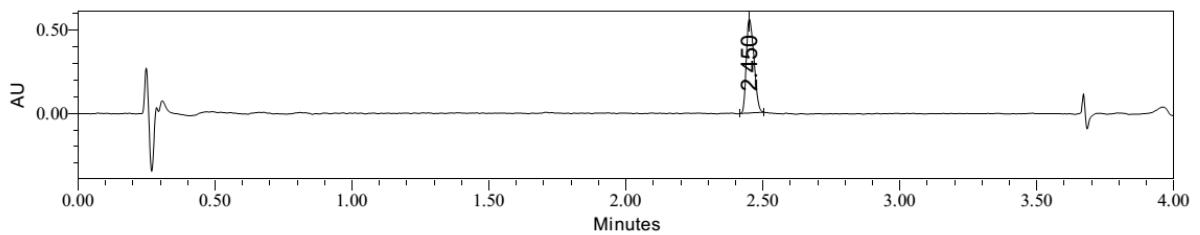
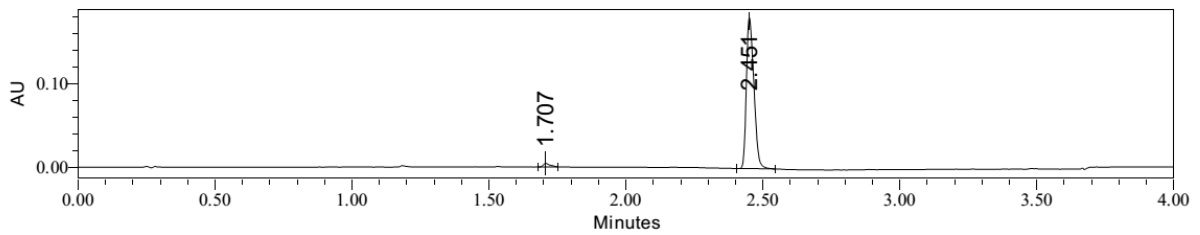






Compound (3p)

LC-MS of Compound 3p

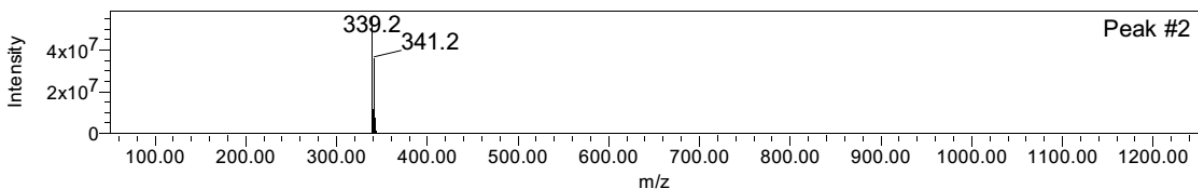


Channel Name 210.0nm; Channel PDA Spectrum

**Peak Results**  
**Channel: PDA Spectrum**

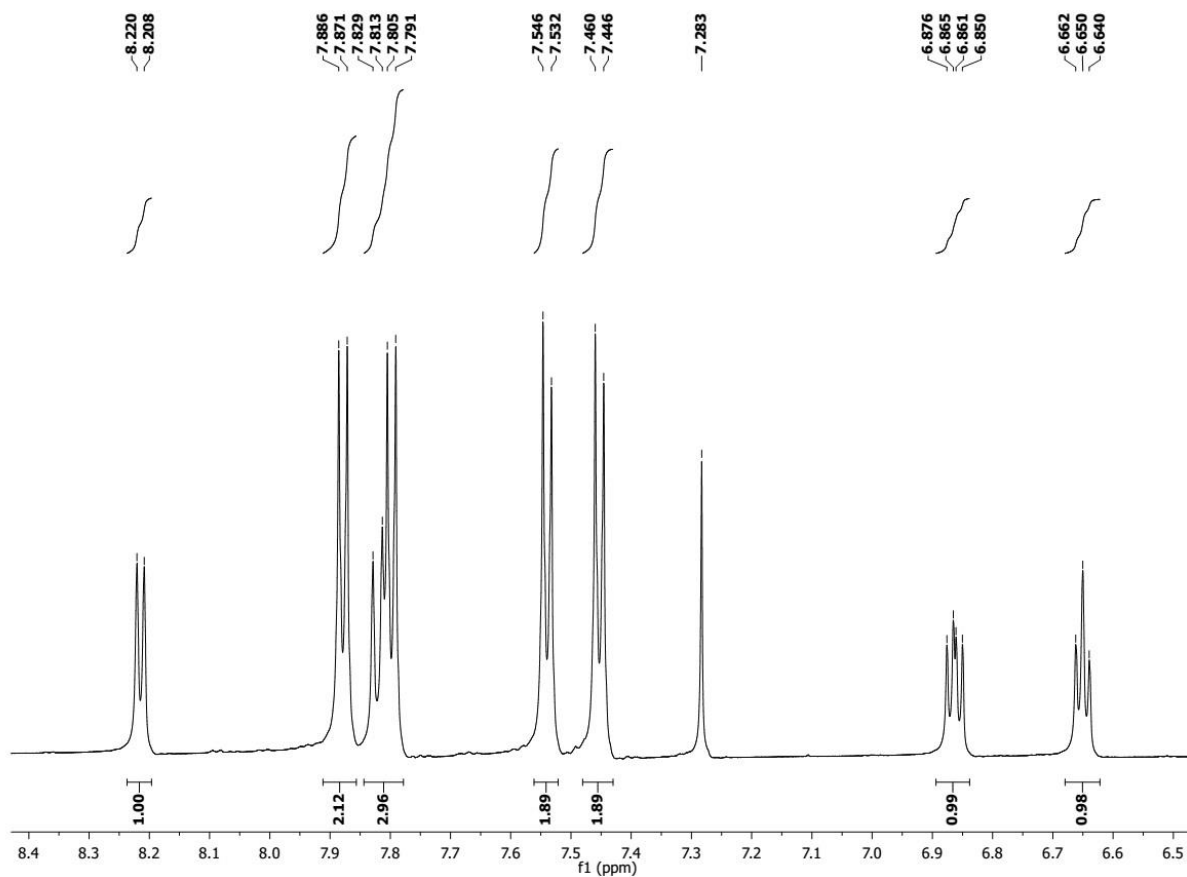
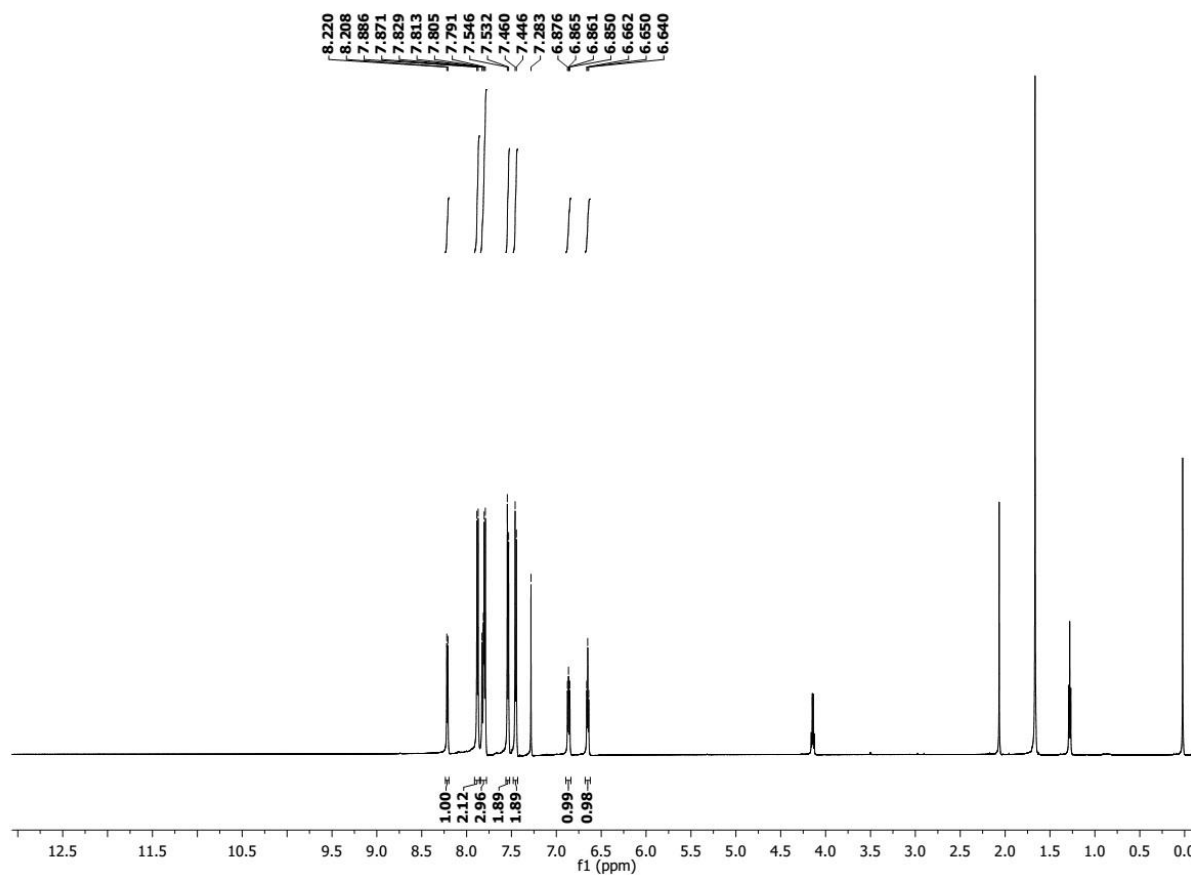
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.707		4294	7433	2.10	PDA Spectrum	254.0nm
2	2.450		564323	1093627	100.00	PDA Spectrum	210.0nm
3	2.451		181645	345792	97.90	PDA Spectrum	254.0nm

Match Plot

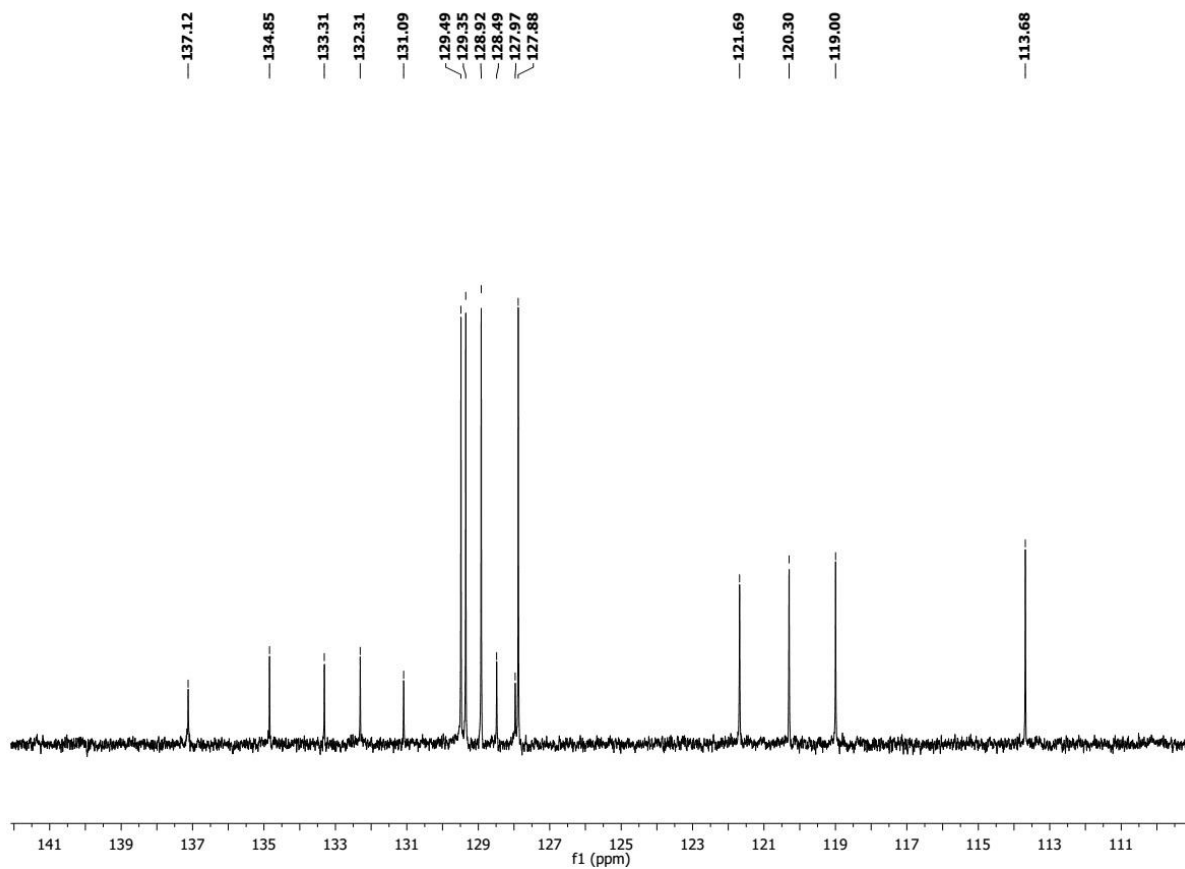
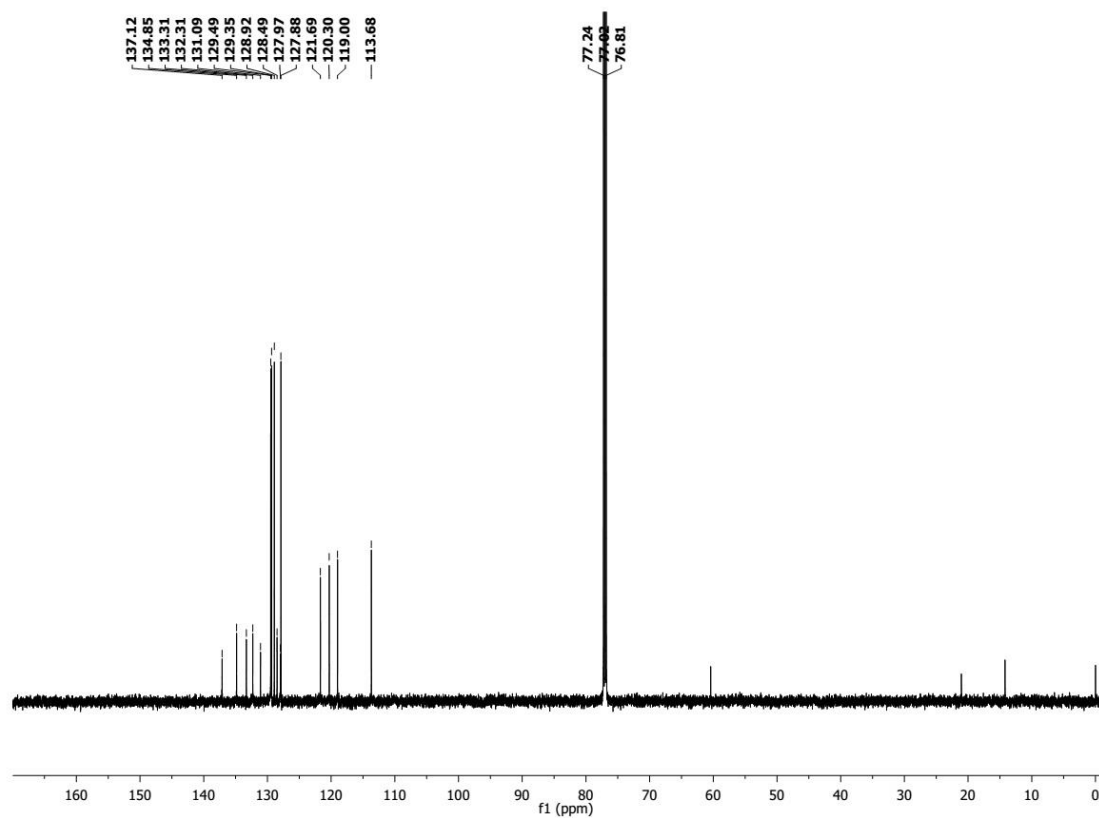


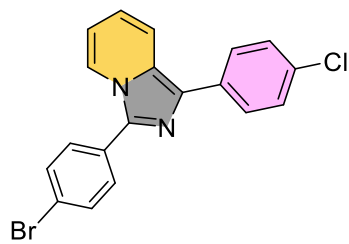
Base Peak 339.18 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (1.8:2.4;0.0:1.6;2.7:3.9) x 20.000 Retention Time 2.479

<sup>1</sup>H NMR of Compound 3p



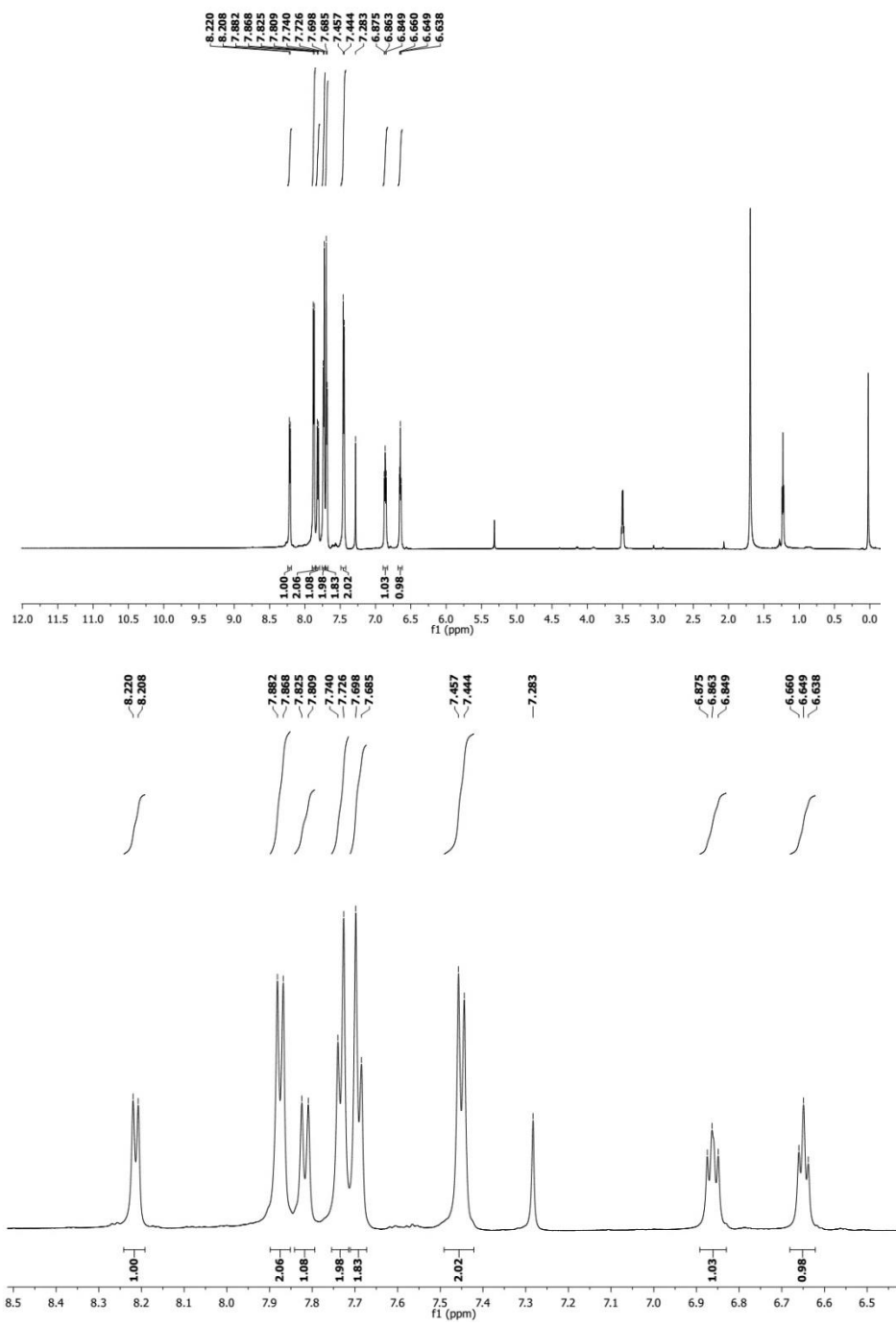
# $^{13}\text{C}$ NMR of Compound 3p



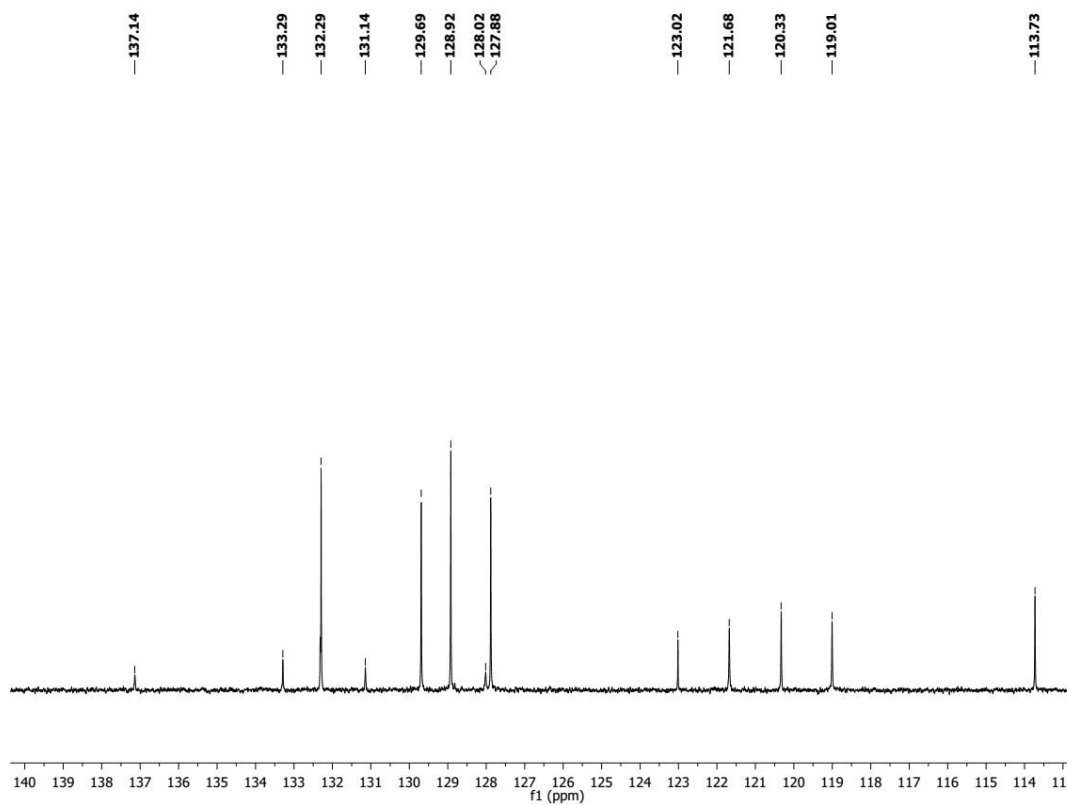
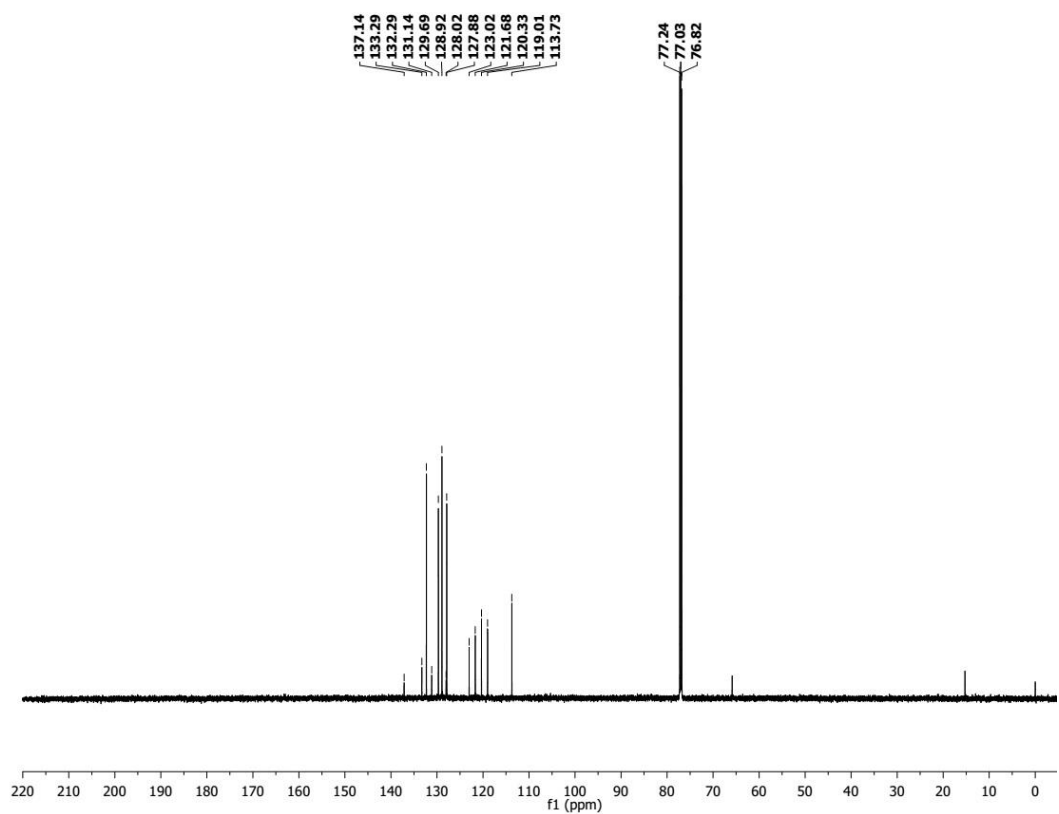


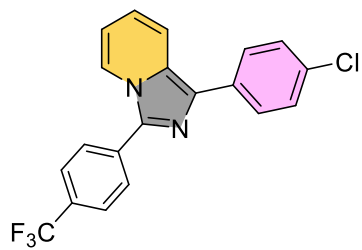
Compound (3q)

$^1\text{H}$  NMR of Compound 3q



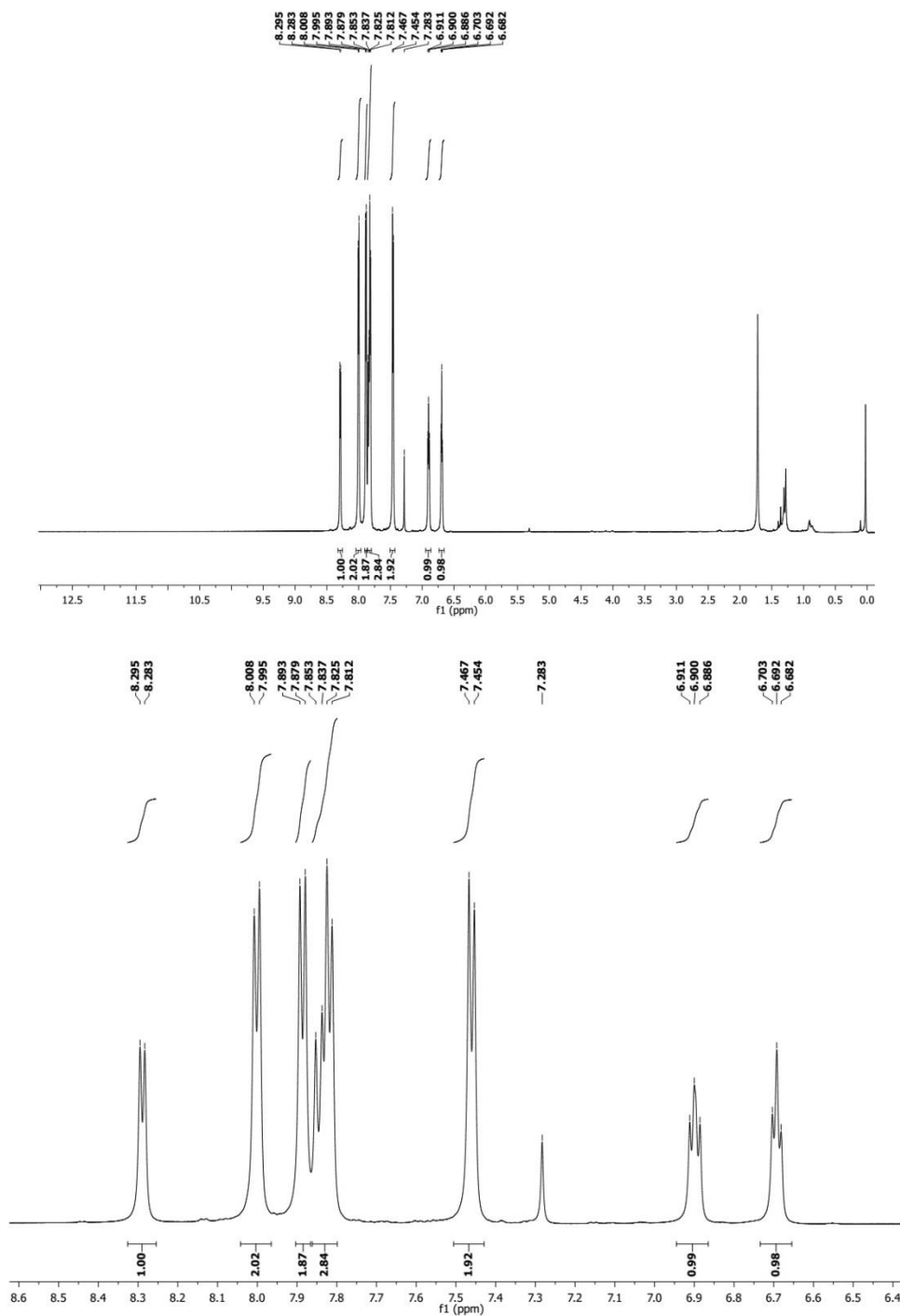
<sup>13</sup>C NMR of Compound 3q



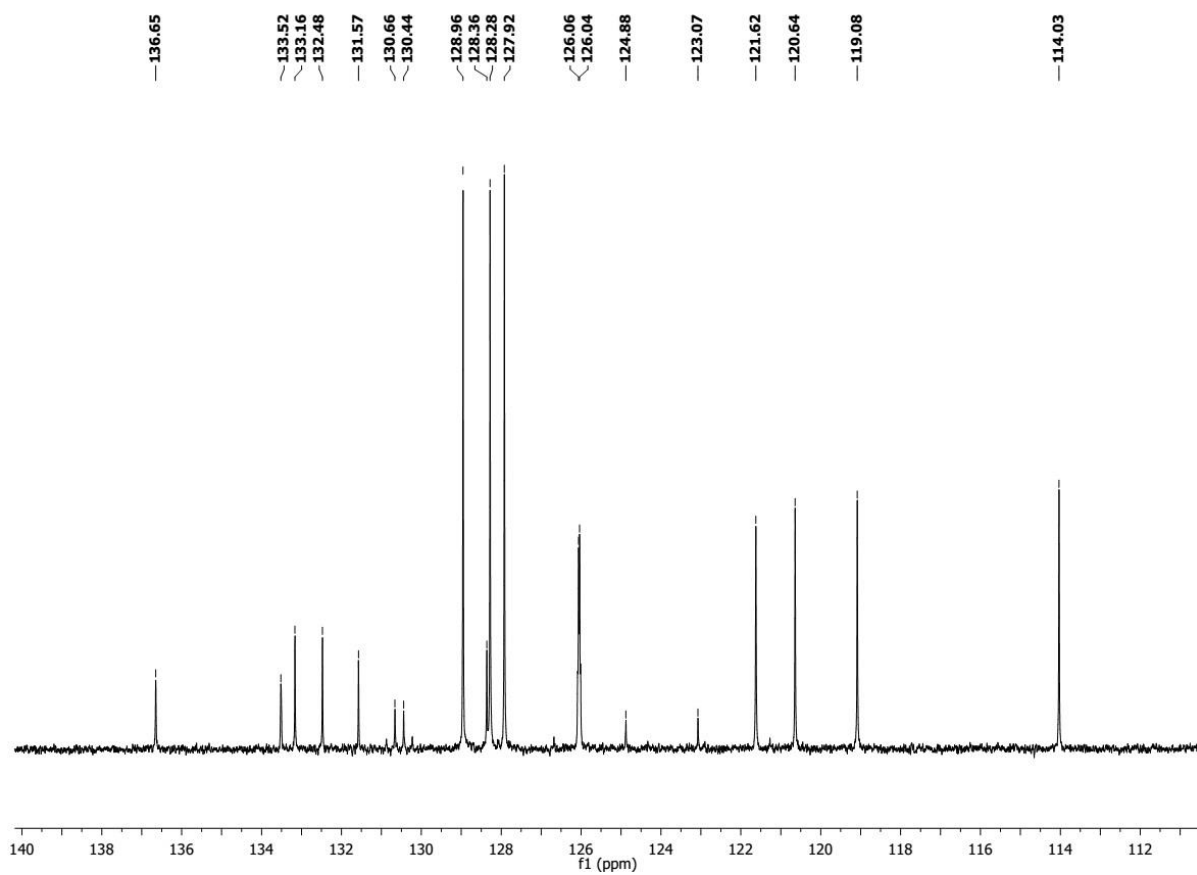
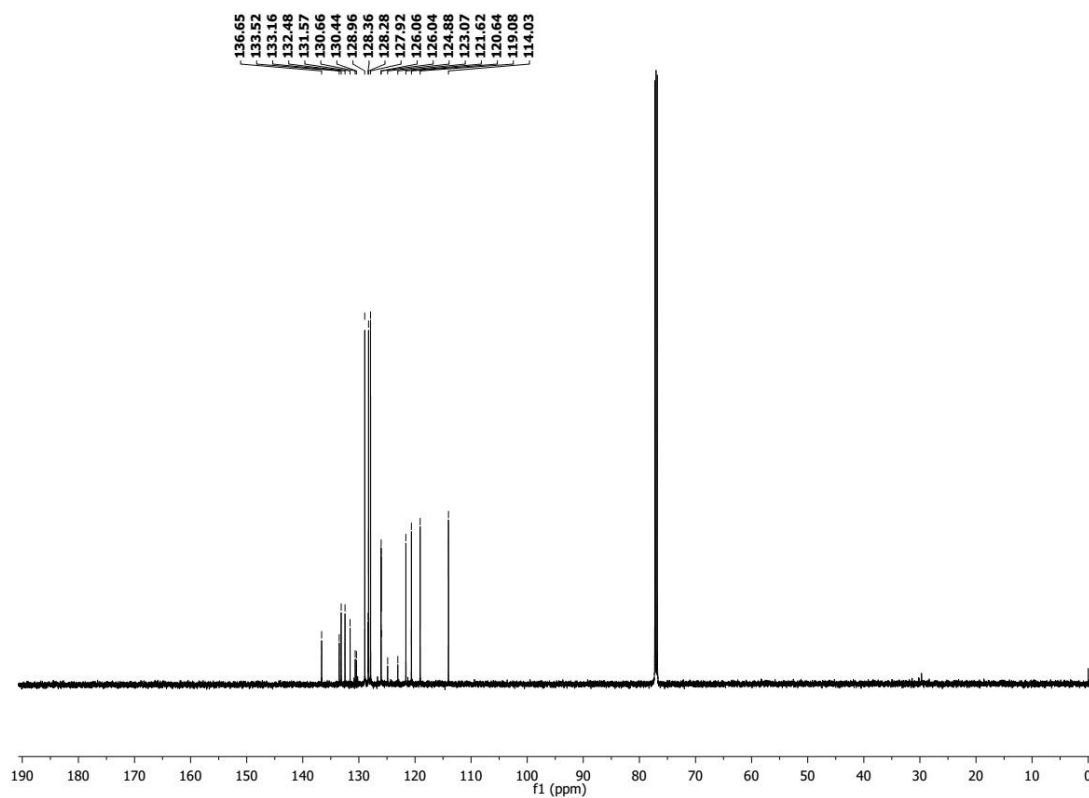


Compound (3r)

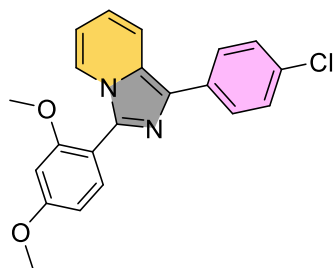
$^1\text{H}$  NMR of Compound 3r



<sup>13</sup>C NMR of Compound 3r

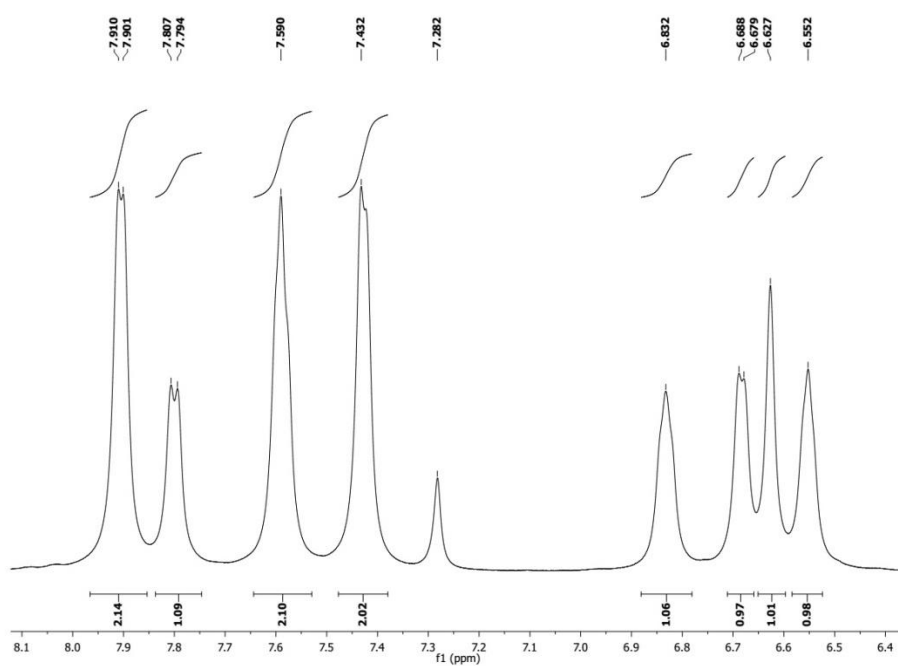
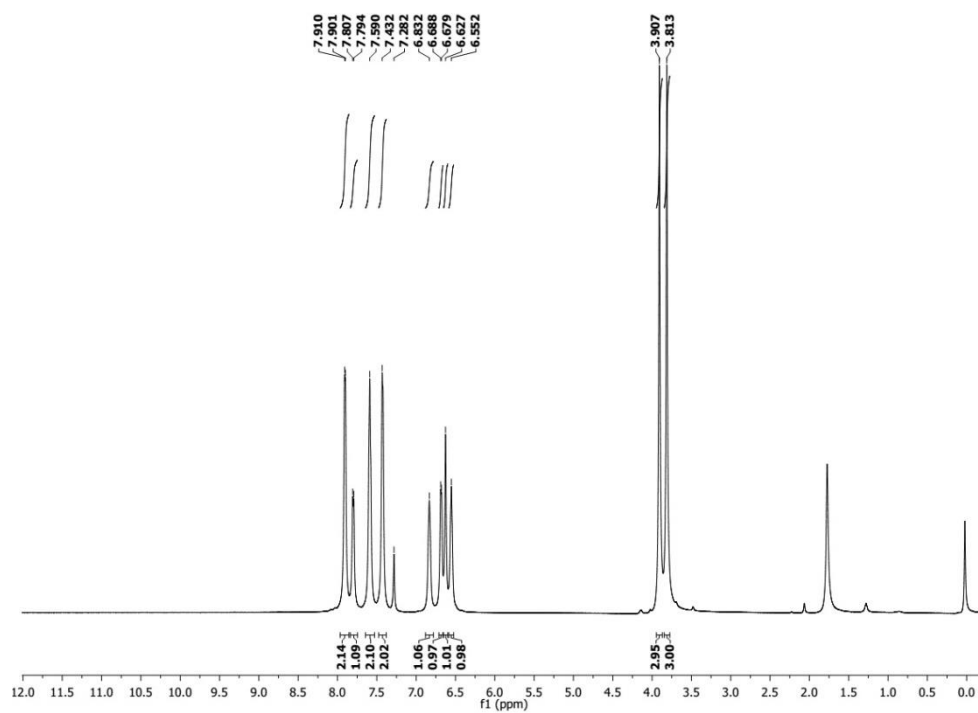


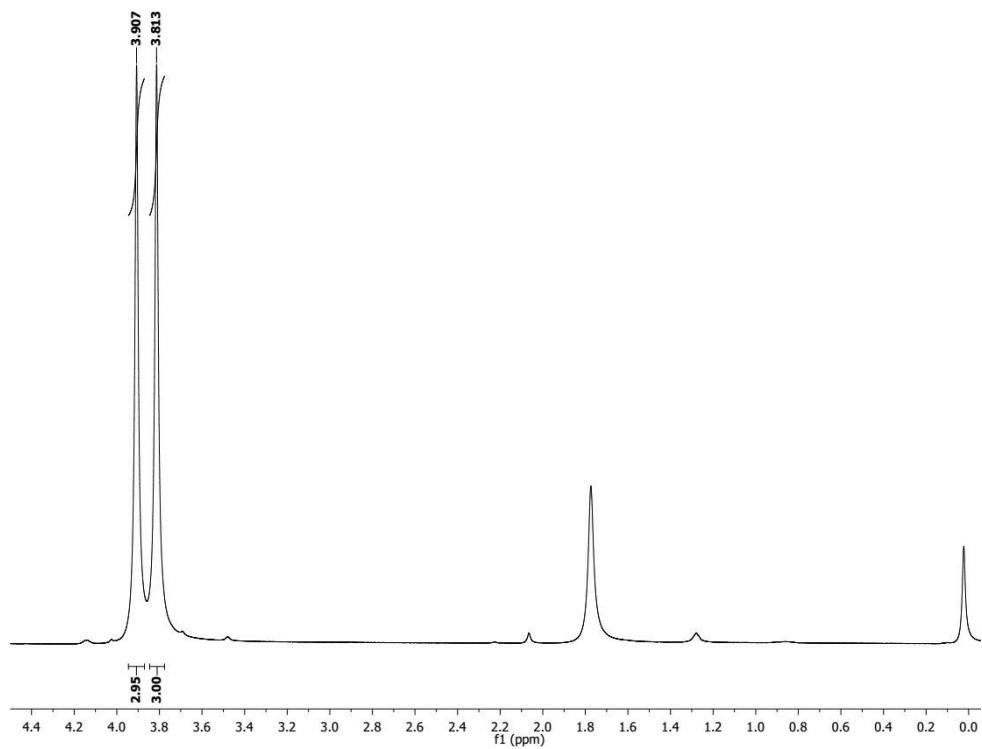




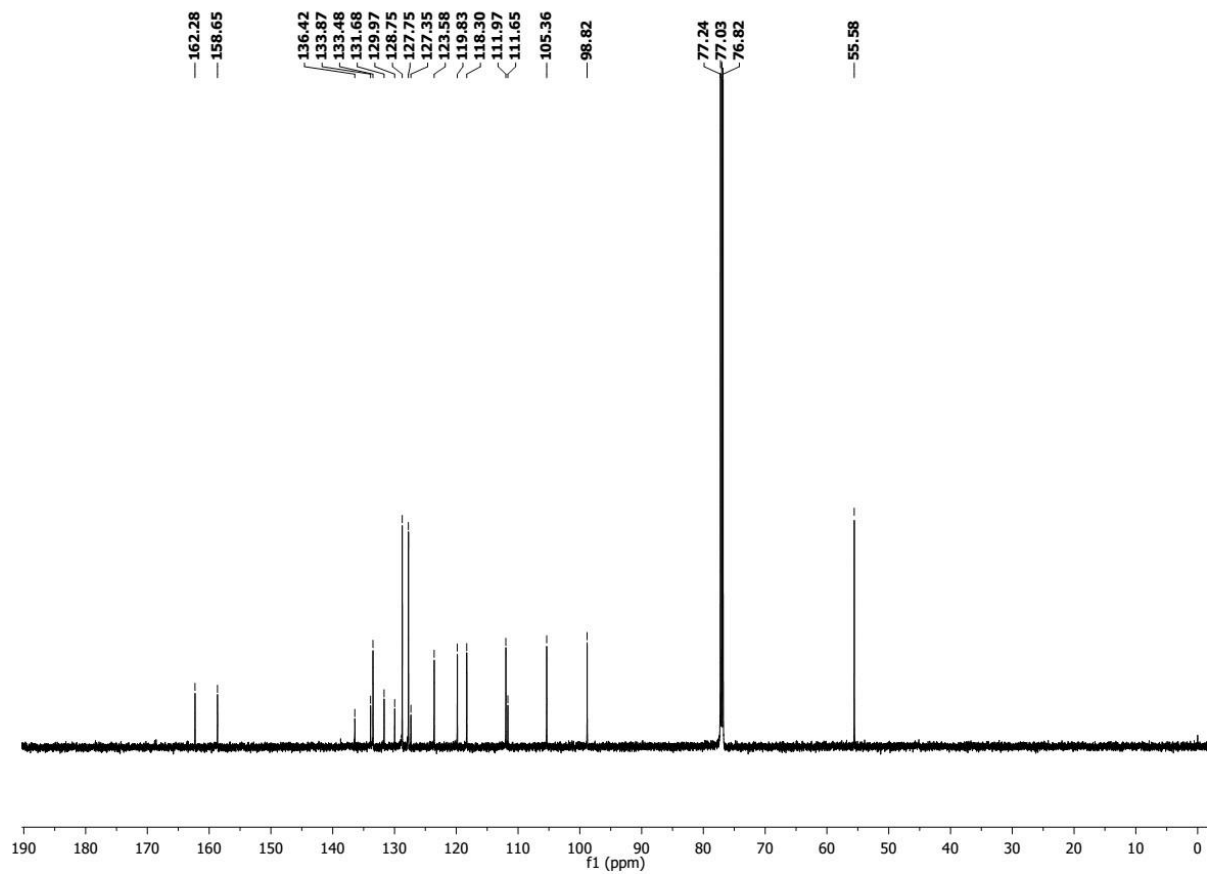
Compound (3s)

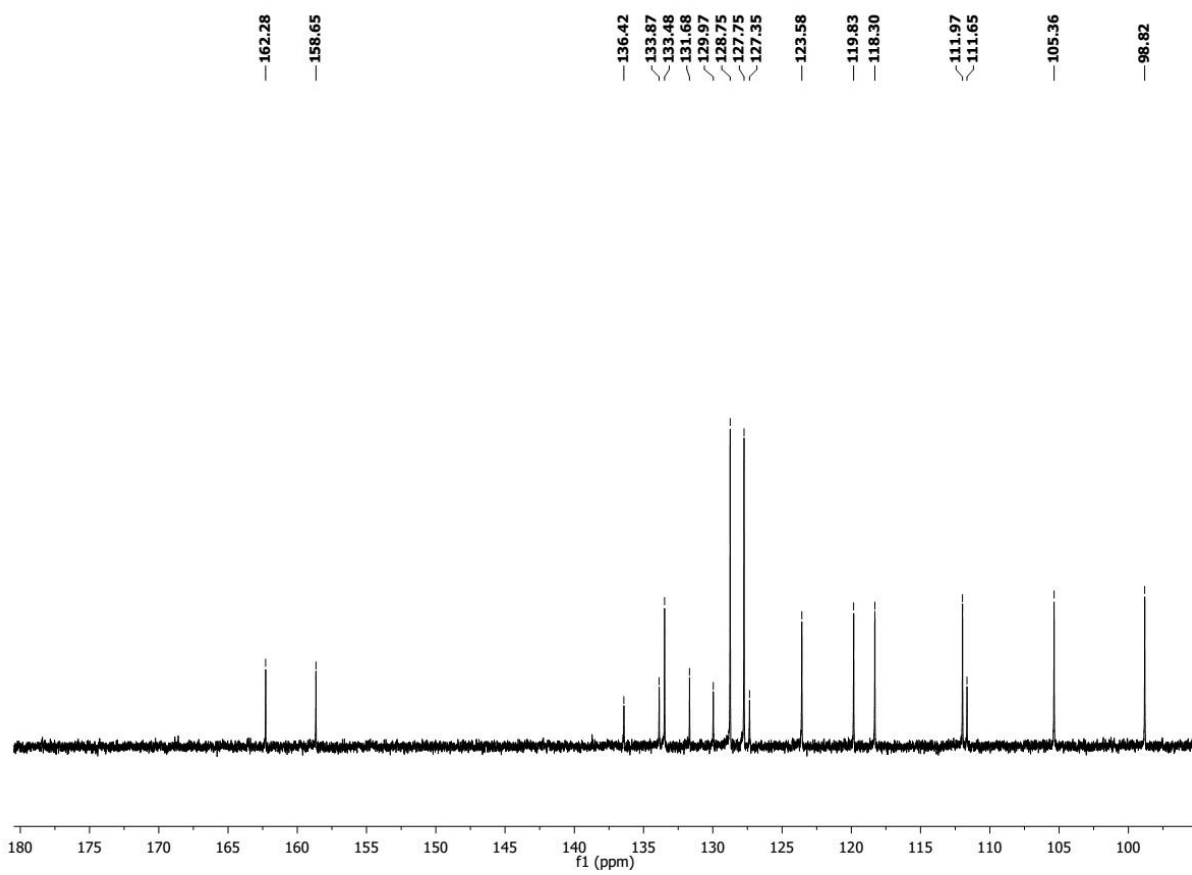
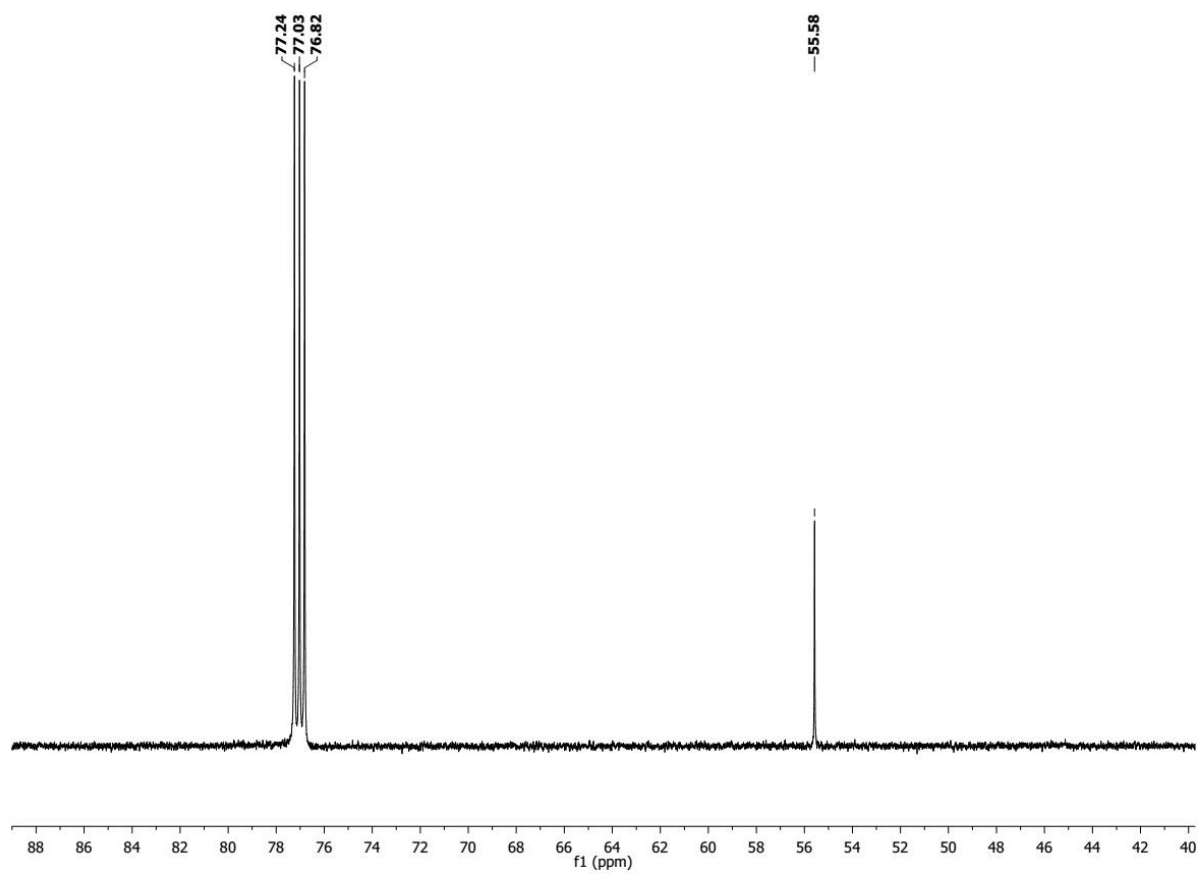
$^1\text{H}$  NMR of Compound 3s

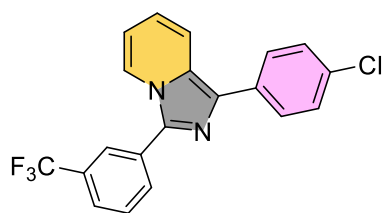




13C NMR of Compound 3s

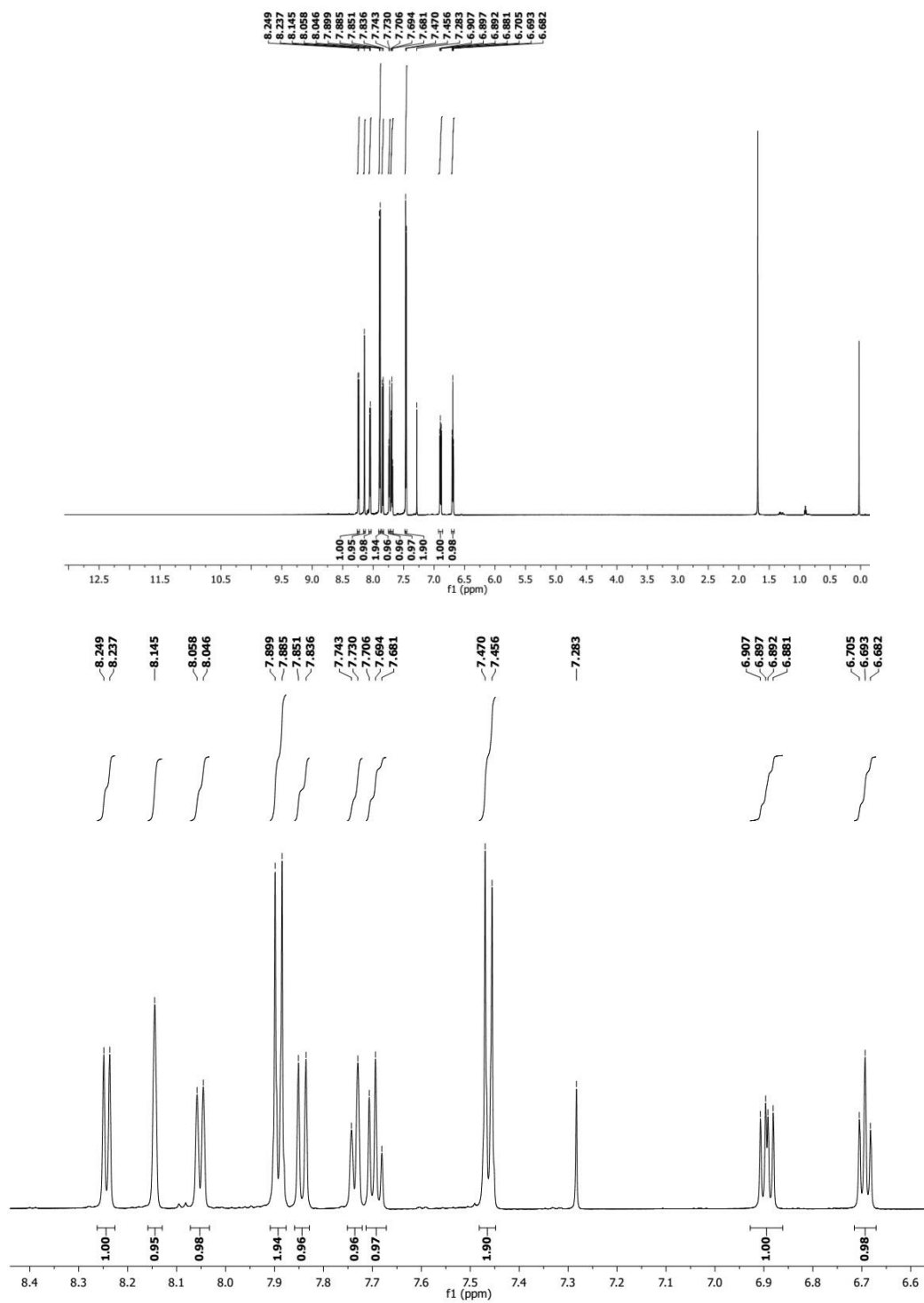




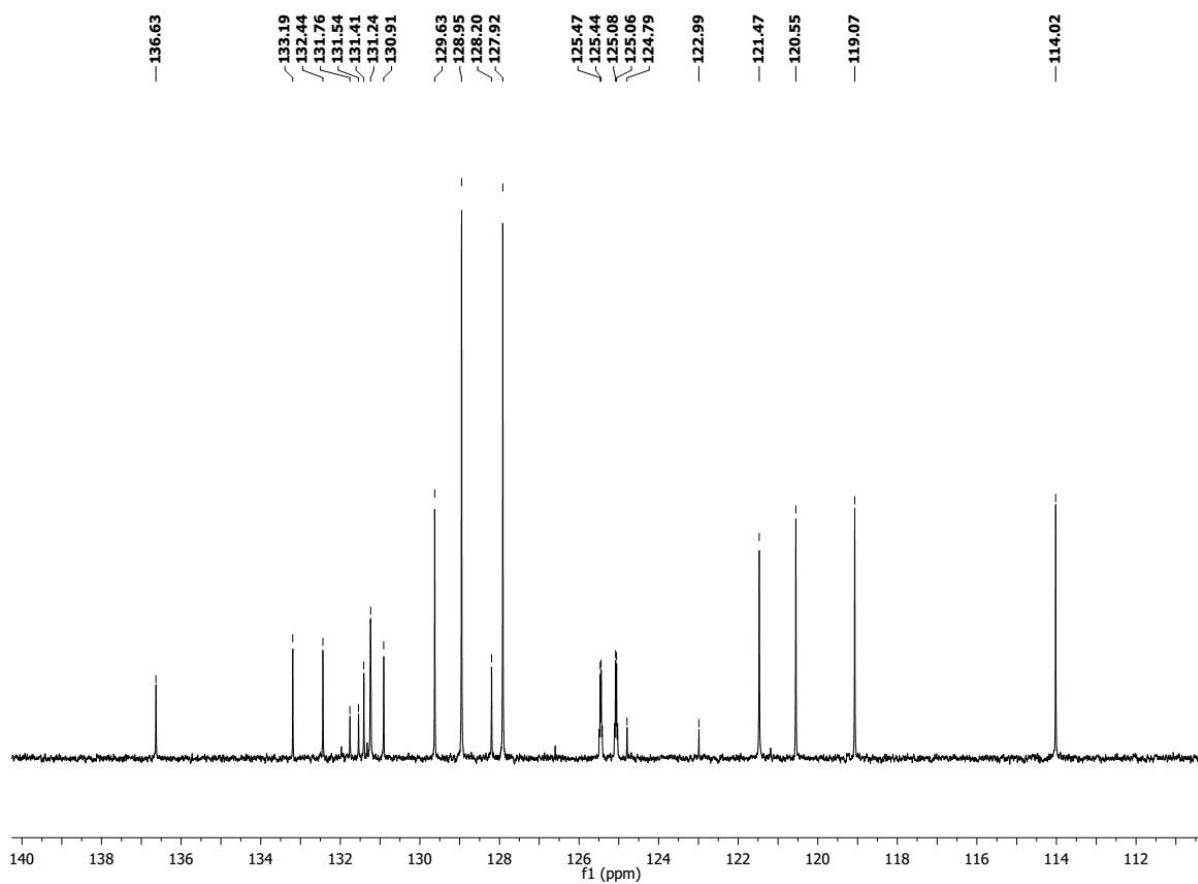
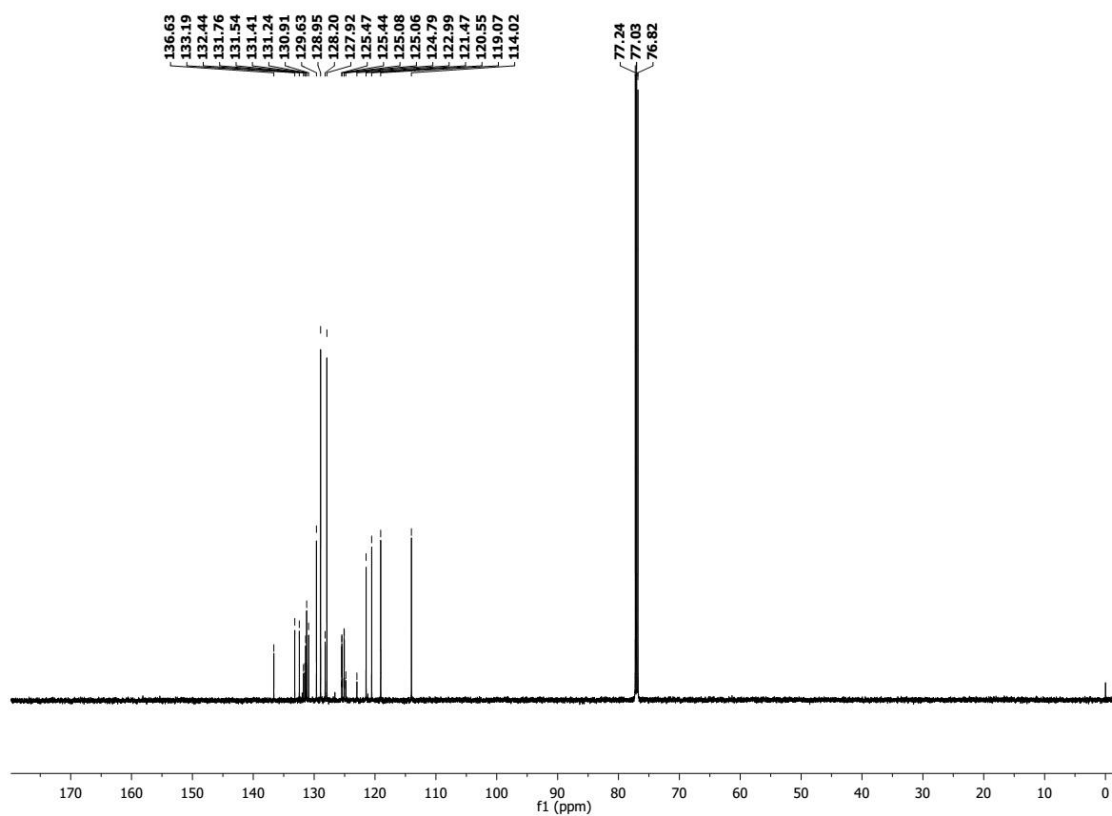


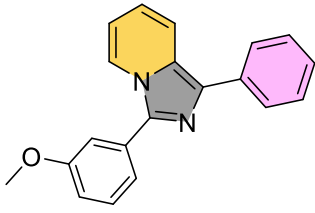
Compound (3t)

<sup>1</sup>H NMR of Compound 3t



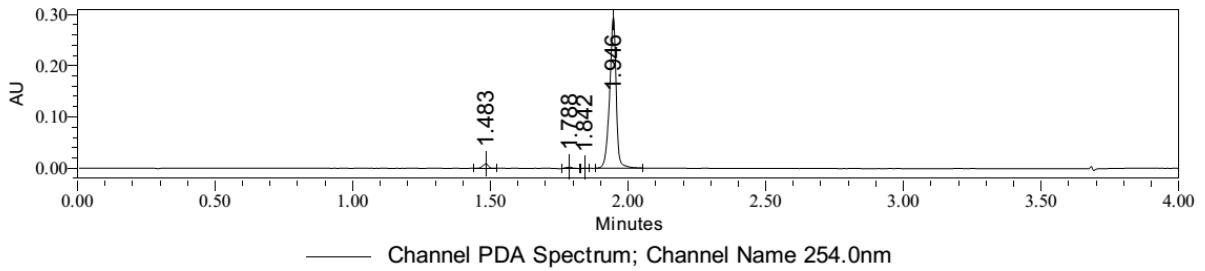
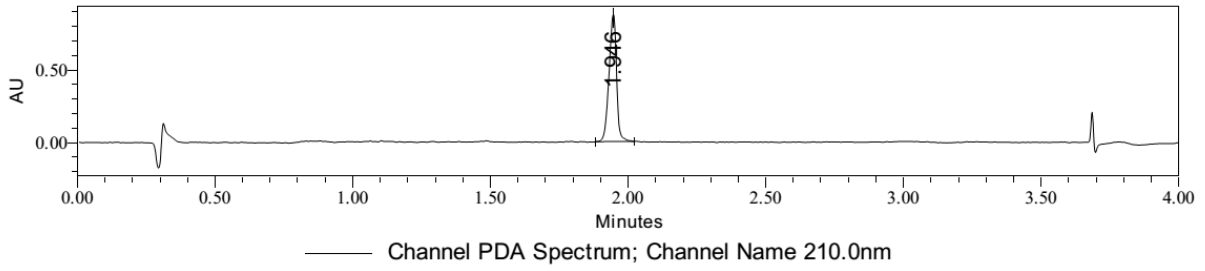
<sup>13</sup>C NMR of Compound 3t





Compound (3u)

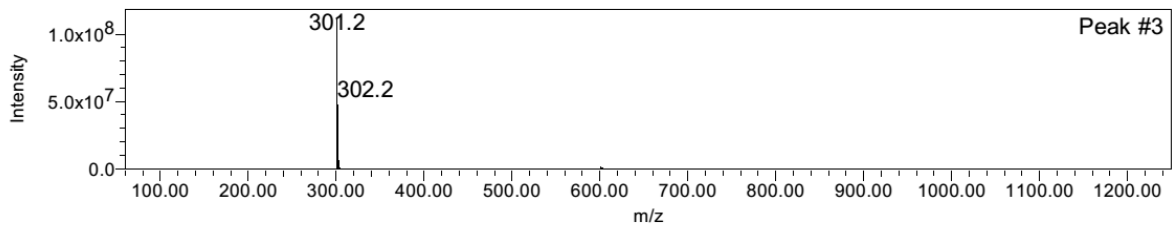
LC-MS of Compound 3u



**Peak Results**  
Channel: PDA Spectrum

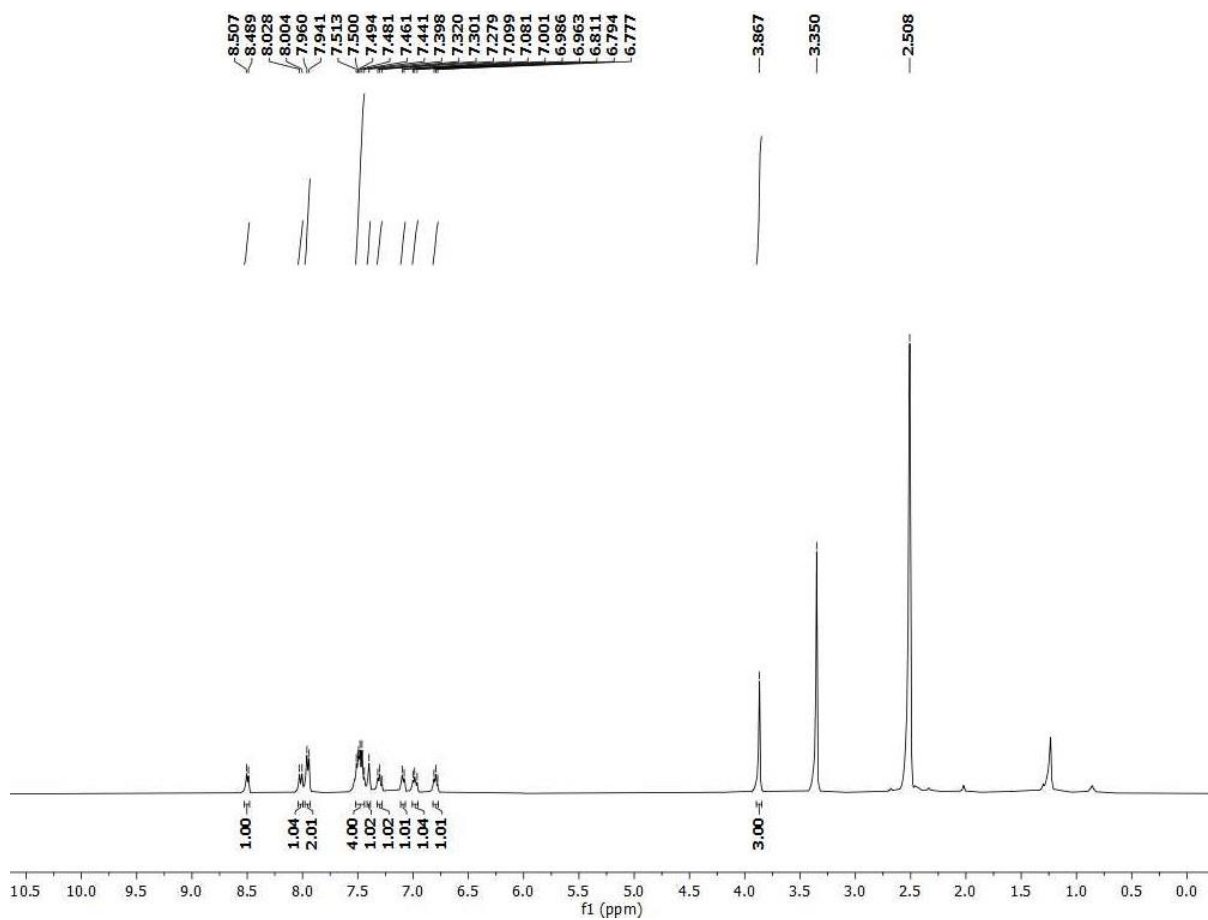
	RT	Base Peak (m/z)	Height	Area	% Area	Channel	Channel Name
1	1.483		9020	11930	2.37	PDA Spectrum	254.0nm
2	1.788		2950	5060	1.00	PDA Spectrum	254.0nm
3	1.842		799	809	0.16	PDA Spectrum	254.0nm
4	1.946		294870	486016	96.47	PDA Spectrum	254.0nm
5	1.946		875637	1666342	100.00	PDA Spectrum	210.0nm

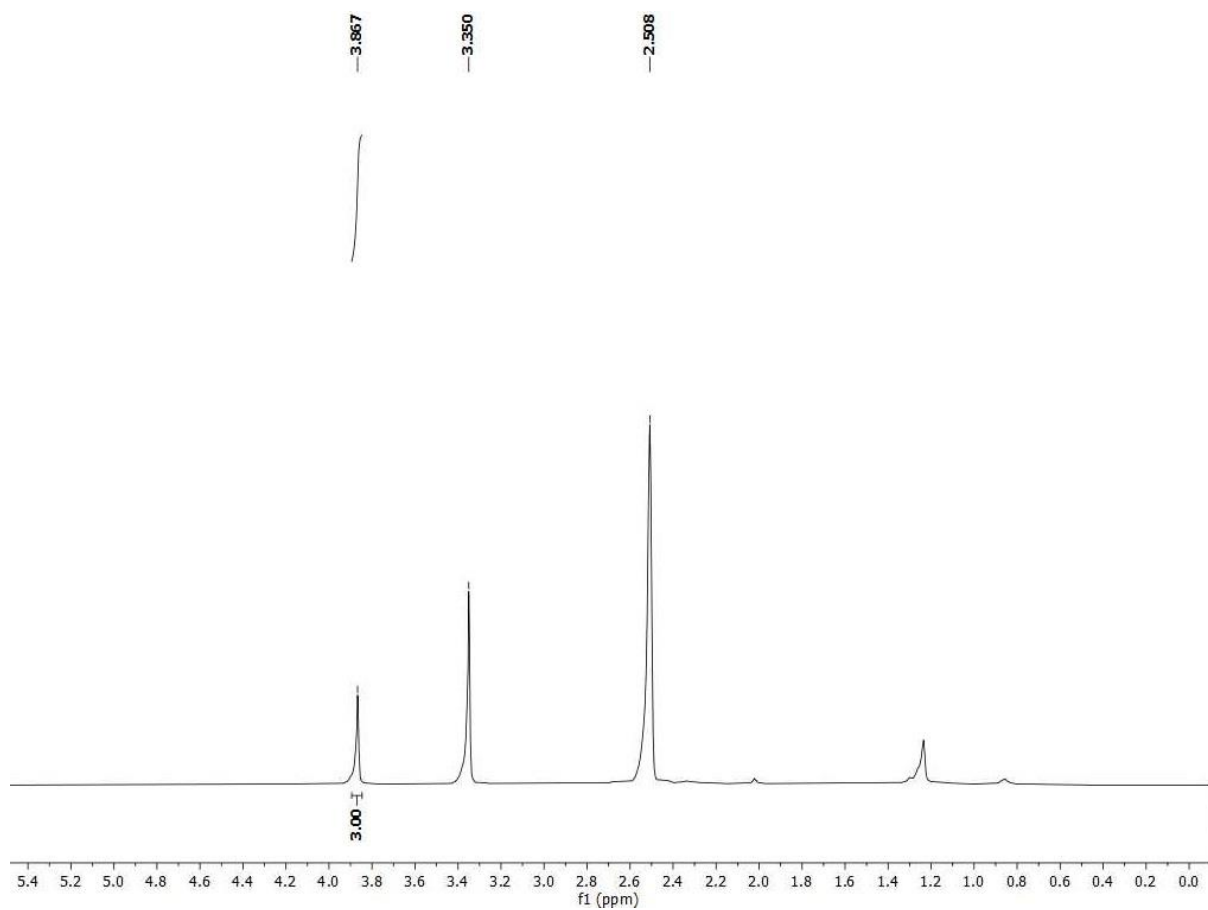
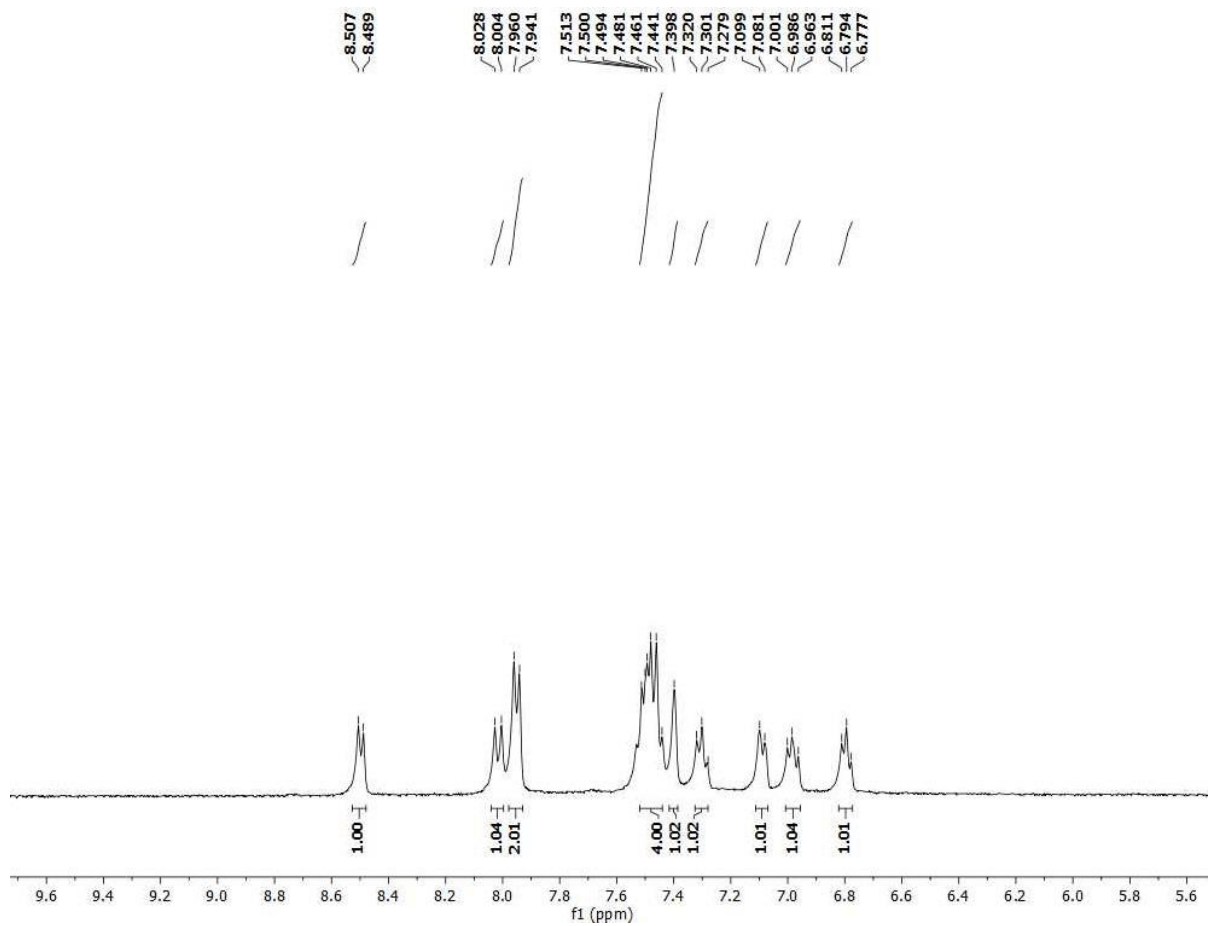
**Match Plot**



Base Peak 301.18 Channel Description 1: QDa Positive(+) Scan (60.00-1250.00)Da, Centroid, CV=10 - AVG (2.7:3.9) 20.000 Th: 0.010 Retention Time 1.966

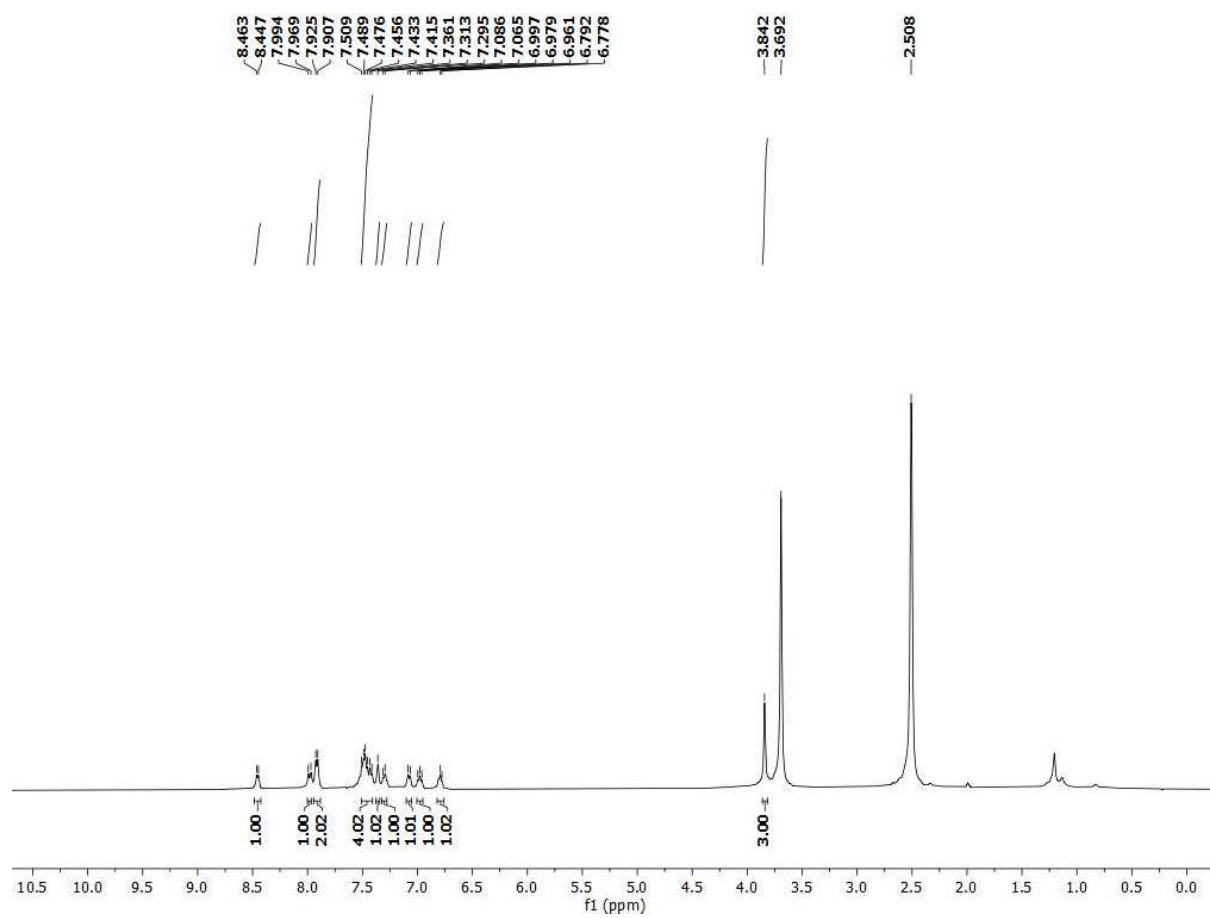
<sup>1</sup>H NMR of Compound 3u

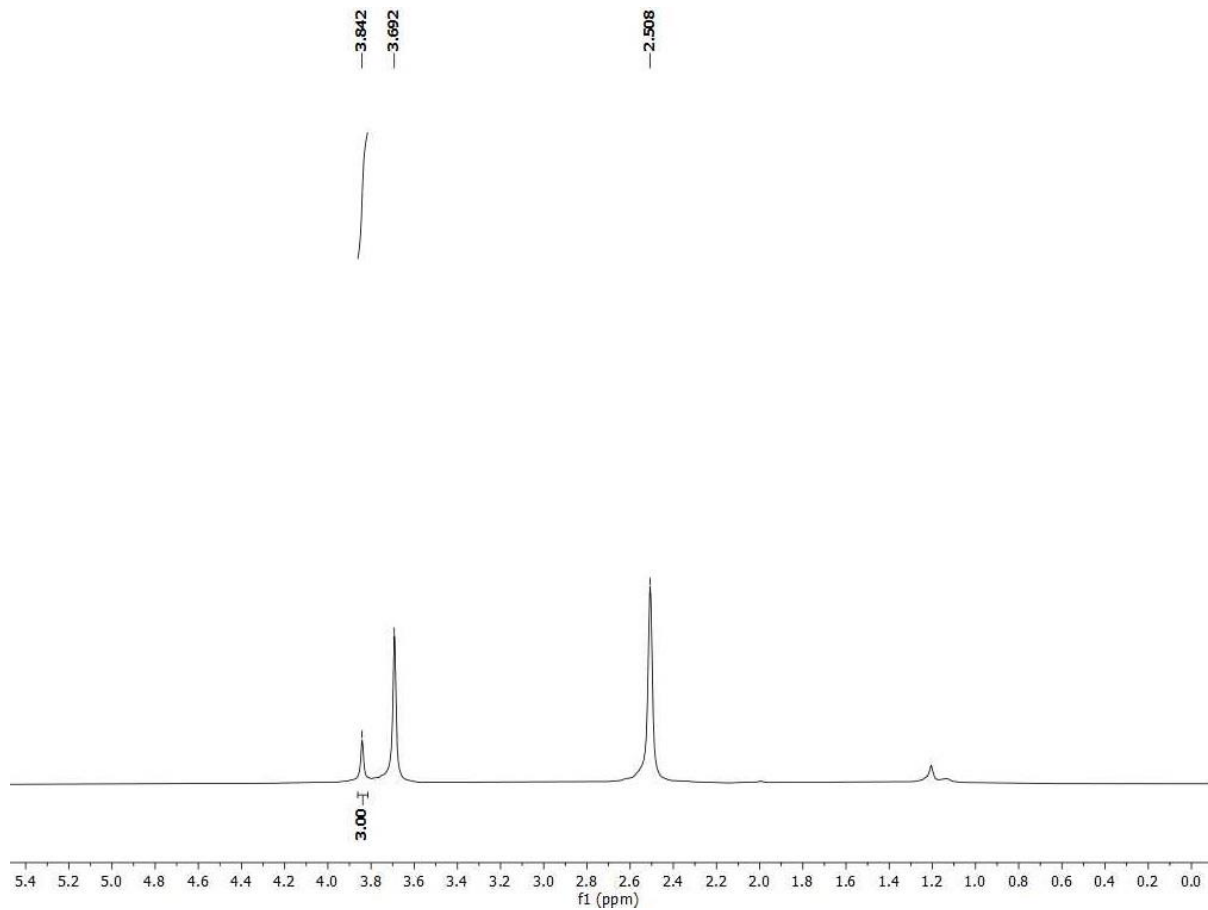
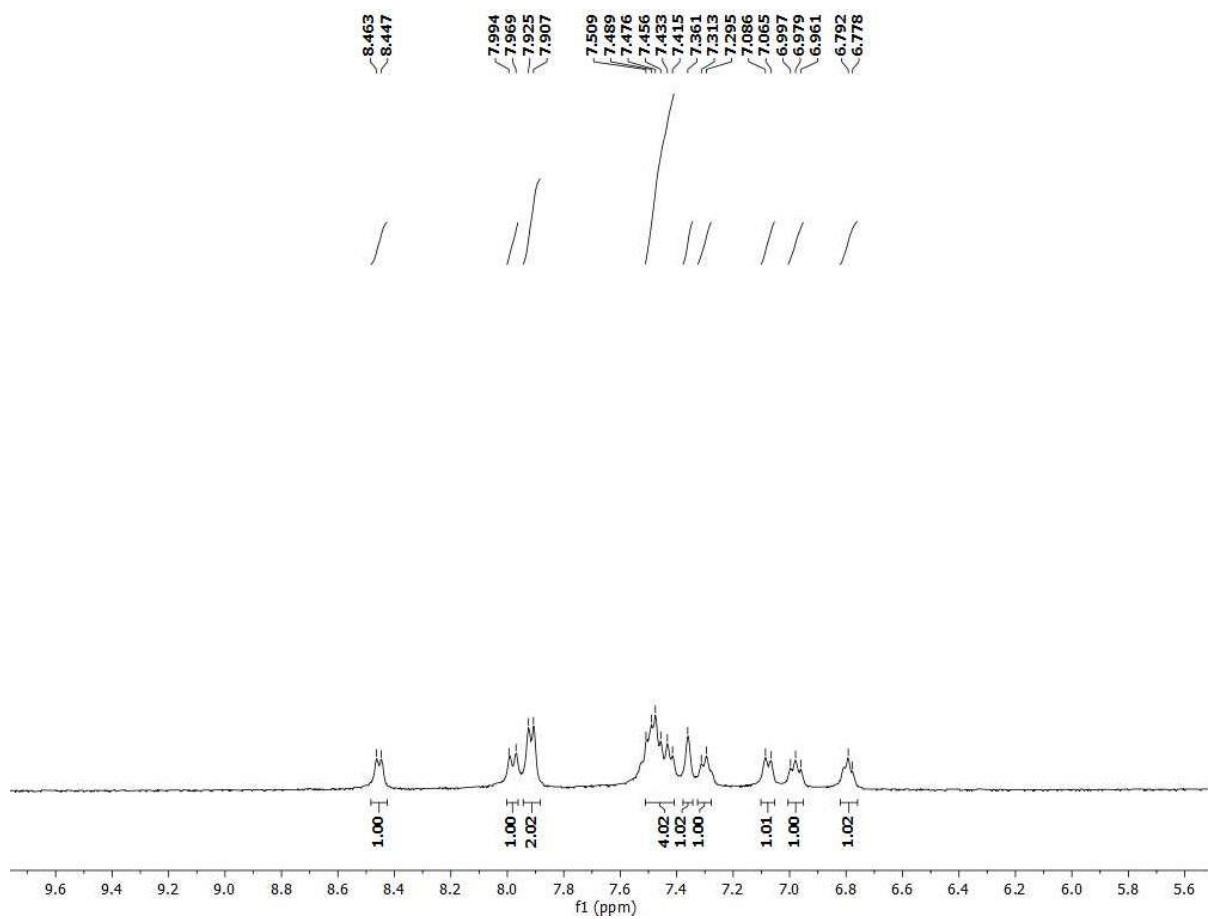




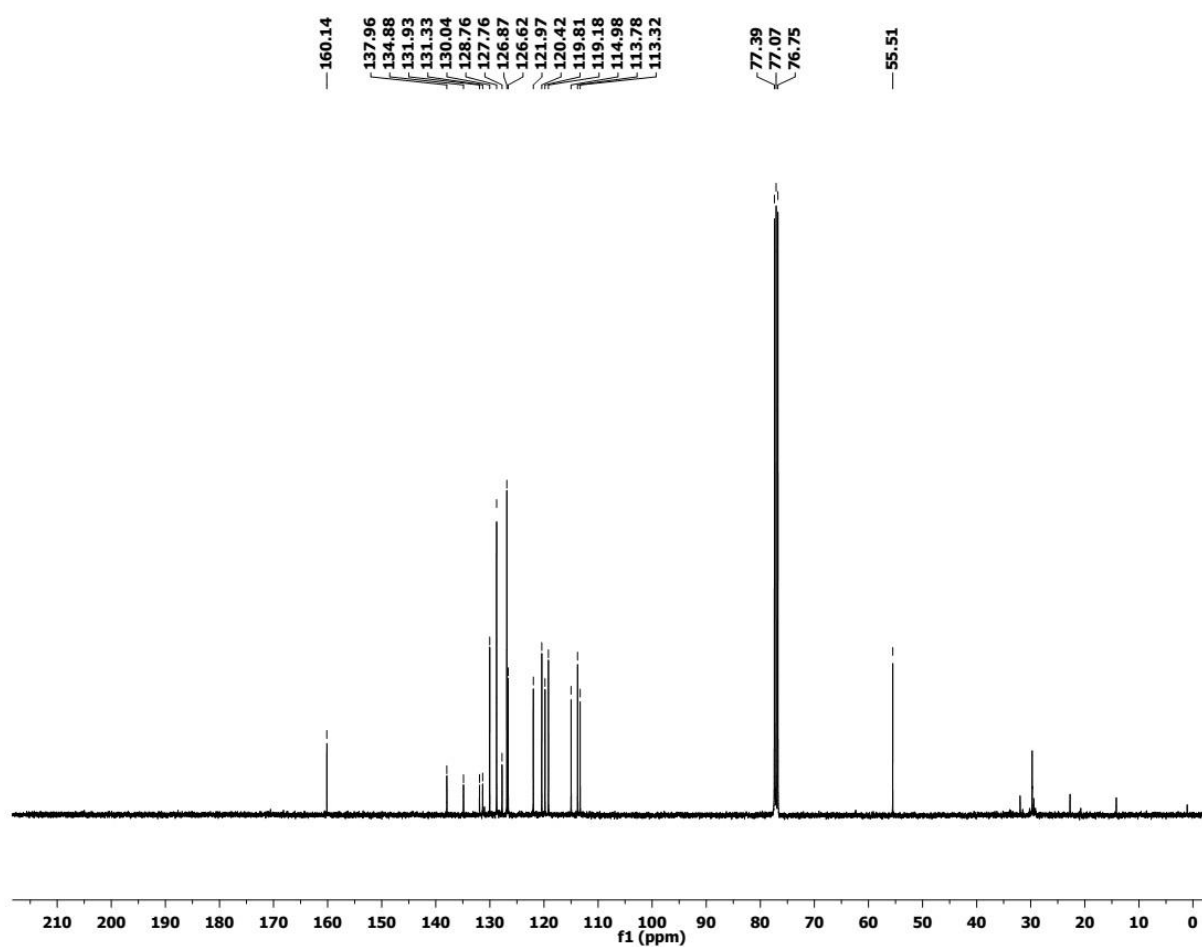


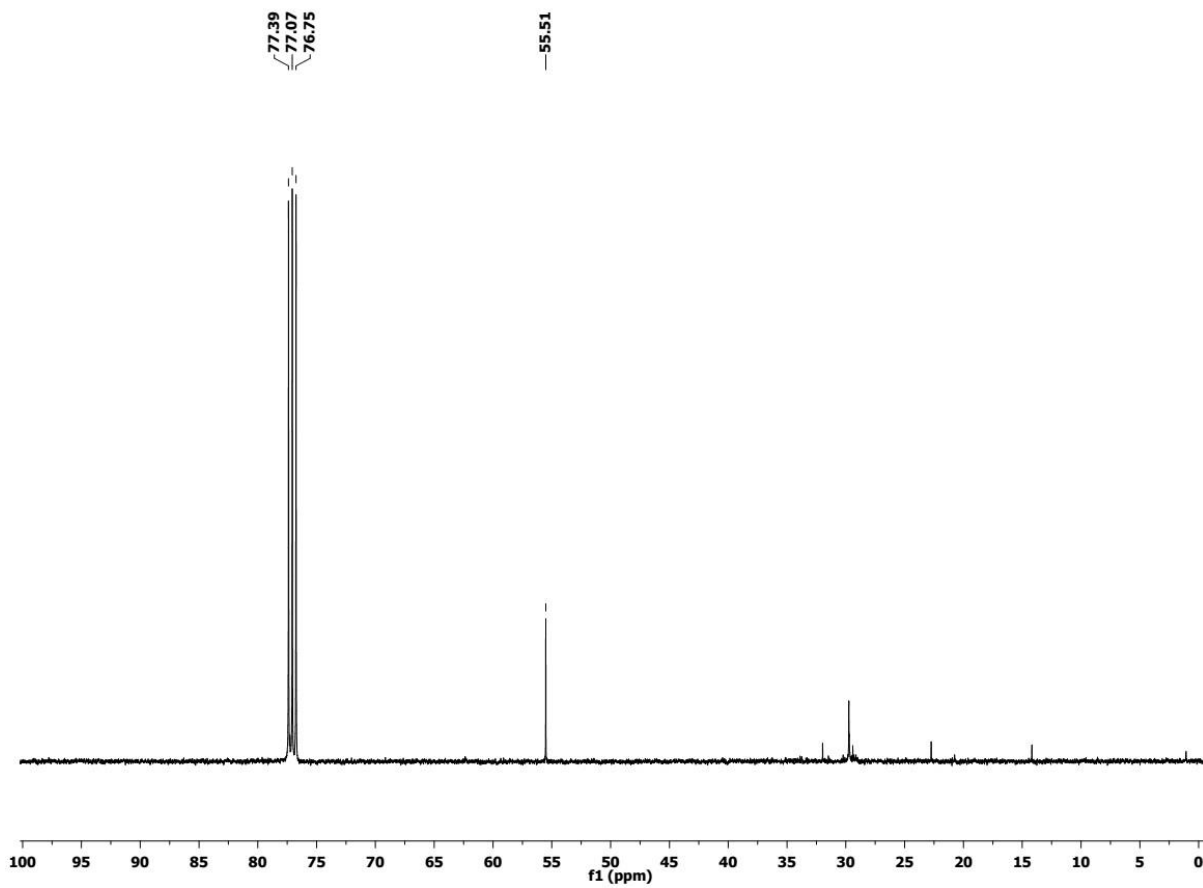
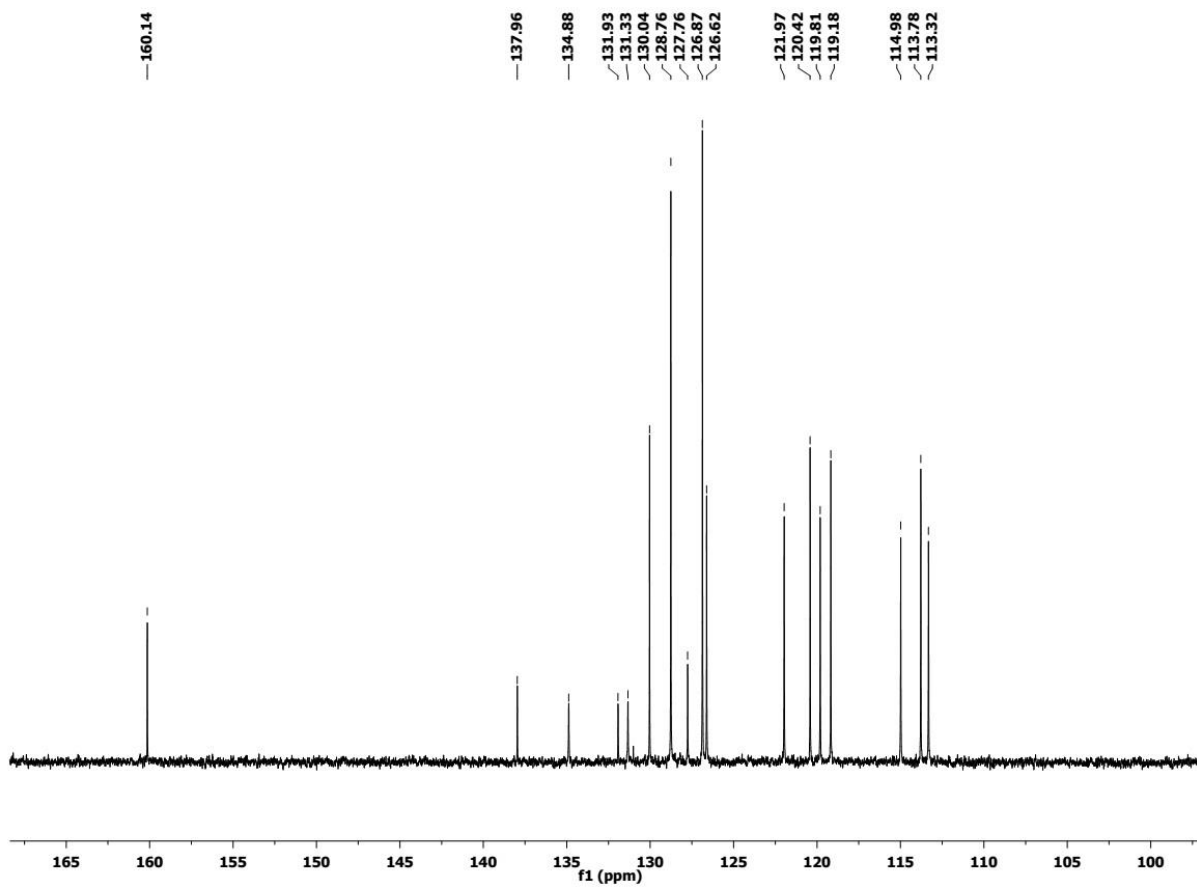
D<sub>2</sub>O NMR of Compound 3u

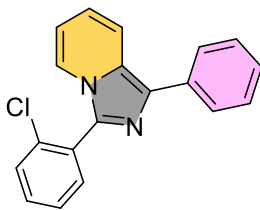




$^{13}\text{C}$  NMR of Compound 3u

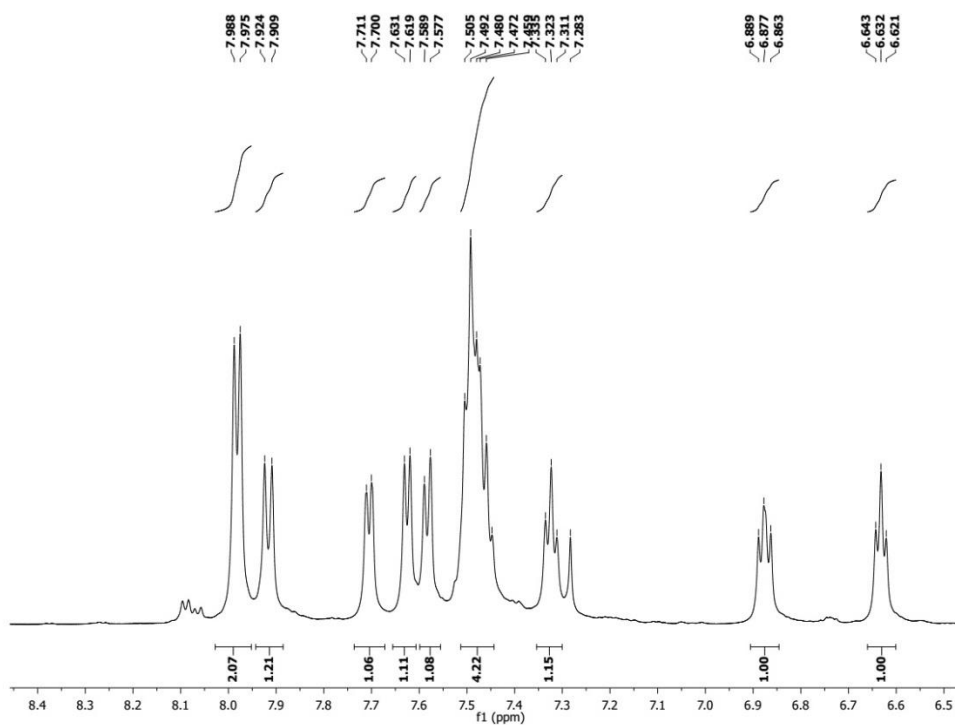
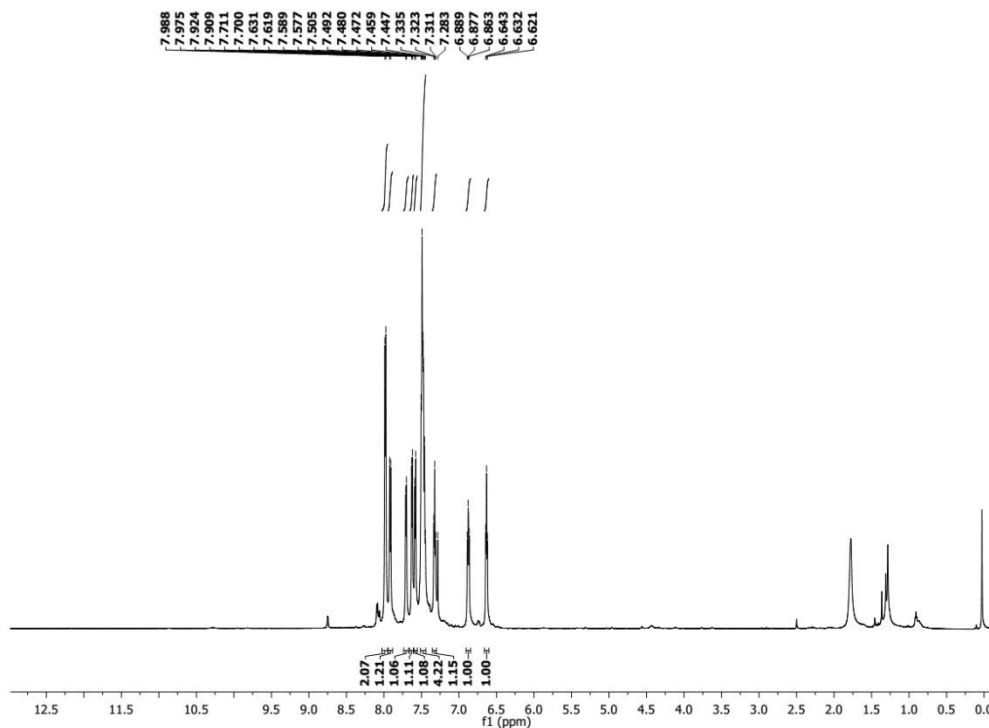




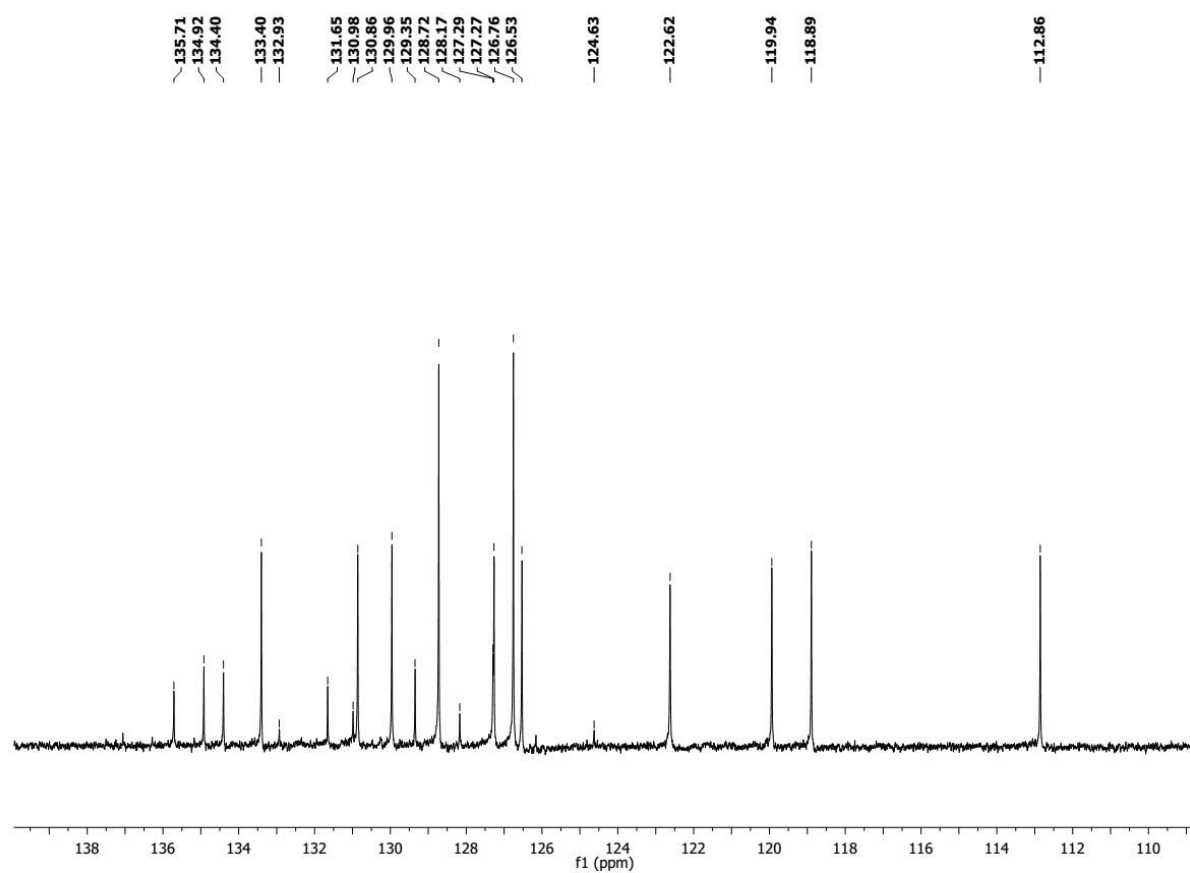
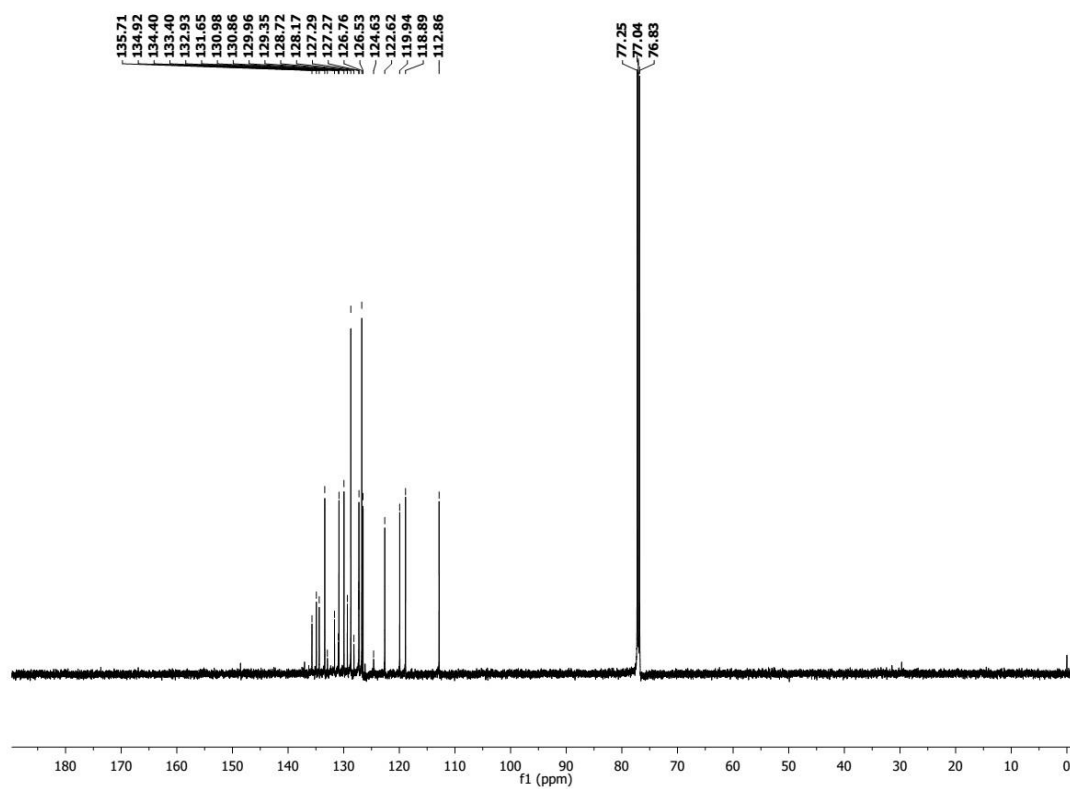


Compound (3v)

$^1\text{H}$  NMR of Compound 3v



# <sup>13</sup>C NMR of Compound 3v



## References

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6. U. Gulati, U. C. Rajesh, D. S. Rawat, *ACS Sustainable Chemistry & Engineering.*, 2018, **6**, 10039-10051.
7. A. Joshi, D. C. Mohan, S. Adimurthy, *J. Org. Chem.*, 2016, **19**, 9461-9469.