

*Supplementary Information*

**A Comparative Study of Two Aldehyde Dehydrogenases from  
*Sphinobium* sp.: Substrate Spectrum and Catalytic Mechanism**

*Siyi Chen, Jieyu Zhou\*, Xiangyuan Gu, Ye Ni\**

Key laboratory of industrial Biotechnology, Ministry of Education, School of Biotechnology,

Jiangnan University, Wuxi 214122, Jiangsu, China

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**Table S1.** Molar absorbance coefficient ( $\epsilon$ ) at different pH conditions.

	pH	equation	R <sup>2</sup>	$\epsilon(\mu\text{L} \cdot \mu\text{mol}^{-1} \cdot \text{cm}^{-1})$
citric acid/sodium citrate buffer (pH 5.0-6.5, 100 mM)	5.0	$y = 0.790x + 0.029$	0.984	1.271
	5.5	$y = 0.784x + 0.028$	0.990	1.260
	6.0	$y = 0.948x + 0.034$	0.983	1.524
	6.5	$y = 1.666x + 0.028$	0.985	2.679
phosphate buffer (pH 6.0-8.0, 50 mM)	6.0	$y = 1.110x + 0.024$	0.982	1.785
	6.5	$y = 2.140x - 0.001$	0.984	3.441
	7.0	$y = 4.886x - 0.066$	0.994	7.856
	7.5	$y = 6.113x + 0.006$	0.993	9.828
	8.0	$y = 7.140x + 0.025$	0.999	11.479
Tris-HCl (pH 7.0-9.0, 50 mM)	7.0	$y = 6.311x - 0.107$	0.992	10.146
	7.5	$y = 5.883x + 0.057$	0.995	9.458
	8.0	$y = 6.377x + 0.065$	0.998	10.252
	8.5	$y = 6.933x + 0.094$	0.998	11.147
	9.0	$y = 8.001x + 0.054$	0.999	12.863
Glycine-NaOH (pH 8.5-11.0, 50 mM)	8.7	$y = 7.890x + 0.039$	0.998	12.685
	9.0	$y = 8.022x + 0.036$	0.998	12.897
	9.6	$y = 7.206x + 0.070$	0.999	11.586
	10.5	$y = 7.310x + 0.094$	0.999	11.753
	11.0	$y = 7.115x + 0.142$	0.999	11.439
	11.5	$y = 6.905x + 0.210$	0.999	11.102

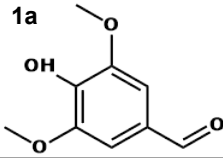
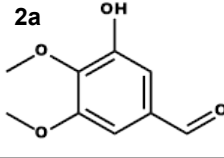
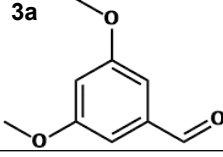
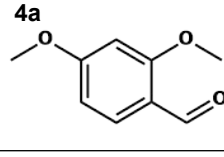
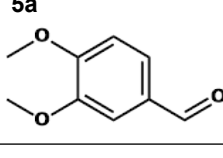
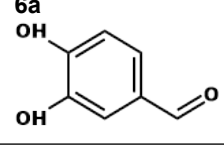
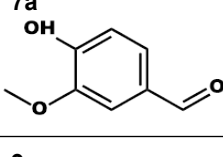
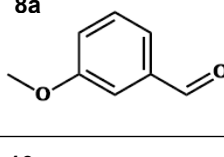
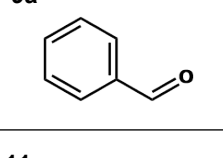
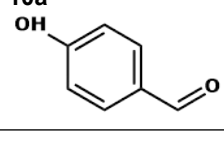
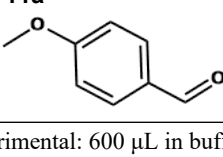
**Table S2.** Coupling reaction between ALDHs and NOX (600µl reaction system (35°C))

	Final concentration
1a	20 mM
NAD <sup>+</sup>	0.5 mM
NOX	0.5 mg/mL
<i>Sp</i> ALDH1/ <i>Sp</i> ALDH2	0.1 mg/mL
Citric acid/sodium citrate buffer (100mM, pH 6.5)	/
/Tris-HCl (100mM, pH 8.0)	/

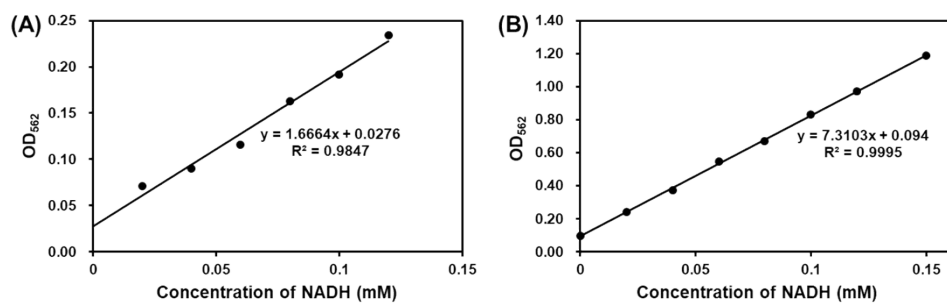
**Table S3.** Coupling reaction between ALDHs and NOX (600µl reaction system (35°C))

	Final concentration
1a	20 mM
NAD <sup>+</sup>	0.5 mM
NOX	1.5 mg/mL
<i>Sp</i> ALDH1/ <i>Sp</i> ALDH2	0.3 mg/mL
Citric acid/sodium citrate buffer (100mM, pH 6.5)	/
/Tris-HCl (100mM, pH 8.0)	/

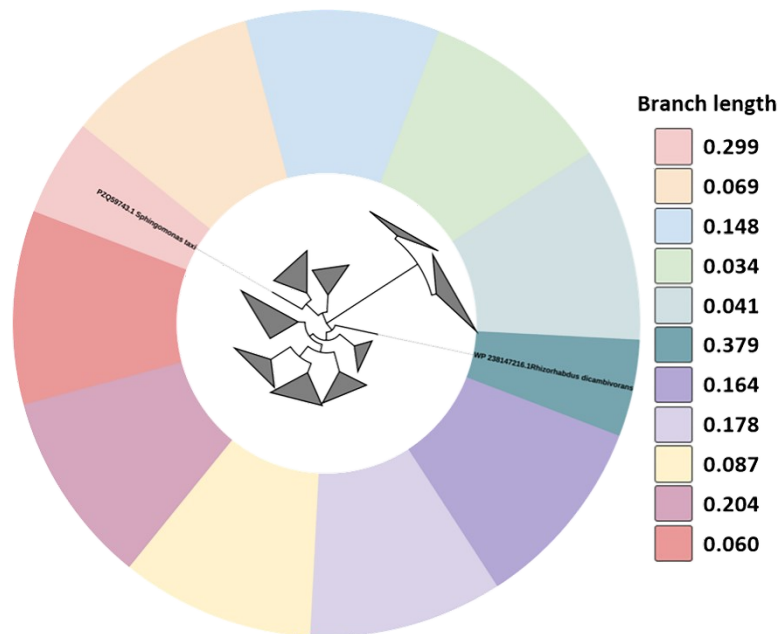
**Table S4.** Biocatalytic oxidation of aromatic aldehydes to carboxylic acids catalyzed by ALDHs.

Substrate	ALDH	Conv. [%]	Substrate	ALDH	Conv. [%]
<b>1a</b> 	<i>Sp</i> ALDH1	> 99	<b>2a</b> 	<i>Sp</i> ALDH1	> 99
	<i>Sp</i> ALDH2	> 99		<i>Sp</i> ALDH2	> 99
<b>3a</b> 	<i>Sp</i> ALDH1	> 99	<b>4a</b> 	<i>Sp</i> ALDH1	3.2
	<i>Sp</i> ALDH2	> 99		<i>Sp</i> ALDH2	> 99
<b>5a</b> 	<i>Sp</i> ALDH1	> 99	<b>6a</b> 	<i>Sp</i> ALDH1	5.3
	<i>Sp</i> ALDH2	> 99		<i>Sp</i> ALDH2	> 99
<b>7a</b> 	<i>Sp</i> ALDH1	> 99	<b>8a</b> 	<i>Sp</i> ALDH1	> 99
	<i>Sp</i> ALDH2	> 99		<i>Sp</i> ALDH2	> 99
<b>9a</b> 	<i>Sp</i> ALDH1	> 99	<b>10a</b> 	<i>Sp</i> ALDH1	> 99
	<i>Sp</i> ALDH2	> 99		<i>Sp</i> ALDH2	> 99
<b>11a</b> 	<i>Sp</i> ALDH1	> 99			
	<i>Sp</i> ALDH2	> 99			

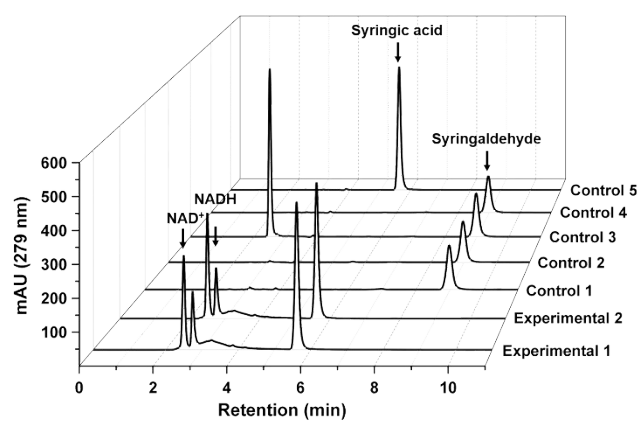
Experimental: 600  $\mu$ L in buffer, T = 35  $^{\circ}$ C, 100 rpm, [substrate] = 5 mM, [ALDH] = 0.2 mg/mL, [NAD<sup>+</sup>] = 10 mM. Citrate buffer (100 mM, pH 6.5) was used to test *Sp*ALDH1, while glycine-NaOH buffer (50 mM, pH 10.5) was employed for *Sp*ALDH2. Conversions were measured by HPLC.



**Fig S1.** Molar absorbance coefficient ( $\epsilon$ ) at different pH conditions. (A) Citric acid/sodium citrate buffer (pH 6.5, 100 mM); (B) Glycine-NaOH (pH 10.5, 50 mM).



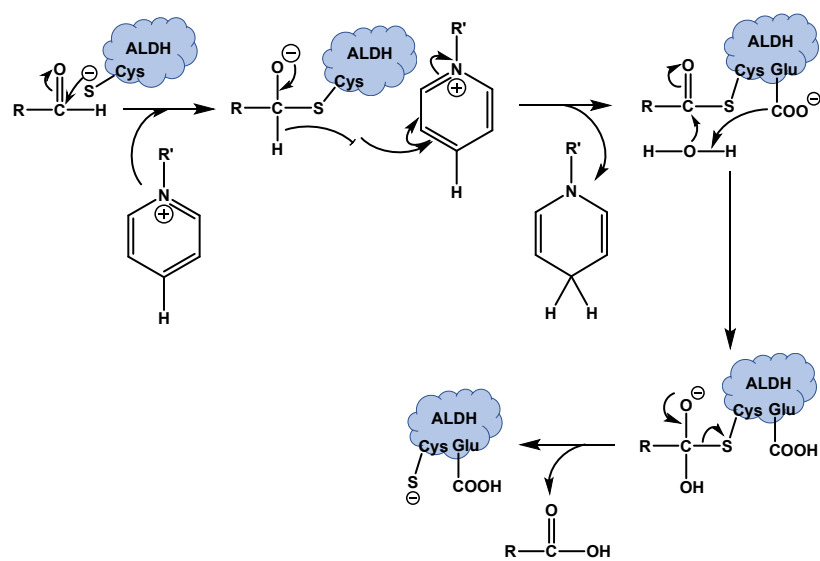
**Fig. S2** Phylogenetic analysis of *SpALDH1*, *SpALDH2* and other bacterial ALDHs. Use differences in branch lengths as a basis for coloring. Auto collapse clade is set to be 0.5.



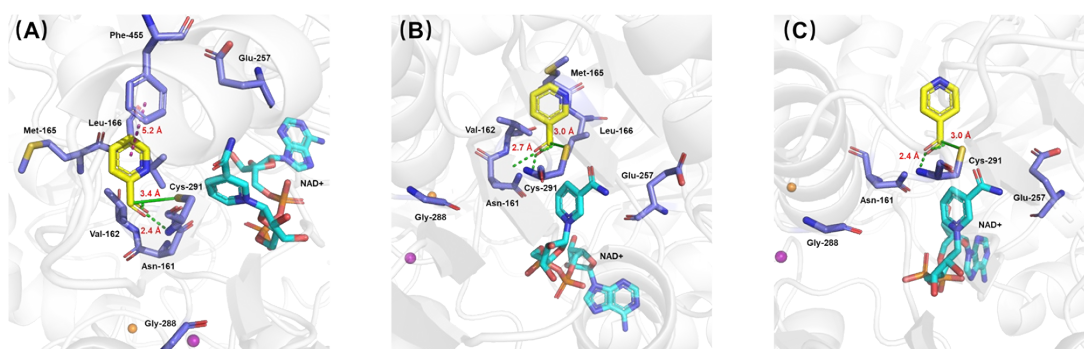
	<i>Sp</i> ALDH1	<i>Sp</i> ALDH2	NAD <sup>+</sup>	Syringaldehyde	Syringic acid	Buffer
Experimental 1	+	-	+	+	-	+
Experimental 2	-	+	+	+	-	+
Control 1	+	-	-	+	-	+
Control 2	-	+	-	+	-	+
Control 3	-	-	+	+	-	+
Control 4	-	-	-	+	-	+
Control 5	-	-	-	-	+	+

**Fig. S3** HPLC analysis of syringaldehyde conversions by purified ALDHs. Controlled experiments of ALDHs biocatalyzed oxidation of syringaldehyde.

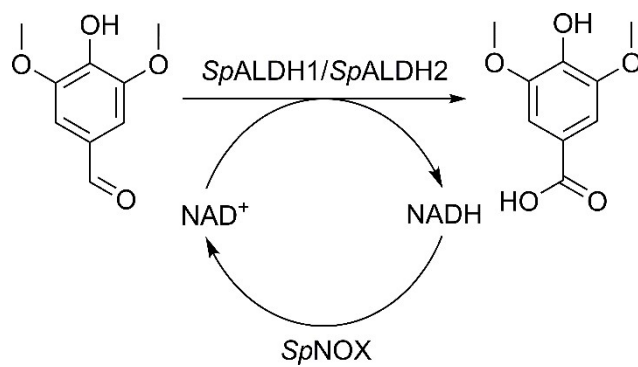




**Fig S4.** Catalytic mechanism of ALDHs toward aldehyde substrates.



**Fig S5.** Molecular docking of *Sp*ALDH1 with pyridine-formaldehydes. (A) 22a; (B) 23a; (C) 24a. Active site residues, substrates and NAD<sup>+</sup> cofactor are showed as sticks, represented by slate (carbon atom), yellow (carbon atom) and cyan (carbon atom), respectively. The ALDHs are depicted as grey. Oxygen atoms (red); Nitrogen atoms (blue); Sulfur atoms (yellow orange); Potassium ions (purple ball); Sodium ions (orange ball). The hydrogen bonds are showed as green dotted lines; the pi-pi stacking interactions are showed as light magenta dotted lines.



**Fig. S6** Preparation of syringic acid by coupling ALDHs with *Sp*NOX.

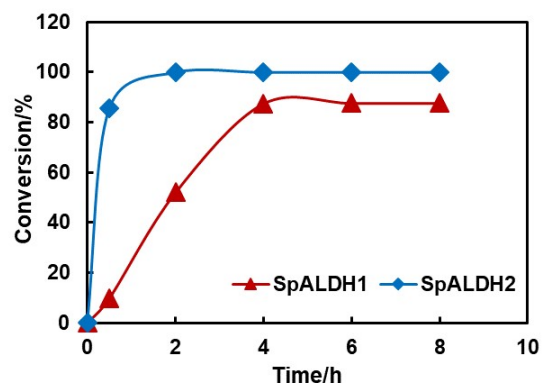


Fig S7. Coupled reaction catalyzed by ALDHs and NOX at substrate **1a** concentration of 20 mM.