

Δ-keto-acid /hydroxy-lactone isomerization in some lichen depsides, depsidones and diphenyl ethers

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SUPPORTING INFORMATION

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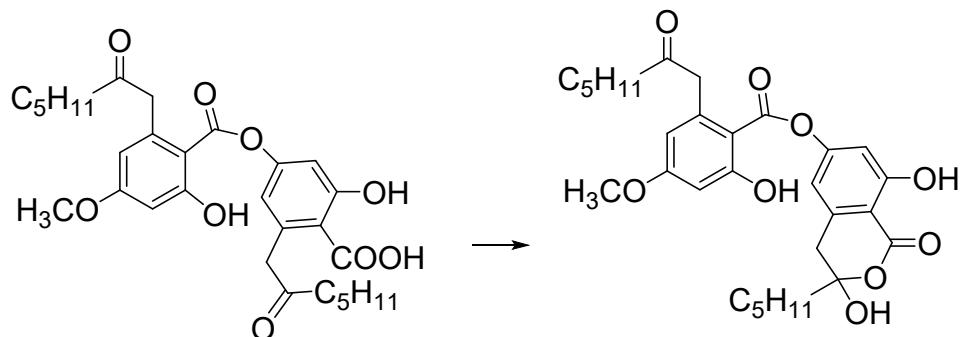
References

- 1 S. Huneck and I. Yoshimura, *Identification of Lichen Substances*, Springer Berlin Heidelberg, Berlin, Heidelberg, 1996.
- 2 S. Ferron, P. Jéhan, X. Guillory and P. Uriac, *Phytochemistry*, 2022, **198**, 113139.
- 3 M. Millot, S. Tomasi, S. Sinbandhit and J. Boustie, *Phytochemistry Letters*, 2008, **1**, 139–143.
- 4 M. Millot.; S.Tomasi; K. Articus; Rouaud, I.; A. Bernard; J. Boustie *J. Nat. Prod.* **2007**, *70* (2), 316–318.

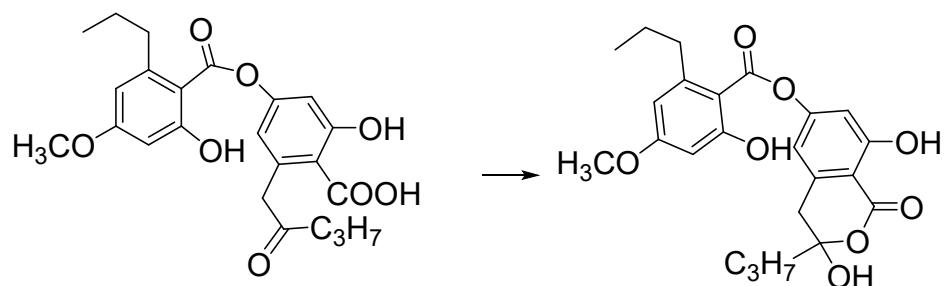
S1. Structure of other lichen specialized metabolites concerned by the study.

On the left the structure proposed in Hüneck and on the right the corrected structure according to the $2\text{H}-1''$ ^1H NMR value and this study.

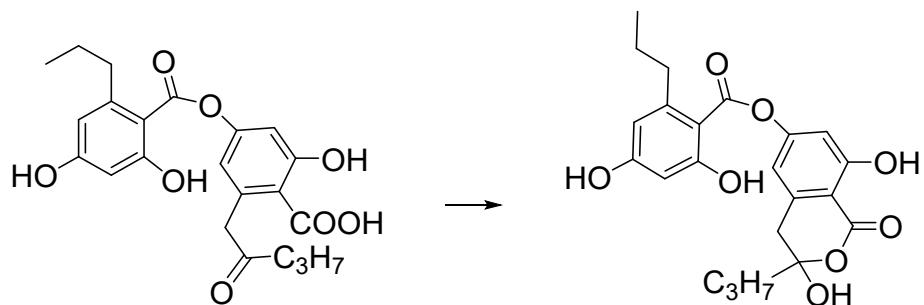
Compound with OH-2': hydroxy-lactone isomer



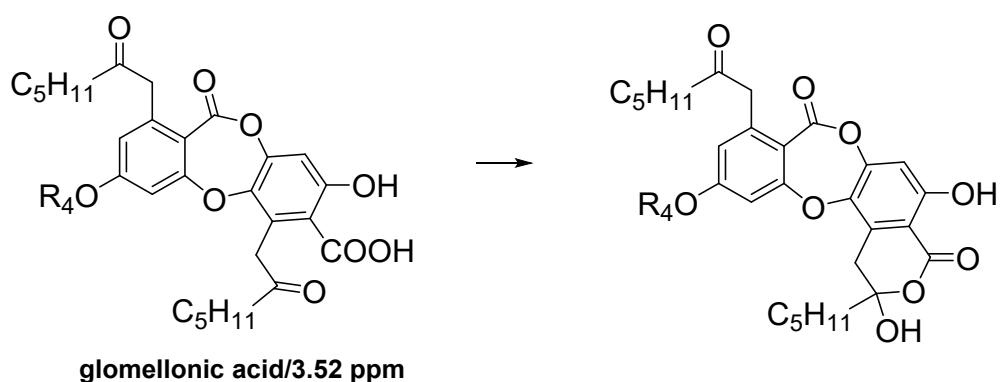
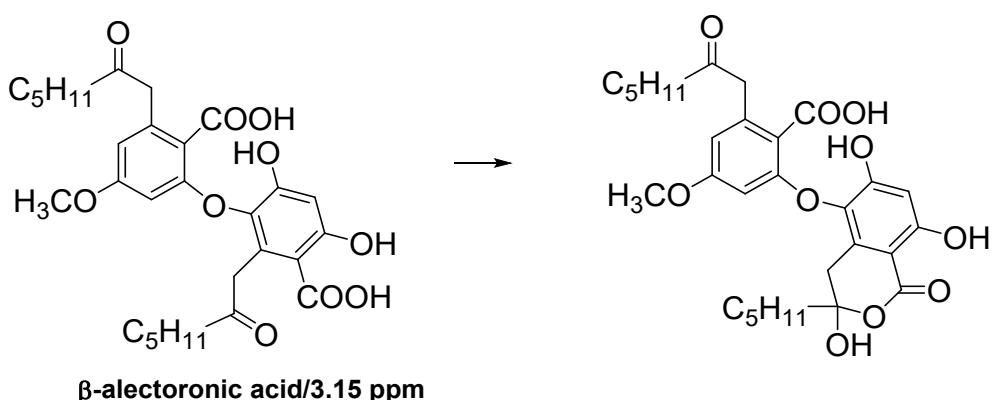
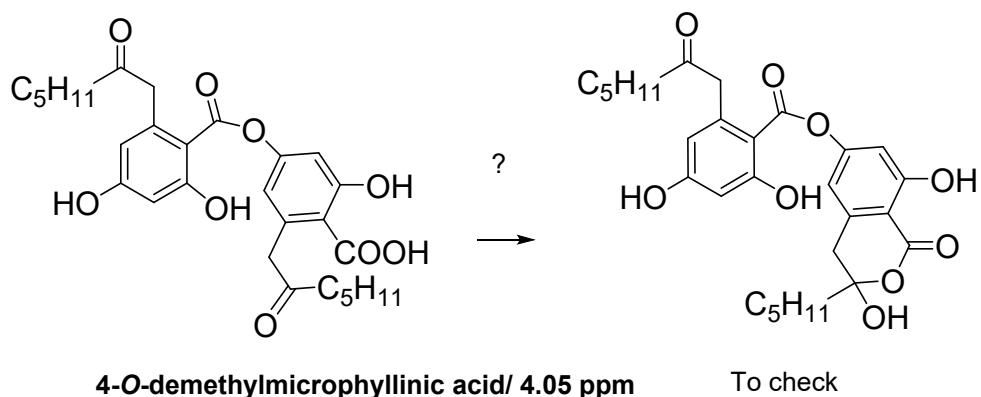
microphyllinic acid/ no data



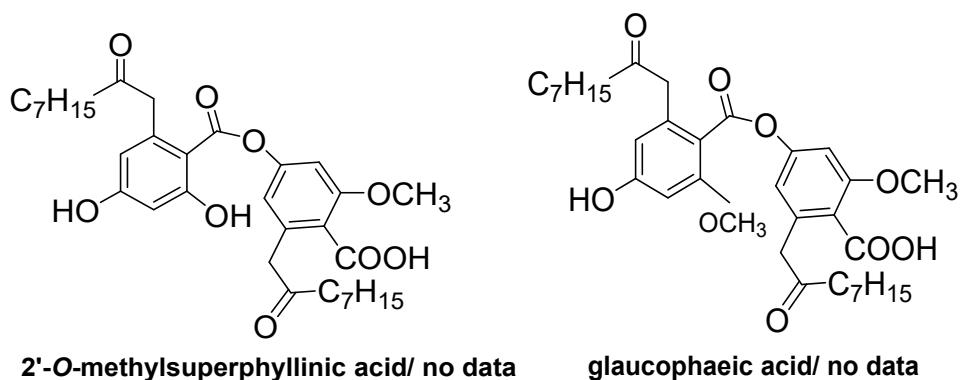
oxostenosporic acid/ 3.22 ppm



4-O-demethylglomellic acid/ 3.40 ppm

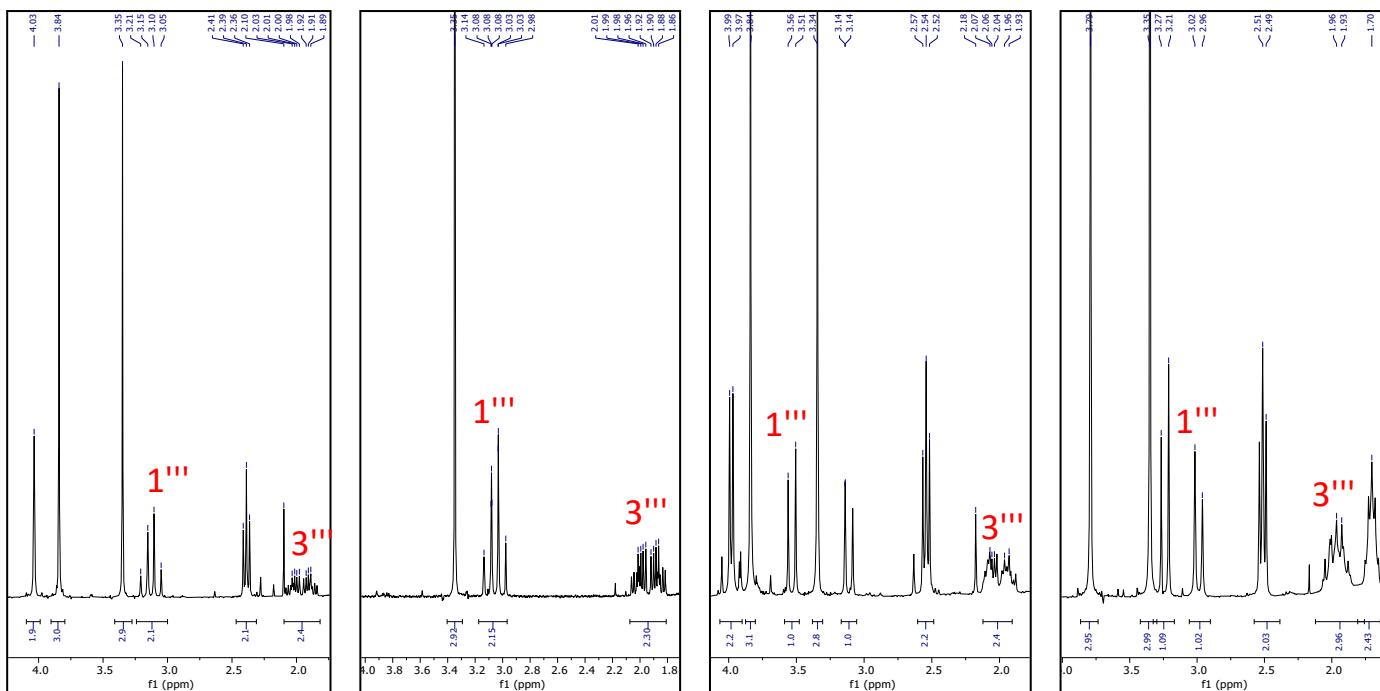


Compound with OCH_3 -2': δ -keto-acid isomer



S2. ^1H NMR data concerning $\text{CH}_2\text{-}1'''$ and $\text{CH}_2\text{-}3'''$ of **hI** and **ka** isomers

Methoxy-lactones hI



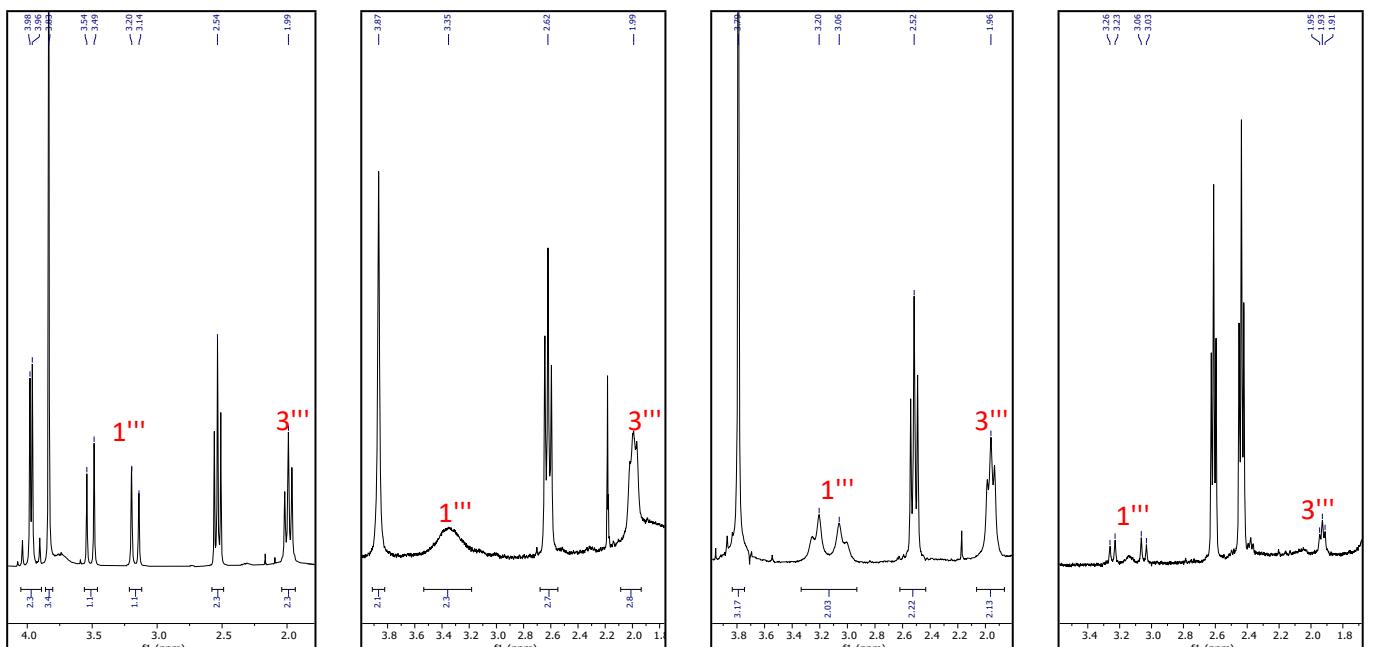
$2'''$ -O-methylglomellic acid **10**

Compound **13'**

$2'''$ -O-methyl- α -collatolic acid **11**

$2'''$ -O-methyl- β -collatolic acid **12**

Hydroxy-lactones hI



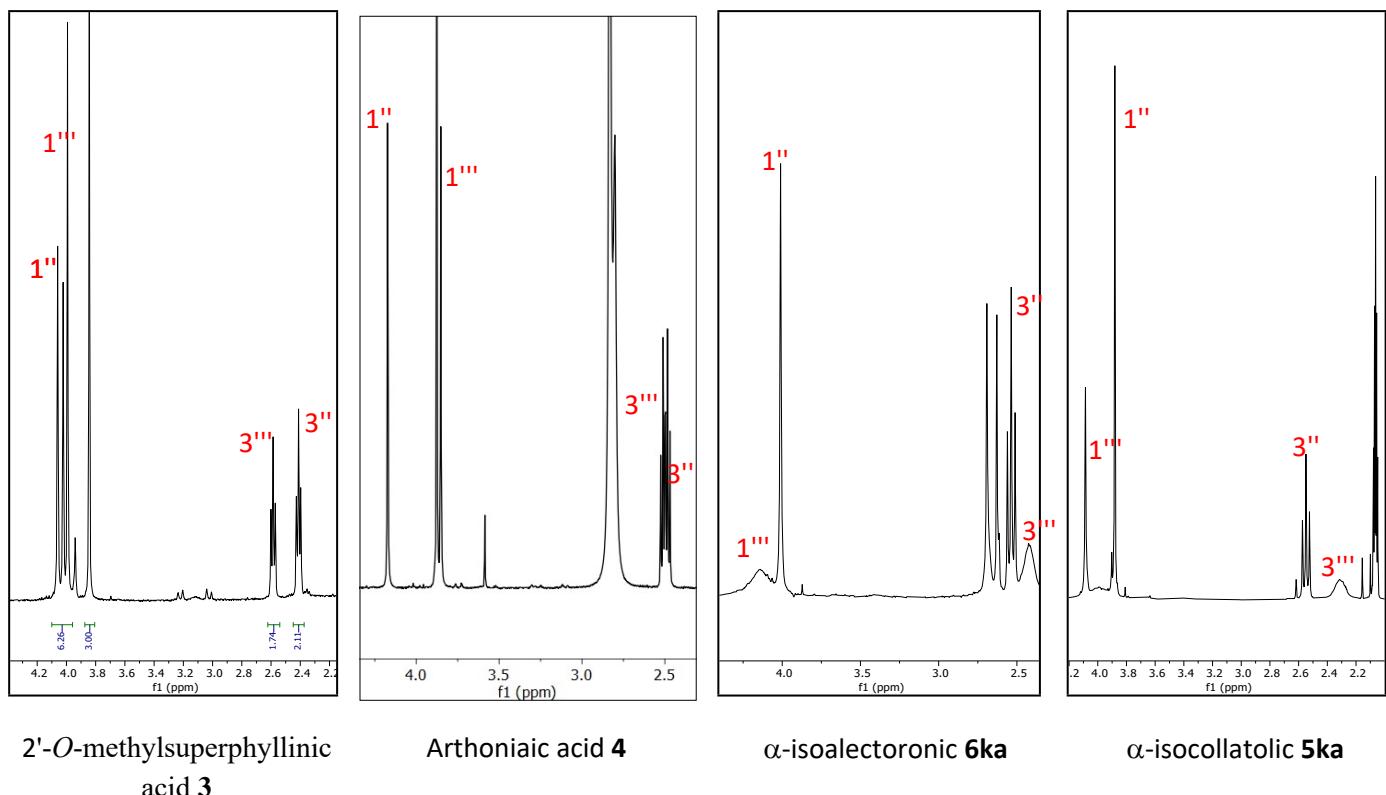
α -collatolic **5**

α -alectoronic **6**

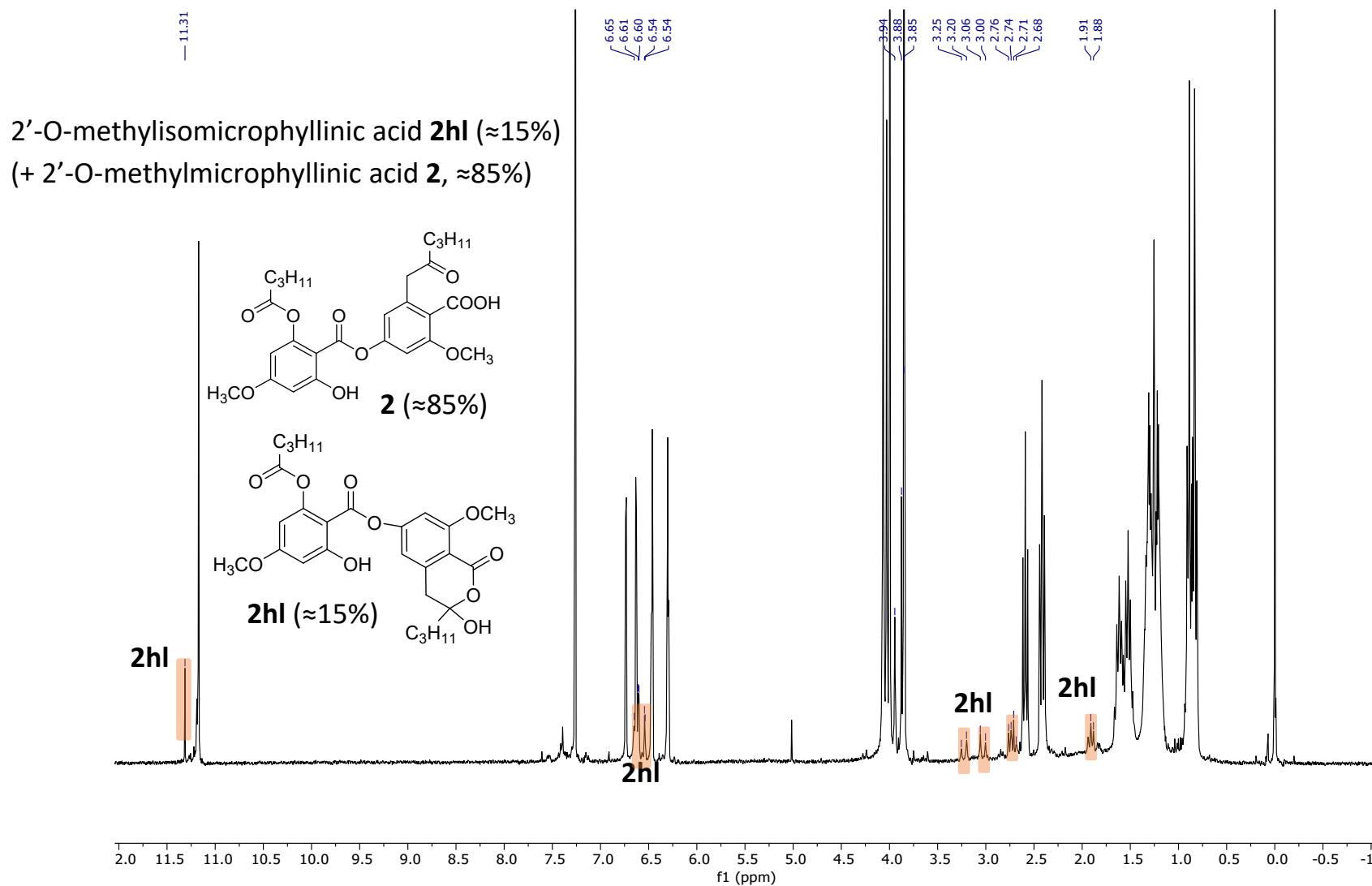
β -collatolic **7**

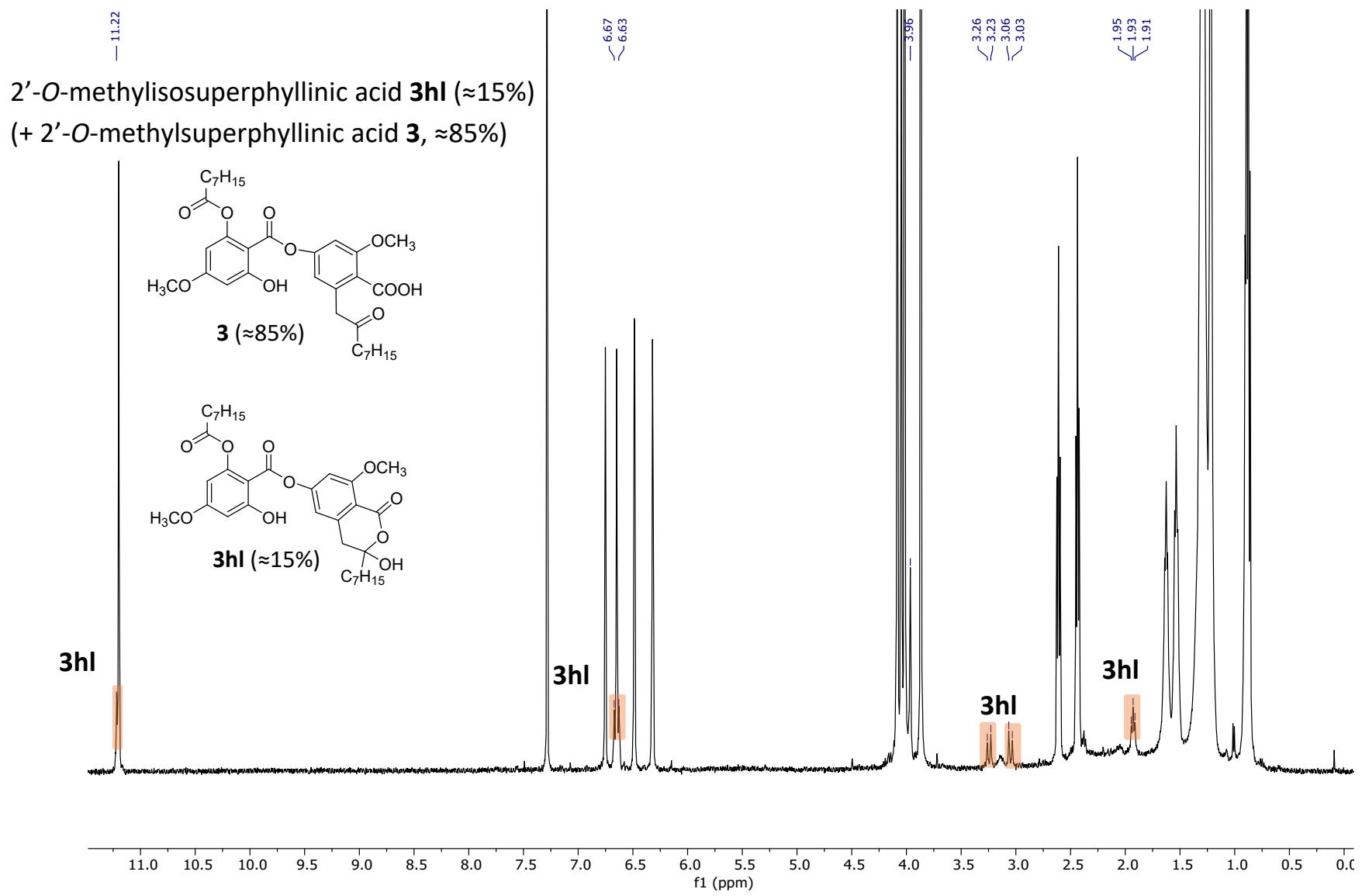
$2'$ -O-methylisomicrophyllinic **2hI**
(idem **3hI**)

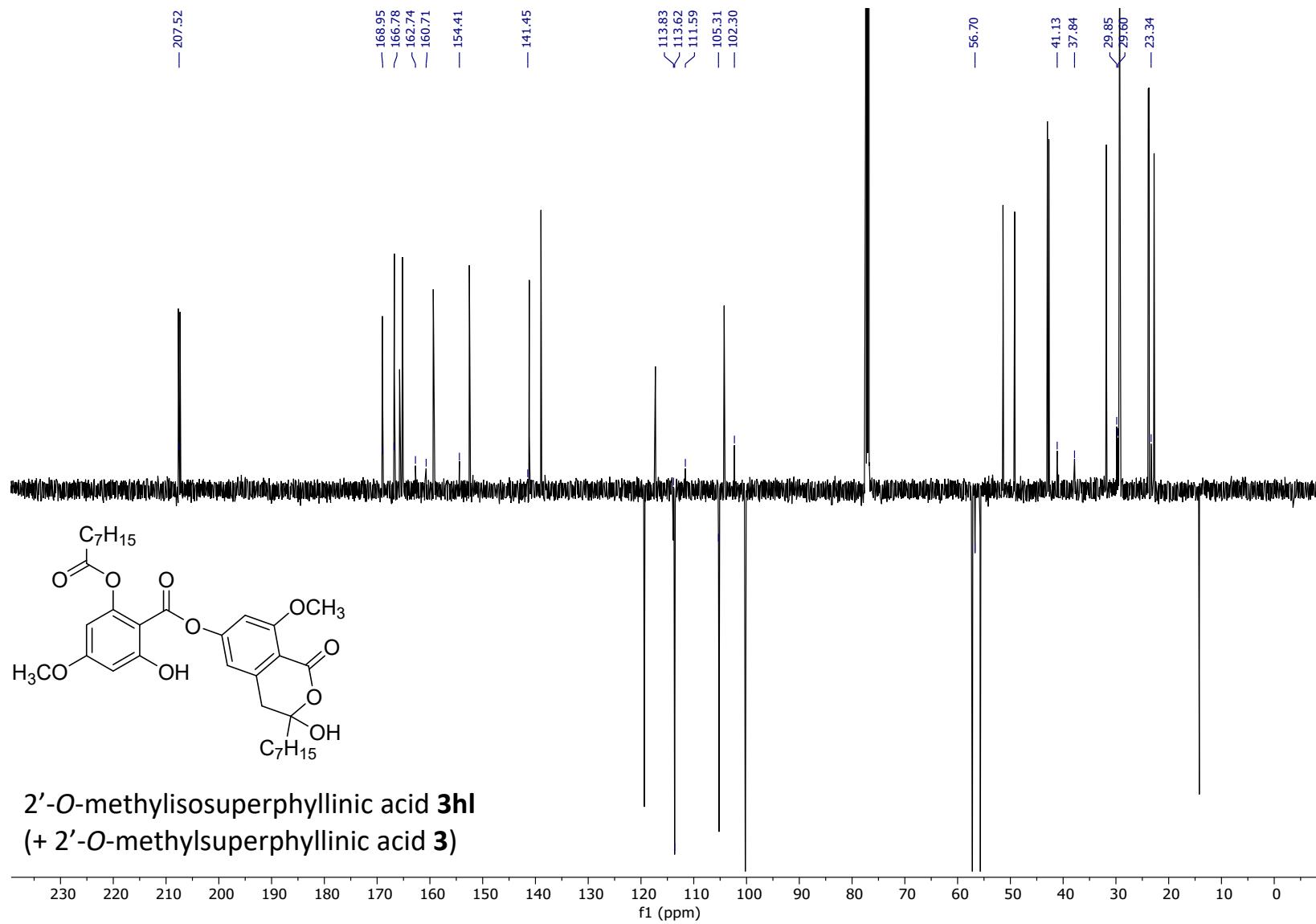
δ -keto-acides ka

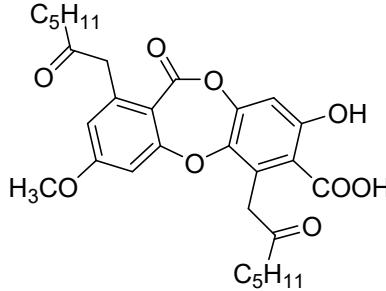


S3. NMR spectra of **2hl (+ **2** major), **3hl** (+ **3** major), **5ka**, **6ka**, **9**, **10**, **11**, **12**, and **13'****

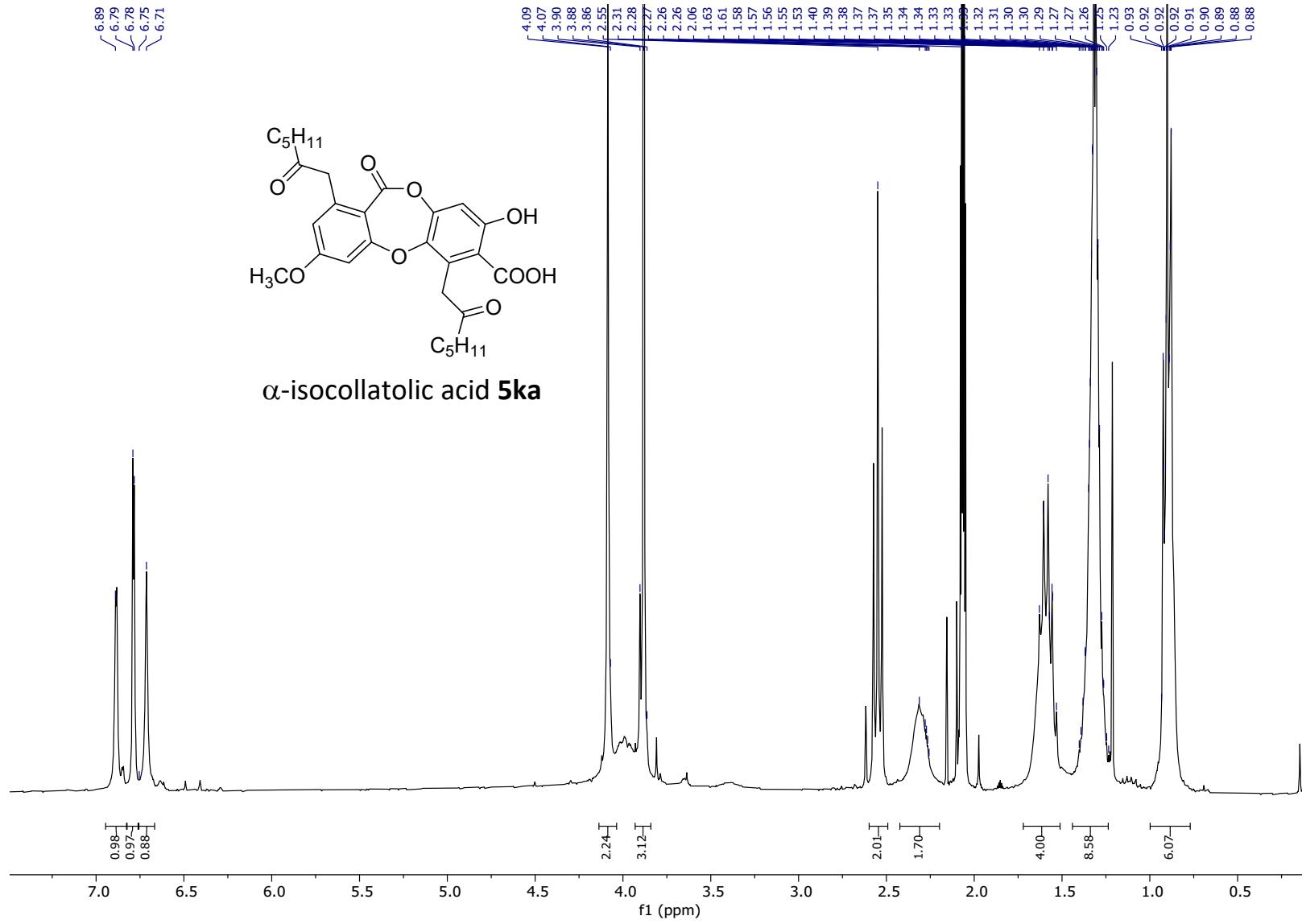


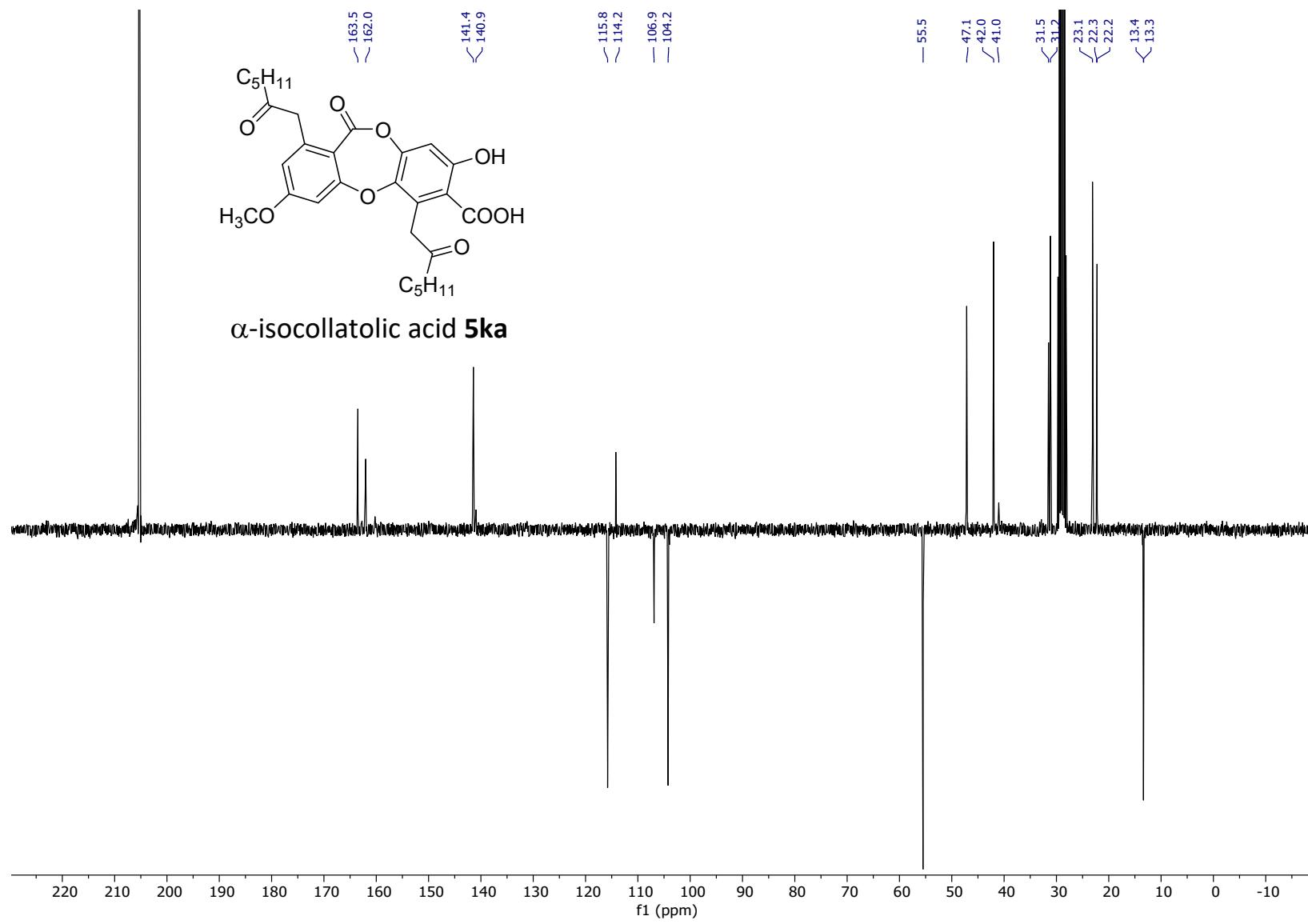


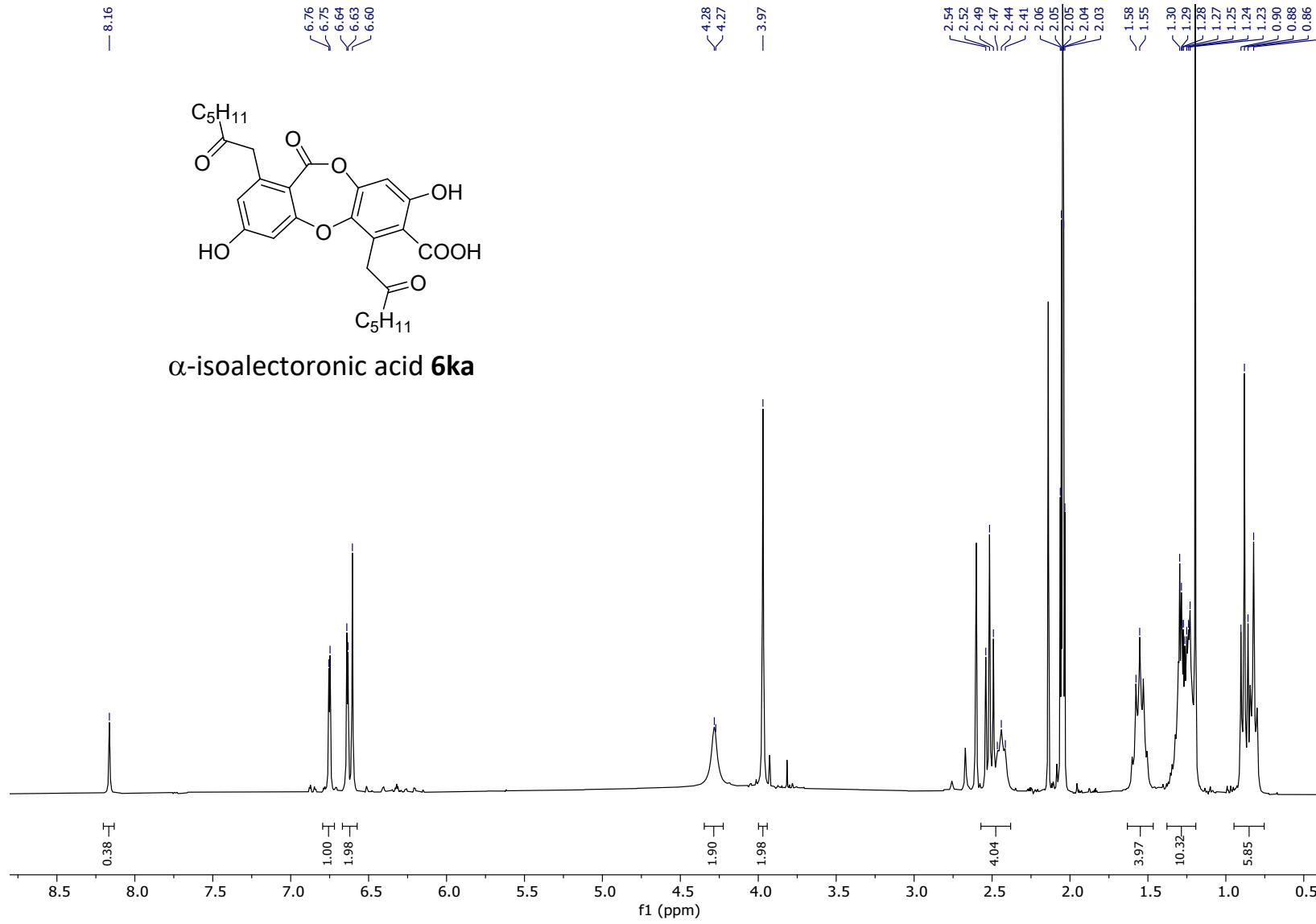


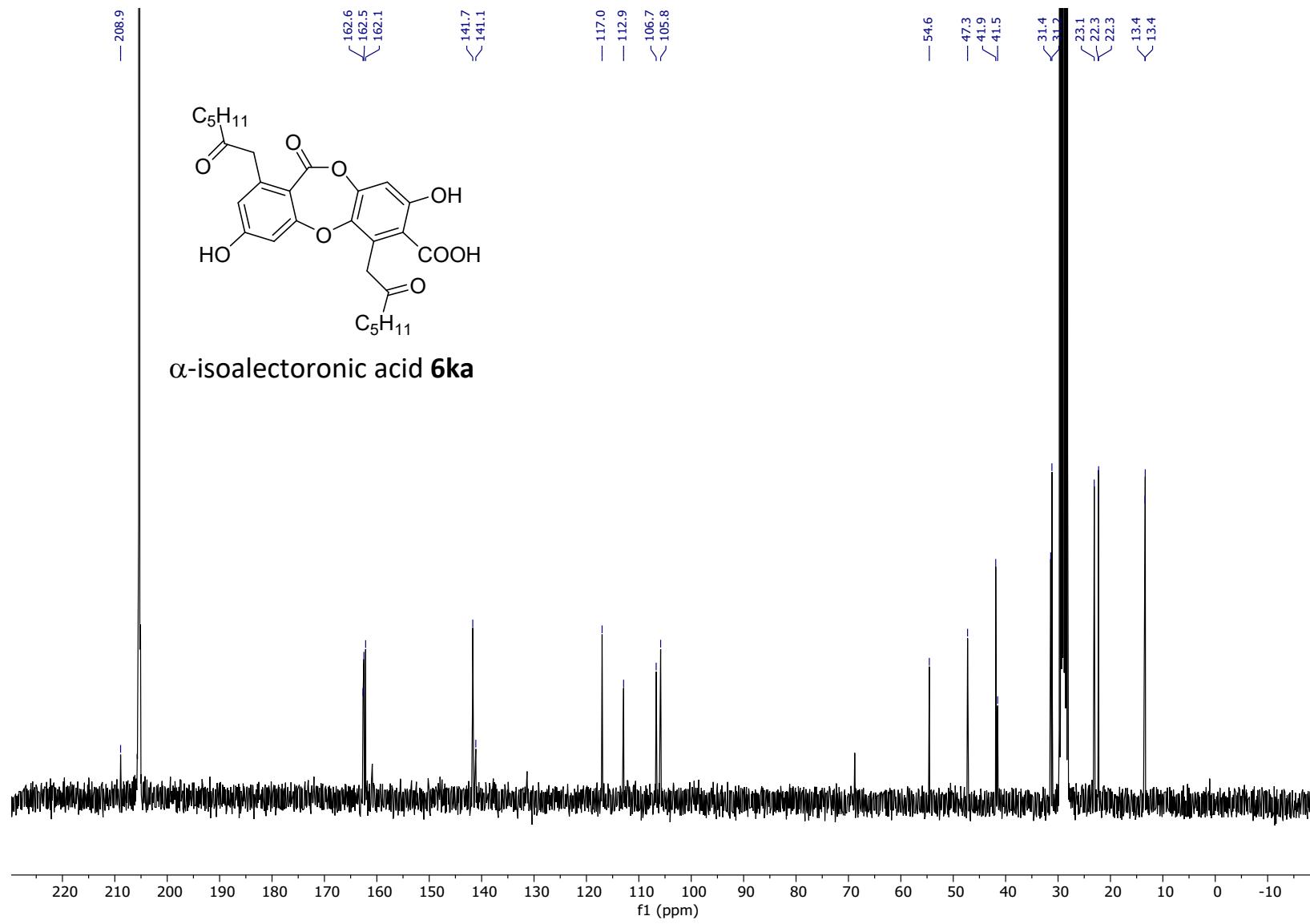


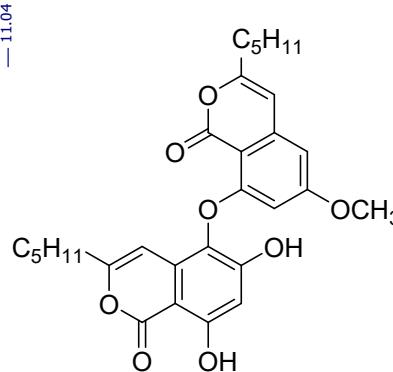
α -isocollatolic acid 5ka



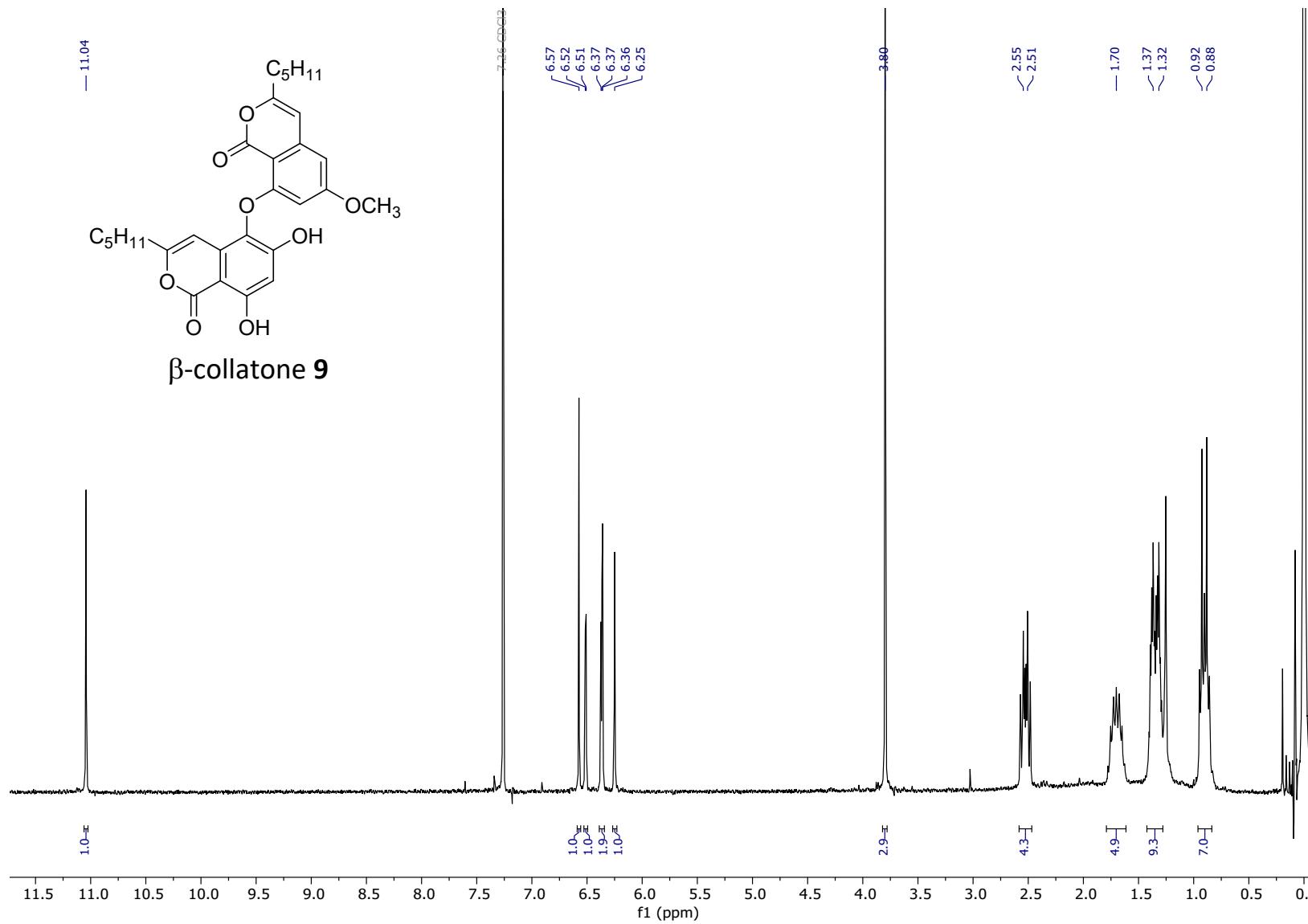


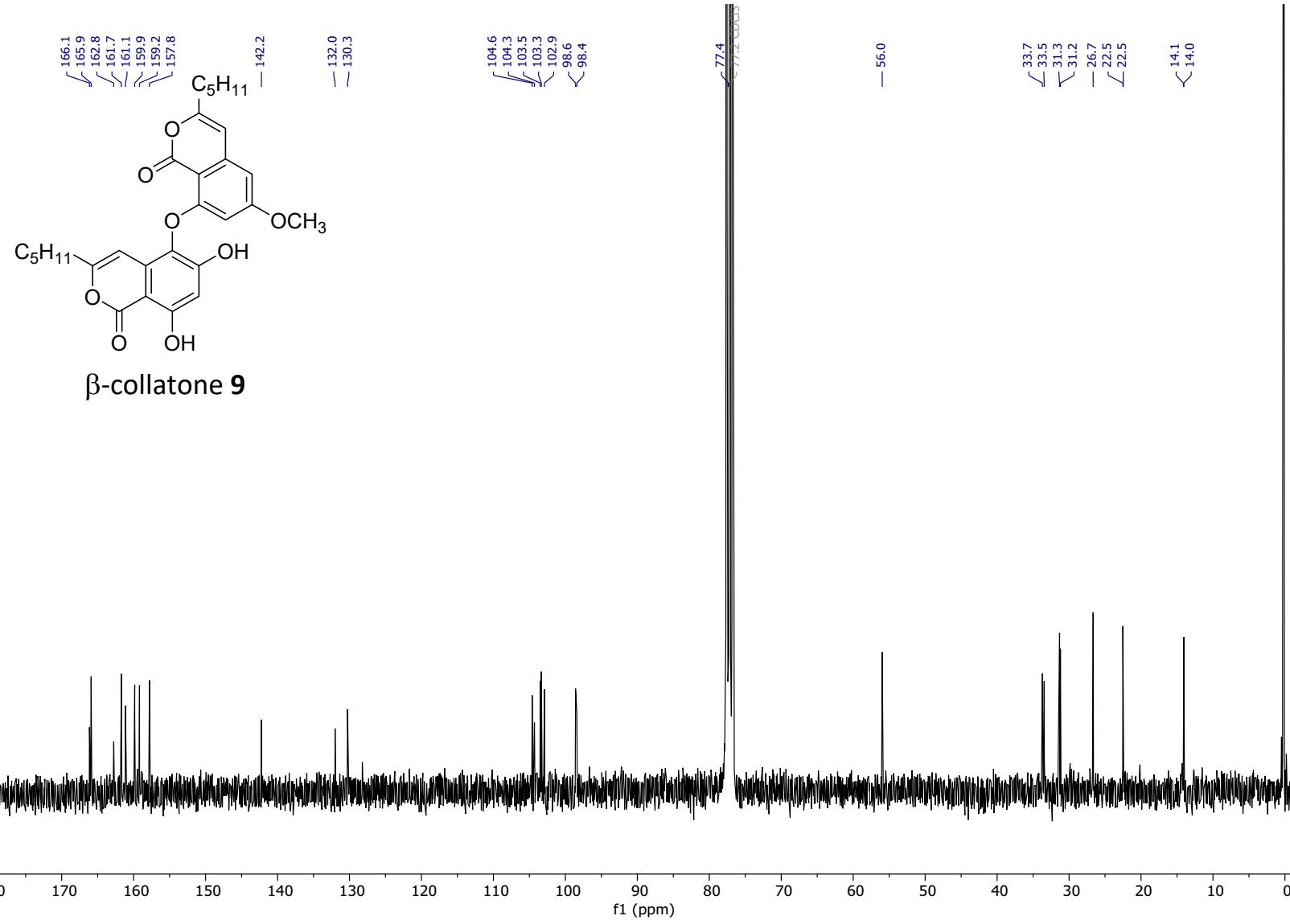


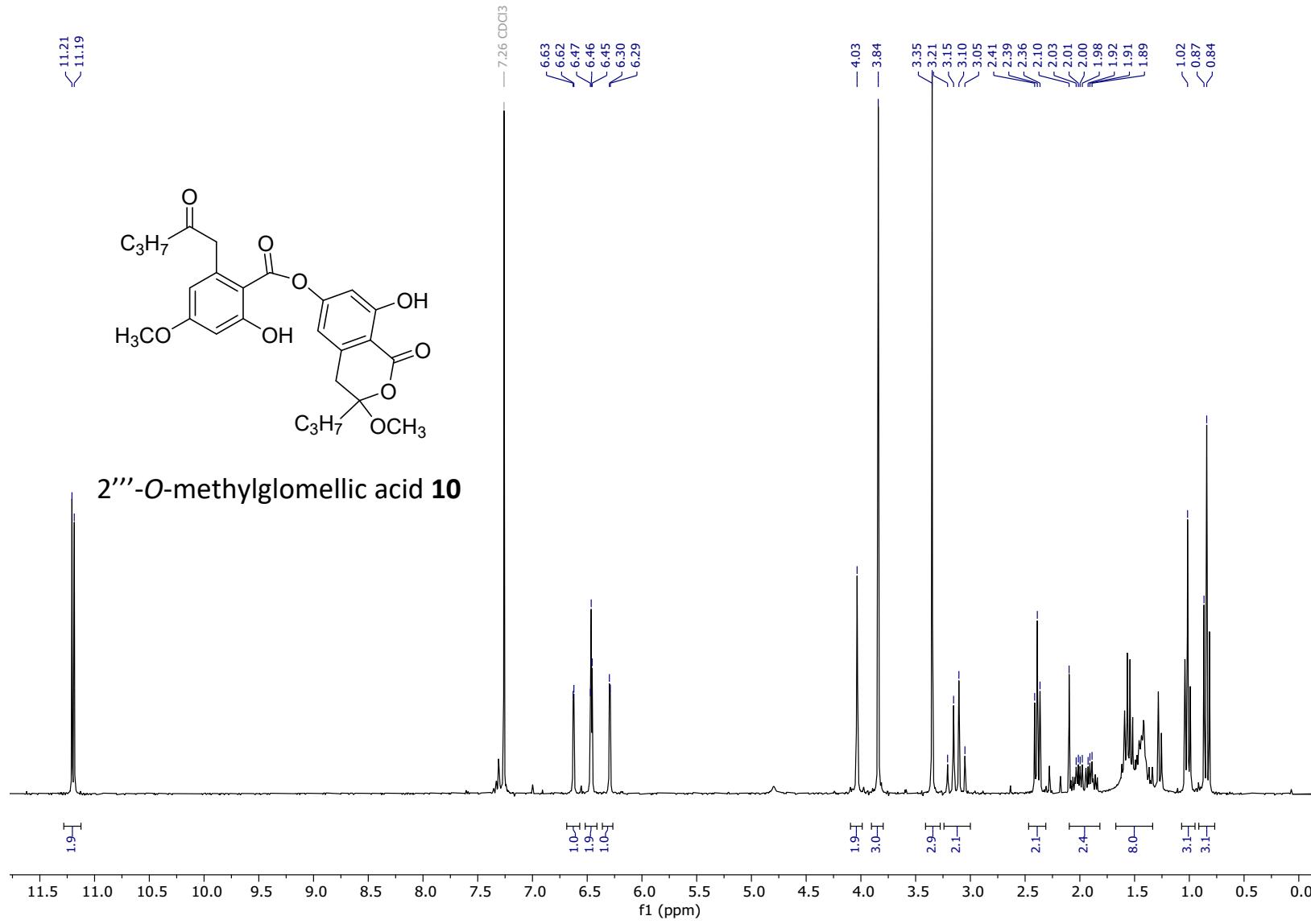


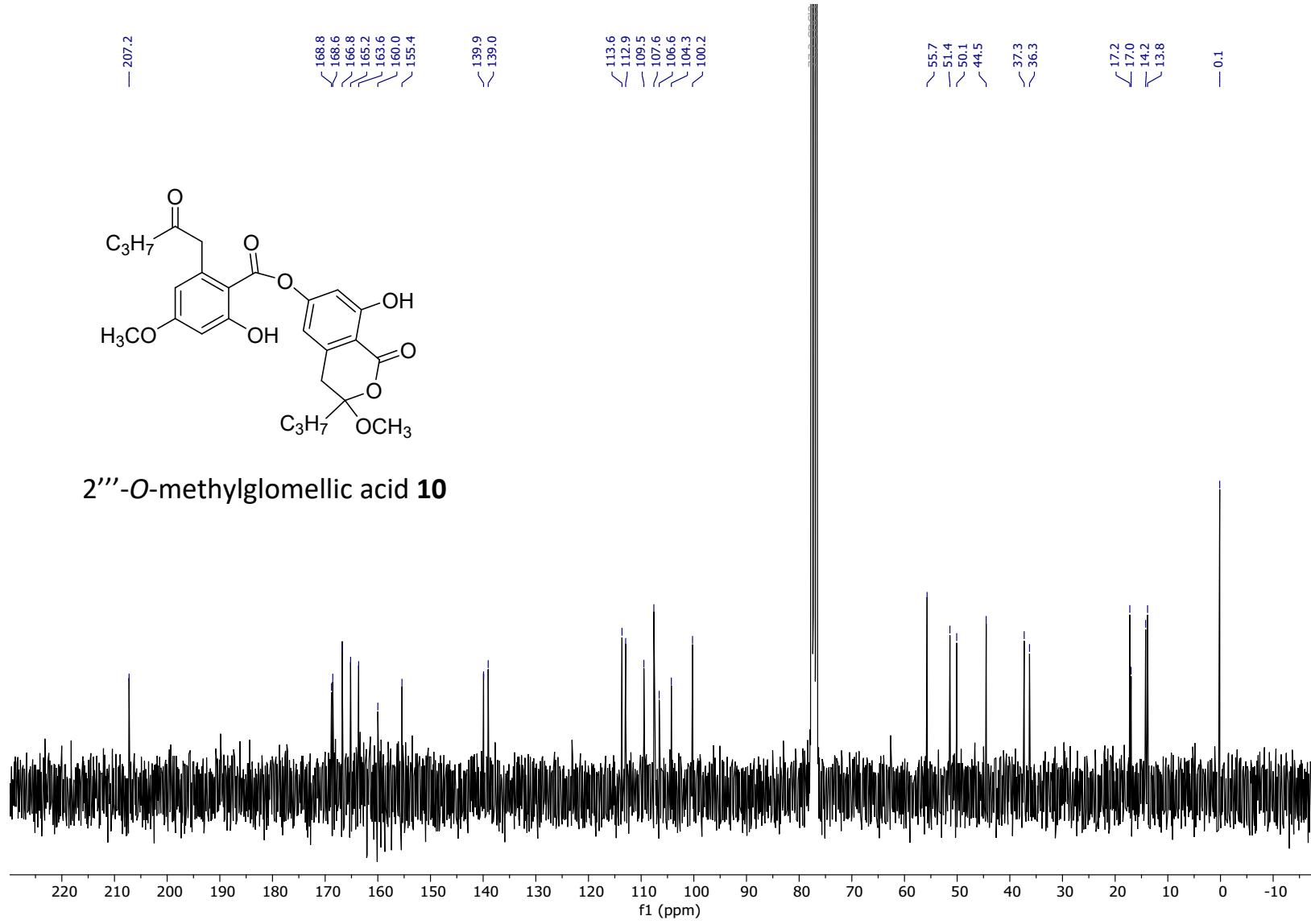


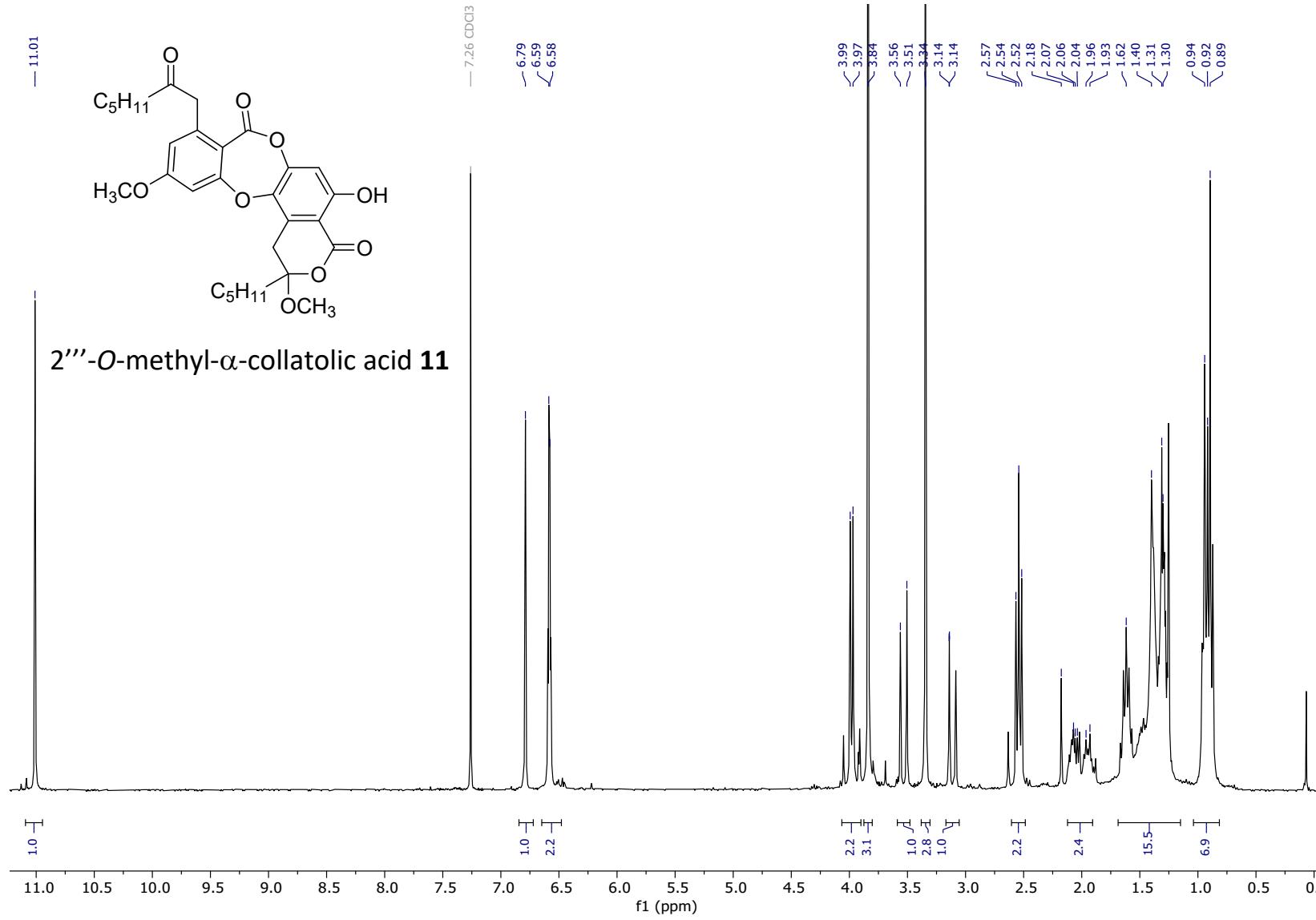
β -collatone 9

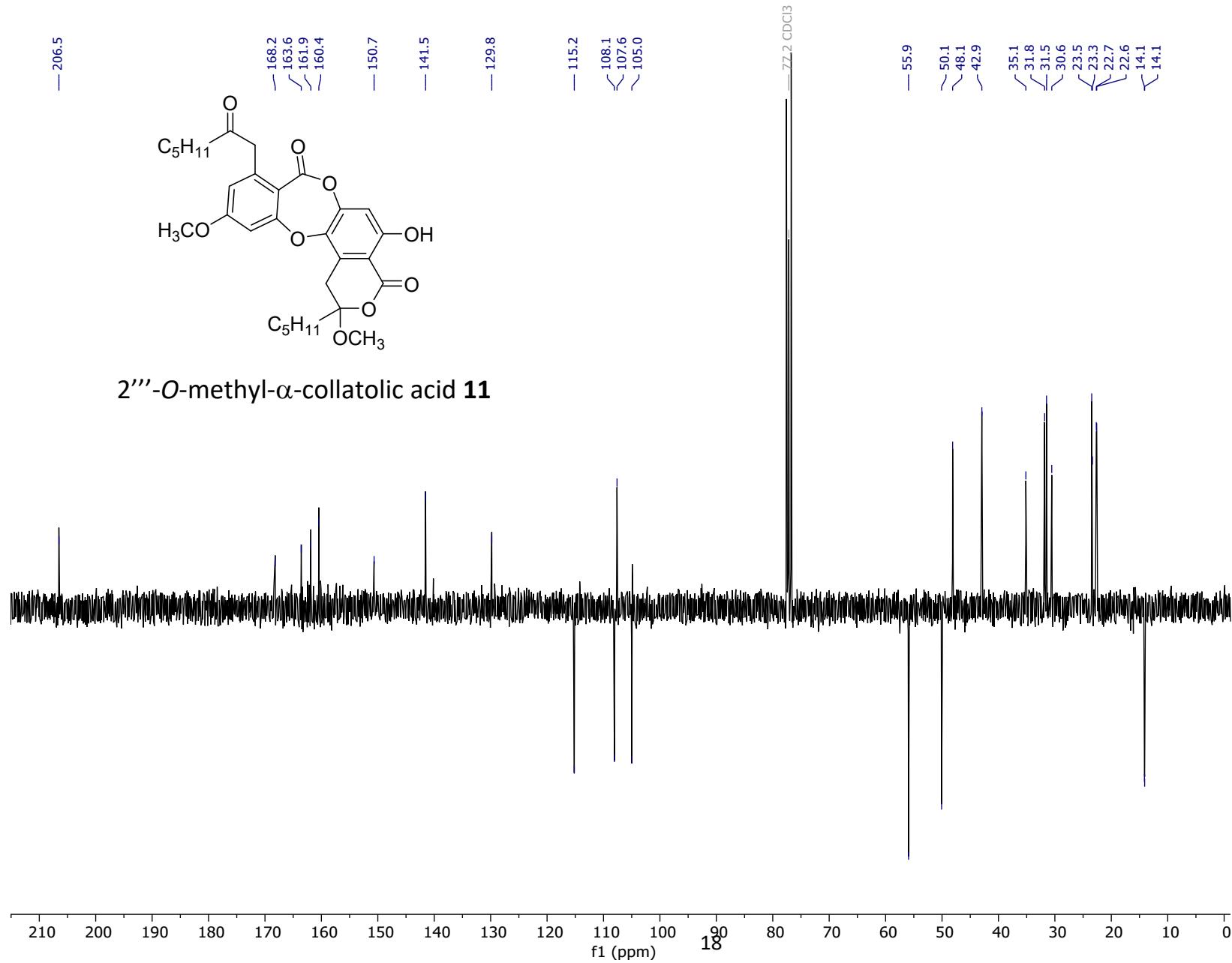


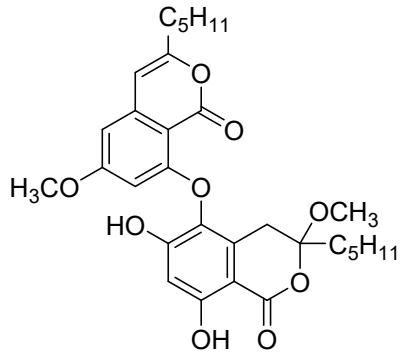




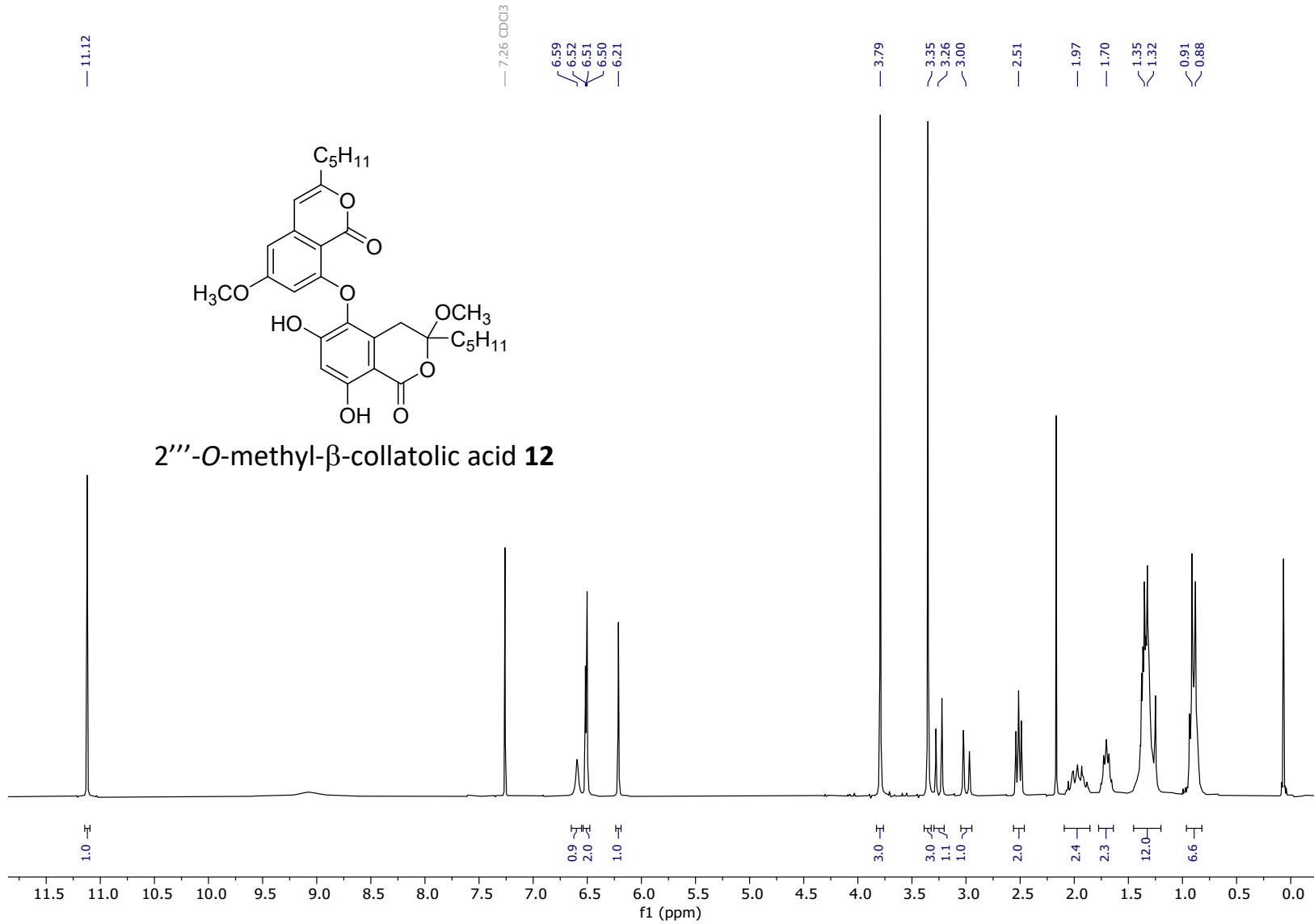








2'''-O-methyl- β -collatolic acid **12**



— 168.4
— 166.1
— 162.8
— 162.3
— 160.8
— 159.6
— 157.1

— 141.8
— 134.5
— 131.3

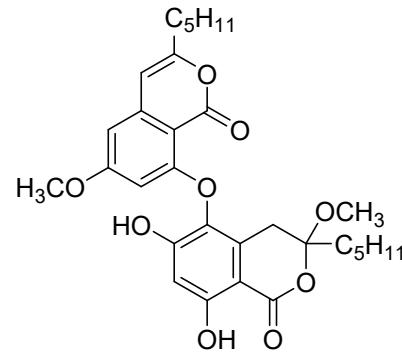
— 107.1
— 104.3
— 104.0
— 103.5
— 103.3
— 103.0
— 100.0

— 56.0

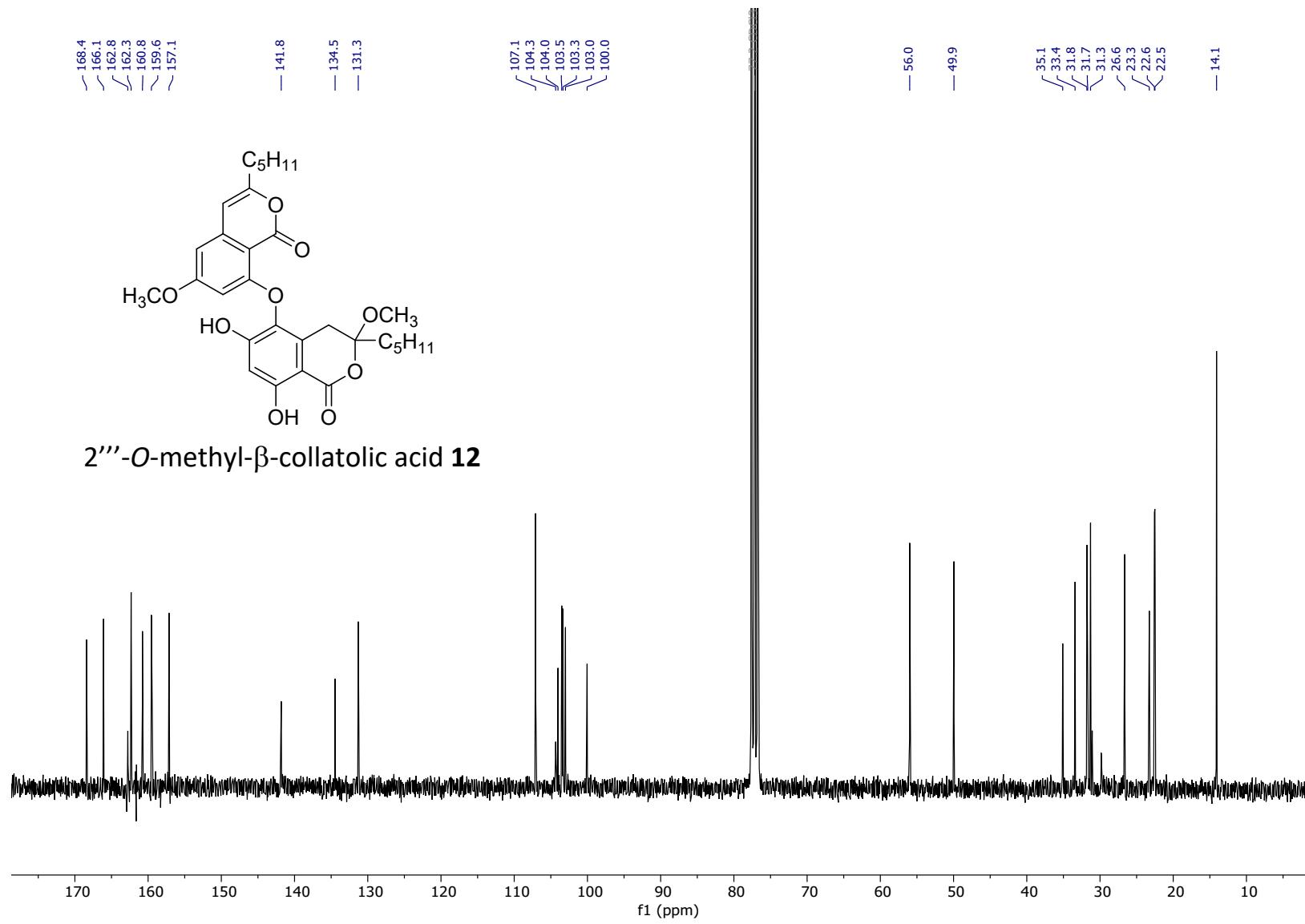
— 49.9

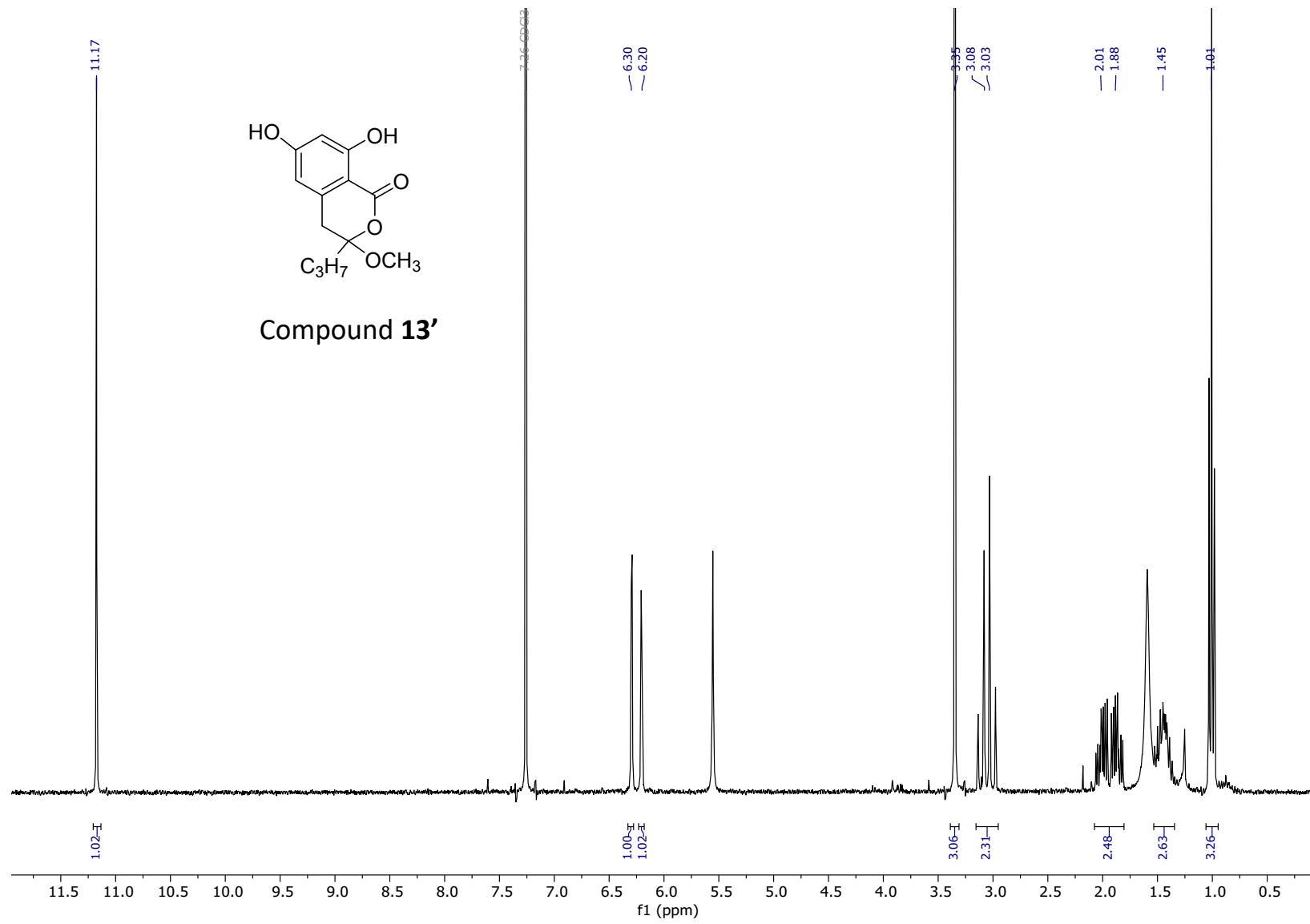
— 35.1
— 33.4
— 31.8
— 31.7
— 31.3
— 26.6
— 23.3
— 22.6
— 22.5

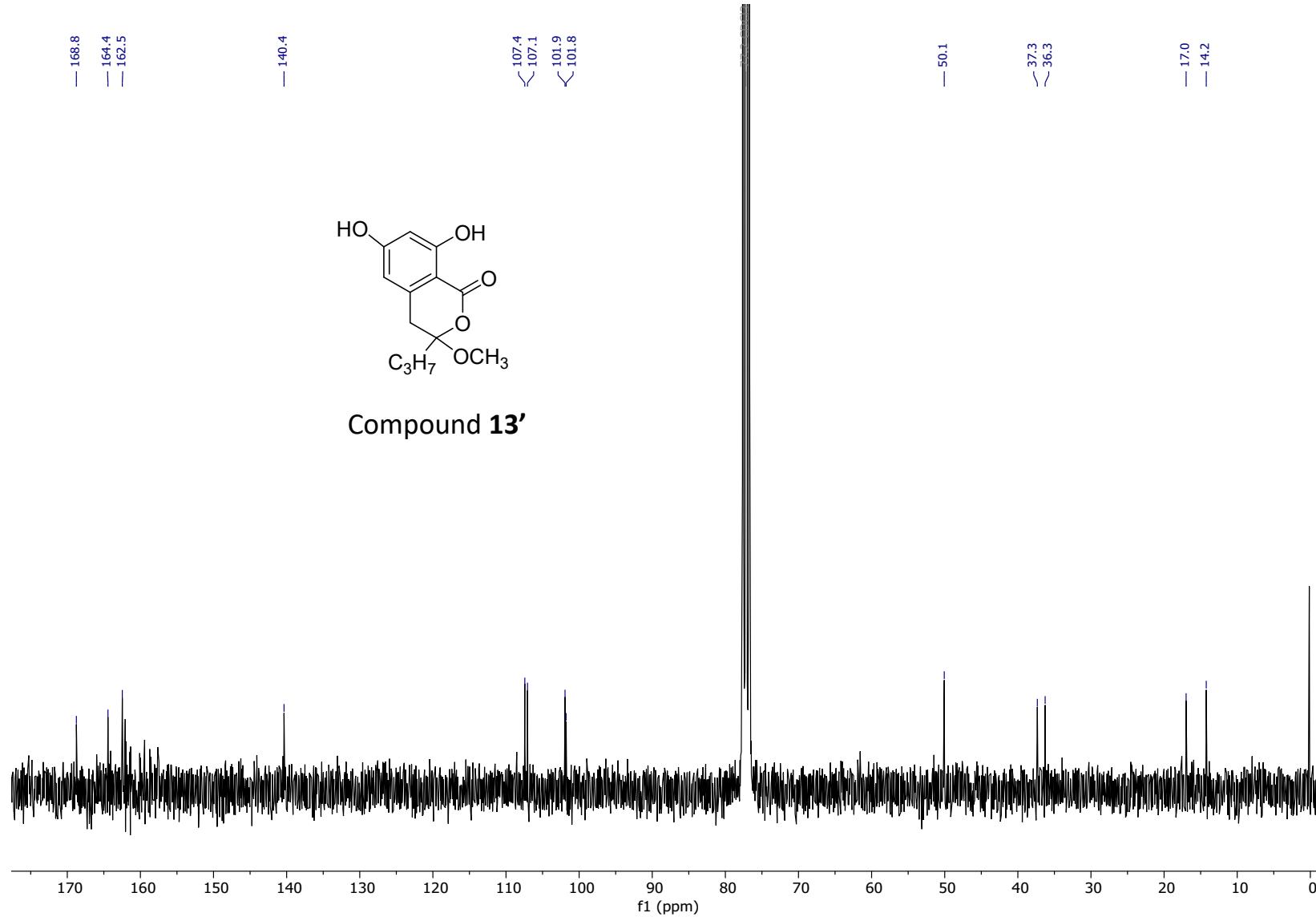
— 14.1



2'''-O-methyl- β -collatolic acid **12**

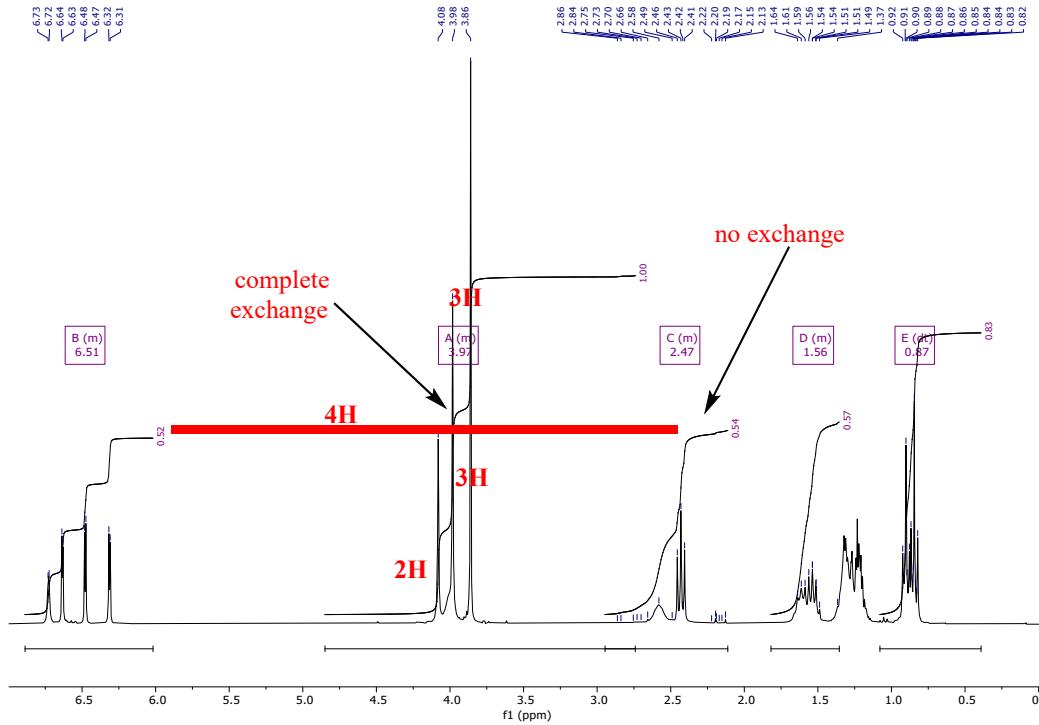






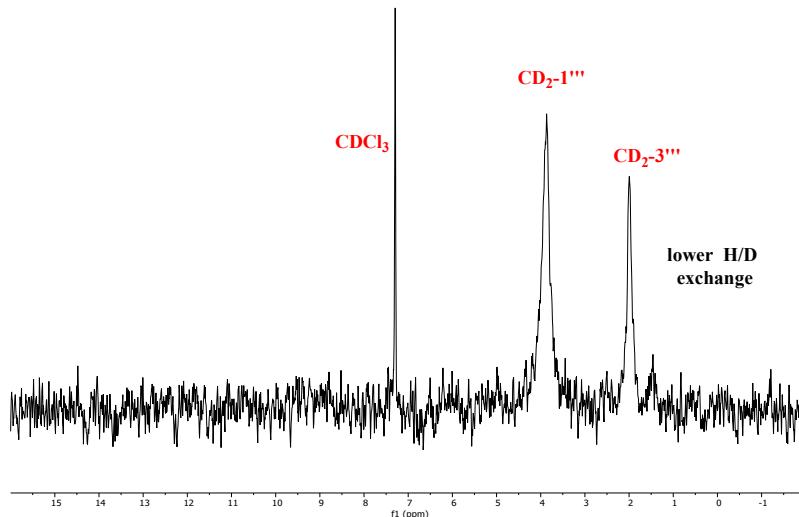
S4. Complementary ^1H and ^2H NMR for **2 and **6ka****

$2'$ -O-methylmicrophyllinic acid in CD_3COCD_3 (exchange H/D) after 30 min

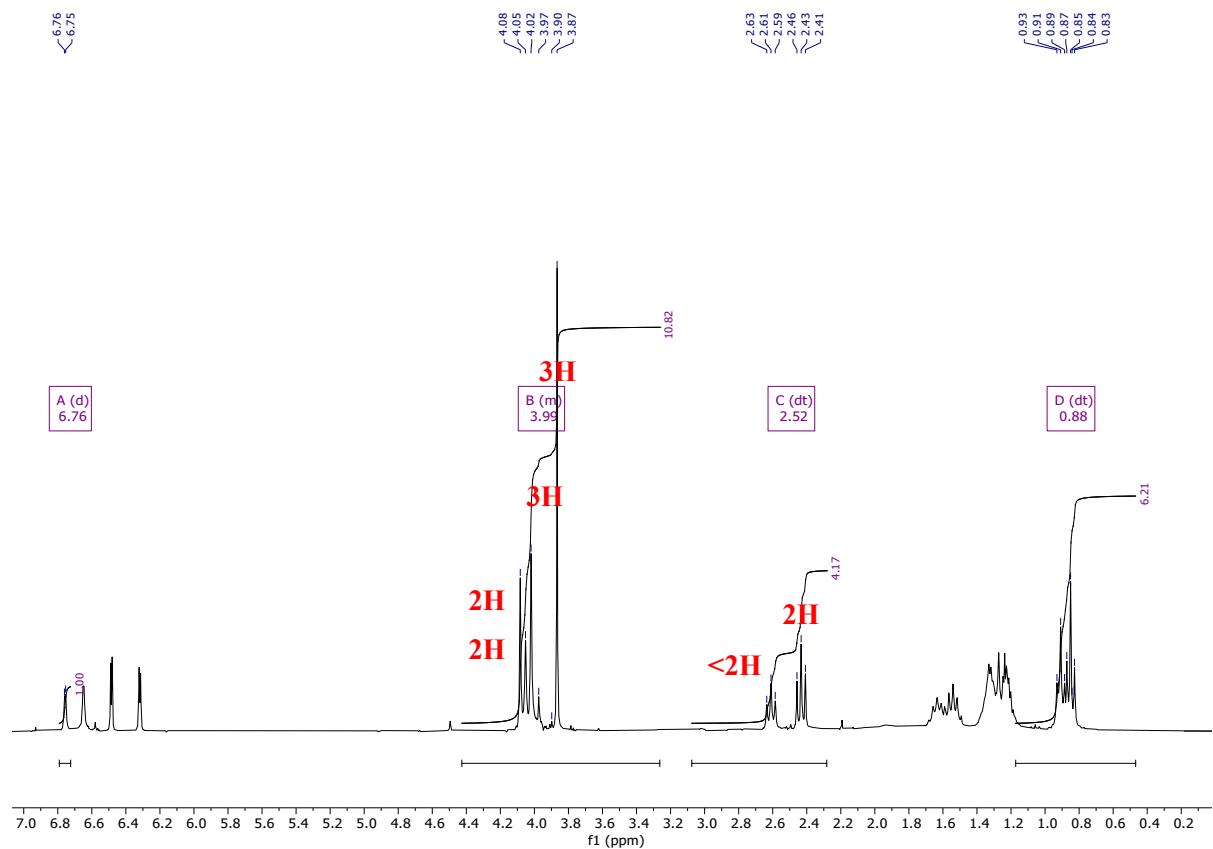


Exchange complete for CH_2 -1''' Lower exchange for CH_2 -3'''

^2H NMR confirmation in CDCl_3



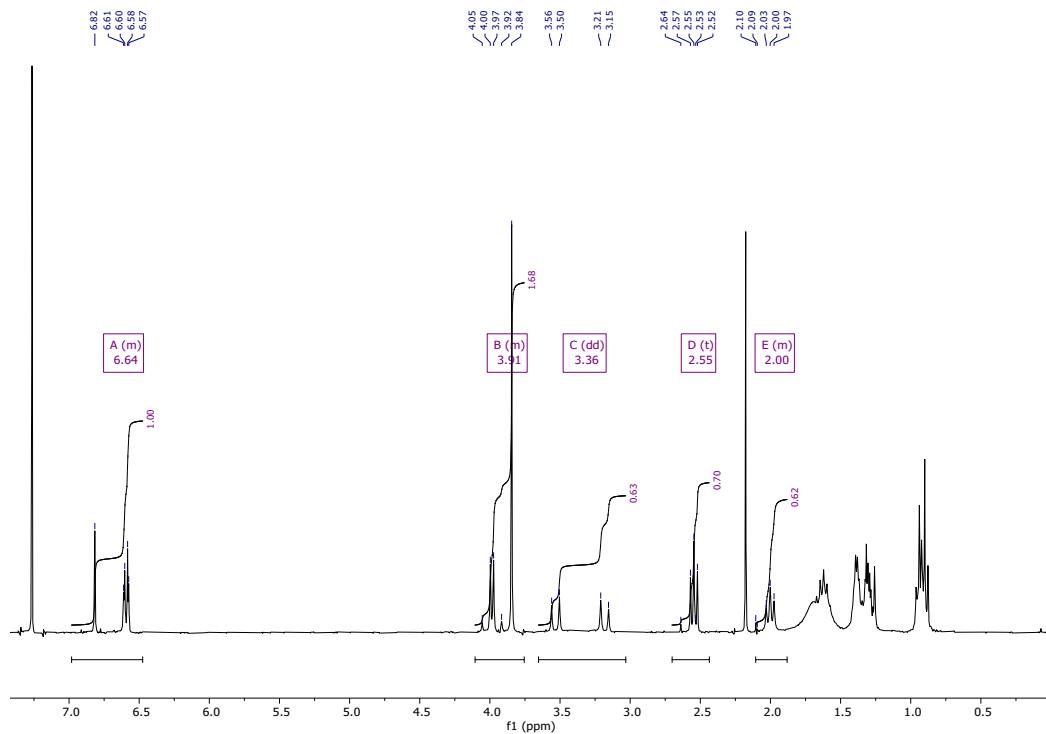
2'-O-methylmicrophyllinic acid in CDCl_3 after evaporation, dissolution in acetone
 (inverse exchange D/H) evaporation and NMR in CDCl_3



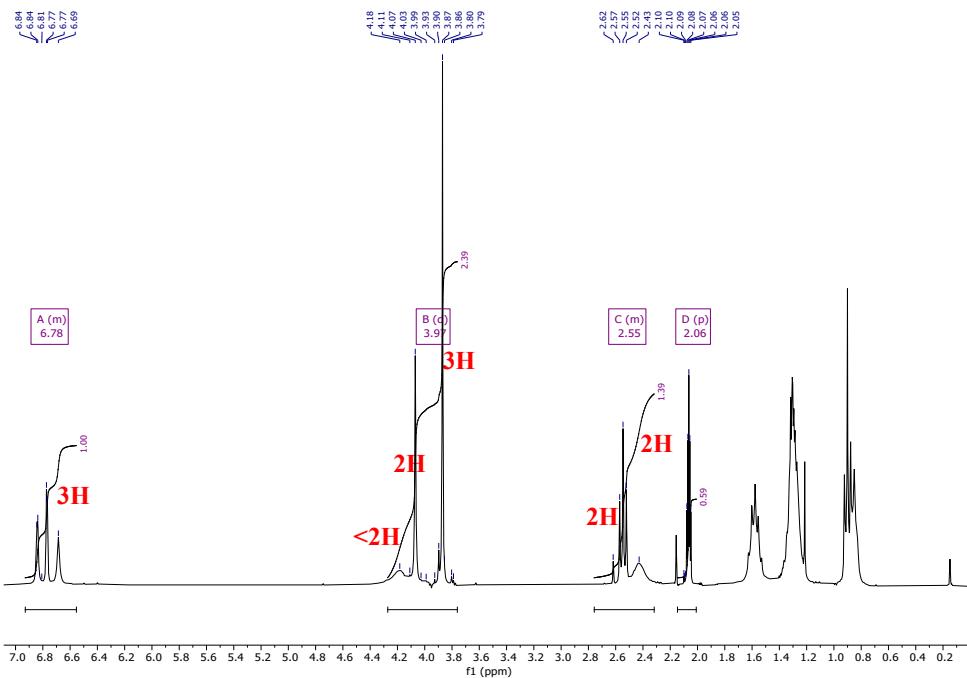
Inverse exchange: complete for $\text{CH}_2\text{-}1'''$ and lower for $\text{CH}_2\text{-}3'''$

The exchange is slow explaining the forms of the signal and the integration values.

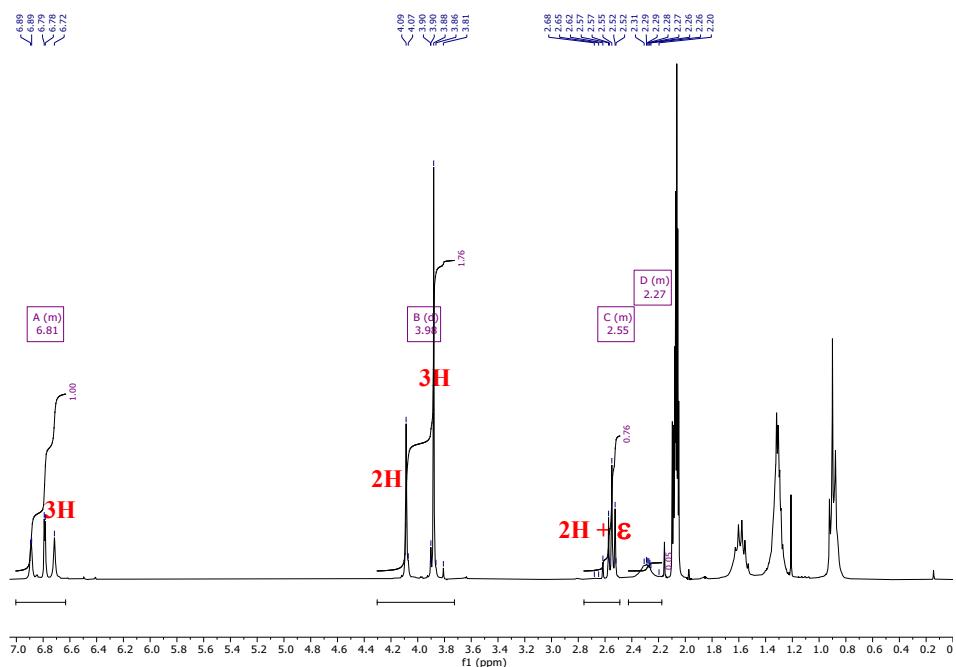
α -collatolic acid (CDCl_3)



α -isocollatolic acid (*d*6-acetone)

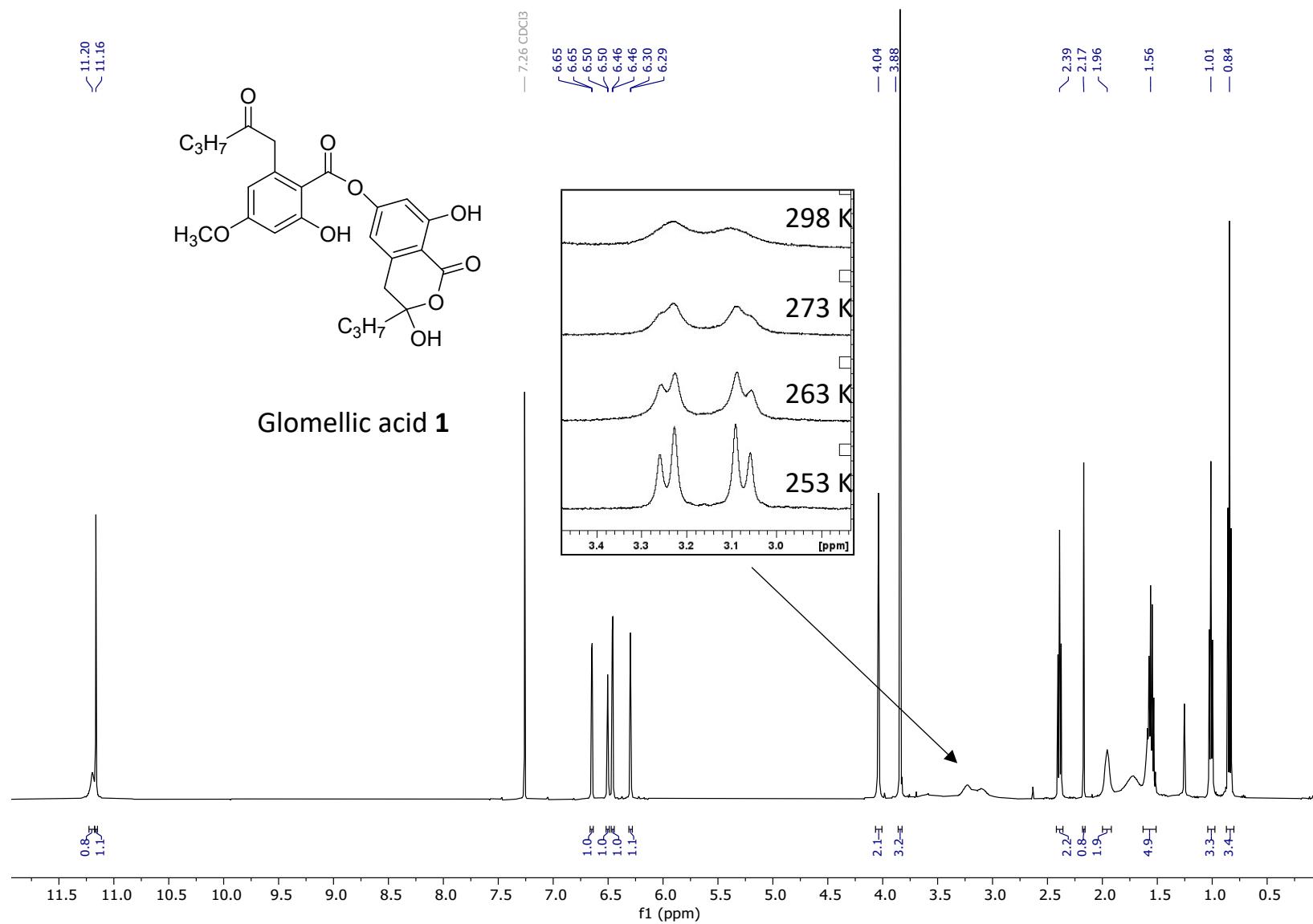


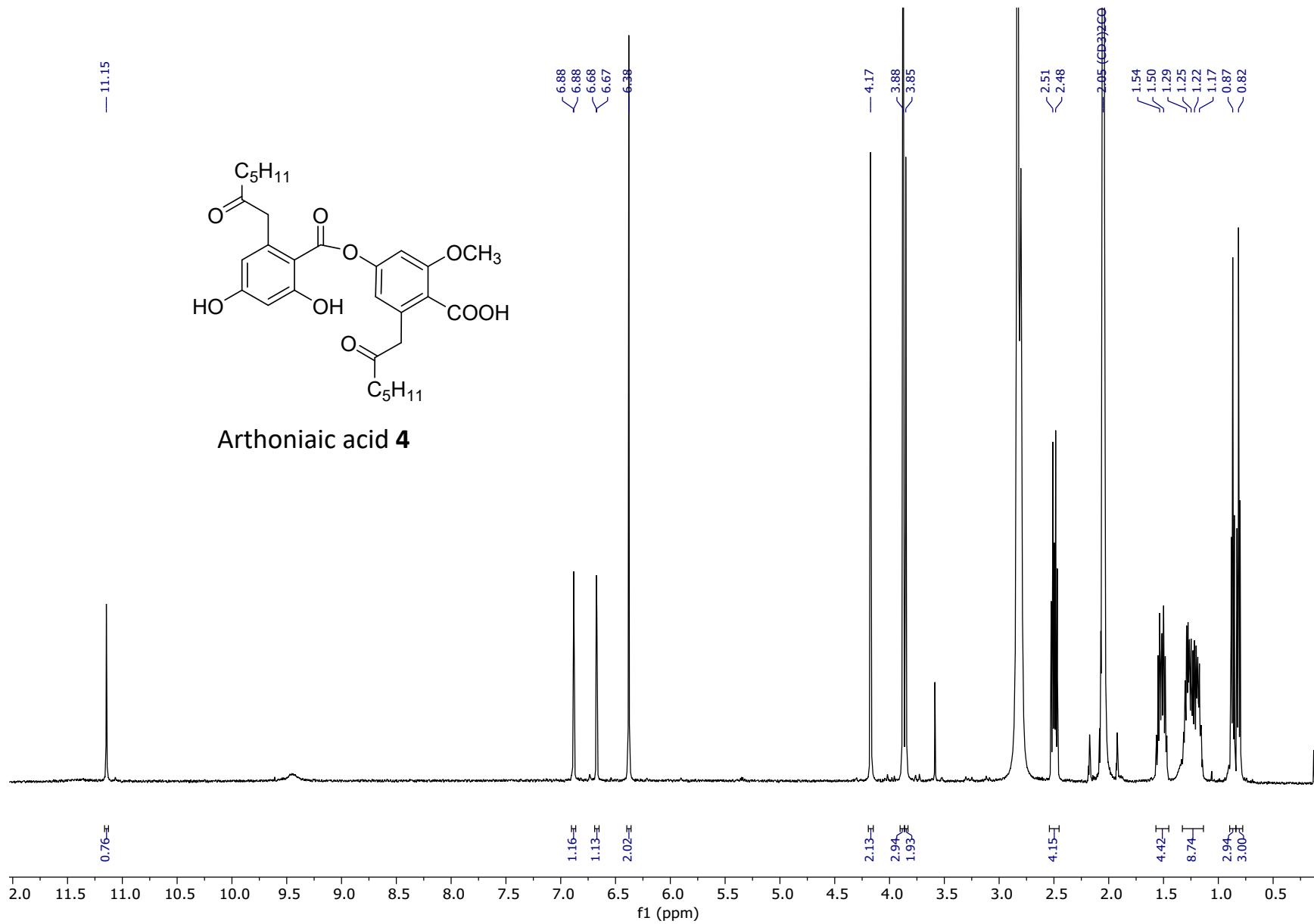
^1H NMR data just after dissolution: partial H/D exchange and coalescence.

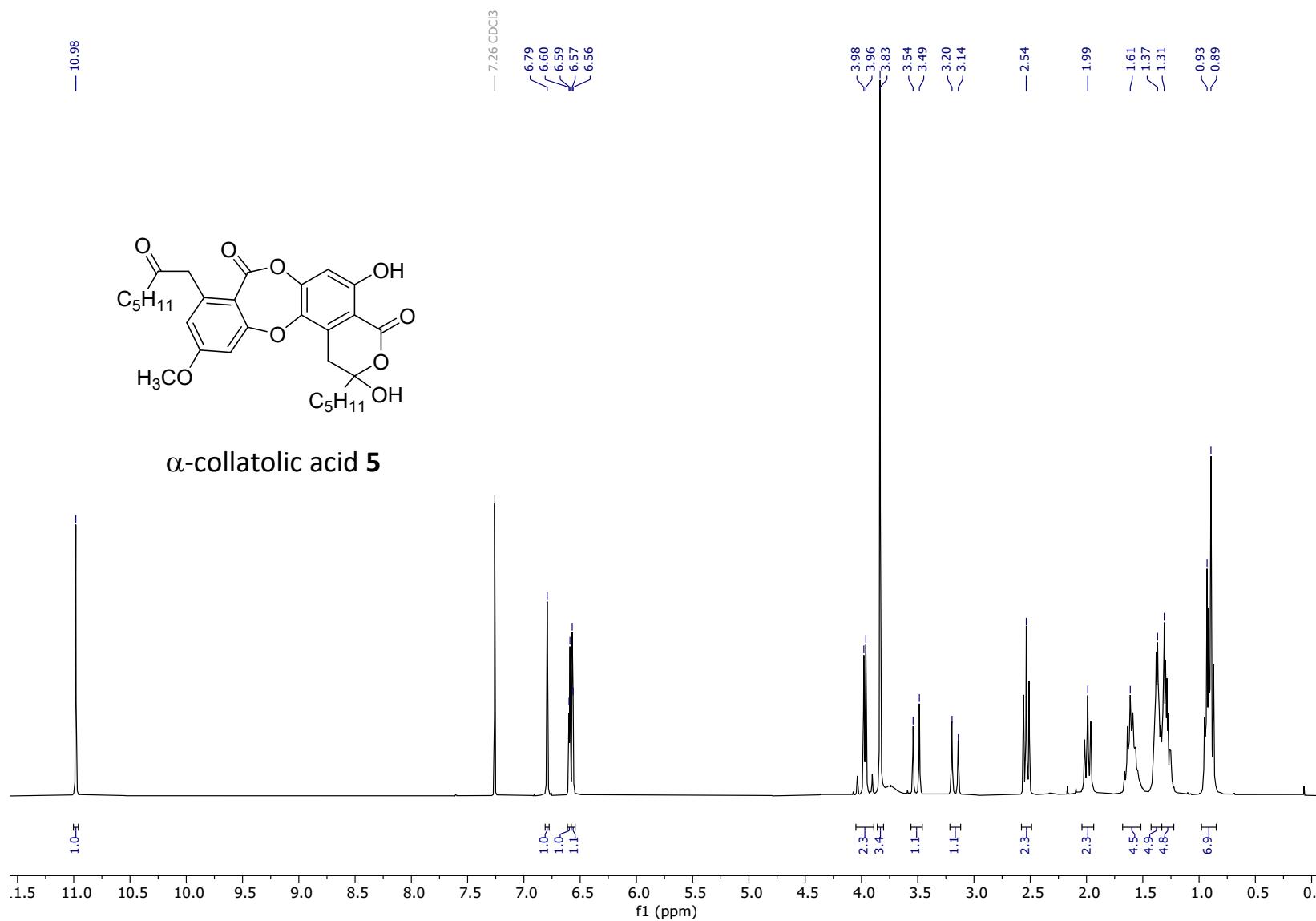


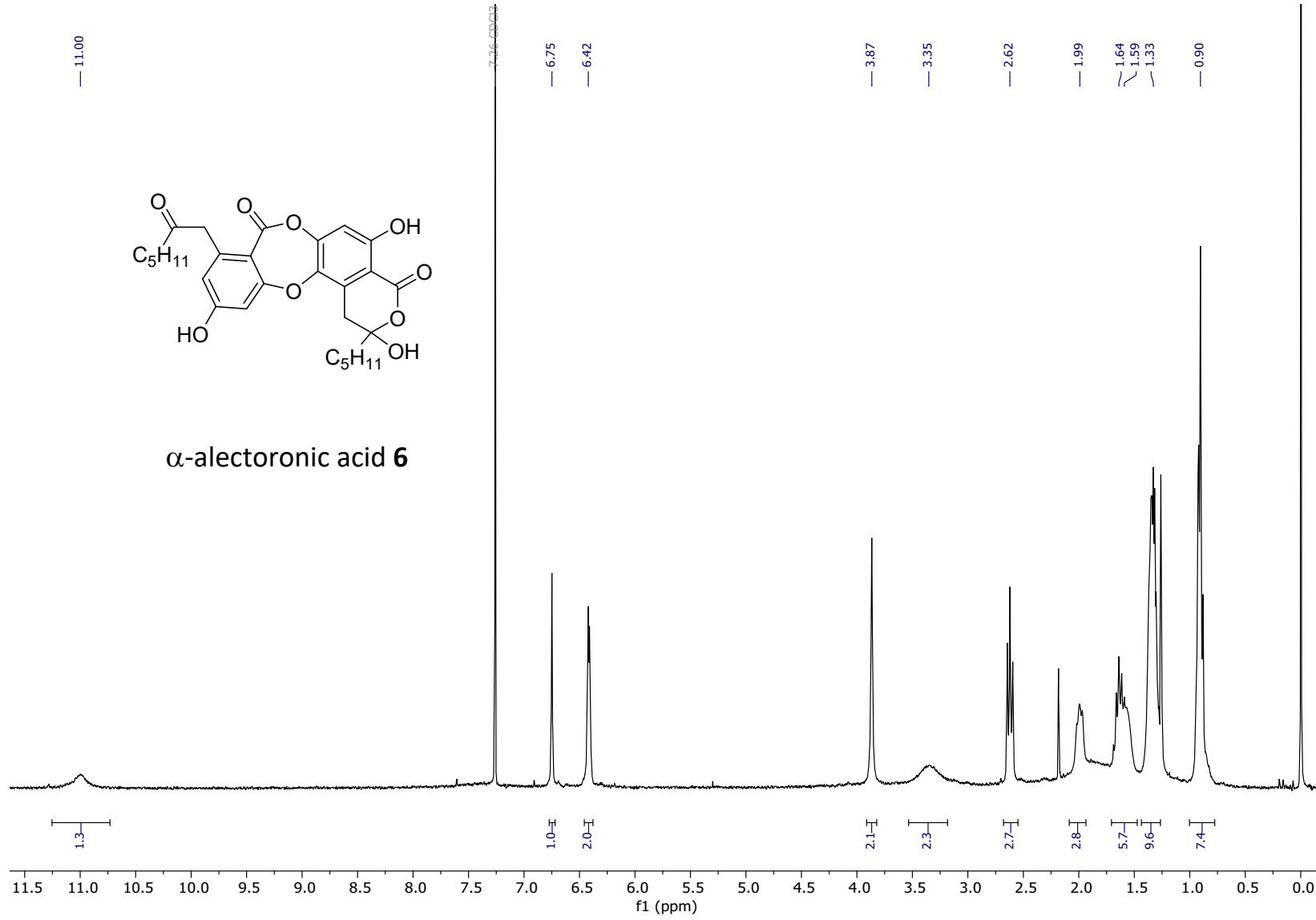
^1H NMR data after 3 days: complete H/D exchange.

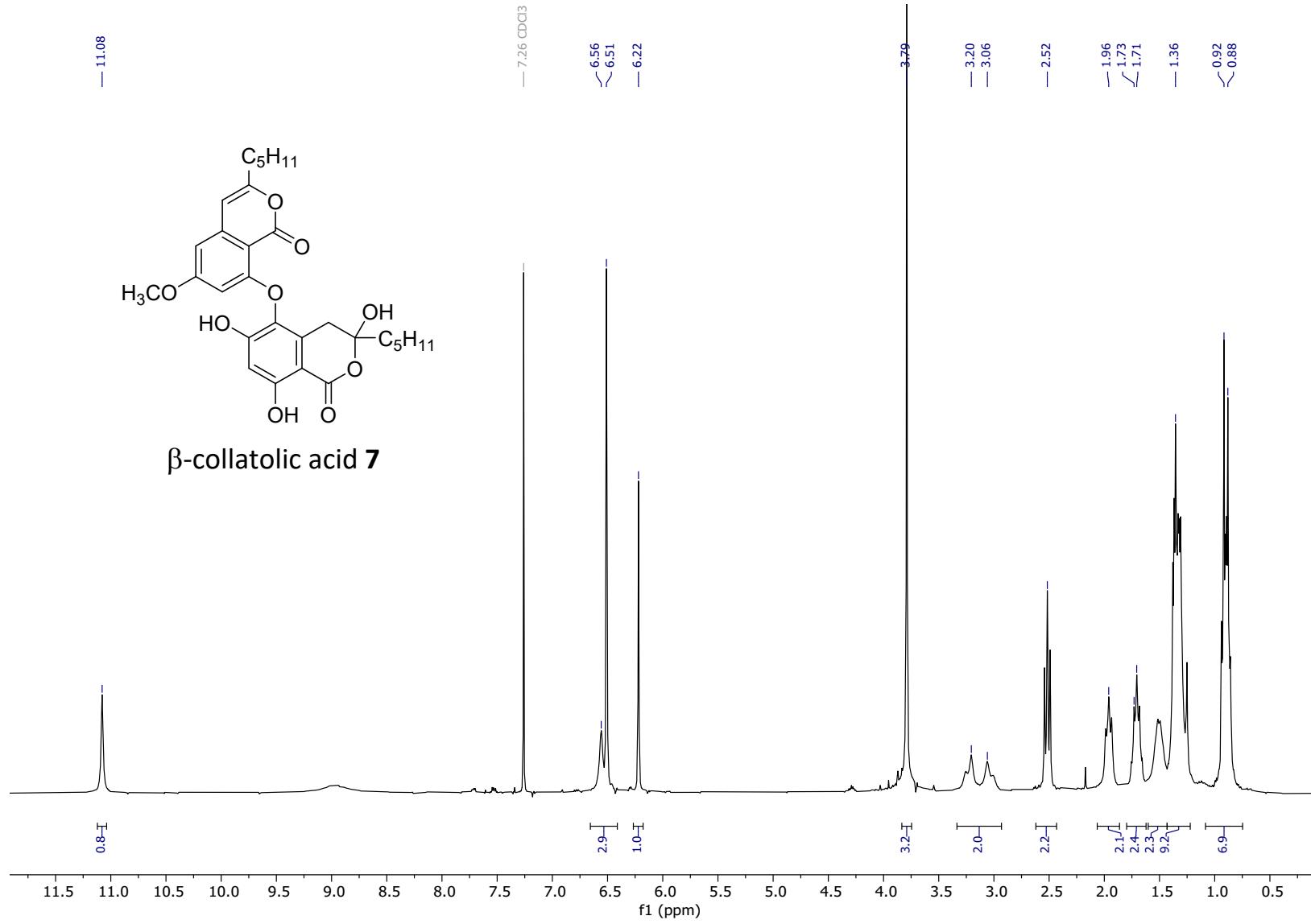
S5. NMR spectra of compounds previously reported^{1–3} : **1**, **4**, **5**, **6**, **7** and of physodic acid and physodone

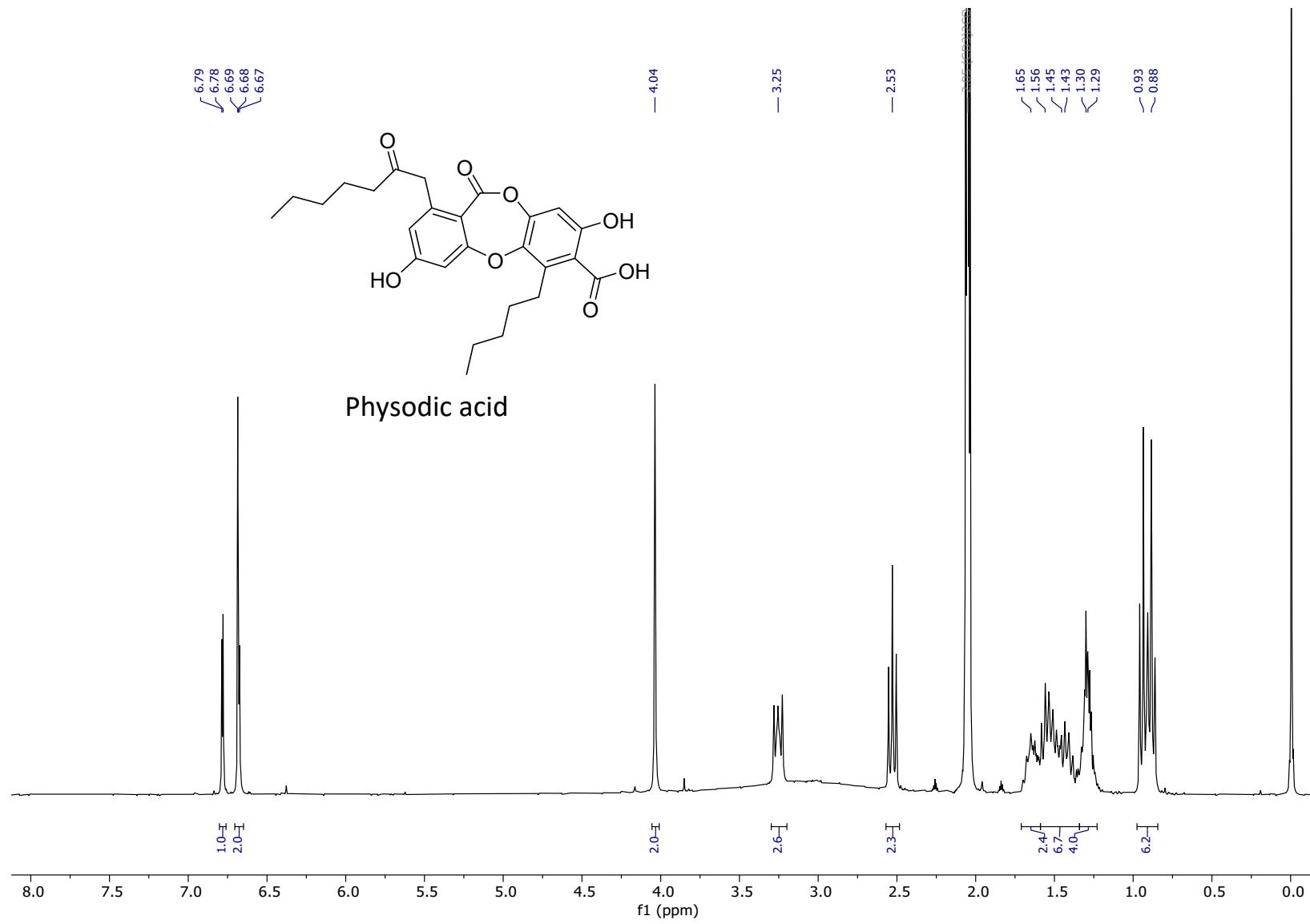


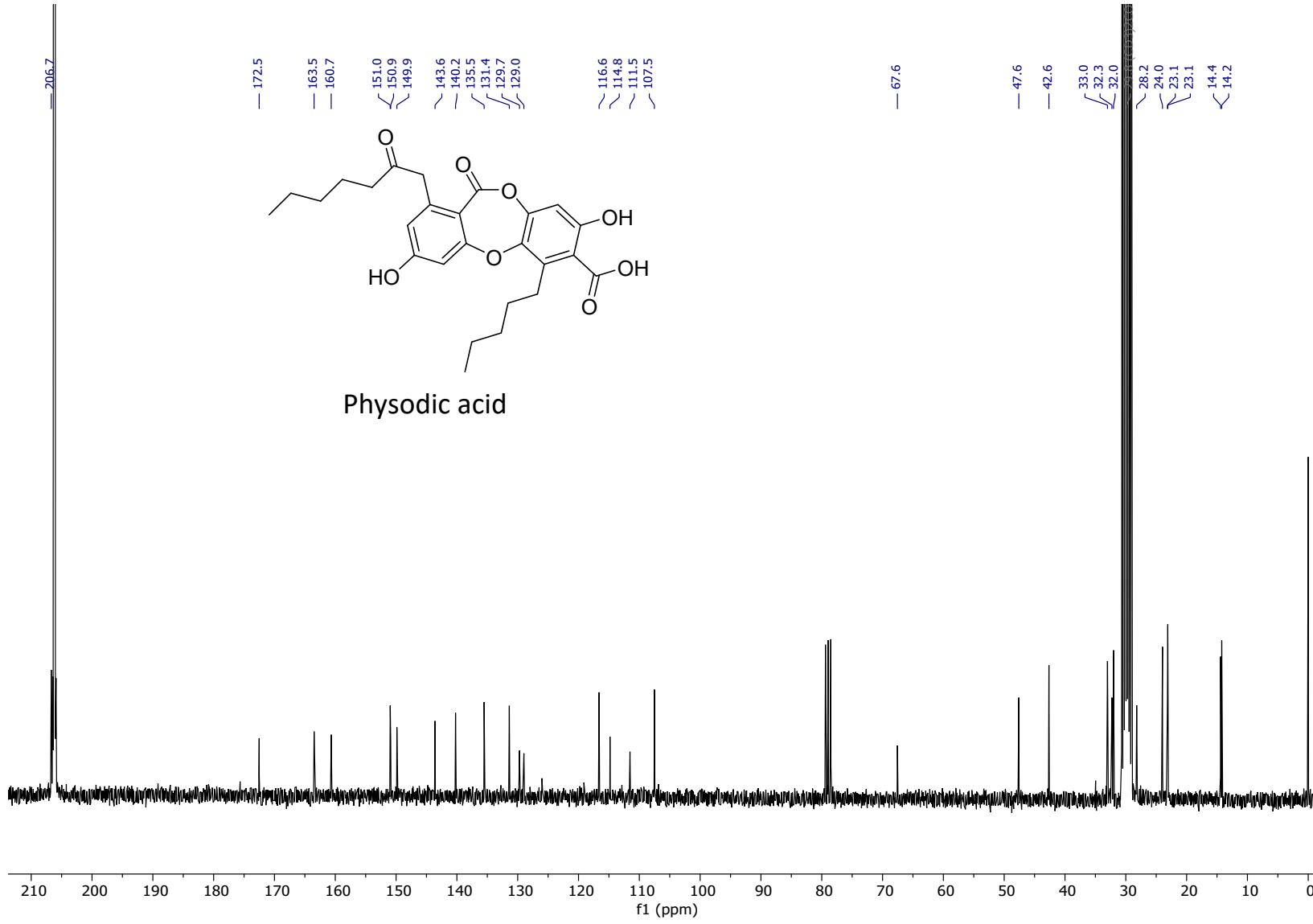


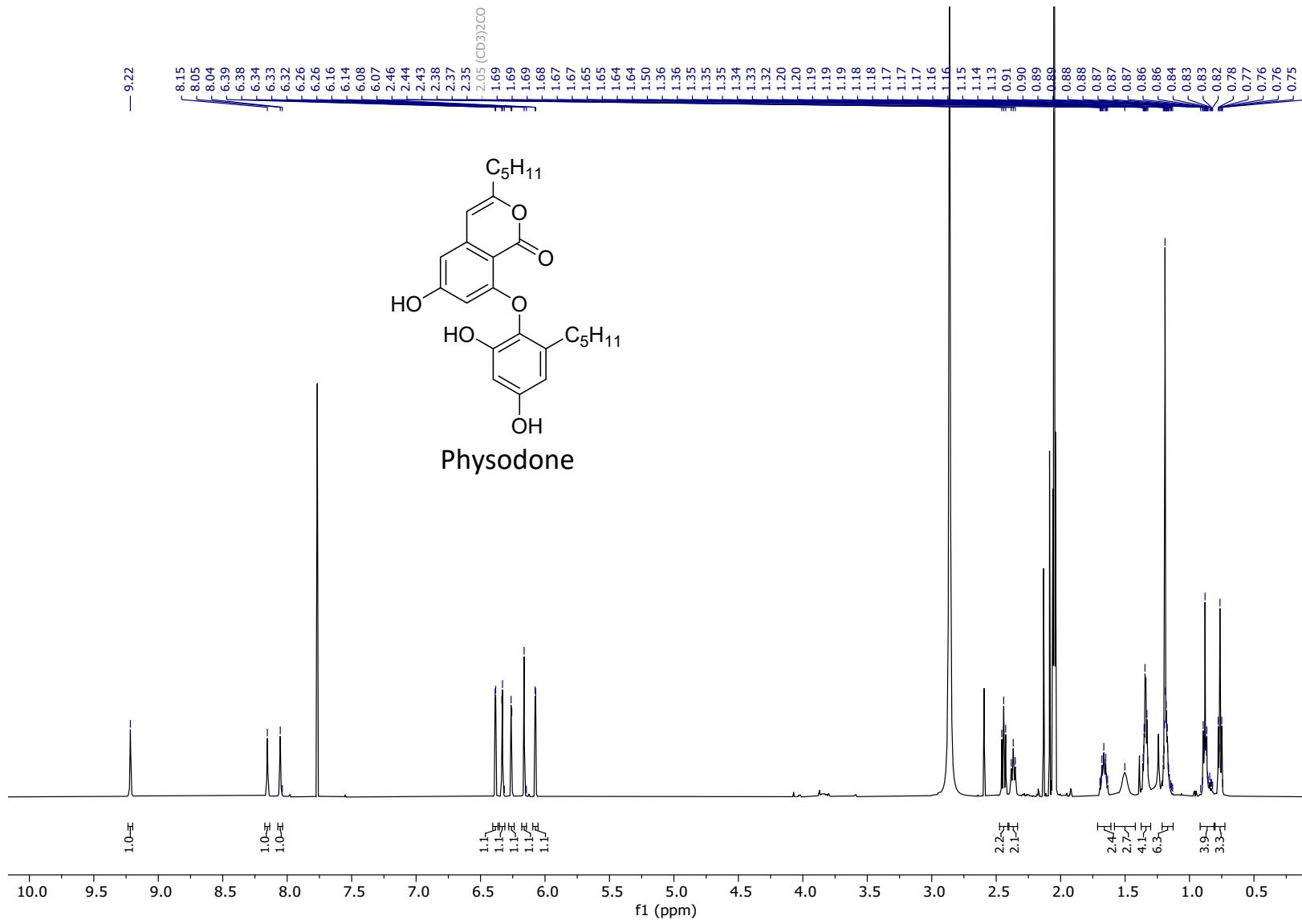


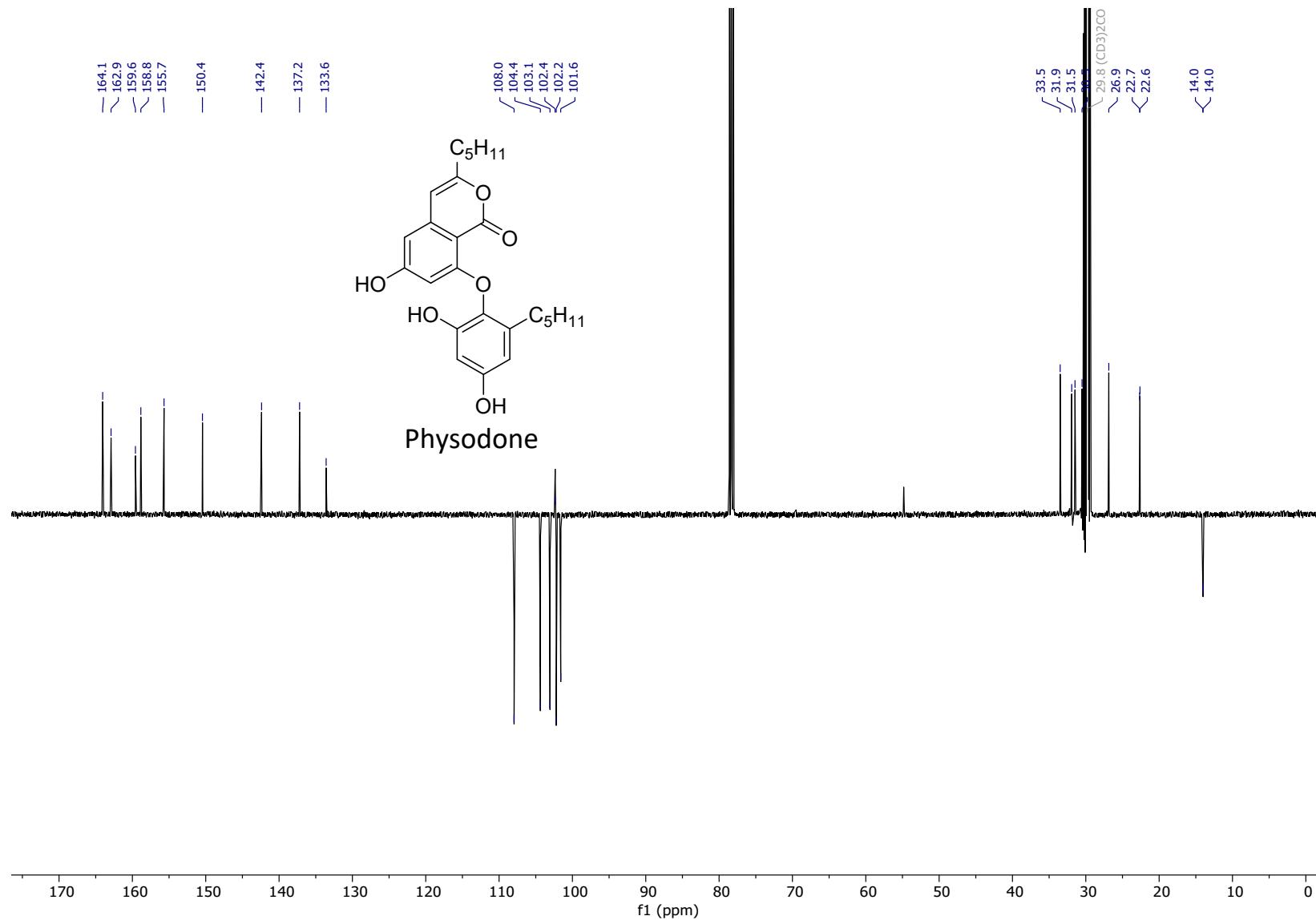






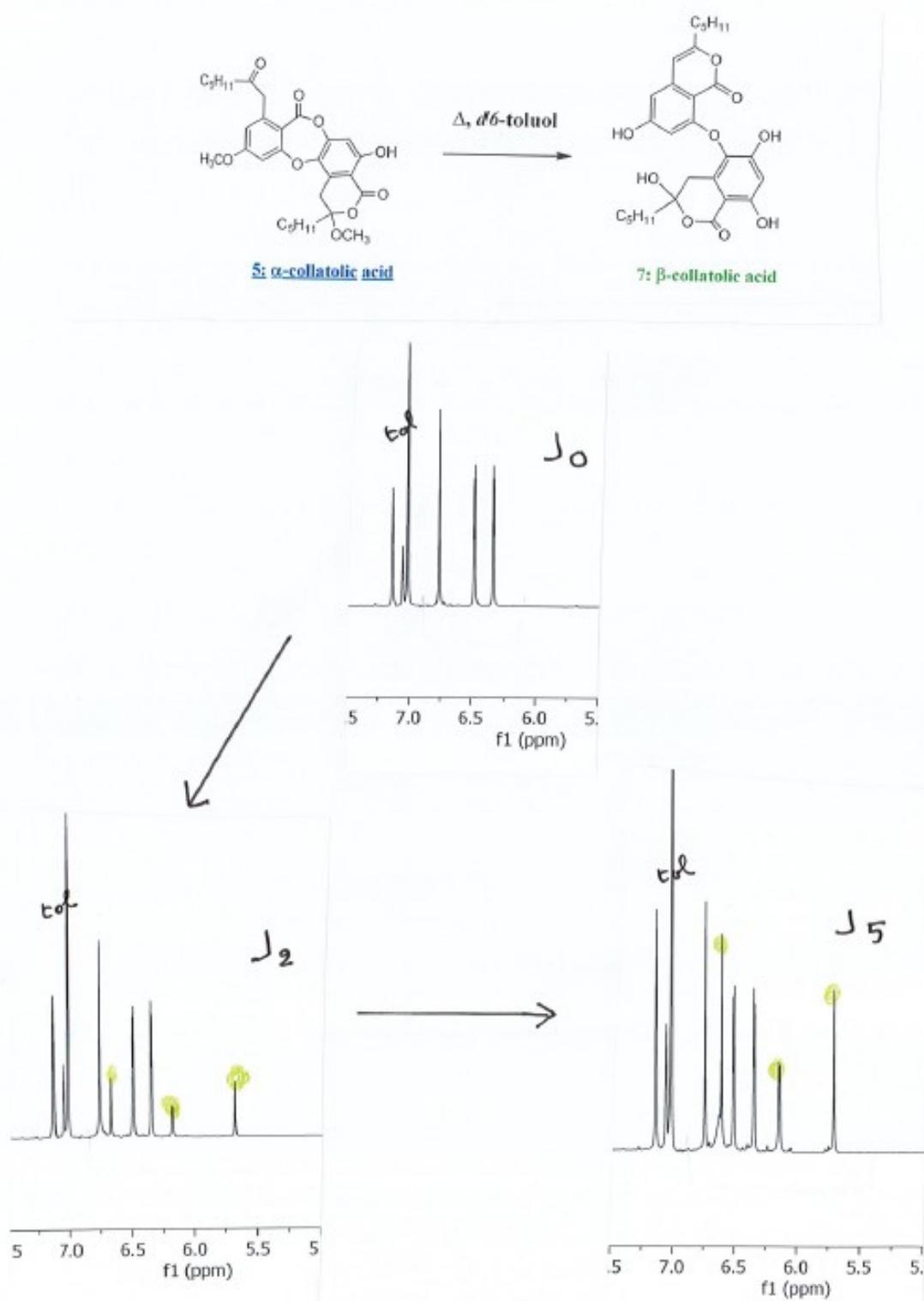




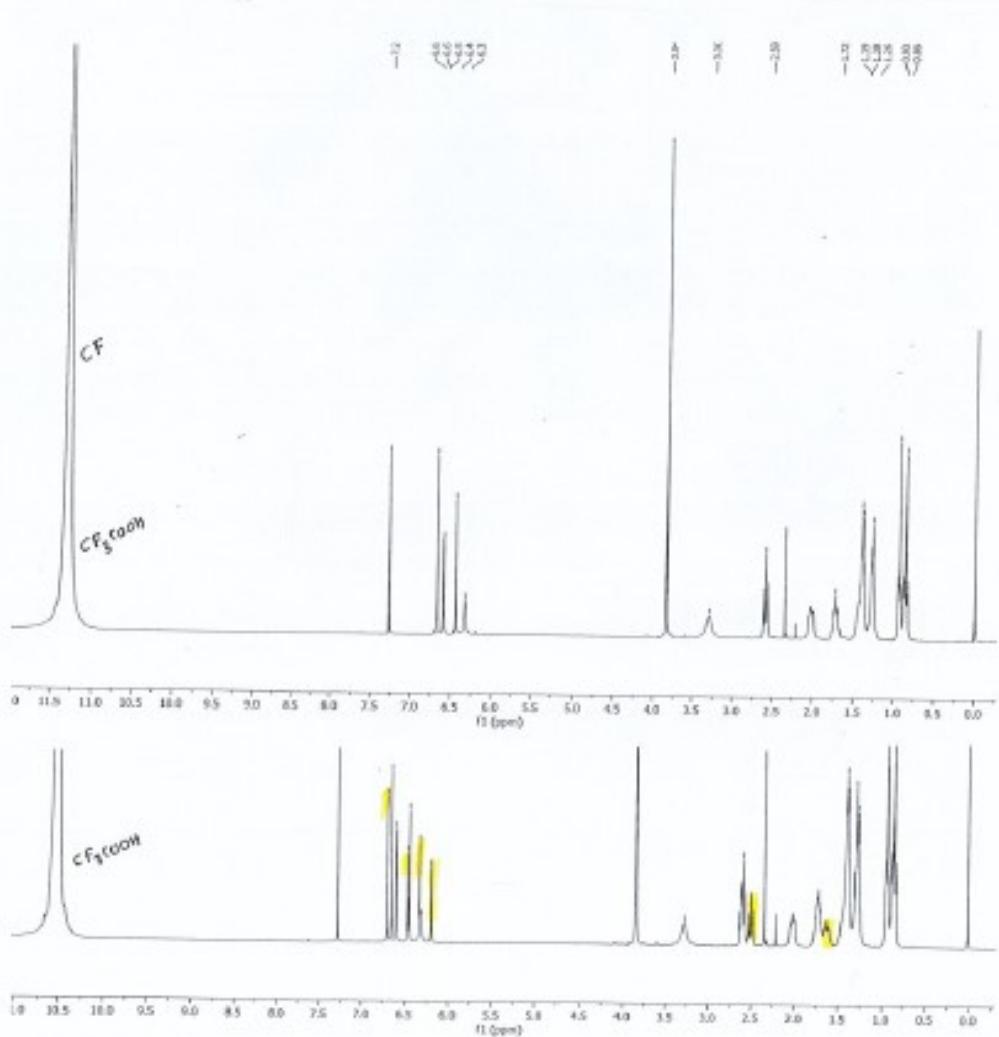
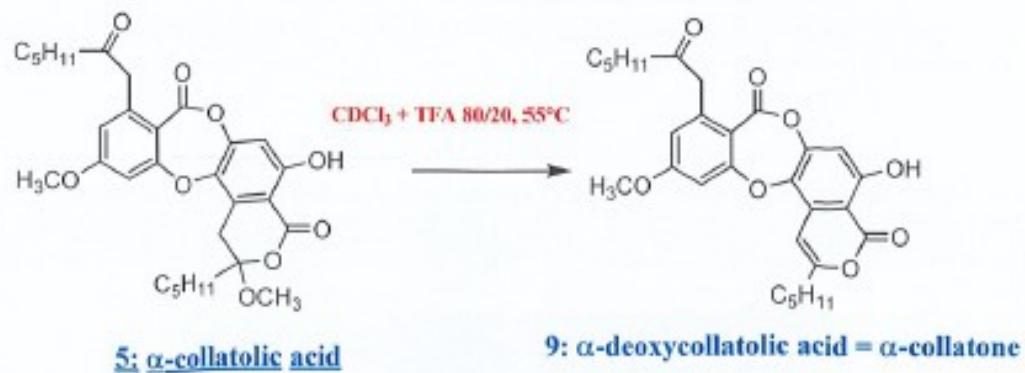


S6. Example of reaction checking by ^1H NMR and keto-enol study

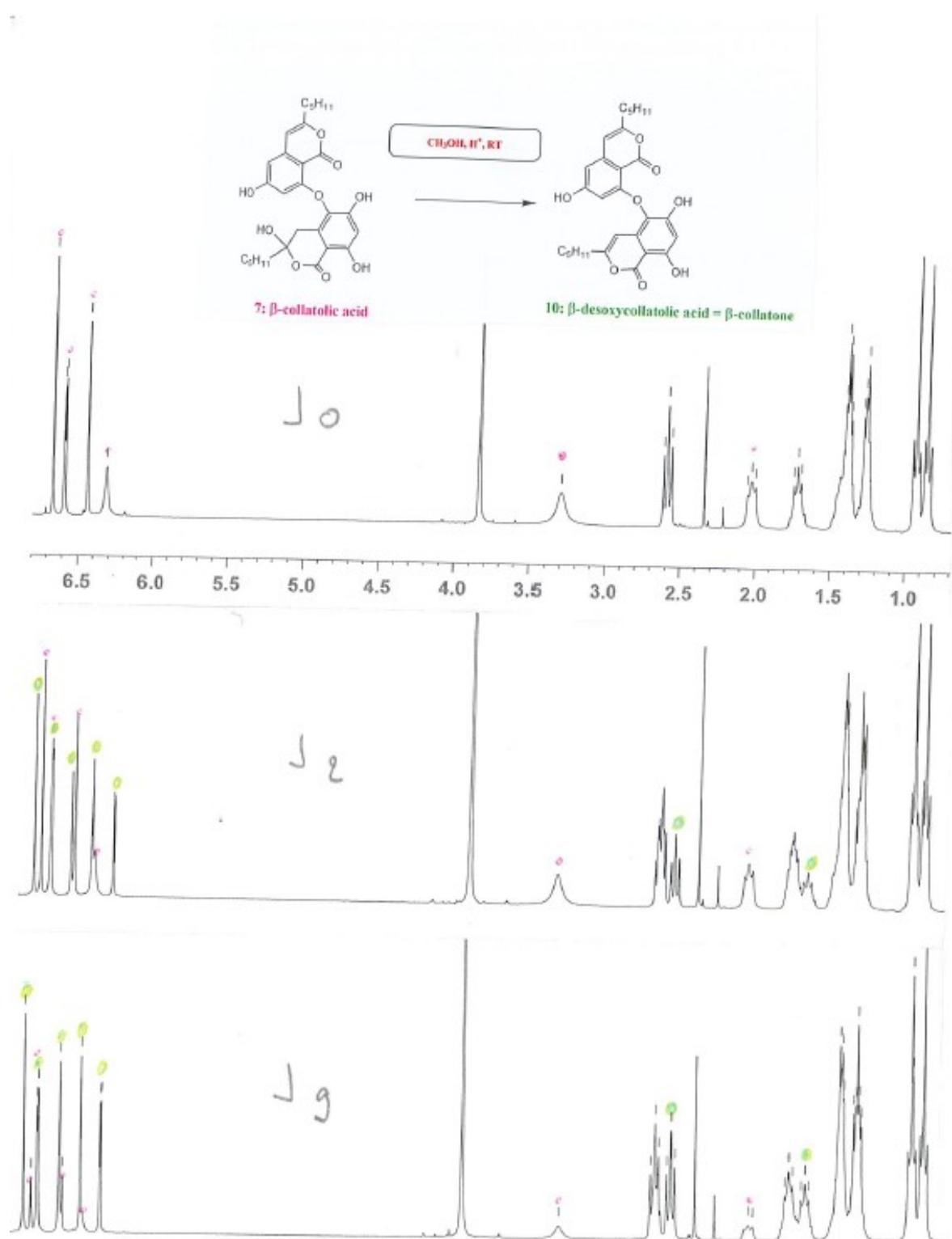
Isomerization of α -collatolic acid 5 into β -collatolic acid 7. Aromatic hydrogen atoms.



Dehydration of α -collatolic acid 5 into α -collatone 8.



Dehydration of β -collatolic acid 7 into β -collatone 9.



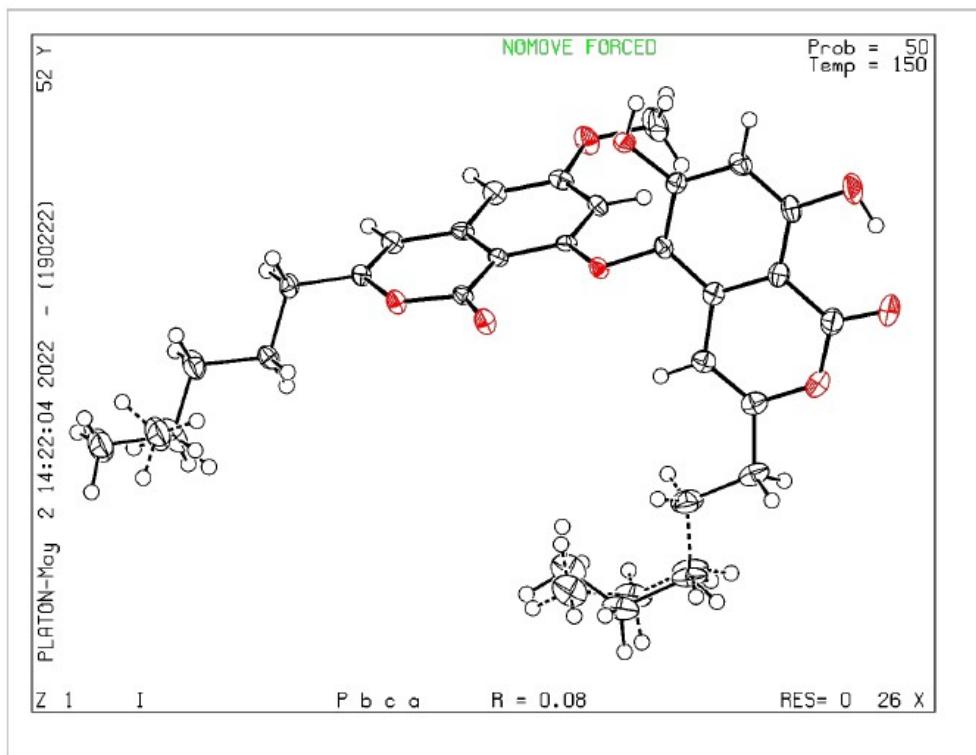
S7. Crystal structure report for β -collatone **9**

(C₂₉H₃₂O₈); M = 508.54.

A suitable crystal for X-ray diffraction single crystal experiment (colourless plate, dimensions = 0.280 x 0.230 x 0.060 mm) was selected and mounted on the goniometer head of a D8 Venture (Bruker-AXS) diffractometer equipped with a CMOS-PHOTON70 detector [*], using Mo-K α radiation (λ = 0.71073 Å, multilayer monochromator) at T = 150(2) K. Crystal structure has been described in orthorhombic symmetry and P b c a (I.T.#61) centric space group. Cell parameters have been refined as follows: a = 14.8340(12), b = 12.7431(9), c = 27.038(2) Å, V = 5110.9(7) Å³. Number of formula unit Z is equal to 8 and calculated density d and absorption coefficient μ values are 1.322 g.cm⁻¹ and 0.096 mm⁻¹ respectively. Crystal structure was solved by dual-space algorithm using SHELXT program [1], and then refined with full-matrix least-squares methods based on F2 (SHELXL program [2]). All non-Hydrogen atoms were refined with anisotropic atomic displacement parameters. Except Hydrogen atoms linked to Oxygen atoms that were introduced in the structural model through Fourier difference maps analysis, H atoms were finally included in their calculated positions and treated as riding on their parent atom with constrained thermal parameters. A final refinement on F2 with 5857 unique intensities and 355 parameters converged at $\omega R(F2)$ = 0.1704 (RF = 0.0840) for 5062 observed reflections with (I > 2 σ).

1. G. M. Sheldrick, Acta Cryst. A71 2015 3-8
2. Sheldrick G.M., Acta Cryst. C71 2015 3-8

Structure visualization



Data collection strategy details

Software : BIS V6.2.15/2021-03-15 && APEX4 2021.10-0
Number of scans : 3
Total number of frames [*] : 302
Total length of scans [*] : 151.00 (deg.)
Rotation speed [*] : ?
Total exposition time [*] : 1 h 40.7 min.

[*] fast scan not included

Scan	Time(s)	Width	DX(mm)	Frames	Range	2Theta	Omega	Phi	Chi	T(K)
1	Fast	4.0	1.00	34.0	180	180.0	0.0	0.0	0.0	54.7 150.0
2	Phi	20.0	0.50	34.0	302	151.0	0.1	270.3	176.5	24.0 150.0

Structural data

... Crystal data ...

Empirical formula	C ₂₉ H ₃₂ O ₈
Formula weight	508.54 g/mol
Temperature	150(2) K
Radiation type	Mo-K α
Wavelength	0.71073 Å
Crystal system, space group	orthorhombic, P b c a (I.T.#61)
Unit cell dimensions	a = 14.8340(12) Å b = 12.7431(9) Å c = 27.038(2) Å α = 90° β = 90° γ = 90°
Volume	5110.9(7) Å ³
Z, Calculated density	8, 1.322 g·cm ⁻³
Absorption coefficient	0.096 mm ⁻¹
F(000)	2160
Crystal size	0.280 x 0.230 x 0.060 mm
Crystal color	colourless
Crystal description	plate

... Data collection ...

Diffractometer	DB Venture (Bruker-AXS)
Detector	CMOS-PHOTON70
θ range for data collection	2.590 to 27.503°
(sin θ /λ) _{max} Å ⁻¹	0.650
h _{min} , h _{max}	-16, 19
k _{min} , k _{max}	-16, 16
l _{min} , l _{max}	-35, 34
Reflections collected / unique	27684 / 5857 [^a R(int) = 0.0529]
Reflections [I > 2σ]	5062
Completeness to θ_{max}	0.997
Absorption correction type	multi-scan
Max. and min. transmission	0.994, 0.852

... Refinement ...

Refinement method	Full-matrix least-squares on F ²
H-atom treatment	H-atom parameters treated by a mixture of independant and constrained refinement
Data / restraints / parameters	5857 / 2 / 355
^b Goodness-of-fit	1.275
Shelxl weighting scheme parameters	a = 0.0479, b = 6.3571
Final R indices [(I > 2σ)]	^c R ₁ = 0.0840, ^d wR ₂ = 0.1704
Final R indices [all data]	^c R ₁ = 0.0959, ^d wR ₂ = 0.1758
$\Delta\rho_{\text{max}}$, $\Delta\rho_{\text{min}}$	0.496, -0.341 e·Å ⁻³

$$^aR_{\text{int}} = \frac{\sum |F_o^2 - \langle F_o^2 \rangle|}{\sum F_o^2}$$

$$^bS = \left\{ \frac{\sum |w(F_o^2 - F_c^2)|^2}{n-p} \right\}^{1/2}$$

$$^cR_1 = \frac{\sum |F_o| - |F_c|}{\sum |F_o|}$$

$$^d wR_2 = \left\{ \frac{\sum |w(F_o^2 - F_c^2)|^2}{\sum |w(F_o^2)|^2} \right\}^{1/2}$$

$$w = 1. / [\sigma(F_o^2) + (aP)^2 + bP] \text{ with } P = [2F_c^2 + \text{Max}(F_o^2, 0)] / 3$$