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Supporting Information

Synthesis of Photodynamic Intramolecularly Crosslinked

Polyether-Copolymers In-Chain Functionalized with

Hexaarylbiimidazoles

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Experimental Section

Materials

All chemicals were used as received unless otherwise noted.

Synthesis of 2_{OTs}.

Into a flask, THF solution (10 mL) of $\mathbf{2}_{OH}$ (4.2 g, 2.10 mmol, 16.8 mmol for OH groups) and aqueous solution (7 mL) of NaOH (6.72 g, 168 mmol) were added. The flask was cooled in an ice bath and a THF solution (15 mL) of TsCl (6.41 g, 33.6 mmol) was slowly added. The mixture was stirred for 19.5 h at room temperature, and then water (ca. 50 mL) was added. After 1 h, the reaction mixture was poured into H₂O/DCM (1/1, v/v), and the DCM phase was washed with H₂O. The organic phase was dried over Na₂SO₄ and concentrated to dryness to give $\mathbf{2}_{OTs}$ as a light yellow solid. The yield was 6.09 g (91%).

Synthesis of 2_{Az}.

Into a flask was added $\mathbf{2}_{OTs}$ (2.0 g, 0.625 mmol, 5.00 mmol for OTs groups), NaN₃ (975 g, 15.0 mmol), and DMF (18 mL). The mixture was stirred at 85 °C for 5 h. After cooling to room temperature, the mixture was poured into an excess amount of H₂O/DCM (1/1, v/v). The organic phase was washed with H₂O, dried over Na₂SO₄, and concentrated to dryness to give light yellowish oily $\mathbf{2}_{Az}$. The yield was 1.36 g (99%).

Synthesis of 2_L.

Into a flask was added $\mathbf{2}_{Az}$ (1.2 g, 0.545 mmol, 4.60 mmol for N₃ groups), \mathbf{L}_E (2.29 g, 6.55 mmol), CuBr (313 mg, 2.18 mmol), PMDETA (456 μ L, 2.18 mmol) and DMF (5

mL). The mixture was stirred at 25 °C for 3 h. After cooling to room temperature, the mixture was diluted with CHCl₃ and passed through a plug of alumina. The eluent was concentrated and precipitated in diethyl ether. The precipitates were dissolved in CHCl₃ and concentrated to dryness to give 2_L as a light yellowish green solid. The yield was 2.6 g (95%). Number- and weight-averaged molecular weights (M_n and M_w), peak molecular weight (M_p), and dispersity ($D = M_w/M_n$) measured by SEC calibrated with polystyrene standards; $M_n = 3900$, $M_w = 4200$, $M_p = 4200$, and D = 1.09.

Synthesis of 2_H.

A DCM solution (250 mL) of $\mathbf{2}_{L}$ (860 mg, 0.172 mmol, 1.38 mmol for lophine groups) was prepared in a flask. To this mixture, an aqueous solution (100 mL) of potassium hydroxide (1.42 g, 27.5 mmol) and potassium ferricyanide (9.06 g, 27.5 mmol) was added, and the resulting mixture was vigorously stirred in the dark. After 3 h, the DCM phase was washed with H_2O , dried over Na_2SO_4 , and concentrated to dryness to give $\mathbf{2}_{H}$ as a green solid. The yield was 856 mg (>99%). M_n , M_w , M_p , and D measured by SEC calibrated with polystyrene standards; $M_n = 4300$, $M_w = 5200$, $M_p = 4000$, and D = 1.22.

NMR measurements.

¹H NMR spectra were recorded on a Bruker AVANCE III spectrometer operating at 500 MHz. CDCl₃ was used as the solvent, and chemical shifts were reported relative to tetramethylsilane (TMS) ($\delta = 0.00$ ppm) or solvent residual signals.

FT-IR measurements.

Fourier transform infrared (FT-IR) spectra were recorded on a Bruker Alpha II FT-IR spectrometer in the range of 4000–400 cm⁻¹ with an attenuated total reflection (ATR)

module.

SEC measurements.

SEC measurements were performed using a Waters e-2695 high-speed liquid chromatograph equipped with RI and UV detectors. A Shodex KF-603 column (flow rate: 0.50 mL/min) was employed with THF as the eluent at 40 °C.

Photoirradiation experiments.

Photoirradiation tests were carried out using an Asahi Spectra MAX-350 xenon lamp equipped with an Asahi Spectra quartz light guide, and an Asahi Spectra LX0365 bandpass filter was used for transmitting only the wavelength of 365 nm. For demonstration, a household blue laser pointer ($\lambda \sim 405$ nm) was used.

ESR measurements.

ESR spectra were recorded on a JEOL JES-TE3000 spectrometer at room temperature. For photoirradiation experiments, photoirradiation was performed from a window equipped with a sample insertion opening.

Scheme S1 Synthetic route for NPC $(\mathbf{2}_H)$ from the starting PGL-based polyether-copolymer $(\mathbf{2}_{OH})$.

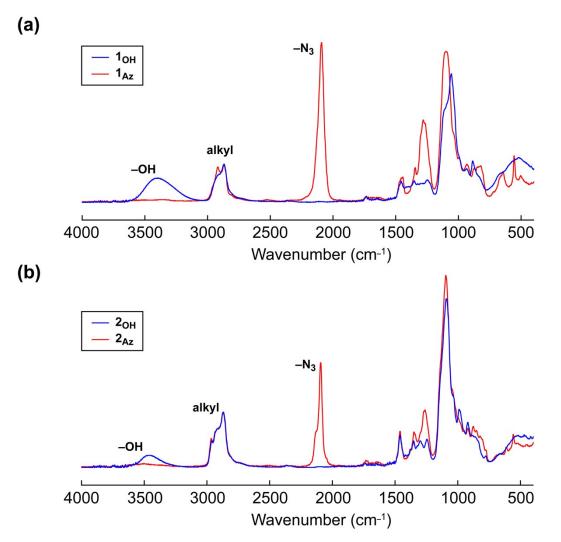


Figure S1 FT-IR spectra of (a) $\mathbf{1}_{OH}$ (blue line) and $\mathbf{1}_{Az}$ (red line) and (b) $\mathbf{2}_{OH}$ (blue line) and $\mathbf{2}_{Az}$ (red line).

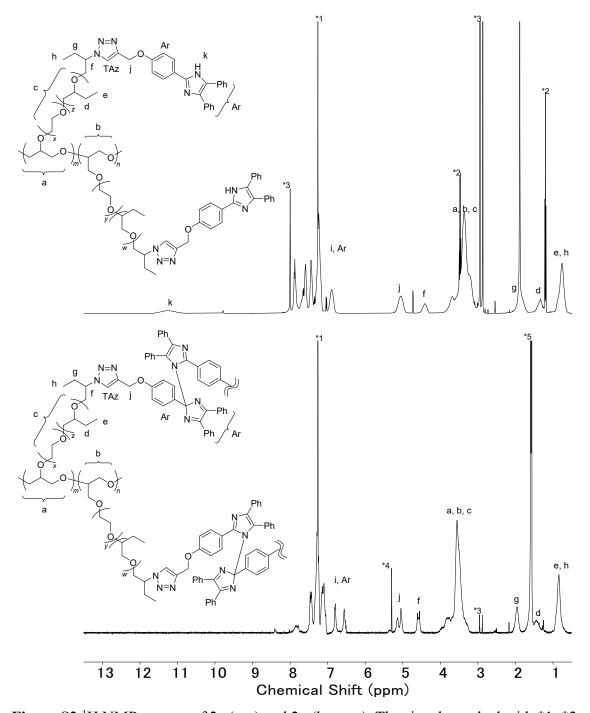


Figure S2 ¹H NMR spectra of 2_L (top) and 2_H (bottom). The signals marked with *1, *2, *3, *4, and *5 are CHCl₃, THF, DMF, DCM, and H₂O, respectively, which remained in the product due to difficulty in complete removal.

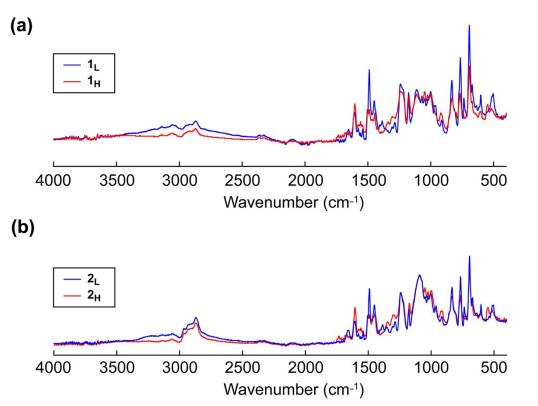


Figure S3 FT-IR spectra of (a) 1_L (blue line) and 1_H (red line) and (b) 2_L (blue line) and 2_H (red line).

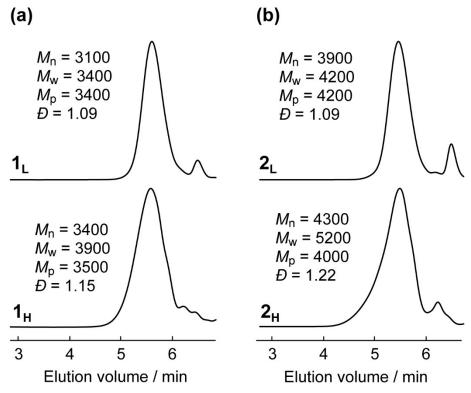


Figure S4 SEC traces of (a) 1_L (top) and 1_H (bottom) and (b) 2_L (top) and 2_H (bottom).