

Supporting Information

One-pot Catalyst-switching Synthesis of Thermoresponsive Amphiphilic Diblock Copolymers Consisting of Poly(*N,N*-diethylacrylamide) and Biodegradable Polyesters

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1. Table of PDEAm_x-OH, PDEAm_x-*b*-PTMC_y, and PDEAm_x-*b*-PLLA_y

Table S1. Preparation for a stock solution of PDEAm-OH in CH₂Cl₂ by i) the equimolar hydrosilylation of MAm-OTBDMS and Me₂EtSiH using B(C₆F₅)₃^a and ii) the GTP of DEAm using Me₃SiNTf₂,^b followed by iii) the deprotection using TBAF^c

PDEAm _x -OH	[DEAm] ₀ /[SKAm ^{Me₂Et} -OTBDMS] ₀	Time / h	<i>M</i> _{n,calcd} ^d / kg mol ⁻¹	<i>M</i> _{n,SEC} (<i>M</i> _w / <i>M</i> _n) ^e / kg mol ⁻¹	<i>T</i> _{cp} ^f
PDEAm ₃₀ -OH	30	6	3.8	3.6 (1.07)	53.8
PDEAm ₄₀ -OH	40	6	5.1	4.8 (1.08)	52.5
PDEAm ₅₀ -OH	50	6	6.4	6.2 (1.08)	51.9
PDEAm ₆₀ -OH	60	6	7.7	7.9 (1.07)	51.5
PDEAm ₇₀ -OH	70	6	8.9	8.7 (1.09)	51.0
PDEAm ₈₀ -OH	80	7	10.1	9.9 (1.10)	50.5
PDEAm ₉₀ -OH	90	8	11.4	10.5 (1.10)	49.9

^a Solvent, CH₂Cl₂; room temperature; argon atmosphere; MAm-OTBDMS, 0.22 mmol; [MAm-OTBDMS]₀/[Me₂EtSiH]₀/[B(C₆F₅)₃]₀ = 1.1/1.0/0.25. ^b [DEAm]₀, 1.0 mol L⁻¹; SKAm^{Me₂Et}-OTBDMS, 0.20 mmol; [SKAm^{Me₂Et}-OTBDMS]₀/[Me₃SiNTf₂]₀, 1/0.05; monomer conversion determined by ¹H NMR spectra in CDCl₃, >99%. ^c Tetrabutylammonium fluoride (TBAF) solution, 1.0 mol L⁻¹ in THF. ^d *M*_{n,calcd} = (MW of PDEAm-OH) + [TMC]₀/[PDEAm-OH]₀ x (monomer conversion) x (MW of TMC) + (M.W. of H) x 2. ^e Determined by SEC equipped with a RI detector in DMF containing LiCl (0.01 mol L⁻¹) using PMMA standards. ^f Determined by UV-vis measurements in water (3 g L⁻¹).

Table S2. Synthesis of PDEAm-*b*-PTMC by the ring-opening polymerization (ROP) of trimethylene carbonate (TMC) using PDEAm-OH as the macroinitiator and *t*-Bu-P₂ as the organocatalyst ^a

Sample code	[TMC] ₀ /[PDEAm-OH] ₀	$M_{n,calcd}$ ^b	$M_{n,SEC} (M_w/M_n)$ ^c
		/ kg mol ⁻¹	/ kg mol ⁻¹
PDEAm ₃₀ - <i>b</i> -PTMC ₇₀	70	11.0	11.5 (1.10)
PDEAm ₄₀ - <i>b</i> -PTMC ₆₀	60	11.2	11.7 (1.18)
PDEAm ₅₀ - <i>b</i> -PTMC ₅₀	50	11.5	11.2 (1.17)
PDEAm ₆₀ - <i>b</i> -PTMC ₄₀	40	11.7	12.5 (1.10)
PDEAm ₇₀ - <i>b</i> -PTMC ₃₀	30	12.0	12.2 (1.09)
PDEAm ₈₀ - <i>b</i> -PTMC ₂₀	20	12.2	13.2 (1.09)
PDEAm ₉₀ - <i>b</i> -PTMC ₁₀	10	12.5	13.5 (1.19)

^a [TMC]₀, 1.0 mol L⁻¹; solvent, CH₂Cl₂; room temperature; argon atmosphere; polymerization time, 24 h; monomer conversion determined by ¹H NMR spectra in CDCl₃, >99%. ^b $M_{n,calcd} = (\text{MW of PDEAm-OH}) + [\text{TMC}]_0/[\text{PDEAm-OH}]_0 \times (\text{monomer conversion}) \times (\text{MW of TMC}) + (\text{M.W. of H}) \times 2$. ^c Determined by SEC equipped with a RI detector in DMF containing LiCl (0.01 mol L⁻¹) using PMMA standards.

Table S3. Synthesis of PDEAm-*b*-PLLA by the ring-opening polymerization (ROP) of L-lactide (L-LA) using PDEAm-OH as the macroinitiator and *t*-Bu-P₂ as the organocatalyst ^a

Sample code	[L-LA] ₀ /[PDEAm-OH] ₀	$M_{n,calcd}$ ^b	$M_{n,SEC} (M_w/M_n)$ ^c
		/ kg mol ⁻¹	/ kg mol ⁻¹
PDEAm ₃₀ - <i>b</i> -PLLA ₇₀	70	13.9	11.6 (1.13)
PDEAm ₄₀ - <i>b</i> -PLLA ₆₀	60	13.7	11.2 (1.16)
PDEAm ₅₀ - <i>b</i> -PLLA ₅₀	50	13.5	12.5 (1.17)
PDEAm ₆₀ - <i>b</i> -PLLA ₄₀	40	13.4	11.8 (1.15)
PDEAm ₇₀ - <i>b</i> -PLLA ₃₀	30	13.2	13.9 (1.19)
PDEAm ₈₀ - <i>b</i> -PLLA ₂₀	20	13.0	12.9 (1.20)
PDEAm ₉₀ - <i>b</i> -PLLA ₁₀	10	12.9	14.3 (1.14)

^a [L-LA]₀, 1.0 mol L⁻¹; solvent, CH₂Cl₂; room temperature; argon atmosphere; polymerization time, 24 h; monomer conversion determined by ¹H NMR spectra in CDCl₃, >99%. ^b $M_{n,calcd} = (\text{MW of PDEAm-OH}) + [\text{L-LA}]_0/[\text{PDEAm-OH}]_0 \times (\text{monomer conversion}) \times (\text{MW of L-LA}) + (\text{MW of H}) \times 2$. ^c Determined by SEC equipped with a RI detector in DMF containing LiCl (0.01 mol L⁻¹) using PMMA standards.

2. SEC(RI) traces of PDEAM_x-*b*-PTMC_y and PDEAM_x-*b*-PLLA_y

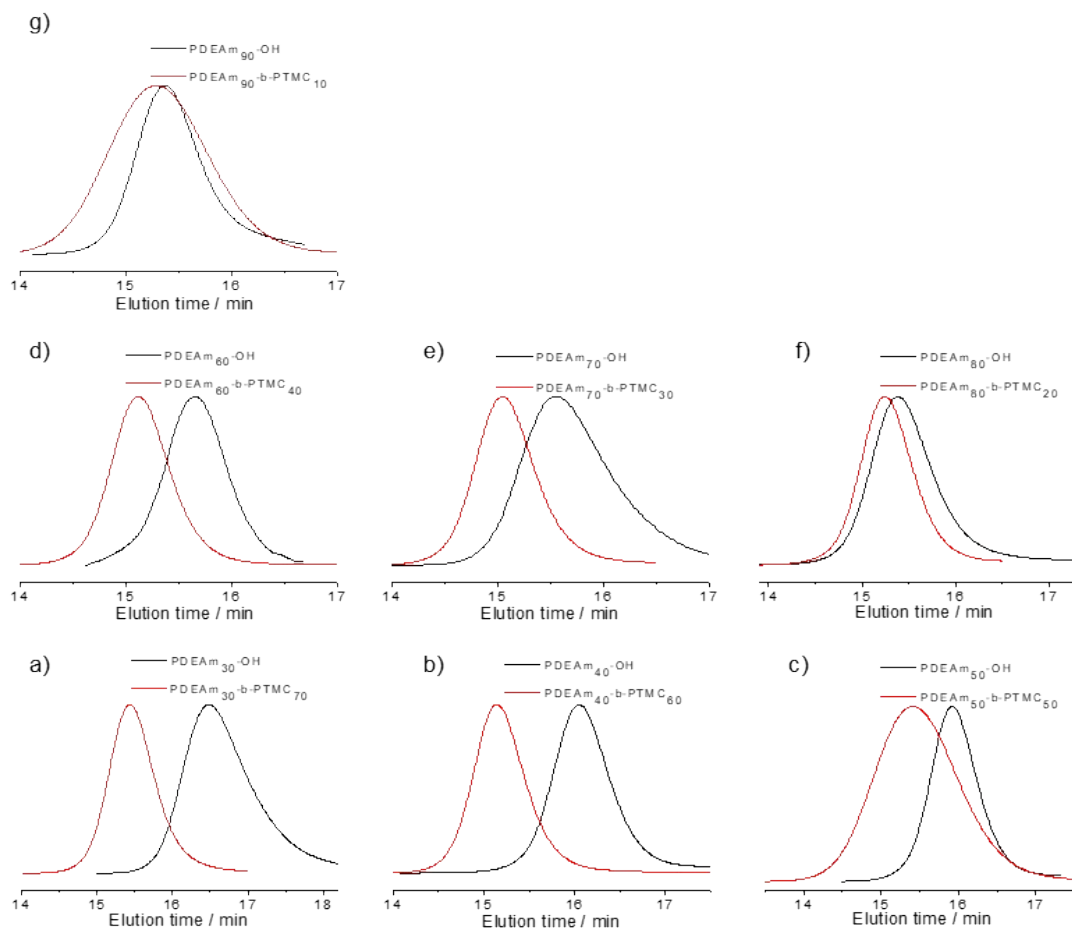


Figure S1. SEC traces of PDEAM and PDEAM-*b*-PTMC in CDCl₃.

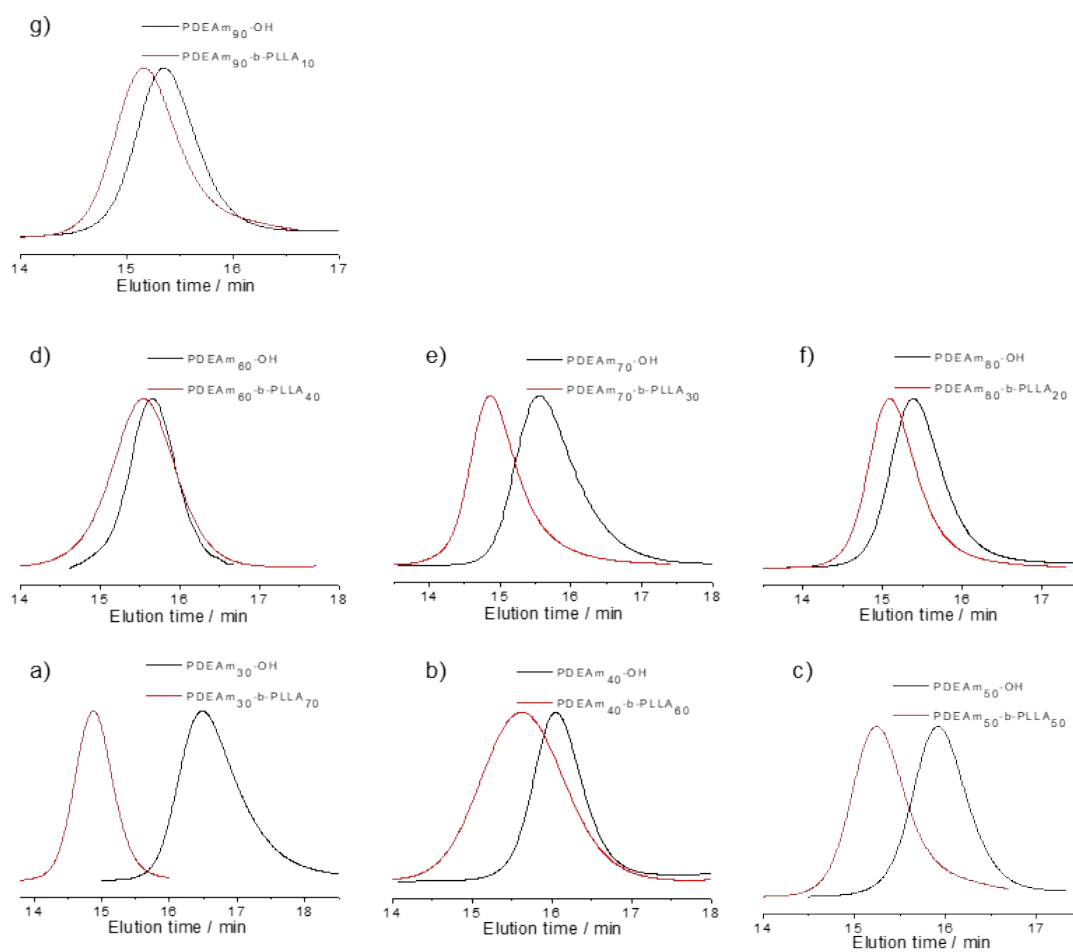


Figure S2. SEC traces PDEAm and of PDEAm-*b*-PLLA in CDCl₃.

3. ^1H NMR spectra of PDEAm-*b*-PTMC and PDEAm-*b*-PLLA

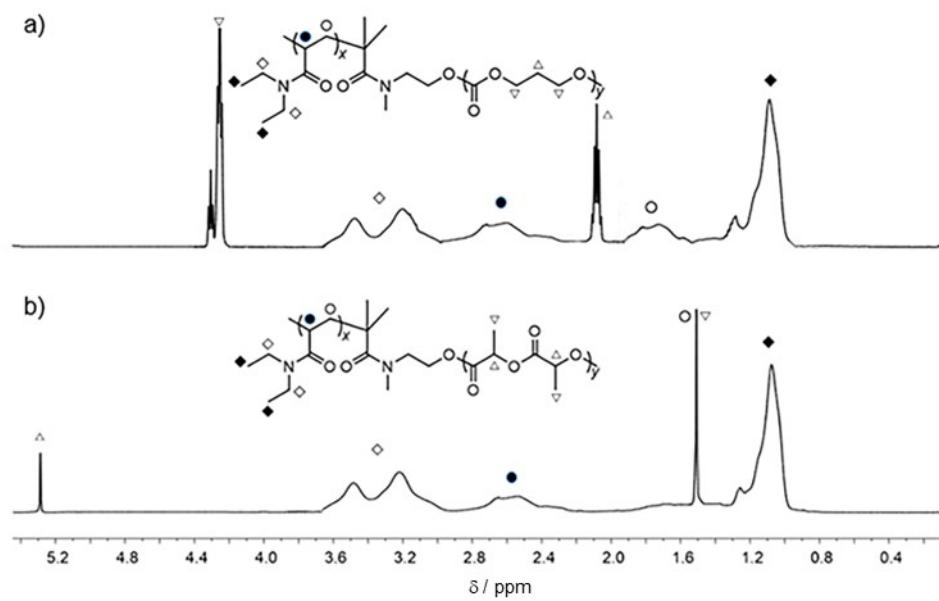


Figure S3. ^1H NMR spectra of a) PDEAm-*b*-PTMC and b) PDEAm-*b*-PLLA in CDCl_3 .

4. Dynamic ^1H NMR spectroscopy of $\text{PDEAm}_{30}\text{-}b\text{-PTMC}_{70}$ and $\text{PDEAm}_{30}\text{-}b\text{-PLLA}_{70}$

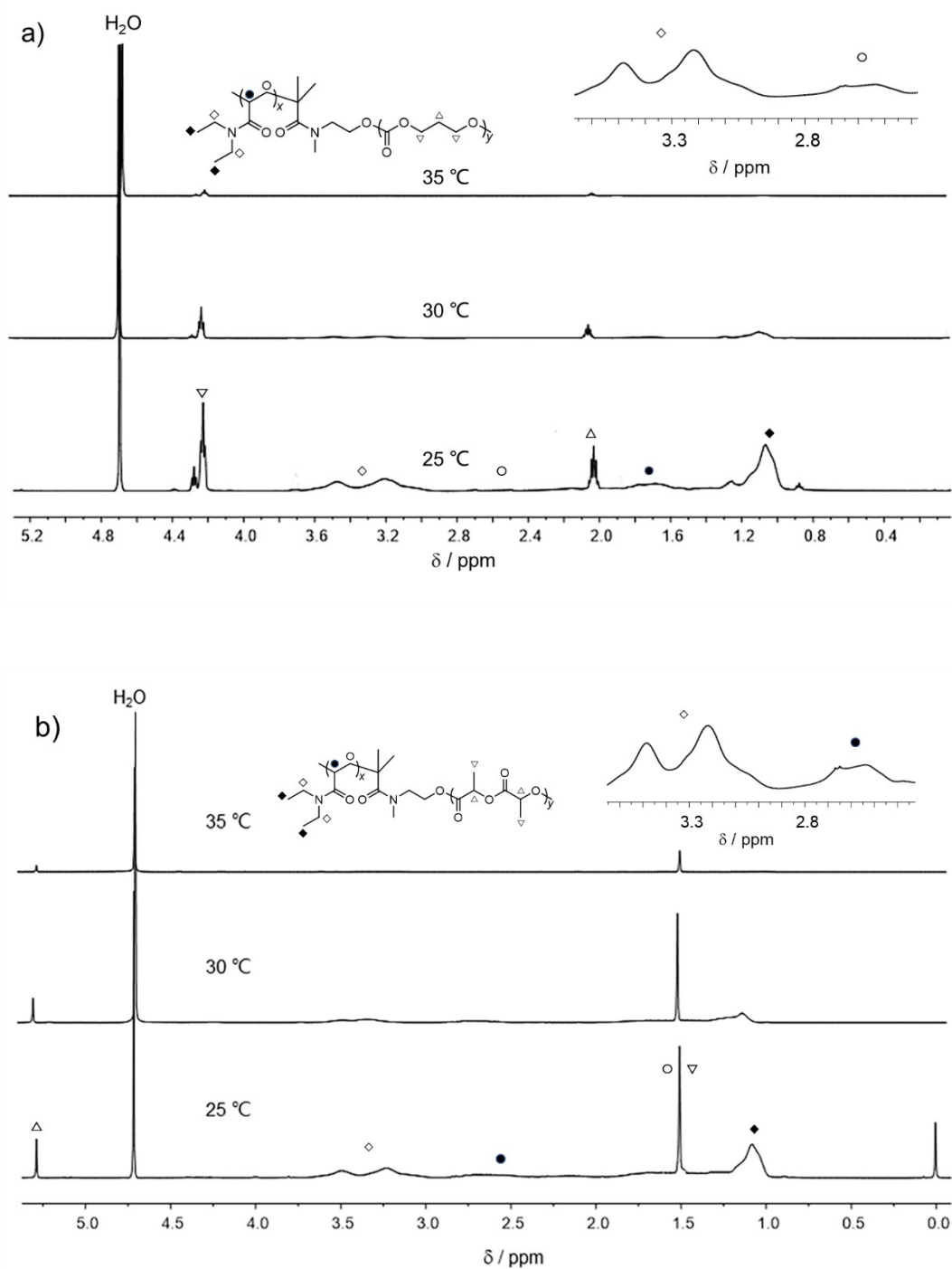


Figure S4. ^1H NMR spectra of a) $\text{PDEAm}_{30}\text{-}b\text{-PTMC}_{70}$ and b) $\text{PDEAm}_{30}\text{-}b\text{-PLLA}_{70}$ measured at 25, 30, 35 $^\circ\text{C}$ in D_2O .

5. Distribution of hydrodynamic radius (Rh) of PDEAM_x-b-PCL_y, PDEAM_x-b-PTMC_y, and PDEAM_x-b-PLLA_y at 25°C and 55°C

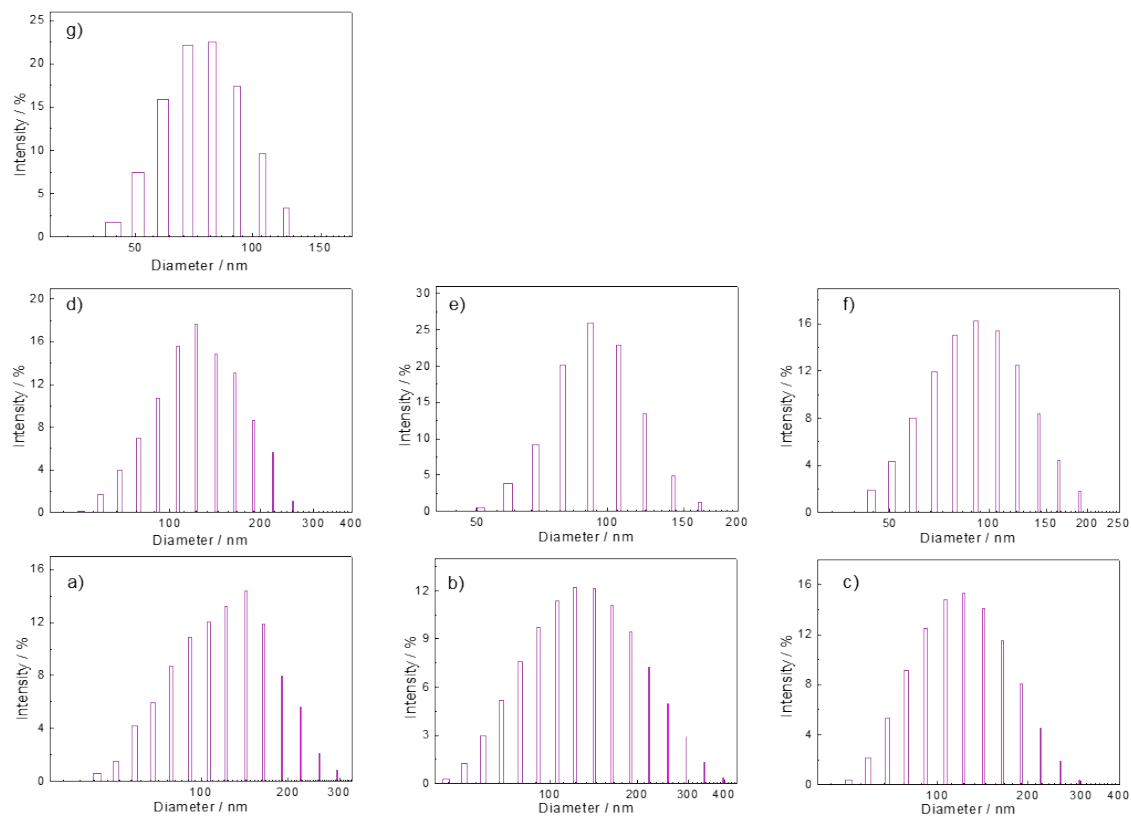


Figure S5. Distribution of hydrodynamic radii for PDEAM_x-b-PCL_y with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 25°C.

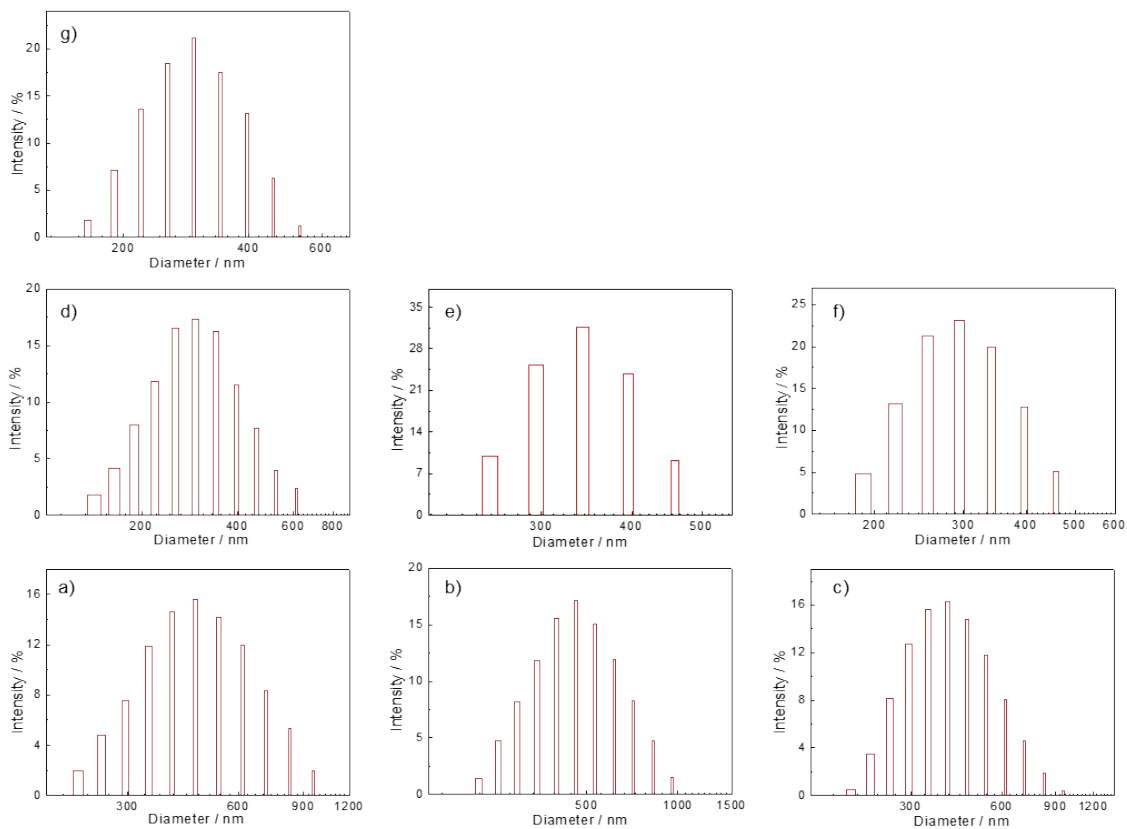


Figure S6. Distribution of hydrodynamic radii for PDEAm_x-b-PCL_y with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 55°C.

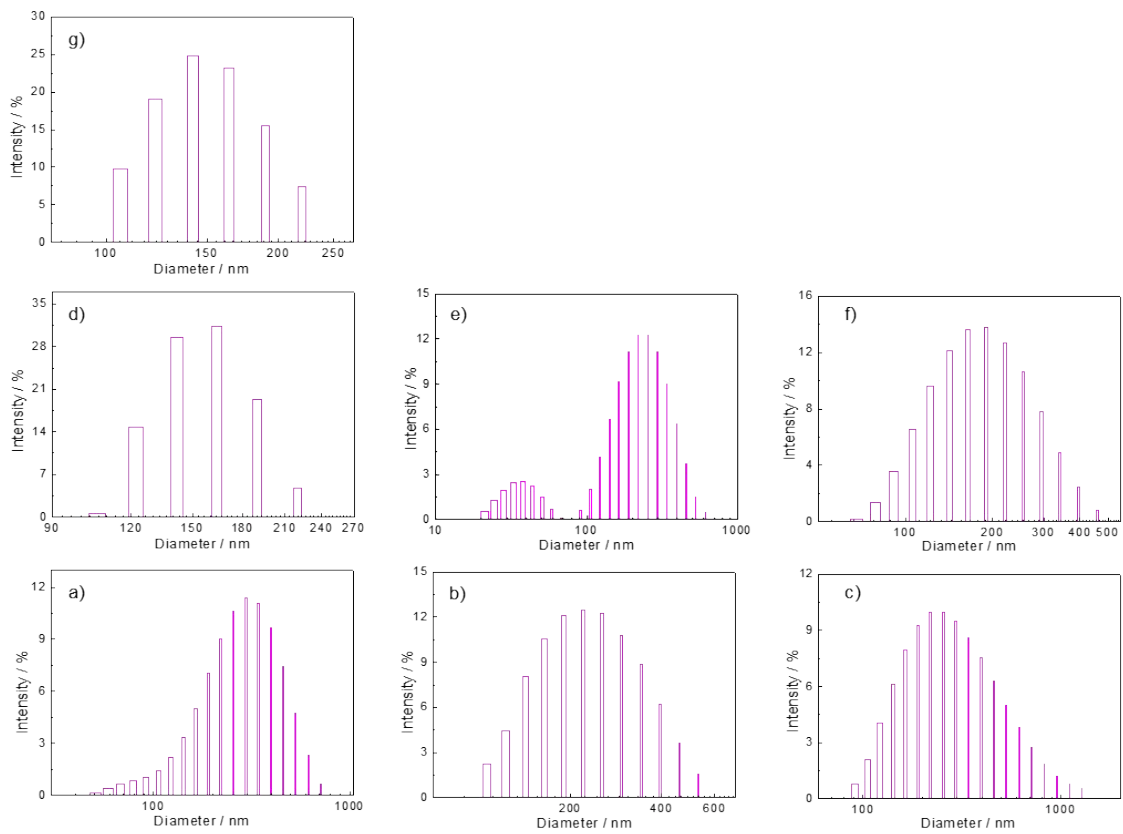


Figure S7. Distribution of hydrodynamic radii for PDEAm_x-b-PTMC_y with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 25°C.

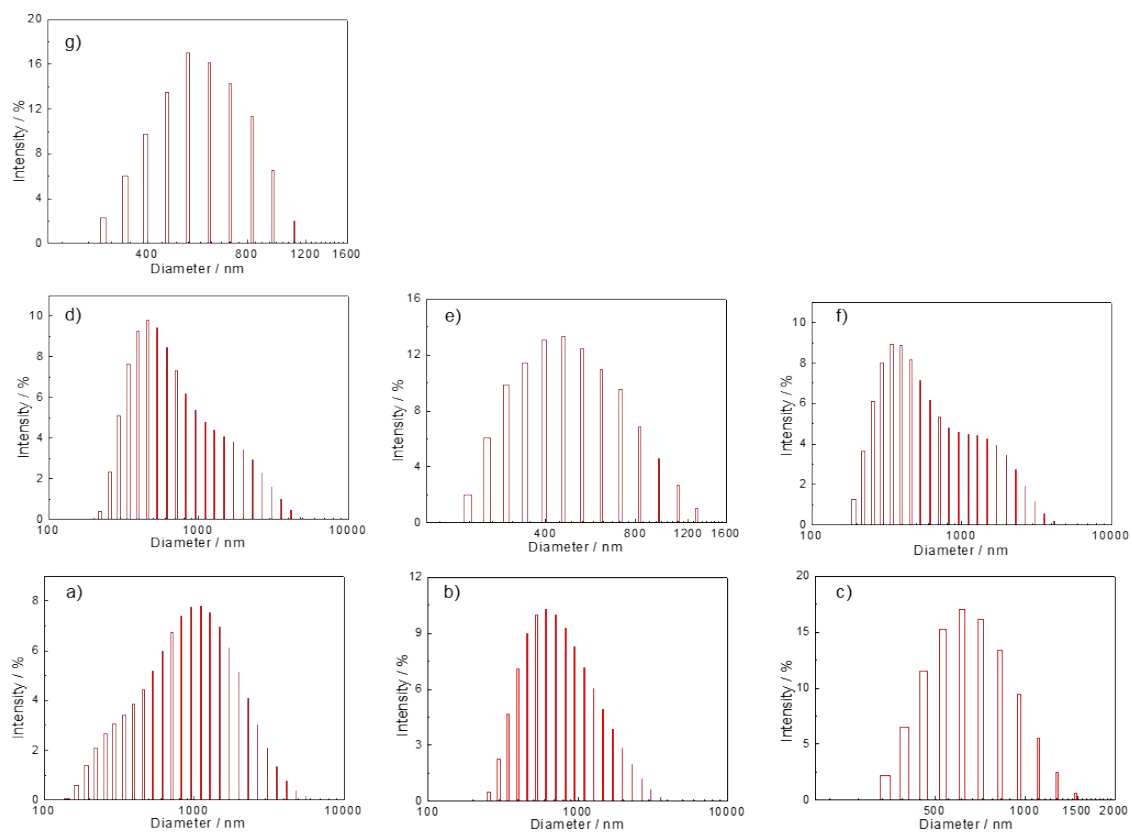


Figure S8. Distribution of hydrodynamic radii for PDEAm_x-b-PTMC_y, with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 55°C.

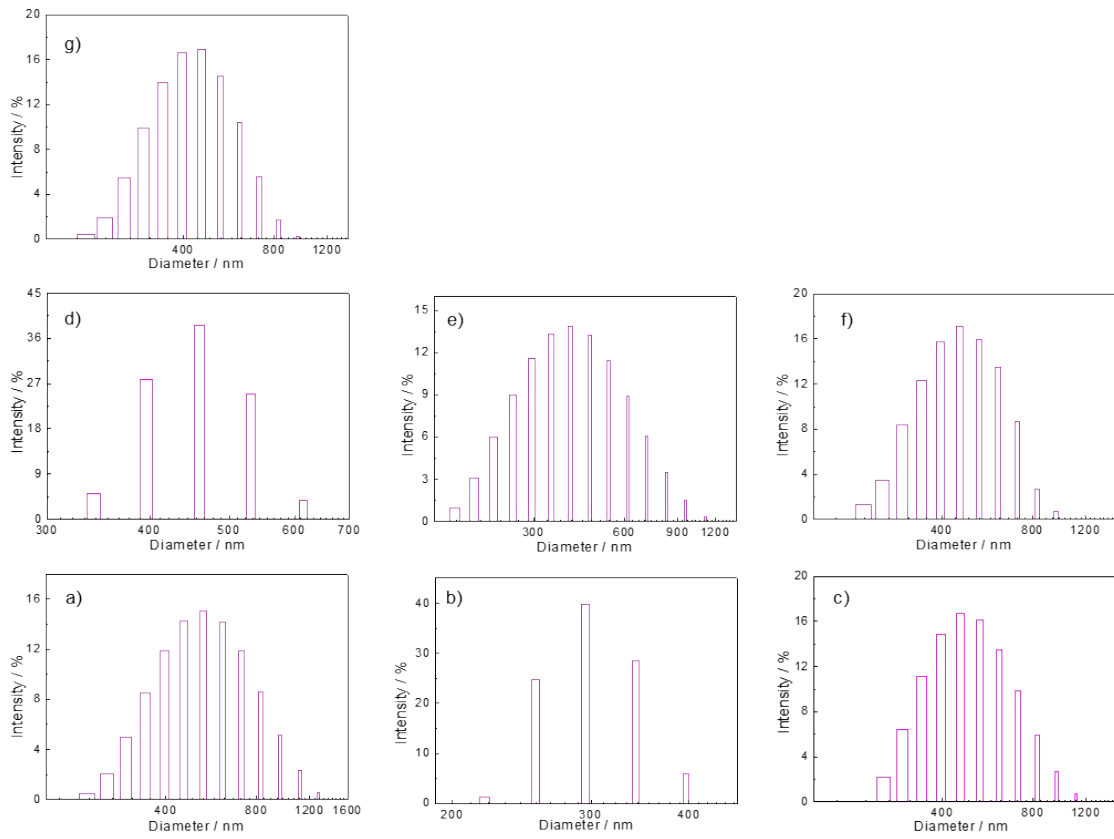


Figure S9. Distribution of hydrodynamic radii for PDEAm_x-b-PLLA_y with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 25°C.

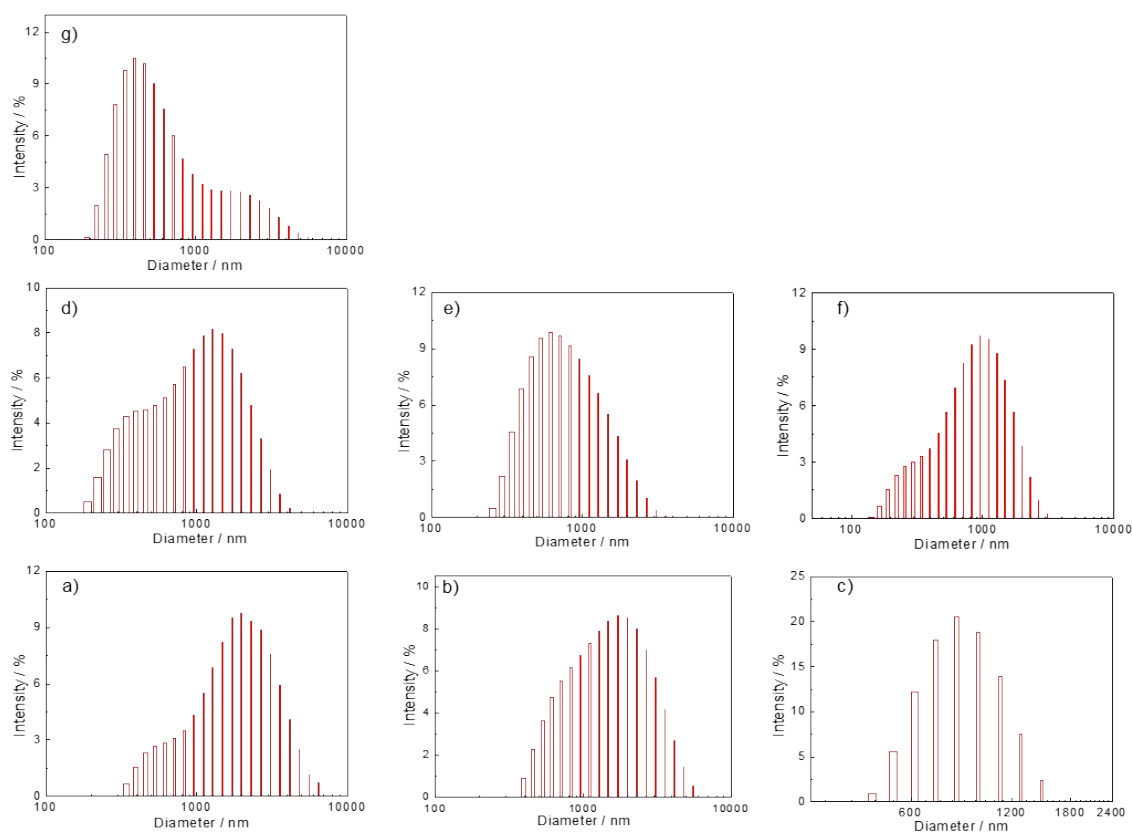


Figure S10. Distribution of hydrodynamic radii for PDEAm_x-b-PLLA_y, with x/y ratios of (a) 30/70, (b) 40/60, (c) 50/50, (d) 60/40, (e) 70/30, (f) 80/20, and (g) 90/10 at 55°C.