

## Supporting Information

# Narrowing Down Chain Length Effects on the Antibacterial Action of Guanylated Oligomers

*Dries Wyers,<sup>a</sup> Thanavit Jirapanjawat,<sup>b,c</sup> John F. Quinn,<sup>d,e</sup> Michael R. Whittaker,<sup>d</sup> Chris  
Greening,<sup>b,c</sup> Tanja Junkers<sup>a,\*</sup>*

<sup>a</sup> *Polymer Reaction Design Group, School of Chemistry, Monash University, 19 Rainforest Walk,  
Clayton VIC 3800, Australia*

<sup>b</sup> *Centre to Impact AMR, Monash University, Clayton, VIC 3800, Australia*

<sup>c</sup> *Department of Microbiology, Biomedicine Discovery Institute, Clayton, VIC 3800, Australia*

<sup>d</sup> *Drug Delivery, Disposition and Dynamics Theme, Monash Institute of Pharmaceutical Sciences,  
Monash University, Parkville, Victoria 3052, Australia*

<sup>e</sup> *Department of Chemical Engineering, Faculty of Engineering, Monash University, Clayton,  
Victoria 3800, Australia*

# $^1\text{H}$ and $^{13}\text{C}$ NMR analysis monomer

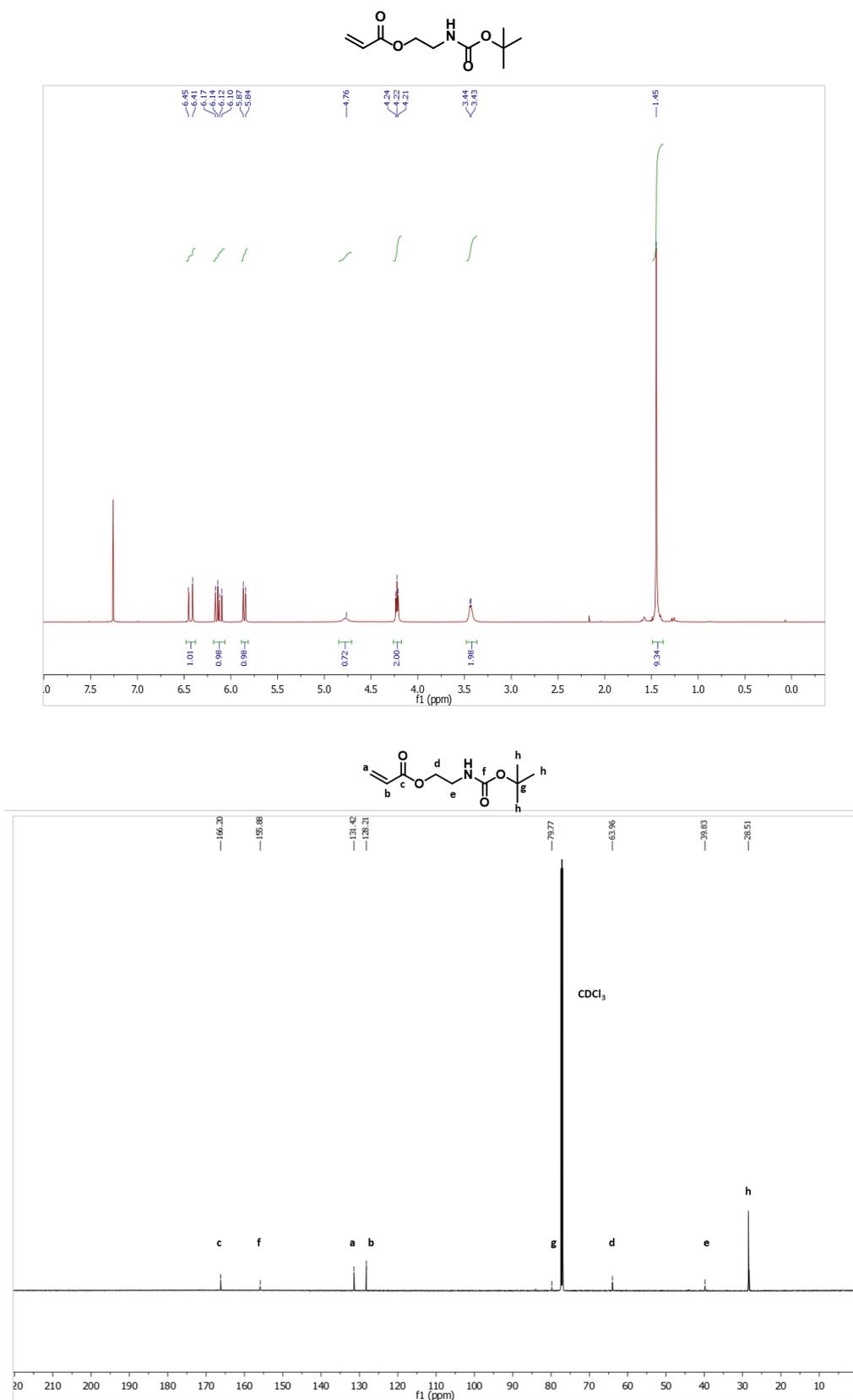
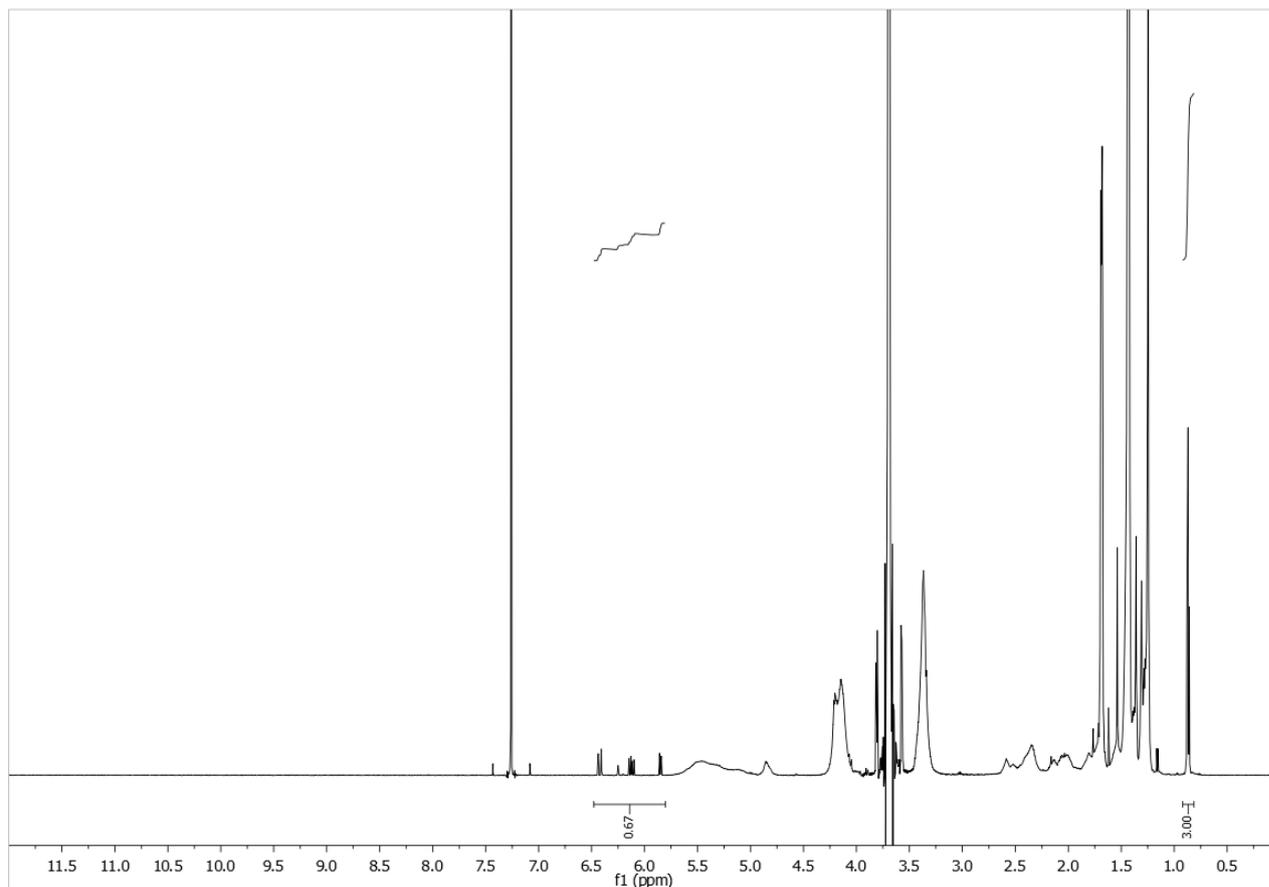


Figure S1:  $^1\text{H}$  NMR (upper) and  $^{13}\text{C}$  NMR (lower) of 2-Boc aminoethyl acrylate

## **<sup>1</sup>H NMR analysis of discrete and disperse oligomers before/after thiol-ene reaction and deprotection**



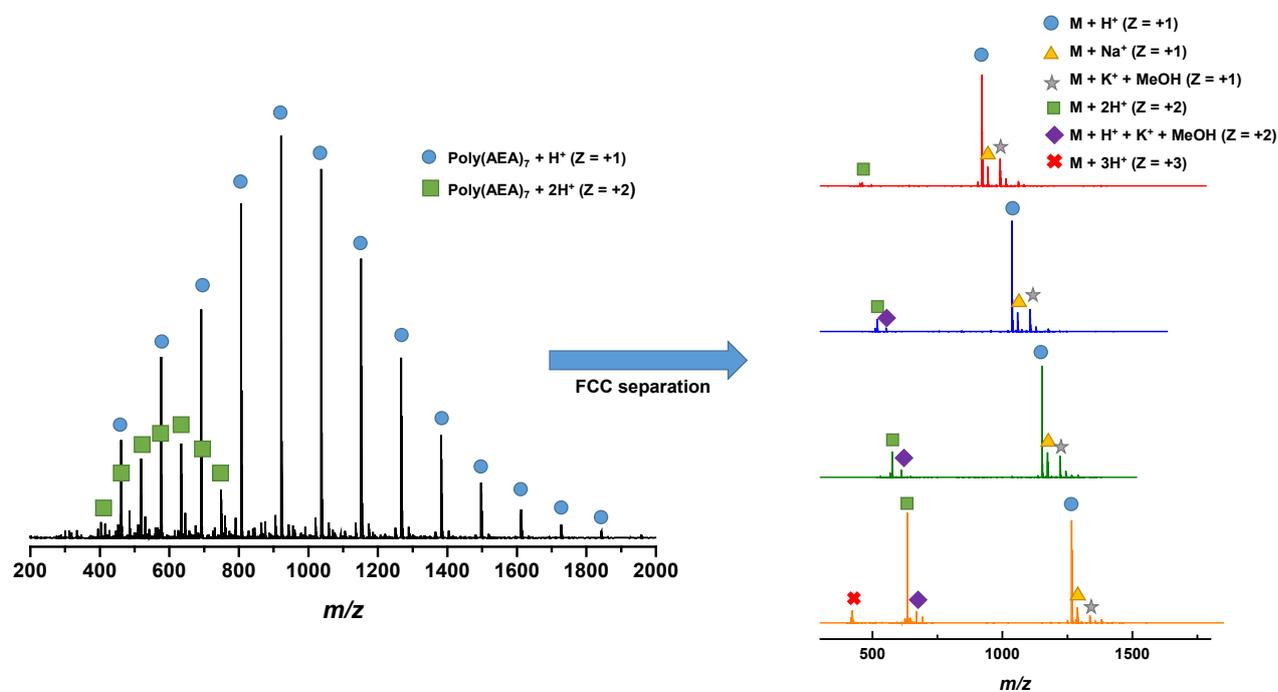
**Figure S2:** Determination of the monomer conversions of homopolymers of the different model monomers via <sup>1</sup>H NMR in CDCl<sub>3</sub>: comparing the average integration of the vinyl peaks (monomer) to the integration of the endgroup as reference integrated for three protons. A reference sample at 0 minutes reaction time (T<sub>0</sub>) was taken to which the vinyl integration were compared. The monomer conversion (p) was calculated using the following formula:

$$p = 1 - \left( \frac{\int I_{\text{vinyl peaks } T_r}}{\int I_{\text{vinyl peaks } T_0}} \right)$$

T<sub>r</sub> represents the reaction time after which a sample was taken.

**Table S1:** Summarizing table of the DP, theoretical and exact masses of monodisperse oligomers isolated through FCC separation and boc deprotection. Exact masses were found on ESI-MS.

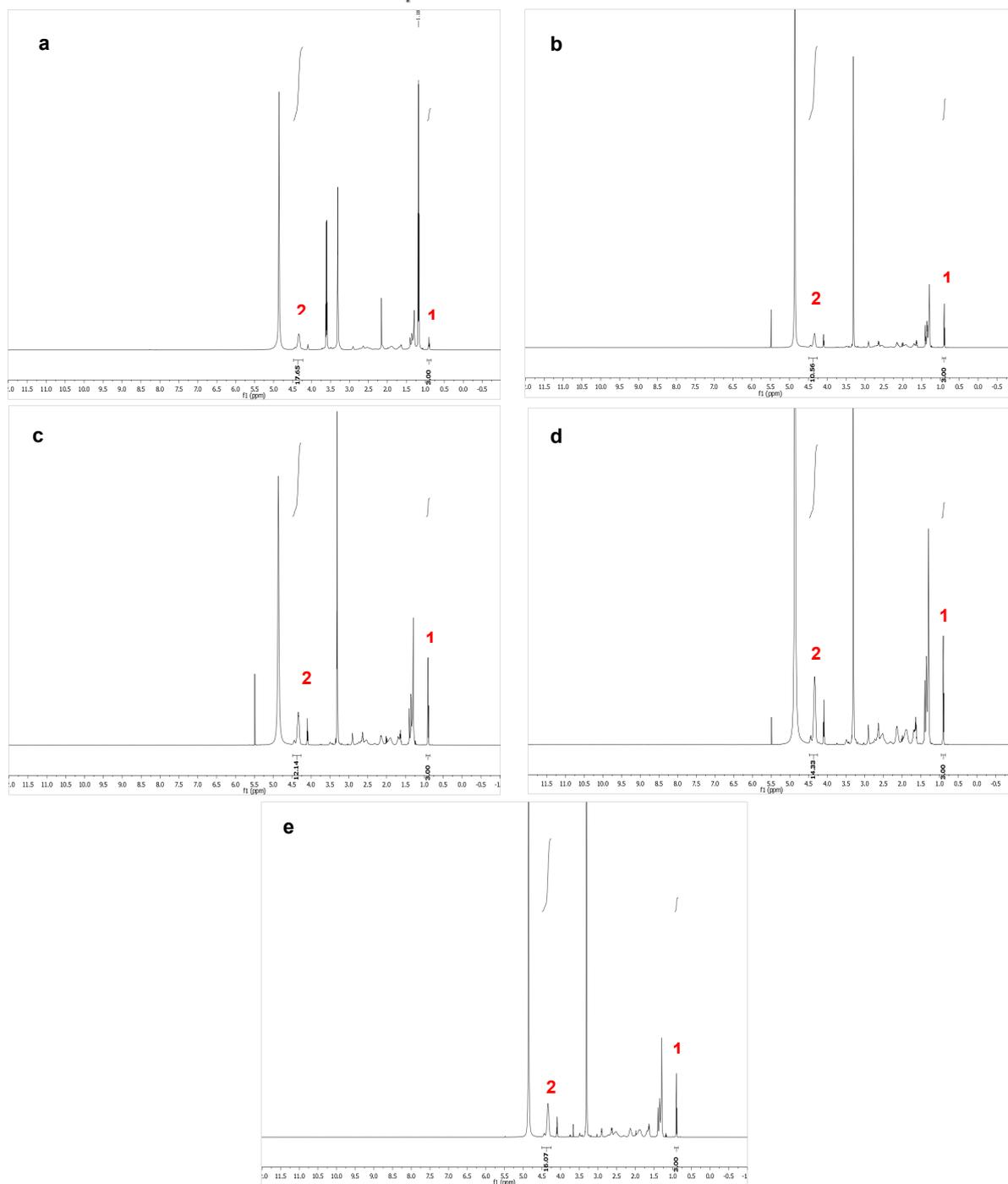
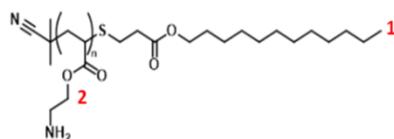
Insert	DP	Theoretical mass	Exact mass
		(M + H <sup>+</sup> )	(M + H <sup>+</sup> )
1	5	921.486	921.474
2	6	1036.549	1036.550
3	7	1151.612	1151.613
4	8	1266.676	1266.674



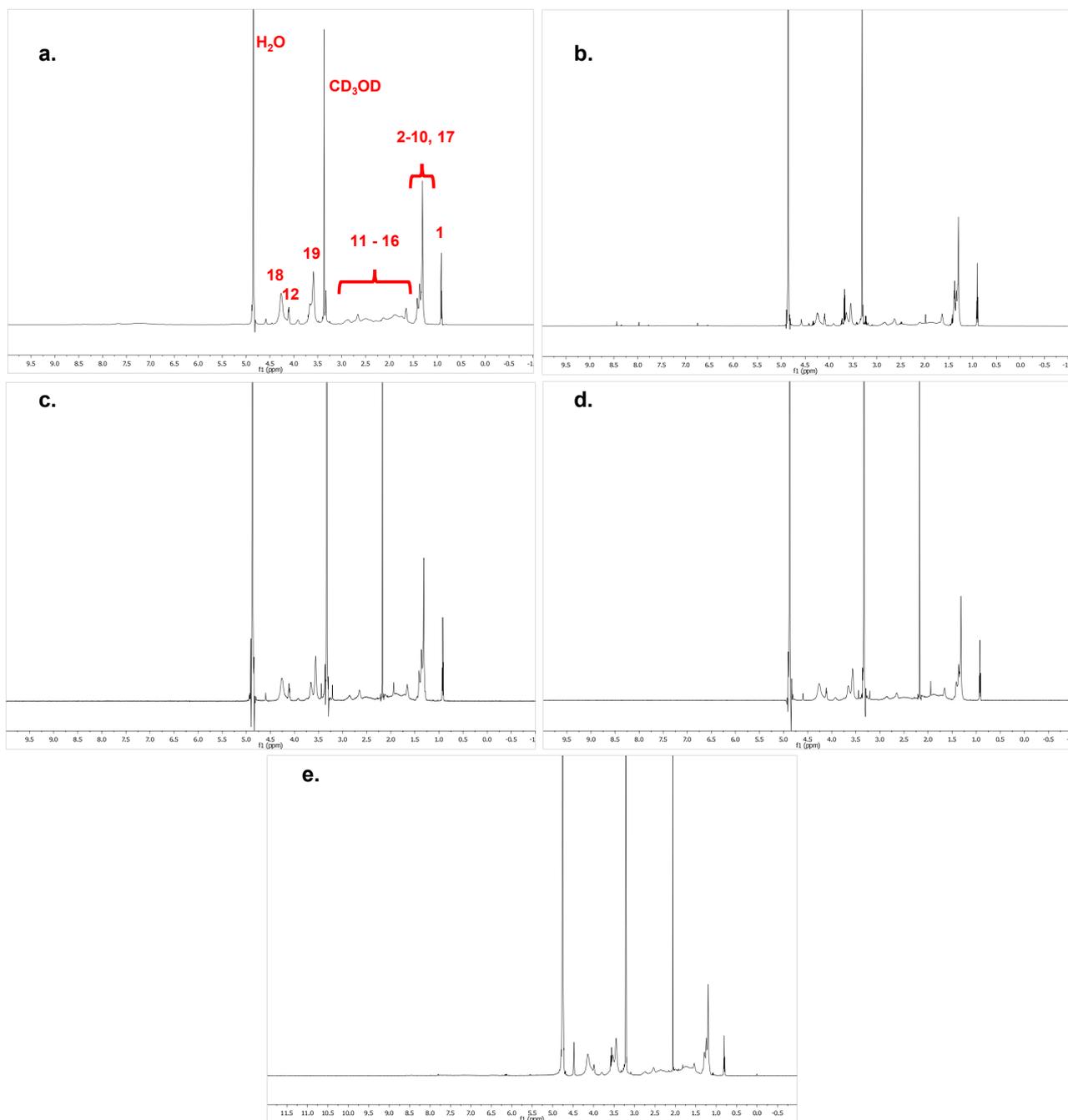
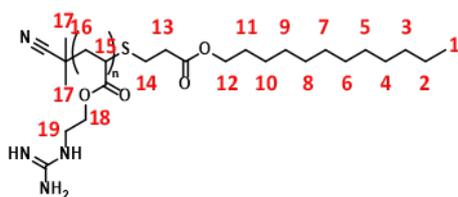
**Figure S3:** ESI-MS spectra of both crude (left) poly(AEA)<sub>7</sub> and isolated monodisperse oligomers (right) with 5 to 8 monomer insertions. Boc protection group removal was required prior to ESI-MS analysis to ensure proper ionization of the compounds.

**Table S2:** Summarizing table of the DP, theoretical and exact masses of monodisperse oligomers after end group modification through aminolysis and *in situ* thiolene coupling and subsequent boc deprotection.

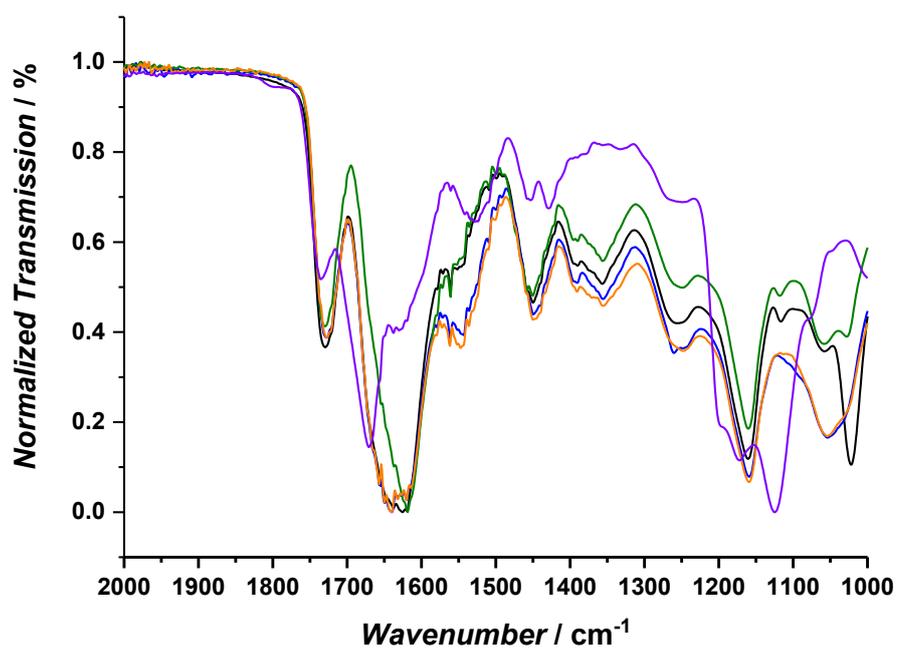
<b>Insert</b>	<b>DP</b>	<b>Theoretical mass (M + H<sup>+</sup>)</b>	<b>Exact mass (M + H<sup>+</sup>)</b>
<b>1</b>	5	917.5628	917.5684
<b>2</b>	6	1032.6261	1032.6351
<b>3</b>	7	1147.6894	1147.6991
<b>4</b>	8	1262.7528	1262.7425



**Figure S4:** <sup>1</sup>H NMR of disperse (a) and discrete oligomers with 5 (b), 6 (c), 7 (d) or 8 (e) monomer insertions after end-group modification through aminolysis and a *in situ* thiolene coupling with dodecyl acrylate and subsequent deprotection with TFA. Complete Boc removal is confirmed by the absence of a singlet at 1.45 ppm indicative of the tert-butyl protons of Boc protection groups. All NMRs are measured in CD<sub>3</sub>OD. The peaks at 1.18 and 3.6 ppm, in spectrum a, and at 5.49 ppm in spectrum c and d are residual peaks of diethyl ether and DCM respectively.



**Figure S5:**  $^1\text{H}$  NMR of disperse (a) and discrete oligomers with 5 (b), 6 (c), 7 (d) or 8 (e) monomer insertions after guanylation. Conversion of primary amines to guanidines is confirmed by a shift in the proton signals adjacent to the functional groups (19) from from  $\sim 3.30$  ppm (adjacent to amine) to 3.54 ppm (adjacent to guanidine). All NMRs are measured in  $\text{CD}_3\text{OD}$ . The peak at 2.16 ppm in spectrum c and d is a residual peak of acetone.



**Figure S6:** ATR FT-IR spectra of primary amine oligomer (purple) and of guanidine modified oligomers (black – orange). After guanylation, a new peak appears around 1639 cm<sup>-1</sup> which is indicative of C=N stretching. In addition, no evidence of base induced isomerization of the oligomers is present with the FT-IR spectrum retaining a C=O stretch signal at 1718 cm<sup>-1</sup> and C-O stretch at 1146 cm<sup>-1</sup>, both indicating the presence of esters.