SUPPORTING INFORMATION

A BINARY NEODYMIUM CATALYST FOR THE POLYMERIZATION OF LACTONES

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Figure S1. ¹H and NMR spectrum of polycaprolactone obtained with NdCl₃·3TEP/TIBA catalytic system

From the slopes of the plot in Figure S2 (c), k_{app} (kapp = k x [M*], where k = rate constant and [M*]=concentration of active species, which is constant for a *living* polymerization) was calculated at the three temperatures and E_a was determined from the Arrhenius plot (Figure S2 (d)) derived from the Arrhenius equation shown below,

$$\ln k = -\frac{E_a}{RT} + \ln A$$

where k = rate constant E_a = activation energy R= universal gas constant T= absolute temperature A= Arrhenius constant



Figure S2. (a) % Conversion vs. time, (b) $\ln\{[M]_0/[M]\}$ vs. time, (c) $\ln\{[M]_0/[M]\}$ vs. time for conversions < 20%, at different temperatures and (d) Arrhenius plot for the polymerization of ε -CL with NdCl₃·3TEP/TIBA catalytic system at a [ε -CL]:[Nd]:[Al] ratio of 500:1:5



Figure S3. SEC traces for (a) PCL before and after chain extension obtained at a [ϵ -CL1]: [ϵ -CL2]:[Nd]:[A1] ratio of 50:100:1:5 and (b) PVL and PVL-*b*-PCL (c) PBrCL and PBrCL-*b*-PCL (d) PMEEECL and PMEEECL-*b*-PCL obtained at a [M]: [ϵ -CL]:[Nd]:[A1] ratio of 50:100:1:5 (e) PCL and PCL-*b*-PVL at a [ϵ -CL]: [δ -VL]:[Nd]:[A1] ratio of 200:1000:1:5 (f) PCL and PCL-*b*-PBnCL at a [ϵ -CL]: [δ -BnCL]:[Nd]:[A1] ratio of 75:25:1:5 at 80 °C (M = δ -VL or MEEECL)



Scheme S1. Schematic representation of the synthesis of (a) PVL-*b*-PCL, (b) PBrCL-*b*-PCL, (c) PMEEECL-*b*-PCL, d) PCL-*b*-PVL, and e) PCL-*b*-PBnCL



Figure S4. DSC traces for PBrCL-*b*-PCL, PMEEECL-*b*-PCL, PCL, PVL and PVL-*b*-PCL for the temperature range of -100 °C to 100 °C



Figure S5. ¹H NMR spectrum of PVL-*b*-PCL. ($M_n = 12,100$ Da, D = 1.4, composition for δ -VL and ϵ -CL is 29 and 71 mol% respectively)



Figure S6. ¹H NMR spectrum of PBrCL-*b*-PCL. ($M_n = 8,100$ Da, D = 1.4, composition for BrCL and ε -CL is 27 and 73 mol% respectively)



Figure S7. ¹H NMR spectrum of PCL-*b*-PVL. ($M_n = 90,100$ Da, D = 1.6, composition for ϵ -CL and δ -VL is 15 and 85 mol% respectively)



Figure S8. ¹H NMR spectrum of PCL-*b*-PBnCL. ($M_n = Da, D = 1$, composition for ε -CL and δ -BnCL is 70 and 30 mol% respectively)



Figure S9. 2D DOSY plot for PCL (purple), PBrCL (red) and PBrCL-*b*-PCL (blue)



Figure S10. ¹H NMR of PMEEECL-*b*-PCL. ($M_n = 12,100$ Da, D = 1.6, composition for MEEECL and ε -CL is 17 and 83 mol% respectively)



Scheme S2. Synthesis of γ -bromo- ε -caprolactone using different synthetic routes



Figure S11. ¹H NMR spectrum of poly(γ -benzyl- ϵ -caprolactone). ($M_n = 17,600$ Da, D = 1.6)



Figure S12. ¹H NMR spectrum of poly(γ -bromo- ϵ -caprolactone). ($M_n = 23,400$ Da, D = 1.8)



Figure S13. ¹H NMR spectrum of poly(γ -phenylbutyrate- ϵ -caprolactone). ($M_n = 22,000$ Da, D = 1.5)



Figure S14. ¹H NMR spectrum of poly(γ -acetate- ϵ -caprolactone). ($M_n = 11,300$ Da, D = 1.3)



Figure S15. ¹H NMR spectrum of polyvalerolactone. ($M_n = 18,300 \text{ Da}, D = 1.5$)



Figure S16. ¹H NMR spectrum of poly (γ -2-[2-(2-methoxyethoxy)ethoxy]ethoxy- ϵ -caprolactone). (M_n = 8,300 Da, D = 1.1)



Figure S17. ¹H NMR spectrum of 4-bromocyclohexanone



Figure S18. ¹H NMR spectrum of γ -bromo- ϵ -caprolactone



Figure S19. ¹H NMR spectrum of 4-oxocyclohexyl acetate



Figure S20. ¹H NMR spectrum of γ -acetate- ϵ -caprolactone