Supporting Information

Hofmeister Effect-Driven Superlattice ConstructionviaHydrophilic/HydrophobicTransitionofPoly(ethylene glycol) Ligands

Yanqiu Du,^a Haidong Li,^a Yang Jiang,^a Yunchao Xiao,^a Jipeng Guan,^a Xuejie Liu,^{*a} and Nan

Yan*b

^a College of Materials and Textile Engineering, Jiaxing University, Jiaxing 314001, China

^b Department of Chemistry, Changchun Normal University, Changchun 130032, China

* Corresponding author: E-mail: <u>yannan@ccsfu.edu.cn</u>; 00008109@zjxu.edu.cn

Supplementary Figures



Fig. S1 (a-b) TEM and SEM images of the monodispersed AuNPs_{19.3}@PEG_{1k} building blocks. (c) The size distribution histogram of corresponding AuNPs_{19.3}@PEG_{1k}. (d) UV-Vis spectra of the monodispersed AuNPs_{19.3}@PEG_{1k}.



Fig. S2 TGA analysis of the AuNPs_{19.3}@PEG_{1k} nanoparticles.



Fig. S3 (a-b) SEM imges of the AuNPs_{19.3} in a 0.60 M K₂CO₃ solution and deionized water, respectively; (c) SEM image of the AuNPs_{19.3}@PEG_{1k} in deionized water.



Fig. S4 (a) SEM image of the 2D superlattice sheets at low magnification. (b) The size distribution

histogram of corresponding monolayer superlattice sheets.



Fig. S5 STEM image of the 2D monolayer superlattice sheets under low magnification.



Fig. S6 The interparticle centerpoint-to-centerpoint distances (*d*) in the formed 2D monolayer superlattices are determined by TEM.



Fig. S7 SEM images of 2D monolayer superlattices (a) and 3D superlattices self-assembled assembled from $AuNPs_{19.3}@PEG_{1k}$ (b) building blocks in a 0.60 M and 0.69 M Na_2CO_3 solution, respectively.



Fig. S8 SEM image of hexagonal monolayer superlattices formed in a 0.69 M K₂CO₃ solution for

12 h.



Fig. S9 A sequence of SEM images showing the formation of hexagonal monolayer superlattices

formed in a 0.69 M K_2CO_3 solution.



Fig. S10 SEM image of 3D single crystals in a 0.69 M K₂CO₃ solution.



Fig. S11 UV-Vis spectra of dispersed AuNPs_{19.3}@PEG_{1k} (black line) and single crystals selfassembled from AuNPs_{19.3}@PEG_{1k} building blocks (blue line).



Fig. S12 SEM image of the single crystal obtained in a $0.69 \text{ M K}_2\text{CO}_3$ solution when the growth time is 6 h.



Fig. S13 SEM image of the single crystal obtained in a 0.69 M K_2CO_3 solution when the growth time is 13 h.



Fig. S14 SEM image of the single crystal obtained in a $0.69 \text{ M K}_2\text{CO}_3$ solution when the growth time is 24 h.



Fig. S15 The formation of multiply twinned superlattices. SEM images of multiply twinned superlattices with quasi-icosahedron (a), a single twin plane (b) and several parallel twin planes (c) that are generated in a 0.85 M K₂CO₃ solution.



Fig. S16 (a-b) TEM and SEM images of the monodispersed AuNPs with the diameter of 16.6 nm. (c) The size distribution histogram of corresponding AuNPs_{16.6}. (d) UV-Vis spectra of the monodispersed AuNPs_{16.6}.



Fig. S17 (a-b) TEM and SEM images of the monodispersed AuNPs with the diameter of 28.0 nm. (c) The size distribution histogram of corresponding AuNPs_{28.0}. (d) UV-Vis spectra of the monodispersed AuNPs_{28.0}.



Fig. S18 (a-b) TEM and SEM images of the monodispersed AuNPs with the diameter of 31.2 nm. (c) The size distribution histogram of corresponding AuNPs_{31.2}. (d) UV-Vis spectra of the monodispersed AuNPs_{31.2}.



Fig. S19 SEM images of the assembly behavior of AuNPs_{16.6}@PEG_{2k} at λ of 0.96 in different K₂CO₃ concentration solution, from left to right are dispersed nanoparticles, monolayer superlattices and disordered aggregates.



Fig. S20 SEM images of the assembly behavior of $AuNPs_{16.6}@PEG_{5k}$ at λ of 1.55 in different K_2CO_3 concentration solution, from left to right are dispersed nanoparticles, monolayer superlattices and disordered aggregates.



Fig. S21 SEM images of the assembly behavior of AuNPs_{19.3}@PEG_{2k} at λ of 0.83 in different K₂CO₃ concentration solution, from left to right are dispersed nanoparticles, monolayer superlattices and disordered aggregates.



Fig. S22 SEM images of the assembly behavior of AuNPs_{19.3}@PEG_{5k} at λ of 1.35 in different K₂CO₃ concentration solution, from left to right are dispersed nanoparticles, monolayer

superlattices and disordered aggregates.



Fig. S23 SEM images of the assembly behavior of $AuNPs_{28.0}@PEG_{5k}$ at λ of 0.93 in different K_2CO_3 concentration solution, from left to right are dispersed nanoparticles, monolayer superlattices and disordered aggregates.



Fig. S24 SAXS pattern of the disordered aggregates assembled from $AuNPs_{16.6}@PEG_{1k}$ building blocks in 1.00 M K₂CO₃ solution.