

Supporting Information

Photoreforming of polyester plastics into added-value chemicals coupling with H₂ evolution over a Ni₂P/ZnIn₂S₄ catalyst

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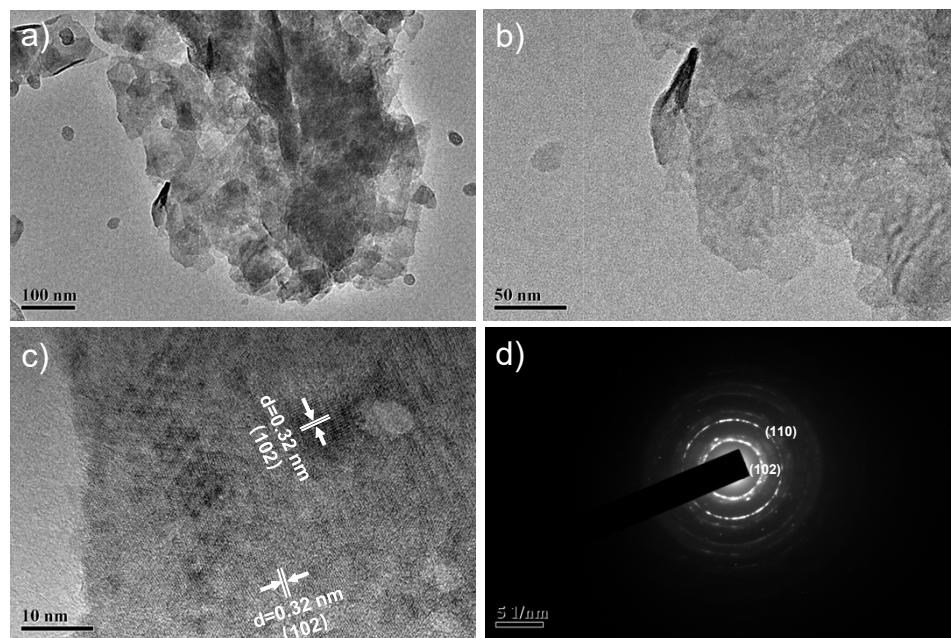


Fig. S1 a-b) TEM and c) high-resolution TEM images of pristine ZnIn_2S_4 nanosheets; d) SAED pattern of pristine ZnIn_2S_4 nanosheets.

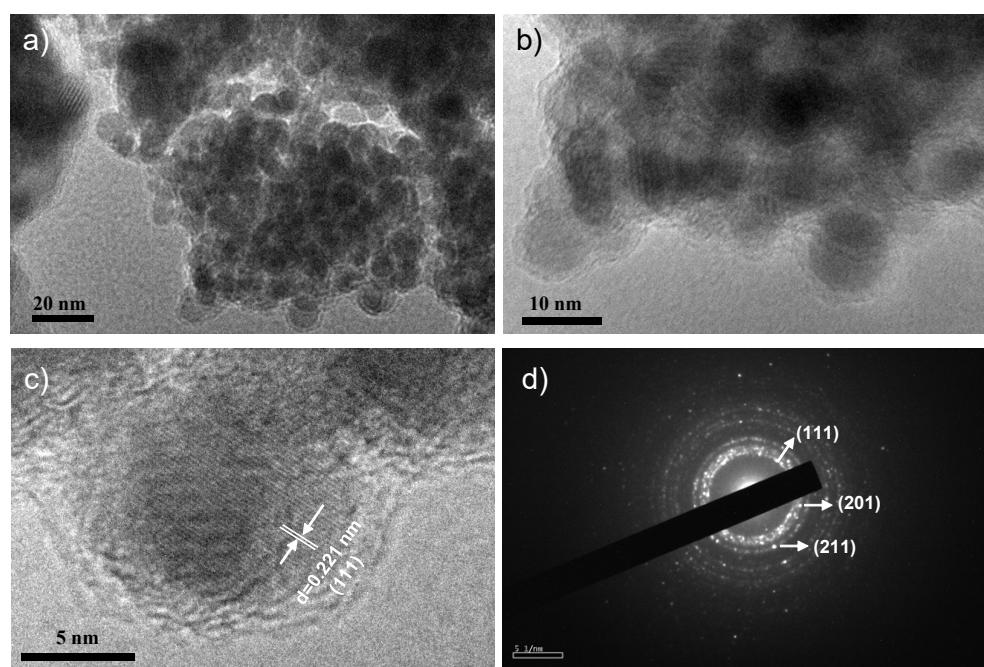


Fig. S2 a-b) TEM and c) high-resolution TEM images of pure Ni_2P nanoparticles; d) SAED pattern of pure Ni_2P nanoparticles.

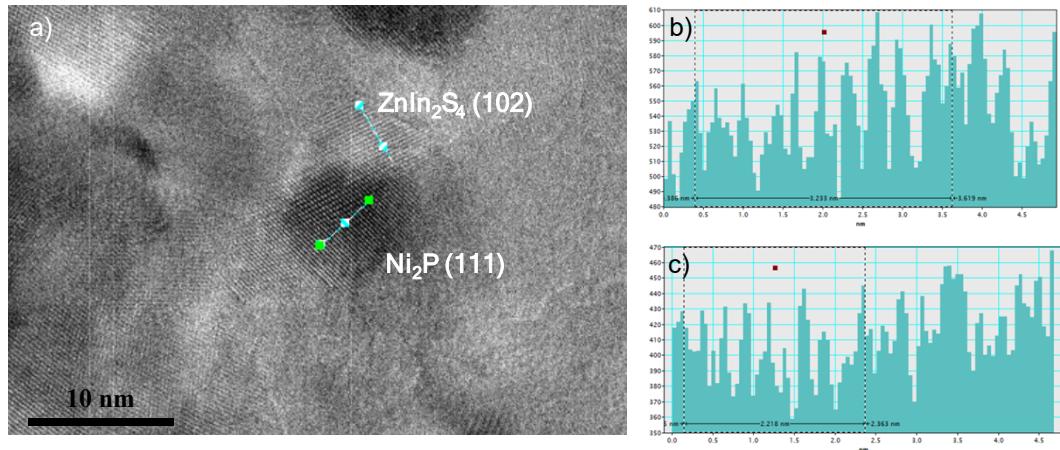


Fig. S3 a) high-resolution TEM image of $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$; b-c) intensity profile of ZnIn_2S_4 and Ni_2P crystal face.

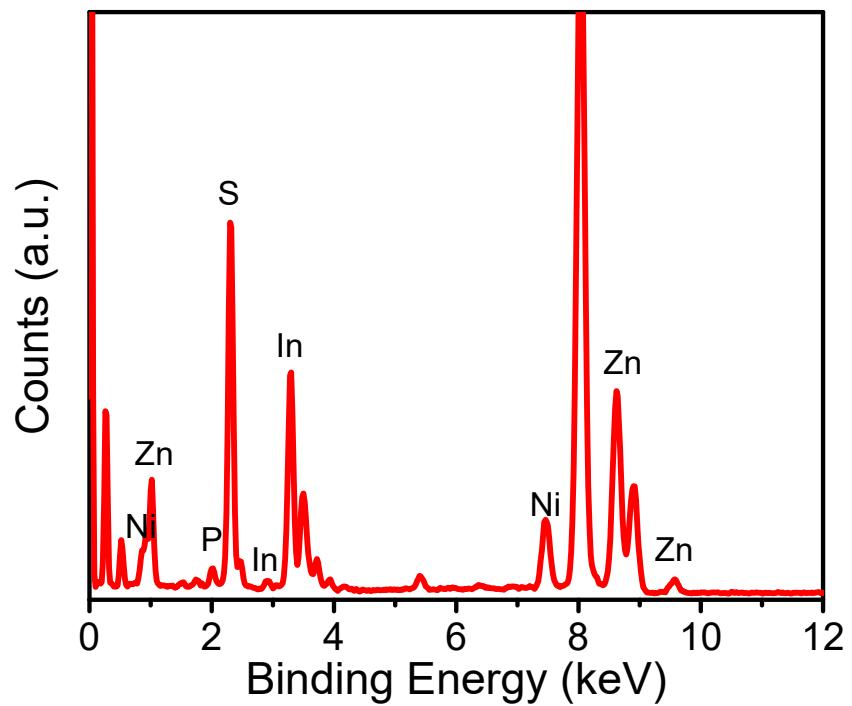


Fig. S4 EDX spectrum of the $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ composites.

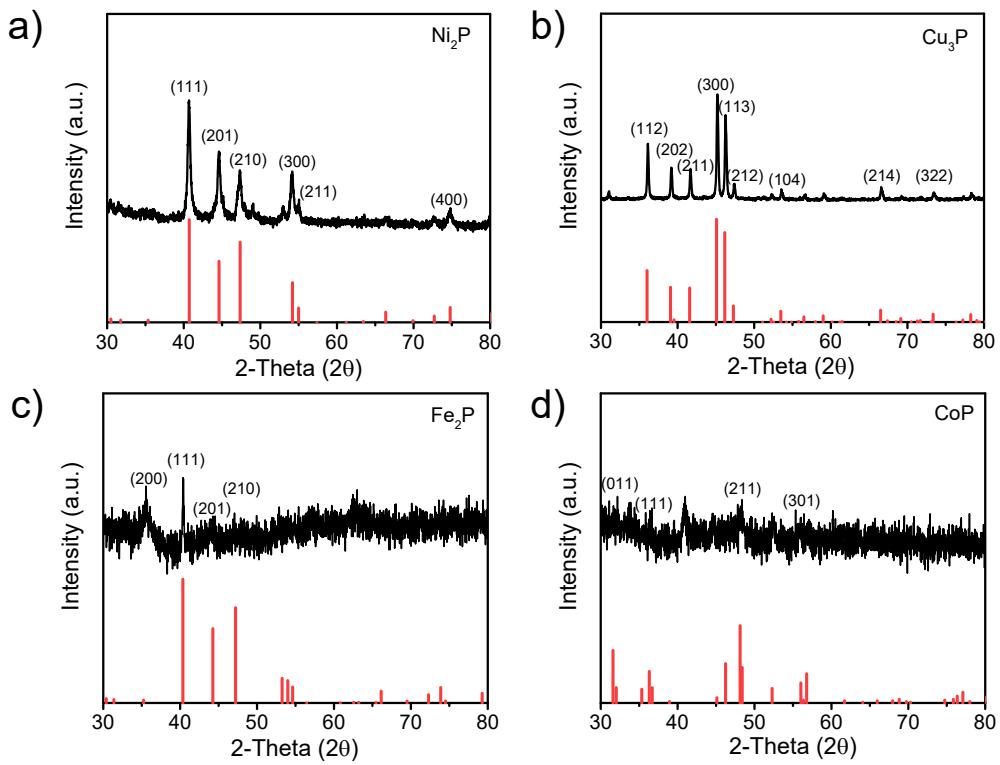


Fig. S5 The XRD patterns of a) Ni₂P; b) Cu₃P; c) Fe₂P; d) CoP samples.

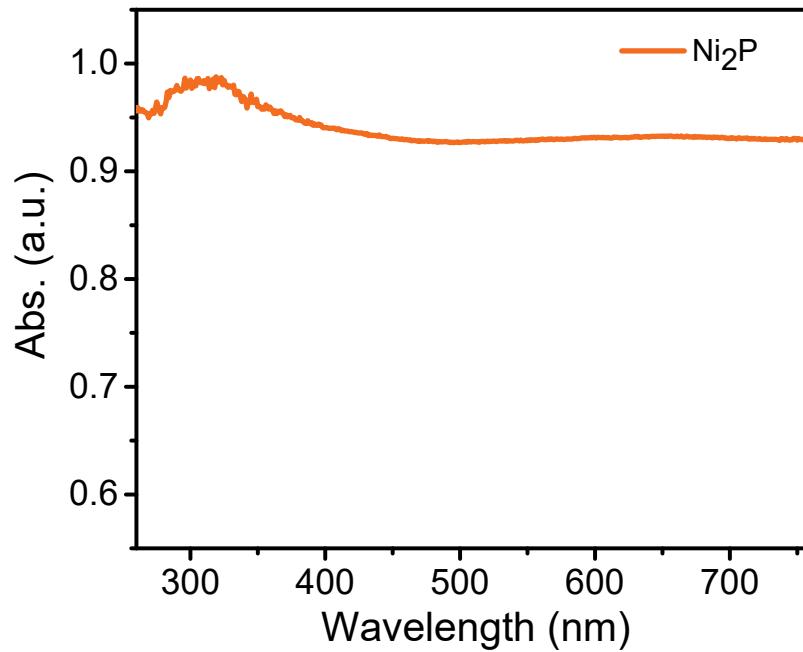


Fig. S6 UV-vis DRS spectrum of Ni₂P nanoparticles.

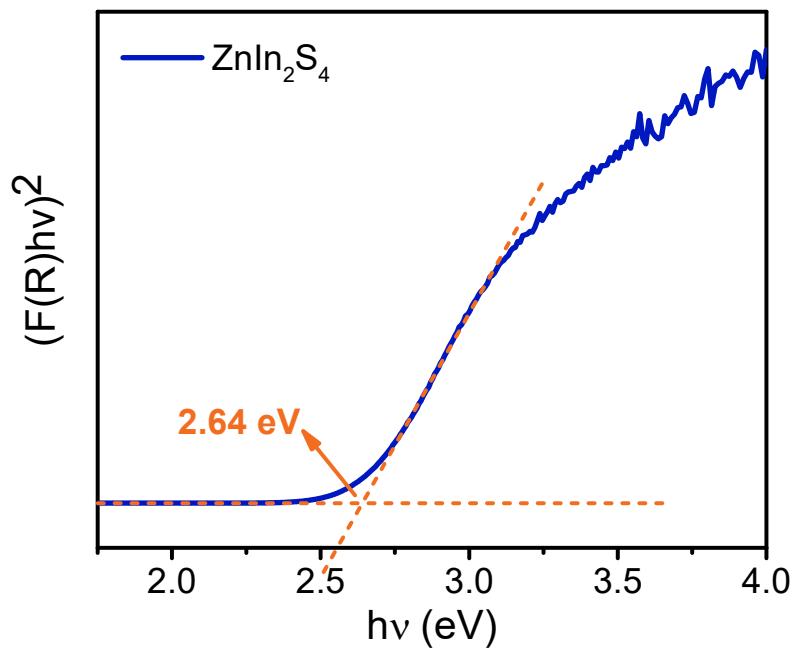


Fig. S7 Curve of $(hv)^2$ versus the energy of the exciting light ($h\nu$) of $ZnIn_2S_4$.

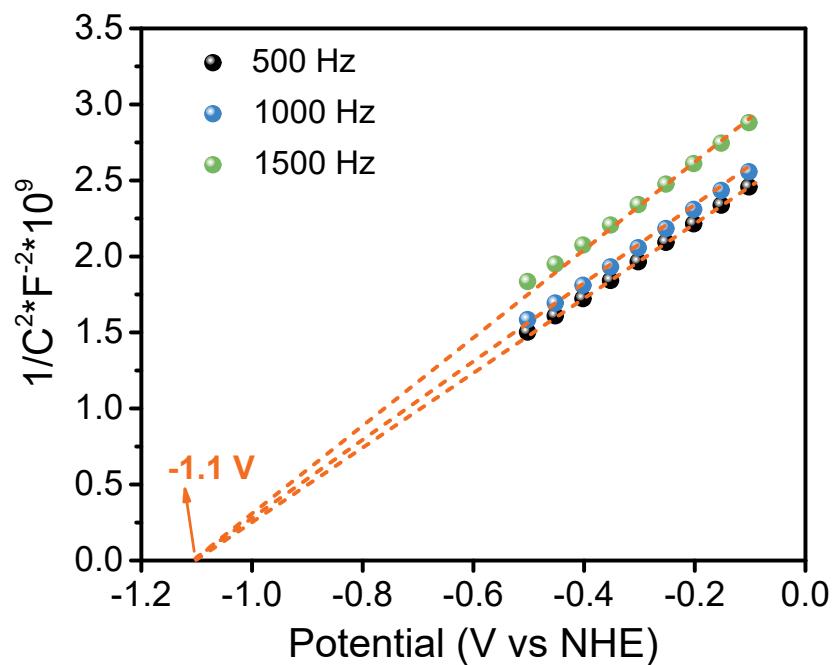


Fig. S8 Mott-Schottky plots of $ZnIn_2S_4$.

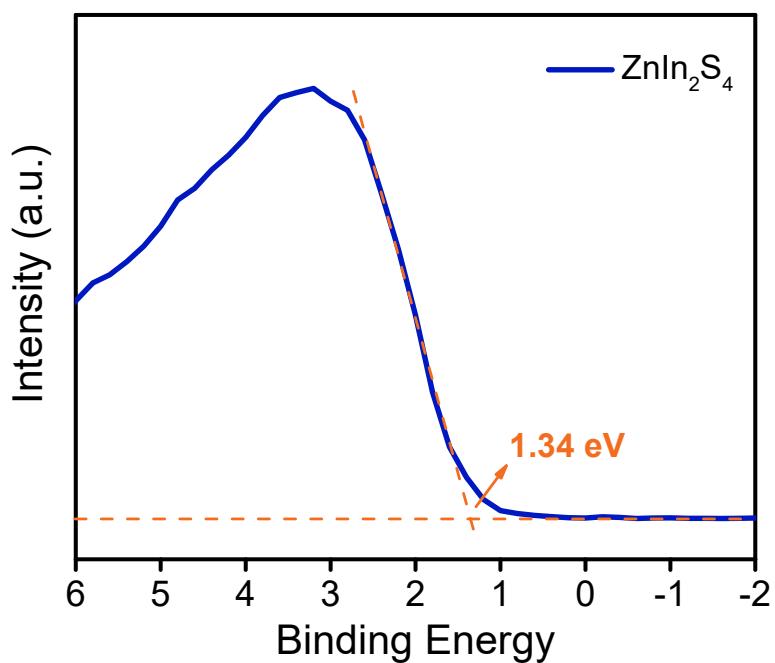


Fig. S9 The valence band spectrum of ZnIn_2S_4 .

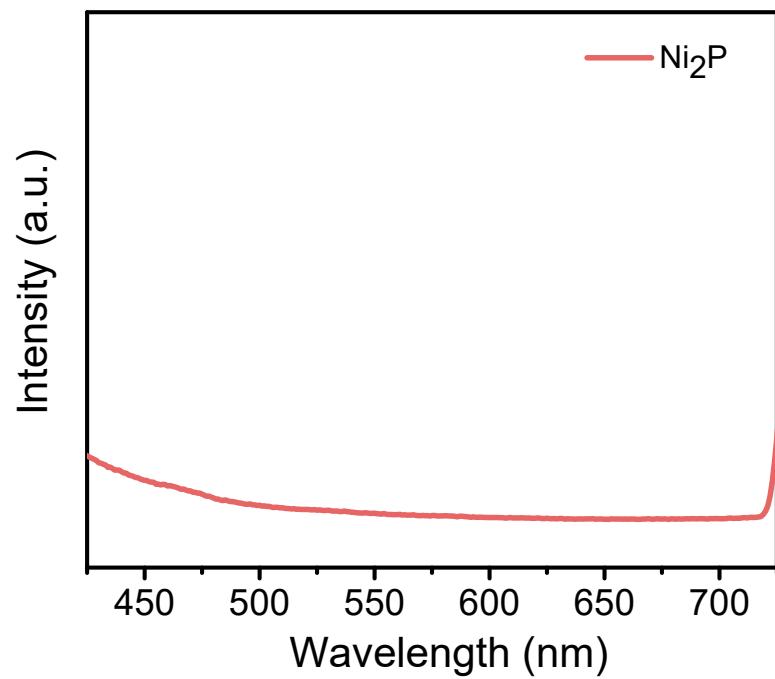


Fig. S10 PL spectrum of Ni_2P nanoparticles.

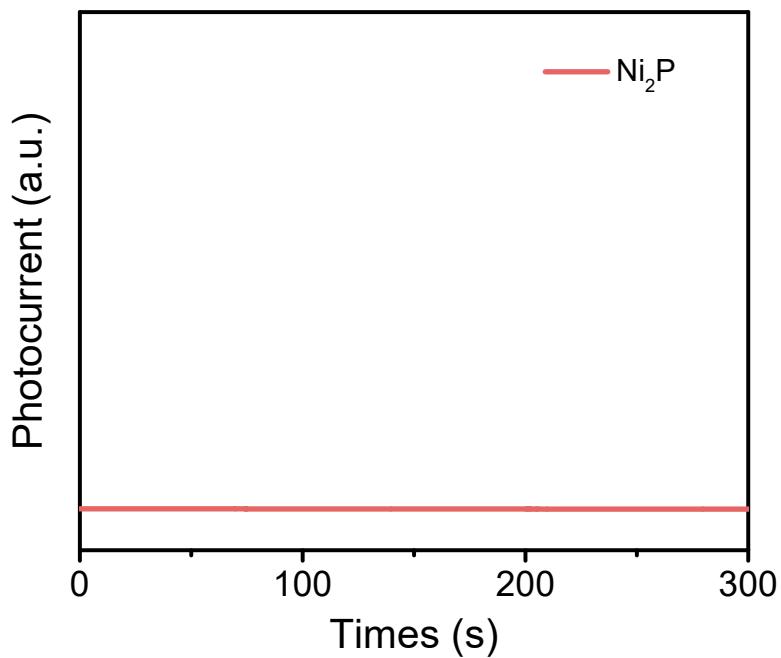


Fig. S11 The photocurrent response of Ni_2P nanoparticles.

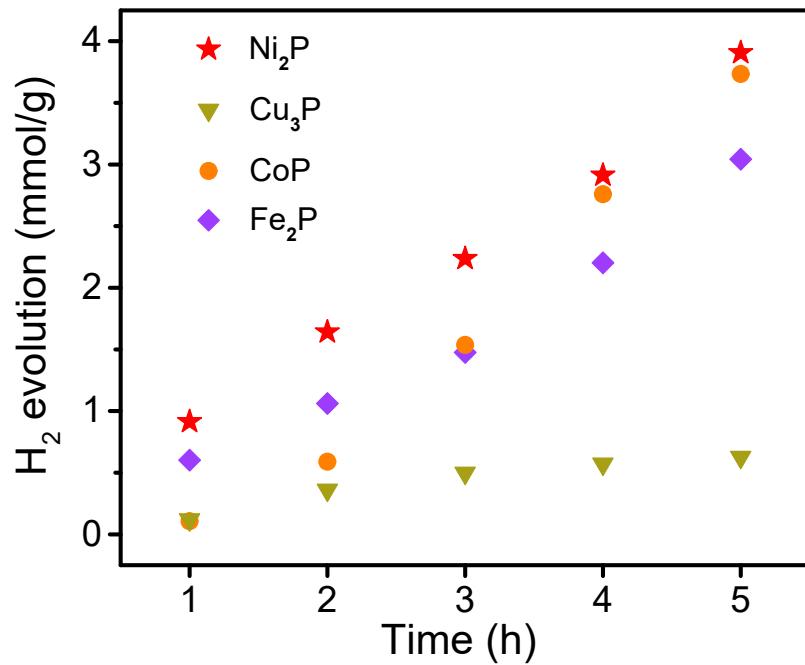


Fig. S12 Time-dependent H_2 evolution plots of $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$, $\text{CoP}/\text{ZnIn}_2\text{S}_4$, $\text{Fe}_2\text{P}/\text{ZnIn}_2\text{S}_4$, $\text{Cu}_3\text{P}/\text{ZnIn}_2\text{S}_4$ under visible light irradiation ($\lambda > 420$ nm) in 0.1 M PLA hydrolysate.

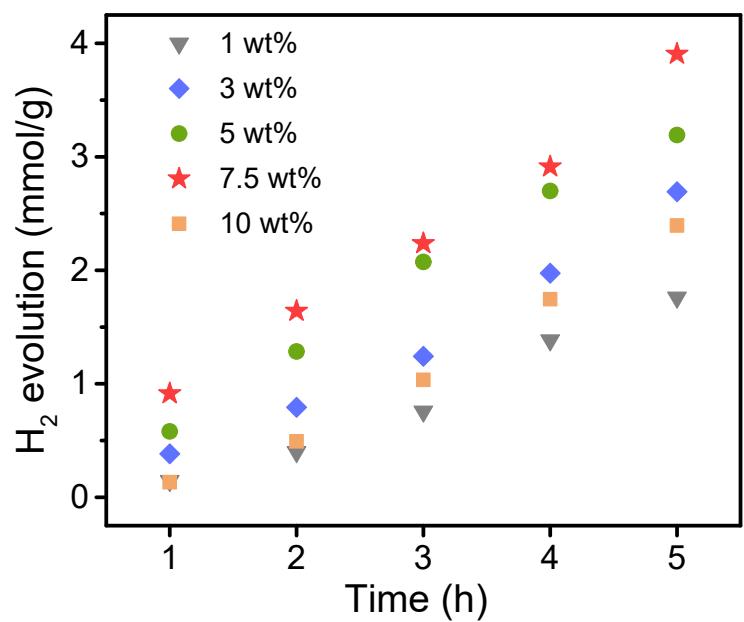


Fig. S13 Time-dependent H_2 evolution plots of $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ with different mass ratio under visible light irradiation ($\lambda > 420 \text{ nm}$) in 0.1 M PLA hydrolysate.

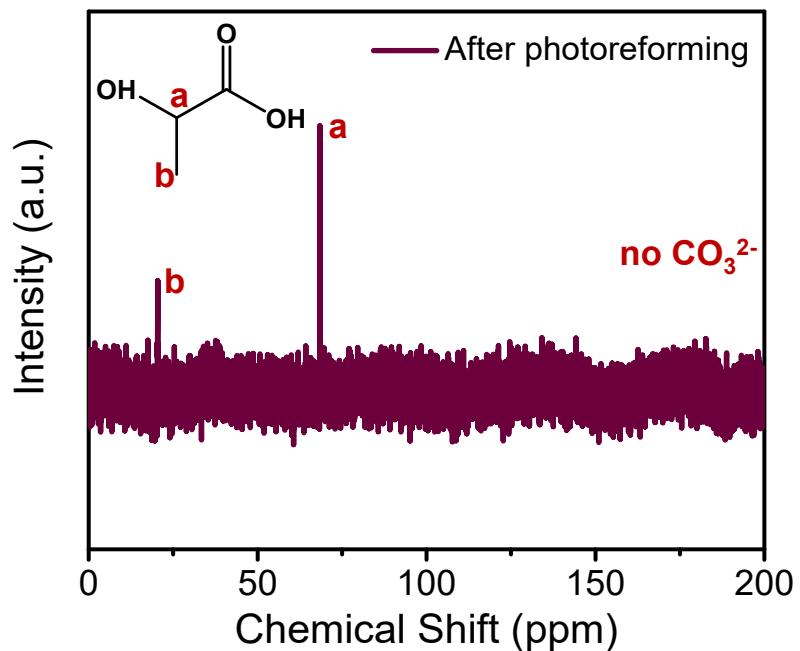


Fig. S14 ^{13}C NMR of pretreated PLA after photoreforming for 24h.

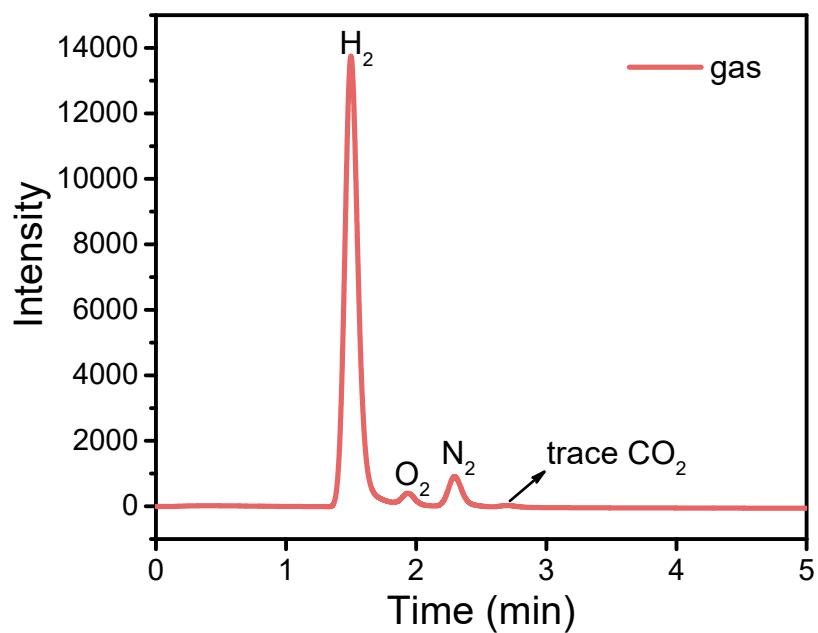


Fig. S15 GC spectrum of the generated gas after photoreforming.

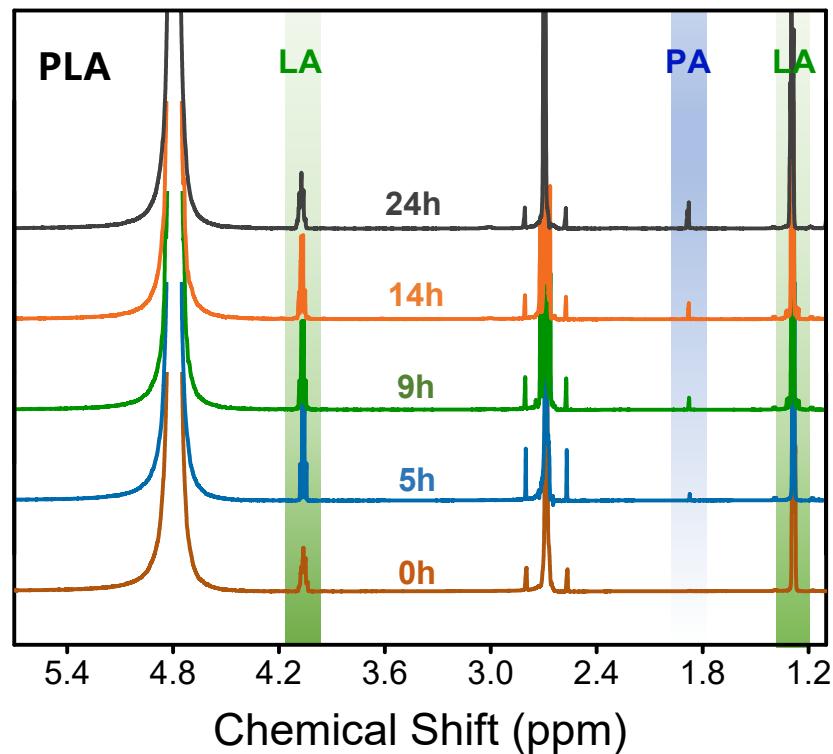


Fig. S16 ¹H NMR measurements of PLA oxidation.

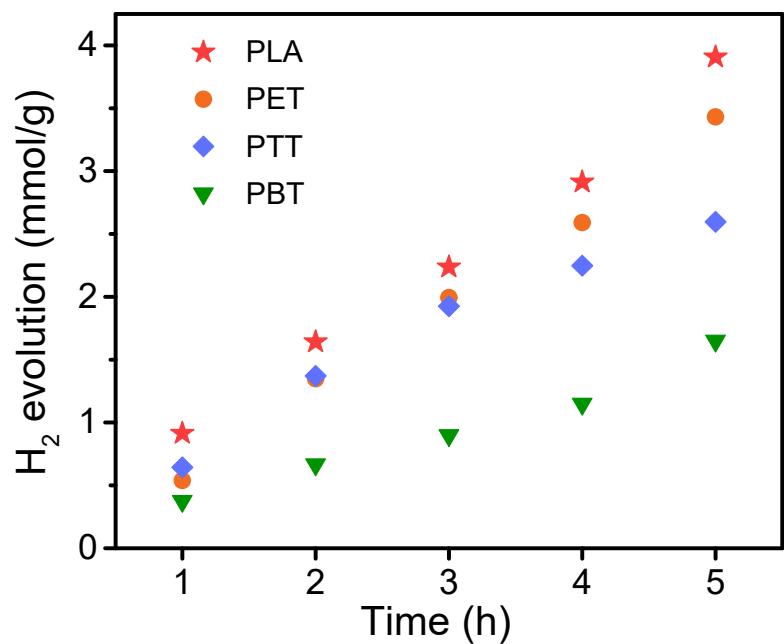


Fig. S17 Time-dependent H_2 evolution plots of 7.5-Ni₂P/ZnIn₂S₄ under visible light irradiation ($\lambda > 420$ nm) in 0.1 M different plastic hydrolysates.

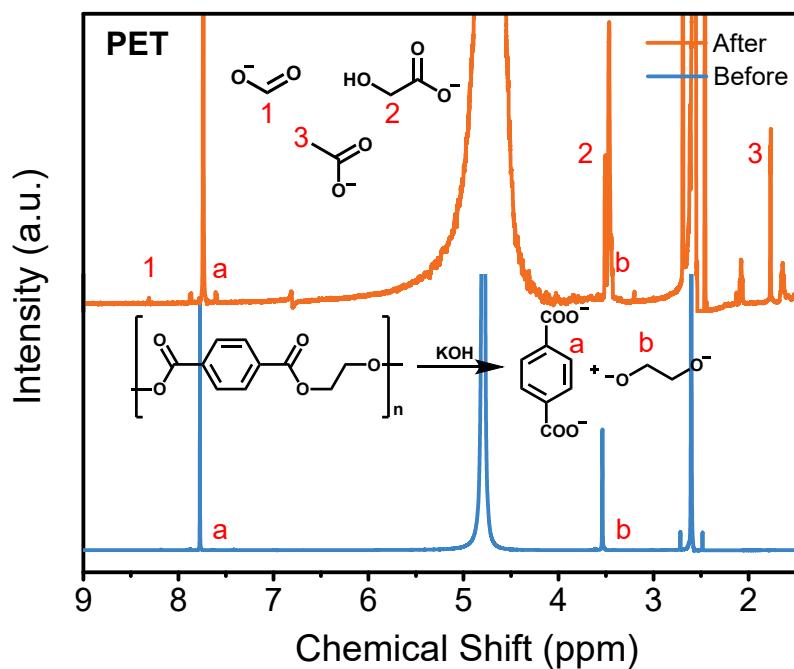


Fig. S18 ¹H NMR spectra of products before and after photoreforming of PET.

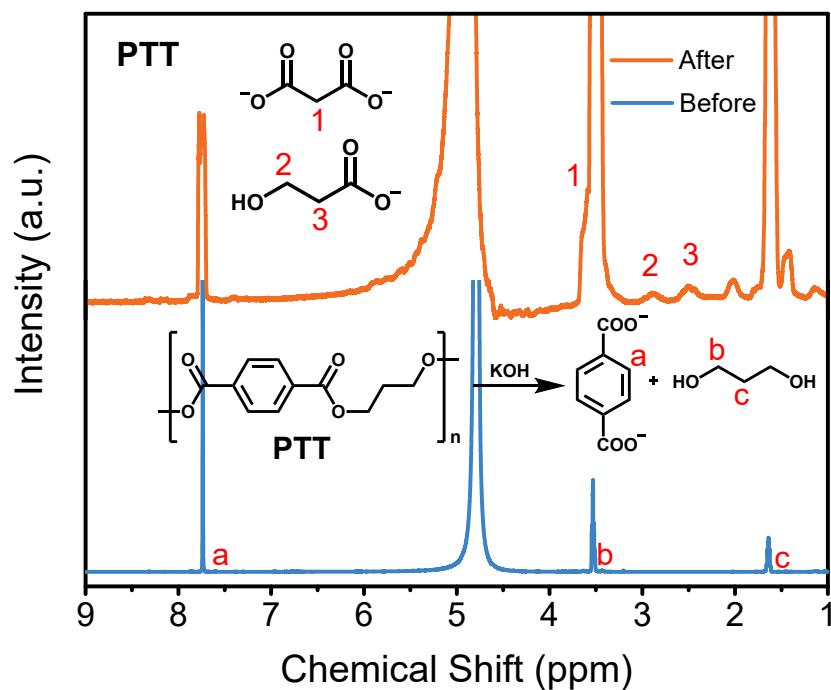


Fig. S19 ^1H NMR spectra of products before and after photoreforming of PTT.

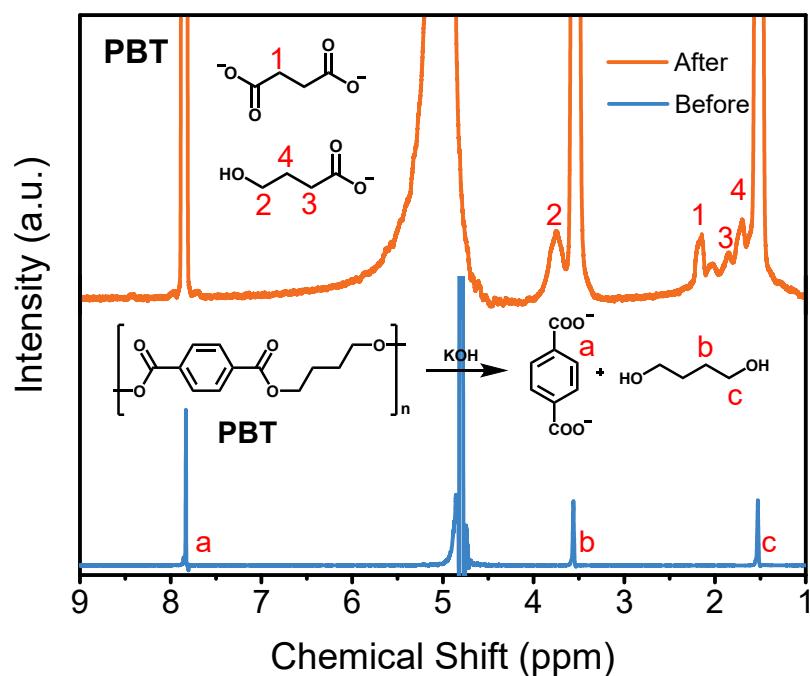


Fig. S20 ^1H NMR spectra of products before and after photoreforming of PBT.

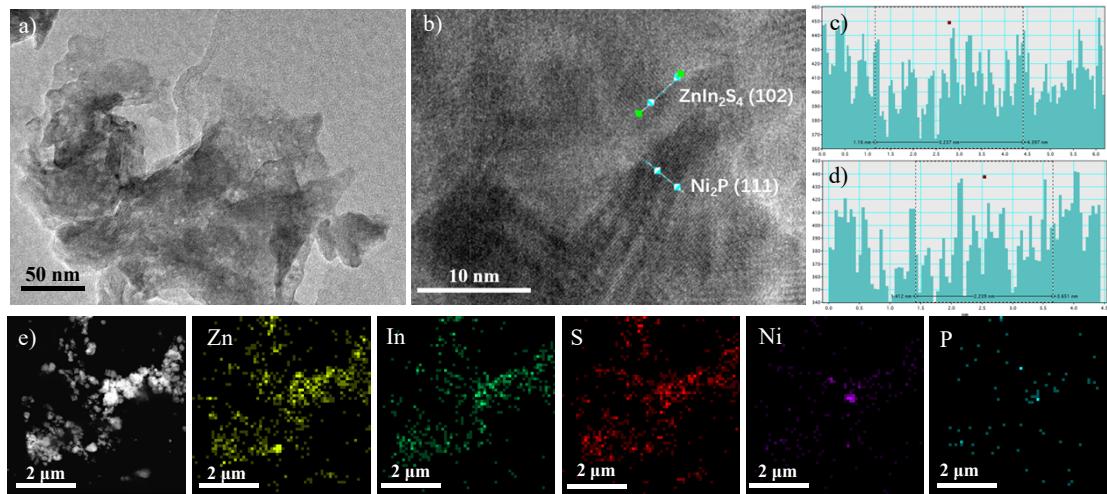


Fig. S21 The morphology of the samples after a long-term photoreforming.

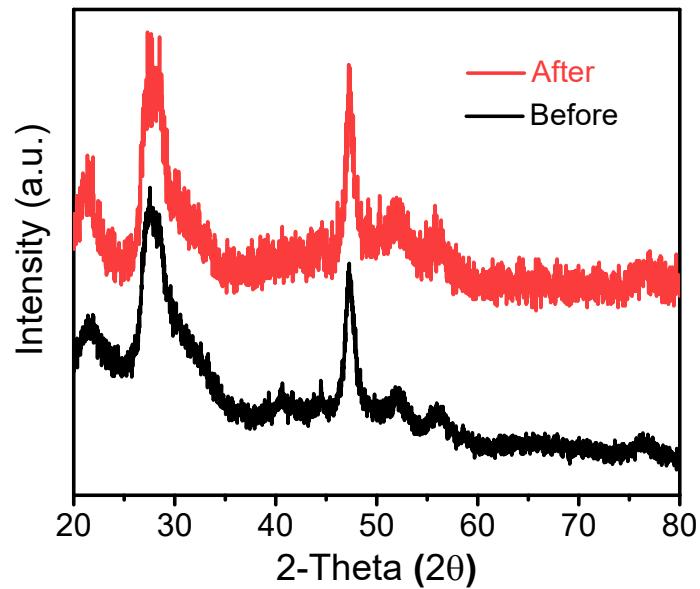


Fig. S22 XRD of $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ before and after photocatalytic reaction.

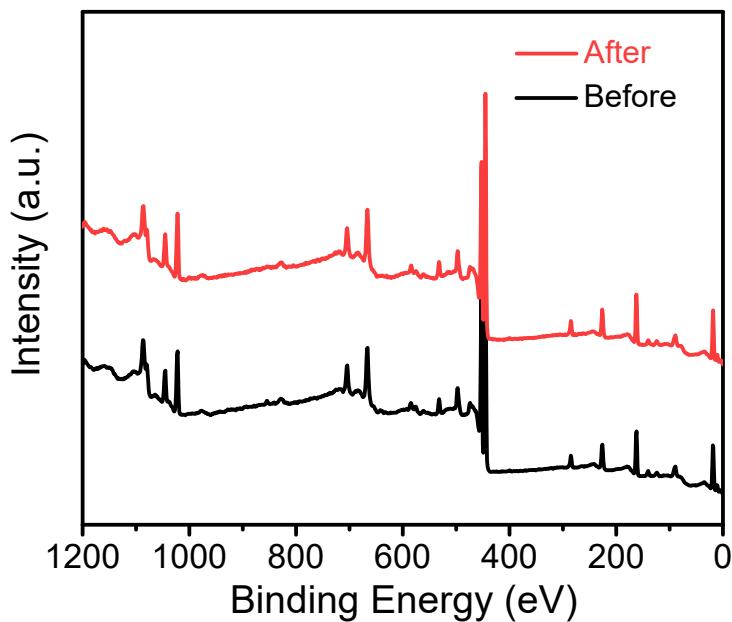


Fig. S23 XPS of $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ before and after photocatalytic reaction.

Table S1. EDX results of elements in the 7.5- $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ composites.

Element	Weight%	Atomic%
P K	1.13	2.27
S K	21.18	41.05
Ni K	7.87	8.33
Zn K	25.80	24.53
In L	44.02	23.82
Totals	100.00	

Table S2. ICP results of elements in the $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$ composites.

	1- $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$	3- $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$	5- $\text{Ni}_2\text{P}/\text{ZnIn}_2\text{S}_4$
Zn (wt%)	13.471	13.818	13.985
In (wt%)	55.743	53.405	51.850
Ni (wt%)	0.800	1.803	3.591