

Supporting Information

Guest-induced proton conductivity of two-dimensional layered hydrogen-bonded organic frameworks

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Table S1. Crystallographic data of BPPA-azo, BPPA-gua and BPPA^[1].

	BPPA-azo	BPPA-gua	BPPA
empirical formula	C ₁₄ H ₁₅ N ₃ O ₆ P ₂	C ₁₃ H ₁₇ N ₃ O ₆ P ₂	C ₁₂ H ₁₂ O ₆ P ₂
formula weight	383.28	373.24	314.16
temperature (K)	293	293	293
crystal system	triclinic	monoclinic	monoclinic
space group	<i>P</i> -1	<i>P</i> 2 ₁ /c	<i>P</i> 2 ₁ /c
<i>a</i> (Å)	9.5241(3)	16.0962(3)	13.0552(18)
<i>b</i> (Å)	10.4051(6)	14.3676(3)	7.0852(10)
<i>c</i> (Å)	17.8775(8)	6.77280(10)	6.7287(13)
α (deg)	77.976(4)	90	90
β (deg)	78.130(3)	97.203(2)	98.176(14)
γ (deg)	72.036(4)	90	90
volume (Å ³)	1629.15(14)	1553.94(5)	616.07(17)
<i>Z</i>	2	2	2
ρ calc (g/cm ³)	1.562	1.595	1.694
μ (mm ⁻¹)	2.792	2.905	0.377
<i>F</i> (000)	792.0	781.0	324.0
Data/params	6113/491	2923/253	1246/109
GOF on <i>F</i> ²	1.041	1.077	0.976
final <i>R</i>	<i>R</i> ₁ = 0.0366	<i>R</i> ₁ = 0.0379	<i>R</i> ₁ = 0.0358
indices [<i>I</i> > 2σ(<i>I</i>)]	<i>wR</i> ₂ = 0.1010	<i>wR</i> ₂ = 0.0941	<i>wR</i> ₂ = 0.0789

Table S2. The lengths (\AA) and angles ($^\circ$) of hydrogen bonds of BPPA.

Donor-H \cdots Acceptor	D-H	H \cdots A	D \cdots A	D-H \cdots A
O (1)-H (1) \cdots O (2)	0.83	1.72	2.546	170
O (3)- H (2) \cdots O (2)	0.83	1.74	2.568	170

Table S3. The lengths (\AA) and angles ($^\circ$) of hydrogen bonds of BPPA-azo (* denotes charge assisted hydrogen bonds between cationic species and phospholates).

Donor- H \cdots Acceptor	D-H	H \cdots A	D \cdots A	D-H \cdots A
*N(1)-H(1A) \cdots O(3)	0.94	1.76	2.681	165
O(2)-H(2A) \cdots O(12)	0.94	1.64	2.566	169
N(2)-H (2B) \cdots O(7)	0.94	1.74	2.639	160
*N(4)-H (4A) \cdots O(9)	0.94	1.78	2.706	169
O(5)-H (5) \cdots O(3)	0.95	1.52	2.472	175
N(5)-H (5B) \cdots O(4)	0.93	1.78	2.690	165
O(6)-H (6) \cdots O(7)	0.94	1.62	2.550	168
O(8)-H (8) \cdots O (4)	0.94	1.65	2.591	173
O(10)-H(10) \cdots O(1)	0.94	1.56	2.483	170
O(11)-H(11) \cdots O(9)	0.94	1.62	2.554	176

Table S4. The lengths (\AA) and angles ($^\circ$) of hydrogen bonds of BPPA-gua (* denotes charge assisted hydrogen bonds between cationic species and phospholates).

Donor-H \cdots Acceptor	D-H	H \cdots A	D \cdots A	D-H \cdots A
*N(1)-H(1A) \cdots O(1)	0.91	2.16	2.979	150
N(1)-H(1B) \cdots O(4)	0.92	2.06	2.968	171
*N(2)-H(2A) \cdots O(2)	0.91	2.00	2.910	173
N(2)-H(2B) \cdots O(5)	0.91	2.05	2.936	163
O(3)-H(3) \cdots O(1)	0.86	1.71	2.569	177
*N(3)-H(3B) \cdots O(2)	0.91	2.33	2.978	128
*N(3)-H(3C) \cdots O(1)	0.91	2.15	2.967	148
O(5)-H(5) \cdots O(2)	0.86	1.59	2.453	175
O(6)-H(6) \cdots O(4)	0.86	1.63	2.494	173

Table S5. Humidity-dependent proton conductivity (S cm^{-1}) of three HOFs.

Conditions	BPPA-azo	BPPA-gua	BPPA
80°C, 60%RH	6.88×10^{-8}	4.31×10^{-8}	1.35×10^{-3}
80°C, 70%RH	2.87×10^{-7}	2.21×10^{-7}	2.34×10^{-3}
80°C, 80%RH	5.63×10^{-7}	5.32×10^{-7}	5.50×10^{-3}
80°C, 90%RH	1.85×10^{-6}	9.43×10^{-7}	2.35×10^{-2}
80°C, 95%RH	9.43×10^{-6}	3.21×10^{-6}	5.14×10^{-2}

Table S6. Temperature-dependent proton conductivity (S cm^{-1}) of three HOFs.

Conditions	BPPA-azo	BPPA-gua	BPPA
40°C, 95%RH	2.01×10^{-6}	4.06×10^{-7}	2.83×10^{-2}
50°C, 95%RH	3.09×10^{-6}	5.02×10^{-7}	3.33×10^{-2}
60°C, 95%RH	4.27×10^{-6}	5.99×10^{-7}	3.90×10^{-2}
70°C, 95%RH	6.38×10^{-6}	7.66×10^{-7}	4.72×10^{-2}
80°C, 95%RH	9.59×10^{-6}	3.05×10^{-6}	5.02×10^{-2}

Table S7. Parameters of equivalent circuit model for HOFs (80 °C, 95% RH).

	R1 (Ω)	C1 (F)	R2 (Ω)	W-R (Ω)	W-T	W-P
BPPA	52	0.05	125	48	18	0.23
BPPA-azo	119	0.21	724580	1023	43	0.61
BPPA-gua	98	0.18	638690	4524	88	0.32

Table S8. The proton conductivities (S cm^{-1}) of reported HOF materials.

	Compound Name	Conductivity	Conditions	References
1	FCF-1 ^[2]	1.2×10^{-1}	100 °C, 98% RH	<i>ACS Appl. Mater. Interfaces</i> 2019
2	UPC-H3 ^[3]	9.0×10^{-2}	80 °C, 99% RH	<i>Cryst. Growth Des.</i> 2020
3	HOF-H4TCPB ^[4]	5.9×10^{-2}	90 °C, 100% RH	<i>J. Membr. Sci.</i> 2022
4	BPPA	5.1×10^{-2}	80 °C, 95% RH	<i>This work</i>
5	BIP ^[5]	3.2×10^{-2}	95 °C, 95% RH	<i>J. Am. Chem. Soc.</i> 2019
6	HOF-GS-11 ^[6]	1.8×10^{-2}	30 °C, 95% RH	<i>Angew. Chem. Int. Ed.</i> 2016
7	HOF-GS-10 ^[6]	7.5×10^{-3}	85 °C, 90% RH	<i>Angew. Chem. Int. Ed.</i> 2016
8	CPOS-2 ^[7]	3.7×10^{-3}	100 °C, 98% RH	<i>Angew. Chem. Int. Ed.</i> 2018
9	CB[6]·1.2H ₂ SO ₄ ·6.4H ₂ O ^[8]	1.3×10^{-3}	25 °C, 98% RH	<i>Angew. Chem. Int. Ed.</i> 2011
10	(H ₁₂ RCC1) ¹²⁺ 12Cl ⁻ ·4H ₂ O ^[9]	1.1×10^{-3}	30 °C, 95% RH	<i>Nat. Commun.</i> 2016
11	CPOS-1 ^[7]	6.0×10^{-4}	30 °C, 98% RH	<i>Angew. Chem. Int. Ed.</i> 2018
12	HOF 1 ^[10]	4.8×10^{-4}	100 °C, 98% RH	<i>New J. Chem.</i> 2023
13	MA-B-BDC ^[11]	4.3×10^{-4}	50 °C, 98% RH	<i>Cryst. Growth Des.</i> 2021
14	CPOS-3 ^[7]	3.7×10^{-4}	30 °C, 98% RH	<i>Angew. Chem. Int. Ed.</i> 2018
15	MA-TMA ^[11]	3.1×10^{-4}	70 °C, 98% RH	<i>Cryst. Growth Des.</i> 2021
16	HOF-H ₃ L ^[12]	6.9×10^{-5}	30 °C, 98% RH	<i>New J. Chem.</i> 2019
17	CPOS-4 ^[7]	5.6×10^{-5}	30 °C, 98% RH	<i>Angew. Chem. Int. Ed.</i> 2018
18	BPPA-azo	9.6×10^{-6}	80 °C, 95% RH	<i>This work</i>
19	HOF 6 ^[13]	3.4×10^{-6}	27 °C, 97% RH	<i>Cryst. Growth Des.</i> 2016
20	BPPA-gua	3.2×10^{-6}	80 °C, 95% RH	<i>This work</i>

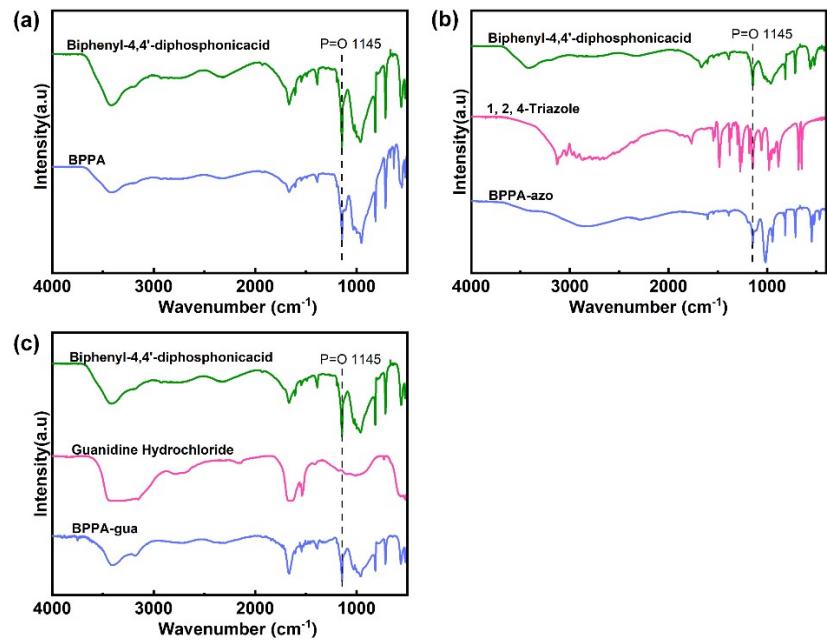


Figure S1. FT-IR spectra of biphenyl-4,4-bisphosphonic acid, 1,2,4-triazole and BPPA
(a), biphenyl-4,4-bisphosphonic acid, guanidine hydrochloride and BPPA-azo (b);
biphenyl-4,4-bisphosphonic acid and BPPA-gua (c).

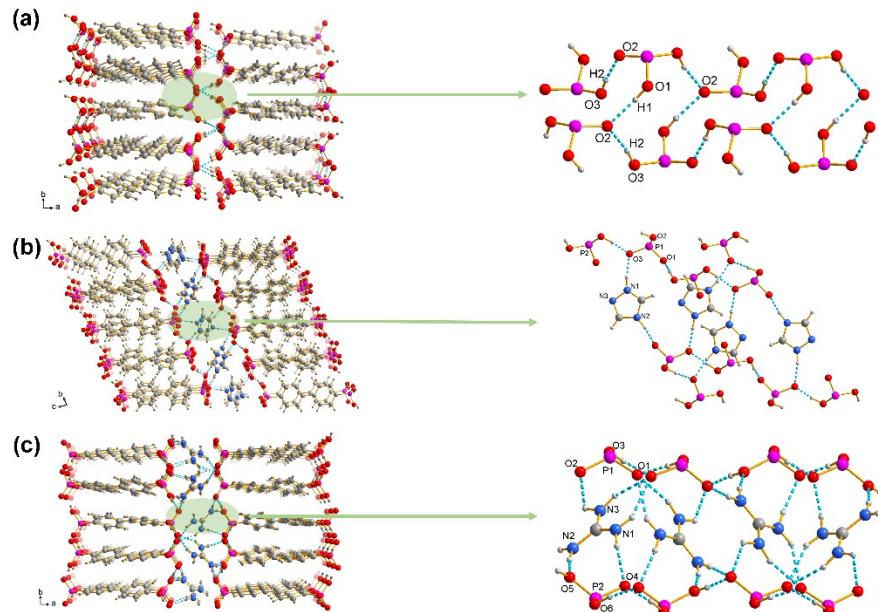


Figure S2. The stacking diagrams and hydrogen bonding networks of (a) BPPA, (b)
BPPA-azo and (c) BPPA-gua.

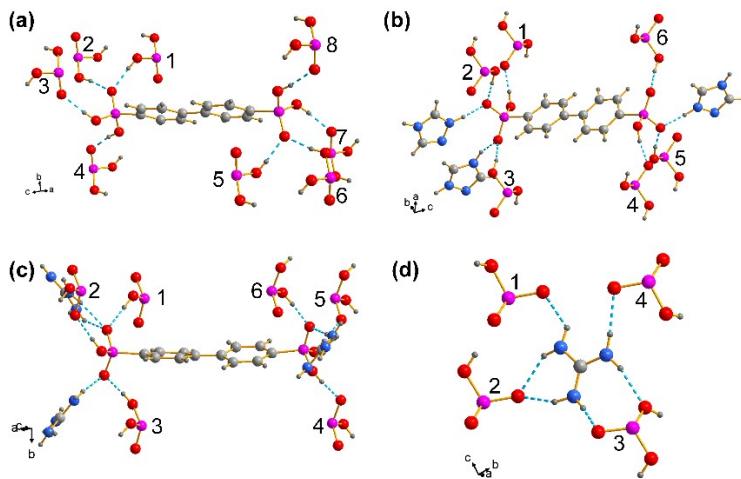


Figure S3. The hydrogen bonds of H_4BPPA molecules in BPPA (a), H_3BPPA^- anions in BPPA-azo (b), H_3BPPA^- anions in BPPA-gua (c), and HGUA^+ cations in BPPA-gua (d).

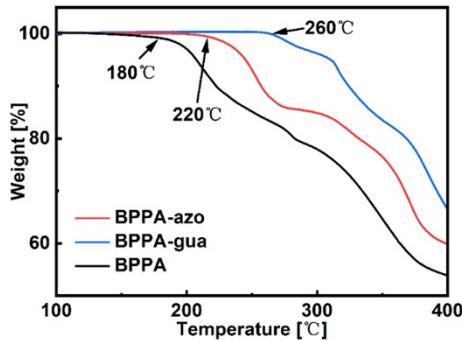


Figure S4. TGA curves of BPPA, BPPA-azo and BPPA-gua.

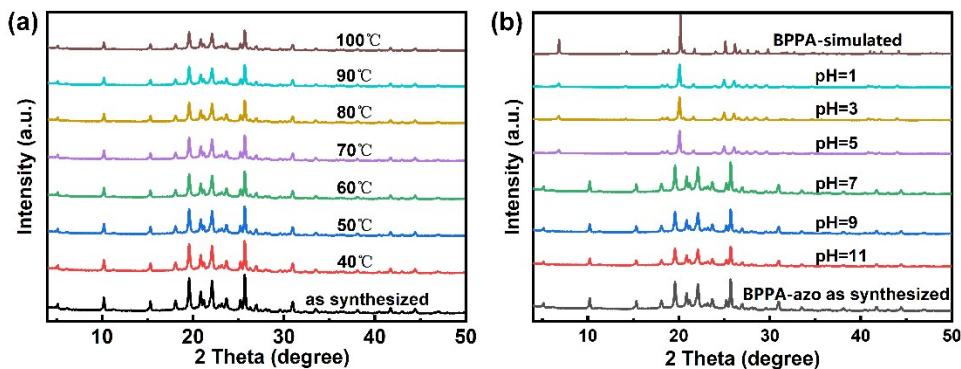


Figure S5. PXRD patterns of BPPA-azo (a) after immersing in solutions of different pH values for 3 d, (b) after immersing in aqueous solutions at different temperatures for 3 d.

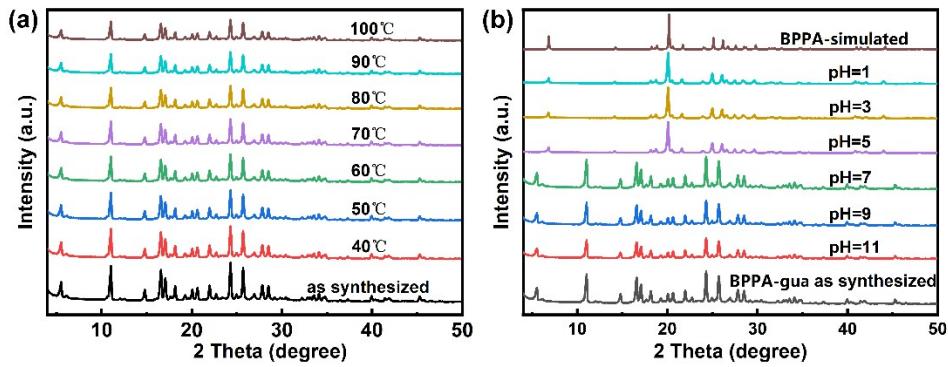


Figure S6. PXRD patterns of BPPA-gua (a) after immersing in solutions of different pH values for 3 d, (b) after immersing in aqueous solutions at different temperatures for 3 d.

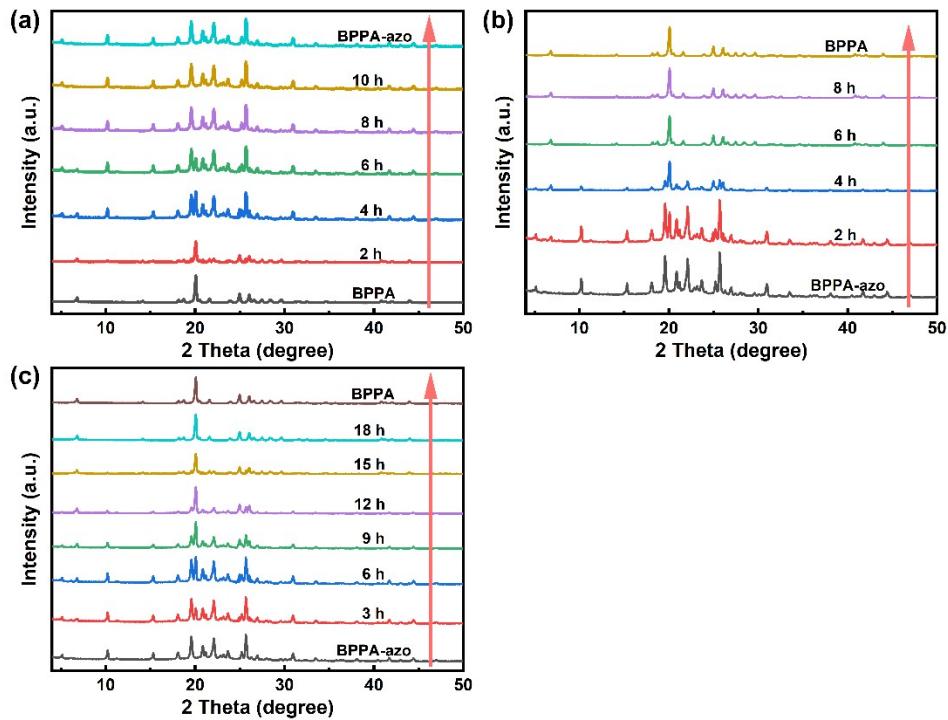


Figure S7. PXRD patterns showing the SCSC transformation time (a) from BPPA to BPPA-azo, (b) from BPPA-azo to BPPA at 100 °C; (c) from BPPA-azo to BPPA at room temperature.

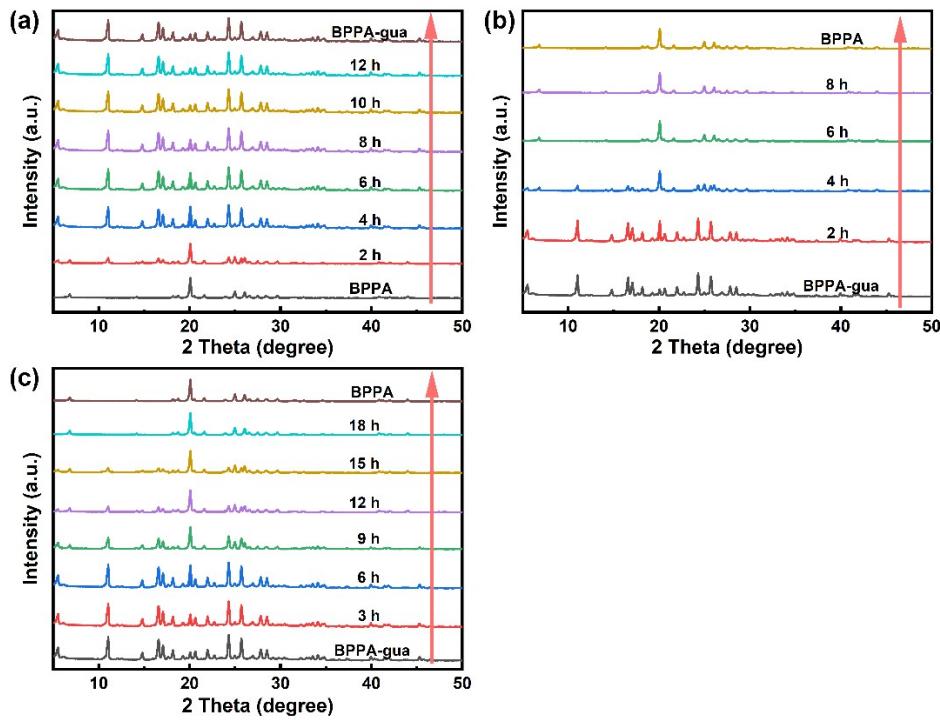


Figure S8. PXRD patterns showing the SCSC transformation time (a) from BPPA to BPPA-gua, (b) from BPPA-gua to BPPA at 100 °C; (c) from BPPA-gua to BPPA at room temperature.

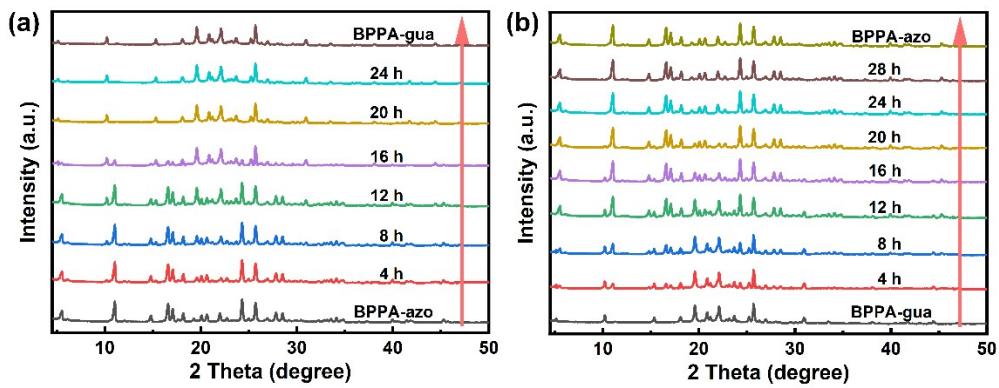


Figure S9. PXRD patterns showing the SCSC transformation time (a) from BPPA-azo to BPPA-gua, (b) from BPPA-gua to BPPA-azo.

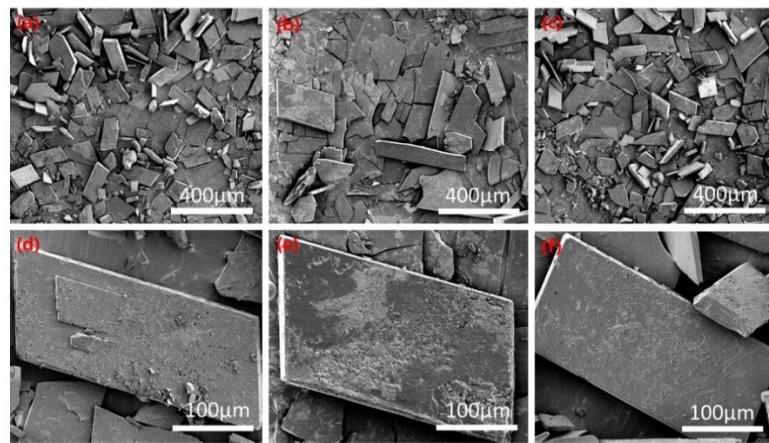


Figure S10. SEM images of (a) BPPA-azo, (b) BPPA-gua and (c) BPPA, Single crystal SEM images of (d) BPPA-azo, (e) BPPA-gua and (f) BPPA.

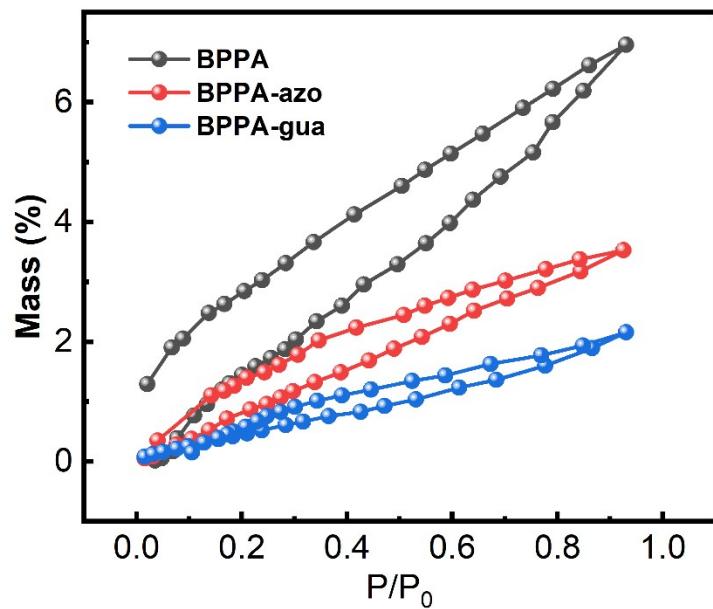


Figure S11. Water sorption isotherms of BPPA, BPPA-azo and BPPA-gua.

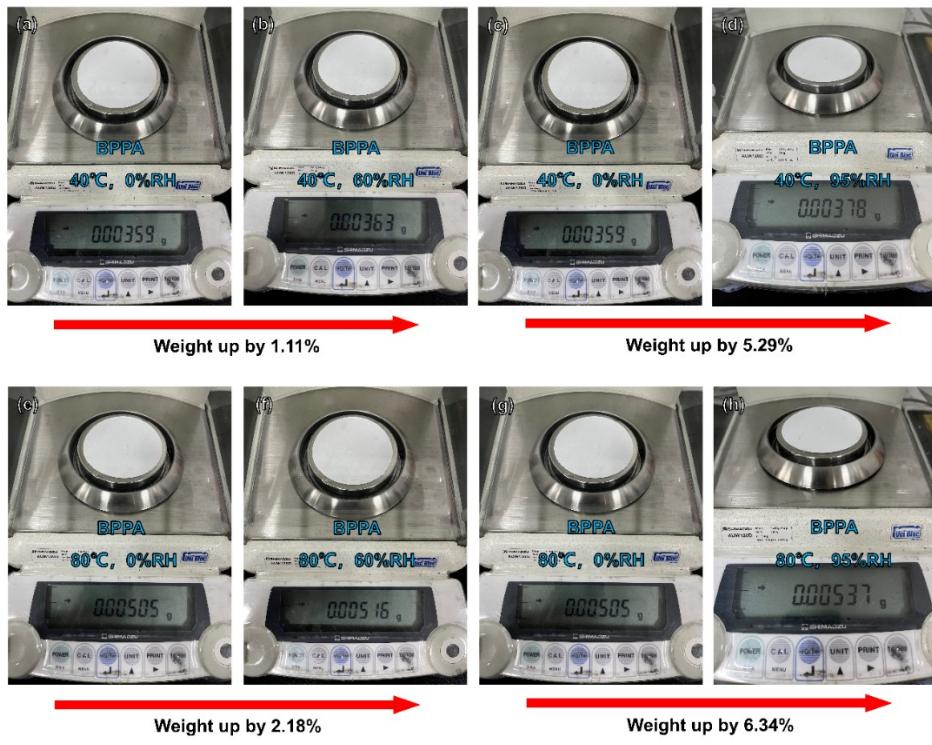


Figure S12. Quality of BPPA pellets at 40 °C, 0% R.H. (a); at 40 °C, 60% RH (b); at 40 °C, 0% RH (c); at 40 °C, 95% RH (d); at 80 °C, 0% RH (e); at 80 °C, 60% RH (f); at 80 °C, 0% RH (g); at 80 °C, 95% RH (h). Note: the quality at 0% RH was obtained by weighing immediately after taking the pellet out of the vacuum drying oven, and the quality at other humidity was obtained by weighing immediately after standing the pellets for 2 hours under the related humidity and temperature.

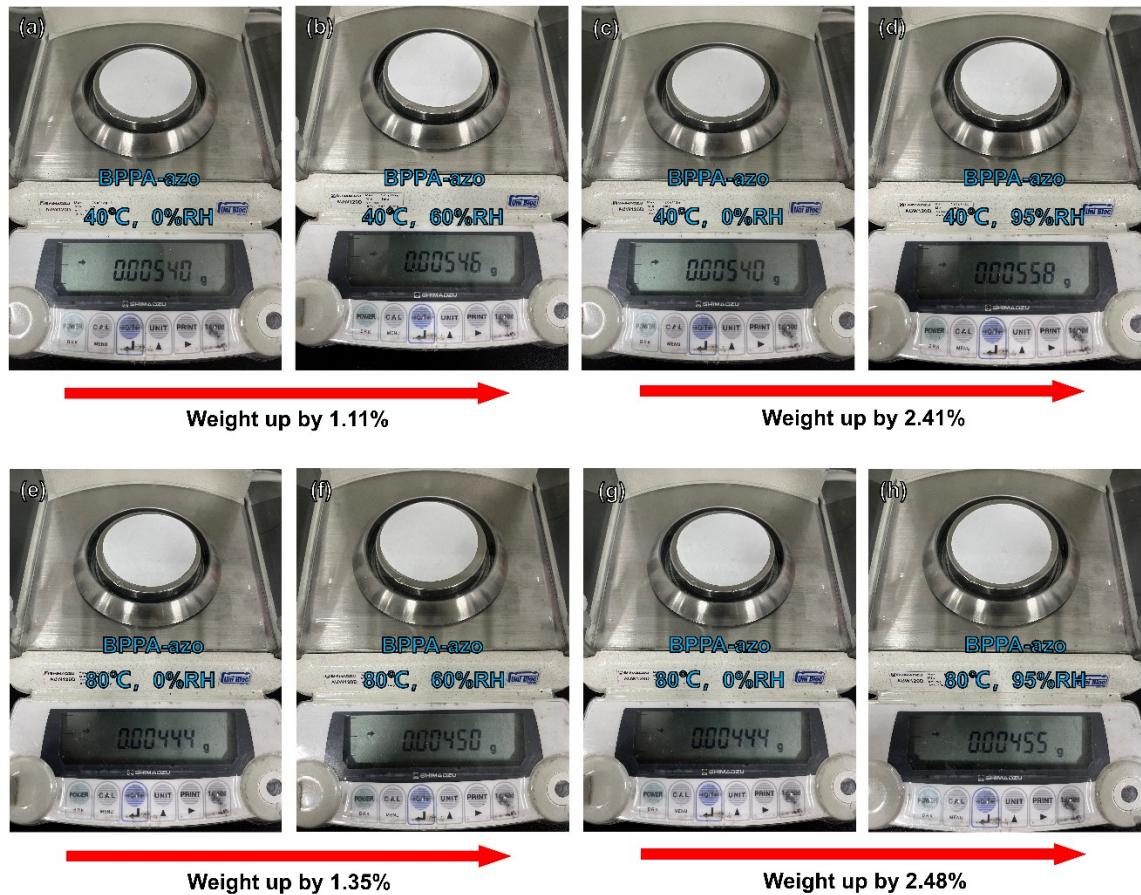


Figure S13. Quality of BPPA-azo pellets at 40 °C, 0% R.H. (a); at 40 °C, 60% RH (b); at 40 °C, 0% RH (c); at 40 °C, 95% RH (d); at 80 °C, 0% RH (e); at 80 °C, 60% RH (f); at 80 °C, 0% RH (g); at 80 °C, 95% RH (h). Note: the quality at 0% RH was obtained by weighing immediately after taking the pellet out of the vacuum drying oven, and the quality at other humidity was obtained by weighing immediately after standing the pellets for 2 hours under the related humidity and temperature.

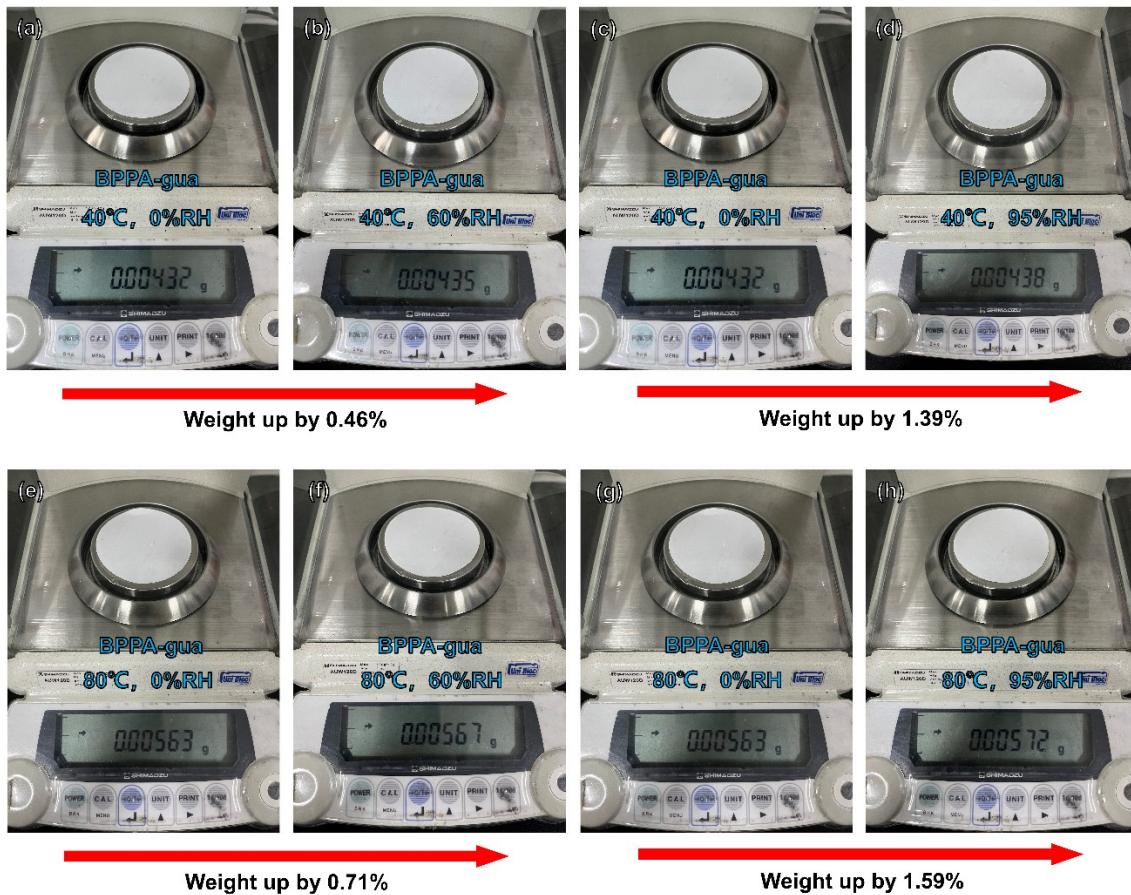


Figure S14. Quality of BPPA-gua pellets at 40 °C, 0% R.H. (a); at 40 °C, 60% RH (b); at 40 °C, 0% RH (c); at 40 °C, 95% RH (d); at 80 °C, 0% RH (e); at 80 °C, 60% RH (f); at 80 °C, 0% RH (g); at 80 °C, 95% RH (h). Note: the quality at 0% RH was obtained by weighing immediately after taking the pellet out of the vacuum drying oven, and the quality at other humidity was obtained by weighing immediately after standing the pellets for 2 hours under the related humidity and temperature.

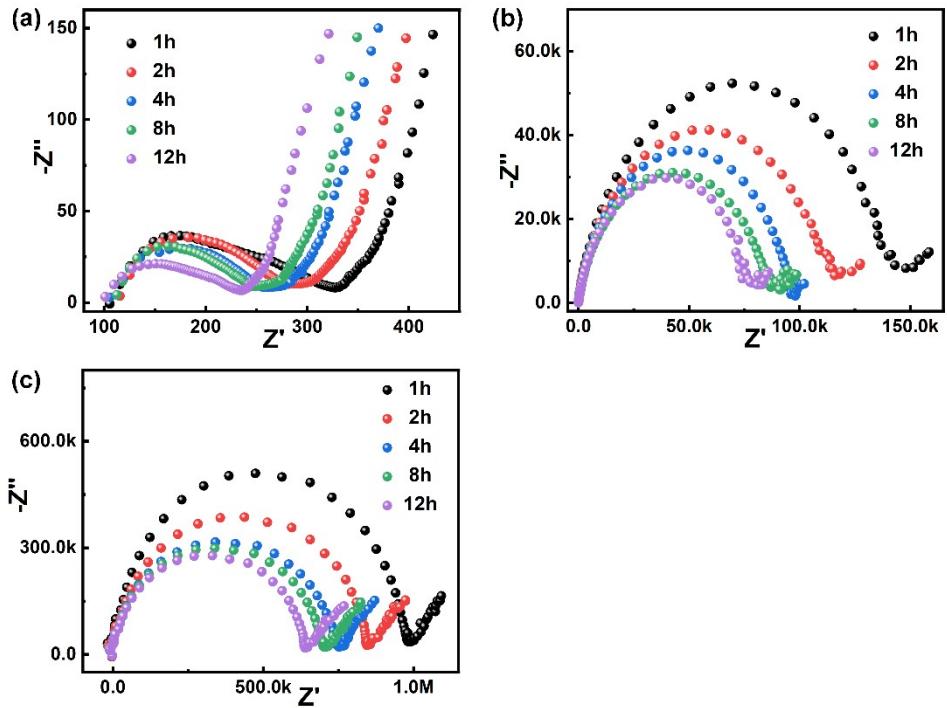


Figure S15. The time-dependent Nyquist plots of (a) BPPA, (b) BPPA-azo, (c) BPPA-gua at 80 °C and 95% RH.

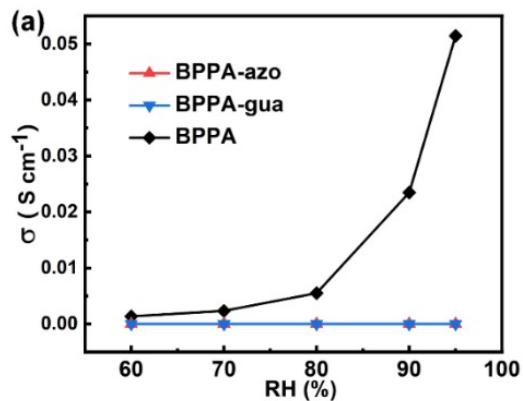


Figure S16. Proton conductivities of BPPA, BPPA-azo and BPPA-gua at 80 °C under different RH.

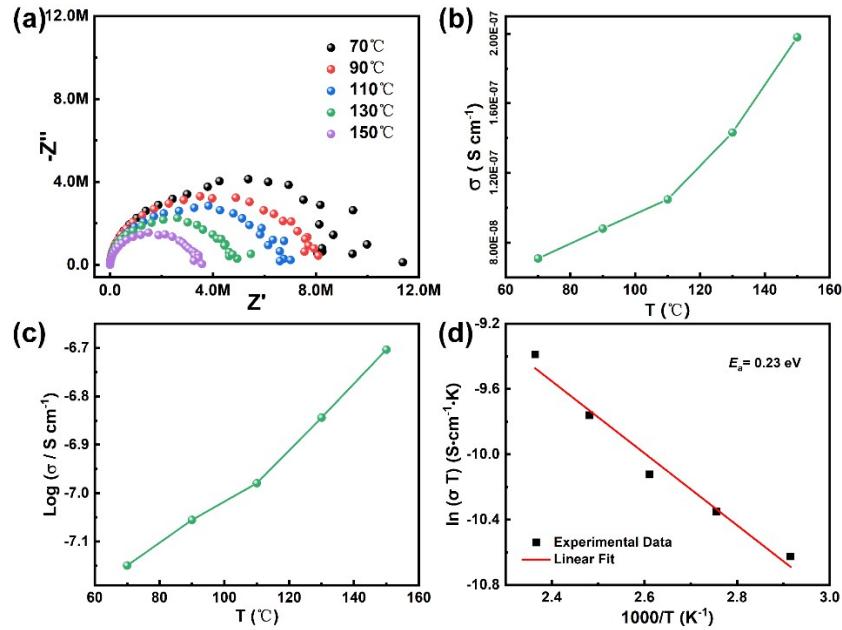


Figure S17. (a) Nyquist plots, (b) Proton conductivities, (c) Log-scaled proton conductivities, and (d) Arrhenius plots of BPPA at various temperatures under anhydrous conditions.

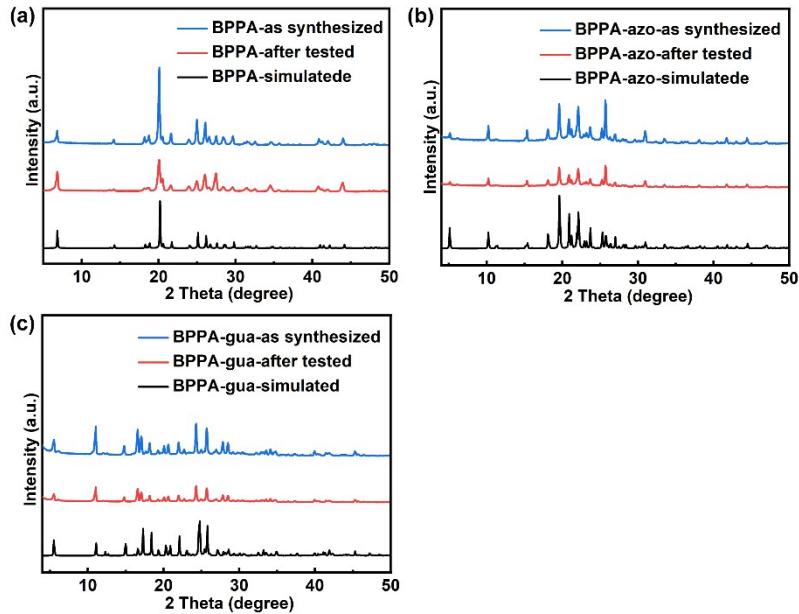


Figure S18. PXRD patterns of (a) BPPA (b) BPPA-azo and (c) BPPA-gua after multiple cyclic humidity-dependent impedance tests at 80 °C and 95% RH.

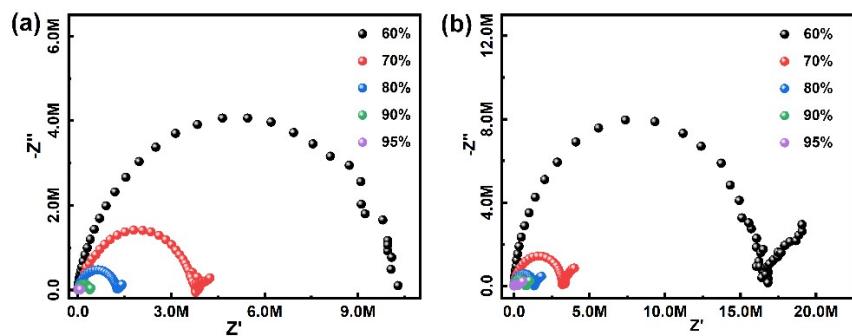


Figure S19. Nyquist plots of (a) BPPA-azo and (b) BPPA-gua under 95% RH at different temperatures.

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