

Electronic Supplementary Information (ESI)

Achieving Broadband Ultraviolet to Mid-Infrared Transparency in Germanate-based Nonlinear Optical Crystals Cs₃REGe₃O₉ (RE = Y, Gd)

Jinmiao Jiao,^a Conggang Li,^{*a} Yuheng She,^a Ning Ye,^a Zhanggui Hu,^{*a} and Yicheng Wu^a

^aTianjin Key Laboratory of Functional Crystal Materials, Institute of Functional Crystal, Tianjin University of Technology, Tianjin 300384, China.

*Corresponding author, E-mail address: cgli@email.tjut.edu.cn

1. **Fig. S1** Powder XRD patterns of the Cs₃YGe₃O₉ and Cs₃GdGe₃O₉ polycrystalline powder before and after melting at 1250 °C, respectively.
2. **Fig. S2** EDS of Cs₃YGe₃O₉ (a) atomic ratio of various elements, (b) surface morphology, (c-f) Cs, Ge, Y, and O mapping results, respectively.
3. **Fig. S3** EDS of Cs₃GdGe₃O₉ (a) atomic ratio of various elements, (b) surface morphology, (c-f) Cs, Ge, Gd, and O mapping results, respectively.
4. **Table S1** Selected bond lengths (Å) and angles (deg.) for Cs₃YGe₃O₉.
5. **Table S2** Selected bond lengths (Å) and angles (deg.) for and Cs₃GdGe₃O₉.
6. **Table S3** The final atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for Cs₃YGe₃O₉. U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor, and the bond valence sum for each atom in asymmetric unit.
7. **Table S4** The final atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{Å}^2 \times 10^3$) for Cs₃GdGe₃O₉. U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor, and the bond valence sum for each atom in asymmetric unit.
8. **Table S5** The detailed infrared peaks of CYGO and CGGO, respectively.

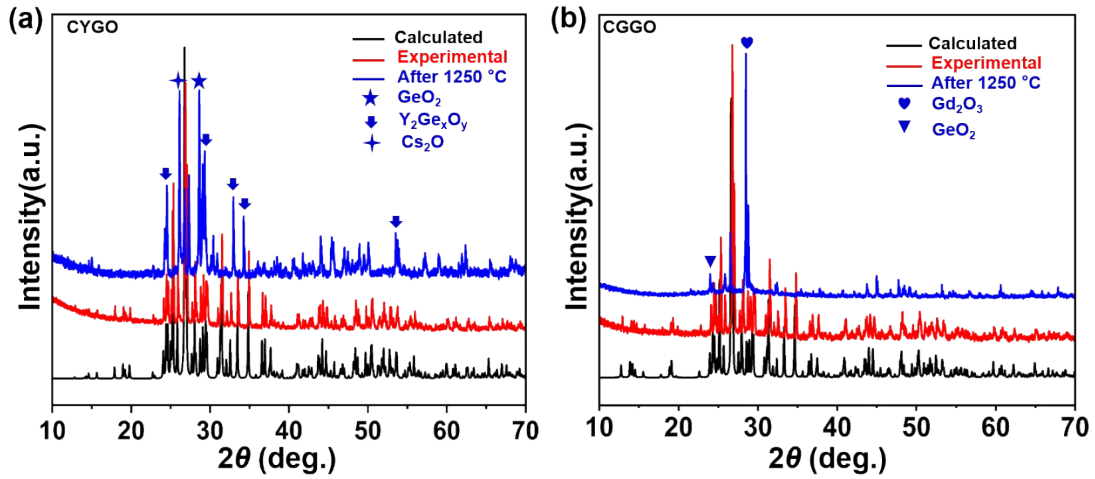


Fig. S1(a-b) Comparison of theoretical XRD spectra, experimentally obtained polycrystalline pure phase XRD spectra, and XRD spectra after calcination at 1250 °C for CYGO and CGGO, respectively.

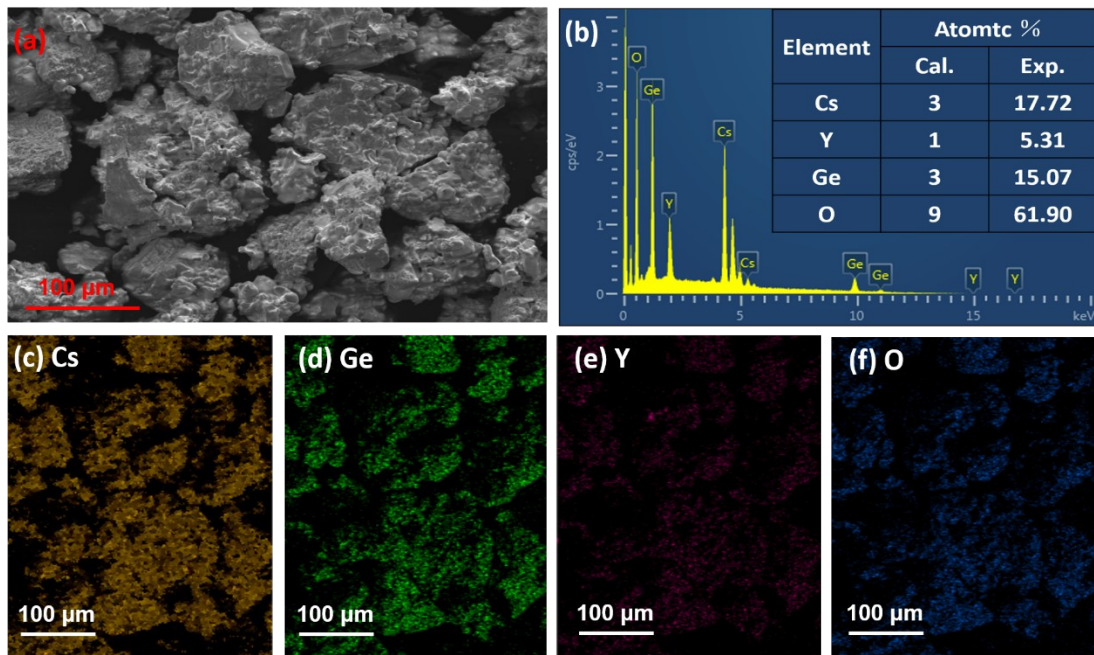


Fig. S2 EDS of $\text{Cs}_3\text{YGe}_3\text{O}_9$. (a) Atomic ratio of various elements, (b) Surface morphology, (c-f) Cs, Ge, Y, and O mapping results, respectively.

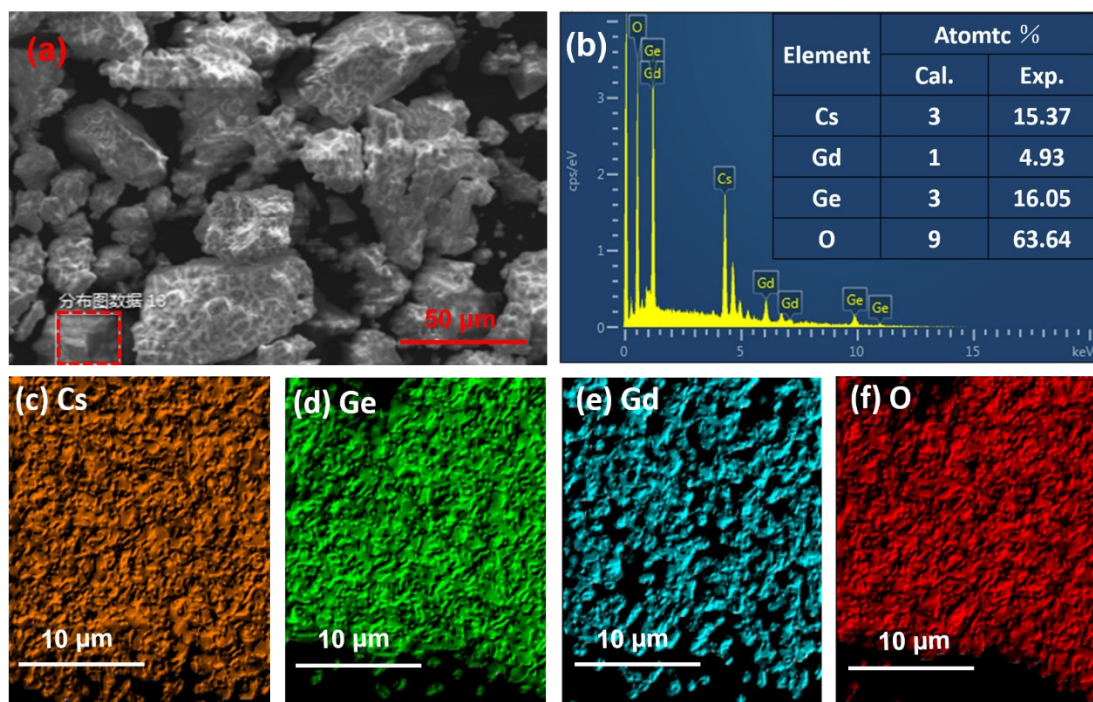


Fig. S3 EDS of $\text{Cs}_3\text{GdGe}_3\text{O}_9$. (a) Atomic ratio of various elements, (b) Surface morphology, (c-f) Cs, Ge, Gd, and O mapping results, respectively.

Table S1 Selected bond lengths (Å) and angles (deg.) for Cs₃YGe₃O₉.

Cs(1)-O(1)	2.993(6)	O(1)-Cs(3A)#9	3.004(19)
Cs(1)-O(5)#1	3.079(7)	O(4)-Cs(3A)#6	3.037(15)
Cs(1)-O(8)#2	3.098(7)	O(9)-Cs(3A)#9	3.064(16)
Cs(1)-O(4)#3	3.194(7)	O(3)-Cs(3A)#6	3.201(17)
Cs(1)-O(2)	3.381(7)	O(3)-Cs(3A)#9	3.53(3)
Cs(1)-O(9)#2	3.397(6)	O(7)-Cs(3A)	3.78(3)
Cs(2)-O(3)	3.041(6)	O(2)-Cs(3A)	2.855(17)
Cs(2)-O(8)#5	3.134(7)	Y(1)-O(8)#8	2.241(6)
Cs(2)-O(6)#2	3.146(7)	Y(1)-O(2)#3	2.244(7)
Cs(2)-O(5)#5	3.156(7)	Y(1)-O(3)#1	2.250(6)
Cs(2)-O(9)#2	3.230(7)	Y(1)-O(4)#3	2.259(6)
Cs(2)-O(7)#2	3.262(7)	Y(1)-O(1)	2.270(6)
Cs(3)-O(2)	2.925(9)	Y(1)-O(5)#1	2.281(6)
Cs(3)-O(4)#3	3.026(8)	Ge(1)-O(1)	1.721(6)
Cs(3)-O(1)#7	3.099(8)	Ge(1)-O(2)	1.722(7)
Cs(3)-O(9)#7	3.114(9)	Ge(1)-O(7)	1.786(6)
Cs(3)-O(3)#3	3.152(9)	Ge(1)-O(9)	1.789(6)
Cs(3)-O(7)	3.618(9)	Ge(3)-O(3)	1.712(6)
Cs(3)-O(3)#7	3.721(10)	Ge(3)-O(4)	1.723(6)
Ge(2)-O(8)	1.730(6)	Ge(3)-O(6)	1.763(6)
Ge(2)-O(5)	1.732(6)	Ge(3)-O(9)	1.774(6)
Ge(2)-O(7)#10	1.769(6)		
Ge(2)-O(6)	1.779(6)		
O(1)-Cs(1)-O(5)#1	61.80(15)	O(2)-Cs(3)-O(3)#7	56.22(19)
O(1)-Cs(1)-O(8)#2	103.52(15)	O(4)#3-Cs(3)-O(3)#7	127.1(2)
O(5)#1-Cs(1)-O(8)#2	157.69(18)	O(1)#7-Cs(3)-O(3)#7	81.4(2)
O(1)-Cs(1)-O(4)#3	60.96(16)	O(9)#7-Cs(3)-O(3)#7	47.97(17)
O(5)#1-Cs(1)-O(4)#3	60.90(17)	O(3)#3-Cs(3)-O(3)#7	138.58(17)

O(8)#2-Cs(1)-O(4)#3	129.15(17)	O(7)-Cs(3)-O(3)#7	99.51(18)
O(1)-Cs(1)-O(2)	53.96(15)	O(8)#8-Y(1)-O(2)#3	88.3(3)
O(5)#1-Cs(1)-O(2)	114.64(15)	O(8)#8-Y(1)-O(3)#1	86.7(2)
O(8)#2-Cs(1)-O(2)	57.48(15)	O(2)#3-Y(1)-O(3)#1	91.1(3)
O(4)#3-Cs(1)-O(2)	78.30(16)	O(8)#8-Y(1)-O(4)#3	97.6(2)
O(1)-Cs(1)-O(9)#2	114.69(16)	O(2)#3-Y(1)-O(4)#3	87.0(2)
O(5)#1-Cs(1)-O(9)#2	83.15(15)	O(3)#1-Y(1)-O(4)#3	175.3(2)
O(8)#2-Cs(1)-O(9)#2	89.02(15)	O(8)#8-Y(1)-O(1)	86.6(2)
O(4)#3-Cs(1)-O(9)#2	141.80(16)	O(2)#3-Y(1)-O(1)	172.2(2)
O(2)-Cs(1)-O(9)#2	132.69(16)	O(3)#1-Y(1)-O(1)	94.4(2)
O(3)-Cs(2)-O(8)#5	59.87(16)	O(4)#3-Y(1)-O(1)	87.9(2)
O(3)-Cs(2)-O(6)#2	115.89(17)	O(8)#8-Y(1)-O(5)#1	170.3(3)
O(8)#5-Cs(2)-O(6)#2	131.97(19)	O(2)#3-Y(1)-O(5)#1	99.2(3)
O(3)-Cs(2)-O(5)#5	103.39(16)	O(3)#1-Y(1)-O(5)#1	87.1(2)
O(8)#5-Cs(2)-O(5)#5	54.51(15)	O(4)#3-Y(1)-O(5)#1	88.9(2)
O(6)#2-Cs(2)-O(5)#5	85.26(19)	O(1)-Y(1)-O(5)#1	86.5(2)
O(3)-Cs(2)-O(9)#2	159.22(16)	O(1)-Ge(1)-O(2)	115.5(3)
O(8)#5-Cs(2)-O(9)#2	140.23(16)	O(1)-Ge(1)-O(7)	112.7(3)
O(6)#2-Cs(2)-O(9)#2	49.80(16)	O(2)-Ge(1)-O(7)	107.7(3)
O(5)#5-Cs(2)-O(9)#2	91.14(15)	O(1)-Ge(1)-O(9)	108.7(3)
O(3)-Cs(2)-O(7)#2	137.93(17)	O(2)-Ge(1)-O(9)	111.5(3)
O(8)#5-Cs(2)-O(7)#2	112.52(19)	O(7)-Ge(1)-O(9)	99.7(3)
O(6)#2-Cs(2)-O(7)#2	99.29(17)	O(8)-Ge(2)-O(5)	112.6(3)
O(5)#5-Cs(2)-O(7)#2	101.70(17)	O(8)-Ge(2)-O(7)#10	112.4(3)
O(9)#2-Cs(2)-O(7)#2	49.76(16)	O(5)-Ge(2)-O(7)#10	111.3(3)
O(2)-Cs(3)-O(4)#3	88.5(2)	O(8)-Ge(2)-O(6)	108.4(3)
O(2)-Cs(3)-O(1)#7	135.9(3)	O(5)-Ge(2)-O(6)	113.5(3)
O(4)#3-Cs(3)-O(1)#7	111.3(2)	O(7)#10-Ge(2)-O(6)	97.7(3)
O(2)-Cs(3)-O(9)#7	97.1(2)	O(3)-Ge(3)-O(4)	113.3(3)

O(4)#3-Cs(3)-O(9)#7	163.6(2)	O(3)-Ge(3)-O(6)	114.3(3)
O(1)#7-Cs(3)-O(9)#7	54.66(18)	O(4)-Ge(3)-O(6)	107.1(3)
O(2)-Cs(3)-O(3)#3	143.1(2)	O(3)-Ge(3)-O(9)	108.7(3)
O(4)#3-Cs(3)-O(3)#3	55.30(18)	O(4)-Ge(3)-O(9)	114.0(3)
O(1)#7-Cs(3)-O(3)#3	64.11(18)	O(6)-Ge(3)-O(9)	98.8(3)
O(9)#7-Cs(3)-O(3)#3	115.9(2)	O(1)#7-Cs(3)-O(7)	166.7(2)
O(2)-Cs(3)-O(7)	49.94(18)	O(9)#7-Cs(3)-O(7)	116.50(19)
O(4)#3-Cs(3)-O(7)	78.9(2)	O(3)#3-Cs(3)-O(7)	119.4(2)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,z-1/2	#2 x+1/2,-y+1/2,z	#3 -x+1,-y+1,z-1/2
#4 x+1/2,-y+3/2,z	#5 x+1/2,-y-1/2,z	#6 -x+1,-y+1,z+1/2
#7 x,y+1,z	#8 -x+1/2,y+1/2,z-1/2	#9 x,y-1,z
#10 -x+1/2,y-1/2,z+1/2	#11 x-1/2,-y+3/2,z	#12 -x+3/2,y+1/2,z-1/2
#13 x-1/2,-y+1/2,z	#14 x-1/2,-y-1/2,z	#15 -x+1,-y,z+1/2

Table S2 Selected bond lengths (Å) and angles (deg.) for Cs₃GdGe₃O₉.

Cs(1)-O(4)	2.91(2)	Cs(2A)-O(9)#10	2.69(3)
Cs(1)-O(2)#7	3.08(2)	Cs(2A)-O(6)#3	2.97(3)
Cs(1)-O(9)#8	3.13(2)	Cs(2A)-O(3)	2.97(3)
Cs(1)-O(3)#8	3.17(2)	Cs(2A)-O(8)	3.25(3)
Cs(1)-O(1)#7	3.23(2)	Cs(2A)-O(7)#3	3.27(4)
Cs(1)-O(5)	3.61(2)	Cs(2A)-O(4)#10	3.42(3)
Cs(1)-O(1)#8	3.69(2)	Cs(2A)-O(5)	3.49(3)
Cs(1A)-O(4)	2.86(3)	Gd(1)-O(4)#1	2.28(2)
Cs(1A)-O(9)#8	2.94(3)	Gd(1)-O(2)#1	2.286(19)
Cs(1A)-O(2)#7	3.03(3)	Gd(1)-O(1)	2.29(2)
Cs(1A)-O(3)#8	3.24(3)	Gd(1)-O(9)#2	2.31(2)
Cs(1A)-O(1)#7	3.30(3)	Gd(1)-O(8)#3	2.331(19)
Cs(1A)-O(1)#8	3.34(4)	Gd(1)-O(6)#2	2.33(2)
Cs(2)-O(9)#10	3.00(2)	Ge(1)-O(1)	1.71(2)
Cs(2)-O(6)#3	3.07(2)	Ge(1)-O(2)	1.73(2)

Cs(2)-O(8)	3.07(2)	Ge(1)-O(7)#3	1.75(2)
Cs(2)-O(2)#11	3.23(2)	Ge(1)-O(3)	1.78(2)
Cs(2)-O(4)#10	3.42(2)	Ge(2)-O(4)	1.73(2)
Cs(2)-O(3)	3.44(2)	Ge(2)-O(5)	1.75(2)
Cs(3)-O(1)#10	3.070(18)	Ge(2)-O(9)	1.76(2)
Cs(3)-O(8)#12	3.16(2)	Ge(2)-O(3)	1.769(19)
Cs(3)-O(6)#12	3.17(2)	Ge(3)-O(6)	1.71(2)
Cs(3)-O(7)#3	3.18(2)	Ge(3)-O(8)	1.746(19)
Cs(3)-O(3)	3.22(2)	Ge(3)-O(7)	1.80(2)
Cs(3)-O(5)	3.31(3)	Ge(3)-O(5)	1.82(2)
Cs(3A)-O(1)#10	2.70(3)	Cs(3A)-O(6)#12	3.03(4)
Cs(3A)-O(8)#12	2.92(4)	Cs(3A)-O(2)#10	3.75(4)
Cs(3A)-O(7)#3	2.98(4)	Cs(3A)-O(1)#10	2.70(3)
O(4)#1-Gd(1)-O(2)#1	86.9(8)	O(4)-Cs(1)-O(2)#7	89.0(6)
O(4)#1-Gd(1)-O(1)	90.8(8)	O(4)-Cs(1)-O(9)#8	137.3(6)
O(2)#1-Gd(1)-O(1)	174.4(7)	O(2)#7-Cs(1)-O(9)#8	110.1(6)
O(4)#1-Gd(1)-O(9)#2	173.2(8)	O(4)-Cs(1)-O(3)#8	98.9(6)
O(2)#1-Gd(1)-O(9)#2	88.3(8)	O(2)#7-Cs(1)-O(3)#8	163.1(6)
O(1)-Gd(1)-O(9)#2	94.4(8)	O(9)#8-Cs(1)-O(3)#8	54.5(5)
O(4)#1-Gd(1)-O(8)#3	99.7(8)	O(4)-Cs(1)-O(1)#7	142.8(6)
O(2)#1-Gd(1)-O(8)#3	87.7(7)	O(2)#7-Cs(1)-O(1)#7	54.2(5)
O(1)-Gd(1)-O(8)#3	87.7(7)	O(9)#8-Cs(1)-O(1)#7	64.0(5)
O(9)#2-Gd(1)-O(8)#3	84.8(8)	O(3)#8-Cs(1)-O(1)#7	115.2(5)
O(4)#1-Gd(1)-O(6)#2	88.3(8)	O(4)-Cs(1)-O(5)	49.0(5)
O(2)#1-Gd(1)-O(6)#2	97.4(8)	O(2)#7-Cs(1)-O(5)	79.2(5)
O(1)-Gd(1)-O(6)#2	87.6(7)	O(9)#8-Cs(1)-O(5)	167.0(6)
O(9)#2-Gd(1)-O(6)#2	87.6(8)	O(3)#8-Cs(1)-O(5)	117.2(5)
O(8)#3-Gd(1)-O(6)#2	170.8(8)	O(1)#7-Cs(1)-O(5)	118.7(6)
O(4)-Cs(1)-O(1)#8	57.6(6)	O(4)-Cs(1A)-O(9)#8	152.0(11)

O(2)#7-Cs(1)-O(1)#8	128.3(5)	O(4)-Cs(1A)-O(2)#7	90.8(8)
O(9)#8-Cs(1)-O(1)#8	81.6(5)	O(9)#8-Cs(1A)-O(2)#7	117.1(9)
O(3)#8-Cs(1)-O(1)#8	48.6(5)	O(4)-Cs(1A)-O(3)#8	98.2(9)
O(1)#7-Cs(1)-O(1)#8	139.0(3)	O(9)#8-Cs(1A)-O(3)#8	55.4(6)
O(5)-Cs(1)-O(1)#8	99.9(5)	O(2)#7-Cs(1A)-O(3)#8	160.7(16)
O(9)#10-Cs(2)-O(6)#3	104.2(6)	O(9)#8-Cs(1A)-O(1)#7	65.2(7)
O(9)#10-Cs(2)-O(8)	62.1(5)	O(2)#7-Cs(1A)-O(1)#7	53.9(6)
O(6)#3-Cs(2)-O(8)	157.6(6)	O(3)#8-Cs(1A)-O(1)#7	111.6(9)
O(9)#10-Cs(2)-O(2)#11	61.7(6)	O(4)-Cs(1A)-O(1)#8	62.7(8)
O(6)#3-Cs(2)-O(2)#11	130.7(6)	O(9)#8-Cs(1A)-O(1)#8	90.8(9)
O(8)-Cs(2)-O(2)#11	61.0(5)	O(2)#7-Cs(1A)-O(1)#8	146.5(11)
O(9)#10-Cs(2)-O(4)#10	53.9(5)	O(3)#8-Cs(1A)-O(1)#8	51.5(6)
O(6)#3-Cs(2)-O(4)#10	59.1(6)	O(1)#7-Cs(1A)-O(1)#8	155.7(8)
O(8)-Cs(2)-O(4)#10	114.7(5)	O(4)-Cs(1A)-O(1)#7	141.6(12)
O(2)#11-Cs(2)-O(4)#10	78.3(6)	O(1)#10-Cs(3)-O(8)#12	103.7(5)
O(9)#10-Cs(2)-O(3)	113.2(6)	O(1)#10-Cs(3)-O(6)#12	61.6(6)
O(6)#3-Cs(2)-O(3)	88.1(5)	O(8)#12-Cs(3)-O(6)#12	54.0(5)
O(8)-Cs(2)-O(3)	82.3(5)	O(1)#10-Cs(3)-O(7)#3	117.0(6)
O(2)#11-Cs(2)-O(3)	141.2(5)	O(8)#12-Cs(3)-O(7)#3	84.7(6)
O(4)#10-Cs(2)-O(3)	132.5(5)	O(6)#12-Cs(3)-O(7)#3	132.5(7)
O(9)#10-Cs(2A)-O(6)#3	115.7(9)	O(1)#10-Cs(3)-O(3)	159.9(5)
O(9)#10-Cs(2A)-O(3)	144.7(11)	O(8)#12-Cs(3)-O(3)	91.2(5)
O(6)#3-Cs(2A)-O(3)	99.6(9)	O(6)#12-Cs(3)-O(3)	138.2(5)
O(9)#10-Cs(2A)-O(8)	62.9(7)	O(7)#3-Cs(3)-O(3)	50.2(6)
O(6)#3-Cs(2A)-O(8)	151.0(12)	O(1)#10-Cs(3)-O(5)	137.2(6)
O(3)-Cs(2A)-O(8)	87.3(8)	O(8)#12-Cs(3)-O(5)	101.6(5)
O(9)#10-Cs(2A)-O(7)#3	155.1(11)	O(6)#12-Cs(3)-O(5)	110.2(6)
O(6)#3-Cs(2A)-O(7)#3	53.9(7)	O(7)#3-Cs(3)-O(5)	99.2(6)
O(3)-Cs(2A)-O(7)#3	51.3(7)	O(3)-Cs(3)-O(5)	49.2(5)

O(8)-Cs(2A)-O(7)#3	138.5(9)	O(1)#10-Cs(3A)-O(8)#12	121.2(13)
O(9)#10-Cs(2A)-O(4)#10	55.9(7)	O(1)#10-Cs(3A)-O(7)#3	139.2(15)
O(6)#3-Cs(2A)-O(4)#10	59.9(7)	O(8)#12-Cs(3A)-O(7)#3	92.7(11)
O(3)-Cs(2A)-O(4)#10	159.4(10)	O(1)#10-Cs(3A)-O(6)#12	67.5(8)
O(8)-Cs(2A)-O(4)#10	110.2(9)	O(8)#12-Cs(3A)-O(6)#12	57.7(9)
O(7)#3-Cs(2A)-O(4)#10	110.1(9)	O(7)#3-Cs(3A)-O(6)#12	150.4(14)
O(9)#10-Cs(2A)-O(5)	96.0(9)	O(1)#10-Cs(3A)-O(3)	146.7(16)
O(6)#3-Cs(2A)-O(5)	147.2(10)	O(8)#12-Cs(3A)-O(3)	85.3(9)
O(3)-Cs(2A)-O(5)	48.9(6)	O(7)#3-Cs(3A)-O(3)	45.8(7)
O(8)-Cs(2A)-O(5)	51.6(6)	O(6)#12-Cs(3A)-O(3)	123.0(12)
O(7)#3-Cs(2A)-O(5)	93.8(8)	O(1)#10-Cs(3A)-O(2)#10	49.8(7)
O(4)#10-Cs(2A)-O(5)	151.4(9)	O(8)#12-Cs(3A)-O(2)#10	146.1(15)
O(1)-Ge(1)-O(2)	113.7(11)	O(7)#3-Cs(3A)-O(2)#10	89.5(10)
O(1)-Ge(1)-O(7)#3	114.1(10)	O(6)#12-Cs(3A)-O(2)#10	114.9(9)
O(2)-Ge(1)-O(7)#3	105.2(12)	O(3)-Cs(3A)-O(2)#10	118.9(10)
O(1)-Ge(1)-O(3)	110.3(11)	O(6)-Ge(3)-O(8)	112.3(11)
O(2)-Ge(1)-O(3)	112.1(10)	O(6)-Ge(3)-O(7)	107.7(12)
O(7)#3-Ge(1)-O(3)	100.7(11)	O(8)-Ge(3)-O(7)	114.8(9)
O(4)-Ge(2)-O(5)	105.8(11)	O(6)-Ge(3)-O(5)	113.1(11)
O(4)-Ge(2)-O(9)	114.2(10)	O(8)-Ge(3)-O(5)	111.2(11)
O(5)-Ge(2)-O(9)	114.7(11)	O(7)-Ge(3)-O(5)	96.7(11)
O(4)-Ge(2)-O(3)	110.5(11)	O(5)-Ge(2)-O(3)	101.0(11)
O(9)-Ge(2)-O(3)	109.8(10)		

Symmetry transformations used to generate equivalent atoms:

#1 $x, y+1, z$	#2 $-x+1, -y+2, z+1/2$	#3 $-x+1/2, y+1/2, z+1/2$
#4 $x+1/2, -y+3/2, z$	#5 $-x+1, -y+1, z+1/2$	#6 $-x+1/2, y-1/2, z-1/2$
#7 $-x+1, -y+1, z-1/2$	#8 $x, y-1, z$	#9 $x+1/2, -y+1/2, z$
#10 $x-1/2, -y+3/2, z$	#11 $-x+1/2, y+1/2, z-1/2$	#12 $-x+1/2, y-1/2, z+1/2$
#13 $x-1/2, -y+1/2, z$	#14 $-x+1, -y+2, z-1/2$	

Table S3 The final atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters

($\text{\AA}^2 \times 10^3$) for $\text{Cs}_3\text{YGe}_3\text{O}_9$. U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor, and the bond valence sum for each atom in asymmetric unit.

	x	y	z	U(eq)	BVS
Y(1)	5381(1)	3462(1)	2259(1)	7(1)	3.107
Ge(1)	4114(1)	3958(1)	4781(1)	9(1)	3.947
Ge(2)	2244(1)	-1076(1)	7751(1)	10(1)	3.961
Ge(3)	3953(1)	1544(1)	6882(1)	8(1)	4.063
Cs(1)	6719(1)	5445(1)	4680(1)	32(1)	0.810
Cs(2)	6774(1)	501(1)	6220(1)	34(1)	0.824
Cs(3)	4370(3)	8944(7)	4025(5)	32(1)	0.970
Cs(3A)	4540(20)	9010(20)	4160(20)	59(3)	1.076
O(1)	5028(4)	3112(9)	3991(5)	14(1)	-2.154
O(2)	4414(5)	5866(10)	5549(5)	22(2)	-2.249
O(3)	4709(4)	-366(8)	6901(6)	15(1)	-2.155
O(4)	4401(5)	3440(9)	7573(5)	18(1)	-2.092
O(5)	3042(4)	-2955(9)	7716(6)	19(1)	-1.836
O(6)	2778(5)	1095(9)	7365(6)	26(2)	-2.020
O(7)	3047(5)	4551(11)	4059(5)	24(2)	-2.015
O(8)	1224(4)	1451(10)	6989(5)	20(1)	-1.895
O(9)	3652(4)	2062(9)	5558(5)	16(1)	-2.340

Table S4 The final atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{Cs}_3\text{GdGe}_3\text{O}_9$. U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor, and the bond valence sum for each atom in asymmetric unit.

	x	y	z	U(eq)	BVS
Gd(1)	4615(1)	13471(2)	7453(1)	11(1)	3.150
Ge(1)	3947(2)	8468(4)	7073(2)	14(1)	4.068
Ge(2)	4101(2)	6054(4)	4978(2)	13(1)	3.947
Ge(3)	2750(2)	6079(4)	2956(3)	16(1)	3.786
Cs(1)	4378(6)	1066(5)	4259(4)	55(2)	0.895
Cs(1A)	4900(30)	950(30)	4520(20)	55(2)	1.025
Cs(2)	1720(2)	10453(4)	4889(2)	43(1)	0.787
Cs(2A)	1591(15)	8800(40)	5410(20)	43(1)	1.272
Cs(3)	1774(2)	5513(5)	6445(3)	55(1)	0.771
Cs(3A)	1580(20)	5580(40)	7150(30)	55(1)	1.187
O(1)	4695(12)	10350(30)	7089(18)	24(5)	-1.980

O(2)	4383(16)	6570(30)	7766(17)	22(5)	-1.870
O(3)	3643(14)	7870(30)	5768(17)	23(4)	-2.172
O(4)	4398(15)	4140(30)	5735(19)	26(5)	-1.944
O(5)	3069(16)	5410(30)	4280(20)	33(6)	-1.940
O(6)	3730(15)	6400(30)	2168(18)	31(6)	-1.890
O(7)	2196(15)	3900(30)	2590(20)	41(7)	-1.980
O(8)	1979(13)	8000(30)	2950(18)	22(4)	-1.800
O(9)	5050(15)	6900(30)	4205(16)	23(5)	-1.840

Table S5 The infrared feature peak details of CYGO and CGGO.^[1,2]

Compound	Position of peak	Vibration mode
CYGO and CGGO	400 cm ⁻¹	Symmetrical curvature vibration of Ge-O-Ge
	532 cm ⁻¹	Aymmetrical stretching vibrations of Ge-O
	731 cm ⁻¹	Asymmetric stretching of Ge-O-Ge

References:

1. R. Saez-Puche, M. Bijkerk, F. Fernández, E. Baran and I. Botto, Crystallographic data, vibrational spectra and magnetic properties of the lanthanide digermanates Ln₂Ge₂O₇, *J. Alloys Compd.*, 1992, **184**, 25-34.
2. I. I. Leonidov, V. P. Petrov, V. A. Chernyshev, A. E. Nikiforov, E. G. Vovkotrub, A. P. Tyutyunnik and V. G. Zubkov, Structural and vibrational properties of the ordered Y₂CaGe₄O₁₂ germanate: a periodic Ab initio study, *J. Phys. Chem. C*, 2014, **118**, 8090-8101.