Electronic Supplementary Information

Exploring multi-stimuli-responsive Pt(II) complexes: Supramolecular selfassembly, lysosome-specific targeted photodynamic therapy and photodegradation of organic pollutants

Hui Ding,‡^a Xiepeng Deng,‡^b Xiang Liu,‡^{a,d} Wenzhao Shang,^b Xianchao Du,^{*a,d} Sufan Wang,^b Xia

Liu,^c Xiangrong Chen,^b Hongjian Liu,^{*b} and Huifang Su^{*b}

^a College of Chemistry and Pharmaceutical Engineering, Nanyang Normal University, Nanyang, Henan 473061, PR China.

^b Department of Orthopaedic Surgery, The First Affiliated Hospital of Zhengzhou University, Zhengzhou, Henan 450052, PR China.

^c State Key Laboratory of Oncology in South China Collaborative Innovation Center for Cancer Medicine Sun Yat-sen University Cancer Center Guangzhou 510060, PR China.

^d College of Chemistry & Pharmacy, Northwest A&F University, Yangling, Shanxi 712100, PR China.

* Corresponding authors.

E-mail addresses: xianchaodu@163.com (X. Du); hongjianmd@126.com (H. Liu); suhuif@mail2.sysu.edu.cn (H. Su).

[‡]These authors contributed equally to this work.



Scheme S1. Synthetic routes to Pt-Cl and Pt-PF.



Fig. S1 ¹H NMR spectrum (400 MHz, CDCl₃) of compound L1.



Fig. S2 ¹³C NMR spectrum (101 MHz, CDCl₃) of compound L1.



Fig. S3 HRMS spectrum of L1.



Fig. S5 ¹³C NMR spectrum (101 MHz, CDCl₃) of compound L2.



Fig. S6 HRMS spectrum of L2.



Fig. S7 ¹H NMR spectrum (400 MHz, CDCl₃) of compound Pt-Cl.



Fig. S8 ¹³C NMR spectrum (101 MHz, CDCl₃) of compound Pt-Cl.



Fig. S9 HRMS spectrum of Pt-Cl.



Fig. S10 ¹H NMR spectrum (400 MHz, CDCl₃) of compound Pt-PF.



Fig. S11 ¹³C NMR spectrum (101 MHz, CDCl₃) of compound Pt-PF.



Fig. S12 HRMS spectrum of Pt-PF.



Fig. S13 The thermodynamic data of Pt-Cl.



Fig. S14 The thermodynamic data of Pt-PF.



Fig. S15 Normalized absorption spectra and PL spectra of Pt-PF measured in DCM



Fig. S16 Absorbance spectra (A) and PL spectra (B) of **Pt-PF** in different solvents. EA, ethyl acetate; THF, tetrahydrofuran; DCM, dichloromethane; DMSO, dimethylsulfoxide. **Pt-PF** concentration: 10 μM.

	Solvents	$\lambda_{abs}(nm)(\epsilon, 10^4 \text{ M}^{-1} \text{cm}^{-1})$	$\lambda_{em}(nm)$
Pt-PF	Hexane	211(0.21),237(0.067),317(0.038)	/
	Cyclohexane	218(0.115),235(0.069),274(0.047),314(0.054),372(0.028)	535
	Toluene	319(0.124),354(0.106),383(0.09)	556
	DCM	249(0.15348),319(0.24084),360(0.24242),386(0.22193), 464(0.00892)	573
	THF	250(0.22),314(0.251),345(0.24),375(0.207)	/
	EA	250(0.113),312(0.186),345(0.195),373(0.171)	/
	TCM	241(0.208),250(0.22),317(0.316),346(0.25),363(0.222),3 81(0.206)	/
	Acetone	324(0.205),345(0.255),373(0.273)	/
	Methanol	220(0.409),247(0.187),311(0.235),354(0.254),373(0.221)	/
	DMSO	317(0.164),338(0.17),375(0.144)	/

Table S1. Photophysical properties of **Pt-PF** in various solvents

DCM, dichloromethane; THF, tetrahydrofuran; EA, ethyl acetate; TCM, trichloromethane; DMSO, dimethylsulfoxide.



Fig. S17 Solutions of **Pt-Cl** in DMSO/Water mixture (percentage of water in DMSO from up to down: 0, 10, 20, 30, 40, 50, 60, 70, 80, 90, 95%). (A) Absorption and (B) PL spectra of Pt-Cl upon increasing the water content in DMSO. **Pt-Cl** concentration: 10 μM.



Fig. 18 SEM images Pt-Cl at different contents of water (A) 0%; (B) 40%; (C) 90%.



Fig. S19 Loss in fluorescence of U2OS cells stained with Pt-PF and LysoTracker Green with the number of scans of laser irradiation. Concentration: 5 μ M, λ_{ex} : 405 nm; scanning rate: 80 s per, frame scale bar:20 um.



Fig. S20 Live/dead staining of **Pt-PF** (5 μ M) treated U2OS cells with or without light irradiation using Calce-DA/PI (1 μ M). (A) and (B)the fluorescence of U2OS cells stained with Calce-DA/PI in the absence of white light; (C) and (D) the fluorescence of U2OS cells stained with Calce-DA/PI in the presence (light) of white light irradiation for 1 h. scale bar: 200 μ m.



Fig. S21 Cyclic absorption of photocatalytic degradation of RhB. Pt-Cl: 50 mg.