Electronic Supplementary Information (ESI)

1. Materials and methods

1.1 Chemicals and measurement

All the chemicals and reagents were purchased from commercial sources and used as received without further purification. The end-products were collected via silica-gel column chromatography and then further purified via vacuum sublimation to be employed for PL and EL properties investigations. ¹H and ¹³C NMR spectra were measured on a Bruker AV 400/500 spectrometer in CD₂Cl₂ or CDCl₃ at room temperature. High-resolution mass spectra (HRMS) were recorded on a GCT premier CAB048 mass spectrometer operating in MALDI-TOF mode. UV-vis absorption spectra were measured on a Shimadzu UV-2600 spectrophotometer. PL spectra were recorded on a Horiba Fluoromax-4 spectrofluorometer. PL quantum yields were measured using a Hamamatsu absolute PL quantum yield spectrometer C11347 Quantaurus QY. Temperature-dependent transient PL decay in films were measured in Edinburgh Instruments FLS1000 spectrometer. The prompt fluorescence lifetimes were measured by using time-correlated single photon counting (TCSPC) mode with a picosecond light source and delayed fluorescence lifetimes were measured by using multichannel scaling (MCS) mode with a variable pulse lasers (VPL) light source. Transient PL decay spectra of materials in solutions were measured using Quantaurus-Tau fluorescence lifetime measurement system (C11367-03, Hamamatsu Photonics Co., Japan). Thermal gravimetric analysis (TGA) data were collected from a TG209F1 under nitrogen protection at a heating rate of 20 K min⁻¹. Differential scanning calorimetric (DSC) analysis were performed on a DSC 214 Polymer under dry nitrogen at a heating rate of 10 °C min⁻¹. The ground-state geometries were optimized using the density function theory (DFT) method with B3LYP functional at the basis set level of 6-31G, and the ΔE_{ST} values between the S₁ and the T₁ were calculated by time-dependent DFT (TDDFT) method at the M062X/6-311G (d, p) level. All the calculations were performed using Gaussian16 package. Cyclic voltammogram was measured in a solution of tetra-n-butylammonium hexafluorophosphate (Bu₄NPF₆, 0.1 M) in dichloromethane and N, N-dimethylformamide (DMF) containing the sample at a scan rate of 100 mV s⁻¹. Three-electrode system (Ag/Ag⁺, platinum wire and glassy carbon electrode as reference, counter and work electrode respectively) was used in the CV method. $E_{\text{HOMO}} = -[E_{\text{ox}} + 4.8] \text{ eV}$, and $E_{\text{LUMO}} = -[E_{\text{re}} + 4.8] \text{ eV}$. E_{ox} and $E_{\rm re}$ represent the onsets oxidation (measured in dichloromethane) and reduction (measured in DMF) potentials relative to ferrocene.

1.2 Device fabrication

Glass substrates pre-coated with a 95-nm-thin layer of indium tin oxide (ITO) with a sheet resistance of 20 Ω per square were thoroughly cleaned for 10 minutes in ultrasonic bath of acetone, isopropyl alcohol, detergent, deionized water, and isopropyl alcohol and then treated with O₂ plasma for 5 min in sequence in order to improve the hole injection ability. Organic layers were deposited onto the ITOcoated substrates by high-vacuum (< 5 × 10⁻⁴ Pa) thermal evaporation in the Fangsheng OMV-FS380 vacuum deposition system. Organic materials, Liq and Al were deposited at rates of 1~2 A s⁻¹, 0.1 A s⁻¹ and 5 A s⁻¹, respectively. The emission area of the devices is 3 × 3 mm² as shaped by the overlapping area of the anode and cathode. All the device characterization steps were carried out at room temperature under ambient laboratory conditions without encapsulation. The luminance–voltage–current density and EQE were characterized with a dual-channel Keithley 2614B source meter and a PIN-25D silicon photodiode. The EQEs were determined by assuming Lambertian pattern. The electroluminescence spectra were obtained via an Ocean Optics USB 2000+ spectrometer, along with a Keithley 2614B Source Meter. Simulated angle dependent relative luminance of the devices was taken by CS-200 Color and Luminance Meter.

2. Equations for photophysical properties

The quantum efficiencies and rate constants were determined using the following equations according to Adachi's method.^{1–3}

$$\begin{split} \boldsymbol{\Phi}_{\text{prompt}} &= \boldsymbol{\Phi}_{\text{PL}} \boldsymbol{R}_{\text{prompt}} \\ \boldsymbol{\Phi}_{\text{delayed}} &= \boldsymbol{\Phi}_{\text{PL}} \boldsymbol{R}_{\text{delayed}} \\ \boldsymbol{k}_{\text{F}} &= \boldsymbol{\Phi}_{\text{prompt}} / \boldsymbol{\tau}_{\text{prompt}} \\ \boldsymbol{\Phi}_{\text{PL}} &= \boldsymbol{k}_{\text{F}} / (\boldsymbol{k}_{\text{F}} + \boldsymbol{k}_{\text{IC}}) \\ \boldsymbol{\Phi}_{\text{prompt}} &= \boldsymbol{k}_{\text{F}} / (\boldsymbol{k}_{\text{F}} + \boldsymbol{k}_{\text{IC}}) \\ \boldsymbol{\Phi}_{\text{IC}} &= \boldsymbol{k}_{\text{IC}} / (\boldsymbol{k}_{\text{F}} + \boldsymbol{k}_{\text{IC}} + \boldsymbol{k}_{\text{ISC}}) \\ \boldsymbol{\Phi}_{\text{ISC}} &= \boldsymbol{k}_{\text{ISC}} / (\boldsymbol{k}_{\text{F}} + \boldsymbol{k}_{\text{IC}} + \boldsymbol{k}_{\text{ISC}}) = 1 - \boldsymbol{\Phi}_{\text{prompt}} - \boldsymbol{\Phi}_{\text{IC}} \\ \boldsymbol{\Phi}_{\text{RISC}} &= \boldsymbol{\Phi}_{\text{delayed}} / \boldsymbol{\Phi}_{\text{ISC}} \\ \boldsymbol{k}_{\text{RISC}} &= (\boldsymbol{k}_{\text{p}} \boldsymbol{k}_{\text{d}} \boldsymbol{\Phi}_{\text{delayed}}) / (\boldsymbol{k}_{\text{ISC}} \boldsymbol{\Phi}_{\text{prompt}}) \\ \boldsymbol{k}_{\text{p}} &= 1 / \boldsymbol{\tau}_{\text{prompt}}; \ \boldsymbol{k}_{\text{d}} = 1 / \boldsymbol{\tau}_{\text{delayed}} \end{split}$$

3. Additional Figures and Tables





Fig. S1 NMR spectra of *p*-APDC-DTPA and *o*-APDC-DTPA, dissolving in dichloromethane- d_2 and chloroform.



Fig. S2 DSC thermograms recorded under nitrogen at a heating rate of 10 °C min⁻¹.



Fig. S3 (A)PL spectra of *p*-APDC-DTPA in THF/water mixtures with different water fractions (f_w). (B) Plots of I/I₀ values versus f_w of *p*-APDC-DTPA in THF/water mixtures. (C) Plots of I/I₀ values versus f_w of *o*-APDC-DTPA in THF/water mixtures. (I₀ is the PL intensity in pure THF, and I is the PL intensity in the mixtures).



Fig. S4 Fluorescence and phosphorescence spectra of (A) *p*-APDC-DTPA and (B) *o*-APDC-DTPA doped in TPBi films with a concentration of 10 wt% measured at 77 K.



Fig. S5 PL spectra of (A) 10 wt% *p*-APDC-DTPA: TPBi doped film and (B) 10 wt% *o*-APDC-DTPA: TPBi doped film, measured with different delay times at room temperature.



Fig. S6 Transient PL decay spectra of (A) *p*-APDC-DTPA and (B) *o*-APDC-DTPA doped in TPBi films with a concentration of 10 wt% measured at 300 K.



Fig. S7 The NTO analyses of excited singlet and triplet states of *p*-APDC-DTPA with the SOC matrix elements. Calculated energy levels, ΔE_{ST} values, and SOC matrix elements of excited singlet and

triplet states of *p*-APDC-DTPA.



Fig. S8 The NTO analyses of excited singlet and triplet states of *o*-APDC-DTPA with the SOC matrix elements. Calculated energy levels, ΔE_{ST} values, and SOC matrix elements of excited singlet and triplet states of *o*-APDC-DTPA.

	Temperature (K)	<i>p</i> -APDC-DTPA	o-APDC-DTPA		
	300	69.16	69.50		
	250	64.16	80.23	80.23	
$ au_{ m delayed}$	200	87.77	49.61		
(µs)	150	54.90	35.97	35.97	
	100	5.20	21.34		
100 5.20 77 6.25 300 62	6.25	10.17			
	300	62	63		
	250	55	43		
$R_{\rm delayed}$	200	36	3		
(%)	150	13	3		
	100	4	1		
	77	3	1		

Table S1. Temperature-dependent photophysical parameters of doped films in TPBi host with a concentration of 10 wt%.

a) $\tau_{\text{delayed}} = \text{lifetimes calculated from delayed fluorescence decay; } R_{\text{delayed}} = \text{ratio of delayed component.}$



Fig. S9. A) Luminance–voltage–current density, B) current efficiency–luminance–power efficiency and C) external quantum efficiency–luminance of the non-doped OLEDs. Inset in plane E: EL spectra of the non-doped OLEDs.



Fig. S10 A) luminance–voltage–current density, B) current efficiency–luminance–power efficiency and C) external quantum efficiency–luminance of the doped OLEDs with the concentration of 5 wt% in TPBi. Inset in plane E: EL spectra of the doped OLEDs.



Fig. S11 A) luminance–voltage–current density, B) current efficiency–luminance–power efficiency and C) external quantum efficiency–luminance of the doped OLEDs with the concentration of 15 wt% in TPBi. Inset in plane E: EL spectra of the doped OLEDs.



Fig. S12 A) EQE summary based on the dicyanopyrazino phenanthrene core. B) EQE summary of representative non-doped red/NIR TADF-OLEDs with emission peaks from 650 to 750 nm.

Table S2. The EL performance based phenanthrene core and non-doped deep-red/NIR emitters with emission peaks above 600 nm.

Emitter	Year	λ_{EL} (nm)	EQE (%)	CIE (x, y)	Ref.
3,6APDC-DTPA	2022	636	10.5	(0.62, 0.38)	This work
4,5APDC-DTPA	2022	636	19.0	(0.63, 0.37)	This work
1DMAC-BP	2019	560	10.1	(0.43, 0.54)	4
2DMAC-BP	2019	576	11.8	(0.43, 0.54)	4
3DMAC-BP	2019	606	22.0	(0.43, 0.54)	4
Ac-CNP	2016	580	13.3	(0.47, 0.51)	5
Px-CNP	2016	606	3.0	(0.53, 0.44)	5
PyCN-ACR	2016	572	15.6	(0.46, 0.52)	6
DMAC-Ph-DCPP	2017	596	16.9	(0.53, 0.46)	7
DPA-DCPP	2017	616	10.4	(0.61, 0.38)	7
DMAC-DCPP	2017	624	10.1	(0.60, 0.40)	7
DPA-Ph-DCPP	2017	644	15.1	(0.64, 0.36)	7
Cz-DCPP	2017	560	14.8	(0.44, 0.54)	7
Cz-Ph-DCPP	2017	564	11.6	(0.46, 0.52)	7
Ac-CNBQx	2018	585	14.0	(0.51, 0.48)	8
Ac-CNBPz	2018	630	16.2	(0.61, 0.39)	8
Da-CNBQx	2018	617	20.0	(0.59, 0.41)	8
PXZ-DCPP	2018	608	17.4	(0.56, 0.43)	9
DPXZ-BPPZ	2018	612	20.1	(0.60, 0.40)	10
TPA-QCN	2017	644	14.5	(0.62, 0.38)	11
DDTPACz-DCPP	2019	646	13.6	(0.61, 0.38)	12
TPA-DCPP	2015	648	9.6	(0.64, 0.35)	13
TPA-PPDCN	2019	648	18.8	(0.65, 0.35)	14
tDBBPZ-DPXZ	2019	608	17.0	(0.57, 0.43)	15
DBPZ-PDXZ	2019	608	17.8	(0.58, 0.42)	15
mDPBPZ-PXZ	2019	624	21.7	(0.62, 0.38)	16
TAT-DBPZ	2020	604	15.4	-	17
TAT-FDBPZ	2020	611	9.2	-	17

TCQ	2022	574	16.9	(0.48, 0.50)	18
TCPQ	2022	612	21.9	(0.58, 0.41)	18
TPQ	2022	622	21.8	(0.62, 0.39)	18
ANQDC-MeFAC	2020	614	26.3	(0.60, 0.40)	19
ANQDC-DMAC	2020	615	27.5	(0.58, 0.41)	19
T-DA-2	2020	640	26.3	(0.62, 0.37)	20
TPA-PZCN	2019	648	28.1	(0.66, 0.34)	21
DCPPr-a-NDPA	2021	606	31.5	(0.58, 0.42)	22
4,5APDC-DTPA	2022	712	6.6	(0.69, 0.29)	This work
mDPBPZ-PXZ	2019	680	5.2	(0.68, 0.32)	16
BPPZ-PXZ	2019	656	2.5	(0.65, 0.35)	16
TPATCN	2015	675	2.6	(0.67, 0.32)	23
TPA-QCN	2017	728	3.9	(0.69, 0.31)	11
pCNQ-TPA	2021	700	4.6	-	24
TPA-DCPP	2015	708	2.1	(0.70, 0.29)	13
NZ2TPA	2017	696	3.3	(0.70, 0.30)	25
TAPPQ	2019	711	3.5	-	26
TPA-PZCN	2019	680	5.3	(0.69, 0.30)	21
TCPQ	2022	718	5.4	(0.70, 0.30)	18
TPQ	2022	716	4.9	(0.69, 0.30)	18

Table S3. EL performances of the OLEDs based on *p*-APDC-DTPA and *o*-APDC-DTPA fabricated with device structure of ITO/HATCN (5 nm)/NPB (30 nm)/mCP (5 nm)/EML (20 nm)/TPBi (50 nm)/LiF (1 nm)/Al.

Emitter	concentration	$V_{\rm on}{}^{\rm a)}({\rm V})$	$L_{\rm max} ({\rm cd} \ {\rm m}^{-2})$	CE (cd A ⁻¹)	PE (lm W ⁻¹)	EQE (%)	$\lambda_{\rm EL}({\rm nm})$	CIE (x, y)
<i>p</i> -APDC-DTPA	5%	3.2	2593	4.3	3.7	4.2	634	(0.615, 0.376)
	10%	2.8	2645	3.3	3.4	5.3	658	(0.651, 0.343)
	15%	2.8	2598	2.3	2.5	5.1	668	(0.668, 0.327)
o-APDC-DTPA	5%	3.0	5912	11.8	11.3	6.4	610	(0.561, 0.430)
	10%	2.8	5252	13.7	14.6	11.0	628	(0.610, 0.386)
	15%	2.8	5319	11.3	12.7	11.2	640	(0.628, 0.369)

^{a)} Abbreviations: V_{on} = turn-on voltage at 1 cd m⁻²; L_{max} = maximum luminance; CE = maximum current efficiency; PE = maximum power efficiency; EQE = maximum external quantum efficiency; λ_{EL} = electroluminescence peak; CIE = Commission Internationale de l'Eclairage coordinates.

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