

Supporting information

Design and Synthesis of Novel Pomegranate-Like TiN@MXene Microspheres as Efficient Sulfur Hosts for Advanced Lithium Sulfur batteries

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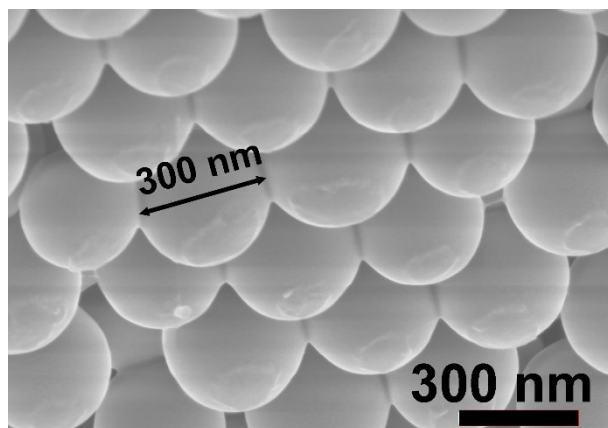


Fig. S1 SEM image of sPS spheres.

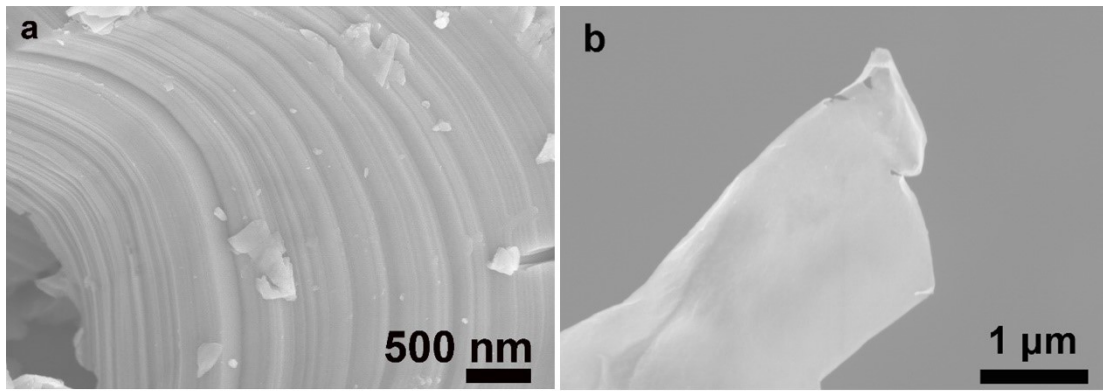


Fig. S2 SEM images of (a) MAX and (b) Ti₃C₂ MXene.

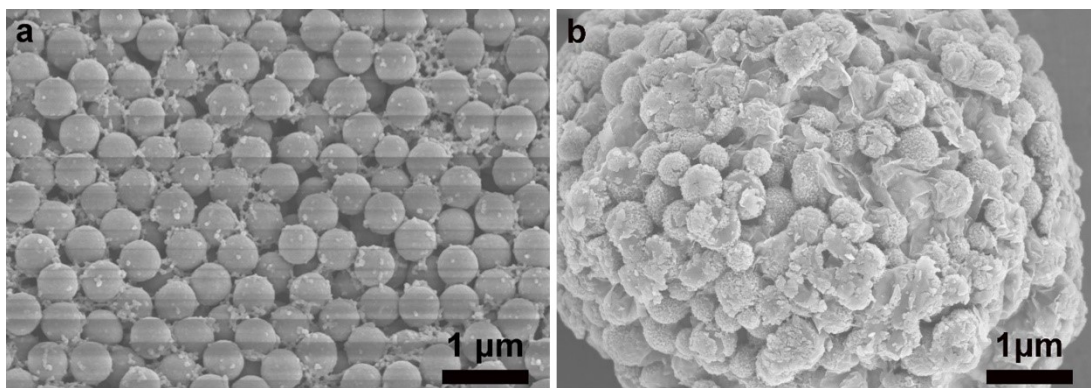


Fig. S3 SEM images of (a) sPS@TiO₂ and (b) sPS@TiO₂@MXene.

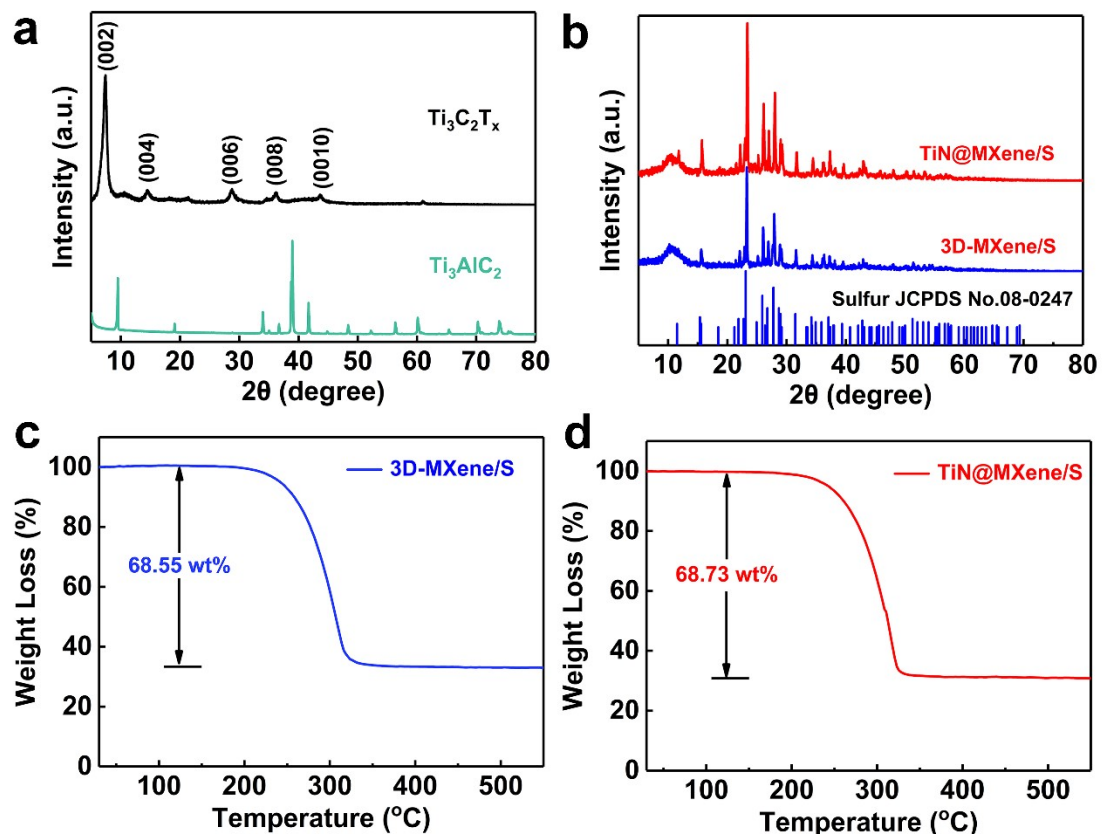


Fig. S4 (a, b) XRD patterns of the samples in synthesis experiments. (c, d) TG curves of the 3D-MXene/S and $TiN@MXene/S$ composites.

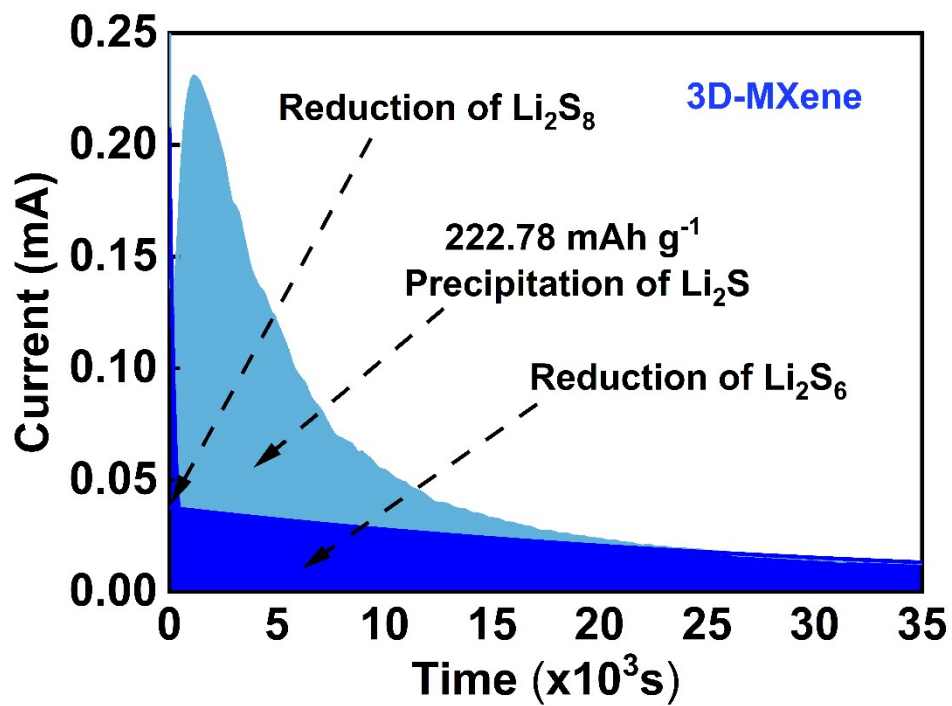


Fig. S5 Constant potential discharge curve of the Li₂S deposition tests using 3D-MXene as substrate.

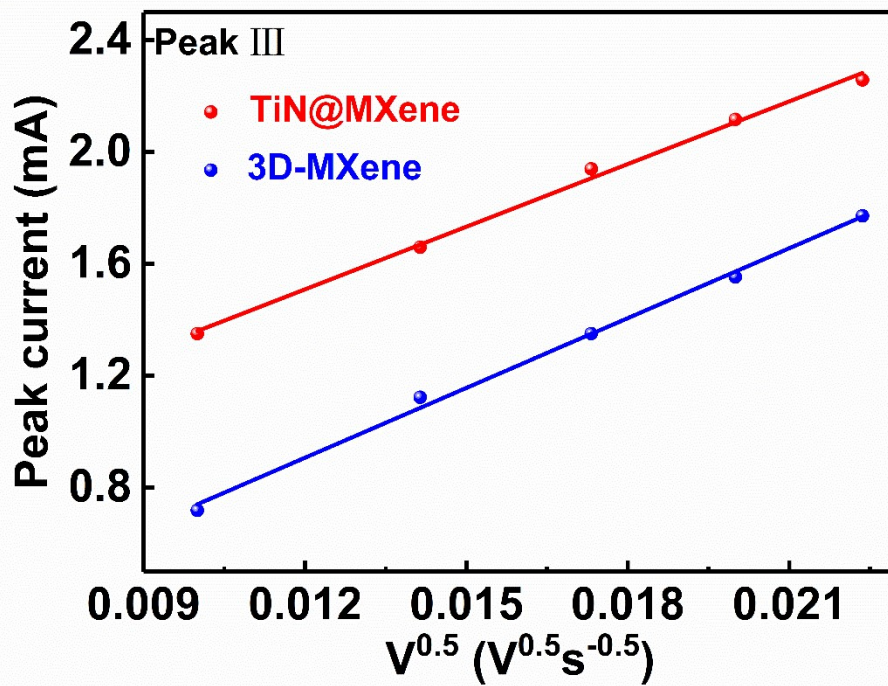


Fig. S6 The Li^+ diffusion coefficient values of the TiN@MXene and 3D-MXene cathodes for the processes represented by peak III.

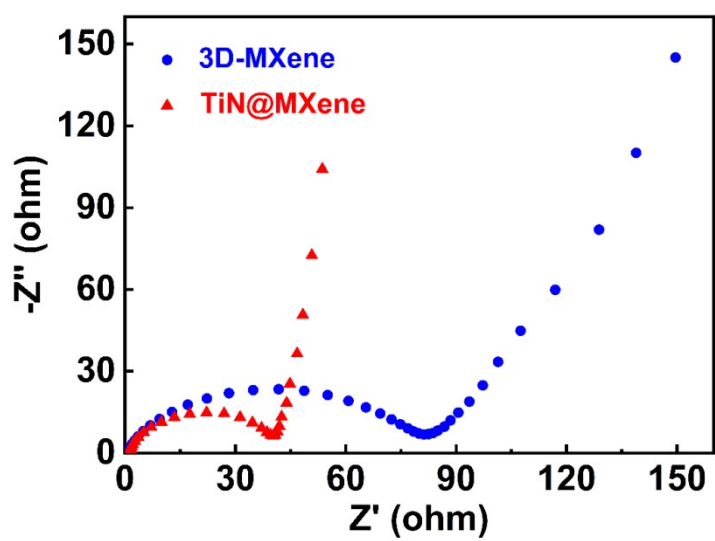


Fig. S7 EIS spectra of the 3D-MXene and TiN@MXene cathodes.

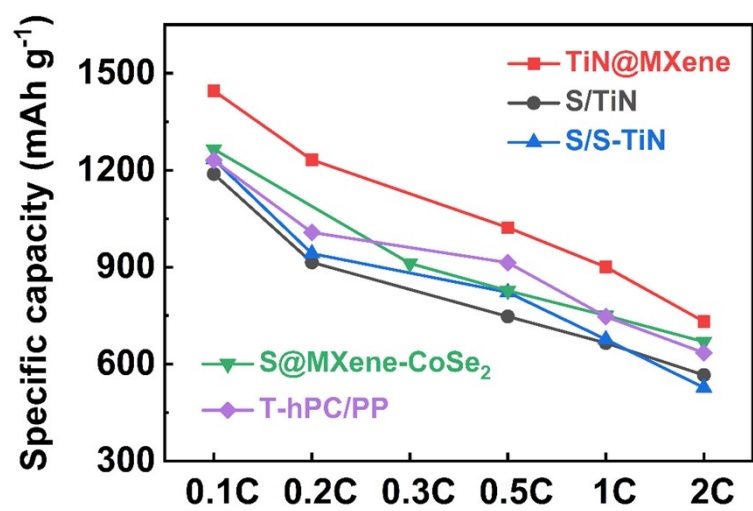


Fig. S8 The rate performance of TiN@MXene cathode was compared with some reported cathodes based on TiN or some other catalytic materials.

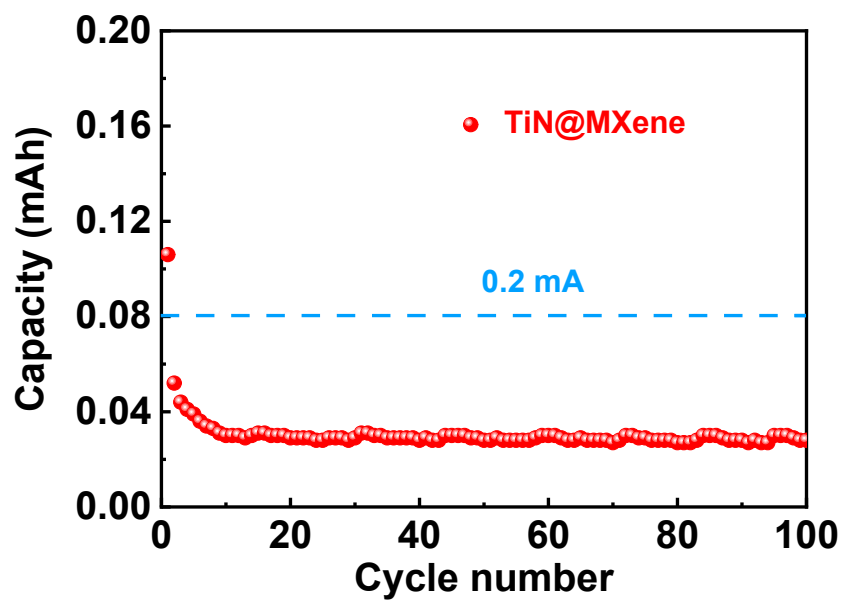


Fig. S9 Cycling performance of TiN@MXene based cathode half-cell at 0.2 mA.

Table S1. Summary of D_{Li} values at peaks I, II, III for 3D-MXene and TiN@MXene cathodes.

CV Peak	TiN@MXene	3D-MXene
Peak I	$8.72 \times 10^{-9} \text{ cm}^2 \text{ s}^{-1}$	$3.40 \times 10^{-9} \text{ cm}^2 \text{ s}^{-1}$
Peak II	$6.73 \times 10^{-9} \text{ cm}^2 \text{ s}^{-1}$	$1.68 \times 10^{-9} \text{ cm}^2 \text{ s}^{-1}$
Peak III	$3.18 \times 10^{-8} \text{ cm}^2 \text{ s}^{-1}$	$2.23 \times 10^{-8} \text{ cm}^2 \text{ s}^{-1}$