

**Highly enhanced electrocatalytic OER activity of water coordinated copper complexes:
Effect of lattice water and bridging ligand**

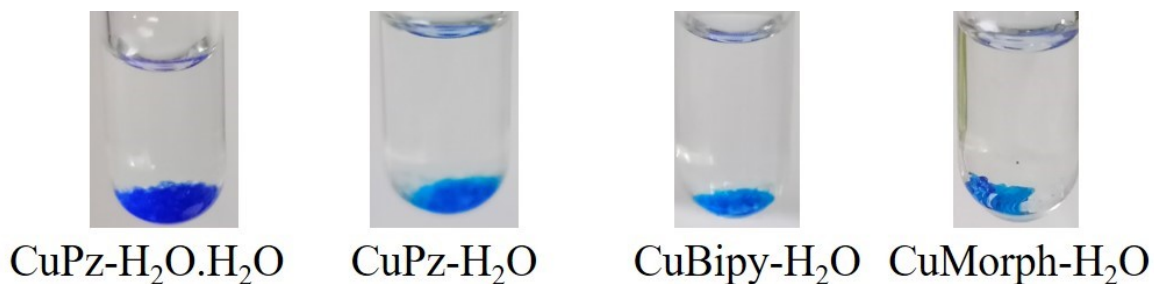


Figure S1. Digital images of copper complex crystals in ethanol.

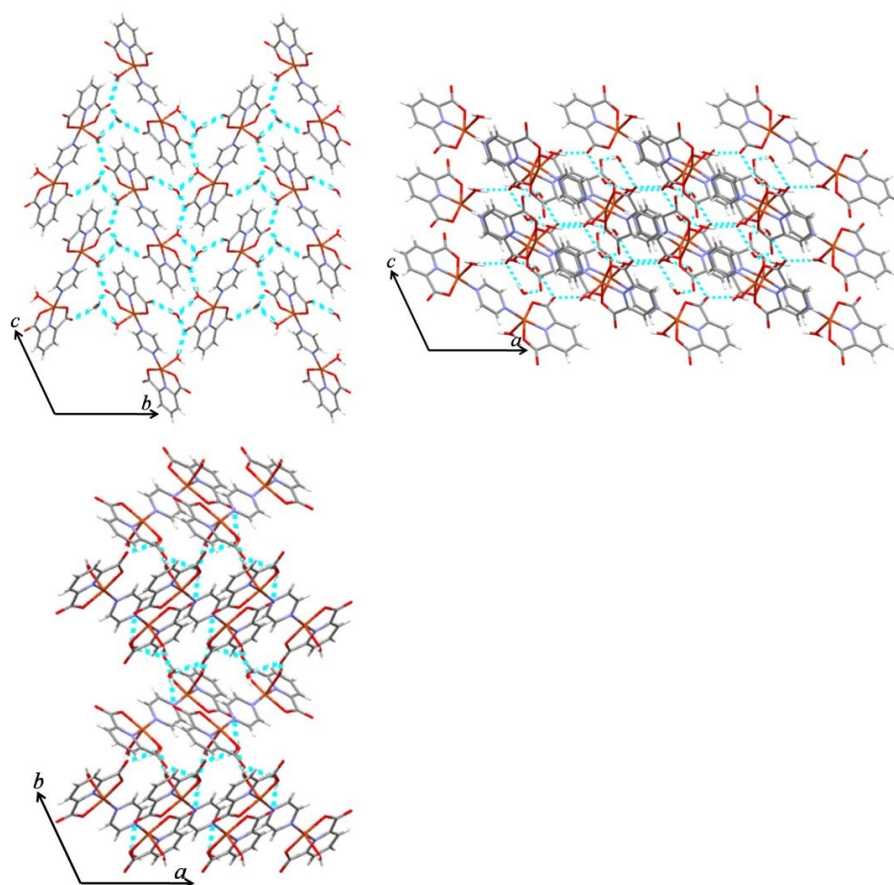


Figure S2. Molecular packing in the crystal lattice of **CuPz-H₂O.H₂O**. C (grey), H (white), N (blue), O (red) and Cu (brown). Dotted lines indicate the hydrogen bonding interactions. H-bond distances ranged between 2.733 and 2.980 Å.

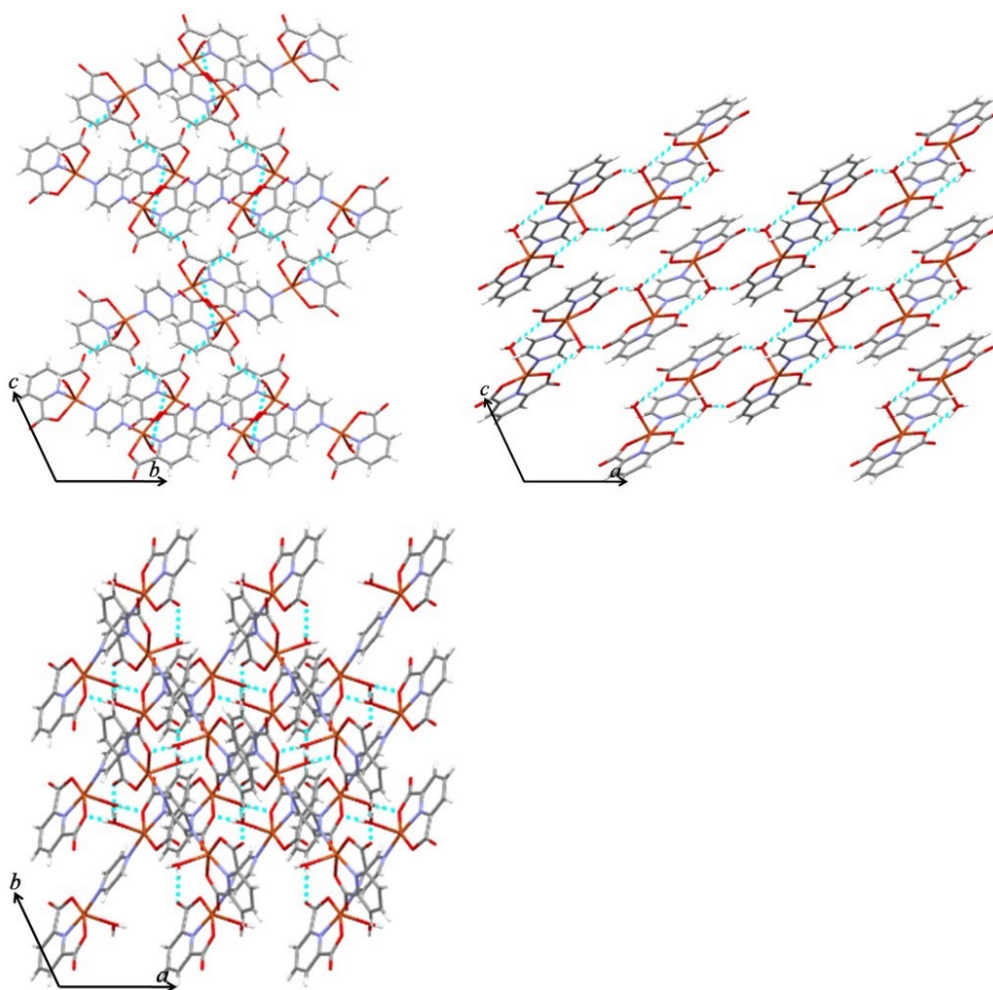


Figure S3. Molecular packing in the crystal lattice of **CuPz-H₂O**. C (grey), H (white), N (blue), O (red) and Cu (brown). C (grey), H (white), N (blue), O (red) and Cu (brown). Dotted lines indicate the hydrogen bonding interactions. H-bond distances ranged between 2.733 and 2.980 Å.

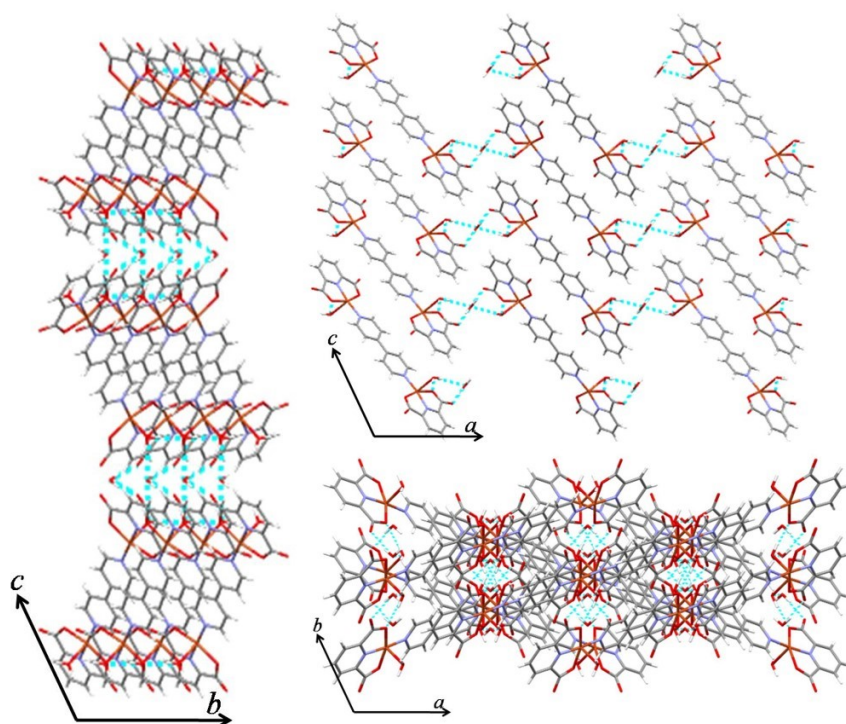


Figure S4. Molecular packing in the crystal lattice of **CuBipy-H₂O.H₂O**. C (grey), H (white), N (blue), O (red) and Cu (brown). C (grey), H (white), N (blue), O (red) and Cu (brown). Dotted lines indicate the hydrogen bonding interactions. H-bond distances ranged between 2.733 and 2.980 Å.

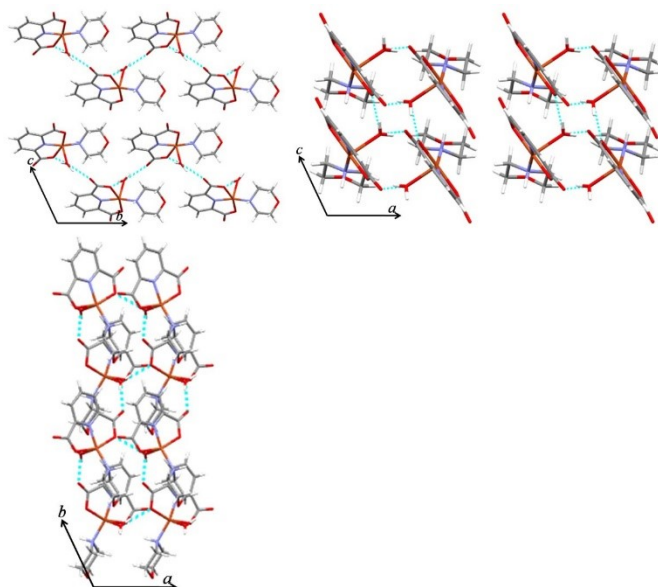


Figure S5. Molecular packing in the crystal lattice of **CuMorph-H₂O**. C (grey), H (white), N (blue), O (red) and Cu (brown). C (grey), H (white), N (blue), O (red) and Cu (brown). Dotted lines indicate the hydrogen bonding interactions. H-bond distances ranged between 2.733 and 2.980 Å.

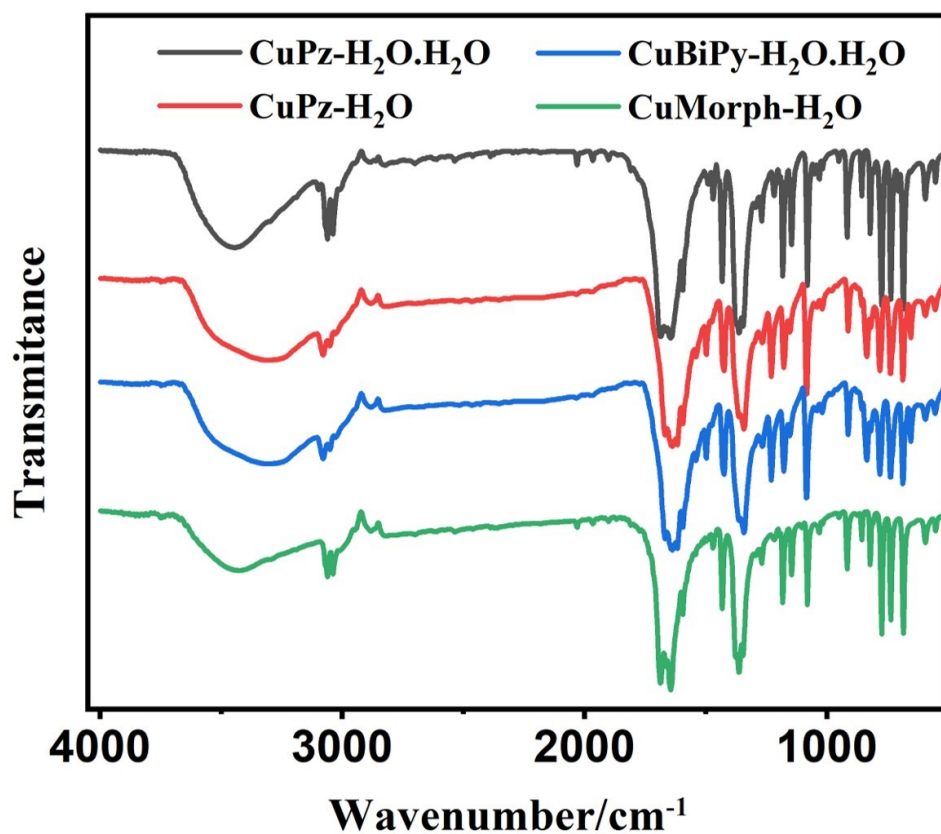


Figure S6. FTIR spectra of copper complexes.

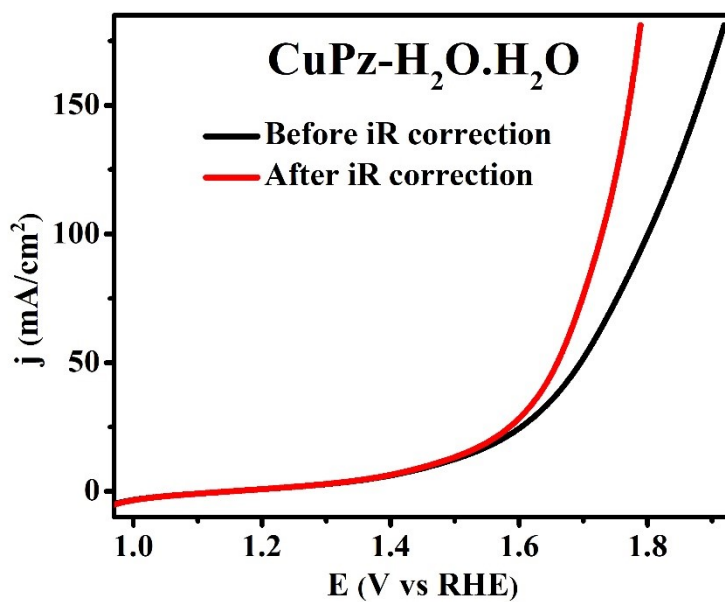


Figure S7. OER polarization curves of CuPz-H₂O.H₂O before and after iR correction.

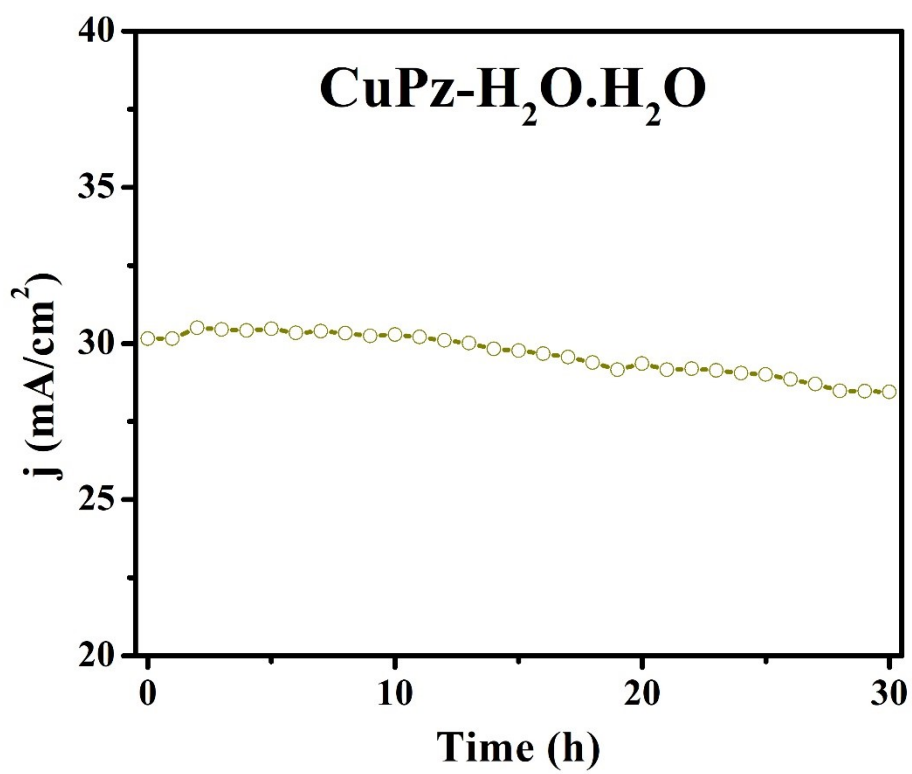


Figure S8. The current-time amperometric test performed at 30 mA/cm².

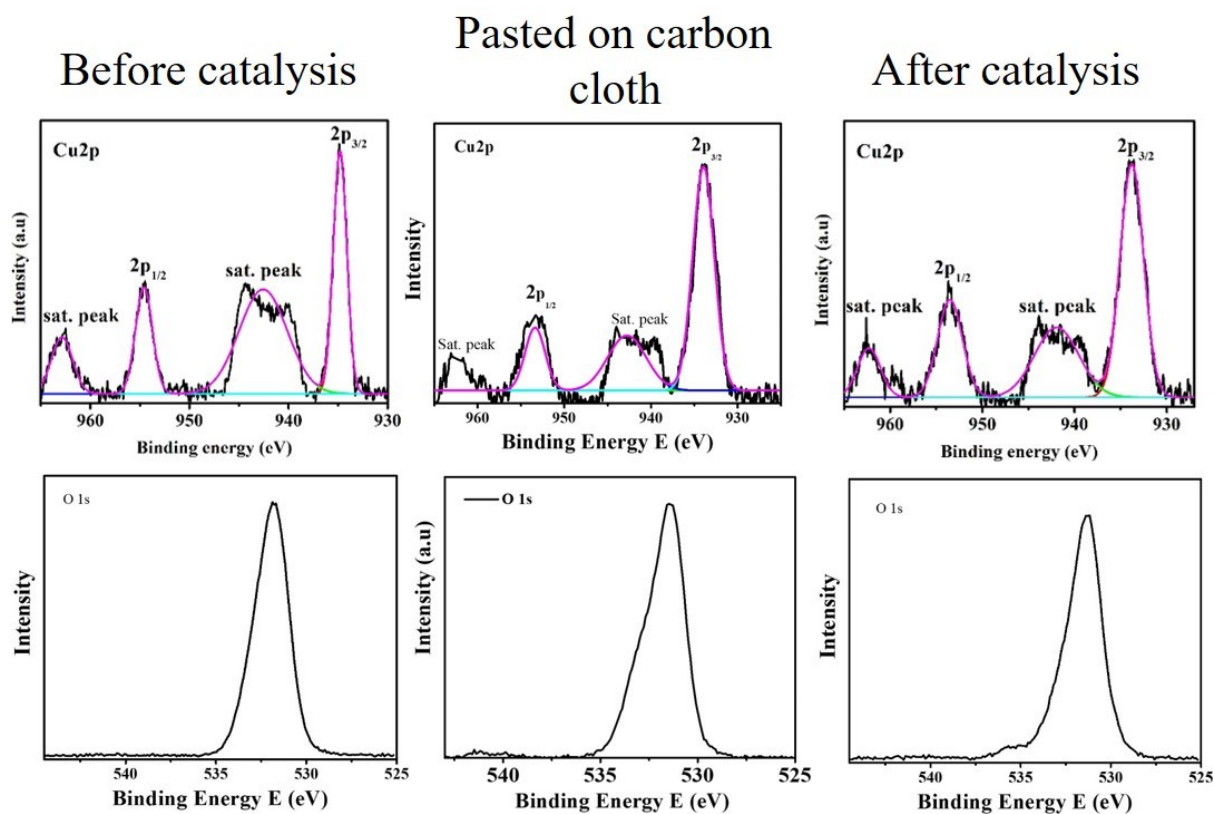


Figure S9. High resolution XPS spectra of Cu 2p and O 1s of $\text{CuPz-H}_2\text{O}\cdot\text{H}_2\text{O}$.

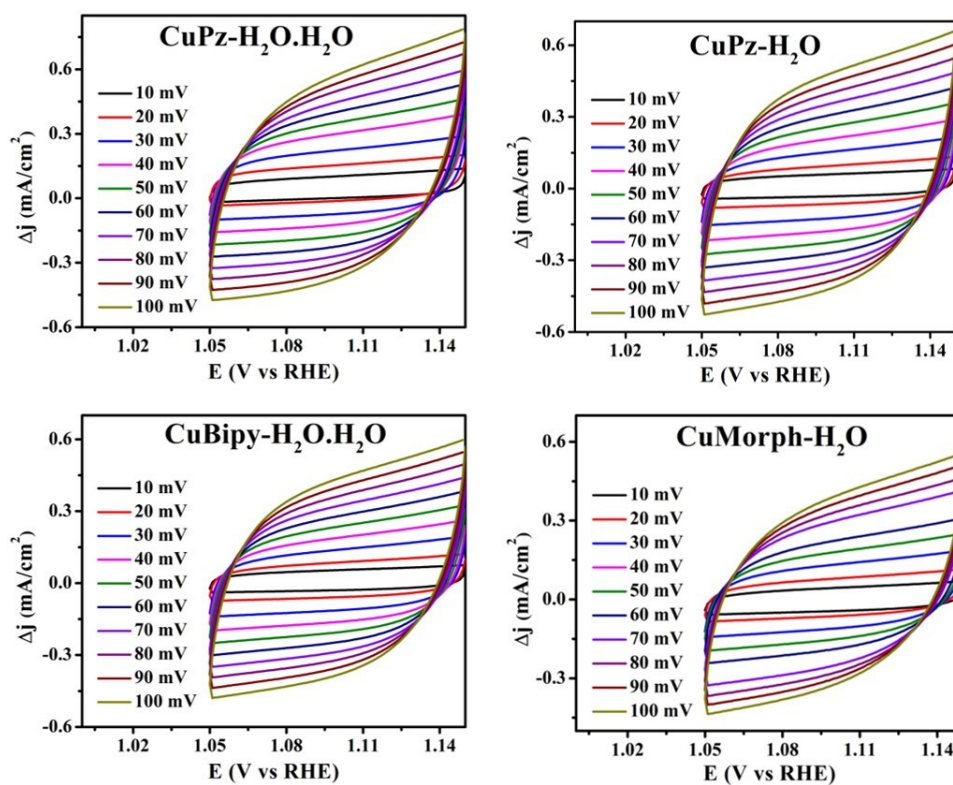


Figure S10. Double layer capacitance and capacitive currents as a functional of scan rate.