

Mechanism of Water Pollutant Photodegradation by Mixed and Core-Shell WO_3/TiO_2 Nanocomposites

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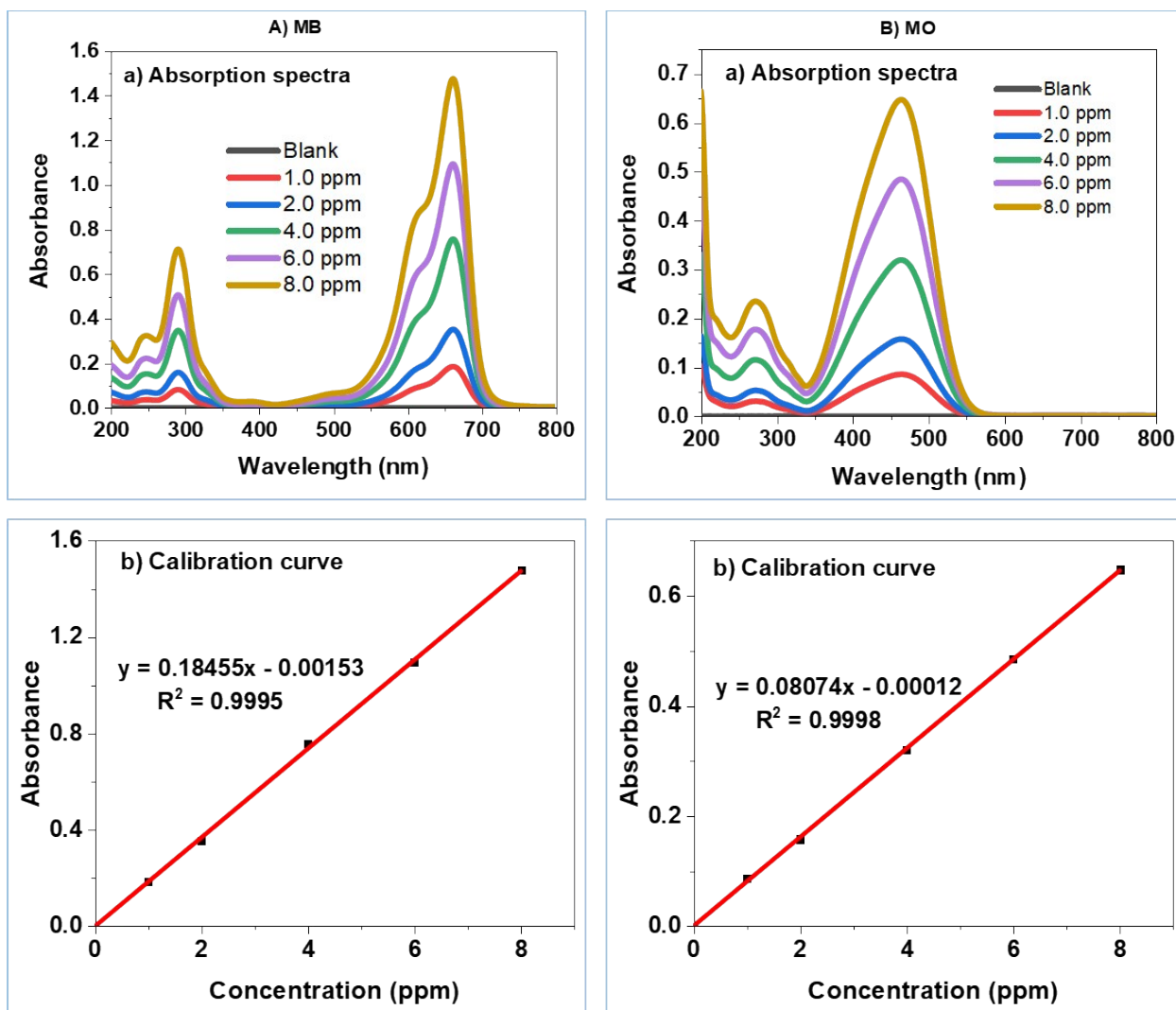


Figure S1. UV-vis absorption spectra and calibration curves for (A) MB⁺ at 662 nm and (B) 464 nm for MO⁻.

Table S1 O 1s peak parameters and % area for each composite.

	Peak Type	Position (eV)	FWHM (eV)	Area	%Area
5 wt% WO ₃ /TiO ₂	Peak 1	530.35	1.242	501.23	68.81
	Peak 2	531.76	1.6	111.29	15.28
	Peak 3	533.42	1.81	115.92	15.91
8 wt% WO ₃ /TiO ₂	Peak 1	529.68	1.3	576.11	72.23
	Peak 2	530.94	1.63	144.13	18.08
	Peak 3	532.75	1.54	77.04	9.668
10 wt% WO ₃ /TiO ₂	Peak 1	529.98	1.26	622.56	71.85
	Peak 2	531.28	1.47	152.49	17.60
	Peak 3	533.01	1.405	91.47	10.56
TiO ₂ @WO ₃	Peak 1	530.00	1.13	477.41	75.15
	Peak 2	531.00	1.37	111.04	17.48
	Peak 3	532.70	1.33	46.86	7.38
WO ₃ @TiO ₂	Peak 1	530.00	1.211	522.03	77.76
	Peak 2	531.32	1.15	74.14	11.04
	Peak 3	532.48	1.384	75.16	11.20

Table S2 Ti 2p peak parameters and % area for each composite.

	Peak Type	Position (eV)	FWHM (eV)	Area	%Area
5 wt%	Ti 2p _{3/2}	459.1	1.17	424.14	64.05
WO ₃ /TiO ₂	Ti 2p _{1/2}	464.77	2.11	238.04	35.95
8 wt%	Ti 2p _{3/2}	458.35	1.326	457.53	71.32
WO ₃ /TiO ₂	Ti 2p _{1/2}	464.05	2.155	183.96	28.68
10 wt%	Ti 2p _{3/2}	458.92	1.15	462.72	65.34
WO ₃ /TiO ₂	Ti 2p _{1/2}	464.62	2.17	245.48	34.66
TiO ₂ @WO ₃	Ti 2p _{3/2}	459.25	1.154759	467.95	66.99
	Ti 2p _{1/2}	464.93	2.089	230.5865	33.01
WO ₃ @TiO ₂	Ti 2p _{3/2}	458.73	1.1	467.6004	69.18
	Ti 2p _{1/2}	464.42	2.05	208.3091	30.82

Table S3 W 4f peak parameters and % area for each composite.

	Peak Type	Position (eV)	FWHM (eV)	Area	%Area	W ⁵⁺ /W ⁶⁺ Ratio	Ti 2p/W 4f Ratio
5 wt% WO ₃ /TiO ₂	W ⁵⁺ 4f _{7/2}	35.56	1.123	46.97	32.80	3.09	4.62
	W ⁵⁺ 4f _{5/2}	37.44	1.206	61.24	42.77		
	W ⁶⁺ 4f _{7/2}	36.48	0.779	15.71	10.97		
	W ⁶⁺ 4f _{5/2}	38.25	1.106	19.27	13.46		
8 wt% WO ₃ /TiO ₂	W ⁵⁺ 4f _{7/2}	34.97	1.45	72.68	39.77	2.42	3.51
	W ⁵⁺ 4f _{5/2}	36.9	1.14	56.70	31.03		
	W ⁶⁺ 4f _{7/2}	36.06	0.91	20.41	11.17		
	W ⁶⁺ 4f _{5/2}	37.6	1.42	32.94	18.03		
10 wt% WO ₃ /TiO ₂	W ⁵⁺ 4f _{7/2}	35.05	1.68	123.9	53.61	4.59	3.06
	W ⁵⁺ 4f _{5/2}	36.95	1.07	65.89	28.50		
	W ⁶⁺ 4f _{7/2}	36.35	0.75	13.43	5.81		
	W ⁶⁺ 4f _{5/2}	37.54	1.12	27.95	12.09		
TiO ₂ @WO ₃	W ⁵⁺ 4f _{7/2}	35.31	1.1	50.21	27.46	2.03	3.82
	W ⁵⁺ 4f _{5/2}	37.45	1.25	72.37	39.57		
	W ⁶⁺ 4f _{7/2}	36.56	1.11	35.10	19.19		
	W ⁶⁺ 4f _{5/2}	38.74	1.21	25.20	13.78		
WO ₃ @TiO ₂	W ⁵⁺ 4f _{7/2}	35.2	1.23	45.46	33.70	13.40	5.00
	W ⁵⁺ 4f _{5/2}	37.12	1.443	80.09	59.36		
	W ⁶⁺ 4f _{7/2}	36.36	0.65	4.83	3.58		
	W ⁶⁺ 4f _{5/2}	38.8	0.941	4.54	3.36		

Table S4 Comparison of photocatalysis performance for MB degradation

Catalyst	Preparation method	Light source	Degradation rate	Rate constant	Reference
TiO ₂	Hydrothermal	Xe lamp, 300 W	75% in 240 min	0.00554 min ⁻¹	W. Wang et al. 2019 ⁴⁶
Core-shell WO ₃ @TiO ₂ (36 wt% of WO ₃)	Sol-gel		100% in 240 min	0.01485 min ⁻¹ 2.68 times > TiO ₂	
TiO ₂	Sol-gel	Halogen lamp, 400 W	75% in 180 min	0.009 min ⁻¹	*W. A. El-Yazeed & Ahmed, 2019
10 wt% WO ₃ /TiO ₂ (mixed)			99% in 180 min	0.017 min ⁻¹ 2 times > TiO ₂	
TiO ₂	Hydrothermal	Xe lamp (350 W with 420 nm cut-off filter)	40.7% in 150 min	-	Q. Wang et al. 2021 ⁴⁸
10 wt% WO ₃ /TiO ₂ (mixed)			87.8% in 150 min	-	
TiO ₂	Precipitation	Halogen lamp, 500 W	12% in 60 min	0.001 min ⁻¹	M. F. Mubarak et al. 2022 ⁶⁵
Core-shell TiO ₂ @CoFe ₃ O ₄	Co-precipitation		91% in 60 min	0.016 min ⁻¹ 16 times > TiO ₂	
TiO ₂	As-purchased	UV light (365 nm)	80% in 120 min	0.0117 min ⁻¹	R. Wahyuono et al. 2019 ⁵¹
25 wt% WO ₃ /TiO ₂ (mixed)	Sol-gel		92% in 120 min	0.0185 min ⁻¹ 1.58 times > TiO ₂	
TiO ₂	Sol-gel	UV LED (365 nm)	24.3% in 120 min	0.0023 min ⁻¹	This work
8 wt% WO ₃ /TiO ₂ (mixed)			94.9% in 120 min	0.0248 min ⁻¹ 10.78 times > TiO ₂	
Core-shell TiO ₂ @WO ₃ (10 wt% WO ₃)	Hydrothermal		95.8% in 60 min	0.0533 min ⁻¹ 23.17 times > TiO ₂	
Core-shell WO ₃ @TiO ₂ (10 wt% WO ₃)			82.5% in 120 min	0.0141 min ⁻¹ 6.13 times > TiO ₂	

* W. A. El-Yazeed and A. I. Ahmed, *Inorganic Chemistry Communications*, 2019, **105**, 102-111.

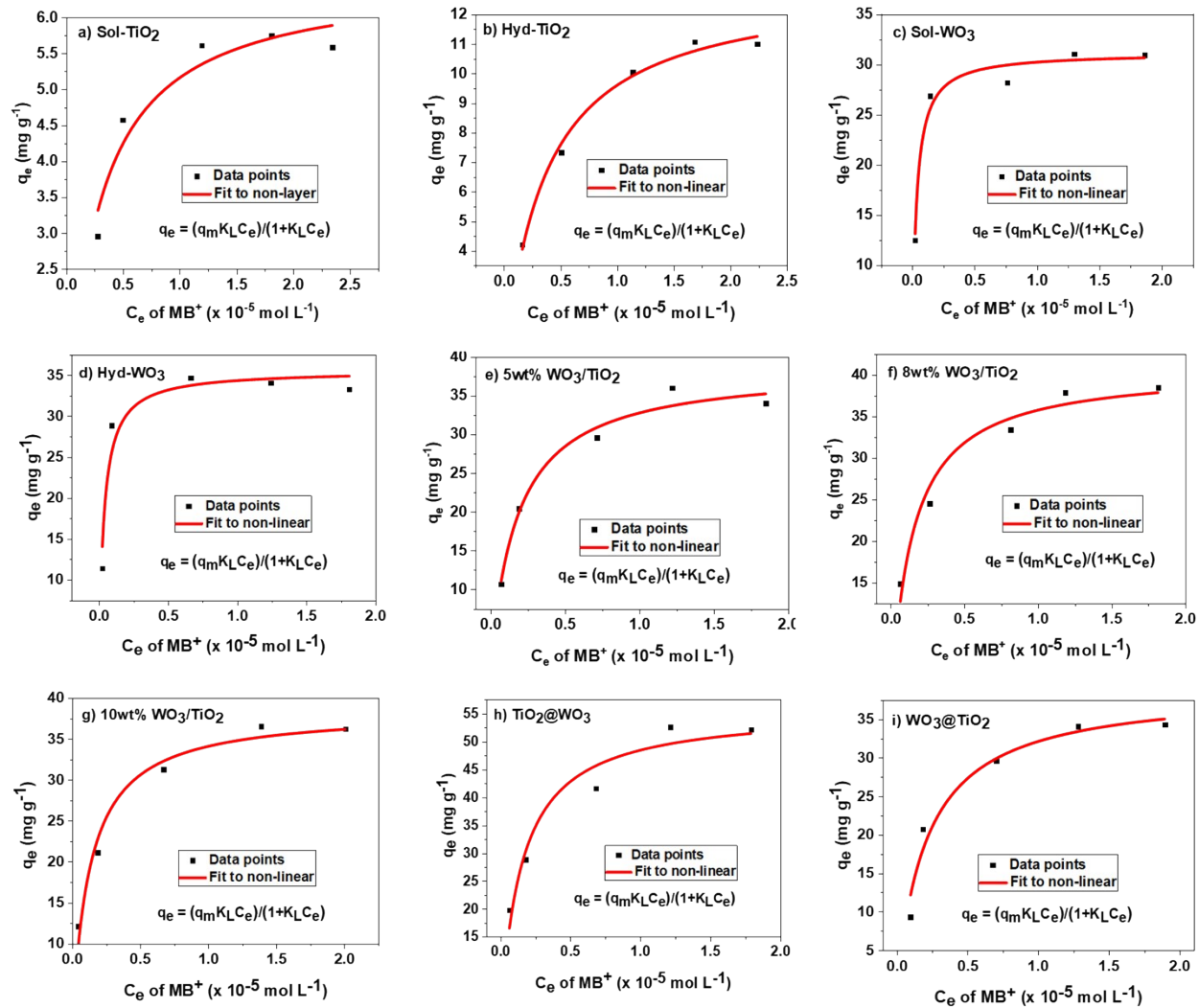


Figure S2. Adsorption behaviors of MB⁺ onto nanocomposites and non-linear Langmuir isotherm fitting.

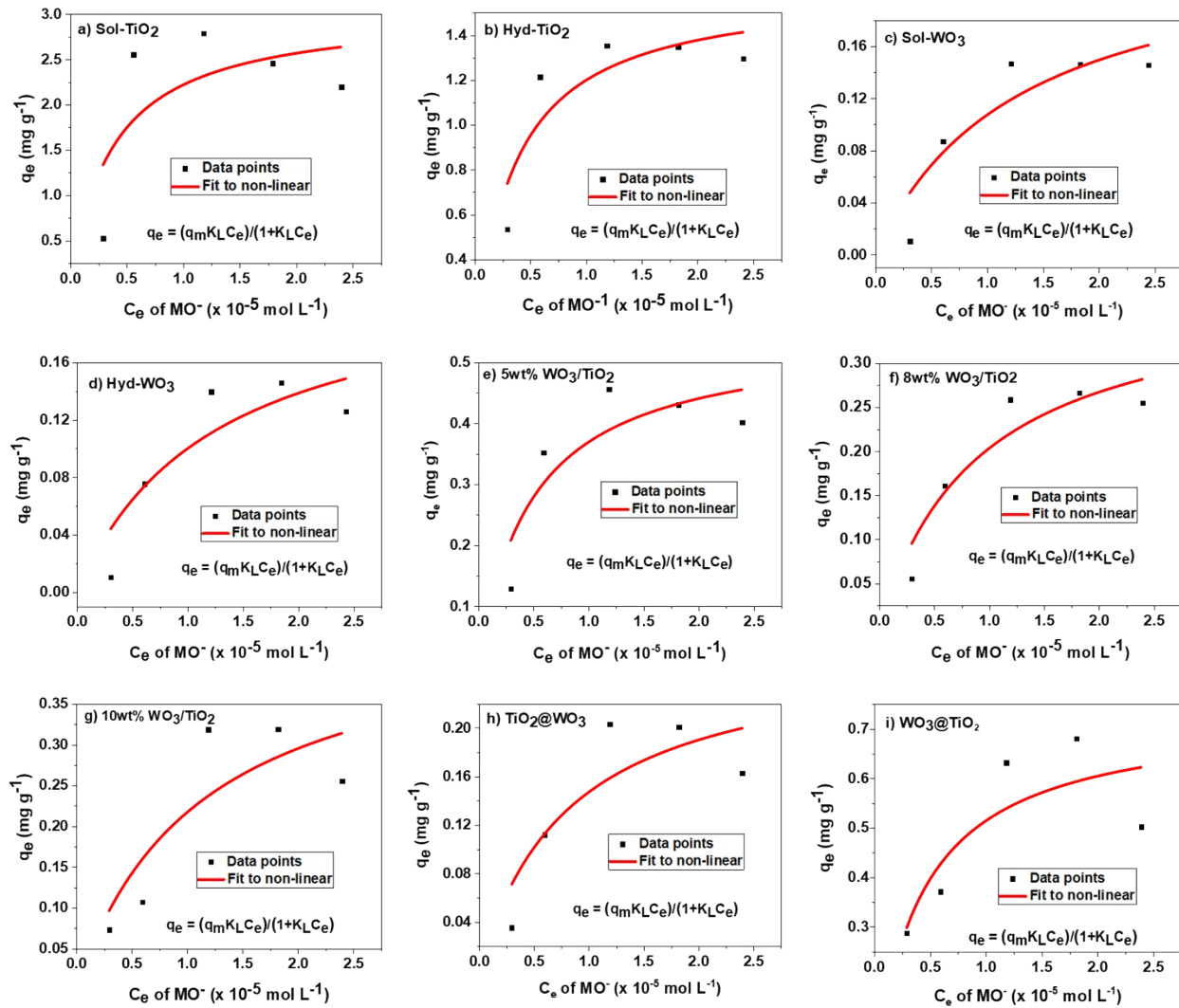


Figure S3. Adsorption behaviors of MO^- onto nanocomposites and non-linear Langmuir isotherm fitting.

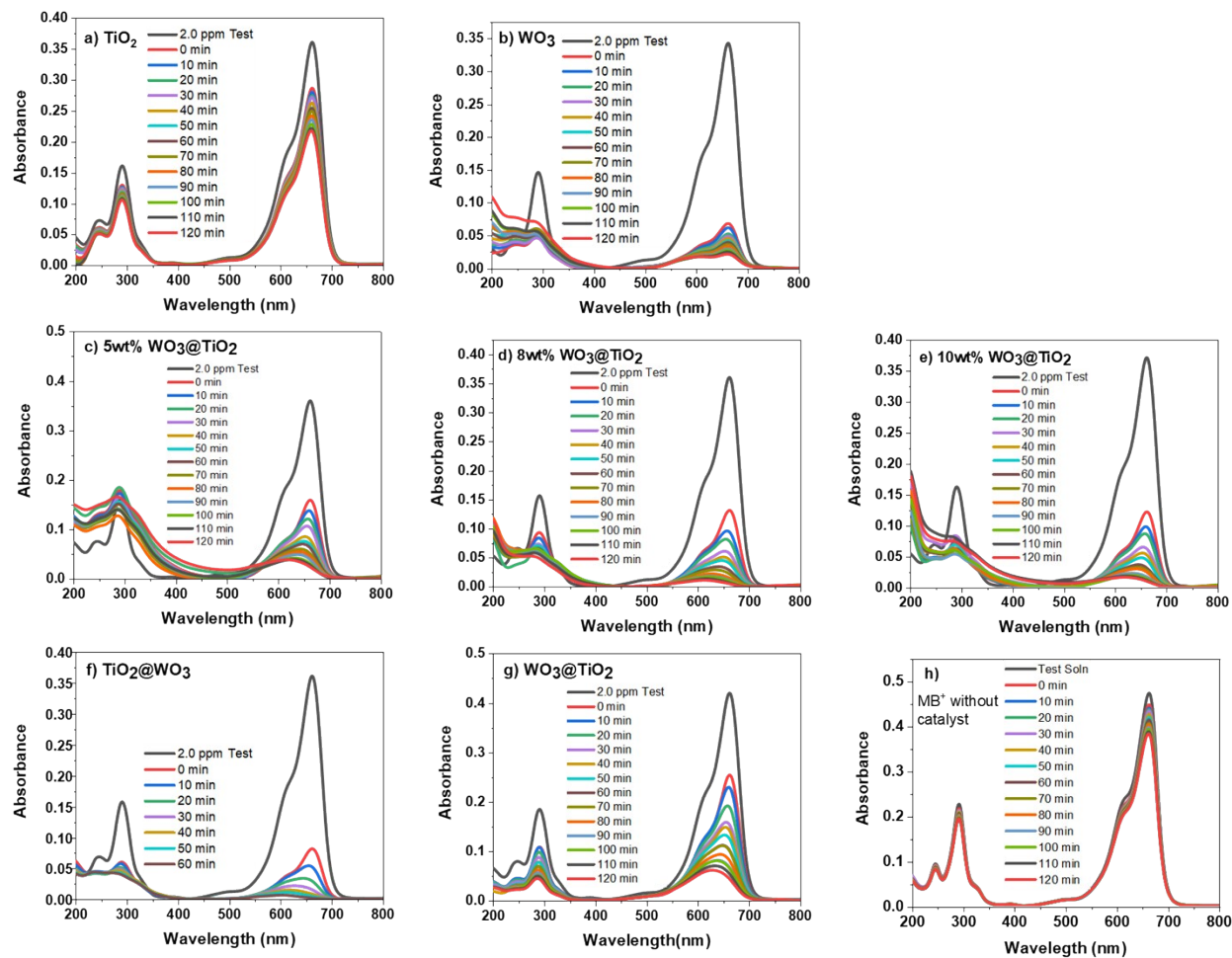


Figure S4. Absorption spectra of MB⁺ with various nanocomposites under UV light and without nanocomposite under UV light.

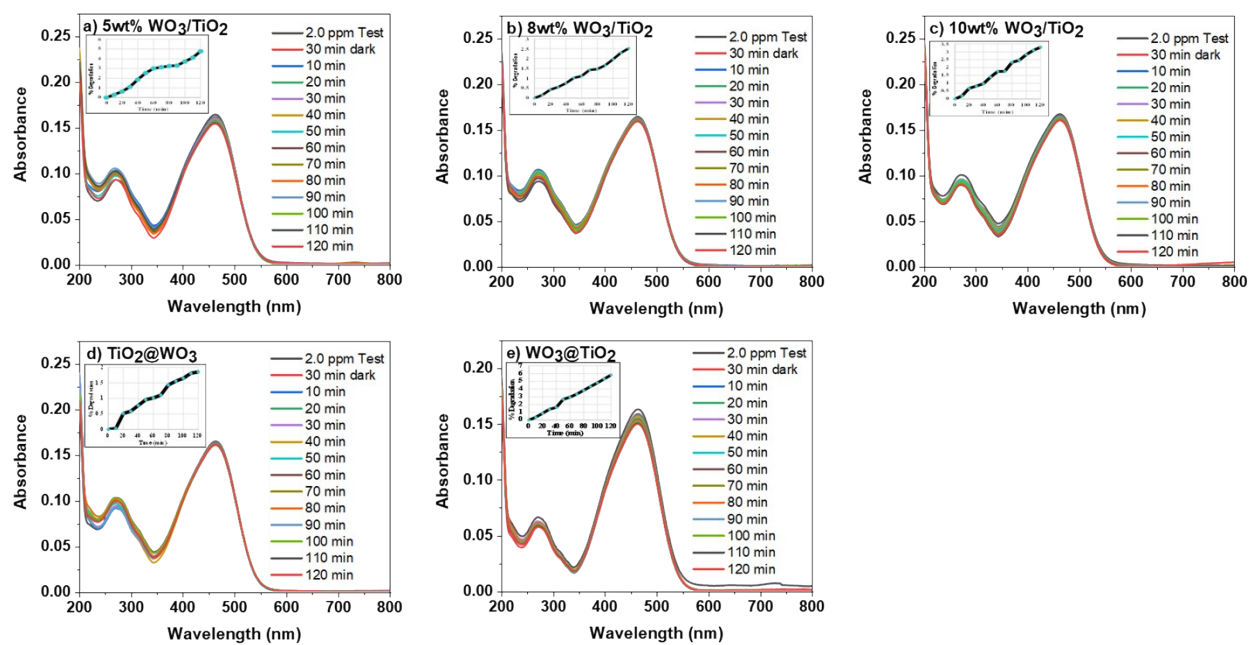


Figure S5. Absorption spectra of MO^- with various nanocomposites under UV light and without nanocomposite under UV light.