

Supplementary Information

A Rationale for Rapid Extraction of Ultra-low level Uranyl ion in Simulated Bioassays regulated by Mn-dopants over Magnetic Nano Particles

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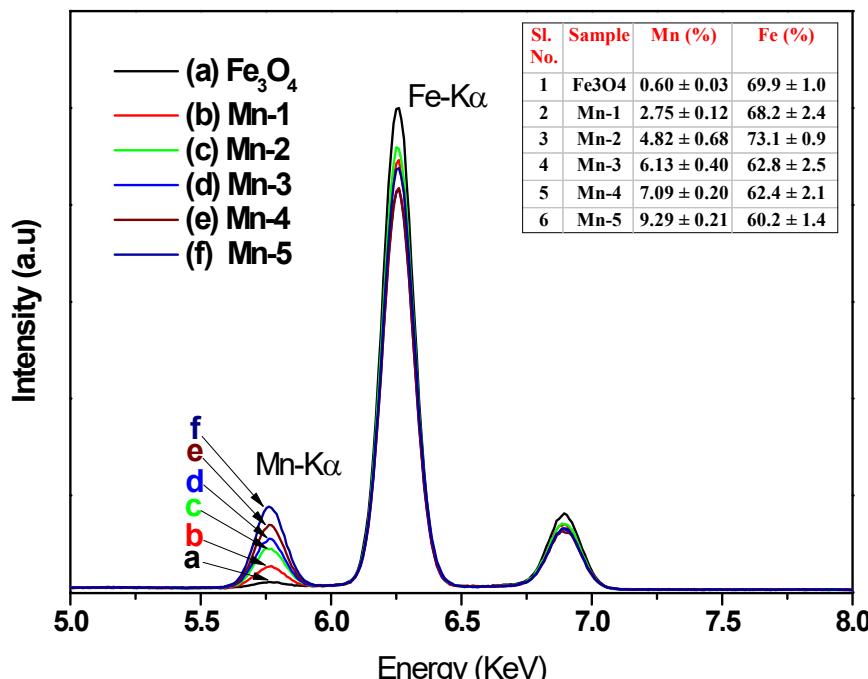


Fig.S-1: ED-XRF data for the different Mn-doped Fe_3O_4 system; a) undoped Fe_3O_4 ; b) Mn-1
c) Mn-2 d) Mn-3 e) Mn-4 f) Mn-5

Table S-1: Atomic Weight Percentage of Elements as calculated from EDS data

Sample	Fe	O	Mn
Fe_3O_4	32.7	67.3	0
Mn-1	30.6	68.1	1.3
Mn-2	29.7	67.8	2.5
Mn	27.6	67.7	4.7

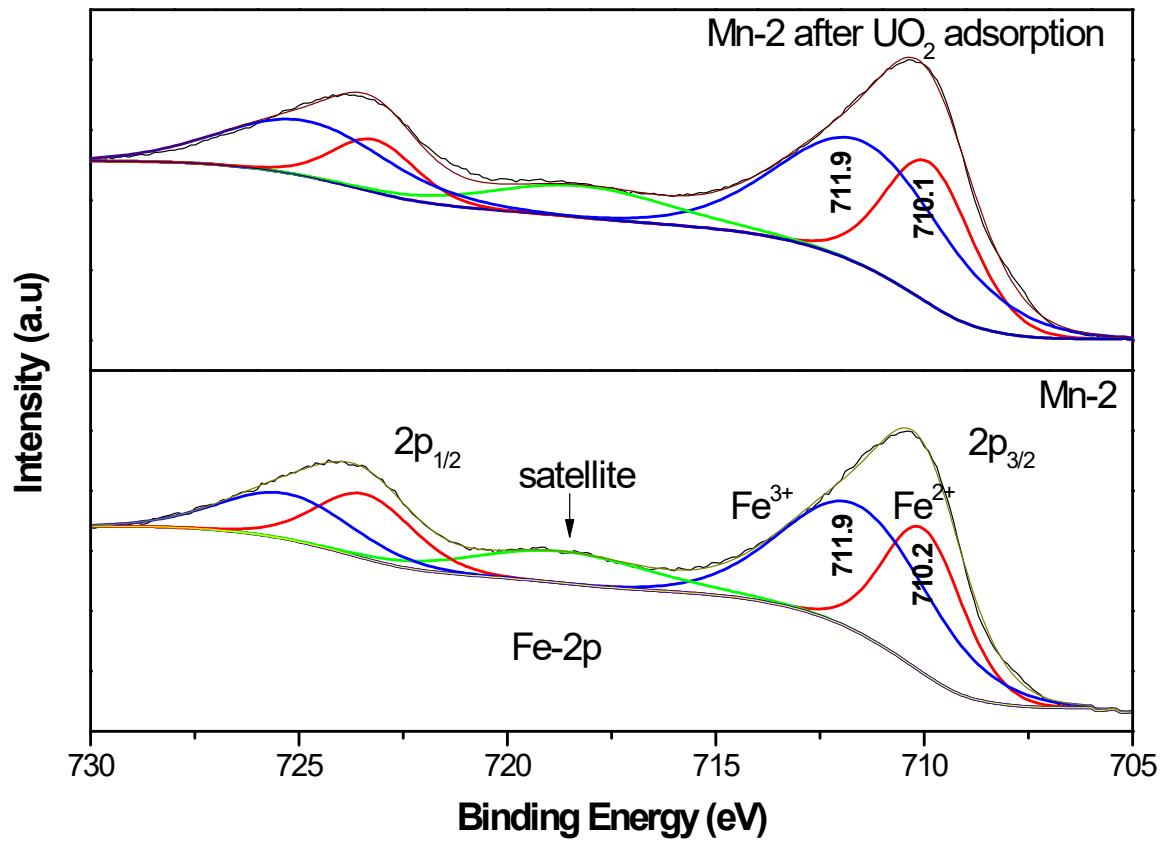


Fig. S-2: XPS Spectra for Fe-2p for comparison of the process of sorption interaction; a) Mn-2 before uranyl sorption and b) Mn-2 after uranyl sorption post drying.

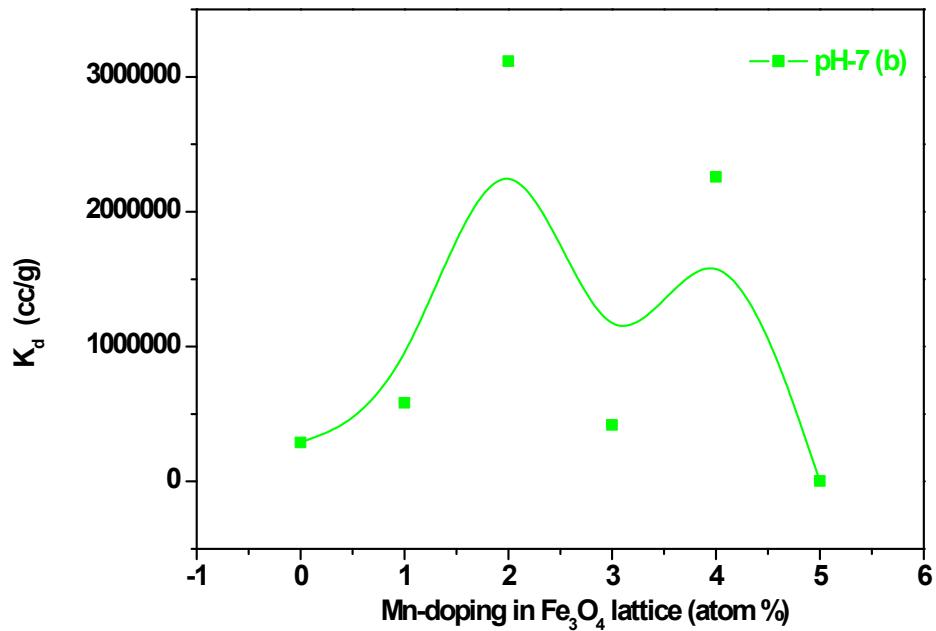


Fig S-3: Plot for K_d versus the Mn-doping (atom %) in the Fe_3O_4 lattice at neutral medium (pH-7).

Table S-2: SAED calculation for Fe₃O₄ system.

1/D (nm ⁻¹)	1/r (nm ⁻¹)	r (nm)	d-spacing (Å)	d-spacing (Å) (JCPDS-75-449)	h k l (JCPDS-75-449)
6.893	3.4465	0.2901	2.9015	2.9380	2 2 0
8.08	4.04	0.2475	2.4752	2.5055	3 1 1
9.655	4.8275	0.2071	2.0715	2.0775	4 0 0
12.66	6.33	0.1580	1.5798	1.5992	5 1 1
13.645	6.8225	0.1466	1.4657	1.4690	4 4 0
18.671	9.3355	0.1071	1.0712	1.1104	6 4 2

Table S-3: SAED calculation for Mn dopped Fe₃O₄ system (Mn-3).

1/D (nm ⁻¹)	1/r (nm ⁻¹)	r (nm)	d-spacing (Å)	d-spacing (Å) (JCPDS-75-449)	h k l (JCPDS-75-449)
6.913	3.4565	0.2893	2.8931	2.9380	220
7.992	3.996	0.2502	2.5025	2.5055	311
9.664	4.832	0.2070	2.0695	2.0775	400
12.843	6.4215	0.1557	1.5573	1.5992	511
13.627	6.8135	0.1468	1.4677	1.4690	440

Table- S-4: Optimisation of the K-points for the bulk calculation of Fe₃O₄

Sl. No.	K -Points	ΔE (eV)/atom	Magnetic Moment
1	1 1 1	-2.487617	59
2	3 3 3	-2.577881	48
3	5 5 5	-2.577813	48

Table -S-5: Comparison of Cell parameters for Fe_3O_4 and Mn-doped Fe_3O_4 system.

Sl. No.	Mn-doping	Cell Parameter ($a=b=c$) (\AA) Experimental	Cell Volume (\AA^3) Experimental	Cell Parameter ($a=b=c$) (\AA) Computational	Cell Parameter ($a=b=c$) (\AA^3) Computational
1	0 (Fe_3O_4)	8.346(2)	581.54(0.28)	8.43036	599.15173
2	Mn-1(Oct)	8.369(4)	586.24(0.51)	8.43877	600.94877
3	Mn-1(tet)	8.369(4)	586.24(0.51)	8.34808	581.78137

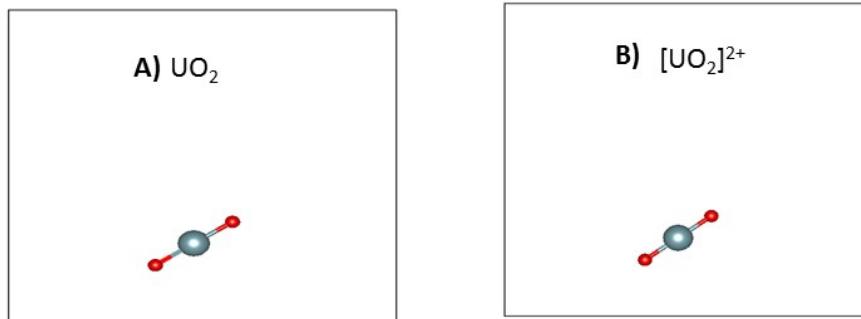


Fig. S-4: Optimised structures for A) UO_2 and B) $[\text{UO}_2]^{2+}$ - ion.

Table -S-6: Energy and bond Length for calculated UO_2 and $[\text{UO}_2]^{2+}$ - ion system.

Sample	Energy (eV)	U-O (length) (\AA)
UO_2	-28.745974	1.8201
$[\text{UO}_2]^{2+}$	-12.759074	1.7284

B. Calculation of Interaction Energy of $[\text{UO}_2]^{2+}$ ion with the Mn-doped Fe_3O_4 (3 1 1) Surface

Interaction Energy = Energy of $[\text{UO}_2]^{2+}$ adsorbed at particular site in Mn- doped Fe_3O_4 (3 1 1) surface - Slab energy Mn- doped Fe_3O_4 (3 1 1) surface
- Energy of $[\text{UO}_2]^{2+}$ ion(4)

Slab energy Mn-doped Fe_3O_4 = -824.80301 eV

Energy of $[\text{UO}_2]^{2+}$ ion = -12.759074 eV

$$\begin{aligned}\text{Interaction energy with } \text{Mn}^{2+} \text{ site} &= -854.61019 + [824.80301 + 12.759074] \text{ eV} \\ &= -17.048144 \text{ eV}\end{aligned}$$

$$\begin{aligned}\text{Interaction Energy with Fe-Tetrahedral } (\text{Fe}^{3+}) \text{ site} &= -853.83122 + \\ &\quad [824.80301 + 12.759074] \text{ eV} \\ &= -16.269136 \text{ eV}\end{aligned}$$

$$\begin{aligned}\text{Interaction energy with Fe-Octahedral } (\text{Fe}^{2+}) \text{ site} &= -853.85384 + \\ &\quad [824.80301 + 12.759074] \text{ eV} \\ &= -16.29179 \text{ eV}\end{aligned}$$

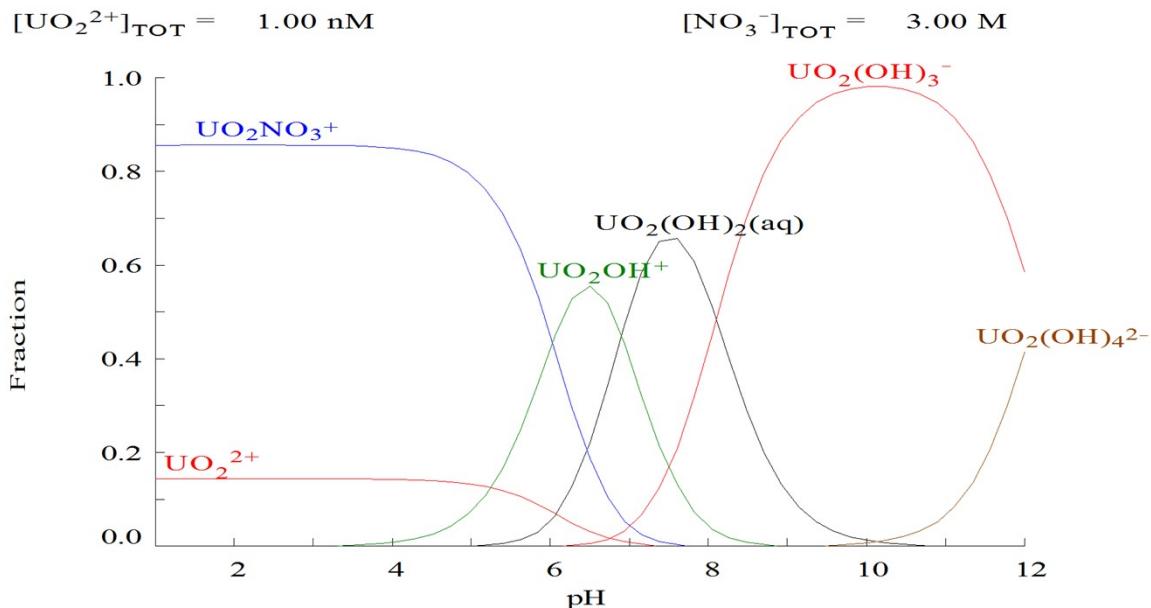


Fig.S-5: Speciation of Uranyl ion at different pH (Concentration of UO_2^{2+} = 1nM and NO_3^- = 3M) as suggested in the experimental section. The Figure is made by the MEDUSSA HYDRA- Software plot.