

3-HYDROXYFLAVONE IS A MILDLY ACTIVE AND SAFE COBALT CHELATOR WHILE COBALT MARKEDLY ENHANCES BAICALEIN TOXICITY TOWARD ERYTHROCYTES

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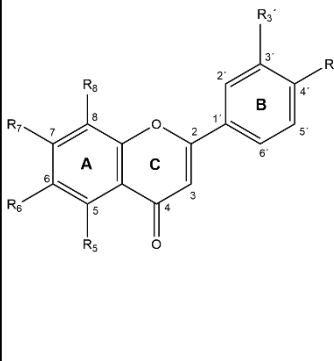
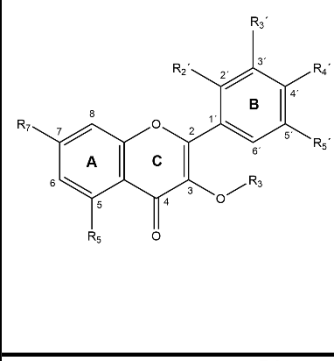
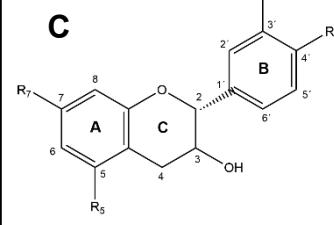
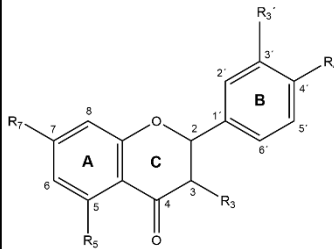
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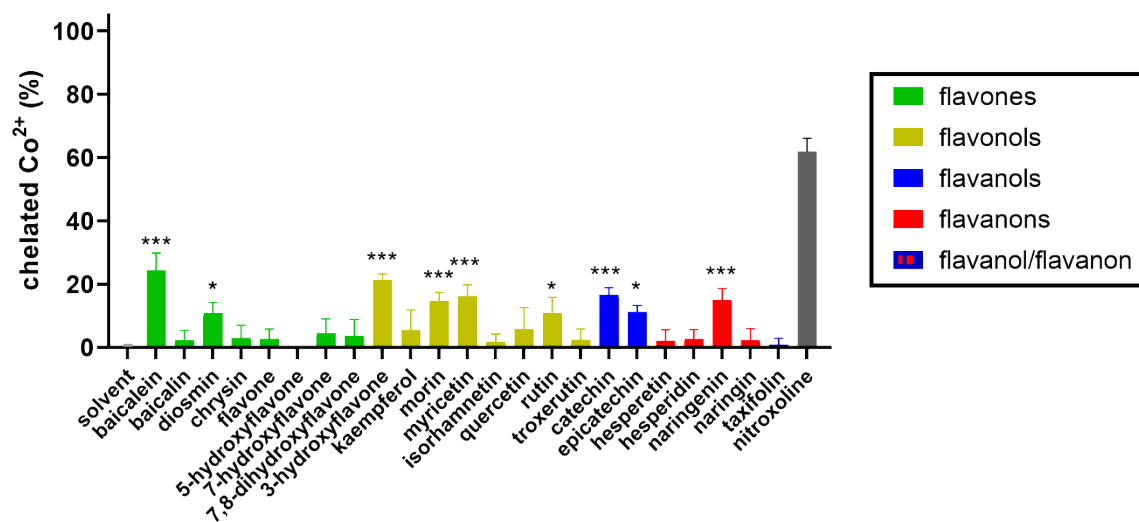
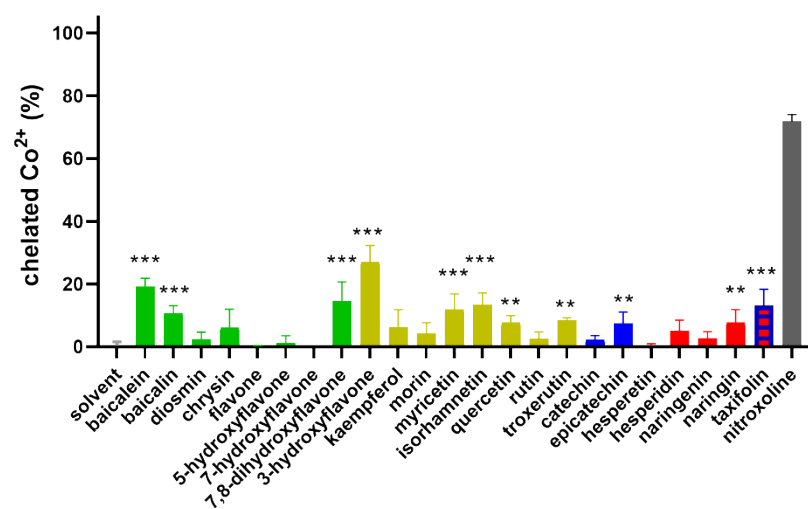
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SUPPLEMENTARY DATA

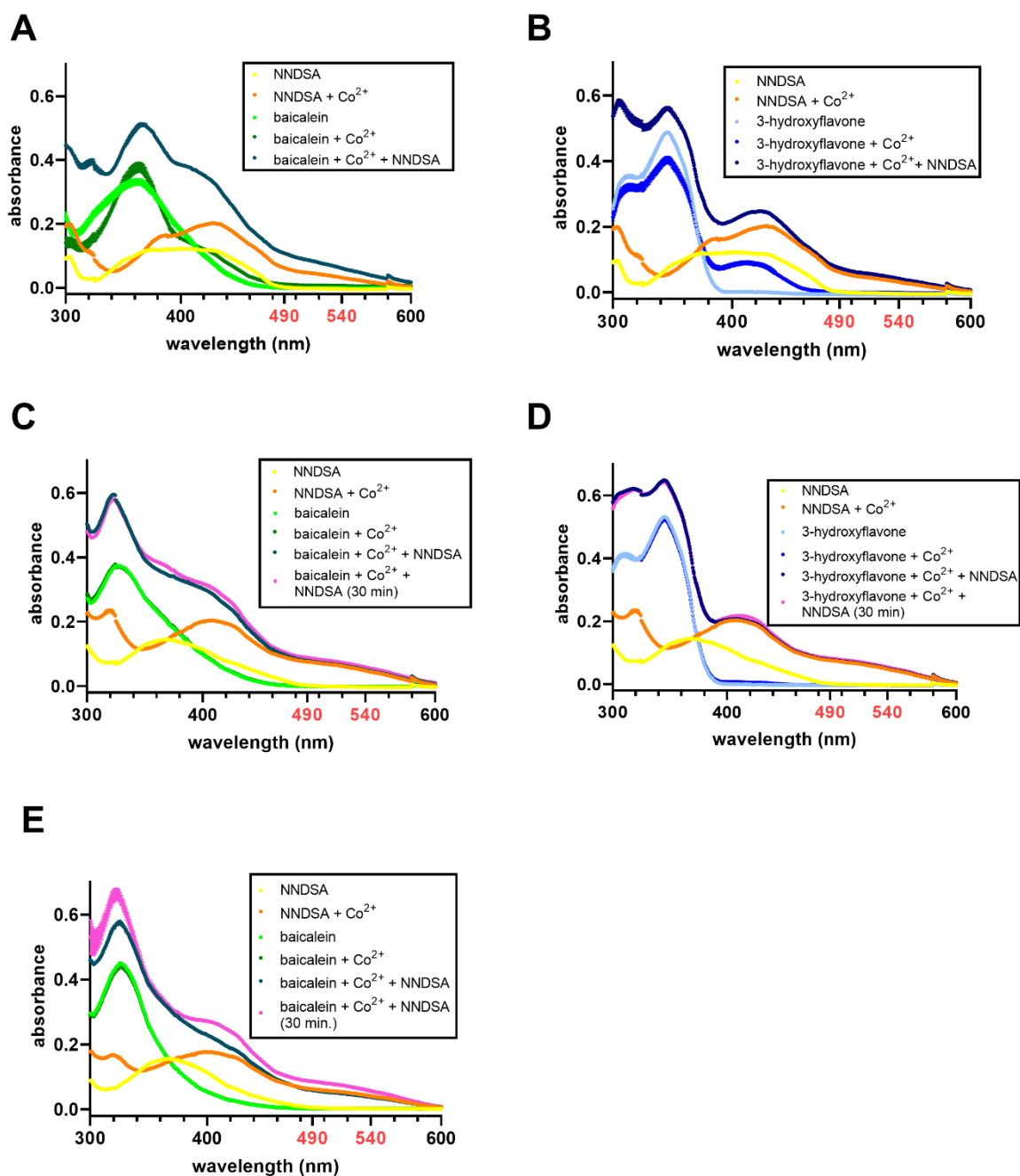
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A 	flavones	R₅	R₆	R₇	R₈	R₃'	R₄'	
	baicalein	OH	OH	OH	H	H	H	
	baicalin	OH	OH	O-Glu	H	H	H	
	diosmin	OH	H	O-Glc-Rha	H	OH	OCH ₃	
	chrysin	OH	H	OH	H	H	H	
	flavone	H	H	H	H	H	H	
	5-hydroxyflavone	OH	H	H	H	H	H	
	7,8-dihydroxyflavone	H	H	OH	OH	H	H	
B 	flavonols	R₃	R₅	R₇	R₂'	R₃'	R₄'	R₅'
	3-hydroxyflavone	H	H	H	H	H	H	H
	kaempferol	H	OH	OH	H	H	OH	H
	morin	H	OH	OH	OH	H	OH	H
	myricetin	H	OH	OH	H	OH	OH	OH
	isorhamnetin	H	OH	OH	H	OCH ₃	OH	H
	quercetin	H	OH	OH	H	OH	OH	H
	rutin	Glc-Rha	OH	OH	H	OH	OH	H
troxerutin	Glc-Rha	OH	O-C ₂ H ₄ -OH	H	O-C ₂ H ₄ -OH	O-C ₂ H ₄ -OH	H	
C 	flavanols	R₅	R₇	R₃'	R₄'	configuration		
	(+)-catechin	OH	OH	OH	OH	2R, 3S		
	(-)-epicatechin	OH	OH	OH	OH	2R, 3R		
D 	flavanones	R₃	R₅	R₇	R₃'	R₄'	configuration	
	hesperetin	H	OH	OH	OH	OCH ₃	2S	
	hesperidin	H	OH	O-Glc-Rha	OH	OCH ₃	2S	
	naringenin	H	OH	OH	H	OH	2RS	
	naringin	H	OH	O-Glc-Rha	H	OH	2RS	
	taxifolin*	OH	OH	OH	OH	OH	2R, 3R	

Suppl. Fig. 1 – Chemical structure of tested flavonoids: A: flavones; B: flavonols; C: flavanols; D: flavanones; *taxifolin can be classified as both, flavanol and flavanone; Glc – glucose, Rha – rhamnose, Glu – glucuronic acid. Structures were created by ChemDraw, version 20.0.

A**B**

Suppl. Fig. 2 – Chelation ability of tested flavonoids at pH 7.5 (A) or at pH 6.8 (B): The ratio of tested flavonoids:cobalt ions was 5:1. Data were analysed from measurement at a wavelength of 540 nm. Nitroxoline was used a positive control, while the solvent DMSO as negative control. All compounds were less active than nitroxoline at both conditions ($p < 0.001$, not shown in the graph). * $p < 0.05$ vs. solvent, ** $p < 0.01$ vs. solvent, *** $p < 0.001$ vs. solvent.



Suppl. Fig. 3 – Spectra of baicalein (A, C, E) and 3-hydroxyflavone (B, D) with/without cobalt and with the indicator NND SA at pH 7.5 (A, B), 6.8 (C, D) and 5.5 (E): The final concentration of cobalt was 50 μM , flavonoids 250 μM (ratio flavonoid:cobalt ions was 5:1), and NND SA 300 μM . Based on the spectra of flavonoids and their complexes with cobalt, no interference with the NND SA competitive method was noticed as there was no absorption at 490 or 540 nm. Spectra are averages of three independent experiments (different solutions).