

Supplementary Information

for

Fine Control for the Preparation of Ceria Nanorods (111)

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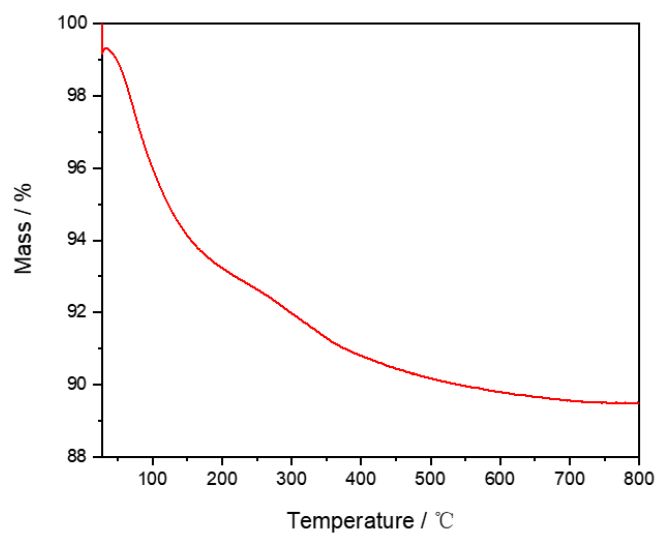


Figure S1. The thermogravimetric curve of the as-prepared product before high temperature calcination. The mass loss is found at a temperature lower than 400 °C, presumably due to dehydration at below 150 °C and dehydroxylation at 200 to 400 °C. The weight loss is not significant above 400 °C while a temperature as high as 700 °C is proved to be successful to transform high energy facets such as (100) and (110) to the thermodynamically stable (111) facets.¹ Thus a calcination temperature of 700 °C is used in the sample preparation process.

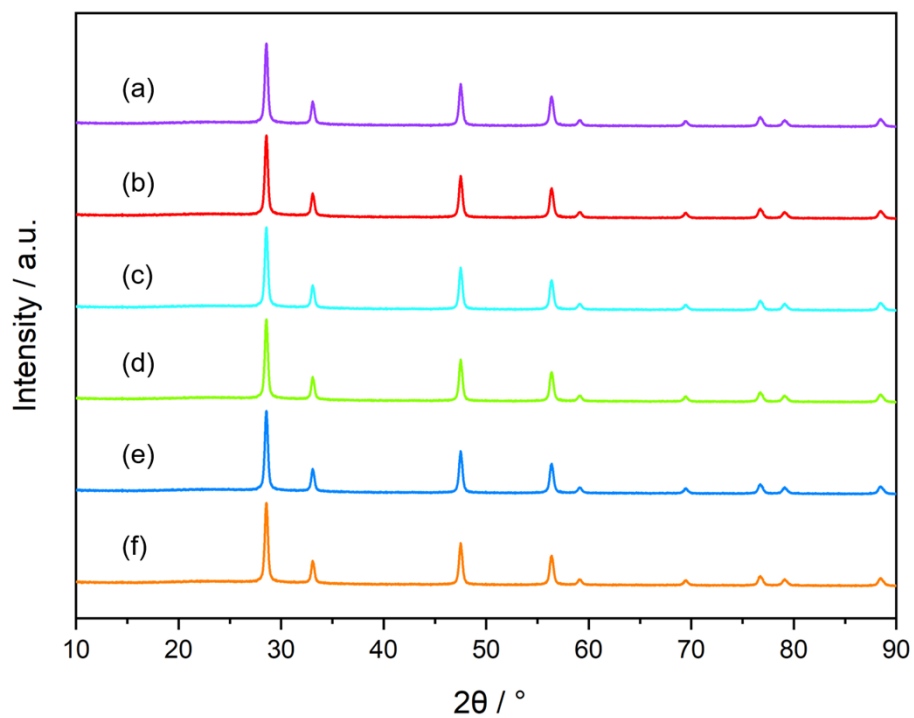


Figure S2. The XRD patterns of CeO_2 nanostructures obtained by pouring $0.05 \text{ mol}\cdot\text{L}^{-1}$ CeCl_3 solution to NaOH solutions with different concentrations. (a) $0.1 \text{ mol}\cdot\text{L}^{-1}$, (b) $1 \text{ mol}\cdot\text{L}^{-1}$, (c) $3 \text{ mol}\cdot\text{L}^{-1}$, (d) $4 \text{ mol}\cdot\text{L}^{-1}$ (e) $5 \text{ mol}\cdot\text{L}^{-1}$ and (f) $6 \text{ mol}\cdot\text{L}^{-1}$. Hydrothermal temperature: 100°C , time: 24 h. Cerium source: $0.05 \text{ mol}\cdot\text{L}^{-1}$ CeCl_3 solution.

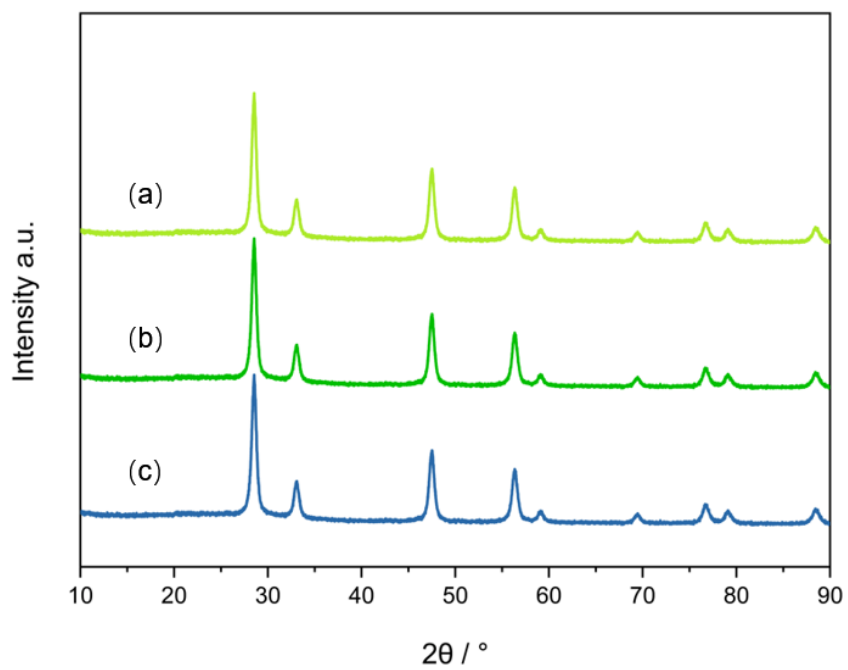


Figure S3. The XRD patterns of CeO₂ nanostructures obtained with different hydrothermal temperatures. (a) 100 °C, (b) 140 °C and (c) 180 °C. Hydrothermal time: 24 h. Cerium source: 0.05 mol·L⁻¹ CeCl₃ solution. NaOH solution concentration: 6 mol·L⁻¹.

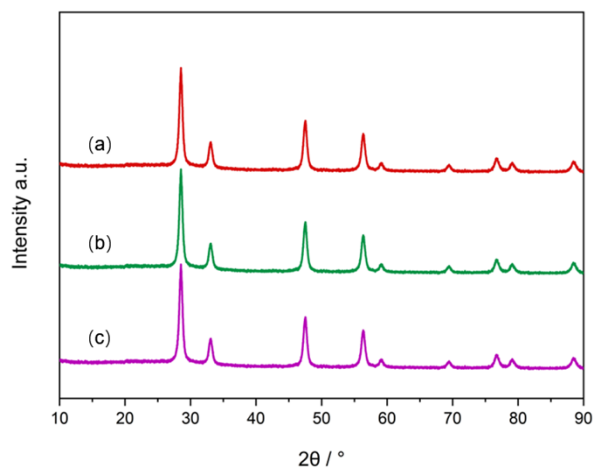


Figure S4. The XRD patterns of CeO_2 nanostructures obtained with different hydrothermal time. (a) 12 h, (b) 24 h and (c) 36 h. Hydrothermal temperature: 100°C . Cerium source: $0.05\text{ mol}\cdot\text{L}^{-1}$ CeCl_3 solution. NaOH solution concentration: $6\text{ mol}\cdot\text{L}^{-1}$.

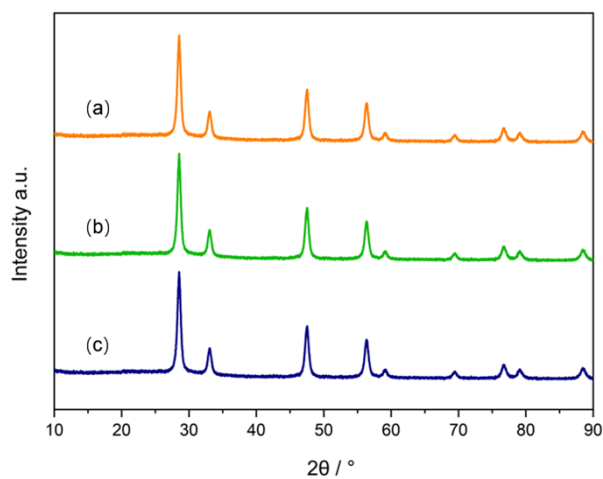


Figure S5. The XRD patterns of CeO_2 nanostructures obtained with different cerium sources. (a) $\text{Ce}(\text{NO}_3)_3$, (b) CeCl_3 , (c) $\text{Ce}(\text{NH}_4)_2(\text{NO}_3)_6$. Hydrothermal temperature and time: 100°C and 24 h. NaOH solution concentration: $6\text{ mol}\cdot\text{L}^{-1}$.

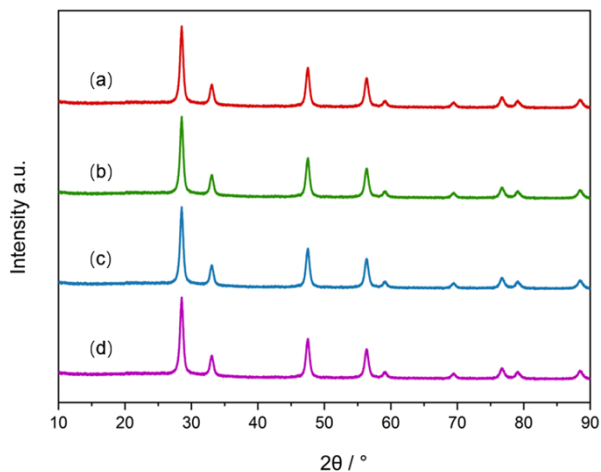


Figure S6. The XRD patterns of CeO₂ nanostructures obtained with different oxidants. (a) H₂O₂, (b) air (reaction time: 120 min), (c) air (reaction time: 30 min) and (d) trace oxygen in N₂ atmosphere. Hydrothermal temperature and time: 100 °C and 24 h. Cerium source: 0.05 mol·L⁻¹ CeCl₃ solution. NaOH solution concentration: 6 mol·L⁻¹.

References

1. B. Sudduth, D. Yun, J. Sun and Y. Wang, *J. Catal.*, 2021, **404**, 96-108.