

Supplementary information

Massively synthesizable Nickel-doped 1T-MoS₂ nanosheet catalyst as an efficient tri-functional catalyst

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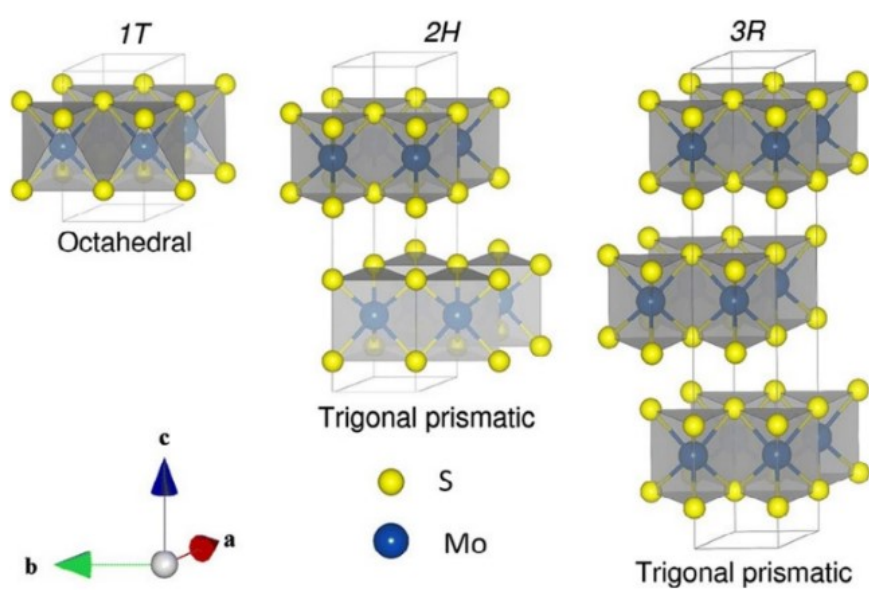


Figure S1. Crystal structure of MoS₂: Octahedral (1T), Trigonal prismatic (2H) and Trigonal prismatic (3R) unit cell structures. Reproduced with permission¹.

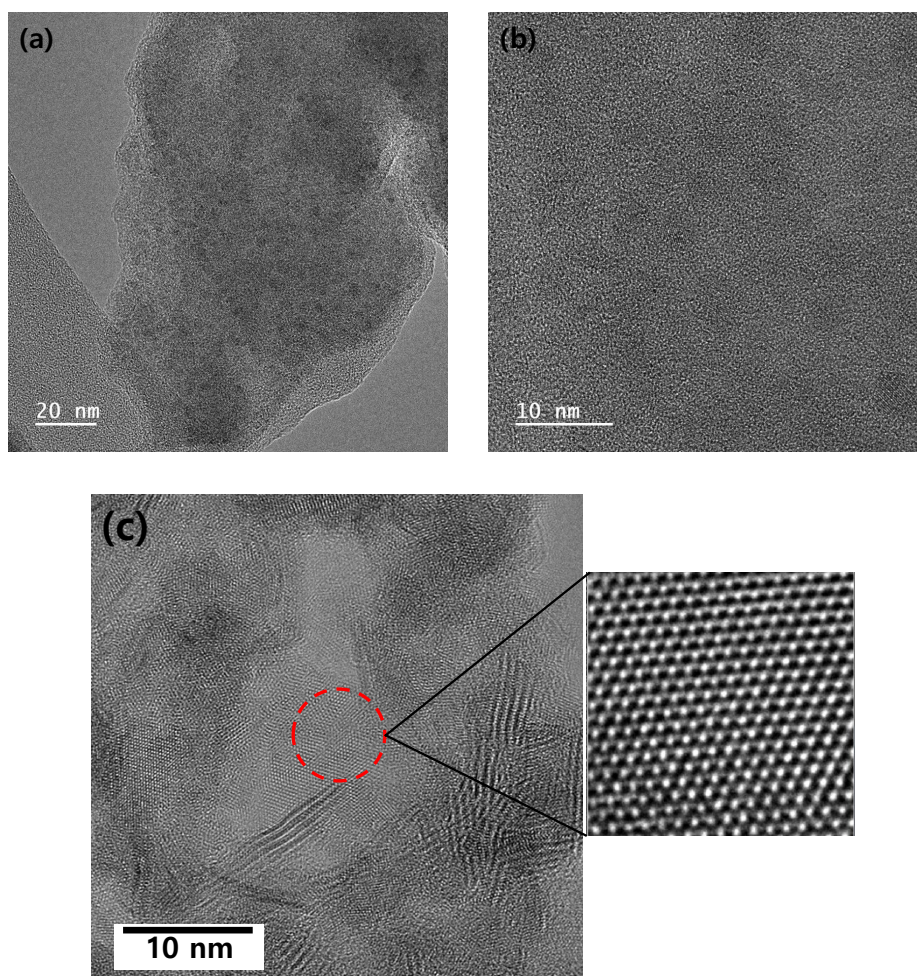


Figure S2. (a) Transmission electron microscopy (TEM) image of as-synthesized Ni-doped MoS₂ nanoparticle on the amorphous 2D Ni-doped MoS₂ sheet, (b) TEM image of Ni-doped MoS₂ crystalline nanoparticle on the amorphous 2D Ni-doped MoS₂ sheet, (c) TEM image of Ni-doped MoS₂ sheet after heat-treatment.

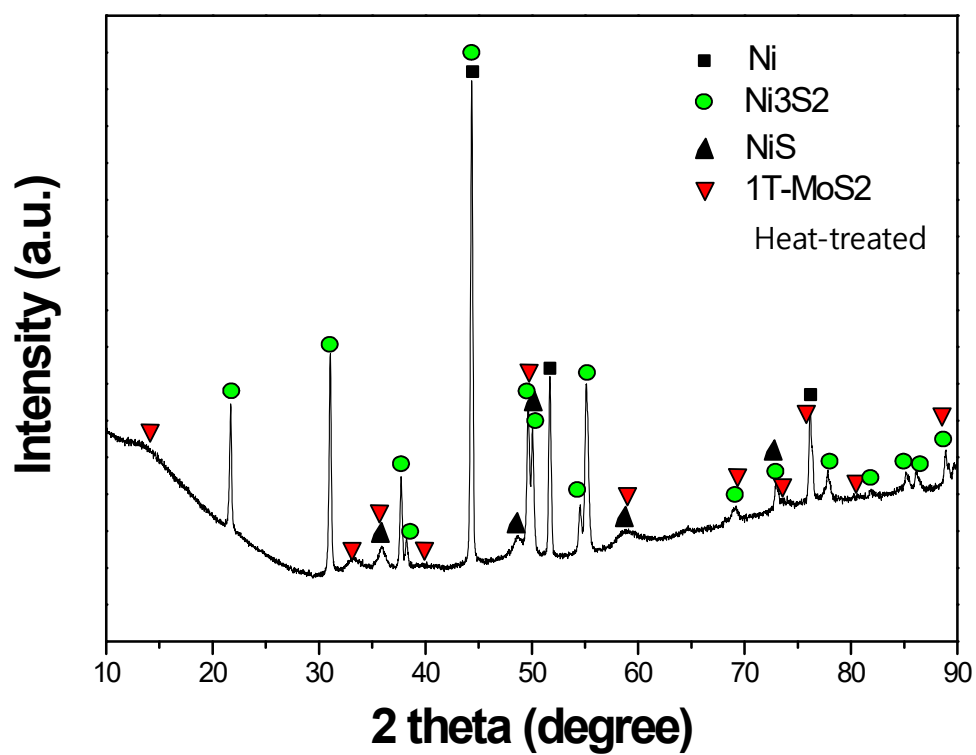


Figure S3. X-ray diffraction analysis result of the Ni-doped 1T-MoS₂

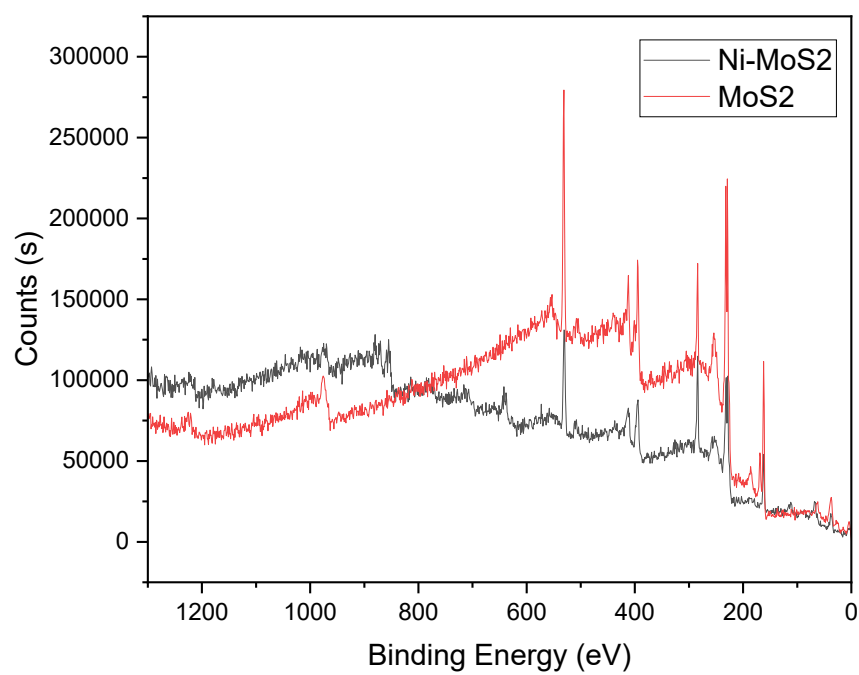


Figure S4. The XPS full spectrum for Ni-MoS₂ and MoS₂.

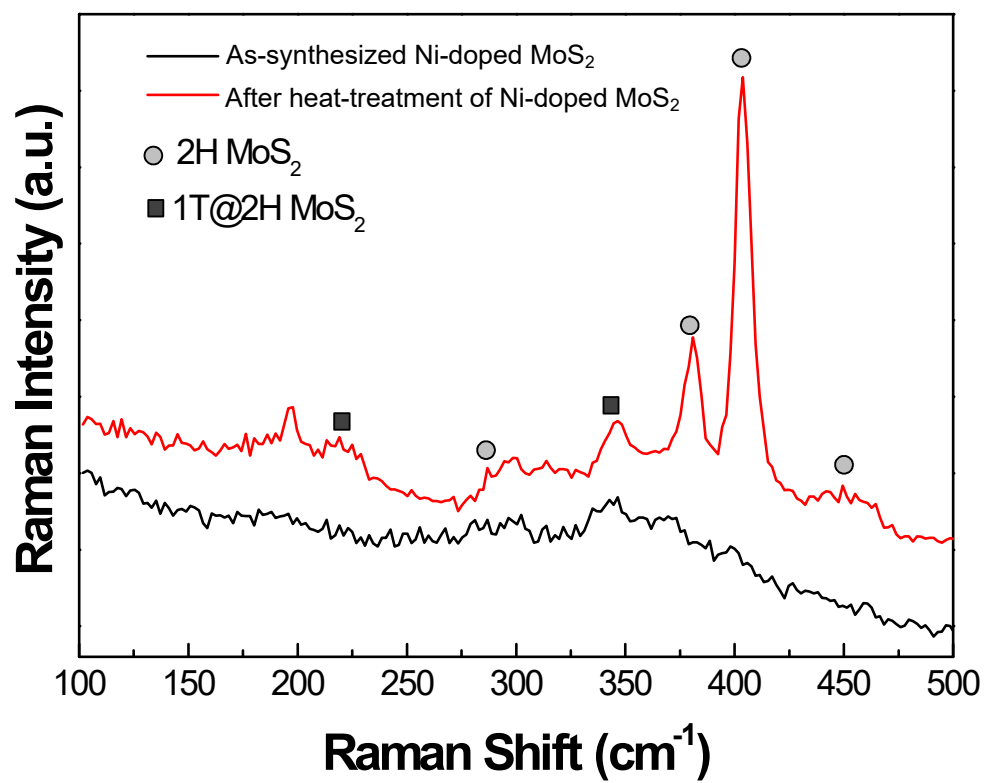


Figure S5. Raman analysis results of Ni-doped MoS_2 before and after heat-treatment

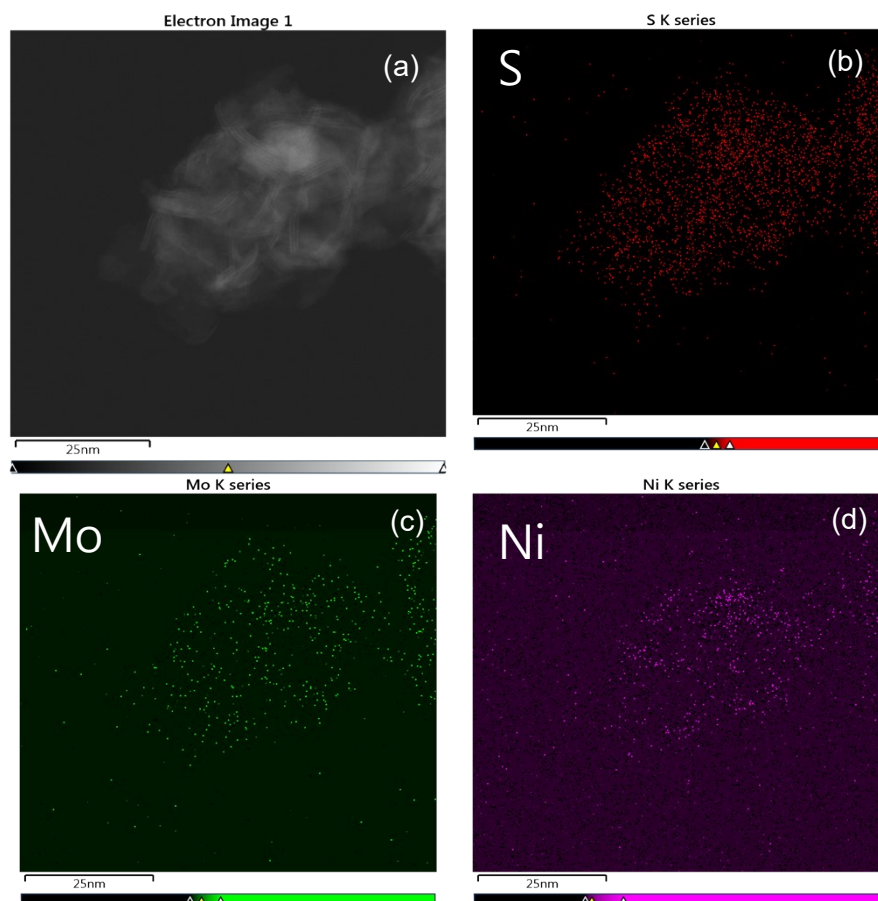


Figure S6. TEM EDS mapping results of Ni-doped 1T-MoS₂. (a) TEM image of Ni-doped MoS₂, (b) TEM EDS mapping result of Sulfur. (c) TEM EDS mapping result of Molybdenum, (d) TEM EDS mapping result of Nickel.

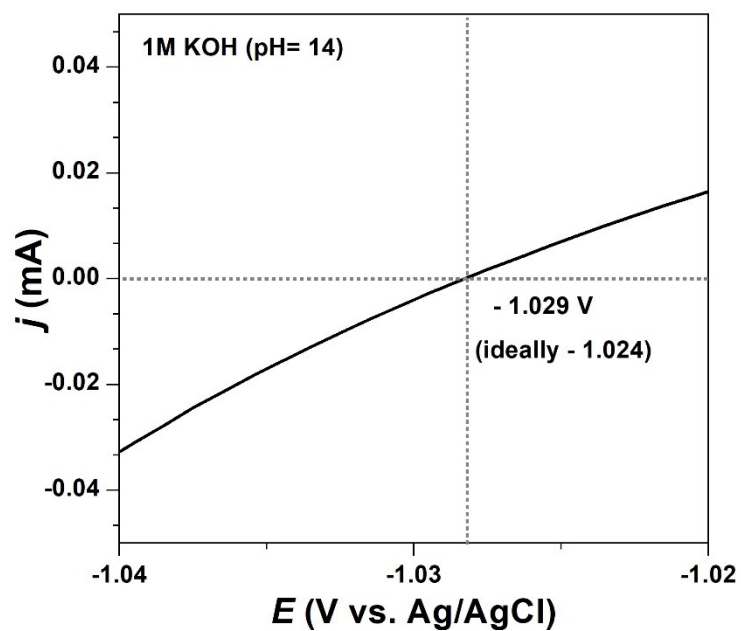


Figure S7. RHE calibration with Pt wires and Ag/AgCl (saturated KCl) reference electrode.

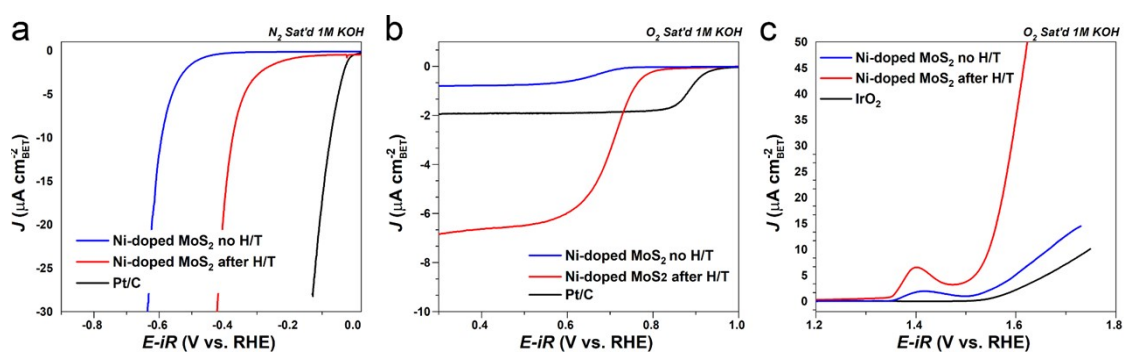


Figure S8. Specific activity normalized to the surface area as a function of potential for (a) HER, (b) ORR, and (c) OER process.

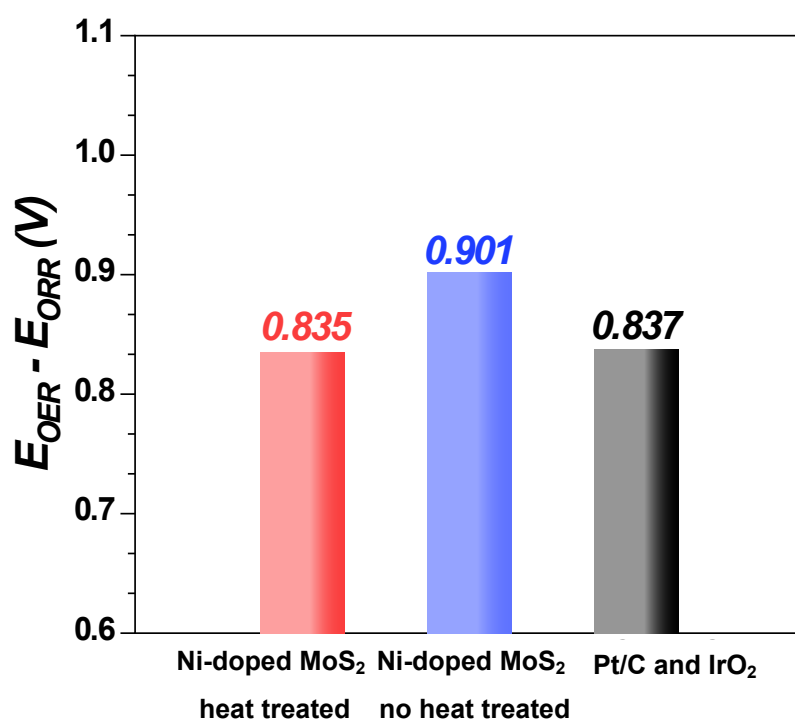


Figure S9. The oxygen evolution reaction (OER) and oxygen reduction reaction (ORR) polarization of Ni-doped 1T MoS₂ (before and after heat-treatment), and the commercial Pt/C and IrO₂.

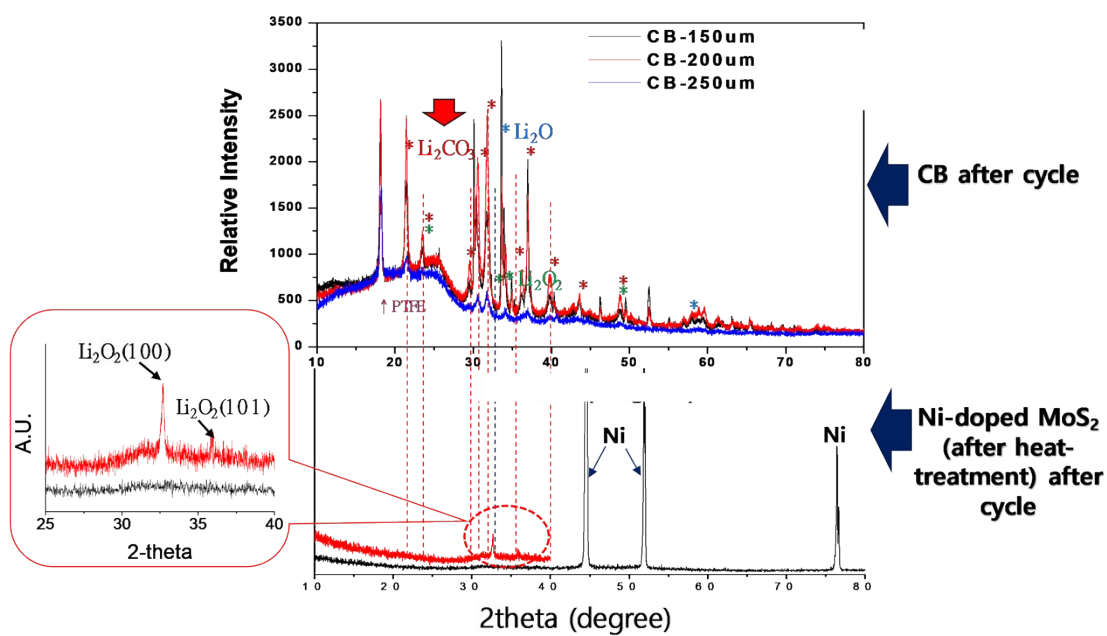


Figure S10. Ni-doped 1T MoS₂ showed no evidence of Li₂CO₃ after the cycle test

References

1. Kuc, A. 2015. Low-dimensional transition-metal dichalcogenides. Pp. 1–29 in M. Springborg, and J. Joswig, ed. Chemical modelling: Volume 11. The Royal Society of Chemistry, London, UK