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## **Characterization of organic release kinetics in particleboard by the dual model fitting methodology**

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**Table S1 Determination conditions of TD-GC-MS**

<b>TD</b>	
Adsorption tube type	Tenax-TA
Tube desorption temperature	280 °C
Tube desorption time	15 min
Cold trap temperature	Low: -20 °C; High: 300 °C
Cold trap desorption time	3 min
Transfer line temperature	150 °C
<b>GC</b>	
Injector temperature	270 °C
Column	HP-5MS, 30 m × 250 μm × 0.25 μm
Carrier gas	He
Constant flow	1.4 mL·min <sup>-1</sup>
Temperature program	The initial temperature was held at 30°C for 2 min, then heated at the rate of 30°C·min <sup>-1</sup> to 250°C and held for 1min., then heated at the rate of 6°C·min <sup>-1</sup> to 300°C and held for 1 min
<b>MS</b>	
Ion source EI	70 eV
Ion source temperature	200 °C
Quadrupole temperature	150 °C
Mass range Scan mode	10-1000 amu
Transfer line temperature	280 °C

**Table S2. Important detectable of particle board**

<b>Compound Name</b>	<b>Boiling Point (°C)</b>	<b>Vapor Pressure (mmHg)</b>	<b>Qualitative ion (m/z)</b>	<b>Quantitative ion (m/z)</b>
Methyl mercaptan	6	1536 ( 20 °C)	48	47
Isoprene	34	550 ( 25 °C)	68	67
Formic acid	100-101	52 ( 37 °C)	46	45
Dimethyl acetal	64	187.1 (25 °C)	75	59
Butenone	80	310 ( 55 °C)	70	55
Butyraldehyde	75	90 ( 20 °C)	72	57
Acetic acid	117-118	11.4 ( 20 °C)	60	45
Trichloromethane	61	160 ( 20 °C)	118,85	83
Tetrahydrofuran	66	<0.01 ( 25 °C)	72	71
Dichloroethane	82-84	83.9 (25 °C)	98	62
1,4-Dioxane	101	27 ( 20 °C)	88	58
Toluene	110-111	22 ( 20 °C)	92	91
Cyclohexanone	155	3.4 ( 20 °C)	98	55
Phenol	182	0.35 ( 25 °C)	94	66
2-Methylphenol	191	0.3 ( 20 °C)	108	107
Octanoic acid	237	1 ( 78 °C)	115	101
Naphthalene	218	0.03 ( 25 °C)	128	127
Nonanoic acid	268-269	<0.1 ( 20 °C)	158	129
p-tert-Butylphenol	236-238	1 ( 70 °C)	150	135
1-Methylnaphthalene	240-243	0.1 (25 °C)	142	141
Phthalic anhydride	284	<0.01 ( 20 °C)	148	104
Decanoic acid	268-270	15 ( 160 °C)	172	129
Dibutyl phthalate	340	1 ( 147 °C)	223,205	149

**Table S3 Biexponential model fitting parameters for toluene release**

Fitting parameter name	Fitting parameter value
Release rate for Phase 1	0.11
Decay constant for Phase 1	1.9
Release rate for Phase 2	0.012
Decay constant for Phase 2	0.003
Emission factor	$E(t)=0.11 * e^{-1.9t}+0.012 * e^{-0.003t}$

**Table S4 Mass transfer model fitting parameters for toluene release**

Parameters set in the experiment						
A(m <sup>2</sup> )	δ(m)	V(m <sup>3</sup> )	Q(m <sup>3</sup> /s)	β		
0.5	0.009	1	1/3600	0.0045		
Theoretical parameters derived from software						
D <sub>s</sub> (m <sup>2</sup> /s) <sup>#</sup>	K	h <sub>m</sub> (m/s)	a	Bi <sub>m</sub> /K	q	G <sub>1</sub>
3.39E-10	7.06E+02	0.000424	66.37	15.94	1.414	102.504
Parameters obtained by fitting the experimental data			The equation corresponding to the linear fit of the data			
C <sub>0</sub> (μg/m <sup>3</sup> )	D <sub>m</sub> (m <sup>2</sup> /s)		b=ln(3.06E-06*C <sub>0</sub> )			
1.7E+06	1.6E-11		k=-24683.9*D <sub>m</sub>			

# Software estimation value of D<sub>m</sub>, the same below.

**Table S5 Biexponential model fitting parameters for dimethyl acetal**

<b>Fitting parameter name</b>	<b>Fitting parameter value</b>
Release rate for Phase 1	0.0677
Decay constant for Phase 1	1.16
Release rate for Phase 2	0.0114
Decay constant for Phase 2	0.01
Emission factor	$E(t)=0.0677*e^{-1.16t}+0.0114*e^{-0.01t}$

**Table S6 Mass transfer model fitting parameters for dimethyl acetal**

<b>Parameters set in the experiment</b>						
A(m <sup>2</sup> )	$\delta$ (m)	V(m <sup>3</sup> )	Q(m <sup>3</sup> /s)	$\beta$		
0.5	0.009	1	1/3600	0.0045		
<b>Theoretical parameters derived from software</b>						
D <sub>s</sub> (m <sup>2</sup> /s)	K	h <sub>m</sub> (m/s)	a	Bi <sub>m</sub> /K	q	G <sub>1</sub>
2.56E-08	73.98	0.000506	0.88	2.4	0.746	1.17426
<b>Parameters obtained by fitting the experimental data</b>			<b>The equation corresponding to the linear fit of the data</b>			
C <sub>0</sub> ( $\mu$ g/m <sup>3</sup> )	D <sub>m</sub> (m <sup>2</sup> /s)		b=ln(7.4E-05*C <sub>0</sub> )			
4.3E+03	3.9E-10		k=-6870.6*D <sub>m</sub>			

**Table S7 Biexponential model fitting parameters for decanoic acid**

Fitting parameter name	Fitting parameter value
Release rate for Phase 1	0.022
Decay constant for Phase 1	0.26
Release rate for Phase 2	0.001
Decay constant for Phase 2	0.005
Emission factor	$E(t)=0.022*e^{-0.26t}+0.001*e^{-0.005t}$

**Table S8 Mass transfer model fitting parameters for decanoic acid**

Parameters set in the experiment						
A(m <sup>2</sup> )	δ(m)	V(m <sup>3</sup> )	Q(m <sup>3</sup> /s)	β		
0.5	0.009	1	1/3600	0.0045		
Theoretical parameters derived from software						
D <sub>s</sub> (m <sup>2</sup> /s)	K	h <sub>m</sub> (m/s)	a	Bi <sub>m</sub> /K	q	G <sub>1</sub>
4.28E-12	5.047E06	0.000352	5257	0.15	0.297	10351.8
Parameters obtained by fitting the experimental data			The equation corresponding to the linear fit of the data			
C <sub>0</sub> (μg/m <sup>3</sup> )		D <sub>m</sub> (m <sup>2</sup> /s)	b=ln(1.3E-09*C <sub>0</sub> )			
5.0E+08		2.3E-09	k=-1089*D <sub>m</sub>			

**Table S9 Biexponential model fitting parameters for phenol**

Fitting parameter name	Fitting parameter value
Release rate for Phase 1	0.0045
Decay constant for Phase 1	0.0048
Release rate for Phase 2	0.003
Decay constant for Phase 2	0.13
Emission factor	$E(t)=0.0045*e^{-0.0048t}+0.003*e^{-0.13t}$

**Table S10 Mass transfer model fitting parameters for phenol**

Parameters set in the experiment						
A(m <sup>2</sup> )	$\delta$ (m)	V(m <sup>3</sup> )	Q(m <sup>3</sup> /s)	$\beta$		
0.5	0.009	1	1/3600	0.0045		
Theoretical parameters derived from software						
D <sub>s</sub> (m <sup>2</sup> /s)	K	h <sub>m</sub> (m/s)	a	Bi <sub>m</sub> /K	q	G <sub>1</sub>
2.83E-10	2.47E+04	0.000439	79.5	0.565	0.534	151.763
Parameters obtained by fitting the experimental data			The equation corresponding to the linear fit of the data			
C <sub>0</sub> ( $\mu$ g/m <sup>3</sup> )		D <sub>m</sub> (m <sup>2</sup> /s)	b=ln(2.95E-07*C <sub>0</sub> )			
7.3E+06		3.98E-10	k=-3520.4*D <sub>m</sub>			