

Supplementary file

Table 1: accuracy and precision of UV-Vis spectroscopy method for curcumine

Standard solution concentration ($\mu\text{g/mL}$)	Accuracy	Precision (RSD %)
1	88.63849765	5
2	109.4209703	5.329774327
3	102.0552947	6.942986042
4	101.971831	6.555547774
5	100.9827856	3.851956345
6	98.23682838	6.77115196
7	99.40532081	3.566206767

Table 2. The type of kinetics equation explained the mechanism release of Cur from the PCL-T-M nano-system.

Model	Equation
First-order	F=100×[1-Exp(-k₁×t)]
zero-order	F=F₀+k₀ ×t
Higuchi	F=F₀+kH×t^{0.5}
Korsmeyer-Peppas	F=F₀+kKP×tⁿ
Hixson-Crowel	F=100×[1-(1-kHC×t)³]
Makoid-Banakar	F= kMB×tⁿ×Exp(-kt)
Weibull	F=100×{1-Exp[-(t^β)/α]}
Logistic	F=100×Exp[α+β×log(t)]/[1+Exp[α+β×log(t)]]
Gompertz	F=100×Exp{-α×Exp[-β×log(t)]}

Table 3. The parameter of the kinetics model at pH 7.4 and 5.5 from PCL-T-M-Cur.

Model	Parameter	PBS (7.4)	PBS (5.5)
First-order	k_1	0.008	0.026
	R^2	0.784	0.929
	AIC	123.91	120.41
Zero-order	k_0	0.461	0.757
	R^2	0.551	0.410
	AIC	138.39	156.58
Higuchi	kH	5.049	8.382
	R^2	0.964	0.935
	AIC	93.43	119.04
Korsmeyer-Peppas	kKP	6.704	13.037
	n	0.436	0.4
	R^2	0.973	0.9652
	AIC	88.99	109.35
Hixson-Crowell	kHC	0.002	0.007
	R^2	0.720	0.887
	AIC	128.361	128.453
Makoid-banakar	kMB	4.495	8.849
	n	0.600	0.563
	k	0.003	0.003
	R^2	0.992	0.988
	AIC	67.659	91.082
Weibull	α	18.04	12.401
	β	0.553	0.668
	R^2	0.987	0.988
	AIC	77.111	91.682
Logistic	α	-3.092	-2.926
	β	1.556	2.223
	R^2	0.993	0.980
	AIC	65.45	99.47
Gompertz	α	4.158	4.655
	β	0.904	1.519
	R^2	0.996	0.963
	AIC	56.66	110.105

