Supporting Information

Using phenolic polymers to control the size and morphology of calcium carbonate microparticles

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Polymers	$M_{ m n}$ ($ imes$ 104)	PDI
P1HS	7.2	1.35
P2HS	8.6	1.23
P3HS	8.0	1.20
P4HS	7.0	1.37
P5HS	14.0	1.68

Table S1. M_n and PDI of the phenolic polymers.

Polymer	Stirring speed (rpm)	Polymer concentration (mg/mL)	Mean value (μm)	Standard deviation (μm)
P1HS	500	0.1	4.53	0.79
P1HS	500	0.3	2.17	0.42
P1HS	500	0.5	2.03	0.42
P1HS	1000	0.1	2.58	0.55
P1HS	1500	0.1	1.76	0.31
P2HS	500	0.1	6.24	0.70
P2HS	500	0.3	6.00	0.62
P2HS	500	0.5	5.79	0.46
P2HS	1000	0.1	4.52	0.61
P2HS	1500	0.1	3.26	0.49
P3HS	500	0.1	7.45	0.81
P3HS	500	0.3	7.40	0.64
P3HS	500	0.5	5.79	0.60
P3HS	1000	0.1	6.33	0.57
P3HS	1500	0.1	4.09	0.41
P4HS	500	0.1	9.04	0.71
P4HS	500	0.3	6.05	0.91
P4HS	500	0.5	4.98	0.67
P4HS	1000	0.1	4.42	0.65
P4HS	1500	0.1	3.56	0.71
P5HS	500	0.1	7.16	0.65
P5HS	500	0.3	5.72	0.61
P5HS	500	0.5	4.94	0.59
P5HS	1000	0.1	4.42	0.65
P5HS	1500	0.1	3.56	0.71

Table S2. Sizes and standard deviations of $CaCO_3$ particles prepared under various conditions.



Fig. S1 (a) SEM image and (b) XRD pattern of CaCO₃ particles obtained in the absence of phenolic polymers.



Fig. S2 The optical microscope image of CaCO₃ particles synthesized in the presence of P4HS. Stirring speed is 500 rpm.



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Fig. S14 (a) SEM image and (b) XRD patterns of CaCO₃ particles prepared in the presence of P4HS ($M_n \sim 12$ kDa). The data for $M_n \sim 8$ kDa is the same as Fig. 1c but also shown here for comparison.



Fig. S15 (a) SEM images and (b) XRD patterns of CaCO₃ particles prepared in the presence of P4HS at different scale of reaction.