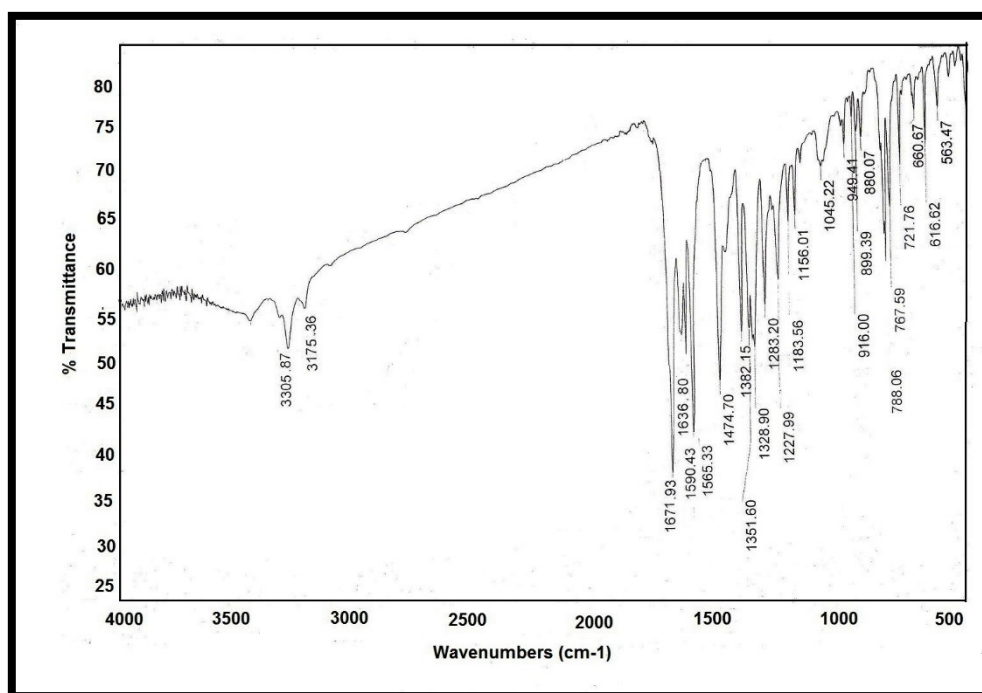
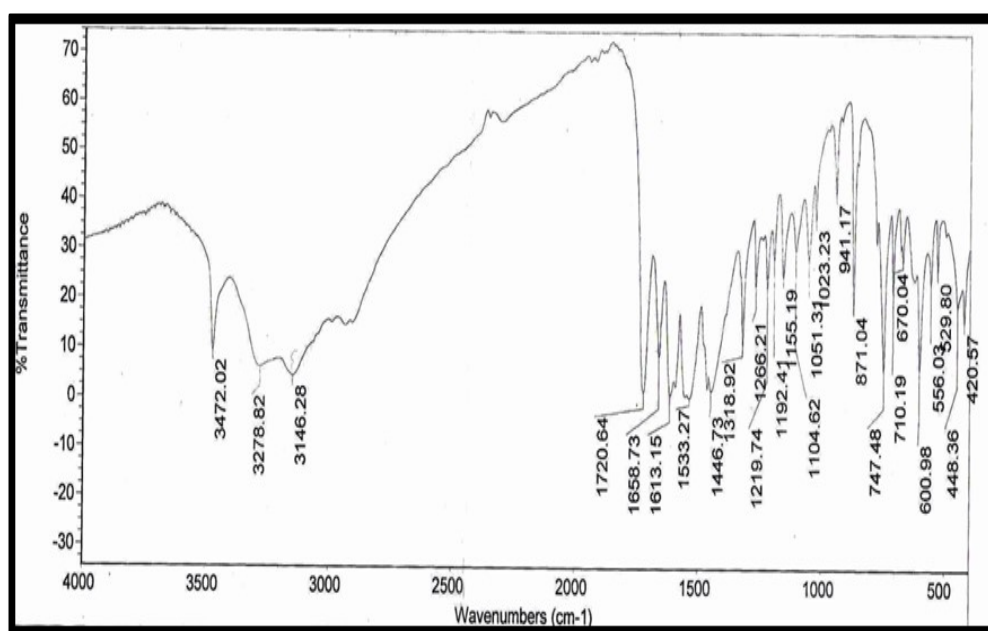
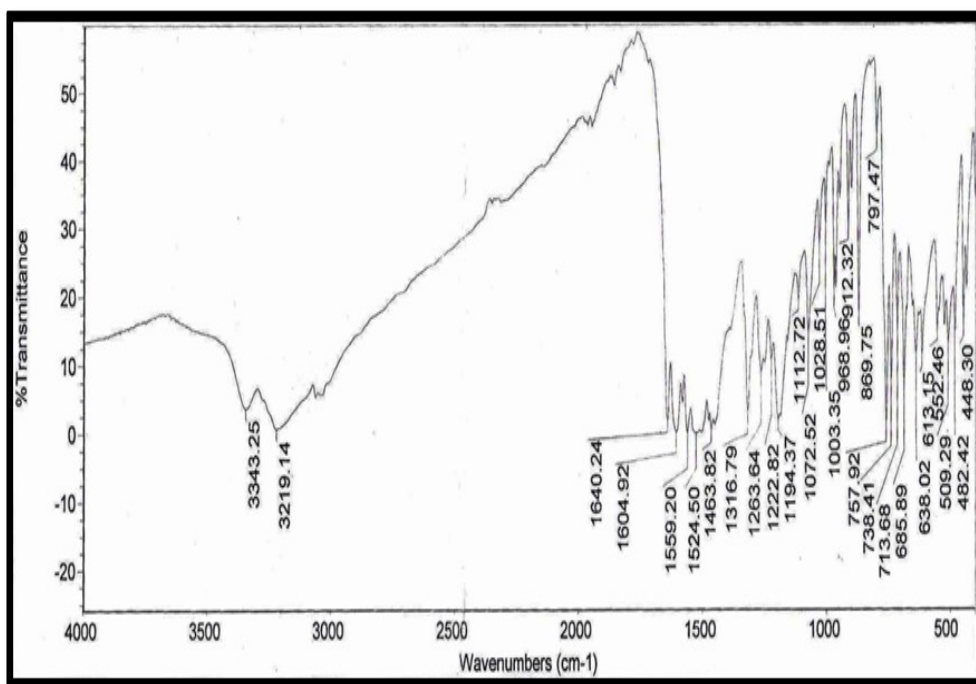




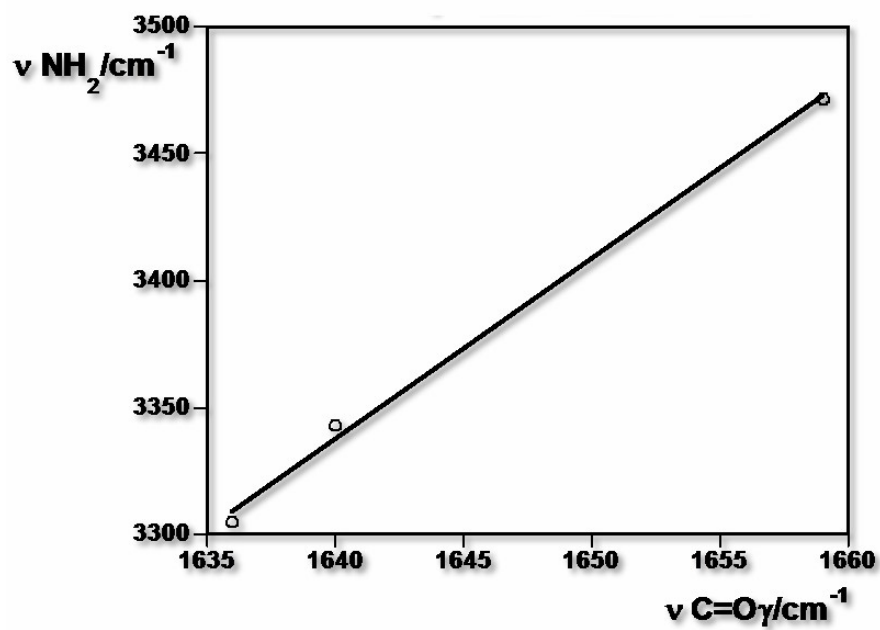
Supplementary Fig 1. Infrared spectrum of ACC.



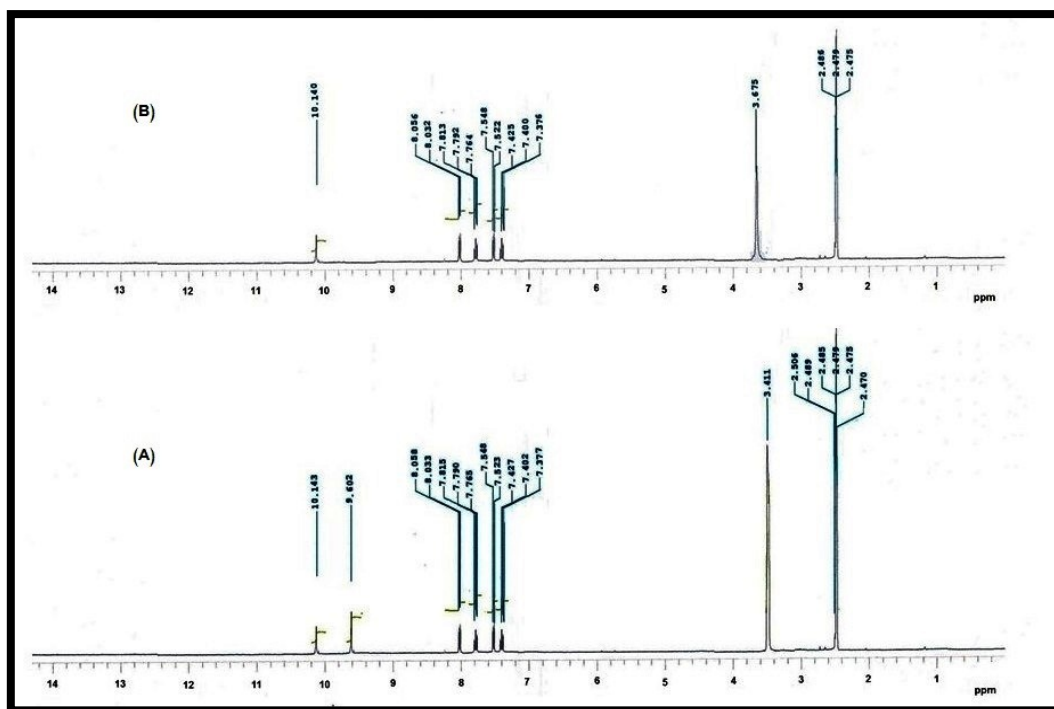
Supplementary Fig 2. Infrared spectrum of ACMHCA.



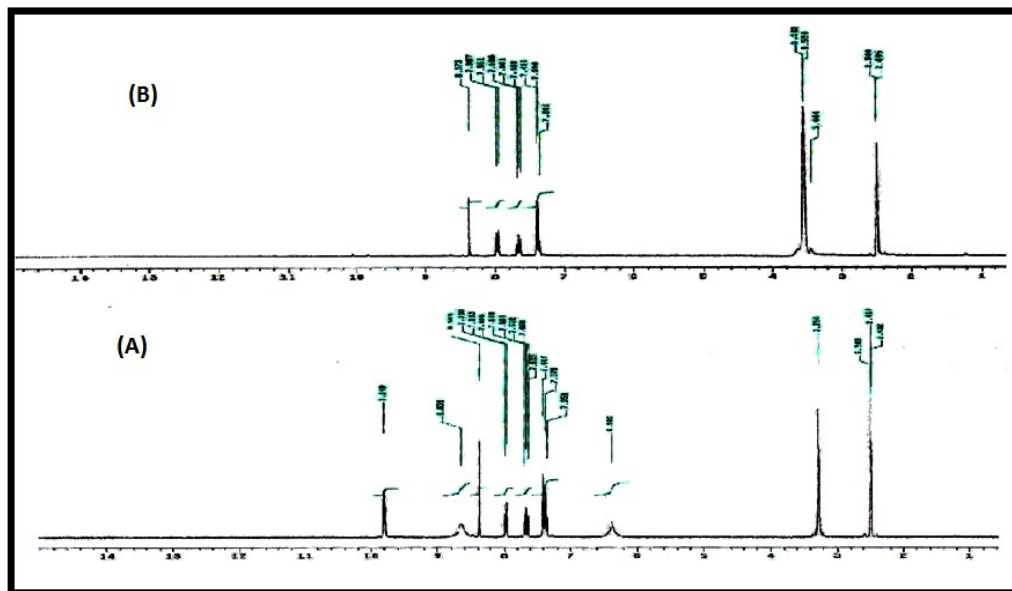
Supplementary Fig 3. Infrared spectrum of ACMNPHTCA.



Supplementary Fig. 4. The relation between $\nu \text{ NH}_2 / \text{cm}^{-1}$ and $\nu \text{ C=O}_{\gamma\text{-pyrone}}$



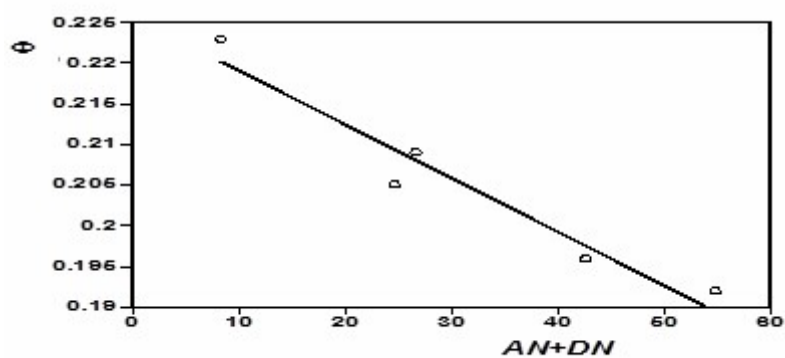
Supplementary Fig. 5. ^1H -NMR spectra of ACC in $(\text{DMSO}-d_6)$ (A) in absence of D_2O and (B) in presence of D_2O .



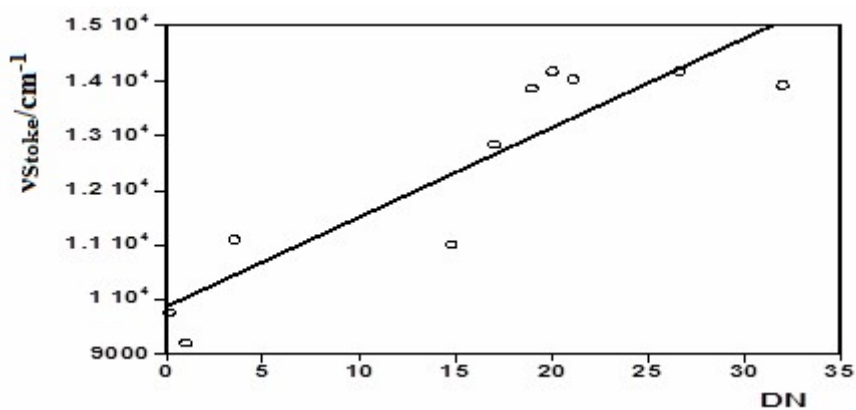
Supplementary Fig. 6. ^1H -NMR spectra of ACMHCA in $(\text{DMSO}-d_6)$ (A) in absence of D_2O and (B) in presence of D_2O .



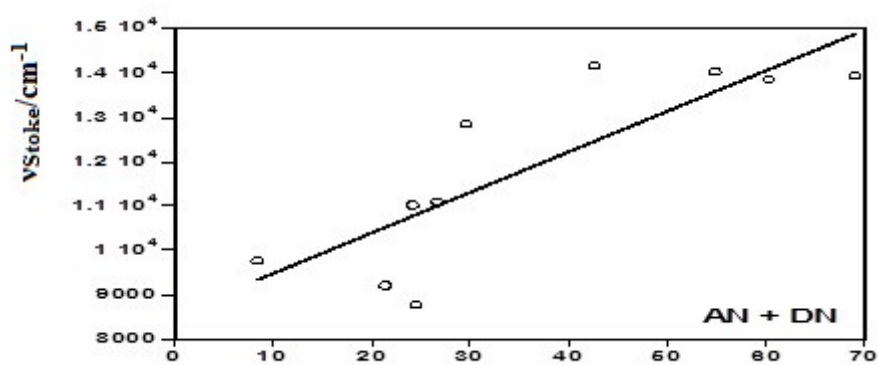
Supplementary Fig. 7. ¹H-NMR spectra of ACMNPHTCA in (DMSO-*d*₆) (A) in absent of D₂O and (B) in presence with D₂O.



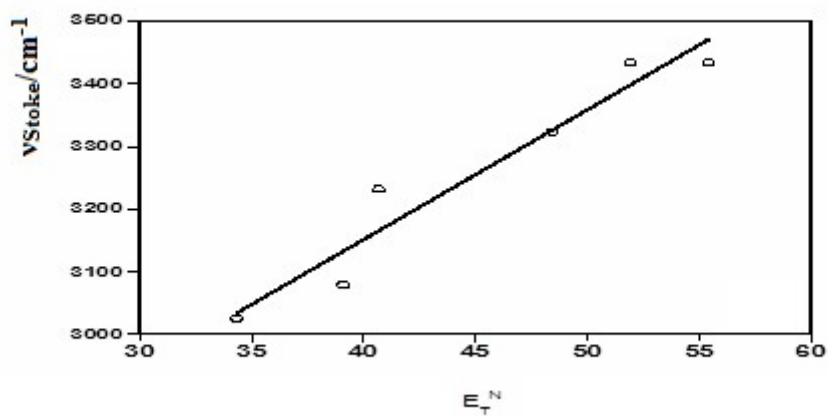
Supplementary Fig. 8. The relation between ϕ_L vs ($AN + DN$)



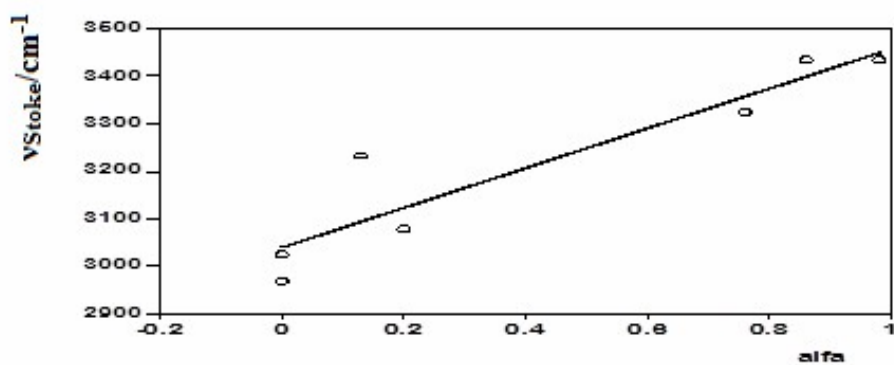
Supplementary Fig. 9. The relation between DN vs ν_{Stoke} shift of ACC.



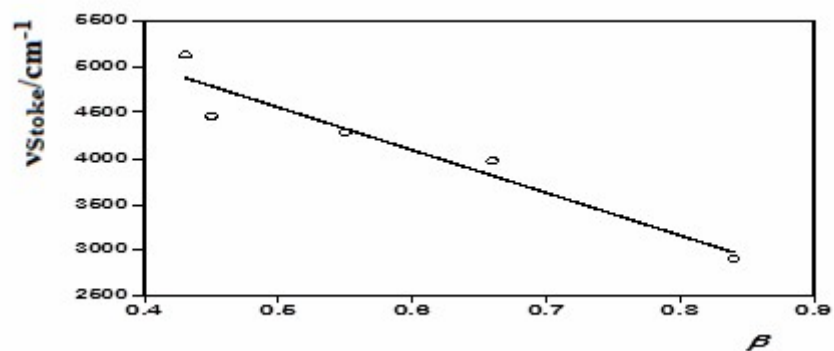
Supplementary Fig. 10. The relation between $\nu_{\text{Stokes shift}}$ vs (AN+DN) of ACC.



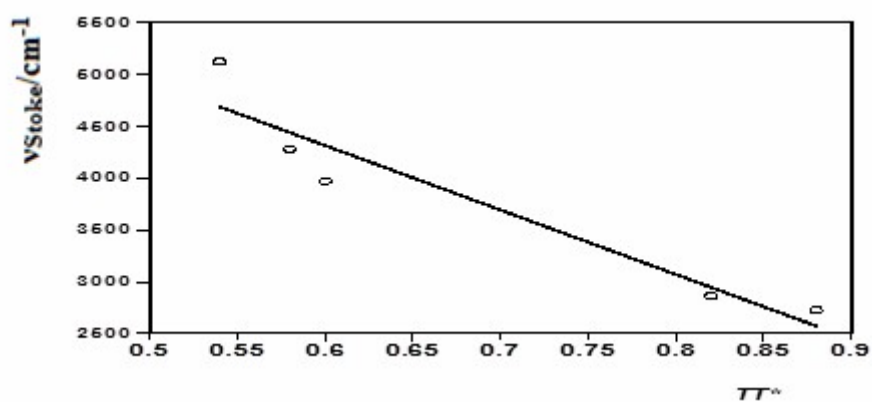
Supplementary Fig. 11. The relation between $\nu_{\text{Stoke shift}}$ vs E_T^n of ACMHCA



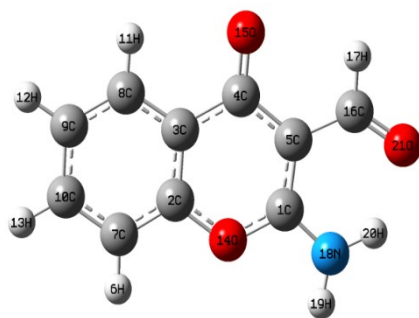
Supplementary Fig. 12. The relation between $\nu_{\text{Stoke shift}}$ vs α of ACMHCA



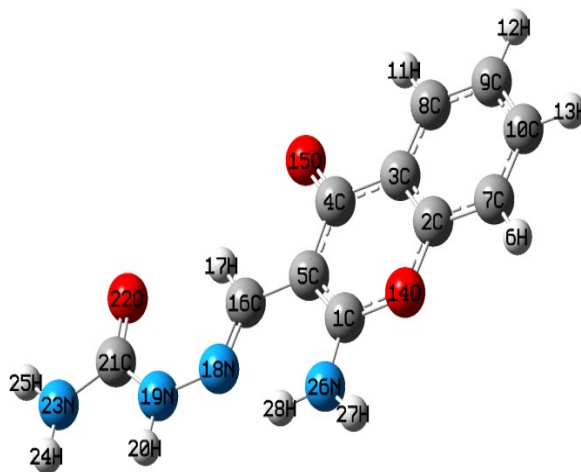
Supplementary Fig. 13. The relation between $\nu_{\text{Stoke shift}}$ vs β of ACMHCA



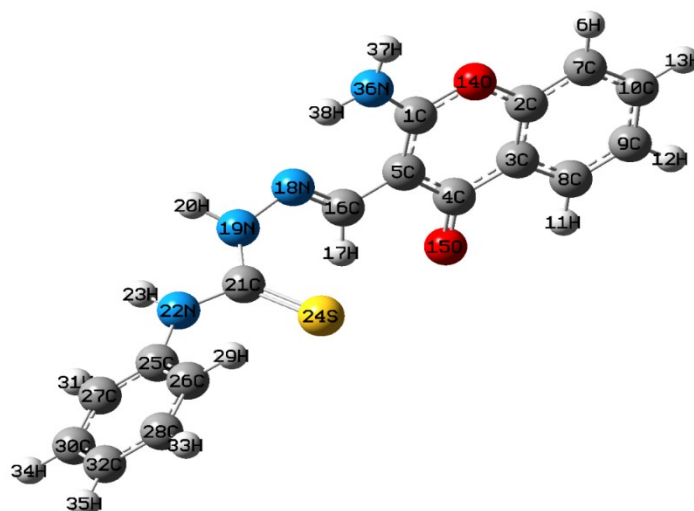
Supplementary Fig. 14. The relation between $\nu_{\text{Stoke shift}}$ vs π^* of ACMNPHTCA.



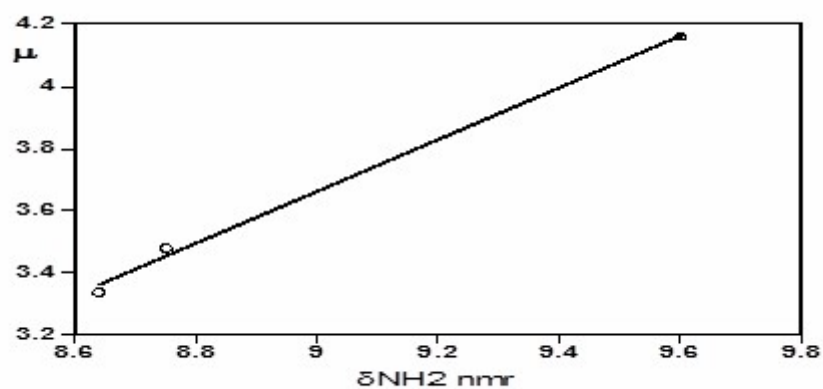
Supplementary Fig. 15. Molecular modeling of ACC.



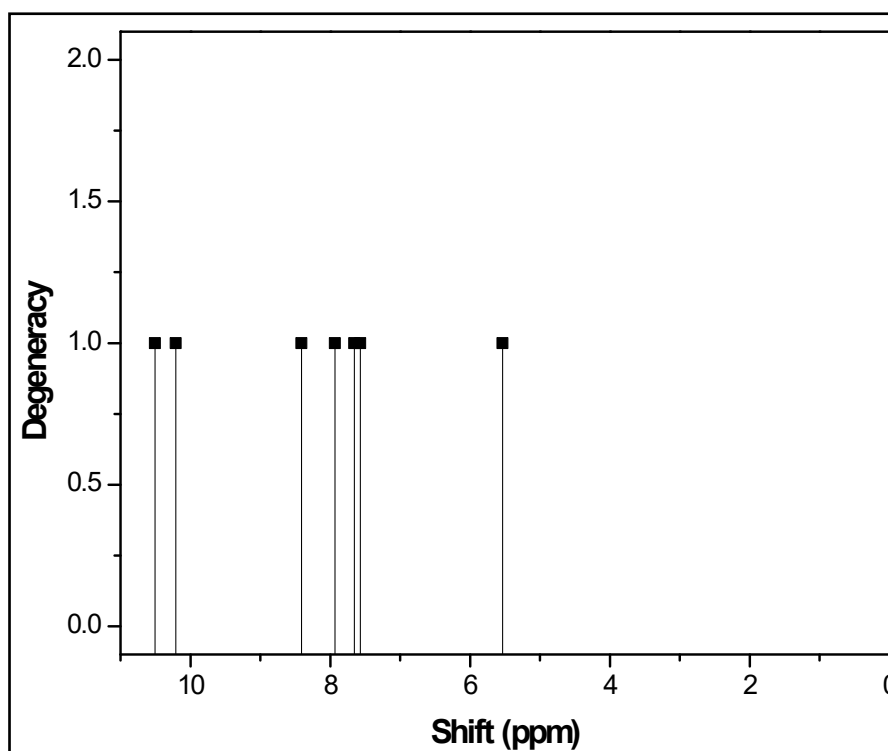
Supplementary Fig.16. Molecular modeling of ACMHCA.



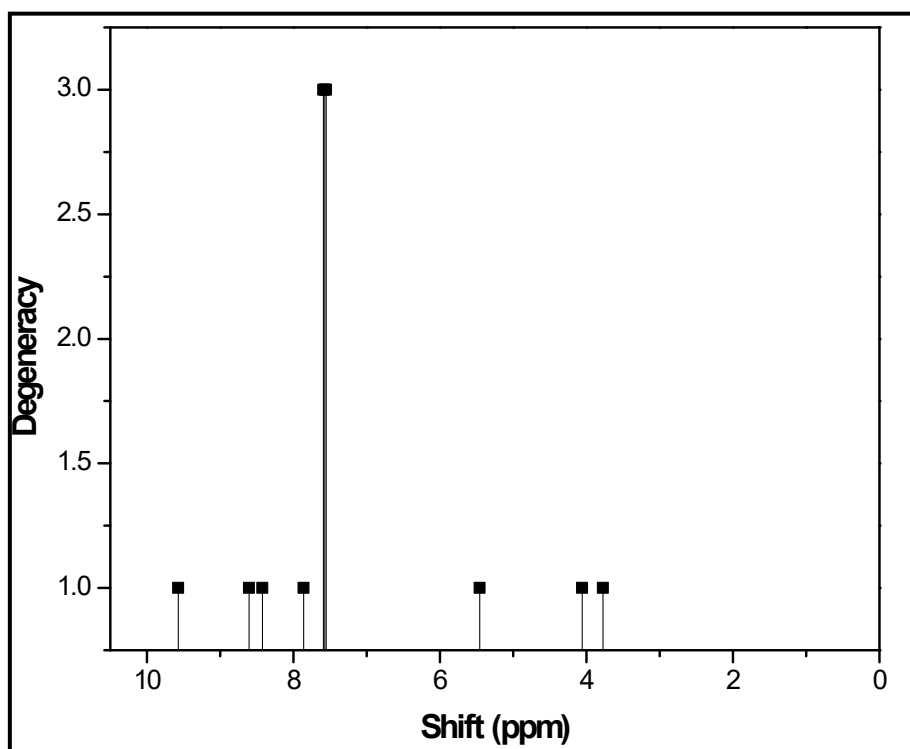
Supplementary Fig.17. Molecular modeling of ACMNPHTCA.



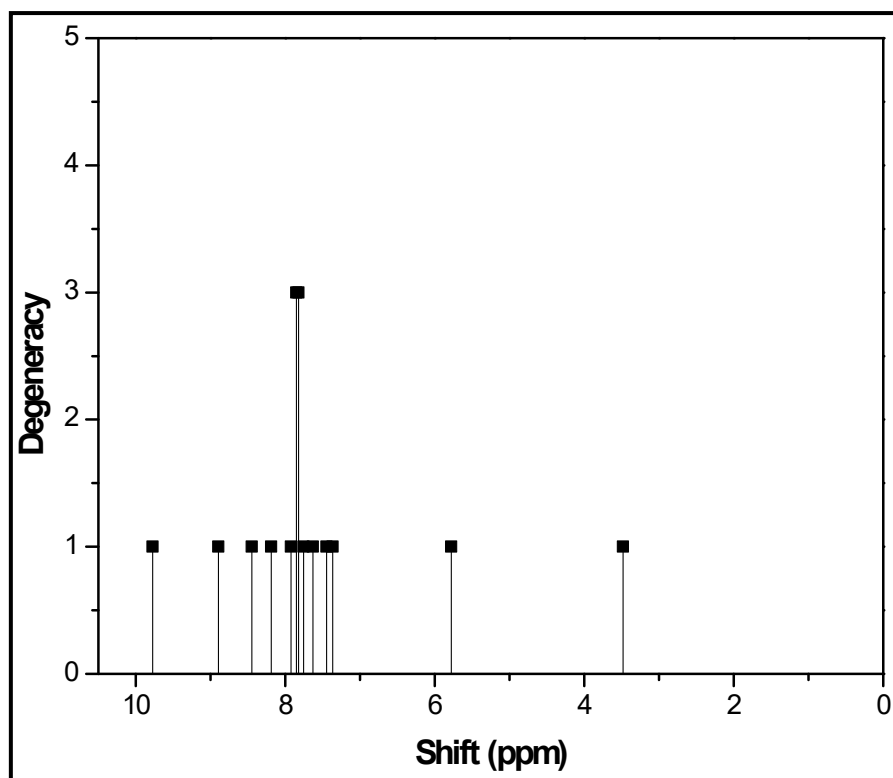
Supplementary Fig.18 The relation between chemical potential (μ) vs δ/ppmNH_2



Supplementary Fig. 19. Calculated ^1H NMR spectrum of compound ACC at B3LYP/6-311++ G(d,p).



Supplementary Fig. 20. Calculated ^1H NMR spectrum of compound ACMHCA at B3LYP/6-311++ G(d,p).



Supplementary Fig. 21. Calculated ^1H NMR spectrum of compound ACMNPHTCA at B3LYP/6-311++ G(d,p).

Supplementary Table 1. Linear regression analysis of the stokes shift *versus* solvent parameters of **ACC** and its hydrazones **ACMHCA** and **ACMNPHTCA**

Compound	Eqn	intercept	slope	<i>r</i>	n
ACC	v stoke ν_{S} DN	9887.3	+162.5	0.90059	10
	vstoke ν_{S} AN	6339.6	+197.95	0.94793	6
	v stoke ν_{S} AN	5682.6	544.81	0.98245	4
	v stoke ν_{S} DN+AN	8579.7	+91.014	0.83312	10
	v stoke ν_{S} E _T	782.77	+256.82	0.81205	9
	v stoke ν_{S} α	10965	+3391	0.8572	7
	v stoke ν_{S} β	9854.8	+5266.9	0.90558	7
	v stoke ν_{S} β	9492.9	+6241.4	0.90355	10
	vstoke ν_{S} π^*	1446	+14930	0.92537	5
	v stoke ν_{S} π^*	21620	-13859	0.89093	6

Continued Supplementary Table 1.

ACMHCA	v stoke vs DN	3039.5	+27.706	0.90494	8
	v stoke vs (DN+AN)	2989.1	+6.6181	0.92913	6
	v stoke vs E _T ⁿ	2325.8	+20.654	0.9673	6
	v stoke vs α	3038.3	+418.81	0.93359	7
	v emission vs β	23732	-333.9	0.80	8
	vstoke vs π^*	1434.6	+2849.7	0.9134	6
	v emission /La vs β, π^*, α	24107	-382 π^*	0.912	
			+412 α ,		
			-1168 β		
	v emission /La vs β, π^*, α	24756	-1423 π^*	0.60	
			-24 α ,		
			-641 β		

Continued Supplementary Table 1.

ACMNPHTCA	ν stoke ν_S π^*	5911.5	-6299.9	0.99584	5
	ν stoke ν_S π^*	8056	-6226.1	0.95496	5
	ν stoke ν_S β	6860.5	-4618.5	0.96233	5
	ν stoke ν_S β	1629.9	+4149.1	0.9259	6
	ν stoke ν_S α	2307.7	+3155	0.95189	6
	ν stoke ν_S E_T	-542.52	+77.803	0.94004	7
	ν stoke ν_S (DN+AN)	1452.6	+45.122	0.91674	7
	ν stoke ν_S AN	1156.1	+102.58	0.98023	5
	ν stoke ν_S DN	1877.7	+91.318	0.8636	7
			+187 π^*		
	ν emission $/L_b$ ν_S $\beta, \pi^* \alpha$	23105	+736 α ,	0.8500	8
			-4451 β		

Supplementary Table 2: Solvent parameters dielectric constant (ϵ), refractive index (n) and microscopic solvent polarities.

No.	Solvent	ϵ	n	φ	E_T^{Ne}	F_{L-M}	F_B	F_{K-C-W}
1	1,4-diox	2.21	1.420	0.619	0.164	0.021	0.042	0.369
2	Benzene	2.30	1.501	-	0.111	0.003	0.004	0.300
3	$CHCl_3$	4.81	1.445	0.9753	0.259	0.148	0.371	0.547
4	Etac	6.02	1.372	0.996	0.228	0.199	0.489	0.564
5	THF	7.58	1.407	1.102	0.209	0.549	0.614	0.207
6	CH_2Cl_2	8.93	1.424	-	0.321	0.219	0.596	0.586
7	2-PrOH	18.00	1.377	1.292	0.540	0.276	0.779	0.712
8	Me_2CO	20.7	1.360	-	0.355	0.285	0.792	0.654
9	EtOH	23.40	1.359	1.305	0.654	0.289	0.813	0.719
10	MeOH	32.63	1.326	1.302	0.762	0.309	0.855	0.722
11	DMF	36.71	1.431	1.419	0.404	0.274	0.836	0.771

Supplementary Table 3: The slope (m), Intercept (C), Correlation coefficient (*r*) and number of data points (n) corresponding to statistical treatment of spectral shifts

Compound	Eqn	intercept	Slope	<i>r</i>	n
ACC	ν stoke ν S FB	8627.9	+17352	0.95238	9
	ν stoke ν S F2	9166.4	+5478.4	0.95262	9
	(ν a+ ν f) ν S F3	34840	-6274.1	0.78065	11
	(ν a+ ν f) ν S F3	34943	-6532.7	0.95264	9
	ν stoke ν S E _T ⁿ	9139.6	+7758.3	0.85647	9
	ν stoke ν S E _T ⁿ	8930.9	+7684.5	0.91198	8
ACMHCA	ν stoke ν S FB	2922	+ 1562.9	0.954	7
	ν stoke ν S F2	2985	+480.97	0.96765	6
	(ν a+ ν f) /2 ν S F3	2558	-574.2	0.84141	7
	(ν a+ ν f)/2 ν S F3	26799	-2571.6	0.95162	7
	ν stoke ν S E _T ⁿ	2453.7	+2490.7	0.9913	4
	ν stoke ν S E _T ⁿ	2983.1	+633.5	0.9736	5

Continued Supplementary Table 3.

ACMNPHTCA	v stoke vs F1	1976.4	+11375	0.98833	5
	v stoke vs F1	1643.5	+6695.2	0.94061	4
	1 st gp F2 vs vstoke	2245.3	+3712.7	0.98212	5
	2 nd gp F2 vs vstoke	1904.5	+2145	0.954	4
	(va+vf)/2 vs F3	25627	-4468.8	0.98586	5
	(va+vf)/2 vs F3	24692	-1206.3	0.91497	4
	v stoke vs E _T ⁿ	2453.7	+2490.7	0.9913	4
	v stoke vs E _T ⁿ	2983.1	+633.5	0.9736	5

Supplementary Table 4. Total energy (au), energy of HOMO, of LUMO, energy gap, , Hardness(η/eV), Electrophilicity (ω/eV), Softness(S/eV^{-1}), chemical potential (μ/eV) and dipole moment (Debye) for **ACC** and its hydrazones in gaseous phase using B3LYP/ 6-311G(d,p) level.

Compound	$E_{\text{T,au}}$ kcal/mol	E_{HOMO} (eV)	E_{LUMO} (eV)	E_{gap} (eV)	$V_{\text{OC/TiO}_2}$ eV	$V_{\text{OC/PCM}}$ eV	S/eV^{-1}	η/eV	ω/eV	μ/eV	μ/D
ACC (keto form)	-665.53	-6.576	-1.740	4.836	2.576(2.26) [#]	2.876 (1.96)	0.413	2.418	3.575	5.911	5.9136
ACC (enol form)	-665.475	-6.338	-2.706	3.632	2.338 (1.294)	2.638 (0.994)	0.275	3.63	2.817	4.158	6.3383
ACMHCA (keto form)	-5456.83	-5.485	-1.192	4.295	1.485 (2.808)	1.785 (2.508)	0.465	2.146	2.596	4.522	7.134
ACMHCA (enol form)	-5456.63	-5.326	-1.186	4.139	1.326 (2.814)	1.626 (2.514)	0.483	2.070	2.561	3.338	5.842
ACMNPHTCA (thione form)	-1423.563	-5.469	-1.493	3.976	1.469 (2.507)	1.769 (2.207)	0.503	1.988	3.049	3.256	6.146
ACMNPHTCA (thiol form)	-1423.546	-5.411	-1.444	3.967	1.411(2.556)	1.711 (2.256)	0.504	1.983	2.961	3.482	4.264

[#]In parenthesis using LUMO instead HOMO