

Supporting Information

Homoleptic and heteroleptic ketodiiminate zinc complexes for the ROP of cyclic *L*-Lactide

*Eduard Glöckler, Leon Kapp, Christoph Wölper, Marcel Schumacher, André H. Gröschel, Stephan Schulz**

Content

I. Spectroscopic Characterization

Figures S1-S3. ^1H , ^{13}C NMR, and IR spectra of L^1H_2 .

Figures S4-S6. ^1H , ^{13}C NMR and IR spectra of L^2H_2 .

Figure S7. ^1H NMR spectrum of the reaction of L^1H_2 with 2 eq. of ZnCp^*Zn .

Figure S8. ^1H NMR spectrum of the reaction of L^1H_2 with 2 eq. of ZnCp_2 .

Figure S9-S10. ^{13}C NMR and IR spectra of **1**.

Figures S11-S13. ^1H , ^{13}C NMR and IR spectra of **2**.

Figures S14-S16. ^1H , ^{13}C NMR and IR spectra of **3**.

II. Crystallographic Details

Table S1 Crystal data for compound **2** and **3**

III. Polymerization Studies

Figure S17. MWDs of cPLA obtained by reaction of *L*-LA and **2** and **3**.

Figure S18. MWDs of cPLA obtained by reaction of *L*-LA and **2** (200:1+200+200).

Figure S19. ^1H NMR (300 MHz, C_6D_6 , 300 K) spectrum of cPLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 200:1 at 100 °C in toluene.

Figure S20. MALDI-ToF spectrum of cyclic-PLLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

Figure S21. IR spectrum of cyclic-PLLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

Figure S22. MALDI-ToF spectrum of linear PLLA obtained by reaction of *L*-LA and **3** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

Figure S23. IR spectrum of linear PLLA obtained by reaction of *L*-LA and **3** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

I. Spectroscopic Characterization

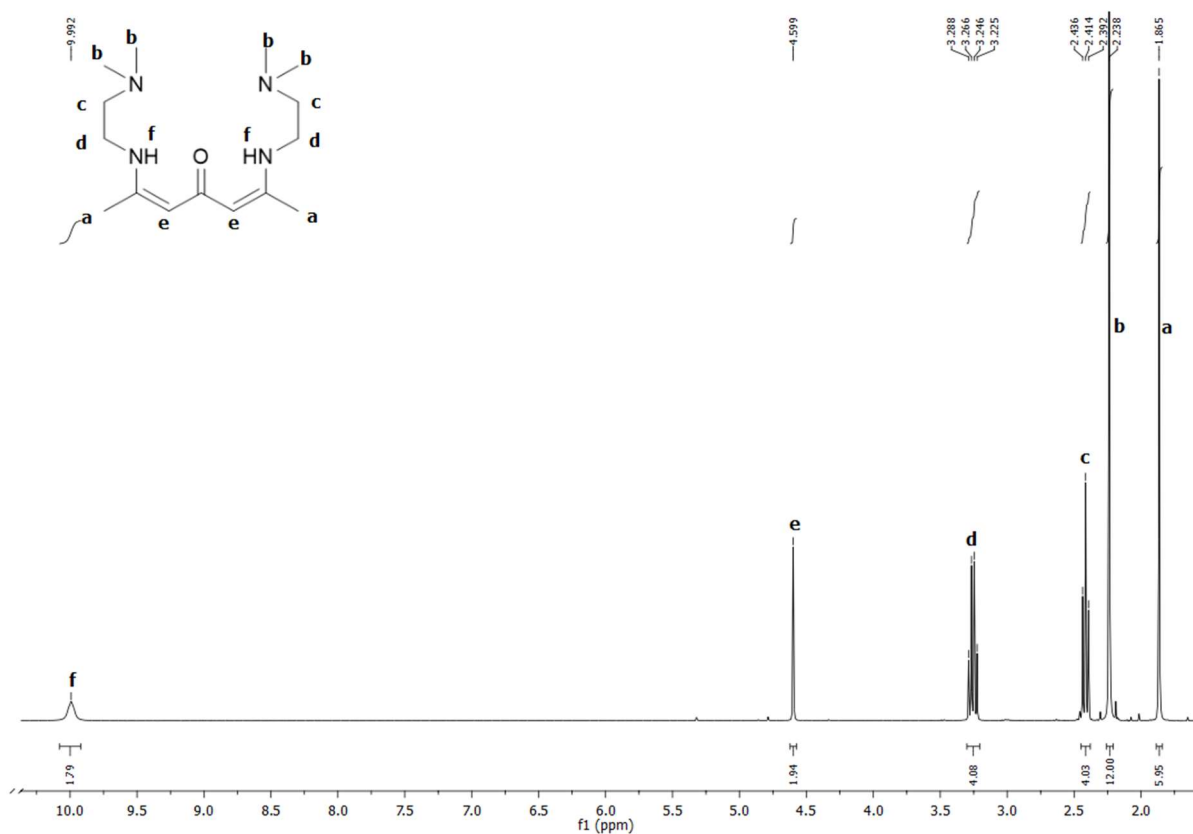


Figure S1. 1H NMR (300 MHz, CD_2Cl_2 , 25 °C) spectrum of L^1H_2 .

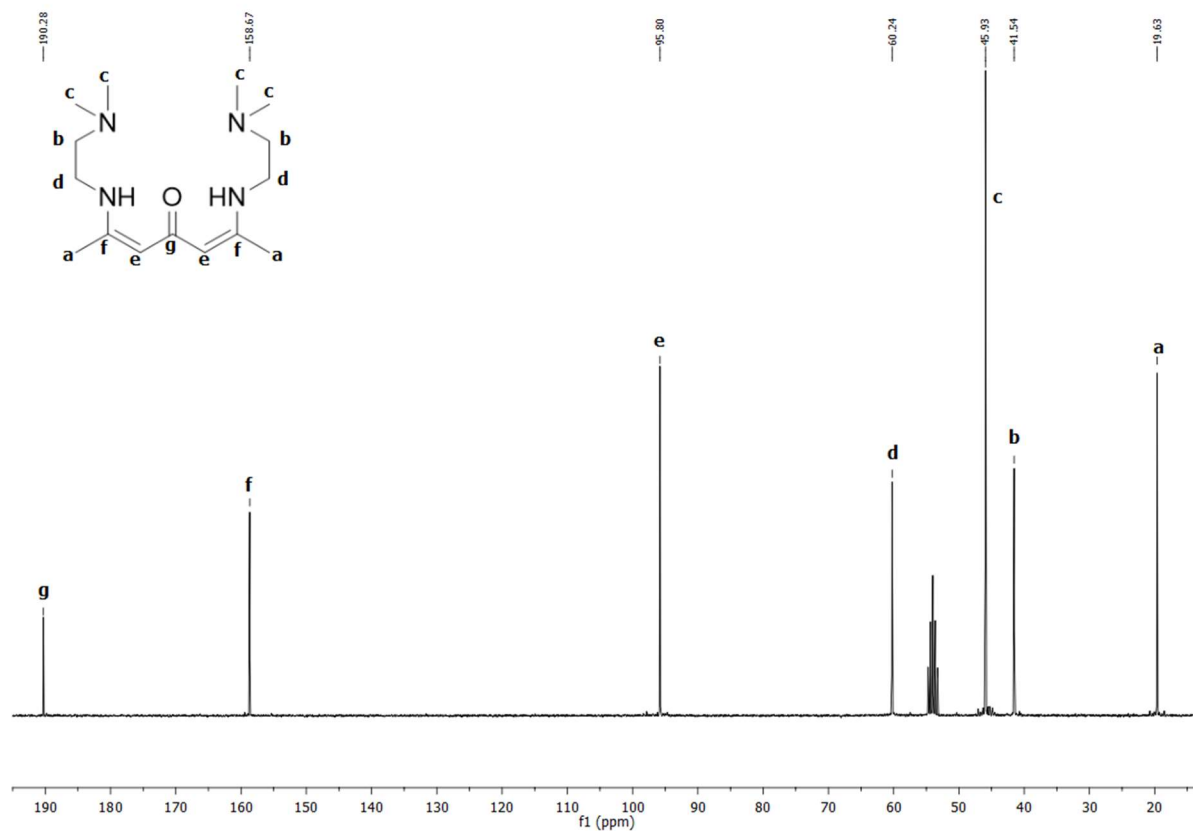


Figure S2. ^{13}C NMR (75 MHz, CD_2Cl_2 , 25 °C) spectrum of L^1H_2 .

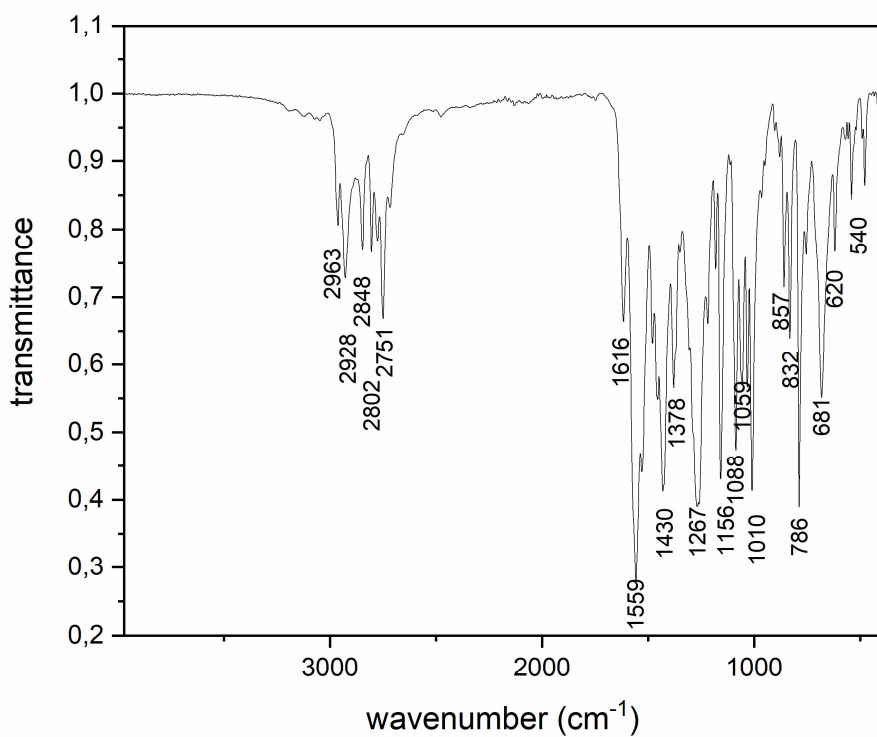


Figure S3. IR spectrum of L^1H_2 .

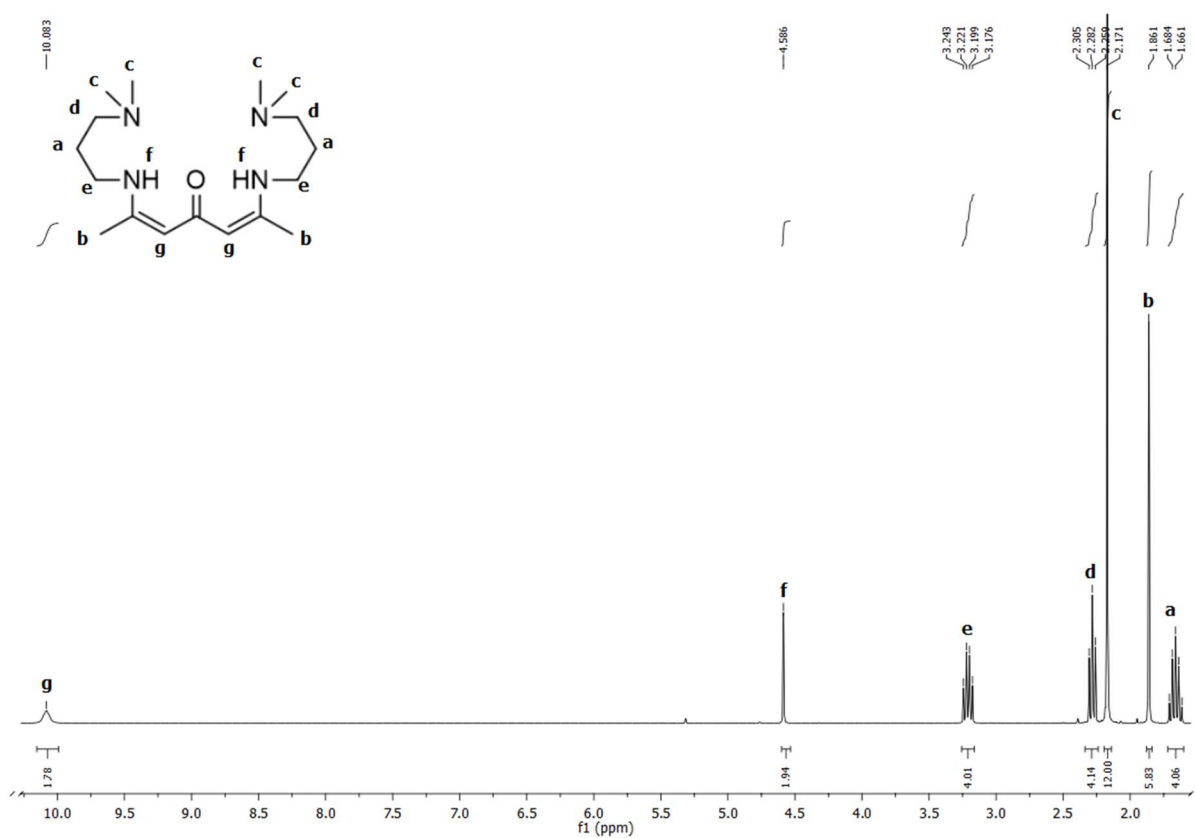


Figure S4. 1H NMR (300 MHz, CD_2Cl_2 , 25 $^{\circ}C$) spectrum of L^2H_2 .

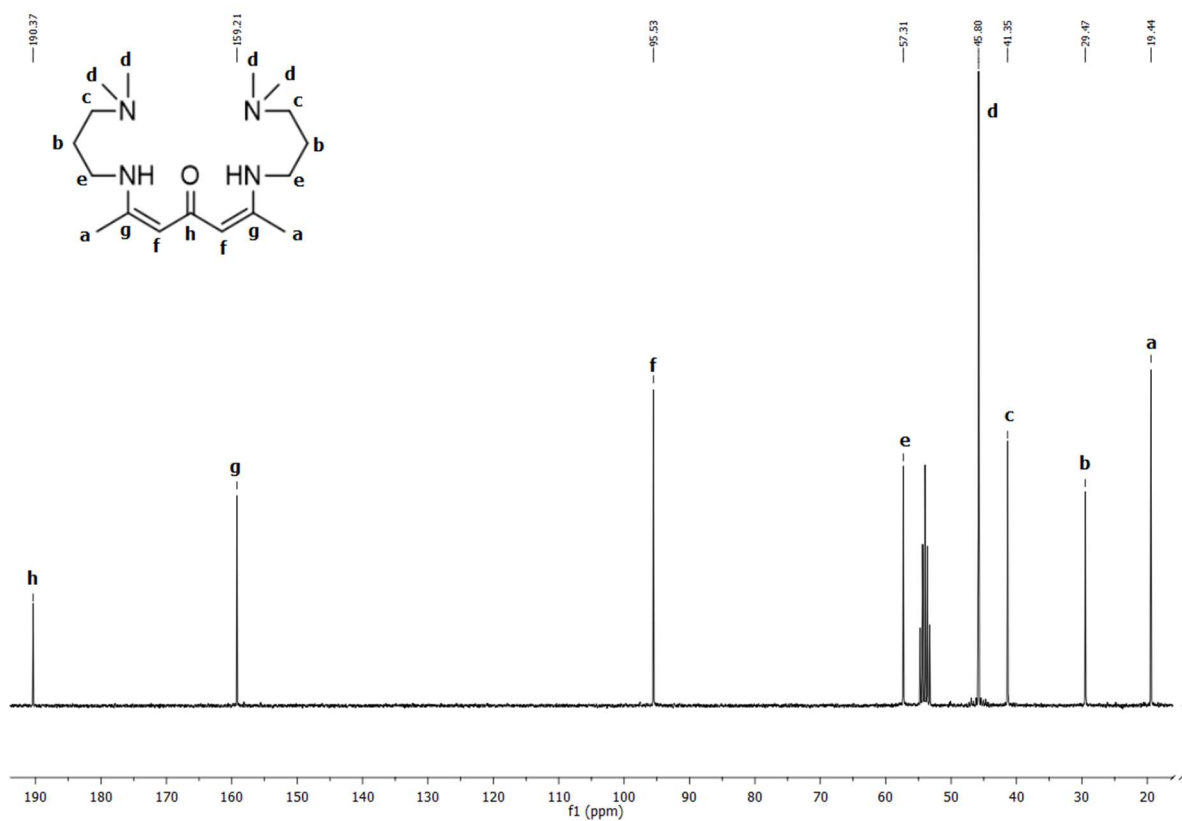


Figure S5. ^{13}C NMR (75 MHz, CD_2Cl_2 , 25 °C) spectrum of L^2H_2 .

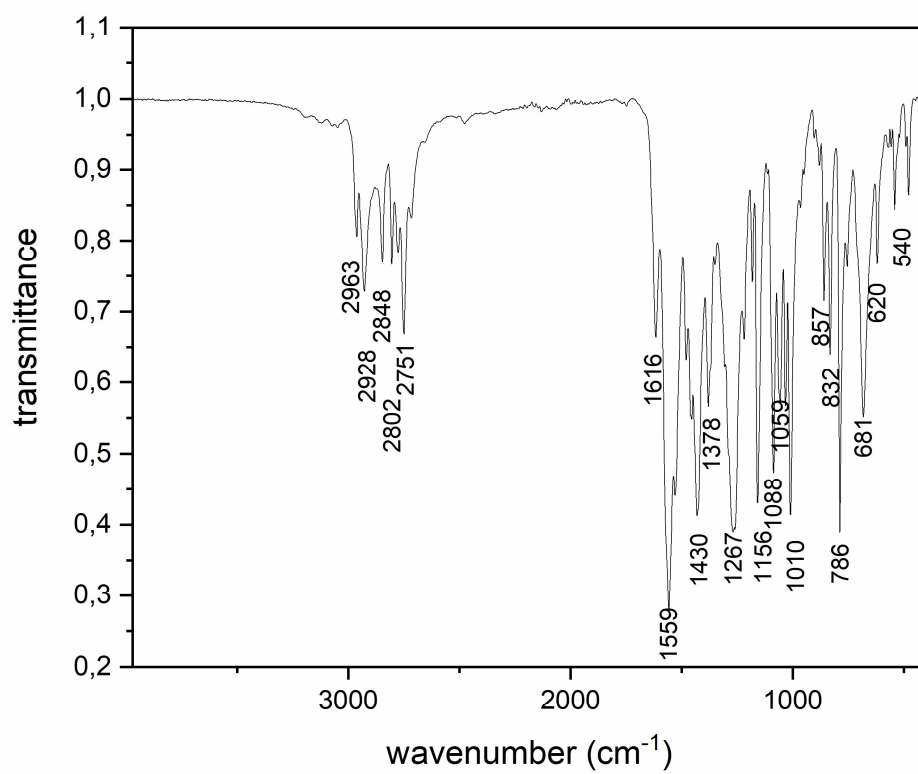


Figure S6. IR spectrum of L^1H_2 .

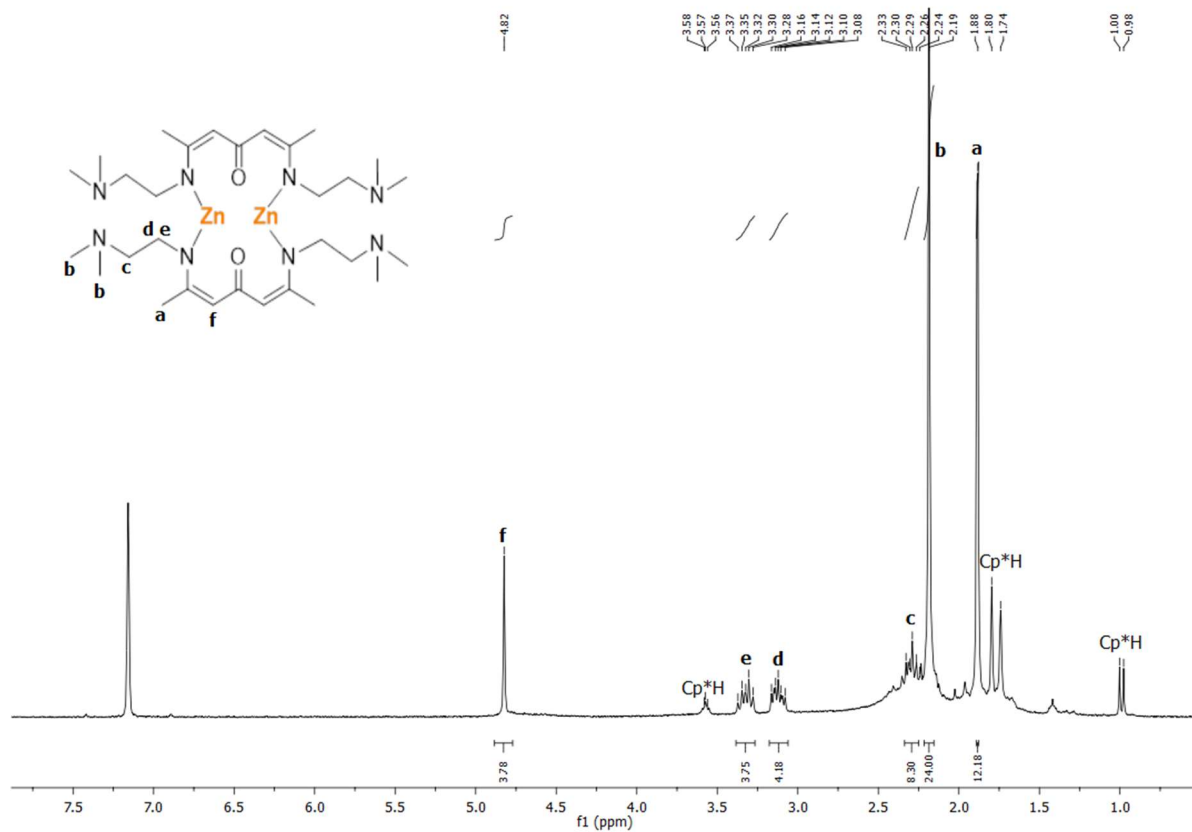


Figure S7. ^1H NMR (300 MHz, CD_2Cl_2 , 25 $^\circ\text{C}$) spectrum of the reaction of L^1H_2 with 2 eq. of ZnCp^*_2 yielding compound 1 and Cp^*H .

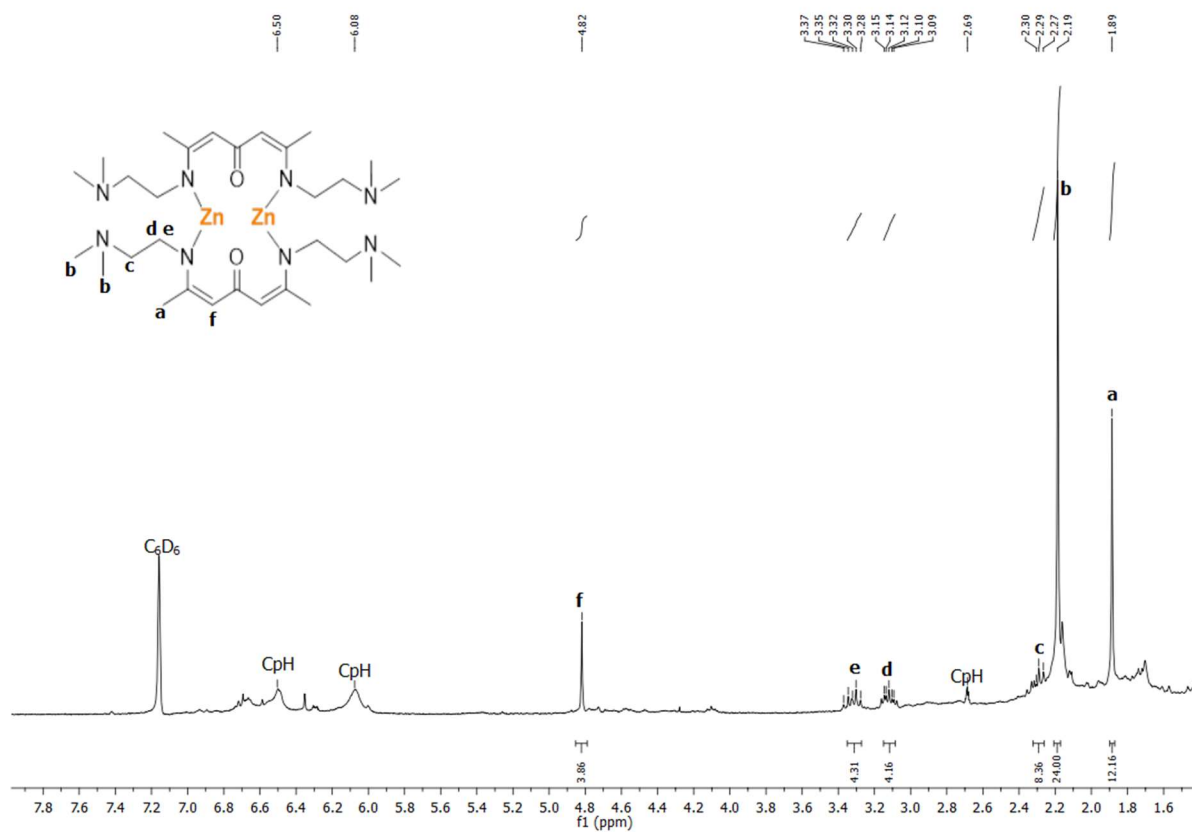


Figure S8. ^1H NMR (300 MHz, CD_2Cl_2 , 25 $^\circ\text{C}$) spectrum of the reaction of L^1H_2 with 2 eq. of ZnCp_2 yielding compound 1 and Cp^*H .

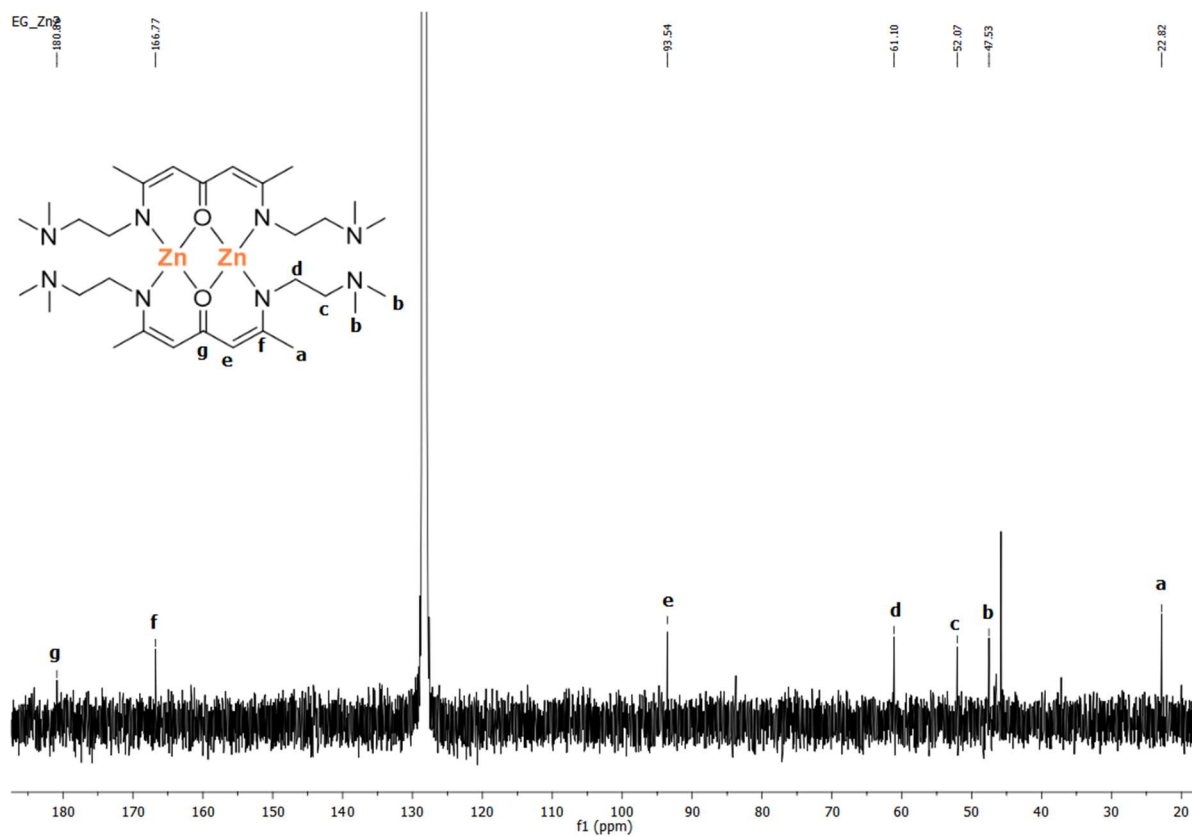


Figure S9. ^{13}C NMR (75 MHz, CD_2Cl_2 , 25 °C) spectrum of **1**.

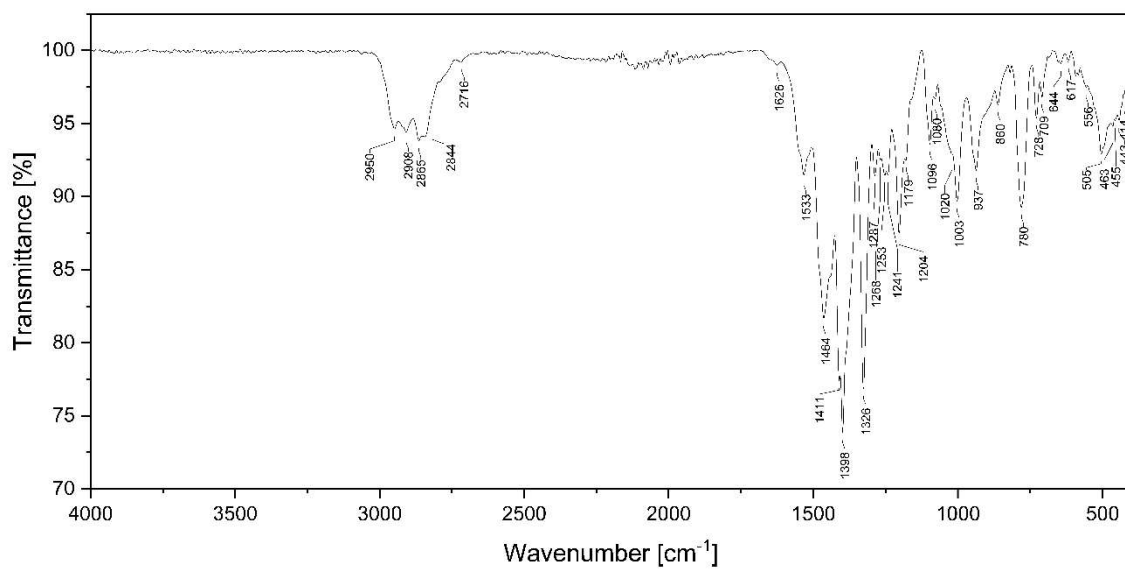


Figure S10. IR spectra of **1**.

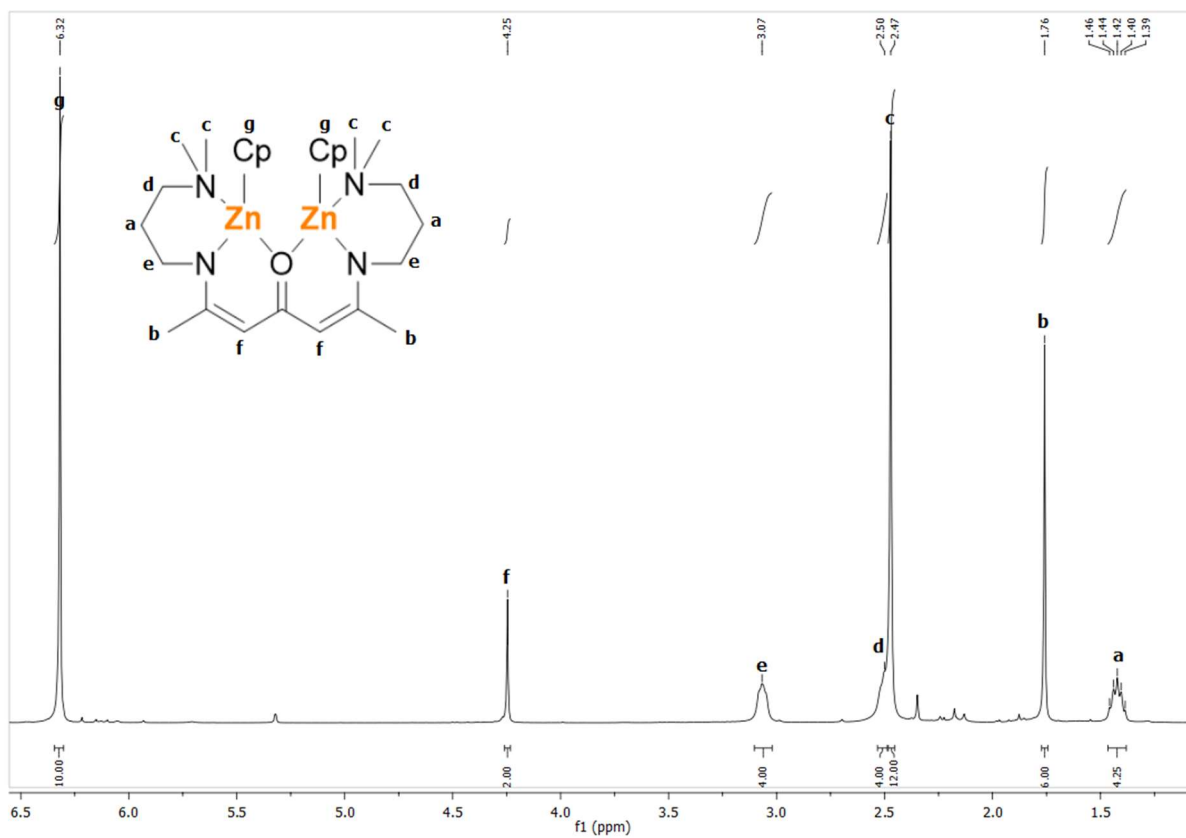


Figure S11. ¹H NMR (300 MHz, toluene-*d*₈, 25 °C) spectrum of **2**.

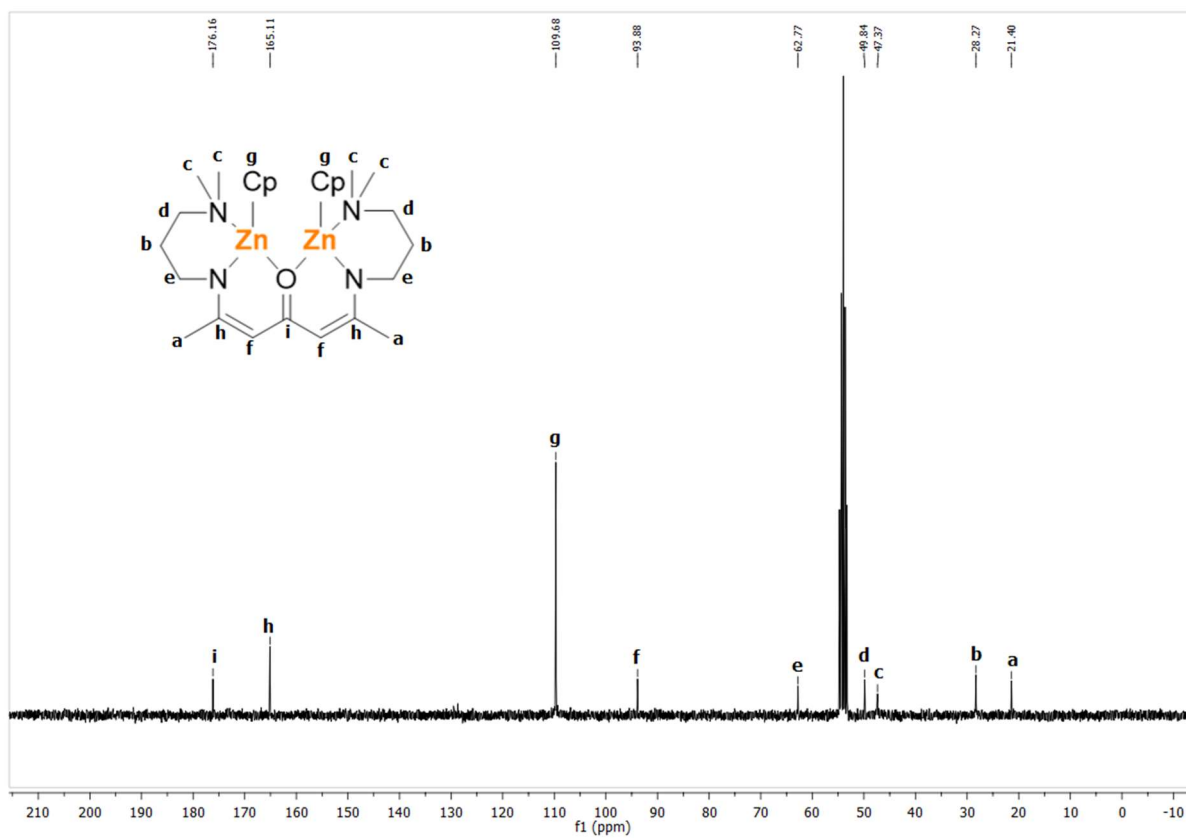


Figure S12. ¹³C NMR (75 MHz, toluene-*d*₈, 25 °C) spectrum of **2**.

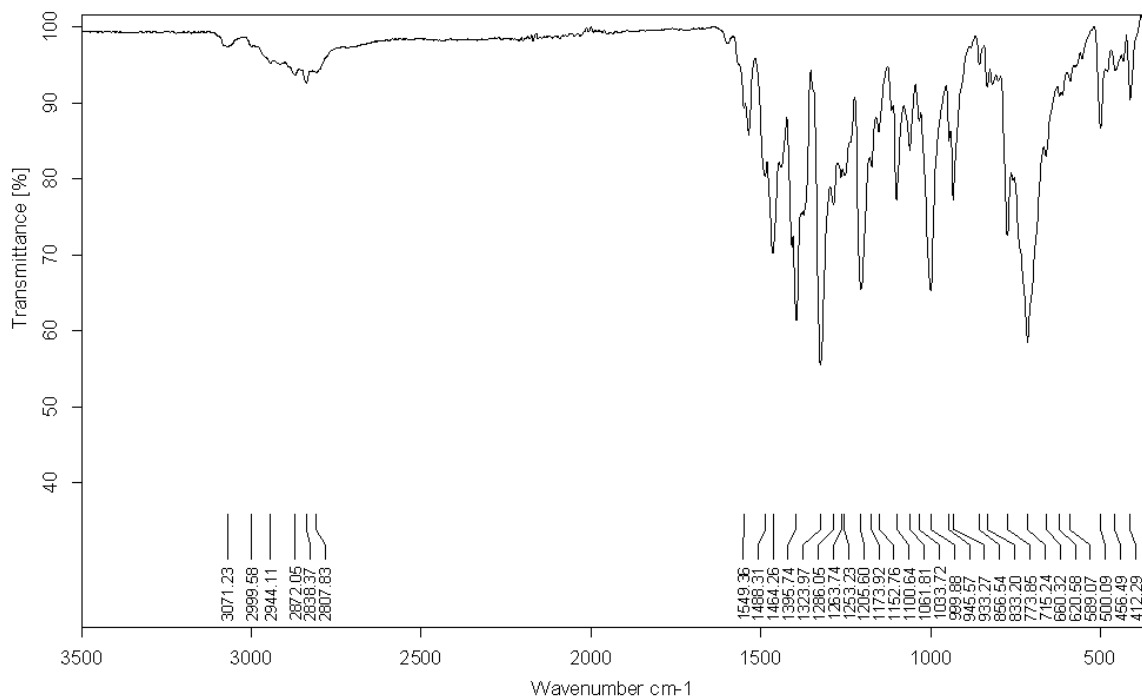


Figure S13. IR spectrum of **2**.

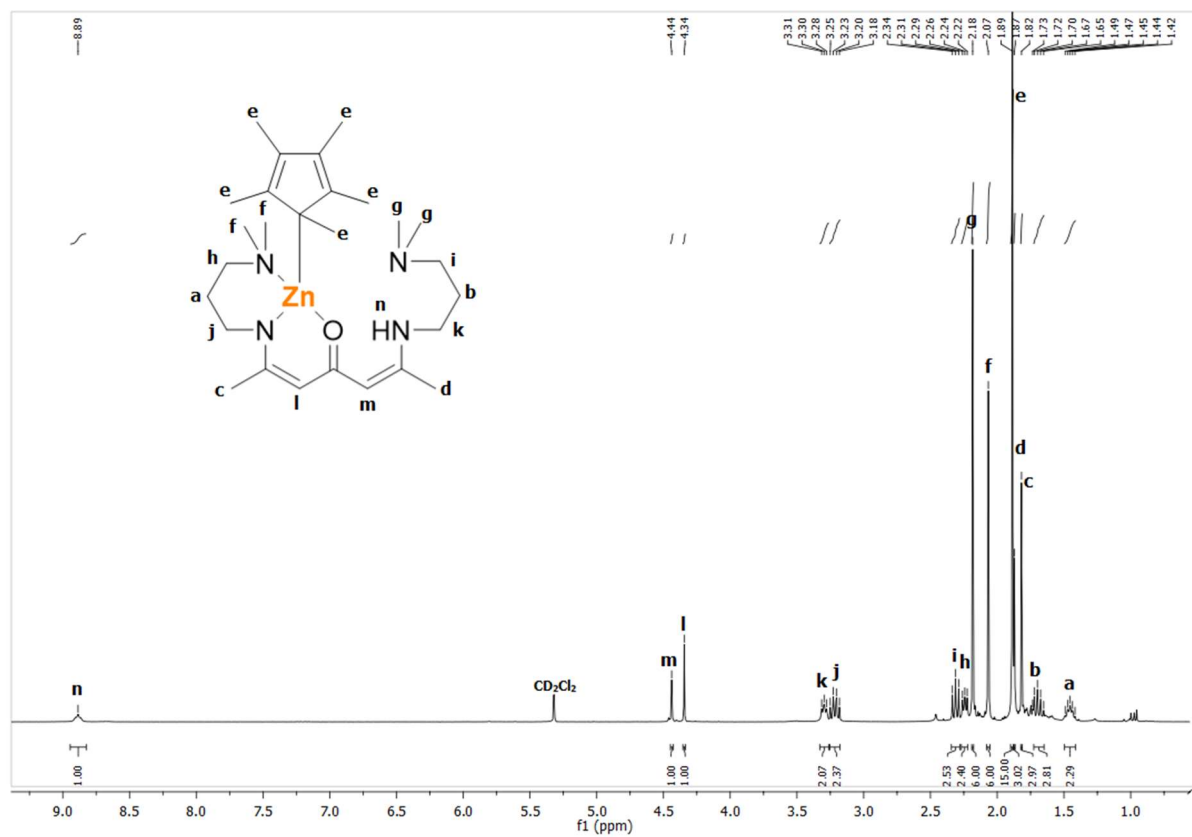


Figure S14. ¹H NMR (300 MHz, C₆D₆, 25 °C) spectrum of **3**.

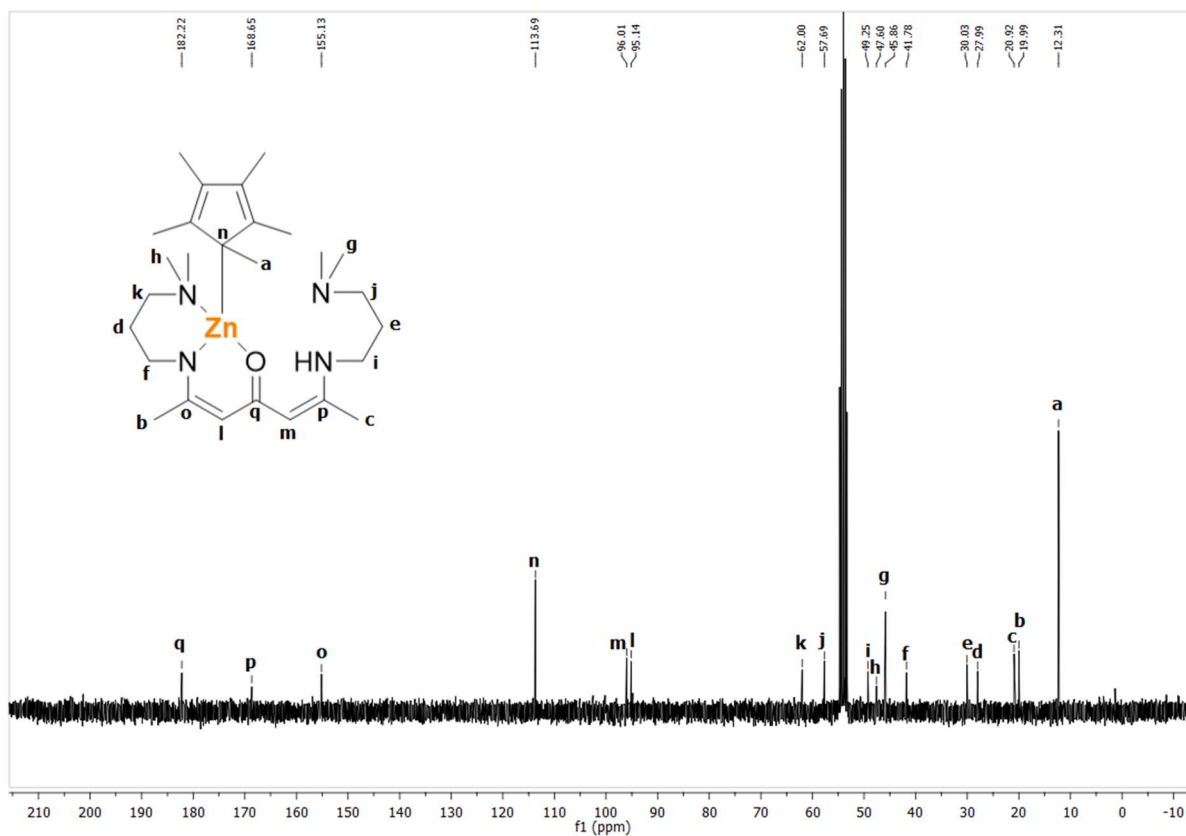


Figure S15. ^{13}C NMR (75 MHz, C_6D_6 , 25 °C) spectrum of **3**.

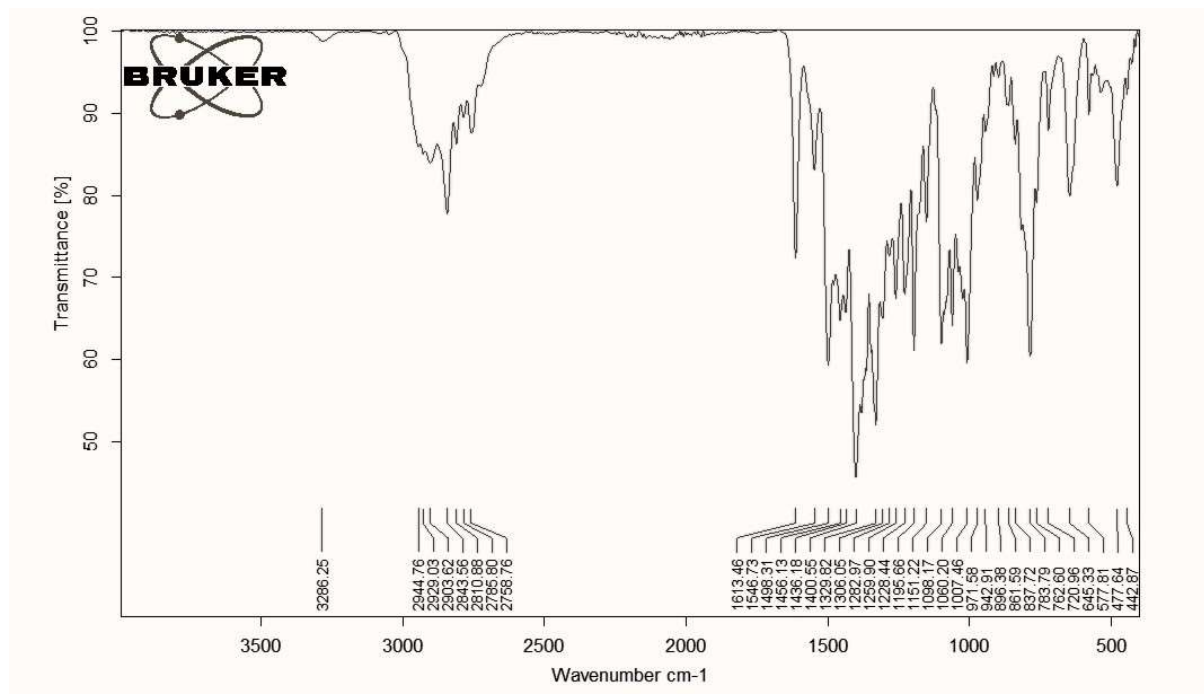


Figure S16. IR spectrum of **3**.

II. Crystallographic Details

Table S1 Crystal data for compounds **2** and **3**

Compound	2	3
Empirical formula	C ₂₇ H ₄₂ N ₄ OZn ₂	C ₂₇ H ₄₈ N ₄ OZn
Formula weight (Da)	569.38	510.06
<i>T</i> (K)	100(2)	100(2)
Wavelength (Å)	0.71073	1.54178
Crystal system	triclinic	triclinic
Space group	<i>P</i> -1	<i>P</i> -1
<i>a</i> /Å	7.9884(6)	8.0513(4)
<i>b</i> /Å	8.0941(7)	12.3749(6)
<i>c</i> /Å	22.3164(18)	14.7581(7)
α (°)	98.388(4)	81.3299(17)
β (°)	95.725(4)	81.6550(17)
γ (°)	106.622(4)	74.9858(17)
<i>V</i> (Å ³)	1352.54(19)	1395.29(12)
<i>Z</i> , Calc. density (g cm ⁻³)	2, 1.398	2, 1.214
Abs. coefficient (mm ⁻¹)	1.799	1.397
Crystal size (mm)	0.255 × 0.119 × 0.05	0.263 × 0.087 × 0.075
Theta range for data collection (°)	2.664°-33.268°	3.048°-79.538°
Reflections collected	92625	71276
Independent reflections	10387	5871
Data/restraints/parameters	10387/0/313	5871/52/362
Goodness-of-fit on <i>F</i> ²	1.040	1.062
Final <i>R</i> indices [<i>I</i> > 2 σ (<i>I</i>)]	<i>R</i> 1 = 0.029	<i>R</i> 1 = 0.038
	<i>wR</i> 2 = 0.064	<i>wR</i> 2 = 0.099
<i>R</i> indices (all data)	<i>R</i> 1 = 0.0406	<i>R</i> 1 = 0.0421
	<i>wR</i> 2 = 0.0679	<i>wR</i> 2 = 0.1040

III. Polymerization Studies

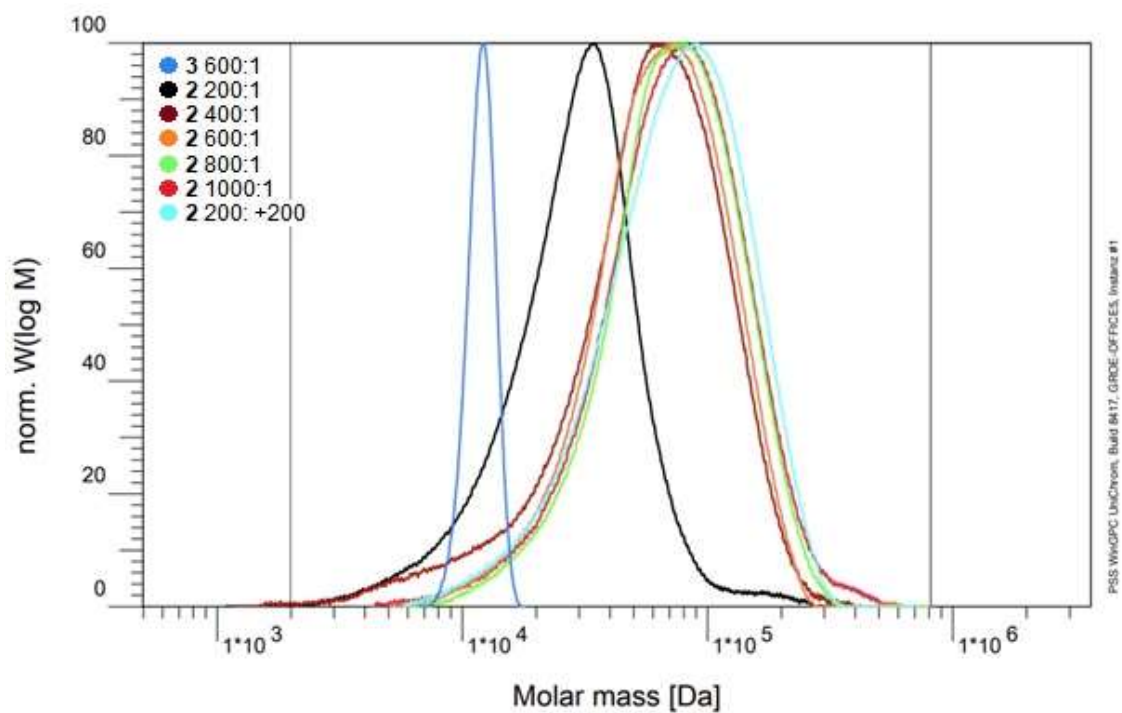


Figure S17. MWDs of cPLA obtained by reaction of L-LA and 2 and 3.

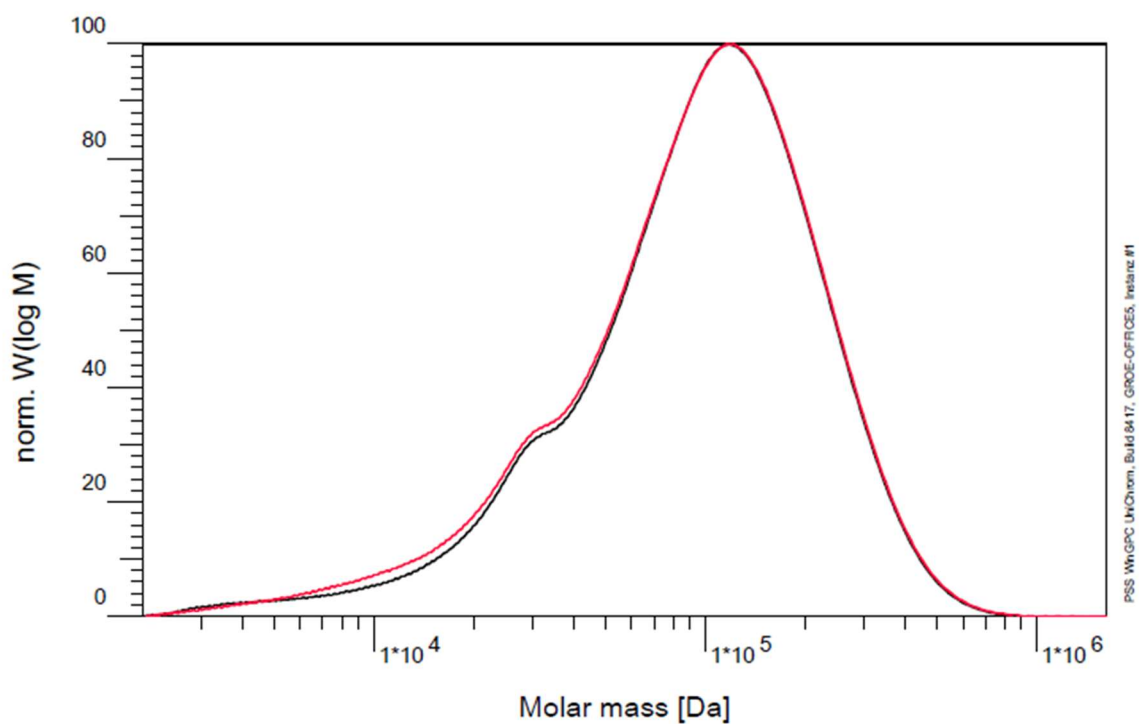


Figure S18. MWDs of cPLA obtained by reaction of L-LA and complex 2 (200:1+200+200).

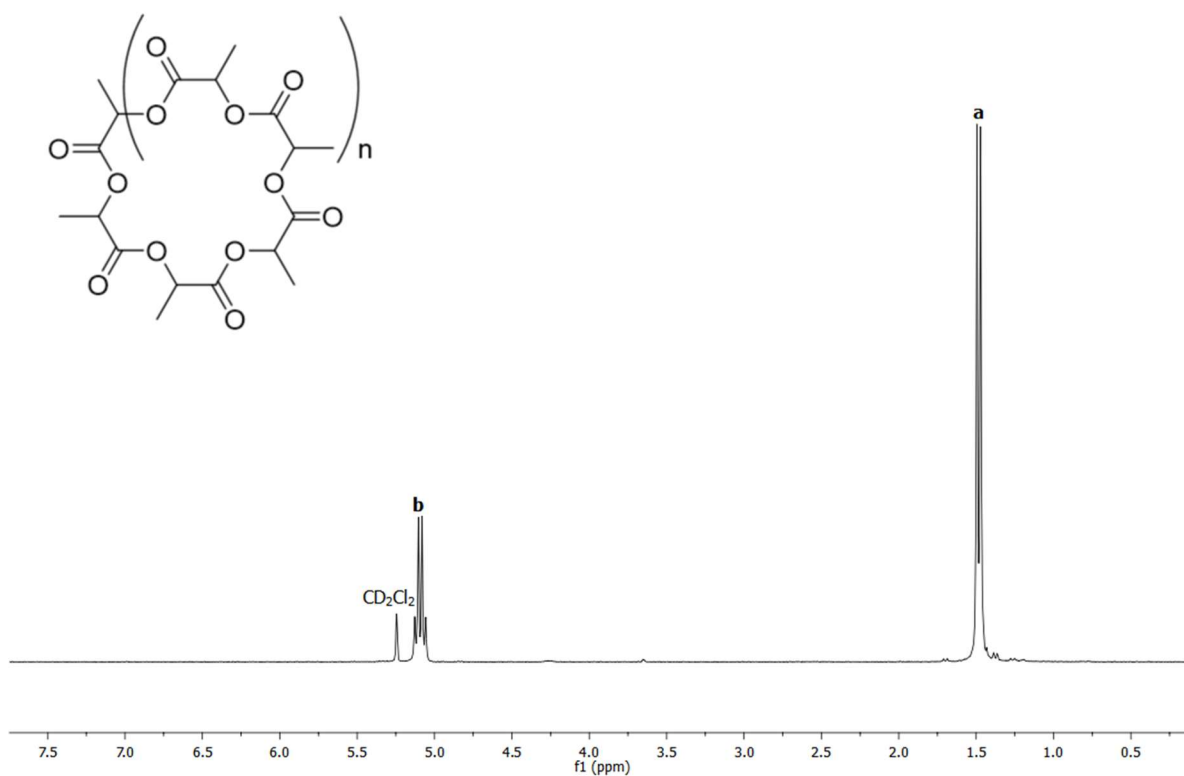


Figure S19. ¹H NMR (300 MHz, C₆D₆, 25 °C) spectrum of cPLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 200:1 at 100 °C in toluene.

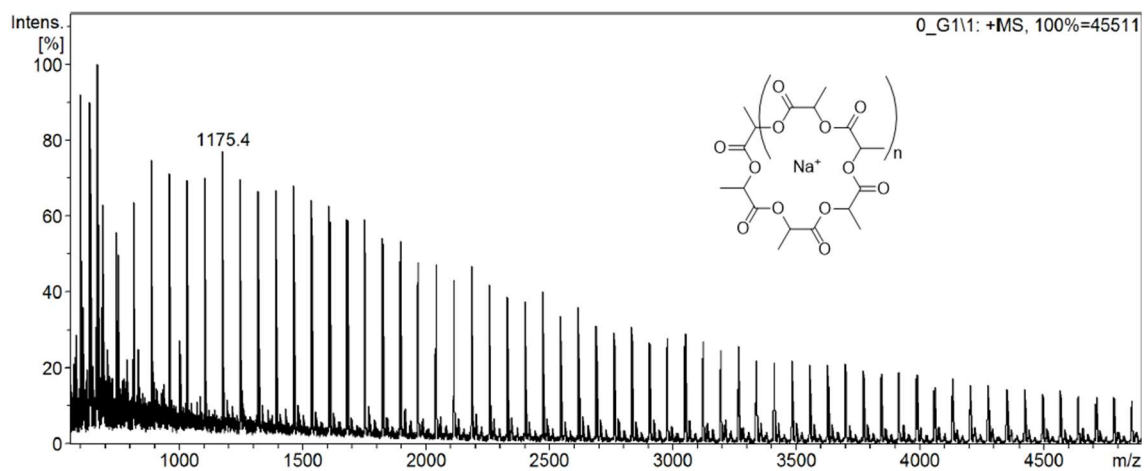


Figure S20. MALDI-ToF spectrum of cyclic-PLLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

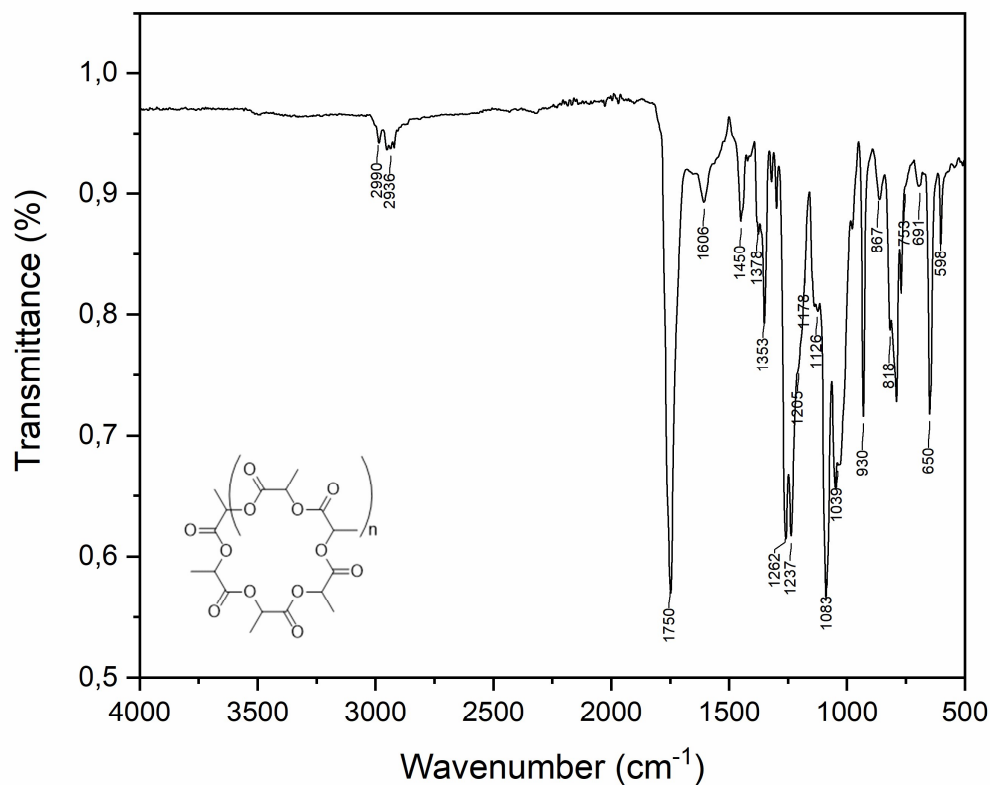


Figure S21. IR spectrum of cyclic-PLLA obtained by reaction of *L*-LA and **2** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

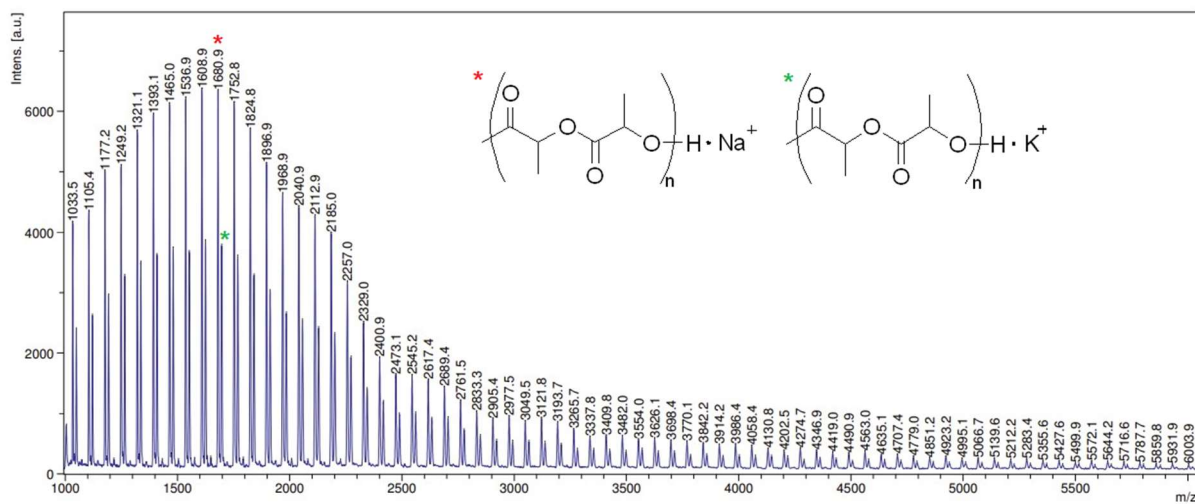


Figure S22. MALDI-ToF spectrum of linear PLLA obtained by reaction of *L*-LA and **3** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.

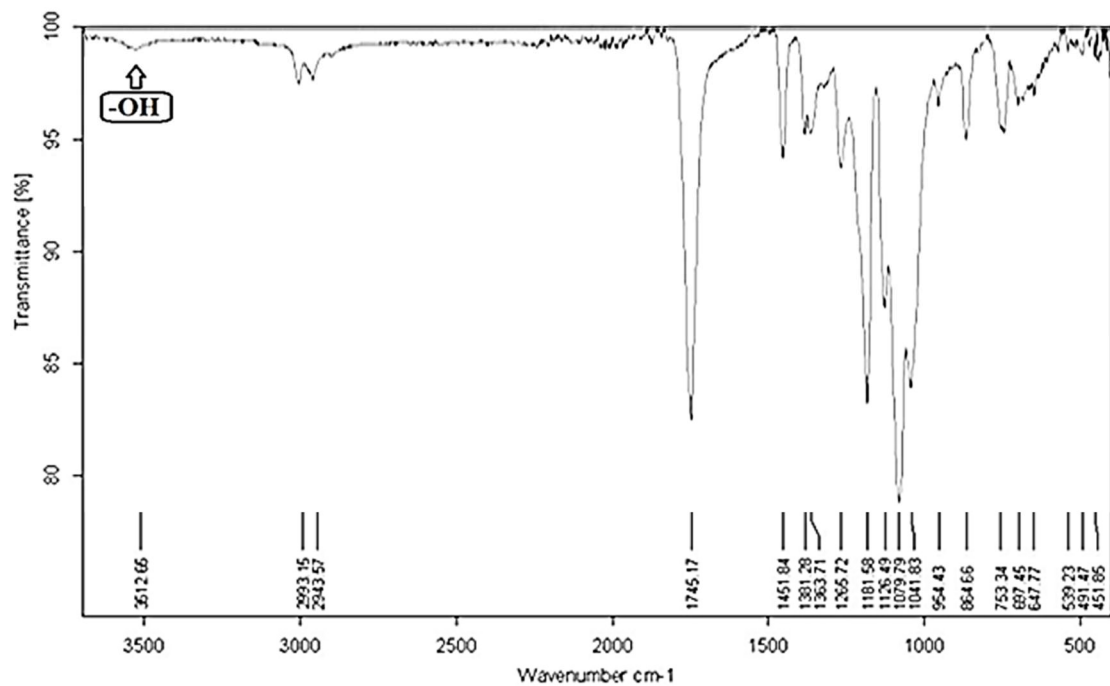


Figure S23. IR spectrum of linear PLLA obtained by reaction of *L*-LA and **3** in ratio [monomer]:[Zn] = 50:1 at 100 °C in toluene.