

Harnessing the power of natural deep eutectic solvents (choline chloride/sucrose) and polypropylene glycol in the formation of aqueous biphasic systems and the application of these systems in drug extraction

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Table S1

The values of parameters of Eq. 10, Eq. 11 and 12 along with the standard deviation of the models, sd , from the experimental values of partition coefficients for the {PPG400 + ChCl(HBA) +sucrose(HBD) + H₂O} at 298.15 K and atmospheric pressure (\approx 85 kPa).

| HBA:HBD | Diamond-Hsu (Eq. 10) | | | |
|---------|-------------------------------|----------------|------------------|------------------|
| | A | $10^3 \cdot B$ | | $100 \cdot sd^a$ |
| | Diclofenac potasium | | | |
| 2:1 | 0.0907 | -0.8209 | | 0.49 |
| 1:1 | 0.0917 | -0.9239 | | 0.38 |
| | Salicylic acid | | | |
| 2:1 | 0.092 | -0.8639 | | 0.30 |
| 1:1 | 0.0899 | -0.8975 | | 0.30 |
| | Acetaminophen | | | |
| 2:1 | 0.1415 | -1.7571 | | 0.59 |
| 1:1 | 0.1669 | -2.4408 | | 0.17 |
| | Modified Diamond-Hsu (Eq. 11) | | | |
| | A_1 | B_1 | $10^4 \cdot C_1$ | $100 \cdot sd$ |
| | Diclofenac potasium | | | |
| 2:1 | 4.4090 | -0.0385 | 1.1532 | 0.11 |
| 1:1 | 3.8590 | -0.0325 | 63.1330 | 0.12 |
| | Salicylic acid | | | |
| 2:1 | 3.3914 | -0.0075 | -1.4389 | 0.02 |
| 1:1 | 2.9721 | -0.0058 | -1.373 | 0.13 |
| | Acetaminophen | | | |
| 2:1 | -4.9348 | 0.2861 | -0.0028 | 0.41 |
| 1:1 | 2.6239 | 0.0824 | -18.0000 | 0.40 |
| | Polynomial Equation (Eq. 12) | | | |

| | A_2 | B_2 | $10^4 \cdot C_2$ | $100 \cdot sd$ |
|----------------------|---------|---------|------------------|----------------|
| Diclofenac potassium | | | | |
| 2:1 | 5.3917 | -0.2759 | 55.6320 | 0.14 |
| 1:1 | 5.5045 | -0.3118 | 64.3750 | 0.08 |
| Salicylic acid | | | | |
| 2:1 | 4.6506 | -0.1868 | 25.2990 | 0.16 |
| 1:1 | 4.6038 | -0.2137 | 37.8300 | 0.10 |
| Acetaminophen | | | | |
| 2:1 | 2.9866 | 0.1712 | -0.0153 | 0.45 |
| 1:1 | 12.1332 | -0.8287 | 95.8130 | 0.22 |

$sd = \left(\sum_{i=1}^N (K^{cal} - K^{exp})^2 / N \right)^{0.5}$ where K and N represented the partition coefficients and number of partition coefficient data, respectively.

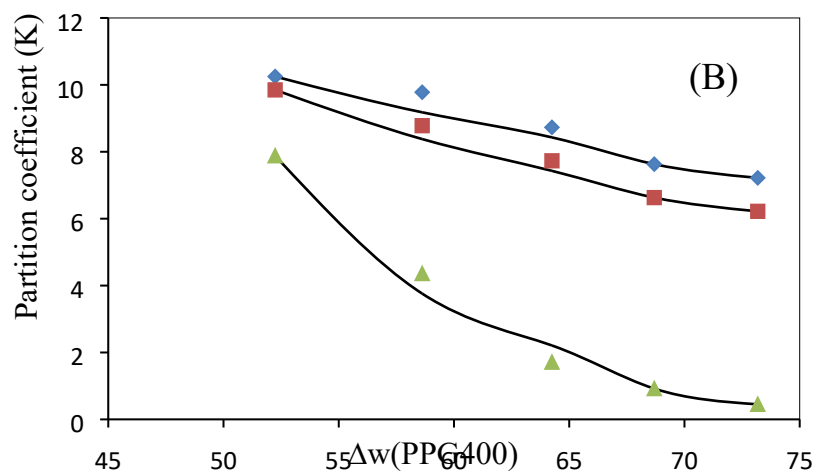
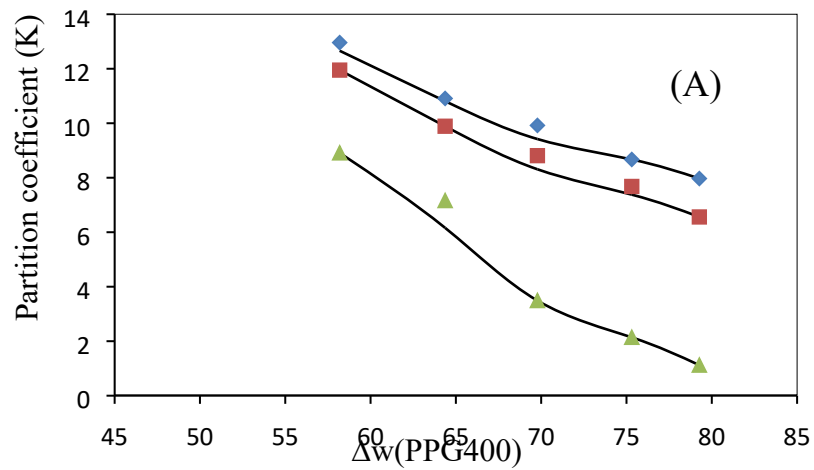


Fig. S1. Partition coefficient, K , as a function of the mass fraction difference of PPG400 in the top and bottom phases for each studied drug in ABS {NADES [ChCl : sucrose] + PPG400 + H₂O} at different ChCl:sucrose molar ratios: (A) 2:1; (B) 1:1. Diclofenac potassium: (◆); Salicylic acid: (■); Acetaminophen: (▲); (-); calculated partition coefficient, K , from Eq. 10.

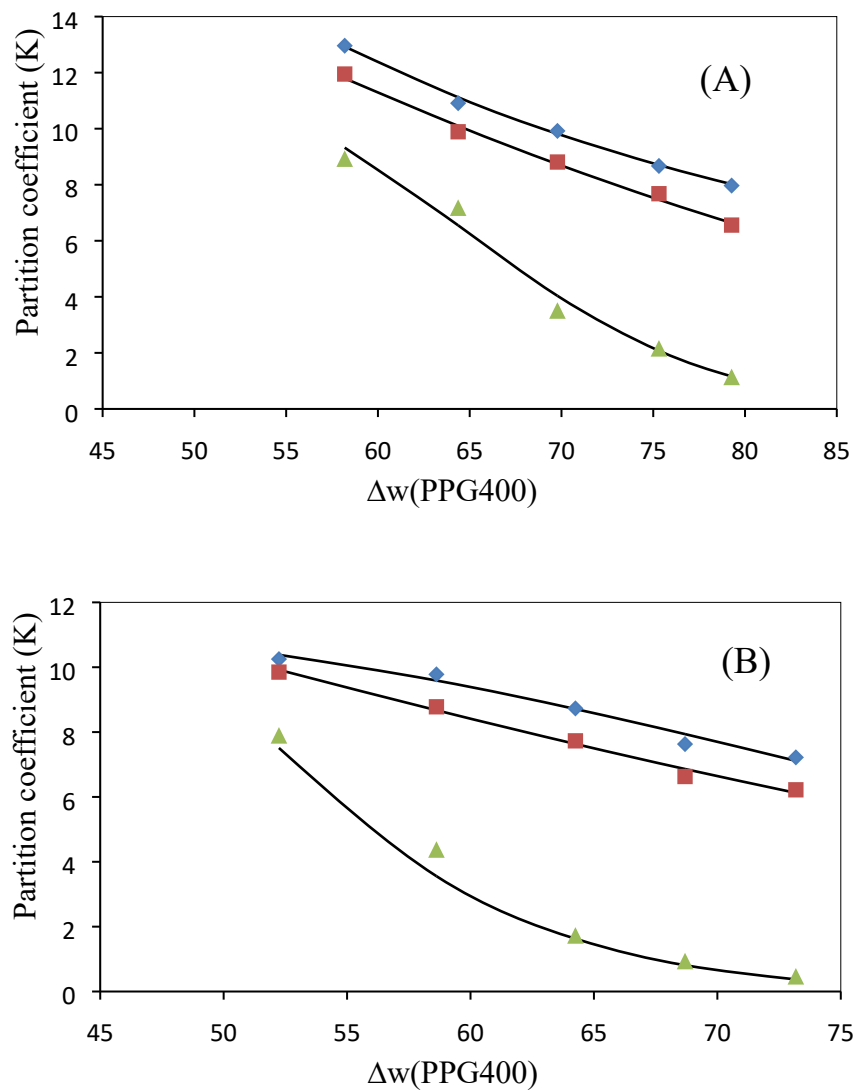


Fig. S2. Partition coefficient, K , as a function of the mass fraction difference of PPG400 in the top and bottom phases for each studied drug in ABS {NADES [ChCl : sucrose] + PPG400 + H₂O} at different ChCl:sucrose molar ratios: (A) 2:1; (B) 1:1. Diclofenac potassium: (◆); Salicylic acid: (■); Acetaminophen: (▲); (-); calculated partition coefficient, K , from Eq. 11.

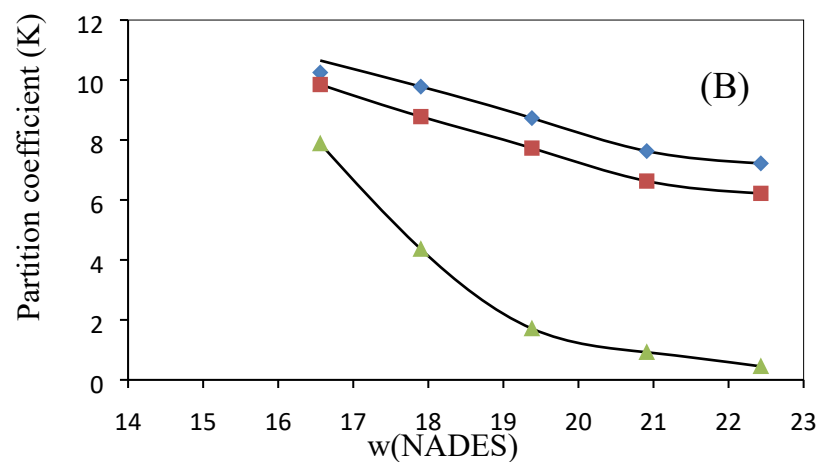
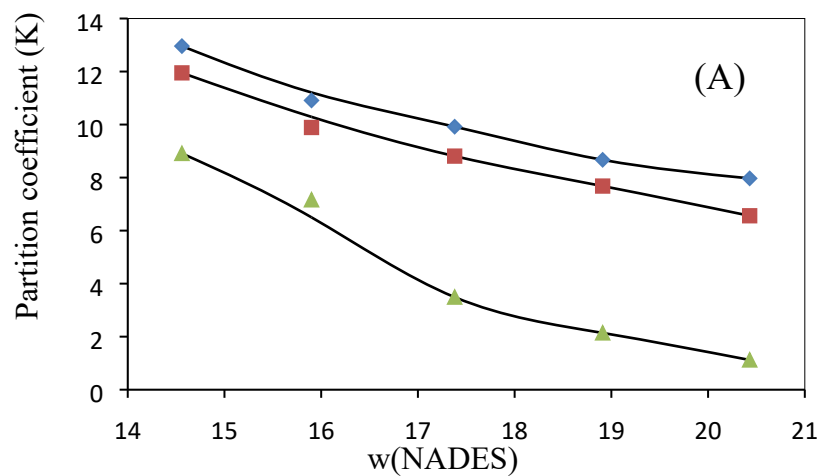


Fig. S3. Partition coefficient, K , as a function of the overall composition of NADES for each studied drug in ABS {NADES [ChCl : sucrose] + PPG400 + H₂O} at different ChCl: sucrose molar ratios: (A) 2:1; (B) 1:1. Diclofenac potassium: (◆); Salicylic acid: (■); Acetaminophen: (▲); (-); calculated partition coefficient, K , from Eq. 12.