

## Supporting Information

### Facile synthesis of morphology-controlled hybrid structure of ZnCo<sub>2</sub>O<sub>4</sub> nanosheets and nanowires for high-performance asymmetric supercapacitors

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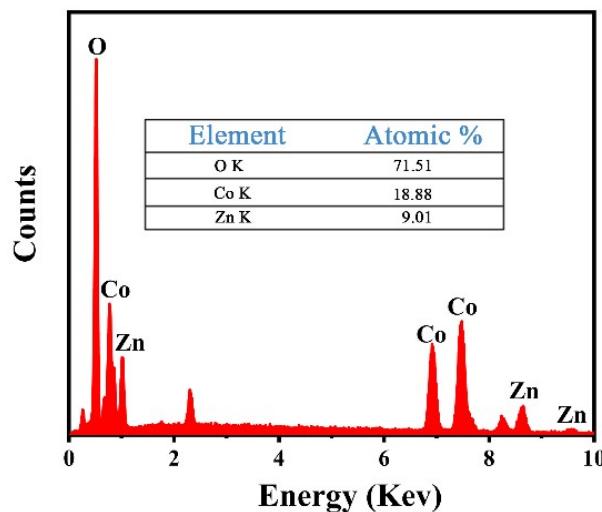


Fig. S1. EDS of the 3D hierarchical ZnCo<sub>2</sub>O<sub>4</sub> nanosheets@nanowires films.

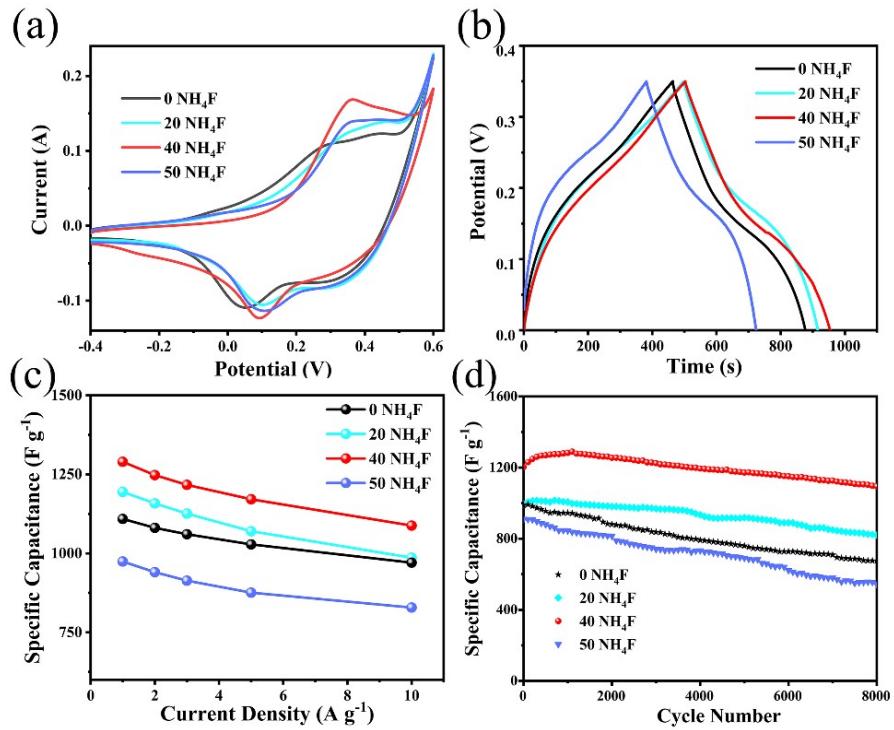


Fig. S2. Electrochemical properties of ZnCo<sub>2</sub>O<sub>4</sub> films electrodes obtained at different concentration of NH<sub>4</sub>F. (a) CV curves at the scan rate of 20 mV s<sup>-1</sup>. (b) Galvanostatic charge-discharge curves at current density of 1 A g<sup>-1</sup>. (c) Specific capacitances at different current densities. (d) Cycling stability at 1 A g<sup>-1</sup>.

Table S1 Parameters obtained by fitting of Nyquist plots in Fig. 7f.

Sample	R <sub>s</sub> (Ω)	R <sub>ct</sub> (Ω)	Z <sub>W</sub> (Ω)	CPE (μF cm <sup>-2</sup> )
Before 10,000 cycles	0.49	0.29	0.51	0.02
After 10,000 cycles	0.51	1.92	2.06	0.03

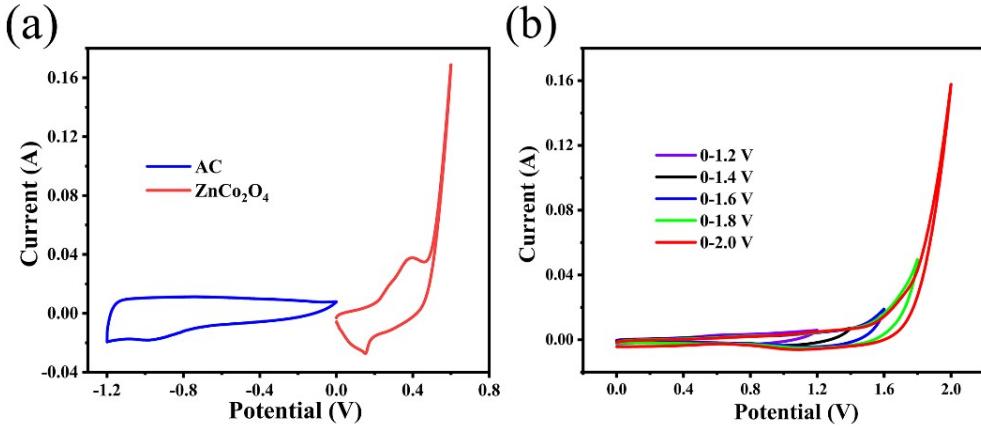


Fig. S3. (a) CV curves of the 3D hierarchical  $\text{ZnCo}_2\text{O}_4$  nanosheets@nanowires films and AC electrodes at the scan rate of  $20 \text{ mV s}^{-1}$ . (b) CV curves of  $\text{ZnCo}_2\text{O}_4/\text{AC}$  ASC device in different potential ranges.

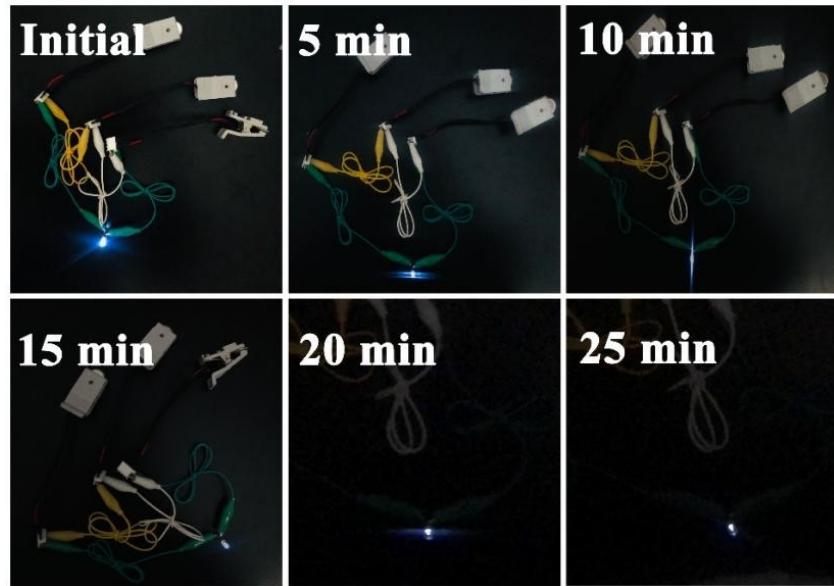


Fig. S4. Optical images of a blue LED powered by three ASC devices connected in series and the lightened blue LED at different stages.

Table S2 Comparison of the electrochemical properties between the as-prepared 3D hierarchical  $\text{ZnCo}_2\text{O}_4$  nanosheets@nanowires films and recent results reported in literature.

Materials	Electrolyte	Voltage window	Specific Capacity	Cycling Stability	Ref.
$\text{ZnCo}_2\text{O}_4$ nanoparticles	1 M KOH	0-0.45 V	$843 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	91.5% after 2,000 cycles at $1 \text{ A g}^{-1}$	[1]
3D hierarchical $\text{ZnCo}_2\text{O}_4$	6 M KOH	0-0.4 V	$421.05 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	88% after 5000 cycles at $5 \text{ A g}^{-1}$	[2]
$\text{ZnCo}_2\text{O}_4$ nanobelts	2 M KOH	0-0.45 V	$776.2 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	84.3% after 1,500 cycles at $3 \text{ A g}^{-1}$	[3]
$\text{ZnCo}_2\text{O}_4$ nanowires	2 M KOH	0-0.4 V	$1099 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	84.8% after 5,000 cycles at $40 \text{ A g}^{-1}$	[4]
$\text{ZnCo}_2\text{O}_4$ microspheres	1 M KOH	0-0.425 V	$344.44 \text{ F g}^{-1}$ at $1 \text{ mA cm}^{-2}$	85% after 500 cycles at $100 \text{ mV s}^{-1}$	[5]
peony-like $\text{ZnCo}_2\text{O}_4$	3 M KOH	0-0.55 V	$440 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	98% after 5,000 cycles at $2 \text{ A g}^{-1}$	[6]
$\text{ZnCo}_2\text{O}_4$ with Zn/Co vacancies	6 M KOH	0-0.4 V	$1608.95 \text{ F g}^{-1}$ at $0.35 \text{ A g}^{-1}$	89% after 3,500 cycles at $5 \text{ A g}^{-1}$	[7]
flower-like $\text{ZnCo}_2\text{O}_4$	KOH	0-0.4 V	$680 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	90% after 2,000 cycles at $5 \text{ A g}^{-1}$	[8]
Porous $\text{ZnCo}_2\text{O}_4$ quasi-cubes	2 M KOH	0-0.45 V	$804 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	79.2% after 3,000 cycles at $5 \text{ A g}^{-1}$	[9]
$\text{ZnCo}_2\text{O}_4$ nanocubes with 3D porous structure	6 M KOH	0-0.5 V	$542.6 \text{ F g}^{-1}$ at $1 \text{ A g}^{-1}$	87% after 8,000 cycles at $15 \text{ A g}^{-1}$	[10]

N-rGO/ZnCo <sub>2</sub> O <sub>4</sub>	1 M KOH	0-0.6 V	950 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	89.6% after 5,000 cycles at 1 A g <sup>-1</sup>	[11]
flower-like ZnCo <sub>2</sub> O <sub>4</sub> /ZnO	6 M KOH	0-0.5 V	803 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	91.04% after 10,000 cycles at 10 A g <sup>-1</sup>	[12]
3D hierarchical ZnCo <sub>2</sub> O <sub>4</sub> nanosheets@nanowires	6 M KOH	0-0.35 V	1289.6 F g <sup>-1</sup> at 1 A g <sup>-1</sup>	86.8% after 10,000 cycles at 1 A g <sup>-1</sup>	(This work)

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