DRGD linked charged EKKE dimeric dodecapeptide: pH-based amyloid nanostructures and their application in lead and uranium binding

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HPLC REPORT

Sample:	Ac-EE-12	Analyzed date:	2019-3-16
Lot. No.:	P190307-MX712437		
Column:	Kromasil-C18, 4.6*250mm, 5µm		
Solvent A	A: 0.1% Trifluoroacetic Acid in 100	% Acetonitrile	
Solvent B	B: 0.1% Trifluoroacetic Acid in 1009	% Water	

	А	В
0.0min	6%	94%
25.0min	31%	69%
25.1min	100%	0%
30.0min		Stop
10µl		
220nm		
1.0ml/min		
	0.0min 25.0min 25.1min 30.0min 10µl 220nm 1.0ml/min	A 0.0min 6% 25.0min 31% 25.1min 100% 30.0min 10µl 220nm 1.0ml/min

Figure S1: HPLC profile of Ac-EKKEDRGDEKKE-CONH₂



400

°C

MASS SPECTROMETRY REPORT

Figure S2: Mass spectrometry report of Ac-EKKEDRGDEKKE-CONH₂

P190307-MX712437 Block Temp.:

Lot. No .:

Table S1: DLS size p	rofile for Ac-EKKEDRGDEKKE-CONH ₂ peptide at pH 3.6, 7.4 and
11.7 at 260	μM concentration.

EKKEDRGDEKKE	1h size profiles	3 days size profiles
peptide	(Z-average)	(Z-average)
рН 3.6	181.8 d.nm, 242.5 d.nm, 226.9	429.1 d.nm, 432 d.nm, 596.9
	d.nm (PdI 0.316)	d.nm (PdI 0.334)
рН 7.4	113.1 d.nm, 196.9 d.nm, 370	412.5 d.nm, 465.8 d.nm, 570
	d.nm (PdI 0.471)	d.nm (PdI 0.515)
рН 11.7	285.73 d.nm, 356.3 d.nm,	413.1 d.nm, 459.8 d.nm, 552.4
	372.1 d.nm (PdI 0.389)	d.nm (PdI 0.496)



Figure S3: [A] Amount of secondary structure elements present at 80 h post dissolution of EKKEDRGDEKKE peptide at pH 3.6, 7.4 and 11.7 and [B] time dependent representation (in %) of the different secondary structure elements for the peptide EKKEDRGDEKKE from the point of incubation at 1h, 8h, 24h, 48h and 80h.

Table S2 (% of secondary structure elements in EKKEDRGDEKKE peptide): Percentage of secondary structure elements present in EKKEDRGDEKKE peptide at the three pH conditions, pH 3.6, pH 7.4 and pH 11.7 across time intervals, 1h, 8h, 24h, 48h and 80h obtained from BestSel analysis of circular dichroism information XPS of peptide and Pb-complexes at three pHs.

Time	EKKEDRGDEKKE pH 3.6	EKKEDRGDEKKE pH 7.4	EKKEDRGDEKKE pH 11.7
1 h	α -helix = 10.5 %	α -helix = 8.7 %	α -helix = 14.5 %
	β -sheets = 50.8 %	β -sheets = 52 %	β -sheets = 0 %
	Others = 38.7 %	Others = 39.3 %	Others = 88.5 %
8 h	α -helix = 7.8 %	α -helix = 5.7 %	α -helix = 26.5 %
	β -sheets = 52.8 %	β -sheets = 54.2 %	β -sheets = 0 %
	Others = 39.4 %	Others = 40.2 %	Others = 73.5 %
24 h	α -helix = 7.8 %	α -helix = 6.9 %	α -helix = 24.7 %
	β -sheets = 52.4 %	β -sheets = 52.4 %	β -sheets = 0 %
	Others = 39.8 %	Others = 40.6 %	Others = 78.3 %
48 h	α -helix = 14.4 %	α -helix = 11.8 %	α -helix = 40.3 %
	β -sheets = 46.7 %	β -sheets = 47.1 %	β -sheets = 0 %
	Others = 38.8 %	Others = 41.1%	Others = 59.7 %
80 h	α -helix = 21.9 %	α -helix = 16.7 %	α -helix = 0 %
	β -sheets = 39.6 %	β -sheets = 37.7 %	β -sheets = 51.5 %
	Others = 38.5 %	Others = 45.6 %	Others = 37.2 %
			β -turn = 11.2 %



Figure S4: XRF of mixed metal ions (100 ppm of each vanadyl sulfate, cadmium chloride, mercury chloride, lead acetate, uranyl nitrate) before and after treated with EKKEDRGDEKKE peptide amyloids at pH 11.7.

Table S3: Relative concentrations of cadmium (Cd^{2+}) , mercury (Hg^{2+}) , lead (Pb^{2+}) , uranyl $(U(O)_2^{2+})$ and vanadyl $(V(O)^{2+})$ ions in solution before and after incubation with the *EKKEDRGDEKKE* peptide amyloids at pH 11.7.

Metal	Before filtration	After filtration
Vanadium	20.258%	25.949%
Cadmium	13.136%	16.274%
Mercury	17.701%	30.845%
Lead	23.032%	15.566%
Uranium	25.832%	11.367%
Total	100%	100%



Figure S5: [A] XPS survey spectra of EKKEDRGDEKKE self-assembled amyloid peptide at the three different pH conditions, pH 3.6, 7.4 and 11.7. [B] XPS survey spectra of EKKEDRGDEKKE self-assembled amyloid peptide incubated with lead acetate metal solution at three different pH conditions, pH 3.6, 7.4 and 11.7.



Figure S6: (C1s XPS of peptide and Pb-complexes at three pHs): C1s spectra of (a) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 (b) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 and 5ppm lead acetate (c) EKKEDRGDEKKE self-assembled peptide at pH 7.4 (d) EKKEDRGDEKKE self-assembled peptide at pH 7.4 (d) EKKEDRGDEKKE self-assembled peptide at pH 7.4 (d) EKKEDRGDEKKE self-assembled peptide at pH 7.4 and lead acetate (e) EKKEDRGDEKKE self-assembled at pH 11.7 (f) EKKEDRGDEKKE self-assembled peptide amyloid at pH 11.7 and 5ppm lead acetate.

Table S4 (C1s XPS of peptide and Pb-complexes at three pHs): C1s binding energy (eV) values for self-assembled EKKEDRGDEKKE peptide and EKKEDRGDEKKE peptide and lead metal complex.

C1s values for EKKEDRGDEKKE peptide		C1s values for EKKEDRGDEKKE peptide and					
				lead metal			
Functional	pH 3.6	pH 7.4	pH 11.7	Functional	pH 3.6 +	pH 7.4 +	pH 11.7 +
Group C(1s)				Group C(1s)	Lead	Lead	Lead
Hydrocarbon	284.8 eV	284.8 eV	284.8 eV	Hydrocarbon	284.7 eV	284.8 eV	284.8 eV
amine(C-N)	286.7 eV	286.8 eV	286.8 eV	amine(C-N)	286.7 eV	286.4 eV	286.6 eV
and alpha				and alpha			
carbon (C-N-				carbon (C-N-			
CO)				CO)			
carboxylate							
(O=C-O ⁻)							
amide	288.9 eV	288.7 eV	288.7 eV	amide	288.8 eV	288.1 eV	288.3 eV
((C=O)-N),				((C=O)-N),			
and				carboxylate			
carboxylic				$(O=C-O^{-})$ and			
acid (O=C-				carboxylic			
OH)				acid (O=C-			
				OH)			
				Lead bound		289.6 eV	289.8 eV
				carboxylate			
				(O=C-O-Pb)			



Figure S7: (O1s XPS of peptide and Pb-complexes at three pHs): O1s spectra of (a) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 (b) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 and 5ppm lead acetate (c) EKKEDRGDEKKE self-assembled peptide at pH 7.4 (d) EKKEDRGDEKKE self-assembled peptide at pH 7.4 and 5ppm lead acetate (e) EKKEDRGDEKKE self-assembled peptide amyloid at pH 11.7 (f) EKKEDRGDEKKE self-assembled peptide amyloid at pH 11.7 and 5ppm lead acetate.

Table S5 (O1s XPS of peptide and Pb-complexes at three pHs). *O1s binding energy (eV) values for self-assembled EKKEDRGDEKKE peptide and EKKEDRGDEKKE peptide and lead metal complex.*

O1s values for EKKEDRGDEKKE peptide		O1s values for EKKEDRGDEKKE peptide and					
				lead metal			
Functional	рН 3.6	pH 7.4	pH 11.7	Functional	pH 3.6 +	pH 7.4 +	pH 11.7 +
Group O(1s)				Group O(1s)	Lead	Lead	Lead
Carboxylate				Carboxylate			529.3 eV
(O=C-O ⁻)				(O=C-O ⁻)			
Carboxylic acid	530.5 eV			Carboxylic	530.0 eV	530.6 eV	
(O=C-OH)				acid (O=C-			
				OH)			
Amide (O=C-	532.2 eV	531.7 eV	532.0 eV	Amide (O=C-	532.2 eV	532.1 eV	531.8 eV
NH-)				NH-)			
Carboxylic acid	534.1 eV			Carboxylic	534.5 eV		
(HO-C=O)				acid (HO-			
				C=O)			
Carboxylate-		535.3 eV	534.4 eV	Carboxylate-			
Na ⁺				Na ⁺			
				Carboxylate-		535.4 eV	535.0 eV
				Pb ²⁺			



Figure S8: N1s spectra of (A) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 (B) EKKEDRGDEKKE self-assembled peptide amyloid at pH 3.6 and 5ppm lead acetate (C) EKKEDRGDEKKE self-assembled peptide amyloid at pH 7.4 (D) EKKEDRGDEKKE self-assembled peptide at pH 7.4 and 5ppm lead acetate (E) EKKEDRGDEKKE self-assembled peptide amyloid at pH 11.7 (F) EKKEDRGDEKKE self-assembled peptide amyloid at pH 11.7 and 5ppm lead acetate.

Table S6: (N1s XPS of peptide and Pb-complexes at three pHs). N1s binding energy (eV) values for

 self-assembled EKKEDRGDEKKE peptide and EKKEDRGDEKKE peptide and lead metal complex.

N1s values for EKKEDRGDEKKE peptide			N1s values for EKKEDRGDEKKE peptide and lead metal				
Functional	pH 3.6	pH 7.4	pH 11.7	Functional	рН 3.6	pH 7.4	pH 11.7
Group N(1s)				Group N(1s)			
Amine (C-NH ₂)		398.1 eV	397.8 eV	Amine (C-NH ₂)			397.6
Amide (O=C-	399.8 eV	399.7 eV	399.7 eV	Amide (O=C-	399.8	399.6	399.6
NH-C)				NH-C)			
Amine+H ⁺ ion	402.5 eV	401.6 eV		Amine+H ⁺ ion	402.5		
(R-NH 3 ⁺)				(R-NH 3 ⁺)			
				Amine + Pb ²⁺		402.0	402.0



Figure S9: Benesi-Hildebrand plot of $[(F_0-F_m)/(F_0-F)]$ (F_0 is the fluorescence intensity without the peptide and F_{min} is the minimum fluorescence intensity upon quenching), versus (a) [1/Q] (Q= peptide concentration in μ M) showcasing 1:1 metal and peptide binding. and (b) $[1/Q^2]$ (Q= peptide concentration) representing 1:2 metal and peptide binding.



Figure S10: U4f XPS spectra of uranyl nitrate (black) at U⁶⁺ oxidation state, uranyl nitrate - EKKEDRGDEKKE complex (red) and uranyl nitrate EKKEDRGDEKKE complex after x-ray exposure for 10 minutes (blue) at U⁵⁺ oxidation state.



Figure S11: Deconvoluted XPS spectra of O and N atoms of uranyl nitrate, EKKEDRGDEKKE peptide and EKKEDRGDEKKE peptide incubated with uranyl nitrate to represent the various contributions of the various bonds based on the binding energy (eV) values.

Table S7: (O1s N1s XPS of peptide and U-complex). O1s and N1s binding energy (eV) values for uranyl nitrate, self-assembled EKKEDRGDEKKE peptide, and EKKEDRGDEKKE peptide - uranyl nitrate complex.

Sample	O1s	N1s
Uranyl nitrate	531.56 eV (U=O); 532.56 eV (NO ₃ ⁻); 532.21 (major)	407.2 eV (NO ₃ ⁻)
EKKEDRGDEKKE	532.01 eV (amide); 534.42 eV (carboxylate-Na ⁺) 532.04 eV (major)	399.7 eV (amide); 397.8 eV (amine)
Uranyl nitrate+ EKKEDRGDEKKE	530.70 eV (carboxylate); 531.86 eV (Uranyl); 531.96 eV (amide); 533.72 eV (carboxylate-uranyl) 531.95 (major)	401.9 eV (uranyl bound amine); 400.1 eV (amide N); 398.2 eV (unbound amine)