

Supporting Information

Two new rare-earth oxyborates $\text{Ba}_4\text{BiTbO}(\text{BO}_3)_4$ and $\text{Ba}_{1.54}\text{Sr}_{2.46}\text{BiTbO}(\text{BO}_3)_4$ and luminescent properties of the $\text{Ba}_4\text{BiTb}_{1-x}\text{Eu}_x\text{O}(\text{BO}_3)_4$ phosphors

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Table S1 Atomic coordinates, site occupancies and equivalent isotropic displacement parameters (\AA^2) for $\text{Ba}_4\text{BiTbO}(\text{BO}_3)_4$ and $\text{Ba}_{1.54}\text{Sr}_{2.46}\text{BiTbO}(\text{BO}_3)_4$.

Atoms	Wyck sites	Site symmetry	x	y	z	Occupancies	U_{eq}
$\text{Ba}_4\text{BiTbO}(\text{BO}_3)_4$							
Ba1	$4f$	C_{3v}	0.6667	0.3333	0.08794(3)	1	0.0146(2)
Ba2	$2d$	D_{3h}	0.6667	0.3333	0.2500	1	0.0160(2)
Ba3	$4e$	C_{3v}	0.0000	0.0000	0.15797(9)	0.5	0.0132(4)
Bi1	$4e$	C_{3v}	0.0000	0.0000	0.17011(6)	0.5	0.0355(6)
Tb1	$2a$	D_{3d}	0.0000	0.0000	0.0000	1	0.00881(19)
B1	$4f$	C_{3v}	0.3333	0.6667	0.0538(6)	1	0.011(2)
B2	$4f$	C_{3v}	0.3333	0.6667	0.1702(5)	1	0.012(2)
O1	$12k$	C_s	0.1871(6)	0.3742(12)	0.0537(2)	1	0.0204(11)
O2	$12k$	C_s	0.0396(14)	0.5198(7)	0.16991(19)	1	0.0218(12)
O3	$2b$	D_{3h}	0.0000	0.0000	0.2500	1	0.029(3)
$\text{Ba}_{1.54}\text{Sr}_{2.46}\text{BiTbO}(\text{BO}_3)_4$							
Ba1/Sr1	$4f$	C_{3v}	0.6667	0.3333	0.08892(5)	0.58(2)/0.42(2)	0.0163(5)
Ba2/Sr2	$2d$	D_{3h}	0.6667	0.3333	0.2500	0.38(3)/0.62(3)	0.0170(7)
Sr3	$4e$	C_{3v}	0.0000	0.0000	0.15810(19)	0.5	0.0062(7)
Bi1	$4e$	C_{3v}	0.0000	0.0000	0.17066(9)	0.5	0.0307(7)
Tb1	$2a$	D_{3d}	0.0000	0.0000	0.0000	1	0.0199(4)
B1	$4f$	C_{3v}	0.3333	0.6667	0.0548(11)	1	0.020(4)
B2	$4f$	C_{3v}	0.3333	0.6667	0.1725(10)	1	0.017(3)
O1	$12k$	C_s	0.1857(12)	0.371(2)	0.0554(4)	1	0.032(2)
O2	$12k$	C_s	0.035(2)	0.5177(11)	0.1721(4)	1	0.026(2)
O3	$2b$	D_{3h}	0.0000	0.0000	0.2500	1	0.029(4)

Note: U_{eq} is defined as one third of the trace of the orthogonalized \mathbf{U} tensor.

Table S2 Selected bond lengths (\AA) and angles ($^\circ$) for $\text{Ba}_4\text{BiTbO}(\text{BO}_3)_4$.

Ba1-O2 $\times 3$	2.780(6)	Bi1-O3	2.1055(15)
Ba1-O1 $\times 6$	2.862(2)	Bi1-O2 $\times 6$	2.7157(5)
Ba2-O2 $\times 6$	2.742(6)	Tb1-O1 $\times 6$	2.255(6)
Ba2-O3 $\times 3$	3.12846(5)	B1-O1 $\times 3$	1.372(6)
Ba3-O3	2.425(2)	B2-O2 $\times 3$	1.379(7)
Ba3-O2 $\times 6$	2.7339(8)	Ba3-Bi1	0.320(3)
O1-Tb1-O1 $\times 3$		O1-B1-O1 $\times 3$	119.999(14)
)
O1-Tb1-O1 $\times 6$	84.8(2)	O2-B2-O2 $\times 3$	119.998(15)
)
O1-Tb1-O1 $\times 6$	95.2(2)		

Table S3 Selected bond lengths (\AA) and angles ($^\circ$) for $\text{Ba}_{1.54}\text{Sr}_{2.46}\text{BiTbO}(\text{BO}_3)_4$.

Ba1/Sr1-O2 $\times 3$	2.763(9)	Bi1-O3	2.068(2)
Ba1/Sr1-O1 $\times 6$	2.827(4)	Bi1-O2 $\times 6$	2.6880(6)
Ba2/Sr2-O2 $\times 6$	2.657(10)	Tb1-O1 $\times 6$	2.250(11)
Ba2/Sr2-O3 $\times 3$	3.09768(11)	B1-O1 $\times 3$	1.372(11)
Sr3-O3	2.396(5)	B2-O2 $\times 3$	1.384(10)
Sr3-O2 $\times 6$	2.7123(16)	Sr3-Bi1	0.328(5)
O1-Tb1-O1 $\times 3$		O1-B1-O1 $\times 3$	119.99(6)
O1-Tb1-O1 $\times 6$	83.3(4)	O2-B2-O2 $\times 3$	119.99(4)
O1-Tb1-O1 $\times 6$	96.7(4)		

Table S4 Rietveld refinement results of $\text{Ba}_4\text{BiTb}_{1-x}\text{Eu}_x\text{O}(\text{BO}_3)_4$ ($0 \leq x \leq 1$) and $\text{Ba}_4\text{BiYO}(\text{BO}_3)_4$.

Eu³⁺ content	x=0	x=0.001	x=0.002	x=0.005	x=0.01
Space group	<i>P</i> 6 ₃ /mmc				
a (Å)	5.42269(15)	5.4231(3)	5.4236(5)	5.4238(4)	5.4243(4)
c (Å)	26.3902(9)	26.4080(14)	26.413(2)	26.411(2)	26.409(2)
V (Å ³)	672.05(4)	672.60(8)	672.83(13)	672.86(12)	672.93(11)
2θ range (°)	10–90	10–90	10–90	10–90	10–90
R_p , %	4.17	4.31	3.95	3.90	4.00
R_{wp} , %	7.89	7.86	7.17	7.10	7.30
GOF	4.77	3.55	3.15	3.11	3.21

Eu³⁺ content	x=0.05	x = 0.1	x = 0.2	x = 1	Ba ₄ BiYO(BO ₃) ₄
Space group	<i>P</i> 6 ₃ /mmc				
a (Å)	5.4245(4)	5.4256(5)	5.4257(4)	5.4322(3)	5.4160(9)
c (Å)	26.410(2)	26.416(2)	26.423(2)	26.5305(16)	26.289(4)
V (Å ³)	673.02(12)	673.43(13)	673.62(11)	677.99(8)	667.8(2)
2θ range (°)	10–90	10–90	10–90	10–90	10–90
R_p , %	3.88	4.13	4.20	5.64	5.97
R_{wp} , %	6.98	7.42	7.63	10.33	10.73
GOF	3.04	3.31	3.41	6.69	6.52

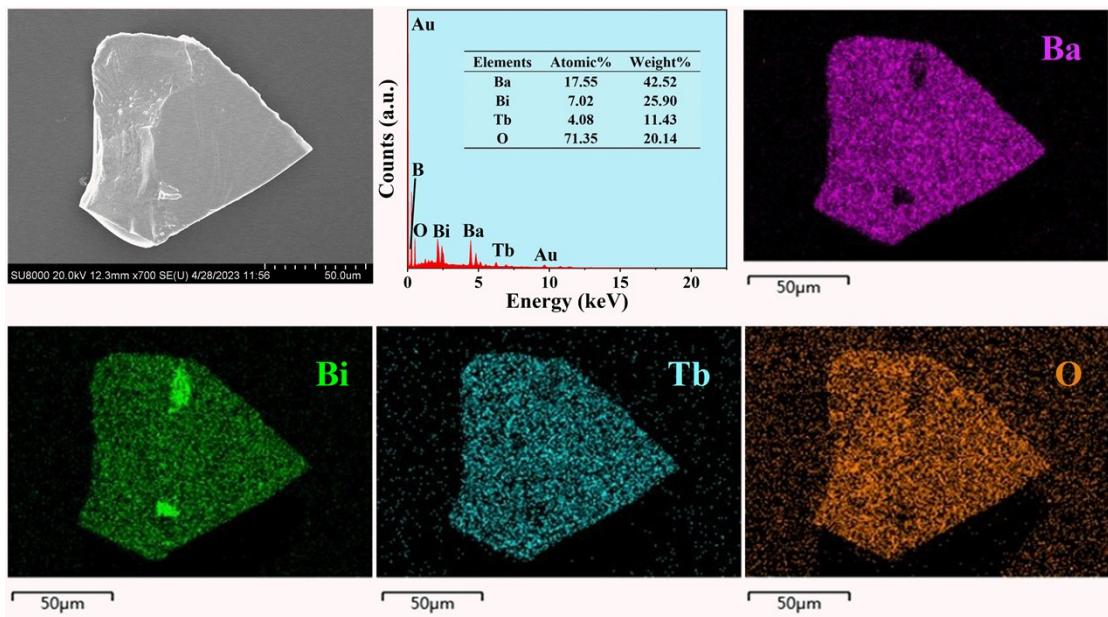


Fig. S1 A FE-SEM image, EDX results and elemental mapping of a typical $\text{Ba}_4\text{BiTbO}(\text{BO}_3)_4$ single crystal. The Au element comes from the pretreatment process (A thin layer of Au was evaporated on the sample surface to provide electrical conductivity).

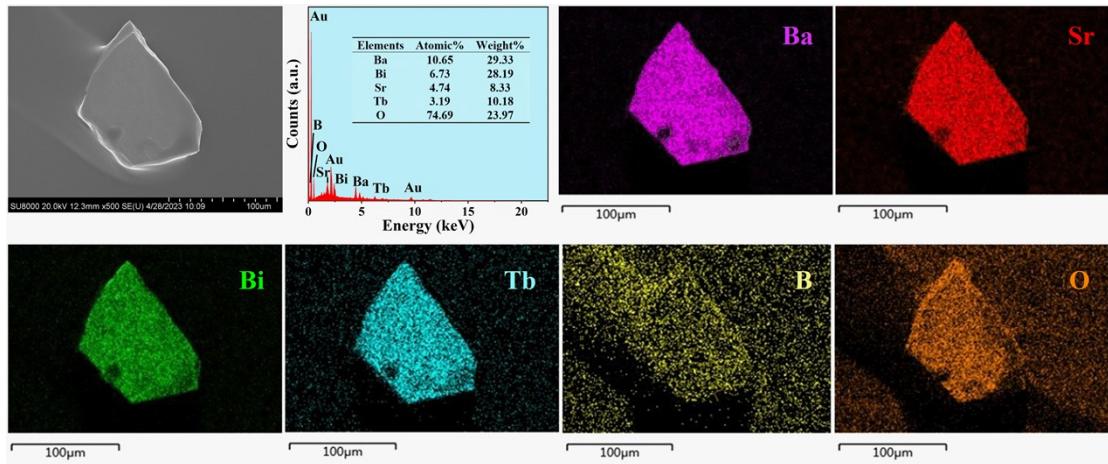


Fig. S2 A FE-SEM image, EDX results and elemental mapping of a typical $\text{Ba}_{1.54}\text{Sr}_{2.46}\text{BiTbO}(\text{BO}_3)_4$ single crystal.

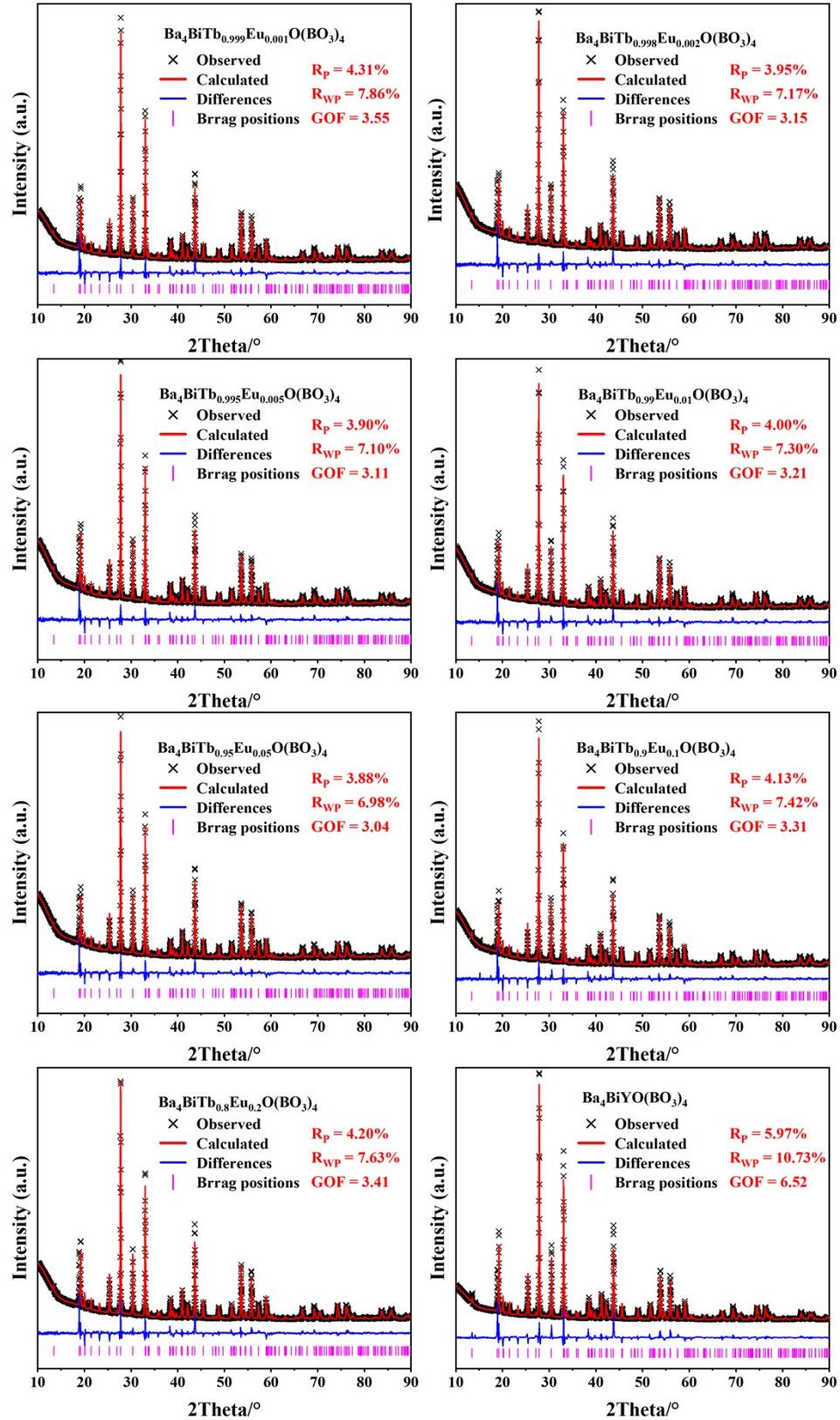


Fig. S3 Rietveld plots of Ba₄BiTb_{1-x}Eu_xO(BO₃)₄ (x = 0.001, 0.002, 0.005, 0.01, 0.05, 0.1, and 0.2) and Ba₄BiYO(BO₃)₄.

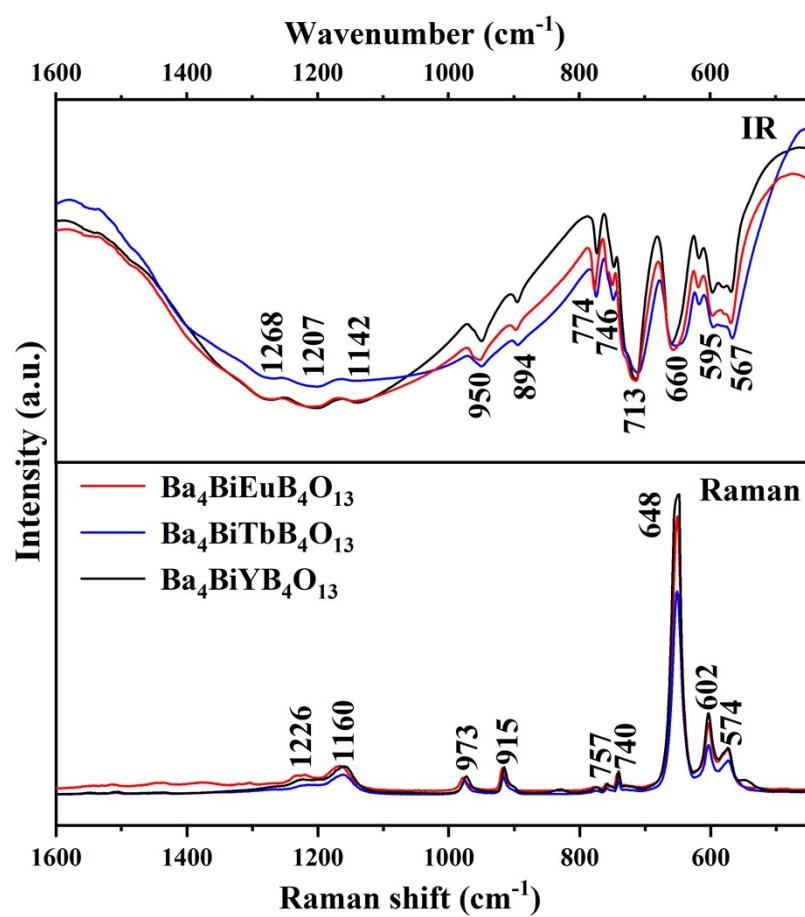


Fig. S4 Infrared and Raman spectra of $\text{Ba}_4\text{BiLnO(BO}_3)_4$ ($\text{Ln} = \text{Y, Tb and Eu}$).

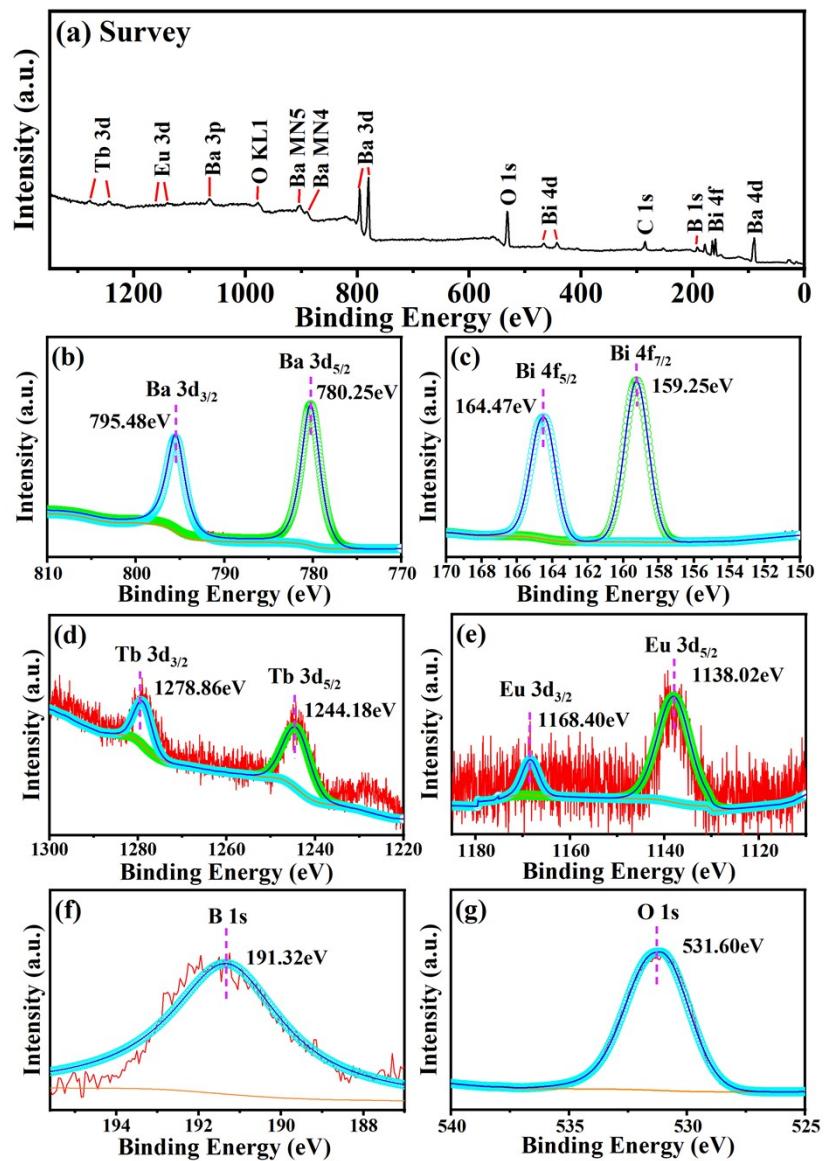


Fig. S5 XPS survey (a) and core-level spectra of Ba 3d (b), Bi 4f (c), Tb 3d (d), Eu 3d (e), B 1s (f) and O 1s (g) for $\text{Ba}_4\text{BiTb}_{0.999}\text{Eu}_{0.001}\text{O}(\text{BO}_3)_4$.

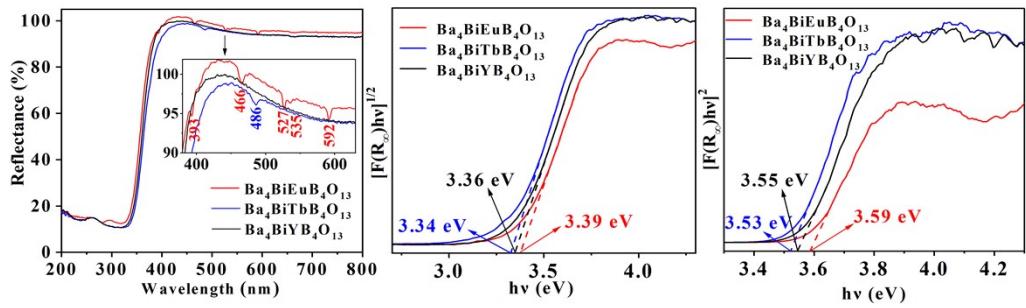


Fig. S6 UV–vis diffuse reflection spectra of $\text{Ba}_4\text{BiLnO(BO}_3)_4$ ($\text{Ln} = \text{Y, Tb and Eu}$) (left). Tauc plots for indirect (middle) and direct (right) transitions are also shown in this figure.