

**Green Synthesis of Surfactant-free Mesoporous Silica with Strong Hydrophilicity  
via Metal Salt Modifications for Moisture Adsorption  
Super-hydrophilic Surfactant-free Mesoporous Silica**

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**Supplementary data**

**Table S1** Percentage of hydrophilic functional groups (Si-O- and Si-OH) of metal salt-  
modified mesoporous silica.

**Figure S1:** TEM images of (a) MPS and NaCl-modified MPS products: (b)  
MPS\_0.125NaCl, (c) MPS\_0.250NaCl, (d) MPS\_0.375NaCl, and (e) MPS\_0.500NaCl.

**Figure S2:** TEM images of (a) MPS and MgCl<sub>2</sub>-modified MPS products: (b)  
MPS\_0.125MgCl<sub>2</sub>, (c) MPS\_0.250MgCl<sub>2</sub>, (d) MPS\_0.375MgCl<sub>2</sub>, and (e)  
MPS\_0.500MgCl<sub>2</sub>.

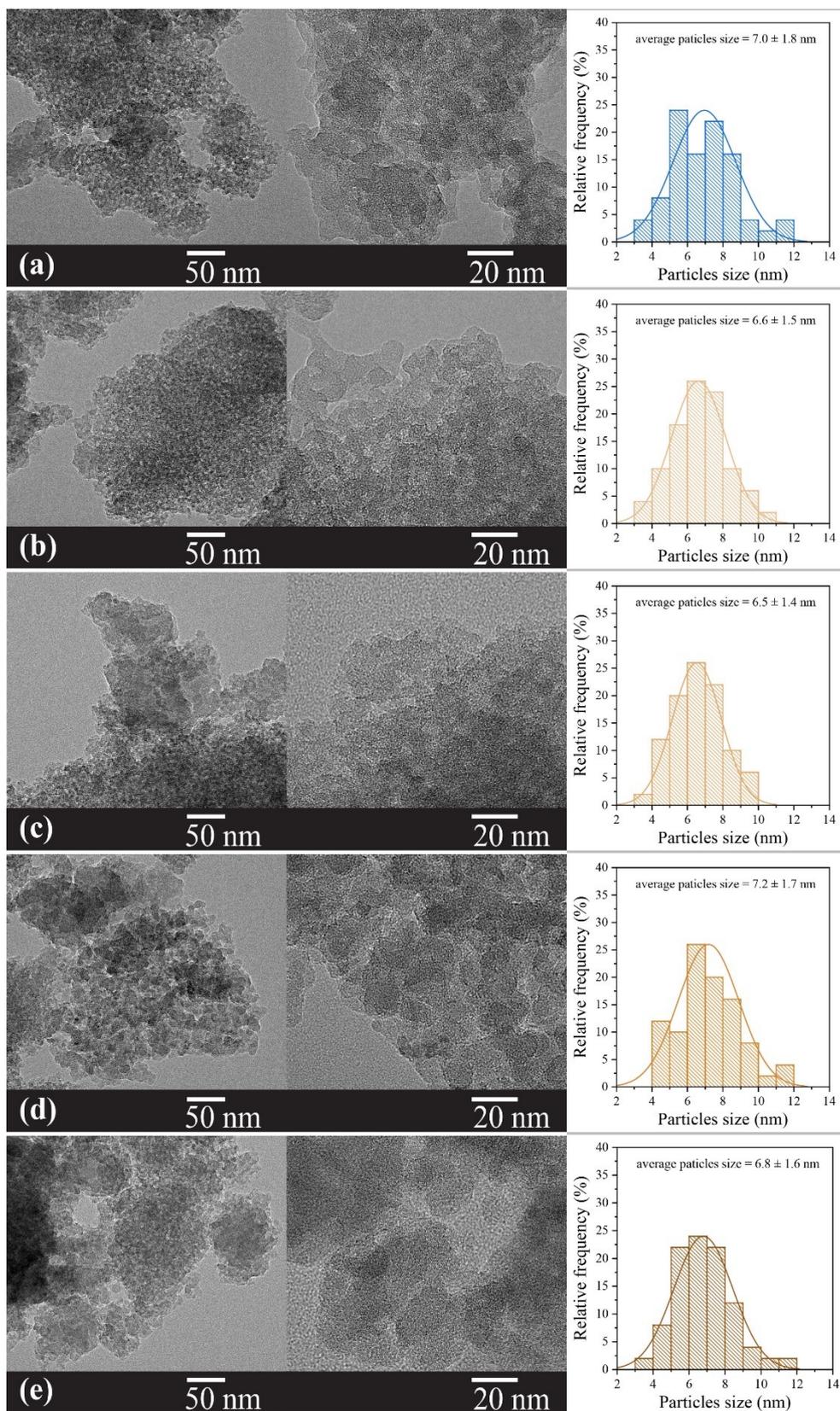
**Figure S3:** FTIR spectra of (a) silica gel, (b) MPS and MPS modified by (A) NaCl, (B)  
KCl, (C) MgCl<sub>2</sub>, and (D) CaCl<sub>2</sub> salts with concentrations of 0.125 (c, g, k, o); 0.250 (d,  
h, l, p); 0.375 (e, i, m, q); and 0.500 (f, j, n, r) g salt/g SiO<sub>2</sub>.

**Figure S4:** DTG curves of silica gel, MPS and MPS modified by (A) NaCl, (B) KCl, (C)  
MgCl<sub>2</sub>, and (D) CaCl<sub>2</sub> salts.

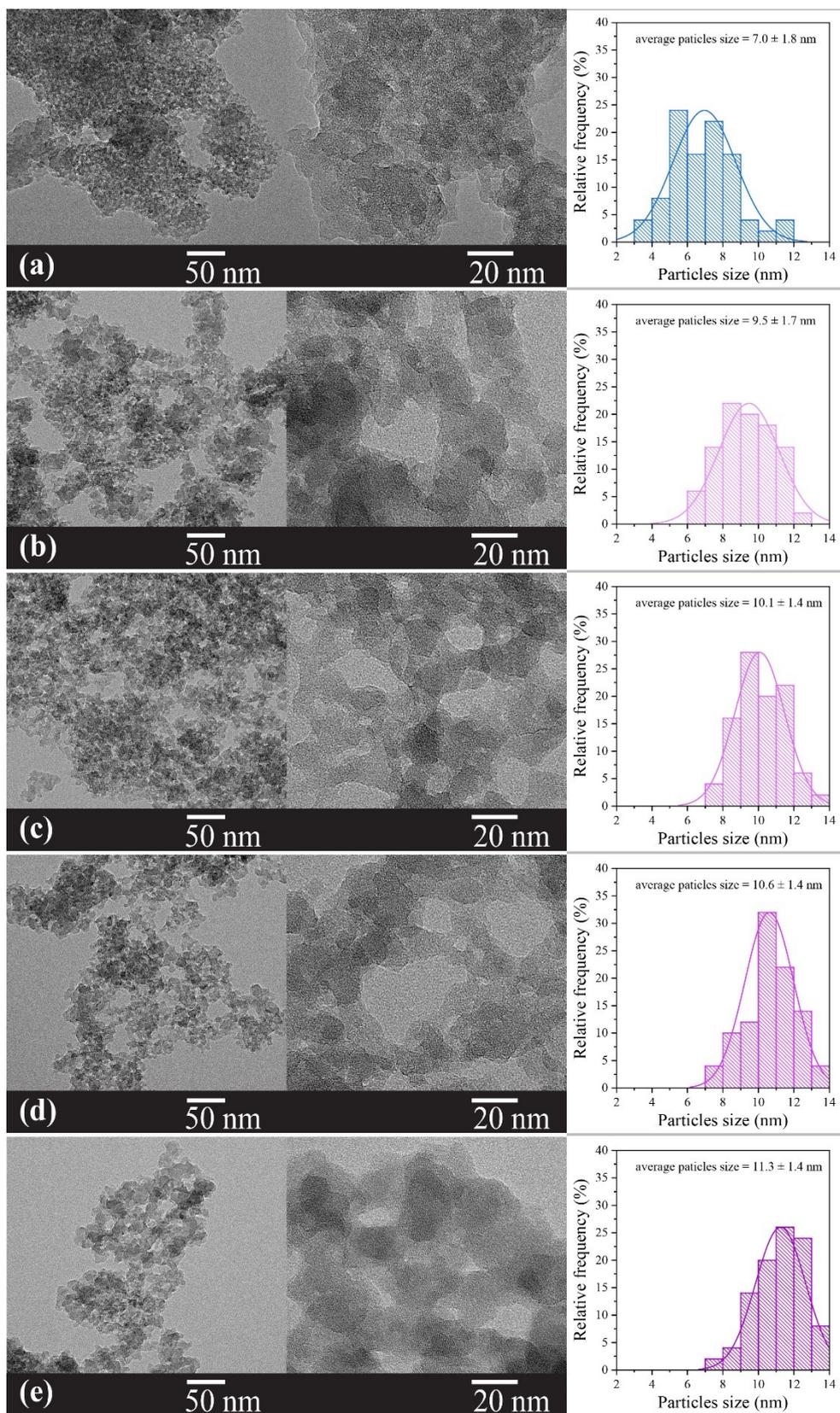
**Table S1** Percentage of hydrophilic functional groups (Si-O- and Si-OH) of metal salt-modified mesoporous silica.

Adsorbents	Percentage of hydrophilic functional group <sup>a</sup> (%)	
	Si-O-	Si-OH
Silica gel	4.91	2.86
MPS	4.48	2.75
MPS_0.125NaCl	4.66	2.50
MPS_0.250NaCl	4.85	2.60
MPS_0.375NaCl	5.09	2.70
MPS_0.500NaCl	5.47	3.04
MPS_0.125KCl	4.28	2.65
MPS_0.250KCl	4.62	2.82
MPS_0.375KCl	4.67	2.83
MPS_0.500KCl	4.93	2.95
MPS_0.125MgCl <sub>2</sub>	4.66	2.50
MPS_0.250MgCl <sub>2</sub>	3.49	2.18
MPS_0.375MgCl <sub>2</sub>	3.03	1.83
MPS_0.500MgCl <sub>2</sub>	2.30	1.40
MPS_0.125CaCl <sub>2</sub>	1.27	0.84
MPS_0.250CaCl <sub>2</sub>	1.22	0.84
MPS_0.375CaCl <sub>2</sub>	1.18	0.78
MPS_0.500CaCl <sub>2</sub>	1.12	0.75

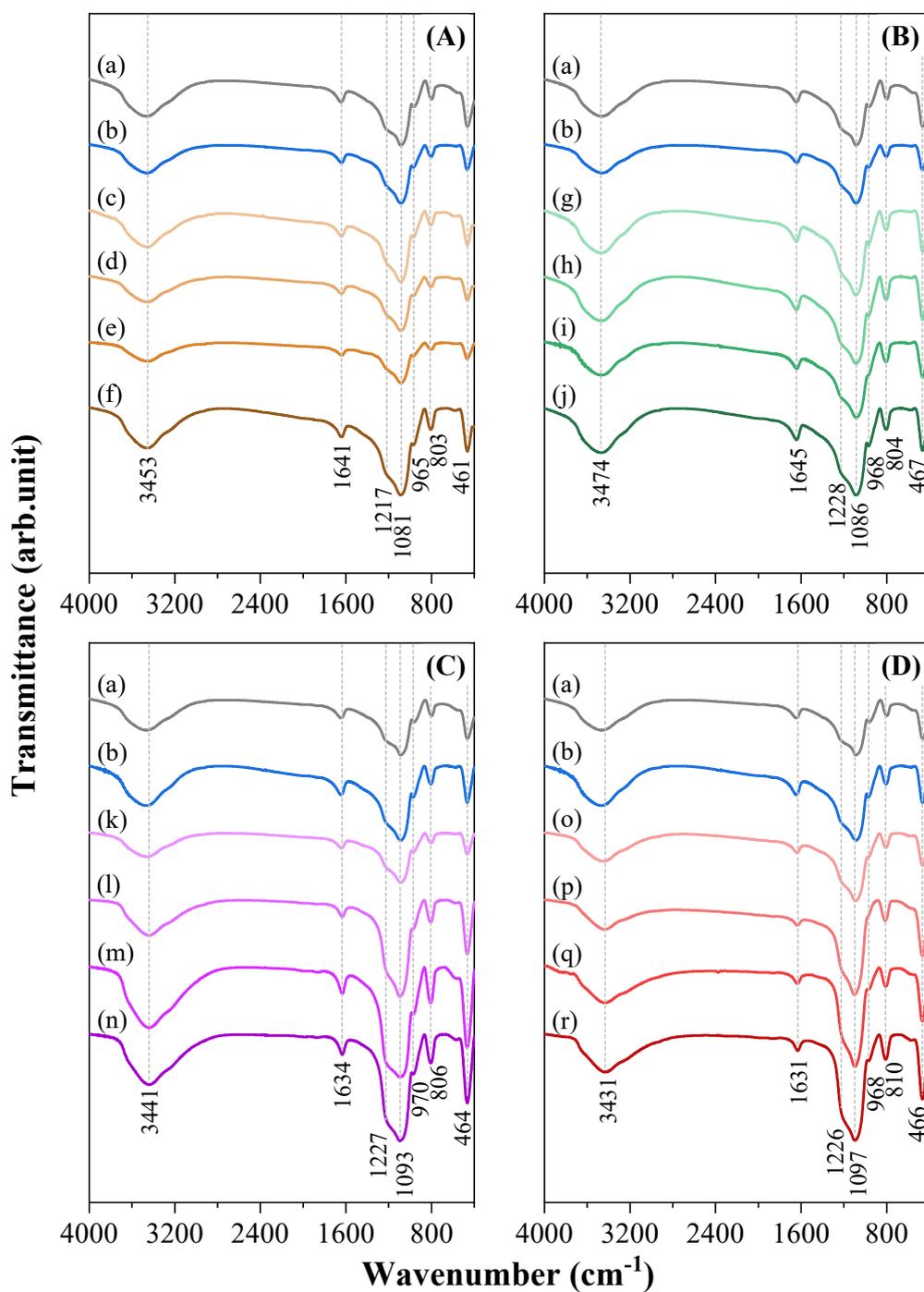
<sup>a</sup> Calculated by using area from deconvolution of FTIR spectra.



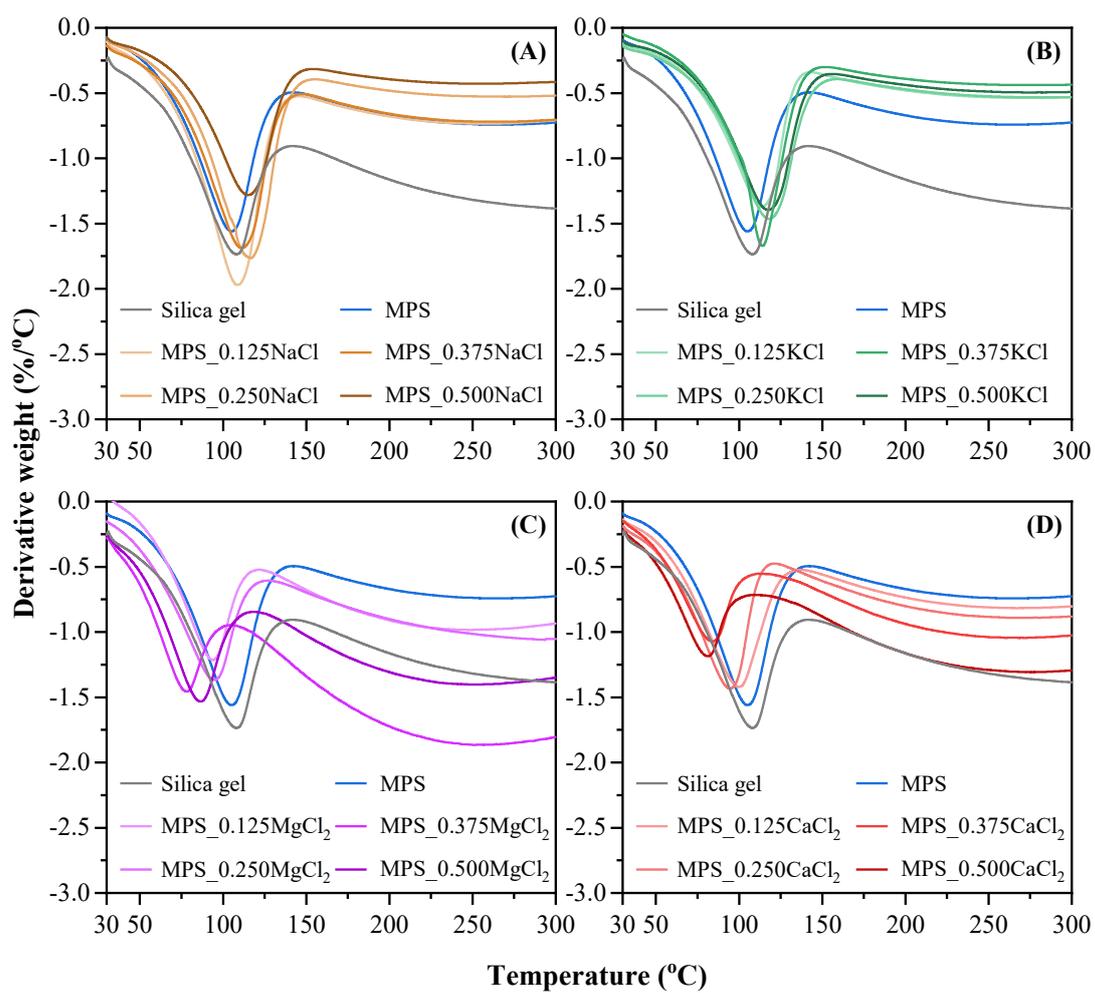
**Figure S1:** TEM images and particle size distributions of (a) MPS and NaCl-modified MPS products: (b) MPS\_0.125NaCl, (c) MPS\_0.250NaCl, (d) MPS\_0.375NaCl, and (e) MPS\_0.500NaCl.



**Figure S2:** TEM images and particle size distributions of (a) MPS and  $\text{MgCl}_2$ -modified MPS products: (b) MPS\_0.125 $\text{MgCl}_2$ , (c) MPS\_0.250 $\text{MgCl}_2$ , (d) MPS\_0.375 $\text{MgCl}_2$ , and (e) MPS\_0.500 $\text{MgCl}_2$ .



**Figure S3:** FTIR spectra of (a) silica gel, (b) MPS and MPS modified by (A) NaCl, (B) KCl, (C)  $\text{MgCl}_2$ , and (D)  $\text{CaCl}_2$  salts with concentrations of 0.125 (c, g, k, o); 0.250 (d, h, l, p); 0.375 (e, i, m, q); and 0.500 (f, j, n, r) g salt/g  $\text{SiO}_2$ .



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**Figure S4:** DTG curves of silica gel, MPS and MPS modified by (A) NaCl, (B) KCl, (C) MgCl<sub>2</sub>, and (D) CaCl<sub>2</sub> salts.