Supporting Information

Visible-light-induced chemo-, diastereo- and enantioselective α -C(sp³)–H functionalization of alkyl silanes

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1. General remarks

10 W or 20 W blue LEDs (Chinese Taobao, https://shop112050379.taobao.com) were used as light sources. ¹H NMR spectra were recorded on Bruker ASCENDTM 400M (400 MHz) or 600M (600 MHz). ¹³C{¹H} NMR data were collected on Bruker ASCENDTM 400M (101 MHz) or 600M (151 MHz) with complete proton decoupling. Chemical shifts were recorded in ppm relative to tetramethylsilane and with the solvent resonance as the internal standard (CDCl₃, δ = 7.26) for ¹H NMR and (CDCl₃, $\delta = 77.0$) for ¹³C{¹H} NMR. Data were reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constants (Hz), integration. ¹⁹F{¹H} NMR spectra were collected on Bruker ASCENDTM 400M (376 MHz) with complete proton decoupling. Chemical shifts δ are given relative to CFCl₃ (external reference, δ 19 F(CFCl₃) = 0). Enantiomeric excesses were determined by chiral HPLC analysis on Daicel Chiralcel AD-H or chiral UPC² analysis on Phenomenex Chiralcel Lux 5u Cellulose at 35 °C with UV detector at 220 nm in comparison with the authentic racemates. Optical rotations were determined after flash column chromatography purification and reported as follows: $[\lambda]_D^T$ (c: g/100 mL, in CH₂Cl₂). HRMS were recorded on Thermo Q-Exactive Focus (FTMS+c ESI). IR spectra were recorded on SHIMADZU IR Tracer-100 FT-IR spectrophotometer. Emission intensities were recorded using a F-7000 FL Spectrophotometer. EPR spectra were recorded at room temperature on a Bruker EPR A300. All reactions were performed in sealed oven-dried glass tubes under an atmosphere of nitrogen unless otherwise noted. CHCl₃ was distilled over CaH₂. All the solvents were purified by usual methods before use. Chromatography: Qingdao Haiyang silica gel, HG/T2354-92, HCP. The chiral N,N'dioxides were prepared according to methods reported in the literature.¹⁻³

2. The synthesis of substrates

2.1 General procedure for the synthesis of N-sulfonyl cyclic ketimines

$$R + O_{D_{2}}O_{D_{$$

Butyllithium (30.75 mmol, 2 M in THF) was added dropwise over a 20 minutes period to a cold (0 °C), mechanically stirred solution of aryl sulfonamide (15 mmol) in anhydrous tetrahydrofuran (80 mL) under a dry nitrogen atmosphere. The mixture was stirred for an additional 25 min at 0 °C and a precipitate was formed. The suspension was cooled to -78 °C and dibenzyl oxalate (45 mmol) in THF (20 mL) was added dropwise over 10 minutes. The cooling bath was removed and the suspension was stirred at ambient temperature for 2 h. The reaction was quenched with 5% HCl (40 mL) and added to water (200 mL). The organic phase was extracted with ether (3×50 mL). The ether phase was washed with brine (200 mL). The solvent was removed and the crude product was obtained used directly in the next step without further purification. To the crude product obtained above, formic acid (25 mL) was added and the suspension was stirred at room temperature under a dry nitrogen atmosphere. After 5 min dissolution occurred. After 24 h the solution was concentrated and the resultant solid was dissolved in CH₂Cl₂ and concentrated to remove traces of formic acid. This afforded the title compound as a solid which was further purified by flash chromatography (DCM/PE = 1/2 - DCM).⁴

2.2 General procedure for the synthesis of alkyl silanes

General procedure I (for 2a, 2b, 2c, 2d, 2f, 2h, 2i)⁵



A flame-dried, three-neck, round-bottom flask, equipped with a mechanical stirrer, reflux condenser, N₂ inlet, and addition funnel, was charged with dry magnesium turnings (0.528 g, 22 mmol), dry tetrahydrofuran (20 mL), and chlorotrimethylsilane (3.26 mL, 30 mmol). The appropriate benzyl chloride (20 mmol) in dry tetrahydrofuran (20 mL) was added slowly, at a rate to maintain gentle reflux. After addition was complete, the mixture was heated under reflux 2 h, cooled, and poured into cold water. Pentane (15 mL) was added and the pentane layer washed three times with cold water (30 mL) and once with saturated NaCl solution (30 mL), dried (MgSO₄), and then rotary evaporated to yield the crude benzyl trimethylsilanes, which was purified by column chromatography to yield the silane derivatives as colorless oils.

General procedure II (for 2g)⁶



To a suspension of Mg-tunings (0.3 g) in Et_2O (10 mL) was added a solution of (2.9 g, 12 mmol) of *p*-bromobenzyl bromide (Aldrich) in Et_2O (20 mL). To the resulting Grignard reagent was added (3.0 mL, 24 mmol) of chlorotrimethylsilane (Aldrich), and the resulting mixture was stirred at 25 °C for 12 h. Addition of ice-water followed by separation, drying, and concentration of the mixture gave an oil which was subjected to molecular distillation (0.05 mm, 40 °C) to the silylbromoarene **2g** as an oil.

General procedure III (for 2j)⁷



A solution of haloarene (1.0 equiv) and Ni(acac)₂ (0.05 equiv) in dry THF (0.24 M) was stirred for 5 min at room temperature and then cooled to 0 °C. A solution Me₃SiCH₂MgCl of in Et₂O (1.5 equiv) was added *via* a cannula, and the mixture was allowed to stir at the indicated temperature for 2 h. At this point, the reaction was quenched with water and extracted with Et₂O three times. The combined organic layers were washed with water, brine, dried over Na₂SO₄ (anhydrous) and then concentrated under reduced pressure. Purification of the crude residue via silica gel flash column chromatography afforded benzyltrimentylsilane.

General procedure IV (for 2k)⁸



Dry magnesium turnings (0.85 g, 35 mmol) and TMSCl (5.7 mL, 44. mmol) was added as a single portion into THF (21 mL). The mixture was cooled to 0 °C and maintained for 15 min. 2-Naphthylbromide (7.00 g, 31.7 mmol) was dissolved in THF (56 mL) and added dropwise *via* syringe pump. The mixture was maintained for 5 h, and then it was allowed to warm to ambient temperature and maintained for 12 h. The mixture was then cooled to 0 °C, quenched and worked up as described in the general procedure I. Flash chromatography (2% Et₂O/petroleum ether) afforded 6.10 g (90%) of **2k** as a white solid.

General procedure V (for 2l, 2m, 2n)⁸



Activated Mg powder (1.1 equiv) was added as a single portion into a solution of aryl chloride in THF (0.8 M). The suspension was cooled to 0 °C and 1 h later, the mixture was allowed to warm to ambient temperature and was maintained for another 1 h. The reaction was then heated to reflux and the indicated chlorosilane (1.4 equiv) was added dropwise. After 12 h, the mixture was cooled to 0 °C and the reaction was quenched with saturated aqueous NH₄Cl. The mixture was allowed to warm to ambient temperature, concentrated in vacuo, then taken up in Et₂O and washed three times with brine. The combined aqueous phases were extracted twice with Et₂O and the combined organic layers were dried over MgSO₄ and then concentrated in vacuo. The arylsilanes products were purified by flash chromatography.

General procedure VI (for 2p)⁹



An oven-dried, argon purged two neck round-bottomed flask fitted with a condenser and septum was added magnesium turnings (750 mg, 2.2 equiv) and anhydrous TFH (8 mL). Chloromethyl trimethylsilane (20 mmol, 2.8 mL, 3 equiv) was added dropwise to maintain reflux. The mixture was cooled to ambient temperature and transferred by cannula into a 50 mL sealable Schlenk tube containing a solution of 3-bromobenzothiophene (1.41 g, 6.65 mmol) and palladium tetrakis-(triphenylphosphine)-palladium (384 mg, 5 mol%) in anhydrous THF (10 mL). The mixture was heated to 80 °C for 48 hours. After the completion of this reaction, the mixture was cooled to room temperature and pour the solution into cold HCl solution (1 M, 50 mL), and then extracted the aqueous phase twice with pentane, the combined organic phase was dried over magnesium sulfate and concentrate in vacuo. The crude product was purified by flash chromatography (hexane 100%) to afford (benzo[b]thiophen-3-ylmethyl)trimethylsilane **2p**.

3. General procedure for the preparation of the racemic products

An oven-dried reaction tube was charged with *N*-sulfonyl cyclic ketimines **1** (0.1 mmol), benzyl silanes **2** (0.1 mmol), $[Ir\{dFCF_3ppy\}_2(bpy)]PF_6$ (1 mol%) and CHCl₃ (1.0 mL). The reaction mixture was positioned in four 20 W blue LEDs lamp under N₂ atmosphere at room temperature After being stirred for the 0.5–24 h (monitored by TLC analysis), the reaction mixture was concentrated and then purified by flash chromatography on silica gel (eluted with PE/EtOAc/DCM = 8:1:1, v/v/v) to afford racemic product *rac*-**3**.

4. General procedure for the catalytic asymmetric reactions

An oven-dried reaction tube was charged with *N*-sulfonyl cyclic ketimines **1** (0.1 mmol), alkyl silanes **2** (0.1 mmol), $[Ir\{dFCF_3ppy\}_2(bpy)]PF_6$ (1 mol%), **L₃-RaPr₂Ad** (10 mol%), Ni(OTf)₂ (10 mol%), 4 Å MS (20 mg) and CHCl₃ (1.0 mL). The reaction mixture was positioned in four 20 W blue LEDs lamp under N₂ atmosphere at room temperature. After being stirred for the 0.5–72 h (monitored by TLC analysis), the reaction mixture was concentrated and then purified by flash chromatography on silica gel (eluted with PE/EtOAc/DCM = 8:1:1, v/v/v). *The reported isolated yields represented the total yield of the two diastereomers*.



Figure S1. Photochemical setup with blue LEDs

5. Optimization of the reaction conditions

0 N + CO ₂ Bn 1a	Photocatalyst (x mol%) L_3 -RaPr ₂ /Ni(OTf) ₂ (1:1, 10 mol%) 20 W blue LEDs, N ₂ , rt, CHCl ₃ , 6 h 2a	NH BnO ₂ C 3a	0 NH BnO ₂ C 3a' L ₃ -RaPr ₂	$R = 2,6-iPr_2C_6H_3$
Entry ^a	photocatalyst (x mol%)	yield/% ^b	ee/%	dr^c
1	xanthone (4 mol%)	0	/	/
2	fac-Ir(dFppy)3(1 mol%)	0	/	/
3	4CzIPN (1 mol%)	0	/	/
4	[Ir[d(Me)ppy]2(dtbbpy)]PF6 (1 mol%)	12	2/-3	54:46
5	[Ir[dF(CF ₃)ppy] ₂ (bpy)]PF ₆ (1 mol%)	69	63/15	66:34
6	[Ir[dF(Me)ppy]2(dtbbpy)]PF6 (1 mol%)	32	32/-6	55:45
7^d	$[Ir[dF(CF_3)ppy]_2(5,5'-dCF_3-bpy)]PF_6 (1 mol$	%) 30(38)	$62/11(29)^d$	64:36
8	Ru(bpz) ₃ PF ₆ (1 mol%)	0	/	/
9	Mes-Acr-Me-ClO ₄ (2 mol%)	0(95)	/(17)	/
10	Anthraquinone (2 mol%)	33	18/-8	55:45
11	Eosin Y (1 mol%)	0	/	/
12	Na ₄ W ₁₀ O ₂₄ (1 mol%)	0	/	/
13	Rhodamine B (1 mol%)	0	/	/

Table S1. Screening of the photocatalysts

^{*a*}All the reactions were performed with Ni(OTf)₂ (10 mol%), L_3 -RaPr₂ (10 mol%), 1a (0.10 mmol), 2a (0.10 mmol) and photocatalyst (x mol%) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis. ^{*d*}Data of desilylation product 3a' in parenthesis.

Table S2. Screening of the ligands



entry ^a	ligand	yield/% ^b	ee/% ^c	$\mathrm{d}\mathbf{r}^{c}$
1	L ₃ -RaPr ₂	69	62/13	66:34
2	L ₃ -PiPr ₂	73	91/44	73:27
3	L ₃ -PrPr ₂	51	57/31	60:40
4	L3-RaEt2	50	38/19	60:40
5	L3-RaMe2	32	7/-8	52:48
6	L ₃ -RaPr ₃	86	91/44	81:19

7	L ₃ -PiPr ₃	56	92/55	78:22
8	L ₃ -RaPr ₂ Ad	75	94/47	86:14
9	L ₃ -PiPr ₂ Ad	79	93/57	77:23
10	L ₃ -RaBn	52	26/-17	55:45
11	L ₃ -RaCy	64	-32/-39	47:53

^{*a*}All the reactions were performed with Ni(OTf)₂ (10 mol %), **ligand** (10 mol %), **1a** (0.10 mmol), **2a** (0.10 mmol) and $[Ir[dF(CF_3)ppy]_2(bpy)]PF_6$ (1 mol %) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis.

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Table S3. Screening of the metal salts

$O_{U}O_{V}O_{V}+O_{CO_2Bn}$	[Ir[dF(CF ₃ Ph TMS <u>L₃-RaPr₂Ad</u> N ₂ , rt, CHC 2a)ppy] ₂ (bpy)]PF ₆ (1 mol%) /metal salts (1:1, 10 mol%) Cl ₃ , 20 W blue LEDs, 6 h	NH BnO ₂ C Ph 3a	N H
entry ^a	metal salts	yield/% ^b	ee/%	$\mathrm{d}\mathbf{r}^{c}$
1	Ni(OTf)2	76	94/48	86:14
2	Mg(OTf) ₂	46	-3/-10	50:50
3	Sc(OTf) ₃	36	11/0	50:50
4	Y(OTf) ₃	41	-13/-13	49:51
5	Fe(OTf) ₂	37	7/-3	54:46
6	Zn(OTf) ₂	71	61/63	63:37
7	Co(OTf) ₂	58	70/6	69:31
8	Cu(OTf) ₂	49	2/-8	48:52
9	Ni(NTf ₂) ₂	73	90/37	82:18
10	Ni(acac) ₂	71	37/5	68:32
11	Ni(ClO ₄) ₂ ·6H ₂ O	39	-6/-10	51:49
12	Ni(BF4)2·6H2O	56	-37/-10	58:42
13	NiCl ₂	57	-8/-2	49:51

^{*a*}All the reactions were performed with metal salt (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (0.10 mmol), 2a (0.10 mmol) and $[Ir[dF(CF_3)ppy]_2(bpy)]PF_6(1 mol\%)$ in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis.

Table S4. Screening of the ratio between 1a and 2a

0 1 N + CO ₂ Bn 1a (x mmol)	Ph TMS [Ir[dF(CF ₃)p] <u>L₃-RaPr₂Ad/</u> N ₂ , rt, CHC 2a (y mmol)	py] ₂ (bpy)]PF ₆ (1 mol%) Ni(OTf) ₂ (1:1, 10 mol%) Pl ₃ , 20 W blue LEDs, 6 h ➤	NH BnO ₂ C Ph 3a	$N \cdot H$ R L_3 -RaPr ₂ Ad: R = 2,6- <i>i</i> Pr ₂ -4- adamantyl-C ₆ H ₂
entry ^a	x:y	yield/% ^b	ee/%	dr^c
1	0.12:0.10	94	94/48	86:14
2	0.10:0.10	76	94/50	86:14
3	0.10:0.12	80	95/52	86:14

^{*a*}All the reactions were performed with Ni(OTf)₂ (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (x mmol), 2a (y mmol) and $[Ir[dF(CF_3)ppy_2](bpy)]PF_6 (1 mol%)$ in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and d.r. values were determined by UPC² analysis.

Table S5.	Screening	of the	solvents and	concentration.
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$ \begin{array}{c} & 0 \\ & N \\ & N \\ & CO_2Bn \end{array} $ 1a 2a	[Ir[dF(CF ₃)ppy] ₂ (bpy)]PF ₆ (1 mol%) L₃-RaPr₂Ad /Ni(OTf) ₂ (1:1, 10 mol%) 20 W blue LEDs, N ₂ , rt, solvent, 6 h	O S NH BnO ₂ C Ph 3a	or NH BnO ₂ C Ph	$\begin{array}{c} & & & \\$
entry ^a	solvents	yield/% ^b	ee/%	$d\mathbf{r}^{c}$
1	CHCl ₃	94	94/48	86:14
2	THF	0(11)	n.a.	n.a.
3^d	Toluene	23(75)	92/35(47)	85:15
4	CH ₃ CN	0(99)	n.a.(36)	n.a.
5	CH ₃ OH	0(98)	n.a.(7)	n.a.
6	Et ₂ O	0	n.a.	n.a.
7^e	CHCl ₃	66	91/37	83.5:16.5
8 ^f	CHCl ₃	85	92/42	84:16

^{*a*}All the reactions were performed with Ni(OTf)₂ (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (0.12 mmol), 2a (0.10 mmol) and $[Ir[dF(CF_3)ppy]_2(bpy)]PF_6(1 mol%)$ in corresponding solvents (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis. ^{*d*}Data of desilylation product 3a' in parenthesis. ^{*e*}CHCl₃ (0.5 mL). ^{*f*}CHCl₃ (2.0 mL).

Table S6. Screening of the amount of chiral catalyst, reaction temperature and time

O S N CO ₂ Bn +	Ph TMS	[Ir[dF(CF ₃)ppy] ₂ (bpy)]PF ₆ (1 mol%) L ₃ -RaPr ₂ Ad/Ni(OTf) ₂ (1:1, x mol%) N ₂ , rt, CHCl ₃ , 20 W blue LEDs, 6 h	NH BnO ₂ C Ph	$\begin{array}{c} & & \\ & & \\ & & \\ & & \\ & \\ & \\ & \\ & $
1a	2a		3a -	adamantyl-C ₆ H ₂
entry ^a	Х	yield/% ^b	ee/%	dr^c
1	10	94	94/48	86:14
2	5	87	92/42	84:16
3	1	43	41/0	61:39
4^d	10	87	93/56	85:15
5^e	10	74	90/49	84:16
6 ^f	10	92	93/46	86:14

^{*a*}All the reactions were performed with Ni(OTf)₂ (x mol%), L₃-RaPr₂Ad (x mol%), 1a (0.12 mmol), 2a (0.10 mmol) and $[Ir[dF(CF_3)ppy]_2(bpy)]PF_6(1 mol%)$ in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 6 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis. ^{*d*}At 10 °C. ^{*e*}At -10 °C. ^{*f*}For 0.5 h.

Table S7. Screening of the additives

$ \begin{array}{c} $	[Ir[dF(C Ph TMS 4.3-RaPr N ₂ , rt, C 2a	F ₃)ppy] ₂ (bpy)]PF ₆ (1 mol%) 2Ad /Ni(OTf) ₂ (1:1, 10 mol%) HCl ₃ , 20 W blue LEDs, 0.5 h additive (1 equiv)	NH BnO ₂ C Ph 3a	N-H H $RL3-RaPr2Ad: R = 2,6-iPr2-4-adamantyl-C6H2$
entry ^a	additives	yield/% ^b	ee/%	$\mathrm{d}\mathbf{r}^{c}$
1	-	94	94/48	86:14
2	NaBr	83	91/35	83:17
3	LiBr	27	83/33	78:22
4	$MgSO_4$	70	87/27	80:20
5	Na_2SO_4	78	89/34	82:18
6	LiCl	79	93/45	85:15
7^d	3 Å MS	93	95/43	86:14
8^d	4 Å MS	95	95/40	86:14
9^d	5 Å MS	94	93/45	86:14
10	H ₂ O	67	84/0	76:24

"All the reactions were performed with Ni(OTf)2 (10 mol%), L3-RaPr2Ad (10 mol%), 1a (0.12 mmol), 2a (0.10 mmol) and [Ir[dF(CF3)ppy]2(bpy)]PF6 (1 mol%) in CHCl3 (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 0.5 h. ^bYield of isolated product. ^cThe ee and dr values were determined by UPC² analysis. ^d20 mg.

6. The limited substrate scope

BnO₂Ć





Ρh

N.R.

7. Control experiments

O N CO ₂ Br 1a	+ Ph ^{^_} TMS - 1 2a	[Ir{dFCF ₃ ppy}; L ₃ -RaPr ₂ Ad/Ni(N ₂ , rt, CHCI ₃ , 2 4 Å N	2(bpy)]PF ₆ (1 mol%) OTf)₂ (1:1, 10 mol%) 0 W blue LEDs, 0.5 h IS (20 mg)	O NH SnO ₂ C Ph 3a	N H
Entry ^a	variation of c	conditions	yield/% ^b	ee/% ^c	d.r. ^c
1	none	5	95	95	86:14
2	without	light	0	/	/
3	withou	ıt Ir	0	/	/
4	without L3-RaPr	2 Ad /Ni(OTf)2	48	/	52:48
5	without L3-l	RaPr ₂ Ad	28	/	47:53
6	without N	i(OTf)2	48	/	51:49
7	in ai	r	0	/	/

^{*a*}All the reactions were performed with Ni(OTf)₂ (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (0.12 mmol), 2a (0.10 mmol) and $[Ir{dFCF_3ppy}_2(bpy)]PF_6$ (1 mol%) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 0.5 h. ^{*b*}Yield of isolated product. ^{*c*}The ee and dr values were determined by UPC² analysis.

8. Mechanistic investigations

8.1 Catalytic reaction interfered with a radical quencher



The reaction was performed with Ni(OTf)₂ (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (0.12 mmol), 2a (0.10 mmol), [Ir{dFCF₃ppy}₂(bpy)]PF₆ (1 mol%), 4 Å MS (20 mg) and TEMPO (3 equiv) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 16 h. No desired product **3a** was observed.

8.2 Detection of the homocoupling product 4a



The reaction was performed with **1a** (0.10 mmol) and $[Ir\{dFCF_3ppy\}_2(bpy)]PF_6$ (50 mol%) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 24 h. Then the reaction mixture was concentrated and purified by flash chromatography on silica gel (eluted with PE:EtOAc = 4:1 to 2:1) to afford product **4a** as a white solid (15.8 mg, 26% yield).

8.3 EPR studies

The reactions were performed with Ni(OTf)₂ (10 mol%), L₃-RaPr₂Ad (10 mol%), 1a (0.12 mmol), 2a (0.10 mmol), [Ir[dF(CF₃)ppy]₂(bpy)]PF₆ (1 mol%), 4Å MS (20 mg) in CHCl₃ (1.0 mL) under the irradiation of 20 W blue LEDs. After being stirred at rt for 15 min, DMPO (5,5-dimethyl-1-pyrroline *N*-oxide, 11.3 mg, 0.10 mmol) was added. The reaction mixture was analyzed by electron paramagnetic resonance (EPR). The existence of radical **6** was further established by HRMS.



Figure S2. EPR spectra of the reaction mixture and 6



Figure S3. HRMS spectra of 6

8.4 Stern-Volmer emission quenching experiments

All [Ir{dFCF₃ppy}₂(bpy)]PF₆ solutions were excited at 370 nm and the emission intensity was collected at 420-600 nm. Stern-Volmer quenching experiments were carried out using a 0.1 mM solution of [Ir{dFCF₃ppy}₂(bpy)]PF₆ and variable concentrations (1, 2, 4, 6, 8 mM) of imine **1a** or benzyltrimethylsilane **2a**. The samples were prepared in 2 mL quartz cuvettes. The intensity of the emission peak at 468 nm ($\lambda_{ex} = 370$ nm) for [Ir{dFCF₃ppy}₂(bpy)]PF₆ expressed as the ratio I₀/I, where I₀ is the emission intensity of [Ir{dFCF₃ppy}₂(bpy)]PF₆ at 468 nm in the absence of a quencher and I is the observed intensity, as a function of the quencher concentration was measured.



Figure S4. Fluorescence quenching spectra of 1a



Figure S5. Fluorescence quenching spectra of 2a



Figure S6. Stern-Volmer quenching study of 1a and 2a

All [Ir{dFCF₃ppy}₂(bpy)]PF₆ solutions were excited at 370 nm and the emission intensity was collected at 420-600 nm. Stern-Volmer quenching experiments were carried out using a 0.1 mM solution of [Ir{dFCF₃ppy}₂(bpy)]PF₆ and variable concentrations (0.5, 1.0, 1.5, 2.0 mM) of imine **1a** or **1a**/Ni(OTf)₂/**L**₃-**RaPr₂Ad**. The samples were prepared in 2 mL quartz cuvettes. The intensity of the emission peak at 470 nm ($\lambda_{ex} = 370$ nm) for [Ir{dFCF₃ppy}₂(bpy)]PF₆ expressed as the ratio I₀/I, where I₀ is the emission intensity of [Ir{dFCF₃ppy}₂(bpy)]PF₆ at 470 nm in the absence of a quencher and I is the observed intensity, as a function of the quencher concentration was measured.



Figure S7. Fluorescence quenching spectra of 1a/Ni(OTf)₂/L-RaPr₂Ad



Figure S8. Stern-Volmer quenching study of 1a and 1a/Ni(OTf)2/L3-RaPr2Ad

8.5 Deuterium experiment



The reaction was performed according to the typical procedure by using deuterated benzyl trimethyl silicon D-2a. The deuterated ratio was determined by ¹H NMR spectrum after chromatography with silica gel column. The peak of H at 3.31 ppm is not observed, which indicates full deuteration of H.





The reaction was performed according to the typical procedure by using deuterated benzyl trimethyl silicon *D*-**2a**. The deuterated ratio was determined by ¹H NMR spectrum after filtration under nitrogen atmosphere. The peak integral at 5.78 is 0.76 and the deuteration rate is 0.24, suggesting the N-H comes from *D*-**2a**.



The deuterated ratio was determined by 1H NMR spectrum after filtration under nitrogen atmosphere. The peak integral at 5.80 is 1.00.



8.6 KIE experiment

Competitive KIE experiments



The reaction was performed according to the typical procedure by using a mixture of **2a** (0.10 mmol) and deuterated borane adduct D-**2a** (0.10 mmol). The deuterated ratio was determined by ¹H NMR spectrum after chromatography with silica gel column.



Parallel KIE experiments



The reaction progress of benzyltrimethylsilane **2a** and *D*-benzyltrimethylsilane **2a** with benzosultam **1a** was monitored in parallel at six distinct periods (10 min, 20 min, 30 min, 40 min, 50 min and 1 h). [Ir{dFCF₃ppy}₂(bpy)]PF₆ (1 mol%), Ni(OTf)₂/L₃-RaPr₂Ad (1:1, 10 mol%), **1a** (0.12 mmol), **2a** or *D*-**2a** (0.10 mmol), and 4 Å MS (20 mg) in CHCl₃ (1.0 mL) at rt under the irradiation of 20 W blue LEDs for 0–1 h. After the reaction was completed, the suspension was filtered through a sand core funnel, and the filtrate was concentrated under vacuum. The residue was detected by ¹H NMR analysis (CH₂Br₂ as the internal standard). The KIE of the reaction, which was calculated by dividing the rate constant of the reaction of benzyltrimethylsilane (Y1:k_H = 1.47) by that of *D*-benzyltrimethylsilane (Y2:k_D = 1.01), was determined to be 1.46.



Figure S9. Parallel KIE experiments

8.7 Cyclic voltammetry analysis

Cyclic voltammograms for mechanistic analysis were carried out with a μ Stat-i 400s potentiostat by a three-electrode cell with a glassy carbon working electrode, a platinum mesh counter electrode, and an Ag/AgNO₃ reference electrode at room temperature in CH₃CN. *n*-Bu₄NPF₆ (0.1 M) was used as the supporting electrolyte, and a Glass Carbon electrode was used as the working electrode. The scan rate was 100 mV s⁻¹. The experimental setup was calibrated using ferrocene (Fc/Fc+) prior to each experiment. Samples were prepared with 0.3 mmol substrate in 10 mL *n*-Bu₄NPF₆ electrolyte $(0.03 \text{ M in CH}_3\text{CN})$ and degassed by sparging with nitrogen gas for 10 min prior to use. The potential $(E_{p/2})$ was determined and converted to SCE as described by Nicewicz¹⁰.



Figure S10. Cyclic voltammogram of 1a and 2a

General Protocol to assess the photon flux of LED:

The photon flux set by the LED ($\lambda = 420$ nm, 20 W) was determined using the standard potassium ferric oxalate photometric method, following a modified literature procedure¹¹.

A 0.018 M potassium ferric oxalate solution was prepared by dissolving 178 mg of potassium ferric oxalate trihydrate and 84 μ L of H₂SO₄ (95–98%) in 20 mL of water. This solution was kept in an amber bottle in the dark. Then a buffer solution was prepared by dissolving 2.5 g of sodium acetate and 0.50 mL of H₂SO₄ (95–98%) in 50 mL of water.

1.0 mL of the 0.018 M potassium ferric oxalate solution was added to a flask containing a stirring bar. Then, the solution was irradiated for 0 s, 30 s, 60 s. Immediately after irradiation, 100 μ L of the solution was transferred to a foil-covered 10 mL volumetric flask in which containing 15 mg of 1,10-phenanthroline dissolved in 3.0 mL of the buffer solution. Then add water to the flask to bring the total volume to 10 ml. The flask was shaken to ensure efficient mixing and the solution was stored in the dark for approximately 20 min. 2.0 mL of the solution was transferred to a quartz cuvette (1.0 mL path length) and the absorbance at $\lambda = 510$ nm was measured by UV/Vis spectroscopy (**Figure S11**). This process was repeated for 30 s, 60 s, and the absorbance of a unirradiated sample was also measured.

UV/Vis spectra 5 4 absorbance 3 2 1 0 350 400 450 500 550 600 wavelength/nm - 30 s — — 60 s 0 s





The number of moles of Fe^{2+} produced by light irradiation was calculated using:

$$mol \ Fe^{2+} = \frac{V_1 V_3 \Delta A \ (510 \ nm)}{V_2 l\varepsilon \ (510 \ nm)}$$

Where

 V_1 = the volume of potassium ferric oxalate solution irradiated (1.0 × 10⁻³ L).

 V_2 = the volume of the solution taken for measurement of the Fe²⁺ ions (1.0 × 10⁻⁴ L).

 V_3 = the final volume of solution after complexation with 1,10-phenanthroline (1.0 × 10⁻² L).

 ΔA (510 nm) = the absorbance difference at $\lambda = 510$ nm between the irradiated solution and the solution kept in dark.

l = the optical path length of the cuvette (1.0 cm).

 ϵ (510 nm) = the molar absorption coefficient of the Fe(phen)₃²⁺ complex at λ = 510 nm (1.11 × 10⁴ L mol⁻¹ cm⁻¹).



The moles of Fe^{2+} were plotted as a function of time (Figure. S12)



The photon flux was then calculated using:

$$photon flux = \frac{mol Fe^{2+}}{\phi t(1-10^{-A})}$$

Where

 Φ = the quantum yield of the potassium ferric oxalate in room temperature at 420 nm is 1.13.¹² t = the irradiated time(s).

A = the potassium ferric oxalate absorbance at 420 nm, which was measured placing 1 mL of the solution in a cuvette which path length is 1 cm by UV/Vis spectrophotometry. We obtained an absorbance value of 1.706.

$$photon flux = \frac{2.0 \times 10^{-7}}{1.13 \times (1 - 10^{-1.706})} = 1.81 \times 10^{-7}$$

Quantum yield of radical reaction:



An oven-dried reaction tube was charged with *N*-sulfonyl cyclic ketimines **1a** (0.12 mmol), alkyl silanes **2a** (0.10 mmol), $[Ir\{dFCF_3ppy\}_2(bpy)]PF_6 (1 mol\%), L_3-RaPr_2Ad (10 mol\%), Ni(OTf)_2 (10 mol\%), 4 Å MS (20 mg) and CHCl_3 (1.0 mL). The mixture was stirred under 420 nm 20 W blue LED irradiation at room temperature. After 10 minutes, the yield of$ **3a**was determined by ¹HNMR, using

dibromaethan as an internal standard. And 2.5×10^{-5} mol **3a** was produced. The quantum yield (ϕ) of this reaction was calculated using the following equation:

$$\phi = \frac{mol \ product}{photon \ flux \ \times \ t \ \times (1 - 10^{-A})}$$

Where

 Φ = the quantum yield of the catalytic reaction in room temperature at 420 nm.

t = the reaction time(s).

A = the catalytic reaction absorbance at 420 nm, which was measured placing 1 mL of the solution in a cuvette which path length is 1 cm by UV/Vis spectrophotometry. We obtained an absorbance value of 1.954

$$\phi = \frac{2.5 \times 10^{-5}}{1.81 \times 10^{-7} \times 600 \times (1 - 10^{-1.954})} = 2.30 \times 10^{-1}$$

8.9 Catalytic cycle



SET/PT mechanism



MS-PCET mechanism

Figure S12. Three possible cycle of reaction



SET oxidation-triggered radical addition mechanism with $\ensuremath{\mathsf{Acr}}^{\mbox{\tiny +}}$ as photocatalyst

9. Product derivation



10. Single-crystal data

Crystallographic Data for	$C_{38}H_{46}N_2O_8S_2Si_2$
Formula	$2(C_{38}H_{46}N_2O_8S_2Si_2)$
Formula mass (amu)	779.07
Space group	P 21 21 21
<i>a</i> (Å)	11.6468(8)
<i>b</i> (Å)	16.5589(12)
<i>c</i> (Å)	21.0550(17)
α (deg)	90
β (deg)	90
γ (deg)	90
$V(Å^3)$	4060.6(5)
Ζ	4
λ (Å)	0.71073
<i>T</i> (K)	173 K
$ ho_{ m calcd}$ (g cm ⁻³)	1.274
$\mu (\mathrm{mm}^{-1})$	0.241
Transmission factors	0.878,0.981
$\theta_{\max}(\deg)$	25.368
No. of unique data, including $F_0^2 < 0$	7419
No. of unique data, with $F_o^2 > 2\sigma(F_o^2)$	7187
No. of variables	485
$R(F)$ for $F_{o}^{2} > 2\sigma(F_{o}^{2})^{a}$	0.0328
$R_{\rm w}(F_{\rm o}^2)^{b}$	0.0759
Goodness of fit	1.103

^{*a*} $R(F) = \sum ||F_{o}| - |F_{c}|| / \sum |F_{o}|.$

^b $R_w(F_o^2) = \left[\sum [w(F_o^2 - F_c^2)^2] / \sum wF_o^4\right]^{1/2}; w^{-1} = [\sigma^2(F_o^2) + (Ap)^2 + Bp], \text{ where } p = \left[\max(F_o^2, 0) + 2F_c^2\right] / 3.$



11. References

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12. Characterization of typical substrates 1a-1h

benzyl benzo[d]isothiazole-3-carboxylate 1,1-dioxide(1a)



HRMS (ESI⁺) m/z calcd for $C_{15}H_{11}NNaO_4S^+$ ([M+Na⁺]) = 324.0301, Found 324.0299.

IR (film, cm⁻¹): 3035, 1736, 1348, 1202, 1178, 1028, 763, 700, 579, 532.

benzyl 5-methylbenzo[d]isothiazole-3-carboxylate 1,1-dioxide(1b)



White solid, m.p. 106 – 108 °C; ¹**H NMR** (400 MHz, CDCl₃) δ 8.04 (s, 1H), 7.80 (d, *J* = 8.0 Hz, 1H), 7.56 (d, *J* = 8.0 Hz, 1H), 7.50-7.47 (m, 2H), 7.44 – 7.37 (m, 3H), 5.49 (s, 2H), 2.51 (s, 3H).; ¹³**C NMR** (101 MHz, CDCl₃) δ 160.4, 160.1, 145.7, 137.4, 134.6, 133.9, 129.1, 128.8, 128.7, 127.9, 122.8, 69.1, 21.8.

HRMS (ESI⁺) m/z calcd for $C_{16}H_{13}NNaO_4S^+$ ([M+Na⁺]) = 338.0458, Found 338.0457.

IR (film, cm⁻¹): 3270, 1743, 1310, 1238, 1186, 1147, 1102, 896, 822, 736, 699, 572.

benzyl 6-methylbenzo[d]isothiazole-3-carboxylate 1,1-dioxide(1c)



White solid, m.p. 107 – 108 °C; ¹**H NMR** (400 MHz, CDCl₃) δ 8.04 (d, *J* = 8.0 Hz, 1H), 7.66 (s, 1H), 7.46 – 7.40 (m, 3H), 7.36-7.29 (m, 3H), 5.41 (s, 2H), 2.46 (s, 3H).; ¹³C NMR (101 MHz, CDCl₃) δ 160.4, 160.0, 146.3, 140.7, 134.9, 133.9, 129.1, 128.8, 127.23, 125.9, 123.7, 69.1, 21.8.

HRMS (ESI⁺) m/z calcd for $C_{16}H_{13}NNaO_4S^+$ ([M+Na⁺]) = 338.0458, Found 338.0454.

IR (film, cm⁻¹): 2963, 1735, 1345, 1206, 1170, 1022, 837, 804, 698, 572.

benzyl 7-chlorobenzo[d]isothiazole-3-carboxylate 1,1-dioxide(1d)



White solid, m.p. $129 - 130 \,^{\circ}\text{C}$; ¹**H NMR** (400 MHz, CDCl₃) δ 8.14 - 8.09 (m, 1H), 7.63 - 7.59 (m, 2H), 7.42 - 7.40 (m, 2H), 7.38 - 7.33 (m, 3H), 5.42 (s, 2H).; ¹³**C NMR** (101 MHz, CDCl₃) δ 160.0, 158.6, 137.6, 135.78, 134.8, 133.6, 131.2, 130.3, 129.2, 128.9, 125.7, 69.3. **HRMS** (ESI⁺) m/z calcd for C₁₅H₁₀³⁵ClNNaO₄S⁺ ([M+Na⁺]) = 357.9912, Found 357.9911.

HRMS (ESI⁺) m/z calcd for $C_{15}H_{10}^{37}ClNNaO_4S^+$ ([M+Na⁺]) = 359.9882, Found 359.9884.

IR (film, cm⁻¹): 2963, 1736, 1356, 1179, 1060, 794, 699, 580.

benzyl 7-(trifluoromethyl)benzo[*d*]isothiazole-3-carboxylate 1,1-dioxide(1e)



White solid, m.p. 101 – 102 °C; ¹H NMR (600 MHz, CDCl₃) δ 8.55 (d, *J* = 6.0 Hz, 1H), 7.99 (d, *J* = 12.0 Hz, 1H), 7.92 (t, *J* = 6.0 Hz, 1H), 7.49 (d, *J* = 6.4 Hz, 2H), 7.50 – 7.48(m, 3H), 5.51 (s, 2H).; ¹³C NMR (151 MHz, CDCl₃) δ 160.0, 158.4, 137.9, 135.3, 133.6, 131.1 (q, *J* = 3.7 Hz), 130.8, 129.5, 129.3, 128.9, 128.9, 127.8 (q, *J* = 36.9 Hz), 122.8, 121.0, 69.5; ¹⁹F NMR (565 MHz,

CDCl₃) δ -60.0. (s, 3F).

HRMS (ESI⁺) m/z calcd for $C_{16}H_{10}F_3NNaO_4S^+$ ([M+Na⁺]) = 392.0175, Found 392.0180.

IR (film, cm⁻¹): 3093, 1739, 1365, 1321, 1180, 1140, 825, 806, 702, 574.

benzyl 7-fluorobenzo[d]isothiazole-3-carboxylate 1,1-dioxide(1f)



White solid, m.p. 123 – 124 °C; ¹**H NMR** (400 MHz, CDCl₃) δ 8.08 (d, *J* = 8.0 Hz, 1H), 7.78 – 7.73 (m, 1H), 7.48 – 7.40 (m, 6H), 5.49 (s, 2H).; ¹³C NMR (101 MHz, CDCl₃) δ 159.9, 158.8, 157.2 (d, *J* = 262.2 Hz), 137.5 (d, *J* = 6.9 Hz), 133.7, 130.9, 129.2, 128.9, 128.9, 126.3 (d, *J* = 21.0 Hz), 123.5 (d, *J* = 3.4 Hz), 122.2 (d, *J* = 19.6 Hz), 69.4; ¹⁹F NMR (377 MHz, CDCl₃) δ -

111.0. (s, 1F)

HRMS (ESI⁺) m/z calcd for $C_{15}H_{10}FNNaO_4S^+$ ([M+Na⁺]) = 342.0207, Found 342.0207.

IR (film, cm⁻¹): 3087, 1737, 1498, 1358, 1261, 1222, 1180, 1108, 841, 755, 700, 522.

benzyl naphtho[2,3-d]isothiazole-3-carboxylate 1,1-dioxide(1g)



White solid, m.p. 207 – 208 °C; ¹H NMR (400 MHz, CDCl₃) δ 8.78 (s, 1H), 8.39 (s, 1H), 8.05 (t, *J* = 8.0 Hz, 2H), 7.82 – 7.74 (m, 2H), 7.54 – 7.51 (m, 2H), 7.45 – 7.40 (m, 3H), 5.54 (s, 2H).; ¹³C NMR (151 MHz, CDCl₃) δ 160.5, 160.0, 134.8, 134.6, 134.1, 133.9, 131.2, 131.0, 123.0, 129.9, 129.8, 129.1, 128.9, 125.0, 124.5, 69.1.

HRMS (FTMS+*c* ESI, m/z) calcd for $C_{19}H_{13}NNaO_4S^+$ ([M+Na⁺]) = 374.0458, Found 374.0455.

IR (neat): (cm⁻¹) 3066, 1740, 1340, 1171, 754, 602, 471.

12. Characterization of the products

benzyl 3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1dioxide(3a)



7.00 (t, J = 6.0 Hz, 2H), 6.91 (t, J = 6.0 Hz, 1H), 5.83 (s, 1H), 5.22 (dd, $J_1 = 12.0$, $J_2 = 30.0$ Hz, 2H), 3.31 (s, 1H), 0.00 (s, 9H).

¹³C NMR (151 MHz, CDCl₃) δ 170.9, 166.6, 164.0, 140.1, 140.0, 136.8, 133.9, 130.7, 129.3, 129.0, 128.9, 128.2, 126.3, 123.4, 123.3, 118.3, 118.1, 113.4, 113.2, 72.0, 70.0, 44.7, 0.9.

IR (film, cm⁻¹): 3280, 2956, 1734, 1307, 1249, 1220, 1169, 845, 754, 703;

4

HRMS (ESI⁺) m/z calcd for $C_{25}H_{27}NNaO_4SSi^+$ ([M]+Na⁺) = 488.1323, found 488.1324.



15.623

207921

10.02

methyl 3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1dioxide(3b)



3.82 (s, 3H), 3.27 (s, 1H), 0.00 (s, 9H).

¹³**C NMR** (101 MHz, CDCl₃) *δ* 172.0, 137.0, 136.6, 134.2, 133.0, 129.9, 127.9, 126.1, 125.9, 120.9, 72.2, 54.3, 44.5, -1.1.

IR (film, cm⁻¹) : 3276, 2956, 1736, 1306, 1246, 1206, 1169, 845, 756, 707;

HRMS (ESI⁺) m/z calcd for $C_{19}H_{23}NNaO_4SSi^+$ ([M]+Na⁺) = 412.1010, found 412.1010.



ethyl 3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-carboxylate 1,1dioxide(3c)



(t, J = 8.0 Hz, 1H), 5.79 (s, 1H), 4.31 – 4.18 (m, 2H), 3.27 (s, 1H), 3.06 (s, 0H), 1.29 (t, J = 8.0 Hz, 3H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 171.4, 137.1, 136.9, 134.3, 132.9, 129.8, 127.8, 126.1, 125.9, 120.9, 72.2, 63.8, 44.5, 13.9, -1.0.

IR (film, cm⁻¹): 3275, 2959, 1730, 1306, 1240, 1206, 1169, 844, 756, 707;

HRMS (ESI⁺) m/z calcd for $C_{20}H_{25}NNaO_4SSi^+$ ([M]+Na⁺) = 426.1166, found 426.1162.



0.50 1.00 1.50 2.00 2.50 3.00 3.50 4.00 4.50 5.00 Minutes 5.50 6.00 6.50 7.00 7.50 8.00 8.50 9.00 9.50

	Retention Time	Area	% Area
1	5.167	471587	20.49
2	5.747	476634	20.71
3	6.746	667927	29.02
4	7.432	685445	29.78



	Retention Time	Area	% Area
1	5.051	987860	3.96
2	5.412	19367332	77.67
3	6.570	1818022	7.29
4	7.228	2761707	11.08

benzyl 5-methyl-3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3d)



Hz, 1H), 5.79 (s, 1H), 5.23 (dd, *J*₁ = 12.0 Hz, *J*₂ = 16.0 Hz, 2H), 3.29 (s, 1H), 2.23 (s, 3H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) *δ* 171.4, 143.8, 137.1, 136.9, 134.0, 131.7, 130.8, 128.9, 128.7, 128.6, 127.8, 126.3, 125.8, 120.6, 72.0, 69.3, 44.3, 21.6, -1.1.

IR (film, cm⁻¹): 3282, 2956, 1734, 1306, 1240, 1222, 1186, 1143, 843, 753, 701;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 502.1479, found 502.1479.



benzyl 6-methyl-3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3e)



5.80 (s, 1H), 5.21 (dd, *J*₁ = 12.0 Hz, *J*₂ = 20.0 Hz, 2H), 3.29 (s, 1H), 2.27 (s, 3H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 171.5, 140.6, 137.2, 134.4, 134.1, 133.8, 128.9, 128.8, 128.6, 127.9, 125.9, 125.8, 120.8, 72.1, 69.4, 44.2, 21.1, -1.0.

IR (film, cm⁻¹): 3279, 2955, 1735, 1307, 1250, 1223, 1161, 1143, 844, 750, 703;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 502.1479, found 502.1476.



	Retention Time	Area	% Area
1	19.530	2317628	18.51
2	20.541	2324535	18.57
3	23.049	3768440	30.10
4	24.913	4109191	32.82



	Retention Time	Area	% Area
1	19.400	36810085	69.48
2	20.543	3206524	6.05
3	23.066	4591198	8.67
4	24.899	8369467	15.80

methyl (S)-5-fluoro-3-((S)-phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3carboxylate 1,1-dioxide(3f)



3H), 3.17 (s, 1H), -0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 171.4, 165.3 (d, *J* = 254.0 Hz), 140.0 (d, *J* = 10.0 Hz), 136.6, 130.4 (d, *J* = 2.0 Hz), 128.1, 126.2, 123.2 (d, *J* = 10.0 Hz), 118.1 (d, *J* = 24.0 Hz), 113.2 (d, *J* = 24.0 Hz), 71.7, 54.5, 44.8, -1.1. ¹⁹F NMR (377 MHz, CDCl₃) δ -103.8. (s, 1F).

IR (film, cm⁻¹): 3279, 2954, 1739, 1302, 1245, 1203, 1173, 851, 754, 710;

HRMS (ESI⁺) m/z calcd for $C_{19}H_{22}FNNaO_4SSi^+$ ([M]+Na⁺) = 430.0916, found 430.0918.



	Retention Time	Area	% Area
1	17.563	4477207	91.52
2	23.139	272720	5.57
3	52.822	52248	1.07
4	57.371	89841	1.84

benzyl 7-fluoro-3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1-dioxide(3g)



Colorless oil, 37.1 mg, 77% yield, 96% ee, 81:19 dr, $[\alpha]_D^{20} = -50.0 \ (c = 0.55, CH_2Cl_2)$; **HPLC** Daicel chiralcel ADH, *n*-hexane/*i*-PrOH 95/5, 1.0 mL/min, $\lambda = 220$ nm, $t_1 = 19.14$ min, $t_2 = 21.54$ min, $t_3 = 23.36$ min, $t_4 = 33.57$ min; ¹**H NMR** (400 MHz, CDCl₃) δ 7.39 – 7.37 (m, 3H), 7.35 – 7.32 (m, 3H), 7.21 (d, J = 8.0 Hz,

2H), 7.09 (d, *J* = 8.0 Hz, 1H), 7.04 (t, *J* = 8.0 Hz, 2H), 6.97 – 6.92 (m, 2H), 5.92 (s, 1H), 5.24

 $(dd, J_1 = 12.0 Hz, J_2 = 24.0 Hz, 2H), 3.29 (s, 1H), 0.00 (s, 9H).$

¹³C NMR (101 MHz, CDCl₃) δ 170.9, 155.8 (d, J = 259.0 Hz), 139.7, 136.7, 135.1 (d, J = 7.2 Hz), 133.8, 129.1, 128.8, 128.7, 128.0, 126.1, 122.7 (d, J = 19.0 Hz), 121.9 (d, J = 3.9 Hz), 116.5 (d, J = 18.3 Hz), 72.5, 69.8, 44.5, -1.1. ¹⁹F NMR (377 MHz, CDCl₃) δ -115.3. (s, 1F)

IR (film, cm⁻¹): 3272, 2957, 1736, 1317, 1253, 1219, 1176, 1143, 844, 755, 701;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}NNaO_4SSi^+$ ([M]+Na⁺) = 506.1229, found 506.1227.





	Retention Time	Area	% Area
1	18.919	2578566	22.95
2	21.136	3022449	26.91
3	22.996	2531615	22.54
4	33.075	3100860	27.60



	Retention Time	Area	% Area
1	19.143	9906889	77.16
2	21.541	470026	3.66
3	23.364	203815	1.59
4	33.569	2258964	17.59

benzyl 7-chloro-3-(phenyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3h)



2H), 6.95 (t, *J* = 8.0 Hz, 1H), 5.92 (s, 1H), 5.22 (dd, *J*₁ = 12.0 Hz, *J*₂ = 24.0 Hz, 2H), 3.29 (s, 1H), 0.00 (s, 9H). ¹³C NMR (101 MHz, CDCl₃) δ 171.0, 139.0, 136.7, 133.8, 133.8, 132.4, 130.6, 129.1, 128.8, 128.7, 128.0, 126.1, 124.5, 71.4, 69.8, 44.2, -1.1.

IR (film, cm⁻¹): 3273, 2957, 1735, 1317, 1250, 1218, 1170, 1143, 847, 754, 701;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{35}$ ClNNaO₄SSi⁺ ([M]+Na⁺) = 522.0933, found 522.0930.

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{37}$ ClNNaO₄SSi⁺ ([M]+Na⁺) = 524.0904, found 524.0904.



	Retention Time	Area	% Area
1	18.311	20320916	73.25
2	20.626	5932726	21.38
3	23.419	443677	1.60
4	26.539	1045813	3.77
benzyl 3-(phenyl(trimethylsilyl)methyl)-7-(trifluoromethyl)-2,3-dihydrobenzo[*d*]isothiazole-3carboxylate 1,1-dioxide(3i)



1H), 7.02 (t, *J* = 8.0 Hz, 2H), 6.93 (t, *J* = 8.0 Hz, 1H), 5.96 (s, 1H), 5.23 (dd, *J*₁ = 12.0 Hz, *J*₂ = 16.0 Hz, 2H), 3.32 (s, 1H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 170.9, 138.9, 136.7, 133.7, 132.9, 130.2, 129.1, 128.8, 128.7, 128.1, 127.8 (d, J = 4.8 Hz), 126.1, 71.7, 69.9, 44.6, -1.1.

¹⁹**F NMR** (377 MHz, CDCl₃) δ -59.6. (s, 3F)

IR (film, cm⁻¹): 3269, 2958, 1736, 1322, 1251, 1219, 1175, 1142, 842, 754, 703;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{26}F_3NNaO_4SSi^+$ ([M]+Na⁺) = 556.1197, found 556.1197.



benzyl 3-(phenyl(trimethylsilyl)methyl)-2,3-dihydronaphtho[2,3-*d*]isothiazole-3-carboxylate 1,1-dioxide(3j)



(d, *J* = 8.0 Hz, 1H), 6.91 (t, *J* = 8.0 Hz, 2H), 6.79 (t, *J* = 8.0 Hz, 1H), 5.83 (s, 1H), 5.19 (dd, *J*₁ = 12.0 Hz, *J*₂ = 44.0 Hz, 2H), 3.41 (s, 1H), 0.00 (s, 9H).

¹³**C NMR** (101 MHz, CDCl₃) *δ* 171.6, 137.2, 135.0, 134.0, 132.8, 132.3, 132.2, 129.1, 128.9, 128.8, 128.7, 128.6, 128.3, 127.9, 127.7, 126.0, 125.8, 121.7, 72.3, 69.5, 45.0, -0.9.

IR (film, cm⁻¹): 3278, 2957, 1735, 1308, 1249, 1219, 1170, 1152, 844, 752, 701;

HRMS (ESI⁺) m/z calcd for $C_{29}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 538.1479, found 538.1480.



	Retention Time	Area	% Area
1	13.798	1009260	21.70
2	16.102	1002391	21.55
3	17.362	1328568	28.56
4	18.928	1311009	28.19



	Retention Time	Area	% Area
1	13.703	6485840	72.01
2	16.122	648127	7.20
3	17.394	687998	7.64
4	18.848	1184836	13.15

benzyl 3-(m-tolyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1dioxide (3k)



5.82 (s, 1H), 5.21 (dd, *J*₁ = 12.0, *J*₂ = 20.0 Hz, 2H), 3.25 (s, 1H), 2.13 (s, 3H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 171.4, 136.8, 136.6, 134.4, 134.0, 132.8, 129.8, 128.9, 128.8, 128.6, 127.6, 126.6, 126.2, 120.9, 72.3, 69.5, 44.2, 21.2, -1.0.

IR (film, cm⁻¹) :3280, 2955, 1734, 1307, 1249, 1219, 1170, 842, 754, 701;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 502.1479, found 502.1478.



	Retention Time	Area	% Area
1	8.015	67402	2.07
2	8.666	2709466	83.16
3	11.368	90023	2.76
4	12.861	391063	12.00

benzyl 3-(p-tolyl(trimethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1dioxide (31)



¹³**C NMR** (101 MHz, CDCl₃) *δ* 171.4, 136.7, 135.3, 134.3, 133.7, 132.9, 129.8, 128.9, 128.8, 128.6, 126.1, 121.0, 72.4, 69.5, 43.8, 20.8, -1.0.

IR (film, cm⁻¹) :3282, 2955, 1735, 1308, 1250, 1220, 1169, 848, 754, 698;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 502.1479, found 502.1480.



1	14.020	31793044	79.15
2	15.467	1167989	2.91
3	16.588	2402958	5.98
4	18.193	4803895	11.96

benzyl 3-((4-methoxyphenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3m)



¹³**C NMR** (101 MHz, CDCl₃) *δ* 171.4, 157.6, 136.7, 134.4, 134.0, 132.9, 129.9, 128.9, 128.8, 128.6, 126.2, 121.0, 113.2, 72.5, 69.5, 54.9, 43.4, -1.0.

IR (film, cm⁻¹):3279, 2956, 1735, 1306, 1248, 1221, 1173, 848, 755, 698;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{29}NNaO_5SSi^+$ ([M]+Na⁺) = 518.1428, found 518.1430.



	Retention Time	Area	% Area
1	10.637	68254	5.88
2	14.576	949378	81.81
3	17.585	34601	2.98
4	24.376	108267	9.33

benzyl 3-((4-fluorophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3n)



¹³C NMR (101 MHz, CDCl₃) δ 171.1, 161.1 (d, J = 244.5 Hz), 139.7, 139.7, 136.3, 134.4, 133.8, 133.0, 132.8 (d, J = 3.3 Hz), 130.1, 129.0, 128.8, 128.7, 126.0, 121.1, 115.2 (d, J = 21.3 Hz), 72.0, 69.7, 44.3, -1.1.

¹⁹**F NMR** (565 MHz, CDCl₃) *δ* -116.8. (s, 1F)

IR (film, cm⁻¹) :3279, 2957, 1735, 1307, 1222, 1220, 1166, 849, 755, 698;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}FNNaO_4SSi^+$ ([M]+Na⁺) = 506.1229, found 506.1230.



benzyl 3-((4-chlorophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3o)



¹³**C NMR** (101 MHz, CDCl₃) *δ* 171.1, 136.3, 135.7, 134.3, 133.8, 133.1, 131.8, 130.1, 129.0, 128.8, 128.7, 128.1, 125.9, 121.2, 72.0, 69.7, 43.9, -1.1.

IR (film, cm⁻¹) :3278, 2956, 1736, 1308, 1251, 1220, 1169, 848, 755, 698;

4

20.635

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}{}^{35}CINNaO_4SSi^+$ ([M]+Na⁺) = 522.0933, found 522.0931.

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{37}CINNaO_4SSi^+$ ([M]+Na⁺) = 524.0904, found 524.0900



986582

12.64

benzyl 3-((4-bromophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3p)



¹³C NMR (101 MHz, CDCl₃) δ 172.2, 137.4, 137.4, 135.5, 135.0, 134.3, 132.1, 131.3, 130.2, 123.0, 129.8, 127.1, 122.3, 121.1, 73.1, 70.8, 45.0, 0.0.

IR (film, cm⁻¹) :3276, 2956, 1735, 1307, 1250, 1218, 1169, 847, 755, 698;

3

4

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{79}BrNNaO_4SSi^+$ ([M]+Na⁺) = 566.0428, found 566.0428.

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{81}BrNNaO_4SSi^+$ ([M]+Na⁺) = 568.0407, found 568.0399.

6.024

8.529



1432255

1434678

28.21

28.26



	Retention Time	Area	% Area
1	4.525	4655471	76.68
2	5.245	104778	1.73
3	6.034	437560	7.21
4	8.541	873695	14.39

benzyl 3-((2-fluorophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3q)



(s, 1H), 5.23 (dd, *J*₁ = 12.0, *J*₂ = 20.0 Hz, 2H), 3.89 (s, 1H), 0.00 (s, 9H).

¹³C NMR (101 MHz, CDCl₃) δ 171.1, 139.7 (d, J = 7.4 Hz), 135.3 (d, J = 191.2 Hz), 133.8, 133.1, 130.1, 129.0, 128.8, 128.7, 125.9, 121.1, 112.9 (d, J = 20.6 Hz), 72.0, 69.7, 44.3, -1.1.

¹⁹**F NMR** (377 MHz, CDCl₃) *δ* -117.3. (s, 1F)

IR (film, cm⁻¹):3281, 2958, 1736, 1308, 1262, 1222, 1166, 844, 734, 700;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}FNNaO_4SSi^+$ ([M]+Na⁺) = 506.1229, found 506.1229.



	Retention Time	Area	% Area
1	5.861	369595	26.03
2	7.152	369302	26.01
3	9.145	346008	24.37
4	11.268	334968	23.59



	Retention Time	Area	% Area
1	5.810	73028	5.51
2	7.104	963238	72.63
3	9.084	115520	8.71
4	11.212	174458	13.15

benzyl 3-((3-fluorophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3r)



(s, 1H), 5.21 (dd, $J_1 = 12.0$, $J_2 = 24.0$ Hz, 2H), 3.29 (s, 1H), 0.00 (s, 9H);

¹³**C NMR** (101 MHz, CDCl₃) *δ* 171.0, 137.9 (d, *J* = 336.8 Hz), 133.8, 133.1, 130.1, 129.1, 128.8, 128.7, 125.9, 121.2, 72.0, 69.7, 44.3, -1.10;

¹⁹**F NMR** (565 MHz, CDCl₃) *δ* -112.2. (s, 1F)

IR (film, cm⁻¹) :3278, 2957, 1735, 1308, 1250, 1220, 1170, 823, 755, 701;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}FNNaO_4SSi^+$ ([M]+Na⁺) = 506.1229, found 506.1229.



	Retention Time	Area	% Area
1	3.463	1311938	28.33
2	4.034	1267104	27.36
3	4.784	1047830	22.63
4	6.845	1004107	21.68



	Retention Time	Area	% Area
1	3.502	441524	73.31
2	4.115	25736	4.27
3	4.869	37461	6.22
4	6.850	97511	16.19

benzyl 3-((3-bromophenyl)(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3s)



 $(dd, J_1 = 12.0, J_2 = 24.0 \text{ Hz}, 2\text{H}), 3.24 (s, 1\text{H}), 0.00 (s, 9\text{H}).$

¹³C NMR (101 MHz, CDCl₃) δ 171.0, 139.6, 136.2, 133.8, 133.1, 130.1, 129.1, 128.8, 128.7, 125.9, 121.2, 72.0,
69.7, 44.3, -1.1.

IR (film, cm⁻¹) :3276, 2956, 1735, 1308, 1251, 1221, 1169, 845, 755, 696;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{79}BrNNaO_4SSi^+$ ([M]+Na⁺) = 566.0428, found 566.0428.

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{81}BrNNaO_4SSi^+$ ([M]+Na⁺) = 568.0407, found 568.0399.



	Retention Time	Area	% Area
1	30.789	23631657	16.37
2	35.093	9315881	6.45
3	37.149	99340194	68.80
4	43.117	12093581	8.38

benzyl 3-(naphthalen-2-yl(trimethylsilyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3t)



¹³C NMR (101 MHz, CDCl₃) *δ* 171.4, 136.6, 134.9, 134.4, 134.0, 132.9, 131.7, 129.9, 129.0, 128.8, 128.6, 125.6, 125.1, 121.0, 72.4, 69.6, 44.4, -1.0.

IR (film, cm⁻¹) :3279, 2956, 1735, 1308, 1250, 1221, 1169, 843, 752, 698;

4

48.084

HRMS (ESI⁺) m/z calcd for $C_{29}H_{29}NNaO_4SSi^+$ ([M]+Na⁺) = 538.1479, found 538.1479.



142976

2.72

benzyl 3-(naphthalen-2-ylmethyl)-2,3-dihydrobenzo[d]isothiazole-3-carboxylate 1,1dioxide(3t')



Colorless oil, 18.3 mg, 41% yield, 47% ee, $[\alpha]_D^{20} = -9.2$ (c = 0.34, CH₂Cl₂); **SFC** Chiralcel IB-3, CO₂/MeOH = 90/10, 1.5 mL/min, $\lambda = 220$ nm, $t_1 = 10.85$ min, $t_2 = 14.19$ min;

¹H NMR (600 MHz, CDCl₃) δ 7.90 (d, J = 6.0Hz, 1H), 7.81 (d, J = 12.0 Hz, 2H),
7.77 - 7.68 (m, 4H), 7.64 - 7.61 (m, 1H), 7.49 - 7.46 (m, 2H), 7.39 - 7.36 (m, 2H), 7.32 (t, J = 6.0 Hz, 2H), 7.23 (d, J = 6.0 Hz, 2H), 5.62 (s, 1H), 5.15 (q, J = 12.0 Hz, 2H), 3.72 (d, J = 12.0 Hz, 1H), 3.34 (d, J = 12.0 Hz, 1H).
¹³C NMR (151 MHz, CDCl₃) δ 168.9, 138.0, 135.5, 133.9, 133.5, 133.2, 132.7, 131.8, 130.7, 129.4, 129.0, 128.8, 128.8, 128.28, 128.0, 127.8, 127.6, 126.1, 126.0, 125.1, 121.6, 69.8, 69.2, 46.3.
IR (film, cm⁻¹) : 3288, 2928, 1738, 1304, 1168, 1222, 1186, 898, 754, 698;

HRMS (ESI⁺) m/z calcd for $C_{26}H_{21}NNaO_4S^+$ ([M]+Na⁺) = 466.1084, found 466.1077.





1 10.852 29962516 73.37	
2 14.190 10873810 26.63	

benzyl 3-(phenyl(triethylsilyl)methyl)-2,3-dihydrobenzo[*d*]isothiazole-3-carboxylate 1,1dioxide(3u)



1H), 5.23 (dd, J₁ = 12.0, J₂ = 42.0 Hz, 2H), 3.45 (s, 1H), 0.86 (t, J = 12.0 Hz, 9H), 0.56 - 0.49 (m, 6H).
¹³C NMR (150 MHz, CDCl₃) δ 171.4, 137.2, 136.7, 134.4, 133.9, 132.8, 129.8, 129.0, 128.8, 128.7, 126.3, 125.8, 120.9, 72.3, 69.6, 41.6, 7.5, 3.3.

IR (film, cm⁻¹) :3283, 3030, 2954, 2877, 1735, 1309, 1219, 1170, 809, 754, 705;

4

HRMS (ESI⁺) m/z calcd for $C_{28}H_{33}NNaO_4SSi^+$ ([M]+Na⁺) = 530.1792, found 530.1792.



752424

5.22

32.584

benzyl 3-((tert-butyldimethylsilyl)(phenyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-

carboxylate 1,1-dioxide(3v)



3.59 (s, 1H), 0.63 (s, 9H), 0.22 (s, 3H), 0.00 (s, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 171.5, 137.4, 136.5, 134.4, 133.9, 132.7, 129.8, 129.0, 128.8, 128.7, 126.2, 126.1, 121.0, 73.3, 69.6, 40.4, 26.7, 17.7, -4.6, -4.8.

IR (film, cm⁻¹) :3279, 3030, 2957, 2931, 2857, 1736, 1309, 1255, 1220, 1169, 828, 756, 705;

HRMS (ESI⁺) m/z calcd for $C_{28}H_{33}NNaO_4SSi^+$ ([M]+Na⁺) = 530.1792, found 530.1792.



	Retention Time	Area	% Area
1	23.694	100544	1.77
2	26.002	5100039	89.99
3	30.040	348371	6.15
4	35.474	118129	2.08

(S)-3-((S)-((chloromethyl)dimethylsilyl)(phenyl)methyl)-2,3-

dihydrobenzo[d]isothiazole-3-carboxylate 1,1-dioxide(3w)



Colorless oil, 33.6 mg, 67% yield, 88% ee, 81:19 dr, $[\alpha]_D^{20} = -36.2$ (c = 0.39, CH₂Cl₂); **SFC** Chiralcel IG-3, CO₂/MeOH = 93/7, 1.5 mL/min, $\lambda = 220$ nm, $t_1 = 14.74$ min, $t_2 = 17.19$ min, $t_3 = 18.85$ min, $t_4 = 21.60$ min;

¹**H NMR** (400 MHz, CDCl3) δ 7.41 (dd, $J_1 = 4.0, J_2 = 8.0$ Hz, 1H), 7.27 – 7.17 (m, 9H), 7.11 – 7.08 (m, 1H), 6.90 – 6.87 (m, 2H), 6.81 (t, J = 4.0 Hz, 1H), 5.83 (s, 1H), 5.11 (s,

2H), 3.46 (s, 1H), 2.47 (dd, $J_1 = 12.0$, $J_2 = 36.0$ Hz, 2H), 0.01 (d, J = 12.0 Hz, 6H).

¹³C NMR (101 MHz, CDCl₃) δ 171.2, 136.2, 136.0, 134.5, 133.8, 133.0, 130.0, 129.0, 128.8, 128.7, 128.2, 126.4, 126.2, 121.0, 72.0, 69.8, 42.0, 29.6, -4.1.

IR (film, cm⁻¹) :3276, 3032, 2956, 2932, 2854, 1735, 1311, 1255, 1222, 1170, 816, 749, 700;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}{}^{35}CINNaO_4SSi^+$ ([M]+Na⁺) = 522.0933, found 522.0935;

HRMS (ESI⁺) m/z calcd for $C_{25}H_{26}^{37}$ ClNNaO₄SSi⁺ ([M]+Na⁺) = 524.0904, found 524.0903.



	Retention Time	Area	% Area
1	15.511	748493	22.36
2	17.346	736586	22.00
3	18.973	930335	27.79
4	21.871	932683	27.86



	Retention Time	Area	% Area
1	14.737	10914629	75.48
2	17.192	674325	4.66
3	18.847	693571	4.80
4	21.600	2178479	15.06

benzyl

benzyl (S)-3-((S)-(dimethylsilyl)(phenyl)methyl)-2,3-dihydrobenzo[d]isothiazole-3-carboxylate 1,1-dioxide(3x)



Colorless oil, 25.2 mg, 56% yield, 80%/43% ee, 73:27 dr, $[\alpha]_D^{20} = -58.4$ (c = 0.25, CH₂Cl₂); **HPLC** Daicel chiralcel IC, *n*-hexane/*i*-PrOH 90/10, 1.0 mL/min, $\lambda = 220$ nm, $t_1 = 29.81$ min, $t_2 = 33.91$ min, $t_3 = 40.66$ min, $t_4 = 44.80$ min; ¹**H NMR** (400 MHz, CDCl3) δ 7.54 (dd, $J_1 = 7.8$, $J_2 = 8.0$ Hz, 2H), 7.45 – 7.33 (m, 8H), 7.16 (d, J = 8.0 Hz, 2H), 7.05 (t, J = 8.0 Hz, 2H), 5.89 (s, 1H), 5.26 (dd, $J_1 = 12.0$, $J_2 = 16.0$

Hz,, 2H), 4.06 – 4.02 (m, 1H), 3.30 (d, *J* = 4.0 Hz, 1H), 0.07 (d, *J* = 4.0 Hz, 3H), -0.00 (d, *J* = 4.0 Hz, 3H).

¹³C NMR (101 MHz, CDCl₃) δ 170.8, 137.7, 136.2, 136.0, 134.6, 134.1, 133.6, 133.0, 130.0, 129.0, 129.0, 128.7, 128.1, 126.3, 125.9, 121.1, 72.3, 69.5, 42.8, -4.4, -4.7.

IR (film, cm⁻¹) :3276, 3035, 2955, 2929, 2855, 1733, 1305, 1259, 1222, 1172, 829, 757, 706;

HRMS (ESI⁺) m/z calcd for $C_{24}H_{25}NNaO_4SSi^+$ ([M]+Na⁺) = 474.1166, found 474.1164.



	Retention Time	Area	% Area
1	29.810	1993913	7.24
2	33.908	2251016	8.18
3	40.658	17555608	63.78
4	44.795	5726485	20.80

dibenzyl [3,3'-bibenzo[d]isothiazole]-3,3'(2H,2'H)-dicarboxylate 1,1,1',1'-tetraoxide (4a)



¹**H NMR** (400 MHz, DMSO-*d*₆) δ 9.24 (s, 2H), 7.89 – 7.87 (m, 2H), 7.68 – 7.66 (m, 2H), 7.54 – 7.51 (m, 4H), 7.29 – 7.27 (m, 6H), 7.20 (dd, J_I = 4.0 Hz, J_2 = 8.0 Hz, 4H), 5.14 – 5.08 (m, 4H). ¹³**C NMR** (101 MHz, DMSO-*d*₆) δ 168.1, 135.8, 135.2, 133.0, 131.8, 131.3, 128.8, 128.7, 128.6, 128.4, 128.2, 127.8, 121.2, 71.9, 68.5.

HRMS (ESI⁺) m/z calcd for $C_{30}H_{24}N_2NaO_8S_2^+$ ([M]+Na⁺) = 627.0867, found 627.0866.

13. Copies of NMR spectra and CD spectra for the reaction substrates and products

















fl (ppm)



















3d





e





3f





3g




-84 -86 -88 -90 -92 -94 -96 -98 -100 -102 -104 -106 -108 -110 -112 -114 -116 -118 -120 -122 -124 -126 -128 -130 -132 -134 -136 -138 -140 -142 -144 -146 -148 -150 fl (ppm)



3h









-58.3 -58.4 -58.5 -58.6 -58.7 -58.8 -58.9 -59.0 -59.1 -59.2 -59.3 -59.4 -59.5 -59.6 -59.7 -59.8 -59.9 -60.0 -60.1 -60.2 -60.3 -60.4 -60.5 -60.6 -60.7 -60.8 -60.9 -61.0 -61.1 -61.2 -61.3 -61.4 -61.5 -61 fl (ppm)





















3m















p

















































x



