Electronic Supplementary Information

Visible-light-driven alkene dicarboxylation with formate and

CO₂ under mild conditions

Fulin Zhang,^a Xiao-Yang Wu,^a Pan-Pan Gao,^a Hao Zhang,^a Zhu Li,^a Shangde Ai^a and Gang Li^{*ab}

^aFrontiers Science Center for Transformative Molecules, Shanghai Key Laboratory for Molecular Engineering of Chiral Drugs, School of Chemistry and Chemical Engineering, Zhang Jiang Institute for Advanced Study, Shanghai Jiao Tong University, Shanghai 200240, China

^bFujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fujian 350002, China

Table of Contents

1. General information	S3
2. Optimization of the reaction conditions	S4
3. General procedures	S8
4. Mechanistic Experiments and Product Application	
5. References	
6. Copies of ¹ H and ¹³ C NMR spectra	S35

1. General information

Chemicals were purchased from commercial suppliers and used without further purification unless otherwise stated. Solvents were generally dried over 4 Å molecular sieves. Unless otherwise indicated, all reactions were set up in glovebox in Schlenk tubes. Flash column chromatography was performed using silica gel 60 (200-300 mesh) or preparative thin layer chromatography. Reaction was monitored by thin layer chromatography (TLC) and was visualized by ultraviolet light (254 nm). Melting points were recorded using a SRS Melting Point thermometer. ¹H and ¹³C NMR spectra were recorded on Bruker-BioSpin AVANCE III HD 400MHz spectrometer and Bruker Avance NEO 500MHz. Chemical shifts are reported parts per million (ppm) referenced to CDCl₃ (δ 7.26 ppm), tetramethylsilane (TMS, δ 0.00 ppm), DMSO-*d*₆ (δ 2.50 ppm), CD₃OD (δ 3.31 ppm) for ¹H NMR; CDCl₃ (δ 77.16 ppm), DMSO- d_6 (δ 40.00 ppm), CD₃OD (δ 49.00 ppm) for ¹³C NMR. Data for ¹H NMR are recorded as follows: chemical shift (δ , ppm), multiplicity (s = singlet; d = doublet; t = triplet; q = quarter; p = pentet; m = multiplet; br = broad), coupling constant (Hz), integration. Data for ¹³C NMR are reported in terms of chemical shift (δ , ppm). HRMS were obtained on SolariX 7.0T and Waters Micromass Q-Tof Premier and XEVO-G2XSQTOF mass spectrometer. CO₂ gas (Purity: 99.999%) was purchased from Air Liquide. The light source used for the photocatalyzed experiment was a 30 W blue LED strips, purchased from Xuzhou Ai Jia electronic technology Co. LTD. (China) and 30W blue LED purchased from Shenzhen Xin Xing Yuan electronic techn -ology Co. LTD. (China)

2. Optimization of the reaction conditions

An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with base (0.6 mmol, 3.0 equiv), HCOOK (50.5 mg, 0.6 mmol, 3.0 equiv), photocatalyst (8.0 mg, 6.2 mol %), HAT (30 mol%), alkene (0.20 mmol, 1.0 equiv, if solid). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the alkene (0.2 mmol, if liquid) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LED strips (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation and the yield of crude products was determined via ¹H NMR using CH₂Br₂ as internal standard.



Photocatalysts:



Table S1. Screening of Bases^a

3DPAFIPN (6.2 mol%) DABCO (30 mol%) HCOOK (3.0 equiv) HOOC HOOC II Base (3.0 equiv)			ноос
Ph ^{AA} Ph 1a	+ CO ₂ DMSO (0.1 M), blue LEDs, (1 atm) then HCl	rt, 24 h Ph Ph 2a	Ph Ph 2a _H
Entry	Base	2a [%] ^b	$2a_{\rm H} [\%]^b$
1	None	11%	66%
2	CsF	23%	54%
3	Cs_2CO_3	40%	12%
4	KF	29%	36%
5	'BuOLi	42%	50%
6	CsOAc	38%	68%
7	NaOAc	44%	31%
8	KOAc	23%	34%
9	LiOAc	13%	40%
10	Na ₂ CO ₃	24%	32%
11	K ₃ PO ₄	44%	50%
12	КОН	27%	54%
13	Na ₃ PO ₄	23%	52%
14	NaOCH ₃	23%	60%
15	DMAP	26%	34%
16	DBU		18%
17	Pyridine	8%	40%
18	1,10-phenanthroline	16%	18%
19 ^c	Cs ₂ CO ₃	59%	
20^{d}	K ₃ PO ₄	64%	

21 ^e	NaOAc	43%	48%
-----------------	-------	-----	-----

^{*a*}Reaction conditions: 1,1-diphenylethylene (0.2 mmol), 3DPAFIPN (6.2 mol%), DABCO (0.06 mmol), Base (0.6 mmol), HCOOK (0.6 mmol) in DMSO (2 mL), 1 atm CO₂, 30 W blue LEDs (450 - 455 nm), rt, 24 h. DABCO = triethylenediamine. ^{*b*}Yield was determined by ¹H NMR with CH₂Br₂ as internal standard. ^{*c*}Cs₂CO₃ (5.0 equiv). ^{*d*}K₃PO₄ (5.0 equiv). ^{*c*}NaOAc (5.0 equiv).

Table S2. Screening of cooperative bases^a

$\begin{array}{c} 3DPAFIPN (6.2 \text{ mol%}) \\ DABCO (30 \text{ mol%}) \\ HCOOK (3.0 \text{ equiv}) \\ Cs_2CO_3 (3.0 \text{ equiv}) \\ Cs_2CO_3 (3.0 \text{ equiv}) \\ Cs_2CO_3 (3.0 \text{ equiv}) \\ DMSO (0.1 \text{ M}), \text{ blue LEDs, rt, 24 h} \end{array} \xrightarrow{HOOC} \begin{array}{c} HOOC \\ Ph \\ Ph \\ Ph \\ Ph \end{array} \xrightarrow{HOC} Ph \\ Ph $			
Entry	cooperative base	2a [%] ^b	$2a_{\rm H} [\%]^b$
1	KPF ₆	55%	
2	KOPiv	31%	38%
3	KF	42%	
4	K ₃ PO ₄	91%(88%)	
5 ^c	K ₃ PO ₄	74%	
6 ^{<i>d</i>}	K ₃ PO ₄	65%	
7^e	K ₃ PO ₄	71%	

^{*a*}Reaction conditions: 1,1-diphenylethylene (0.2 mmol), 3DPAFIPN (6.2 mol%), DABCO (0.06 mmol), Cs₂CO₃ (0.6 mmol), additive (0.4 mmol), HCOOK (0.6 mmol) in DMSO (2 mL), 1 atm CO₂, 30 W blue LEDs (450 - 455 nm), rt, 24 h. DABCO = triethylenediamine. ^{*b*}Yield was determined by ¹H NMR with CH₂Br₂ as internal standard. Numbers in parentheses are referred to isolated yields. ^{*c*}K₃PO₄ (1.0 equiv). ^{*d*}Cs₂CO₃ (2.0 equiv) and K₃PO₄ (1.0 equiv). ^{*c*}Cs₂CO₃ (2.0 equiv).

Table S3. Screening of HATs^a

Ph Ph 1a	3DPAFIPN (6.2 mol%) HAT (30 mol%) HCOOK (3.0 equiv) CS ₂ CO ₃ (3.0 equiv) K ₃ PO ₄ (2.0 equiv) DMSO (0.1 M), blue LEDs, rt, 24 h then HCl	HOOC HOOC Ph COOH Ph Ph Ph 2a	р Рh 2а _н
Entry	HAT	2a [%] ^b	2a _H [%] ^b
1	quinuclidine	40%	
2	Mesna	56%	20%
3	<i>p</i> -Toluenethiol	81%	
4	2-Mercaptobenzoic Acid	77%	
	Methyl Ester		
5	L-Cysteine	50%	48%
6 ^{<i>c</i>}	DABCO	65%	
7 ^{<i>d</i>}	DABCO	71%	

^aReaction conditions: 1,1-diphenylethylene (0.2 mmol), 3DPAFIPN (6.2 mol%), HAT (0.06 mmol), Cs₂CO₃ (0.6 mmol), K₃PO₄ (0.4

mmol), HCOOK (0.6 mmol) in DMSO (2 mL), 1 atm CO₂, 30 W blue LEDs (450 - 455 nm), rt, 24 h. DABCO = triethylenediamine. ^bYield was determined by ¹H NMR with CH₂Br₂ as internal standard. ^cDABCO (10 mol%). ^dDABCO (20 mol%).

Ph Ph 1a	PC (6.2 mol%) DBCO (30 mol%) HCOOK (3.0 equiv) CS₂CO ₃ (3.0 equiv) K ₃ PO ₄ (2.0 equiv) DMSO (0.1 M), blue LEDs, rt, 24 h then HCl	HOOC Ph Ph 2a HOOC Ph Ph Ph 2a 2a	`Ph ⊮
Entry	Photocatalyst (PC)	2a [%] ^b	2a _H [%] ^b
1	None		
2	3DPAFIPN	91%(88%) ^c	
3^d	Ru(bpy) ₃ Cl ₂		
4 ^{<i>d</i>}	Ir(ppy) ₂ (dtbbpy)(PF ₆)	7%	
5^d	[Ir(dF(CF ₃)ppy) ₂ (dtbbpy)]PF ₆		
6 ^{<i>d</i>}	[Ir(dF(Me)ppy) ₂ (dtbbpy)]PF ₆	20%	
7	4DPAIPN	65%	
8	4CzIPN	39%	
9 ^e	3DPAFIPN	70%	
10 ^f	3DPAFIPN	86%	
11 ^g	3DPAFIPN		
12^h	3DPAFIPN	32%	28%
13^i	3DPAFIPN		76%
14 ^j	3DPAFIPN		

Table S4. Screening of photocatalysts^a

^{*a*}Reaction conditions: 1,1-diphenylethylene (0.2 mmol), PC (6.2 mol%), DABCO (0.06 mmol), Cs₂CO₃ (0.6 mmol), K₃PO₄ (0.4 mmol), HCOOK (0.6 mmol) in DMSO (2 mL), 1 atm CO₂, 30 W blue LEDs (450 - 455 nm), rt, 24 h. DABCO = triethylenediamine. ^{*b*}Yield was determined by ¹H NMR with CH₂Br₂ as internal standard. ^cIsolated yield. ^{*d*}PC (5.0 mol%). ^{*c*}3DPAFIPN (2.0 mol%). ^{*f*}3DPAFIPN (4.0 mol%). ^{*g*}In darkness. ^{*b*}Without DABCO. ^{*f*}In N₂ atmosphere. ^{*f*}Without HCOOK.

Table S5. Screening of Solvents

Ph ⁺ CO ₂ + CO ₂ - 1a (1 atm)	3DPAFIPN (6.2 mol%) DABCO (30 mol%) HCOOK (3.0 equiv) Cs ₂ CO ₃ (3.0 equiv) K ₃ PO ₄ (2.0 equiv) Solvent (0.1 M), blue LEDs, rt, 24 h then HCl	HOOC Ph Ph Ph 2a	∠ _{Ph} 2a _H
Entry	Solvent	2a [%] ^b	2a _H [%] ^b
1	DMF	15%	
2	1,4-dioxane		
3	THF	16%	
4	MeCN		
5	DMA	17%	

^aReaction conditions: 1,1-diphenylethylene (0.2 mmol), 3DPAFIPN (6.2 mol%), DABCO (0.06 mmol), Cs₂CO₃ (0.6 mmol), K₃PO₄ (0.4

mmol), HCOOK (0.6 mmol) in slovent (2 mL), 1 atm CO₂, 30 W blue LEDs (450 - 455 nm), rt, 24 h. DABCO = triethylenediamine. b Yield was determined by ¹H NMR with CH₂Br₂ as internal standard.

3. General procedures



An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (2.0 equiv, 0.4 mmol, 85.5 mg), Cs₂CO₃ (3.0 equiv, 0.6 mmol, 195.5 mg), HCOOK (3.0 equiv, 0.6 mmol, 50.5 mg), 3DPAFIPN (6.2 mol%, 8.0 mg, 3DPAFIPN was synthesized following the literature procedures¹), DABCO (30 mol%), alkene **1** (0.20 mmol, 1.0 equiv, if solid). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the alkene (0.2 mmol, if liquid) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LED strips (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product **2**. If there was traces of DMSO, extraction with H₂O.



An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K_3PO_4 (2.0 equiv, 0.4 mmol, 85.5 mg), Cs_2CO_3 (3.0 equiv, 0.6 mmol, 195.5 mg), HCOOK (3.0 equiv, 0.6 mmol, 50.5 mg), 3DPAFIPN (6.2 mol%, 8.0 mg),

DABCO (30 mol%, 6.8 mg), alkene (0.20 mmol, 1.0 equiv, if solid). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the alkene **3** (0.2 mmol, if liquid) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LED strips ($\lambda = 450 - 455$ nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times, then concentrated in vacuo. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product **4** (If traces of DMSO were present with product, extract with H₂O to remove DMSO).



An oven-dried reaction tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (2.0 equiv, 0.4 mmol, 85.5 mg), Cs₂CO₃ (3.0 equiv, 0.6 mmol, 195.5 mg), HCOOK (3.0 equiv, 0.6 mmol, 50.5 mg), 3DPAFIPN (6.2 mol%, 8.0 mg), DABCO (30 mol%, 6.8 mg). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the alkene **1n** or **3m** (0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LED strips (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) light source and stirred at ambient temperature for 24 h. Upon completion of the reaction, then AcOH (0.2 mL) was added and the system was stirred for 10 min at room temperature. The solvent along with other volatile matter was removed directly under reduced pressure at high temperature. The residue was re-dissolved in MeOH (8 mL) and then SOCl₂ (0.8 mL) was added dropwise carefully. The Schlenk tube was capped and stirred at 100 °C for 10 h. After cooled to room

temperature, the mixture was diluted with EtOAc (10 mL) and quenched with saturated aqueous solution of sodium hydrogen carbonate. The reaction was extracted with EtOAc (30 mL \times 3) and the organic layers were combined and dried over Na₂SO₄. After concentrated under reduced pressure, crude ¹H NMR spectrum was taken using CH₂Br₂ as internal standard. The resulting residue was purified by flash silica gel chromatography or preparative thin layer chromatography using petroleum ether/EtOAc (3:1-1:1) as the eluent to give the desired products.

Large Scale Reaction Procedure:



An oven-dried round-bottom flask (250 mL) containing a stirring bar was cooled to room temperature. The round-bottom flask was then introduced in a glovebox, where it was charged with K₃PO₄ (2.0 equiv, 12.0 mmol, 2.55 g), Cs₂CO₃ (3.0 equiv, 18.0 mmol, 5.86 g), HCOOK (3.0 equiv, 18.0 mmol, 1.51 g), 3DPAFIPN (6.2 mol%, 240 mg), DABCO (30 mol%, 202 mg). The flask was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (10 mL, 0.1 M) and 1,1-diphenylethylene (6.0 mmol, 1.0 equiv, 1.08 g) were added into the flask via syringe. Then the flask was evacuated and back-filled with CO₂ for 10 times. Then the reaction flask was attached with a CO₂ gas bag (about 3 L). Finally, the tube was placed under two 30 W blue LEDs (purchased from Shenzhen Xin Xing Yuan electronic techn-ology Co. LTD. Of China; $\lambda = 450 - 455$ nm; 3.0 cm - 4.5 cm away from the LED with a fan to keep the reaction temperature at about 36 °C) light source and stirred at ambient temperature for 52 h. Then, the mixture was quenched with H₂O and 10 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product in 85% yield (1.38 g).



Set-up the reaction outside the glovebox



Take some K₃PO₄, Cs₂CO₃, HCOOK, DABCO and 3DPAFIPN from bottles kept in the glovebox and put them into vials respectively. Then weigh these chemicals using a balance outside the glovebox. To an oven-dried reaction tube containing a stirring bar cooled to room temperature was added K₃PO₄ (2.0 equiv, 0.4 mmol, 85.5 mg), Cs₂CO₃ (3.0 equiv, 0.6 mmol, 195.5 mg), HCOOK (3.0 equiv, 0.6 mmol, 50.5 mg), 3DPAFIPN (6.2 mol%, 8.0 mg) and DABCO (30 mol%, 6.8 mg). Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and 1,1-diphenylethylene (1.0 equiv, 0.2 mmol, 36 mg) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs ($\lambda = 450 - 455$ nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 28 °C) and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product **2a** in 89% yield (48.1 mg).

2,2-diphenylsuccinic acid (2a)²



Product (2a) was obtained as a white solid (47.4 mg, 88%); $R_f = 0.4$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.35 - 7.32 (m, 4H), 7.27 - 7.18 (m, 6H), 3.50 (s, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 176.6, 174.3, 144.4, 129.9, 128.8, 127.8, 58.5, 44.7.

2-phenyl-2-(*o*-tolyl)succinic acid (2b)²



Product (2b) was obtained as a white solid (43.7 mg, 77%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.51 - 7.47 (m, 1H), 7.40 - 7.37 (m, 2H), 7.26 - 7.15 (m, 5H), 7.09 - 7.07 (m, 1H), 3.62 (d, *J* = 15.6 Hz, 1H), 3.41 (d, *J* = 15.6 Hz, 1H), 1.86 (s, 3H).¹³C NMR (100 MHz, CD₃OD) δ 176.1, 174.4, 143.7, 141.7, 138.9, 133.5, 130.0, 129.7, 128.8, 128.3, 127.7, 126.4, 58.7, 45.0, 22.0. **2-(2-fluorophenyl)-2-phenylsuccinic acid (2c)**²



Product (2c) was obtained as a colorless oil (42 mg, 73%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.45 - 7.24 (m, 7H), 7.09 - 6.97 (m, 2H), 3.58 (2H, ABq, J = 17.2 Hz). ¹³C NMR (100 MHz, CD₃OD) δ 176.0, 174.2, 162.3 (d, J = 246 Hz), 141.8, 132.5 (d, J = 3.6 Hz), 131.4 (d, J = 10.8 Hz), 130.2 (d, J = 9.0 Hz), 129.2, 129.0, 128.2, 124.2 (d, J = 3.3 Hz), 116.7 (d, J = 23.2 Hz), 56.6, 41.6 (d, J = 3.0 Hz).¹⁹F NMR (376 MHz, CD₃OD) δ -109.5.

2-phenyl-2-(m-tolyl)succinic acid (2d)



Product (2d) was obtained as a white solid (42 mg, 74%); mp: 151-152 °C; $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.34 - 7.31 (m, 2H), 7.27 - 7.20 (m, 3H), 7.17 - 7.10 (m, 3H), 7.05 - 7.03 (m, 1H), 3.47 (2H, ABq, J = 16.8 Hz), 2.27 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 176.7, 174.4, 144.5, 144.4, 138.4, 130.5, 130.0, 128.71, 128.68, 128.5, 127.8, 126.8, 58.4, 44.8, 21.6. HRMS (ESI) m/z calcd for C₁₇H₁₅O₄⁻ (M-H)⁻ 283.0976; found 283.0967.

2-phenyl-2-(3-(trifluoromethyl)phenyl)succinic acid (2e)²



Product (2e) was obtained as a white solid (41 mg, 62%); $R_f = 0.4$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.52 (s, 1H), 7.35 - 7.29 (m, 2H), 7.24 - 7.20 (m, 1H), 7.15 - 7.03 (m, 5H), 3.46 (d, *J* = 16.4 Hz, 1H), 3.19 (d, *J* = 16.8 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.0, 174.0, 145.9, 144.0, 134.2, 130.9 (q, *J* = 31.6 Hz), 129.4, 129.31, 129.28, 128.4, 127.4 (q, *J* = 3.9 Hz), 125.7 (q, *J* = 270 Hz), 124.5 (q, *J* = 3.8 Hz), 58.5, 44.5. ¹⁹F NMR (376 MHz, CD₃OD) δ -64.0.

2-phenyl-2-(*p*-tolyl)succinic acid (2f)²



Product (2f) was obtained as a white solid (35 mg, 62%); $R_f = 0.3$ (Petroleum ether :

Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.34 - 7.31 (m, 2H), 7.26 - 7.19 (m, 5H), 7.08 - 7.06 (m, 2H), 3.46 (2H, ABq, *J* = 16.4 Hz), 2.29 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 176.8, 174.4, 144.6, 141.5, 137.6, 129.9, 129.7, 129.4, 128.7, 127.7, 58.2, 44.8, 20.9.

2,2-di-p-tolylsuccinic acid (2g)



Product **(2g)** was obtained as a white solid (42 mg, 70%); mp: 117-118 °C; $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.43 (br, 2H), 7.16 - 7.14 (m, 4H), 7.07 - 7.05 (m, 4H), 3.34 (s, 2H), 2.25 (s, 6H). ¹³C NMR (100 MHz, DMSO-*d*₆) δ 174.2, 171.9, 140.6, 135.5, 128.4, 128.2, 56.0, 43.5, 20.5. HRMS (ESI) m/z calcd for C₁₈H₁₇O₄⁻ (M-H)⁻ 297.1132; found 297.1122.

2-(4-methoxyphenyl)-2-phenylsuccinic acid (2h)²



Product (**2h**) was obtained as a white solid (45 mg, 75%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.34 - 7.32 (m, 2H), 7.27 - 7.18 (m, 5H), 6.82 - 6.79 (m, 2H), 3.75 (s, 3H), 3.46 (2H, ABq, J = 16.4 Hz). ¹³C NMR (100 MHz, CD₃OD) δ 176.9, 174.4, 159.8, 144.7, 136.4, 131.0, 129.9, 128.7, 127.8, 114.1, 57.9, 55.6, 44.9.

2,2-bis(4-methoxyphenyl)succinic acid (2i)



Product (2i) was obtained as a colorless oil (48 mg, 74%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.25 - 7.22 (m, 4H), 6.83 - 6.79 (m, 4H), 3.77 (s, 6H), 3.41 (s, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 177.2, 174.5, 159.8, 136.7, 131.0, 114.0, 57.3, 55.6, 45.1. HRMS (ESI) m/z calcd for C₁₈H₁₇O₆⁻ (M-H)⁻ 329.1031; found 329.1018.

2-(4-fluorophenyl)-2-phenylsuccinic acid (2j)²



Product (2j) was obtained as a white solid (45 mg, 78%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.38 - 7.32 (m, 4H), 7.30 - 7.20 (m, 3H), 7.00 - 6.94 (m, 2H), 3.55 (d, J = 16.4 Hz, 1H), 3.42 (d, J = 16.4 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.4, 174.2, 162.9 (d, J = 243.5 Hz), 144.4, 140.5 (d, J= 3.4 Hz), 132.1 (d, J = 7.9 Hz), 129.6, 129.0, 128.0, 115.2 (d, J = 21.3 Hz), 58.0, 44.7. ¹⁹F NMR (376 MHz, CD₃OD) δ -118.3.

2-(3,4-dimethoxyphenyl)-2-phenylsuccinic acid (2k)



Product (2k) was obtained as a colorless oil (29.8 mg, 72%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.36 - 7.34 (m, 2H), 7.28 - 7.19 (m, 3H), 6.97 (s, 1H), 6.88 - 6.82 (m, 2H), 3.80 (s, 3H), 3.69 (s, 3H), 3.47 (s, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 176.8, 174.3, 149.6, 149.5, 144.6, 137.2, 129.9,

128.8, 127.8, 122.4, 114.9, 112.1, 58.2, 56.4, 44.9. HRMS (ESI) m/z calcd for $C_{18}H_{17}O_{6}^{-}$ (M-H)⁻ 329.1031; found 329.1018.

2-phenyl-2-(thiophen-3-yl)succinic acid (2l)²



Product **(21)** was obtained as a white solid (30 mg, 55%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.37 - 7.34 (m, 2H), 7.29 - 7.21 (m, 4H), 7.03 - 7.00 (m, 1H), 6.92 - 6.90 (m, 1H), 3.66 (d, J = 16.8 Hz, 1H), 3.47 (d, J = 16.8 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 175.9, 173.9, 147.5, 144.4, 129.1, 128.4, 128.3, 127.6, 126.7, 126.3, 56.0, 45.6.

2-(benzofuran-2-yl)-2-phenylsuccinic acid(2m)²



Product (2m) was obtained as a white solid (16 mg, 51.6%); $R_f = 0.3$ (Petroleum ether : acetone = 3 : 1); ¹H NMR (500 MHz, CD₃OD) δ 7.97 (d, J = 7.0 Hz, 1H), 7.52 (d, J = 8.0 Hz, 1H), 7.44 (d, J = 8.0 Hz, 2H), 7.32 - 7.27 (m, 4H), 7.22 -7.14 (m, 1H), 5.66 - 5.63 (m, 1H), 3.28 (d, J = 9.0 Hz, 1H), 3.11 (dd, J = 16.0, 7.0 Hz, 1H). ¹³C NMR (126 MHz, CD₃OD) δ 174.7, 166.9, 155.2, 141.8, 129.7, 128.9, 128.2, 127.6, 125.8, 125.0, 123.2, 111.9, 110.0, 41.3, 39.1.

dimethyl 2-methyl-2-(pyridin-3-yl)succinate (2n)



Product (2n) was obtained as a yellow oil (18 mg, 38%); $R_f = 0.5$ (Petroleum ether : ethyl acetate = 1 : 1); ¹H NMR (500 MHz, CDCl₃, TMS) δ 8.62 (s, 1H), 8.52 (d, J = 5.0 Hz, 1H), 7.67 (d, J = 8.0 Hz, 1H), 7.29 - 7.26 (m, 1H), 3.72 (s, 3H), 3.68 (s, 3H),

3.25 (d, J = 16.0 Hz, 1H), 2.88 (d, J = 16.5 Hz, 1H), 1.74 (s, 3H). ¹³C NMR (126 MHz, CDCl₃) δ 174.7, 171.1, 148.7, 147.7, 138.0, 133.8, 123.4, 52.8, 51.9, 47.1, 43.24, 23.2. HRMS (ESI) m/z calcd for C₁₂H₁₅NO₄Na⁺ (M+Na)⁺ 260.0901; found 260.0899.

2-butyl-2-phenylsuccinic acid (20)³



Product (20) was obtained as a white solid (31 mg, 63%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.37 - 7.30 (m, 4H), 7.25 - 7.21 (m, 1H), 3.20 (d, J = 16.0 Hz, 1H), 3.02 (d, J = 16.0 Hz, 1H), 2.24 - 2.09 (m, 2H), 1.20 - 1.01 (m, 2H), 0.87 (t, J = 7.2 Hz, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 178.6, 174.7, 143.2, 129.4, 127.9, 127.4, 52.8, 40.0, 38.9, 18.6, 14.8.

2-methyl-2-phenylsuccinic acid (2p)²

Product (**2p**) was obtained as a white solid (25 mg, 60%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.42 - 7.40 (m, 2H), 7.35 - 7.31 (m,, 2H), 7.26 - 7.22 (m, 1H), 3.21 (d, J = 16.4 Hz, 1H), 2.82 (d, J = 16.4 Hz, 1H), 1.69 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 178.8, 174.7, 144.7, 129.5, 128.0, 126.8, 49.2, 44.3, 24.0.

2-(3-cyanophenyl)-2-methylsuccinic acid (2q)



Product (2q) was obtained as a white solid (31 mg, 68%); mp: 140-141 °C; $R_f = 0.4$

(Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.78 - 7.74 (m, 2H), 7.64 - 7.62 (m, 1H), 7.54 - 7.50 (m, 1H), 3.15 (d, *J* = 16.8 Hz, 1H), 2.93 (d, *J* = 16.8 Hz, 1H), 1.72 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 177.7, 174.1, 146.3, 132.2, 131.8, 131.0, 130.6, 119.7, 113.5, 44.0, 23.8. HRMS (ESI) m/z calcd for C₁₂H₁₀NO₄⁻ (M-H)⁻ 232.0615; found 232.0605.

3-methyl-2,2-diphenylsuccinic acid (2r)



Product (2r) was obtained as a white solid (30 mg, 53%); mp: 138-139 °C; $R_f = 0.4$ (Petroleum ether : Acetone = 2 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.40 - 7.22 (m, 10H), 4.15 (q, *J* = 7.2 Hz, 1H), 1.12 (d, *J* = 7.2 Hz, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 177.8, 177.3, 143.0, 141.7, 132.5, 130.7, 128.8, 128.1, 127.8, 127.7, 63.5, 45.2, 15.2. HRMS (ESI) m/z calcd for C₁₇H₁₆O₄Na⁺ (M+Na)⁺ 307.0941; found 307.0942.

2-(2,2-diphenylvinyl)succinic acid (2s_a)



Product (**2s**_a) was obtained as a white solid (22 mg, 37%); mp: 172-173 °C; $R_f = 0.3$ (Petroleum ether : Acetone = 2 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.43 - 7.33 (m, 3H), 7.28 - 7.18 (m, 7H), 6.04 (d, J = 10.8 Hz, 1H), 3.59 - 3.53 (m, 1H), 2.81 - 2.75 (m, 1H), 2.54 - 2.49 (m, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.7, 174.9, 146.5, 143.3, 140.6, 130.9, 129.4, 129.2, 128.6, 128.4, 126.1, 43.7, 37.9. HRMS (ESI) m/z calcd for C₁₈H₁₆O₄Na⁺ (M+Na)⁺ 319.0941; found 319.0942.

(E)-2,2-diphenylhex-3-enedioic acid (2sb)



Product (2sb) was obtained as a white solid (10 mg, 17%); mp: 155-156 °C; $R_f = 0.4$ (Petroleum ether : Acetone = 2 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.32 - 7.24 (m, 6H), 7.20 - 7.18 (m, 4H), 6.58 (d, J = 15.8 Hz, 1H), 5.07 - 5.00 (m, 1H), 3.13 (dd, J = 7.0, 0.8 Hz, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 177.2, 175.4, 143.1, 138.8, 130.6, 128.8, 128.1, 127.1, 65.2, 38.8. HRMS (ESI) m/z calcd for C₁₈H₁₅O₄⁻ (M-H)⁻ 295.0976; found 295.0961.

2-(tert-butoxycarbonyl)-2-methylsuccinic acid (2t)²



Product (2t) was obtained as a white solid (35 mg, 76%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CDCl₃, TMS) δ 3.01 (2H, ABq, J = 17.2 Hz), 1.54 (s, 3H), 1.46 (s, 9H).¹³C NMR (100 MHz, CDCl₃) δ 176.8, 175.6, 172.0, 83.9, 51.2, 40.1, 27.6, 21.9.

2-((cyclohexyloxy)carbonyl)-2-methylsuccinic acid (2u)²

Product (2u) was obtained as a white solid (42 mg, 82%); $R_f = 0.3$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 4.84 - 4.78 (m, 1H), 2.97 (d, J = 16.8 Hz, 1H), 2.83 (d, J = 17.2 Hz, 1H), 1.80 - 1.70 (m, 4H), 1.53 - 1.44 (m, 6H), 1.42 - 1.29 (m, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 174.6, 173.9, 172.4, 74.7, 52.8, 40.9, 32.0, 26.5, 24.2, 20.8.

2-benzyl-2-(methoxycarbonyl)succinic acid (2v)



Product (2v) was obtained as a white solid (21 mg, 40%); mp: 134-135 °C; $R_f = 0.4$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.29 - 7.21 (m, 3H), 7.12 - 7.10 (m, 2H), 3.73 (s, 3H), 3.38 (dd, J = 31.6, 14.0 Hz, 2H), 2.76 (dd, J = 44.0, 17.6 Hz, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 174.1, 172.9, 172.5, 137.4, 131.1, 129.4, 128.2, 57.7, 53.1, 39.5, 37.3. HRMS (ESI) m/z calcd for C₁₃H₁₃O₆⁻ (M-H)⁻ 265.0718; found 265.0708.

2-phenylsuccinic acid (4a)²



Product **(4a)** was obtained as a white solid (28 mg, 72%); $R_f = 0.3$ (Petroleum ether : Acetone = 2 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.35 - 7.25 (m, 5H), 4.01 (dd, J = 10.0, 5.2 Hz, 1H), 3.10 (dd, J = 16.8, 10.4 Hz, 1H), 2.62 (dd, J = 16.8, 5.2 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.6, 175.2, 139.9, 129.8, 128.9, 128.5, 48.7, 38.7.

2-(*o*-tolyl)succinic acid (4b)²



Product **(4b)** was obtained as a white solid (30 mg, 72%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.23 - 7.11(m, 4H), 4.32 (dd, J = 10.0, 5.2 Hz, 1H), 3.08 (dd, J = 17.2, 10.0 Hz, 1H), 2.57 (dd, J = 16.8, 5.2 Hz, 1H), 2.42 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 176.8, 175.4, 138.3, 137.4, 131.7, 128.3, 127.6, 127.4, 44.2, 38.1, 19.8.

2-(2-methoxyphenyl)succinic acid (4c)



Product (4c) was obtained as a white solid (35 mg, 78%) (42 mg, 94%, note: the reaction temperature is 40 to 45°C); mp: 177-178 °C; $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.27 - 7.23 (m, 1H), 7.18 - 7.16 (m, 1H), 6.98 - 6.96 (m, 1H), 6.92 - 6.88 (m 1H), 4.35 (dd, J = 9.6, 5.2 Hz, 1H), 3.83 (s, 3H), 3.01 (dd, J = 16.8, 9.6 Hz, 1H), 2.52 (dd, J = 16.8, 5.2 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 177.0, 175.7, 158.3, 129.8, 128.4, 121.7, 112.1, 56.0, 43.2, 37.5. HRMS (ESI) m/z calcd for C₁₁H₁₁O₅⁻ (M-H)⁻223.0612; found 223.0602.

2-(2-chlorophenyl)succinic acid (4d)⁴



Product **(4d)** was obtained as a white solid (22 mg, 48%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.44 -7.41 (m, 1H), 7.35 - 7.24 (m, 3H), 4.56 (dd, J = 10.0, 5.2 Hz, 1H), 3.05 (dd, J = 17.2, 10.0 Hz, 1H), 2.61 (dd, J = 16.8, 4.8 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 175.7, 174.9, 137.7, 134.9, 130.9, 130.1, 130.0, 128.5, 45.4, 37.6.

2-(3-methoxyphenyl)succinic acid (4e)⁵



Product (4e) was obtained as a white solid (35 mg, 78%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.25 - 7.21 (m, 1H), 6.89 - 6.81 (m, 3H), 3.99 (dd, J = 10.0, 5.2 Hz, 1H), 3.77 (s, 3H), 3.09 (dd, J = 16.8, 10.0 Hz, 1H), 2.62 (dd, J = 16.8, 5.2 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.5, 175.2, 161.3,

2-(3-chlorophenyl)succinic acid (4f)²



Product **(4f)** was obtained as a white solid (33 mg, 73%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.35 - 7.24 (m, 4H), 4.02 (dd, J = 10.0, 5.6 Hz, 1H), 3.09 (dd, J = 16.8, 9.6 Hz, 1H), 2.65 (dd, J = 16.8, 5.6 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 175.9, 174.8, 142.1, 135.5, 131.3, 129.1, 128.6, 127.4, 48.3, 38.4.

2-(3-cyanophenyl)succinic acid (4g)



Product **(4g)** was obtained as a white solid (31 mg, 71%); mp: 180-181 °C; $R_f = 0.3$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.70 - 7.63 (m, 3H), 7.53 - 7.49 (m, 1H), 4.11 (dd, J = 9.2, 5.6 Hz, 1H), 3.12 (dd, J = 17.2, 9.6 Hz, 1H), 2.69 (dd, J = 16.8, 5.6 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 175.5, 174.7, 141.6, 133.9, 132.8, 132.3, 130.9, 119.5, 113.7, 48.4, 38.2. HRMS (ESI) m/z calcd for C₁₁H₈NO₄⁻ (M-H)⁻ 218.0459; found 218.0446.

2-(*p*-tolyl)succinic acid (4h)²



Product (4h) was obtained as a white solid (25 mg, 62%); $R_f = 0.4$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.20 - 7.12 (m, 4H), 3.97 (dd, J =

10.0, 5.2 Hz, 1H), 3.08 (dd, *J* = 17.2, 10.4 Hz, 1H), 2.59 (dd, *J* = 16.8, 5.2 Hz, 1H), 2.30 (s, 3H). ¹³C NMR (100 MHz, CD₃OD) δ 176.8, 175.2, 138.3, 136.8, 130.4, 128.7, 48.2, 38.8, 21.1.

2-(4-methoxyphenyl)succinic acid (4i)⁶



Product **(4i)** was obtained as a white solid (25 mg, 56%); $R_f = 0.5$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.24 - 7.21 (m, 2H), 6.89 - 6.85 (m, 2H), 3.95 (dd, J = 10.0, 5.2 Hz, 1H), 3.76 (s, 3H), 3.07 (dd, J = 17.2, 10.0 Hz, 1H), 2.59 (dd, J = 17.2, 5.2 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.9, 175.3, 160.5, 131.7, 129.9, 115.1, 55.7, 47.8, 38.8.

2-(4-fluorophenyl)succinic acid (4j)²



Product **(4j)** was obtained as a white solid (25.5 mg, 60%); $R_f = 0.3$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, DMSO-*d*₆) δ 12.39 (br, 2H), 7.35 - 7.31 (m, 2H), 7.17 - 7.13 (m, 2H), 3.91 (dd, *J* = 10.0, 5.2 Hz, 1H), 2.94 (dd, *J* = 17.2, 10.4 Hz, 1H), 2.57 - 2.53 (m, 1H). ¹³C NMR (100 MHz, DMSO-*d*₆) δ 174.0, 172.6, 161.4 (d, *J* = 242 Hz), 134.9 (d, *J* = 3.0 Hz), 129.8 (d, *J* = 8.1 Hz), 115.4 (d, *J* = 21.1 Hz), 46.1, 37.4. ¹⁹F NMR (376 MHz, DMSO) δ -115.6.

2-(4-cyanophenyl)succinic acid (4k)²



Product (4k) was obtained as a white solid (28 mg, 65%); $R_f = 0.4$ (Petroleum ether : Acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.71 - 7.69 (m, 2H), 7.54 - 7.52 (m,

2H), 4.15 - 4.11 (m, 1H), 3.16 - 3.10 (m, 1H), 2.72 - 2.66 (m, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 175.3, 174.6, 145.5, 133.6, 130.2, 119.5, 112.3, 48.6, 48.4.

2-([1,1'-biphenyl]-4-yl)succinic acid (4l)²



Product **(41)** was obtained as a white solid (36 mg, 66%); $R_f = 0.5$ (Petroleum ether : acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.60 - 7.57 (m, 4H), 7.43 - 7.39 (m, 4H), 7.34 - 7.30 (m, 1H), 4.07 (dd, J = 10.0, 5.2 Hz, 1H), 3.14 (dd, J = 16.8, 10.0 Hz, 1H), 2.67 (dd, J = 16.8, 5.2 Hz, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.6, 175.2, 141.9, 141.7, 138.9, 129.9, 129.4, 128.4, 128.3, 127.9, 48.3, 38.7.

dimethyl 2-(pyridin-3-yl)succinate (4m)⁷



Product **(4m)** was obtained as a yellow oil (18 mg, 40%); $R_f = 0.5$ (Petroleum ether : ethyl acetate = 1 : 1); ¹H NMR (500 MHz, CDCl₃, TMS) δ 8.56 - 8.54 (m, 2H), 7.64 - 7.61 (m, 1H), 7.29 - 7.26 (m, 1H), 4.13 (dd, J = 9.5, 6.0 Hz, 1H), 3.69 (d, J = 9.5 Hz, 6H), 3.22 (dd, J = 17.0, 9.5 Hz, 1H), 2.71 (dd, J = 17.0, 6.0 Hz, 1H). ¹³C NMR (126 MHz, CDCl₃) δ 172.8, 171.6, 149.6, 149.3, 135.2, 133.5, 123.8, 52.7, 52.2, 44.8, 37.3.

2-(benzo[b]thiophen-2-yl)succinic acid (4n)²



Product (4n) was obtained as a colourless oil (16 mg, 32%); $R_f = 0.5$ (Petroleum ether : acetone = 3 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.83 - 7.71 (m, 2H), 7.36 - 7.22 (m, 3H), 4.39 (dd, J = 10.0, 5.2 Hz, 1H), 3.21 (dd, J = 16.8, 9.6 Hz, 1H), 2.87 (dd, J = 16.8, 5.2 Hz, 1H). ¹³C NMR (126 MHz, CD₃OD) δ 175.4, 174.7, 142.8, 141.0,

140.8, 125.4, 125.3, 124.4, 123.4, 123.1, 44.9, 39.0. HRMS (ESI) m/z calcd for $C_{12}H_9SO_4^-$ (M-H)⁻ 249.0224; found 249.0222.

4. Mechanistic Experiments and Product Application

Radical clock experiment



4-(2-(1-phenylvinyl)cyclopropyl)-1,1'-biphenyl was synthesized following the literature procedures⁸. An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (85.5 mg, 0.4 mmol, 2.0 equiv), Cs₂CO₃ (195.5 mg, 0.6 mmol, 3.0 equiv), HCOOK (50.5 mg, 0.6 mmol, 3.0 equiv), 3DPAFIPN (8.0 mg, 6.2 mol%), DABCO (6.8 mg, 30 mol%). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the 4-(2-(1-phenylvinyl)cyclopropyl)-1,1'-biphenyl (44.1 mg, 0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs ($\lambda = 450 - 455$ nm, 3.0 cm -4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) light source and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product 6 as a colorless oil; $R_f = 0.4$ (Petroleum ether : ethyl acetate = 1 : 1); ¹H NMR (400 MHz, CD₃OD) δ 7.39 - 7.17 (m, 11.5H), 7.10 - 7.08 (m, 2H), 5.84 (t, J = 7.4 Hz, 0.33H), 5.58 (t, J = 7.3 Hz, 1H), 3.76 (t, J = 7.7 Hz, 0.34H), 3.57 (t, J = 7.6 Hz, 1H), 3.48 (0.71H, ABq, J = 16.0 Hz), 3.25 (s, 1.9H), 3.01 - 2.94 (m, 0.33H), 2.72 - 2.64 (m, 1.34H), 2.48 - 2.41 (m, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 177.1, 177.1, 175.2, 175.1, 143.6, 140.9, 140.5, 140.2, 137.6, 136.1, 129.8, 129.6, 129.5, 129.5, 129.4, 129.2, 129.1, 129.1, 128.3, 128.2,

128.1, 128.0, 127.0, 52.9, 52.6, 36.7, 34.0. HRMS (ESI) m/z calcd for C₁₉H₁₈NaO₄⁺ (M+Na)⁺ 333.1097; found 333.1096.



Figure S1. 1D selective NOE of 6.

Reactions in the presence of D₂O



An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was

charged with K₃PO₄ (85.5 mg, 0.4 mmol, 2.0 equiv), Cs₂CO₃ (195.5 mg, 0.6 mmol, 3.0 equiv), HCOOK (50.5 mg, 0.6 mmol, 3.0 equiv), 3DPAFIPN (8.0 mg, 6.2 mol%), DABCO (6.8 mg, 30 mol%). The tube was taken out of the glovebox. Subsequently, D₂O (40 mg, 10 equiv), the degassed anhydrous DMSO (2 mL, 0.1 M) and the **1a** (36.05 mg, 0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) light source and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired product **2a_H-DH** (30 mg, 65%). ¹H NMR (400 MHz, CD₃OD) δ 7.27 - 7.23 (m, 8H), 7.19 - 7.13 (m, 2H), 4.50 (t, *J* = 8.0 Hz, 0.51H), 3.05 - 3.03 (m, 2H). ¹³C NMR (100 MHz, CD₃OD) δ 175.7, 145.3, 145.2, 129.5, 128.8, 128.7, 127.4, 48.5, 41.5, 41.4.



An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (85.5 mg, 0.4 mmol, 2.0 equiv), Cs₂CO₃ (195.5 mg, 0.6 mmol, 3.0 equiv), HCOOK (50.5 mg, 0.6 mmol, 3.0 equiv), 3DPAFIPN (8.0 mg, 6.2 mol%), DABCO (6.8 mg, 30 mol%). The tube was taken out of the glovebox. Subsequently, D₂O (80 mg, 20 equiv), the degassed anhydrous DMSO (2 mL, 0.1 M) and the **1a** (36.05 mg, 0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with N₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) light source and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation and the

yield of crude products was determined via ¹H NMR using CH₂Br₂ as internal standard.



Figure S2. Reaction in the presence of D_2O .

The reaction in the absence of HCOOK



An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (85.5 mg, 0.4 mmol, 2.0 equiv), Cs₂CO₃ (195.5 mg, 0.6 mmol, 3.0 equiv), HCOOK (50.5 mg, 0.6 mmol, 3.0 equiv), 3DPAFIPN (8.0 mg, 6.2 mol%), DABCO (6.8 mg, 30 mol%). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the **1a** (36.05 mg, 0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs (λ = 450 - 455 nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the reaction temperature at 25 - 33 °C) light source and stirred at ambient temperature for 24 h. Then, the

mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation and the yield of crude products was determined via ¹H NMR using CH₂Br₂ as internal standard.



Figure S3. The reaction in the absence of HCOOK.

¹³C-labelling experiment using commercially available H¹³COONa

Ph Ph +
$$CO_2$$
 + H¹³COONa standard conditions
1a Ph COOH Ph COOH

An oven-dried Schlenk tube (25 mL) containing a stirring bar was cooled to room temperature. The Schlenk tube was then introduced in a glovebox, where it was charged with K₃PO₄ (85.5 mg, 0.4 mmol, 2.0 equiv), Cs₂CO₃ (195.5 mg, 0.6 mmol, 3.0 equiv), H¹³COONa (40.8 mg, 0.6 mmol, 3.0 equiv), 3DPAFIPN (8.0 mg, 6.2 mol%), DABCO (6.8 mg, 30 mol%). The tube was taken out of the glovebox. Subsequently, the degassed anhydrous DMSO (2 mL, 0.1 M) and the **1a** (36.05 mg, 0.2 mmol) were added into the tube via syringe. Then the tube was evacuated and back-filled with CO₂ for 10 times. Finally, the tube was placed under 30 W blue LEDs ($\lambda = 450 - 455$ nm, 3.0 cm - 4.5 cm away from the LEDs, with fans to keep the

reaction temperature at 25 - 33 °C) light source and stirred for 24 h. Then, the mixture was quenched with H₂O and 2 mL of HCl (2 N), extracted with EtOAc three times. The resulting solution was concentrated by rotary evaporation. The residual was purified by column chromatography on silica gel to give the desired products 2a+(2aa/2ab)+2ac (37 mg, 68%)^{·1}H NMR (400 MHz, CD₃OD) δ 7.34 - 7.19 (m, 5H), 3.49 (s, 1H). ¹³C NMR (100 MHz, CD₃OD) δ 176.6, 174.3, 144.5, 129.9, 128.8, 127.8, 58.5, 49.6, 44.7.



Figure S4. The products were detected by HR-MS.

m/z	Abundance
293.0786	875014
294.0819	388090
295.0851	78504



³COOH ¹³COOH 2ac

2a : (2aa+2ab) : 2ac = 81 : 14 : 5

The Application of the Reaction



An oven-dried reaction tube (10 mL) containing a stirring bar was charged with **4a** (0.2 mmol, 38.8 mg), CH₃NH₂·HCl (0.22 mmol, 14.9 mg) and 4Å Molecular sieves powder (200 mg). After degassed and filled with dry nitrogen repeated three times, Et₃N (0.3 mmol, 30.4 mg) and toluene (2 mL) were added via syringe. The reaction solution was stirred for 0.5 h at room temperature and then heated to reflux for 24 h. The reaction mixture was filtered through a short pad of celite. After removing the solvent under reduced pressure, the residual was purified by silica gel column chromatography to obtain product 7 (25 mg, 66%).^{9 1}H NMR (500 MHz, CDCl₃, TMS) δ 7.39 - 7.36 (m, 2H), 7.33 - 7.30 (m, 1H), 7.22 (d, *J* = 7.0 Hz, 2H), 4.03 (dd, *J* = 9.5, 4.5 Hz, 1H), 3.21 (dd, *J* = 18.5, 9.5 Hz, 1H), 3.07 (s, 1H), 2.84 (dd, *J* = 18.5, 5.0 Hz, 1H).¹³C NMR (126 MHz, CDCl₃) δ 177.97, 176.40, 137.18, 129.33, 128.10, 127.51, 46.07, 37.25, 25.35.

Stern-Volmer fluorescence quenching experiments

Fluorescence spectra were collected on Edinburgh FS5 spectrofluorimeter. Samples for the quenching experiments were prepared in a 4 mL glass cuvette with a septum screw cap. 3DPAFIPN was irradiated at 420 nm and the emission intensity at 550 nm was observed. In a typical experiment, the emission spectrum of a 5.0×10^{-5} M solution of 3DPAFIPN in DMSO was collected.

DABCO: A stock solution of DABCO (56.1 mg, 0.5 mmol) in 5 ml of DMSO was prepared. Then, different amounts of this stock solution were added to a solution of the 3DPAFIPN in DMSO (5.0×10^{-5} M).



Figure S5. Stern-Volmer quenching by DABCO.

HCOOK: A stock solution of HCOOK (42.1 mg, 0.5 mmol) in 10 ml of DMSO was prepared. Then, different amounts of this stock solution were added to a solution of the 3DPAFIPN in DMSO (5.0×10^{-5} M).



1a: A stock solution of **1a** (90.1 mg, 0.5 mmol) in 5 ml of DMSO was prepared. Then, different amounts of this stock solution were added to a solution of the 3DPAFIPN in DMSO (5.0×10^{-5} M).



Figure S7. Stern-Volmer quenching by 1a.

5. References

1. Elisabeth, S.; Tillmann, G. *et al.* A toolbox approach to construct broadly applicable metal-free catalysts for photoredox chemistry: deliberate tuning of redox potentials and importance of halogens in donor–acceptor cyanoarenes. *J. Am. Chem. Soc.* **2018**, *140*, 15353-15365.

2. Ju, T.; Zhou, Y.-Q.; Cao, K.-G. *et al.* Dicarboxylation of alkenes, allenes and (hetero)arenes with CO₂ via visible-light photoredox catalysis. *Nat Catal* **2021**, *4*, 304-311.

3. Kofron, W. G.; Wideman, L. G. Specific synthesis and selective alkylation and condensation of monoesters of substituted succinic acids. *J. Org. Chem.* **1972**, *37*, 555-559.

Mąkosza, M.; Marcinowicz, A. Simple synthesis of arylsuccinic acids. *Synthesis*, 2001, 9, 1311-1312.

5. Gan, Z.-J.; Wang, Y.-H.; Xu, Y.-G.; Guo, T.; Wang, J.; Song, Q. *et al.* Discovery, stereospecific characterization and peripheral modification of $1-(pyrrolidin-1-ylmethyl)-2-[(6-chloro-3-oxo-indan)-formyl]-1,2,3,4-tetrahydroisoqui nolines as novel selective <math>\kappa$ opioid receptor agonists. *Org. Biomol. Chem.* **2015**, *13*, 5656-5673.

6. Manoni, F.; Cornaggia, C.; Murray, J.; Tallona, S. J. Catalytic, enantio- and diastereoselective synthesis of γ-butyrolactones incorporating quaternary stereocentres. *Chem. Commun.* **2012**, *48*, 6502-6504.

7. Padmanabhan, M.; James C.; Thirumurugan A. Dalton Trans. 2008, 21, 2809-2811.

8. Combee, L. A.; Johnson, S. L.; Laudenschlager, J. E.; Hilinski, M. K., Rh(II)-catalyzed nitrene-transfer [5 + 1] cycloadditions of aryl-substituted vinylcyclopropanes. *Org. Lett.* **2019**, *21*, 2307-2311.

9. Mao, Z., Huang, S., Gao, L. *et al.* A novel and versatile method for the enantioselective syntheses of tropane alkaloids. *Sci. China Chem.* **2014**, *57*, 252–264.

6. Copies of ¹H and ¹³C NMR spectra





S36




10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -160 -170 -180 -200 -200 -210 fl (ppa)









10 0 -10 -20 -30 -40 -50 -60 -70 -50 -100 -110 -120 -130 -140 -150 -150 -170 -180 -200 -210 fl (ppa)













10 0 -10 -20 -30 -40 -50 -60 -70 -80 -90 -100 -110 -120 -130 -140 -150 -150 -170 -180 -200 -210 f1 (pps)





















zo 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1(ppn)



















20 210 200 190 160 170 160 150 140 130 120 110 100 90 60 70 60 50 40 30 20 10 0 -10 fl (ppm)



















10 0 -10 -20 -30 -40 -50 -60 -70 -60 -90 -100 -110 -120 -130 -140 -150 -150 -170 -180 -190 -200 -210 fl (gpa)



20 210 200 190 160 170 160 150 140 130 120 110 100 90 50 70 60 50 40 30 20 10 0 -10 fl (ppm)




່ 210 ່ 250 ່ 150 ່ 150 ່ 150 ່ 150 ່ 150 ່ 120 ່ 120 ່ 110 ່ 150 ່ 90 ່ 50 ່ 70 ່ 60 ່ 50 ່ 40 ່ 50 ່ 20 ່ 10 ່ 0 ່ -10 . fl (spm)







180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 fl (ppm)



20 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 fl (ppm)





