# Using waste to treat waste: facile synthesis of hollow carbon nanospheres from lignin for water decontamination

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#### 1. Materials.

The chemicals are used without further purification. Peroxydisulfate (K<sub>2</sub>S<sub>2</sub>O<sub>8</sub>, PDS) was bought from Rhawn Co., Ltd. oxytetracycline (OTC) from Tianjin Heowns Biochemical Co., Ltd., phenol, tetracyclines (TC) and chlortetracycline (CTC) from Shanghai Macklin Biochemical Co., 3aminophenol and sulfamethoxazole (SMX) from Aladdin chemical reagent Co., Ltd. sodium hydroxide (NaOH, AR, 96%, Tianjin Kemiou Chemical Reagent Co., Ltd., ethyl alcohol from Tianli Chemical Reagent Co., Ltd., Tert- butylalcohol was purchased from Aladdin Reagent Co., Ltd. Sodium azide (NaN<sub>3</sub>) from Wuhan geao Chemical Technology Co., Ltd. deionized water was utilized in all the experiments.

#### 2. Characterization

Transmission electron microscopy (TEM) were performed by a JEOL-JEM 2100F microscope operated at 200 kV. Energy Dispersive Spectrometer (EDS) Mapping were performed with a TECNAI G2 F20 microscope operated at 200 kV. The morphologies of the catalysts were observed by scanning electron microscopy (SEM, FEI- Quanta 250 FEG). X-ray diffraction (XRD) analyses were measured on a Bruker AXS-D8 Advanced. X-ray photoelectron spectrometry (XPS) was performed on a ThermoFisher Nexsa. Electron paramagnetic resonance (EPR) spectroscopy was conducted on a Bruker A300. Fourier transform infrared spectrometer (FT-IR) were measured on ThermoFisher-IS50. Raman was performed on Horiba scientific-LabRAM HR evolution.

#### 3. Synthesis of HCNs

First, 0.5 g of dealkali lignin was pyrolyzed at 400 °C, 500 °C resp. 600 °C for 4 h in a tubular furnace, with a ramp rate of 2.5 °C·min<sup>-1</sup>. Then, the reaction system was cooled to 30 °C using a cooling rate of 2.5 °C/min. Next, the residue was obtained and washed with deionized H<sub>2</sub>O resp. EtOH, and it was dried overnight at 60 °C. The final products are labeled HCN-400, HCN-500 and HCN-600, respectively.

#### 4. Experimental Procedures.

In a typical procedure, 5 mg of PDS was added into OTC solution (20 mg/L, 50 mL) in a round-

bottom flask at 30 °C. Then, OTC degradation was activated by adding HCN-500 (12.5 mg). During the reaction, a 2 mL of reaction solution was extracted out at determined intervals. It was filtered by 0.22 μm film, and immediately tested by UV-Vis. In the quenching experiments, TBA, EtOH, NaN<sub>3</sub> and BQ, respectively, was injected into the reaction medium, and the OTC degradation was carried out at the same condition. Cycle experiment: The HCN-500 was collected by the simple filtration, washed for use in the next cycle.

The degradation rate was calculated as:

$$\frac{C_0 - C_t}{C_0} \times 100\%$$

Degradation rate (%) =

where  $C_0$  and  $C_t$  are the initial and final pollutant concentration determined on UV absorbance value .



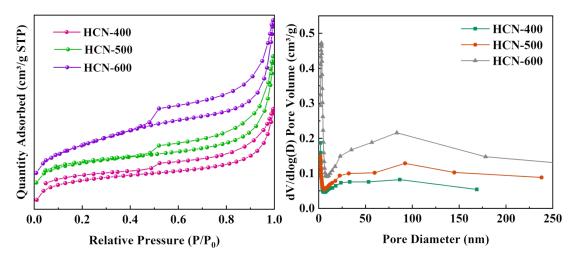


Figure S1. BET of HCNs.

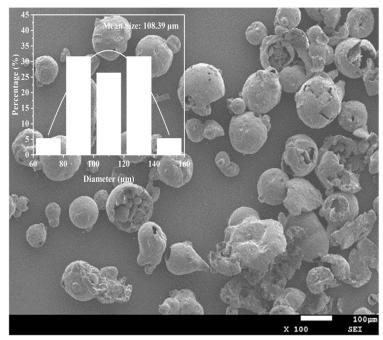


Figure S2. SEM of dealkali lignin

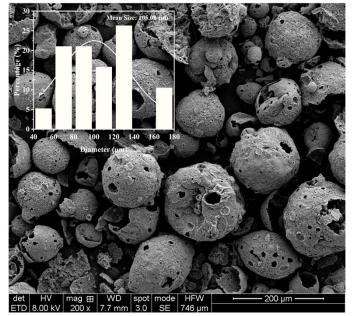


Figure S3. SEM of HCN-400

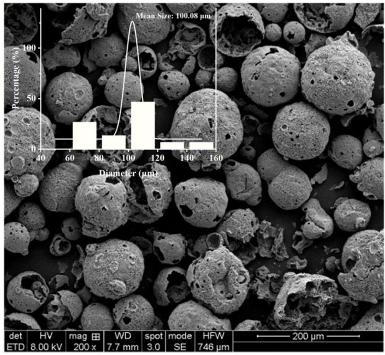


Figure S4. SEM of HCN-500

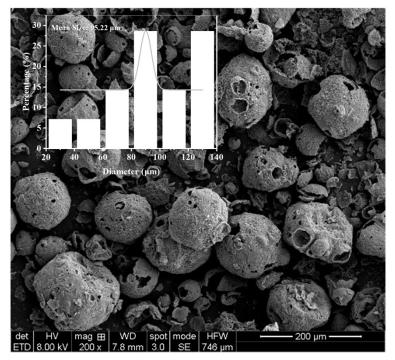
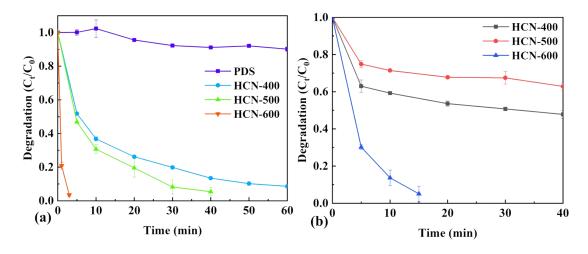
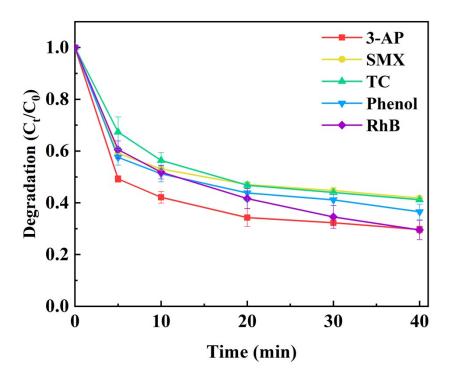


Figure S5. SEM of HCN-600



**Figure S6.** (a) Comparison of OTC degradation catalyzed by HCN-400, HCN-500 and HCN-600. Conditions: 20 mg/L of OTC, 0.1 g/L of PDS and 0.5 g/L of catalyst at 30 °C; (b) Comparison of OTC adsorption by HCN-400, HCN-500 and HCN-600. Conditions: 20 mg/L of OTC and 0.5 g/L of catalyst at 30 °C;



**Figure S7.** The removal of 3-aminophenol (3-AP), sulfamethoxazole (SMX), tetracycline (TC), phenol and Rhodamine B (RhB) over HCN-500/PDS system.

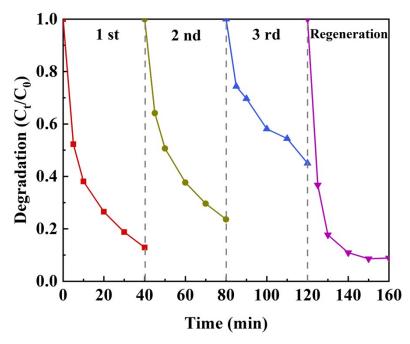


Figure S8. Stability of HCN-500 in OTC degradation.



Figure S9. SEM of reactivated HCN-500

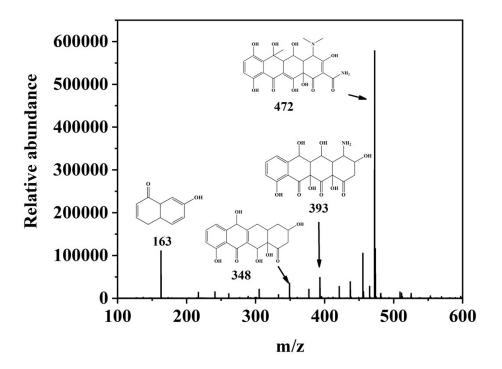


Figure S10. HR-MS of the totally degraded OTC solution

Entry	OTC	Catalyst	Oxidant	Other additive	Degradation efficiency	Time	Ref
1	20 mg/L	HCN-500 (0.5 g/L)	PDS (0. 1 g/L)	-	95%	40 min	This work
2	20 mg/L	BBC <sub>3</sub> (0.3 g/L)	PDS (2 mM)	-	94%	120 min	[31a]
3	5 mg/L	O-Gcn (0.1 g/L)	PMS (1 mM)	-	63.7%	60 min	[31b]
4	20 mg/L	ACN-10 (0.3 g/L)	-	300  W XL ( $\lambda > 420 \text{ nm}$ )	79.3%	60 min	[31c]
5	15 mg/L	N,S-CSs900-10%-OH (0.5 g/L)	PMS (3 mM)	PH=4	95.9%	60 min	[31d]
6	20 mg/L	TCN-5 (0.5 g/L)	-	300  W XL ( $\lambda > 420 \text{ nm}$ )	93%	60 min	[31e]
7	50 mg/L	CB (0.05 g/L)	PDS (1 g/L)	PH=5	60%	40 min	[31f]
8	30 mg/L	$g-C_3N_4(0.5 \text{ g/L})$	-	300  W XL ( $\lambda > 420 \text{ nm}$ )	64.8%	120 min	[31g]
9	50 mg/L	14Fe <sub>3</sub> O <sub>4</sub> -Cs (0.5 g/L)	$H_2O_2$ (10 Mm)	-	90%	120 min	[31h]
10	600 mg/L	nZVI-BC <sub>1000</sub> (0.1 g/L)	PDS (0.5 mM)	-	98.3%	300 min	[31i]
11	20 mg/L	1-B@TBC-600 (0.1 g/L)	PMS (0.5 mM)	-	90%	20 min	[31j]
12	30 mg/L	Co@NC-800 (0.2 g/L)	PMS (0.2 g/L)		91.3%	30 min	[31k]
13	30 mg/L	CN (300 mg/L)	PMS (0.1 g/L)	PH=4.34	69.1%	45 min	[311]
14	20 mg/L	2-EC (0.5 g/L)	H <sub>2</sub> O <sub>2</sub> (10.0 mM)	PH=5	97.1%	120 min	[31m]

### Table S1 Comparison of HCN-500/PDS with other reported catalytic systems for OTC degradation.