Electronic Supplementary Information (ESI)

Significant hydrogen generation via photo-mechanical coupling in

flexible methylammonium lead iodide nanowires

Yucheng Zhang,^a Jiawei Huang,^a Mengya Zhu,^b Zhouyang Zhang,^b Kaiqi Nie,^c Zhiguo Wang,^a Xiaxia Liao,^a Longlong Shu,^a Tingfang Tian,^{*a} Zhao Wang,^{*d} Yang Lu,^e and Linfeng Fei ^{*a}

^a School of Physics and Materials Science, Nanchang University, Nanchang 330031, China

^b Department of Mechanical Engineering, City University of Hong Kong, Kowloon, Hong Kong SAR, China

^c Institute of High Energy Physics, Chinese Academy of Sciences, Beijing 100049, China

^d Hubei Key Laboratory of Micro-& Nanoelectronic Materials and Devices, School of Microelectronics, Hubei University, Wuhan 430062, China

^e Department of Mechanical Engineering, The University of Hong Kong, Hong Kong SAR, China

E-mail: <u>feilinfeng@gmail.com</u> (L. F. Fei); <u>wangzhao@hubu.edu.cn</u> (Z. Wang); <u>tftian@ncu.edu.cn</u> (T. F. Tian)

Contents of Supplementary Information:

Methods	S2
Synthesis of MAPbI $_3$ NWs	S2
Synthesis of MAPbI ₃ NPs	S3
Synthesis of MAPbI $_3$ SCs	S3
General characterizations	S4
In situ PFM test	S4
In situ bending test	S4
Finite element simulations	
Hydrogen evolution test	S5
Calculation of the catalytic efficiency	S6
Supplementary Figures	S7
References	S24

Methods

Synthesis of MAPbl₃ NWs

461.1 mg Pbl₂ powder was added into 1 mL N,N-dimethylformamide (DMF) before the system was heated to 70 °C on a hot plate for 1 h, while in another reactor, 159 mg MAI (C₄H₆IN) was added into 20 mL isopropyl alcohol (IPA) and vigorously stirred at room temperature for 30 min under an argon atmosphere. After that, the Pbl₂ solution was gradually added to the MAI solution. The mixture was vigorously stirred under an atmosphere of argon for 30 minutes; and finally, MAPbl₃ NWs were obtained by centrifuging the mixture.

Synthesis of MAPbl₃ NPs

MAI and PbI₂ of 1:1 molar ratio were dissolved in DMF in a vial and stirred for 2 hours before the formation of a yellow solution. MAPbI₃ NPs were synthesized by adding the precursor solution drop by drop to toluene. The solvent was then removed at a centrifugal rate of 8000 rpm for 5 minutes. The product was then collected after drying at 70 °C in a vacuum oven.

Synthesis of MAPbl₃ SCs

2.0 M Pbl₂ and 2.0 M MAI were dissolved in a mixed solvent of DMSO–DMF (7:3 by volume; DMSO: dimethyl sulfoxide) in a vial. The precursor solution was then filtered with a polytetrafluoroethylene (PTFE) filter before heating on an oil bath at 70 °C for a period of time in order to produce MAPbl₃ SCs. For example, small SCs (average size about 1.5 mm) can be harvested by maintaining 5 ml precursor solution at this temperature for 5 hours.

General characterizations

XRD patterns were obtained on a Rigaku D/Max diffractometer with a Cu K_{α} radiation. SEM images were obtained on a Thermos Scientific Scios2 dual beam electron microscope operated at 20 kV. TEM images and EDS elemental mappings were obtained on a 300 kV double aberration-corrected Titan Themis Z electron microscope. UPS measurements were conducted in the 4B9B beamline at Beijing Synchrotron Radiation Facility with He I source (21.2 eV). The spectra were collected using a Scienta R4000 spectrometer with an energy resolution better than 0.1 eV. For the work function measurements, all samples were biased at –5 eV to acquire the low cutoff of secondary electron. The photon energy was calibrated by the Au $4f_{7/2}$ peak with a binding energy of 84 eV. UV–Vis DRS spectra were recorded on a Hitachi UH-4150 UV–Vis spectrophotometer with spectral-grade BaSO₄ as a reference.

In situ PFM test

In situ PFM test was performed on a NT-MDT scanning probe microscopy (SPM, NT-MDT Ntegra upright) with a heating stage (SU045NTF, NT-MDT, Russia) which was equipped with a Pt resistor to achieve its temperature stability at 0.05 °C. By moving the probe tip of PFM to the middle region of the scanned NW, the displacementvoltage butterfly curve and the phase curve were obtained upon applying a voltage in a certain range at each temperature.

In situ bending test

A batch of NWs was dispersed in isopropanol by stirring and ultrasonication to get a dilute suspension, which was then drop-casted onto a copper grid. Using the Model 450PM Manual Probe Station platform, an appropriate NW was picked with the help of a light microscopy, which was then experienced a force on one tip by a fine tungsten needle.

Finite element simulations

Finite element simulations were performed using a COMSOL software. The length of MAPbI₃ NW was set to 30 μ m, and the diameter was set to 300 nm. The elastic modulus and Poisson's ratio of the MAPbI₃ NW were set to 20 GPa and 0.4, respectively. The NW was simply supported at both ends, and the axial displacement constraint was set at the bottom to ensure that the NW can bend freely. The bending of the NW was realized by applying bending moments at both ends. The bending moment was set to 10^{-12} N m in order to approach the level of strain gradient in our experiments.

Hydrogen evolution test

First, a saturated HI solution was prepared as follows. An H-shaped cell was employed in this preparation to separate the anode and cathode compartment with a Nafion 117 membrane placed between them. 10 mL of 57 wt% HI solution was added in each compartment. 2.97 g PbI₂ was dissolved in the HI at the cathode compartment and stirred at 60 °C before slowly adding 1.025 g MAI. Carbon fiber papers with an active area of 4 cm² were utilized as both the working and counter electrodes. After Ar gas purging into two compartments for 15 min, –800 mA of current was applied on the working electrode for 2 h (**Figure S16**). Thereafter, the saturated hydrogen iodide solution was obtained in the cathode compartment.

Subsequently, the hydrogen production rates were determined from a homemade top-lighting reactor (30 ml; with water cooling). A 300 W Xe-lamp (PLS-SXE 300/300UV, Beijing Perfect Light, China) with a 420 nm cut-off filter was used as a visible light source, and mechanical stimulation was introduced to the reactor by a magnetic stirrer at 300 rpm. Typically, 50 mg MAPbl₃ NWs was dispersed into 5 ml saturated HI solution for a single hydrogen evolution test. The evolved H₂ was

quantitatively analyzed by a gas chromatograph (Shimadzu GC-2014) with Ar as the carrier gas.

Calculation of the catalytic efficiency

By utilizing the concentration of HI solution and the computed free energy associated with splitting the solution, we can determine the potential energy value, denoted as E_{HI} , which equals to 0.33 eV. Utilizing this potential energy value, we can evaluate the solar conversion efficiency of our photomechanocatalytic reaction by utilizing the following equation.^{1, 2}

Solar HI splitting efficiency (%) =

$$\frac{\left[\text{Evolved H}_{2} \text{ (mol)} \times 6.02 \times 10^{23} \times 2 \times 0.330 \text{ (eV)} \times 1.6 \times 10^{-19}\right]}{\left[\text{Psol (W cm}^{-2}) \times \text{Area (cm}^{2}) \times \text{time (s)}\right]} \times 100\%$$

For instance, in the case of the MAPbI₃ NWs deposited, 75.65 μ mol of H₂ could be generated after 2 h of light irradiance at 100 mW cm⁻². Because our light irradiation area was 0.25 cm², the solar HI splitting efficiency could be calculated as 2.67%.

Supplementary Figures



Fig. S1 Statistics of lengths and diameters for $MAPbI_3$ NWs. (a) Length statistics of $MAPbI_3$ NWs.(b)Diameterstatisticsof $MAPbI_3$ NWs.



Fig. S2 Morphological characterization and elemental distributions of MAPbI₃ NWs. (a) TEM image of a NW. (b-f) STEM (i.e., scanning transmission electron microscopy) image and the EDS elemental mappings for Pb, N, I, and C.



Fig. S3 UPS characterizations for (a) $MAPbI_3 SC$ and (b) $MAPbI_3 NPs$.



Fig. S4 Tauc plots for (a) $MAPbI_3 SC$ and (b) $MAPbI_3 NPs$.



Fig. S5 Comparison of the structures of optical bandgaps (versus Fermi level) for MAPbI₃ SC, NWs and NPs.



Fig. S6 In situ PFM response of NWs at a temperature range from RT to 100 °C. (a) The corresponding PFM morphology image for a MAPbI₃ NW. (b) PFM amplitude response for a MAPbI₃ NW at 40 °C. (c) PFM amplitude response for a MAPbI₃ NW at 60 °C. (d) PFM amplitude response for a MAPbI₃ NW at 60 °C. (e) PFM amplitude response for a MAPbI₃ NW at 100 °C. (f) Electromechanical coupling coefficients for the above NW from RT to 100°C.



Fig. S7 The phase-voltage curves of a MAPbI₃ NW tested by PFM at (a) 40 °C, (b) 60 °C, (c) 80 °C, and (d) 100 °C.



Fig. S8 PFM amplitude response of a commercial $LiNbO_3$ sample.



Fig. S9 Bright-field TEM image for $MAPbI_3 NWs$, in which the bending NW can be observed.



Fig. S10 Optical image series showing a MAPbI₃ NW experienced multiple bending-recovery processes before it was broken.



Fig. S11 GC data for hydrogen gas production from the decomposition of saturated HI using MAPbI₃ NWs under photomechanocatalytic, mechanocatalytic, and photocatalytic conditions. (a) MAPbI₃ NWs decompose saturated HI under a photomechanocatalytic condition. (b) MAPbI₃ NWs decompose saturated HI under a mechanocatalytic condition. (c) MAPbI₃ NWs decompose saturated HI under a photocatalytic condition.



Fig. S12 $MAPbI_3$ NPs decompose saturated HI under a photomechanocatalytic condition.



Fig. S13 HI splitting by the NWs in a saturated HI solution at various conditions. (a) Mechanocatalytic HI splitting reaction of the saturated solution in the presence and absence of MAPbI₃ NWs. (b) Photocatalytic HI splitting reaction of the saturated solution in the presence and absence of MAPbI₃ NWs. (c) Photomechanocatalytic HI splitting reaction of the saturated solution in the presence of MAPbI₃ NWs.



Fig. S14 XRD patterns for MAPbI_3 NWs before and after the catalytic reaction.



Fig. S15 Structural characterization of NWs before and after the catalytic reaction. (a) SEM image of MAPbI₃ NWs at 0 min. (b) SEM image of MAPbI₃ NWs after 60 min of catalytic reaction. (c) SEM of MAPbI₃ NWs at 120 min of catalytic reaction.



Fig. S16 Photos for the electrical reduction process of saturated HI.

Catalyst	Light source	Precious	Catalytic	Activity	Year
		metal	Mechanism	$[H_2 \ \mu mol \ g^{-1} \ h^{-1}]$	[Reference]
MAPbl ₃ NWs	300 W Xe lamp		Photo-	756 5	This work
	(λ ≥ 420 nm)		mechano-	10.5	THIS WORK
MAPbl ₃ /Pt	Solar simulator	Pt	Photo-	57.0	2016 ¹
	(λ > 475nm)				
MAPbl ₃ /Pt/TiO ₂	300 W Xe lamp	Pt	Photo-	5293	2018 ³
	$(200 \text{ mW cm}^2,$				
MAPbBr _{3-x} l _x /Pt	300 W Xe lamp	Pt	Photo-	2604	2018 ⁴
	$(\lambda > 420 \text{ nm})$				
MAPbl ₃ /rGO	300 W Xe lamp		Photo-	939	20185
	(120 mW cm ⁻² ,				
	λ ≥ 420 nm)				
MABi ₂ I ₉ /Pt	300 W Xe lamp	Pt	Photo-	169	2019 ⁶
	(λ ≥ 400 nm)				
MAPbl ₃ /black phosphorus	Xe lamp (300		Photo-	3742	20197
	mW cm⁻², λ ≥				
	420 nm)				
Pt/Ta ₂ O ₅ /MAPbB r ₃ /PEDOT:PSS	300 W Xe lamp	Pt	Photo-	619	2019 ⁸
	(150 mW cm ⁻² ,				
MAPbl ₃	$\Lambda \ge 420 \text{ nm}$		Photo-	68.4	2019 ⁹
	$(100 \text{ mW} \text{ cm}^{-2})$				
MAPbl ₃	(100 mW cm)		Piezo-	44.2	2019 ⁹
	(100 mW cm^{-2})				
MAPbl ₃	500 W Xe lamp	1	Piezophoto-	460.7	2019 ⁹
	(100 mW cm ⁻²)				
MAPbl ₃ /Ni ₃ C	300 W Xe lamp		Photo	2362	2019 ¹⁰
	(λ > 420nm)				
MAPb(Br _{0.1} I _{0.9}) ₃ / Pt	300 W Xe lamp	Pt	Photo-	3348	2019 ¹¹
	(λ ≥ 420 nm)				
MAPbl ₃ /CoP	150 W Xe lamp		Photo-	785.9	2020 ¹²
	(λ > 420nm)			026	202013
CS3BI2xSD2-2xI9/Pt	Solar simulator	Ρt	Photo-	926	202013
MAPbl ₃ /Pt/C	דה אה רבה ומנווט (380 <צ<280		Photo-	685	2020 ¹⁴
	nm)				
Cs ₃ Bi ₂ I ₉	450 W Xe lamp		Photo-	225	
	(150 mW cm ⁻²)				2022 ¹⁵

Table. S1 A summary of previously reported typical catalysts and their performancein the hydrogen evolution reaction from HI decomposition.

References

- 1. S. Park, W. J. Chang, C. W. Lee, S. Park, H.-Y. Ahn and K. T. Nam, Photocatalytic hydrogen generation from hydriodic acid using methylammonium lead iodide in dynamic equilibrium with aqueous solution, *Nat. Energy*, 2016, **2**, 16185.
- W. Guan, Y. Li, Q. Zhong, H. Liu, J. Chen, H. Hu, K. Lv, J. Gong, Y. Xu, Z. Kang, M. Cao and Q. Zhang, Fabricating MAPbl₃/MoS₂ composites for improved photocatalytic performance, *Nano Lett.*, 2021, **21**, 597-604.
- X. Wang, H. Wang, H. Zhang, W. Yu, X. Wang, Y. Zhao, X. Zong and C. Li, Dynamic interaction between methylammonium lead lodide and TiO₂ nanocrystals leads to enhanced photocatalytic H₂ evolution from HI splitting, ACS Energy Lett., 2018, **3**, 1159-1164.
- Y. Wu, P. Wang, Z. Guan, J. Liu, Z. Wang, Z. Zheng, S. Jin, Y. Dai, M.-H. Whangbo and B. Huang, Enhancing the photocatalytic hydrogen evolution activity of mixed-halide perovskite CH₃NH₃PbBr_{3-x}I_x achieved by bandgap funneling of charge carriers, ACS Catal., 2018, 8, 10349-10357.
- Y. Wu, P. Wang, X. Zhu, Q. Zhang, Z. Wang, Y. Liu, G. Zou, Y. Dai, M.-H. Whangbo and B. Huang, Composite of CH₃NH₃Pbl₃ with reduced graphene oxide as a highly efficient and stable visible-light photocatalyst for hydrogen evolution in aqueous HI solution, *Adv. Mater.*, 2018, **30**, 1704342.
- Y. Guo, G. Liu, Z. Li, Y. Lou, J. Chen and Y. Zhao, Stable lead-free (CH₃NH₃)₃Bi₂I₉ perovskite for photocatalytic hydrogen generation, ACS Sustainable Chem. Eng., 2019, 7, 15080-15085.
- 7. R. Li, X. Li, J. Wu, X. Lv, Y.-Z. Zheng, Z. Zhao, X. Ding, X. Tao and J.-F. Chen, Fewlayer black phosphorus-on-MAPbI₃ for superb visible-light photocatalytic hydrogen evolution from HI splitting, *Appl. Catal., B*, 2019, **259**, 118075.
- H. Wang, X. Wang, R. Chen, H. Zhang, X. Wang, J. Wang, J. Zhang, L. Mu, K. Wu,
 F. Fan, X. Zong and C. Li, Promoting photocatalytic H₂ evolution on organic– inorganic hybrid perovskite nanocrystals by simultaneous dual-charge transportation modulation, ACS Energy Lett., 2019, 4, 40-47.
- M. Wang, Y. Zuo, J. Wang, Y. Wang, X. Shen, B. Qiu, L. Cai, F. Zhou, S. P. Lau and Y. Chai, Remarkably enhanced hydrogen generation of organolead halide perovskites via piezocatalysis and photocatalysis, *Adv. Energy Mater.*, 2019, 9, 1901801.
- Z. Zhao, J. Wu, Y.-Z. Zheng, N. Li, X. Li and X. Tao, Ni₃C-decorated MAPbl₃ as visible-light photocatalyst for H₂ evolution from HI splitting, *ACS Catal.*, 2019, 9, 8144-8152.

- Z. Zhao, J. Wu, Y.-Z. Zheng, N. Li, X. Li, Z. Ye, S. Lu, X. Tao and C. Chen, Stable hybrid perovskite MAPb(I_{1-x}Br_x)₃ for photocatalytic hydrogen evolution, *Appl. Catal., B*, 2019, **253**, 41-48.
- 12. C. Cai, Y. Teng, J.-H. Wu, J.-Y. Li, H.-Y. Chen, J.-H. Chen and D.-B. Kuang, In situ photosynthesis of an MAPbl₃/CoP hybrid heterojunction for efficient photocatalytic hydrogen evolution, *Adv. Funct. Mater.*, 2020, **30**, 2001478.
- 13. G. Chen, P. Wang, Y. Wu, Q. Zhang, Q. Wu, Z. Wang, Z. Zheng, Y. Liu, Y. Dai and B. Huang, Lead-free halide perovskite $Cs_3Bi_{2x}Sb_{2-2x}I_9$ (x \approx 0.3) possessing the photocatalytic activity for hydrogen evolution comparable to that of (CH₃NH₃)PbI₃, *Adv. Mater.*, 2020, **32**, 2001344.
- 14. F. Wang, X. Liu, Z. Zhang and S. Min, A noble-metal-free MoS₂ nanosheetcoupled MAPbI₃ photocatalyst for efficient and stable visible-light-driven hydrogen evolution, *Chem. Commun.*, 2020, **56**, 3281-3284.
- S. P. Chaudhary, S. Bhattacharjee, V. Hazra, S. Shyamal, N. Pradhan and S. Bhattacharyya, Cs₃Bi₂I₉ nanodiscs with phase and Bi_(III) state stability under reductive potential or illumination for H₂ generation from diluted aqueous HI, *Nanoscale*, 2022, **14**, 4281-4291.
- 16. T. Zhang, Y. Cai, Y. Lou and J. Chen, 1T-2H MoSe₂ modified MAPbl₃ for effective photocatalytic hydrogen evolution, *J. Alloys Compd.*, 2022, **893**, 162329.