

Supporting Information

A benzimidazole-derived fluorescent chemosensor for Cu(II)-selective turn-off and Zn(II)-selective ratiometric turn-on detection in aqueous solution

Wan-Yu Zhu¹, Kai Liu^{1*}, Xuan Zhang^{1,2*}

¹*Key Laboratory of Science and Technology of Eco-Textiles, Ministry of Education, College of Chemistry and Chemical Engineering, Donghua University, Shanghai 201620, China;* ²*National Innovation Center of Advanced Dyeing & Finishing Technology, Tai'an, Shandong 271000, China;*

* Corresponding Author. E-mail address: kailiu@dhu.edu.cn (KL); xzhang@dhu.edu.cn (XZ)

Table S1 The spectral data of BBMP in various solvents

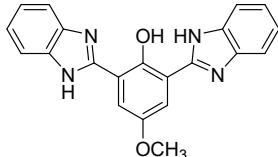
Compound	Solvent	λ_{Abs} (nm)	λ_{Flu} (nm)	Φ_F
	DEE	296/372	547	0.677
	CHCl ₃	298/373	548	0.876
	THF	298//372	548	0.730
	DMSO	298/373	546	0.9900
	DMF	298/373/437	473 (sh.)/543	0.790
	MeOH	296/372/425	452/539	0.689
	NaOH	294/407	491	0.818

Table S2 Comparison of the sensing performance with previously reported Cu²⁺/Zn²⁺ sensors

Solution media	Ratiometric detection available	Linear range (μM)	Detection limit (μM)	Reference
DMSO/H ₂ O (4:1)	no	Cu ²⁺ : 0-15	Cu ²⁺ : 0.023	9
		Zn ²⁺ : 0-20	Zn ²⁺ : 0.068	
THF/H ₂ O (5:95)	yes	Cu ²⁺ : 0-10	Cu ²⁺ : 0.23	10
		Zn ²⁺ : 0-10	Zn ²⁺ : 1.8	
DMSO	no	Cu ²⁺ : no data	Cu ²⁺ : 5.25	11
		Zn ²⁺ : no data	Zn ²⁺ : 29.3	
DMSO/H ₂ O (4:1)	no	Cu ²⁺ : 1-10	Cu ²⁺ : 1.46	16
		Zn ²⁺ : 2-20	Zn ²⁺ : 0.035	
DMF	no	Cu ²⁺ : no data	Cu ²⁺ : 0.18	18
		Zn ²⁺ : no data	Zn ²⁺ : 0.43	
MeCN	no	Cu ²⁺ : no data	Cu ²⁺ : 1.54	19
		Zn ²⁺ : no data	Zn ²⁺ : 3.3	
DMSO/H ₂ O (9:1)	yes	Cu ²⁺ : no data	Cu ²⁺ : 0.001	27
		Zn ²⁺ : no data	Zn ²⁺ : 0.94	
THF/H ₂ O (7:3)	yes	Cu ²⁺ : 0-5	Cu ²⁺ : 0.16	This work
		Zn ²⁺ : 0-10	Zn ²⁺ : 0.1	

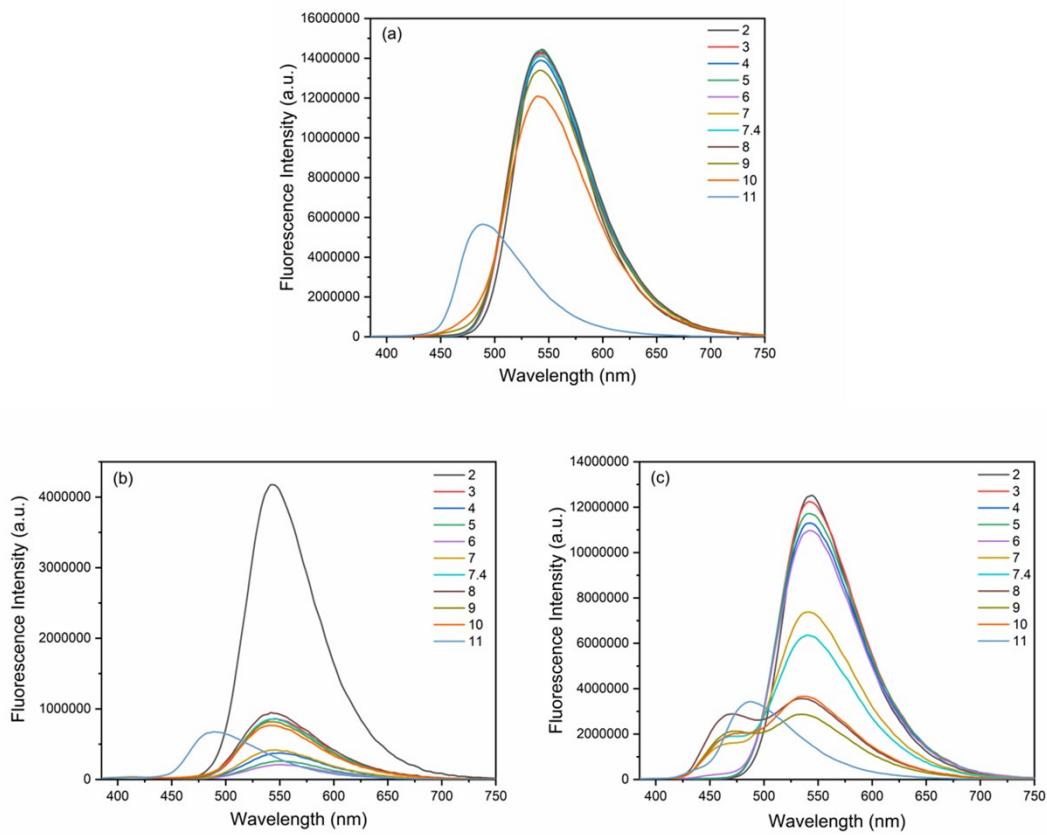


Fig. S1 Fluorescence spectra of probe BBMP (a), BBMP with 1 equiv. Cu²⁺ (b) and BBMP with 1 equiv. Zn²⁺ (c) at different pH values in Tris-HCl buffer. $\lambda_{\text{ex}} = 370 \text{ nm}$.

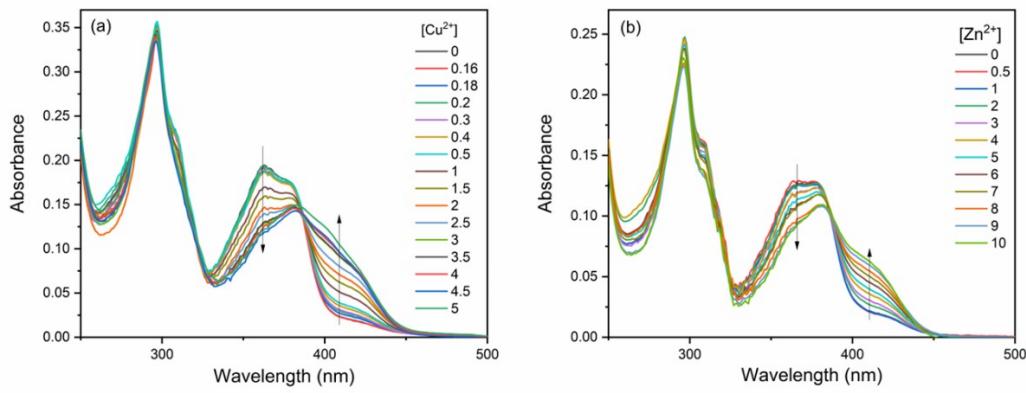


Fig. S2 UV-vis absorption titration spectra of BBMP (10 μM) upon addition of different concentration of Cu²⁺ (a) and Zn²⁺ (b) in aqueous THF solution (7:3 (v/v) THF/Tris-HCl, pH = 7.4).

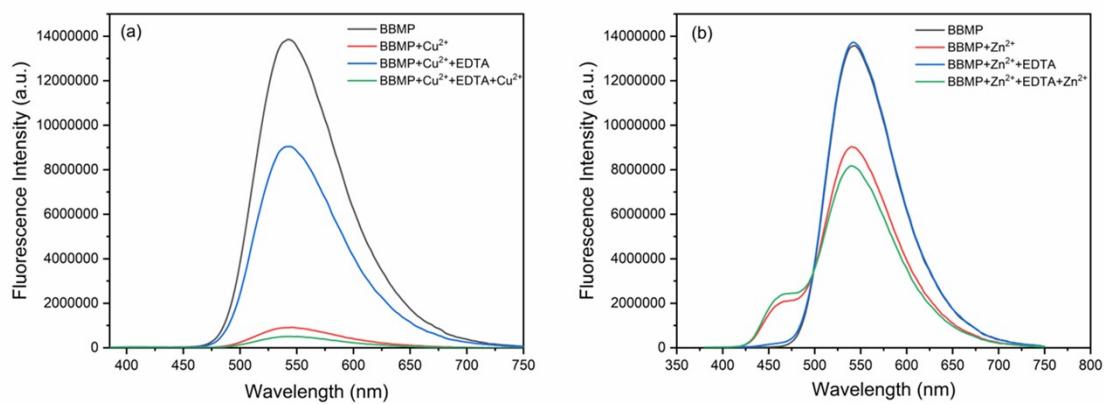
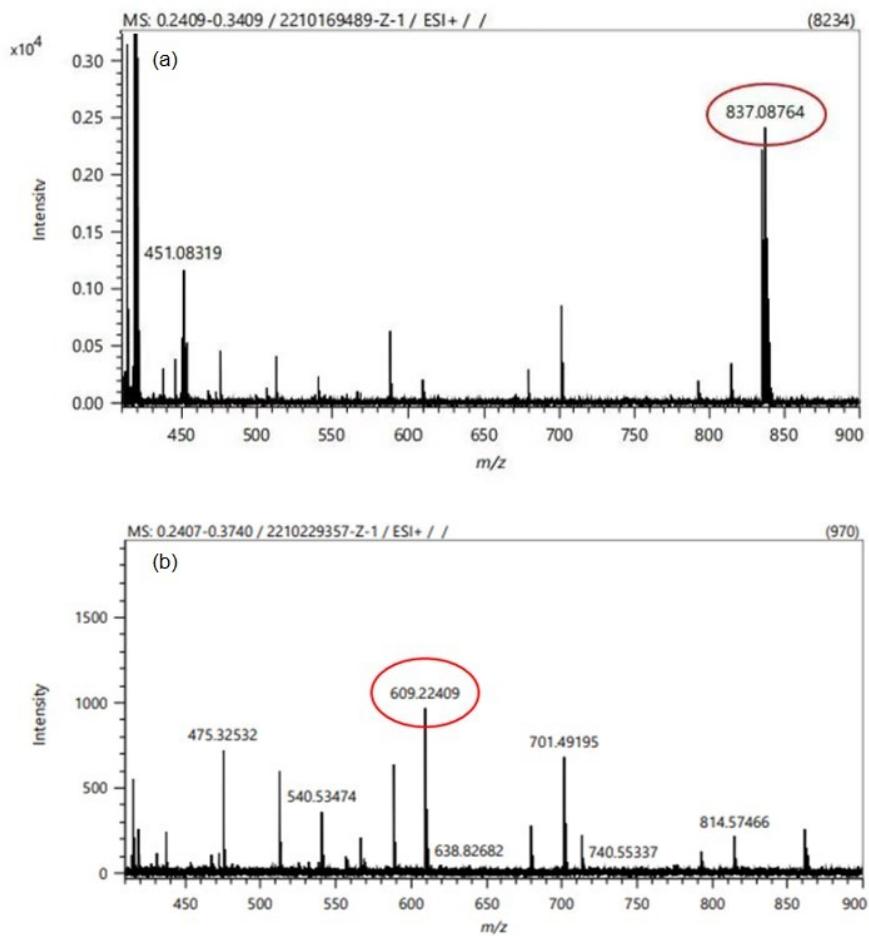


Fig. S3 Fluorescence intensity of BBMP (10 μ M) upon the alternate addition of Cu²⁺ (a), Zn²⁺ (b) and EDTA in aqueous THF solution (7:3 (v/v) THF/Tris-HCl, pH = 7.4).



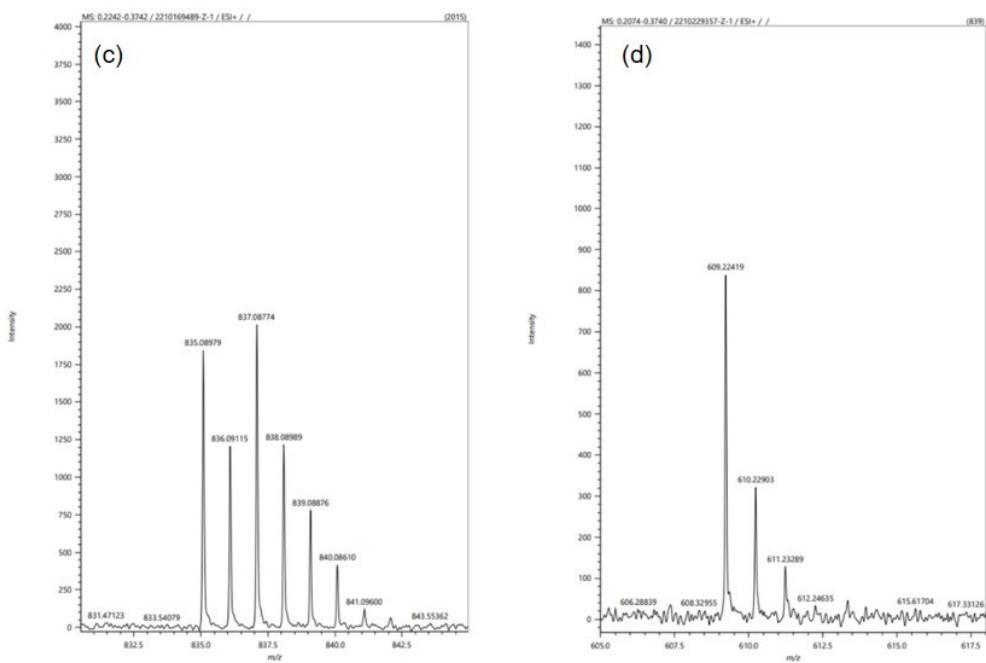


Fig. S4. ESI-MS spectra of the mixture of BBMP with Cu²⁺ (a) and Zn²⁺ (b) in MeOH, where the target peaks associated with complexes were labeled with red circle. ESI-MS spectra of near the target peak of the mixture of BBMP with Cu²⁺ (c) and Zn²⁺ (d).

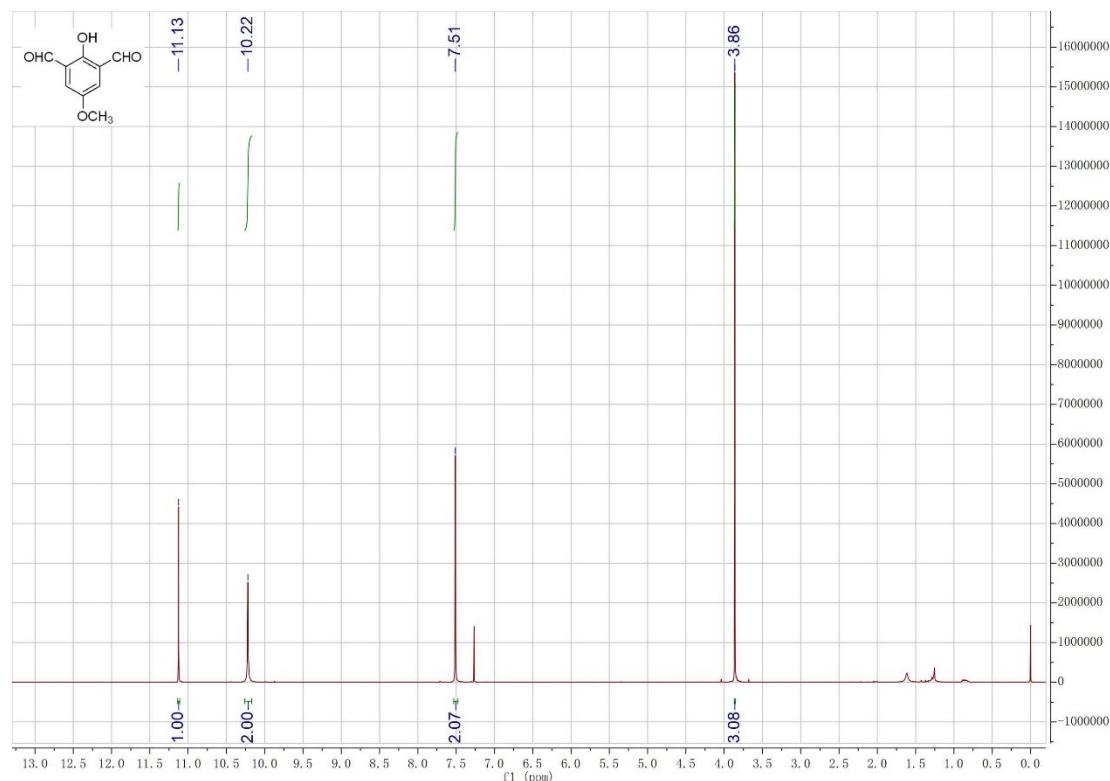


Fig. S5. ¹H NMR spectra of 2,6-diformyl-4-methoxyphenol.

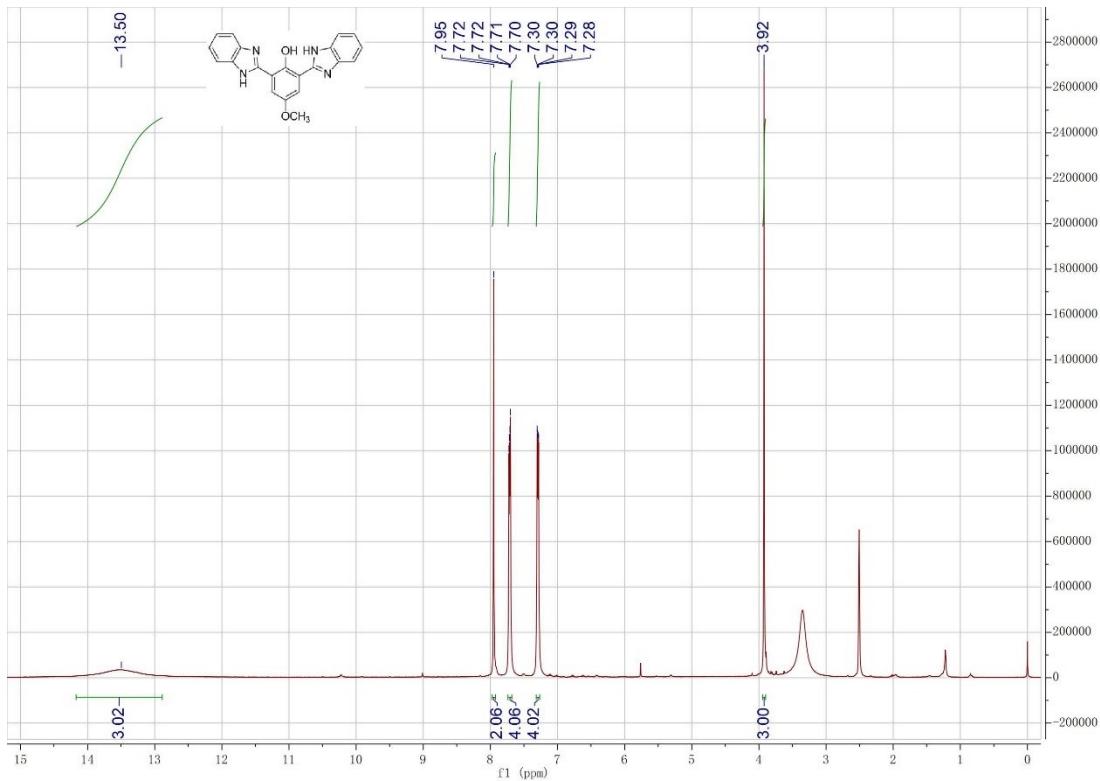


Fig. S6. ¹H NMR spectra of BBMP.

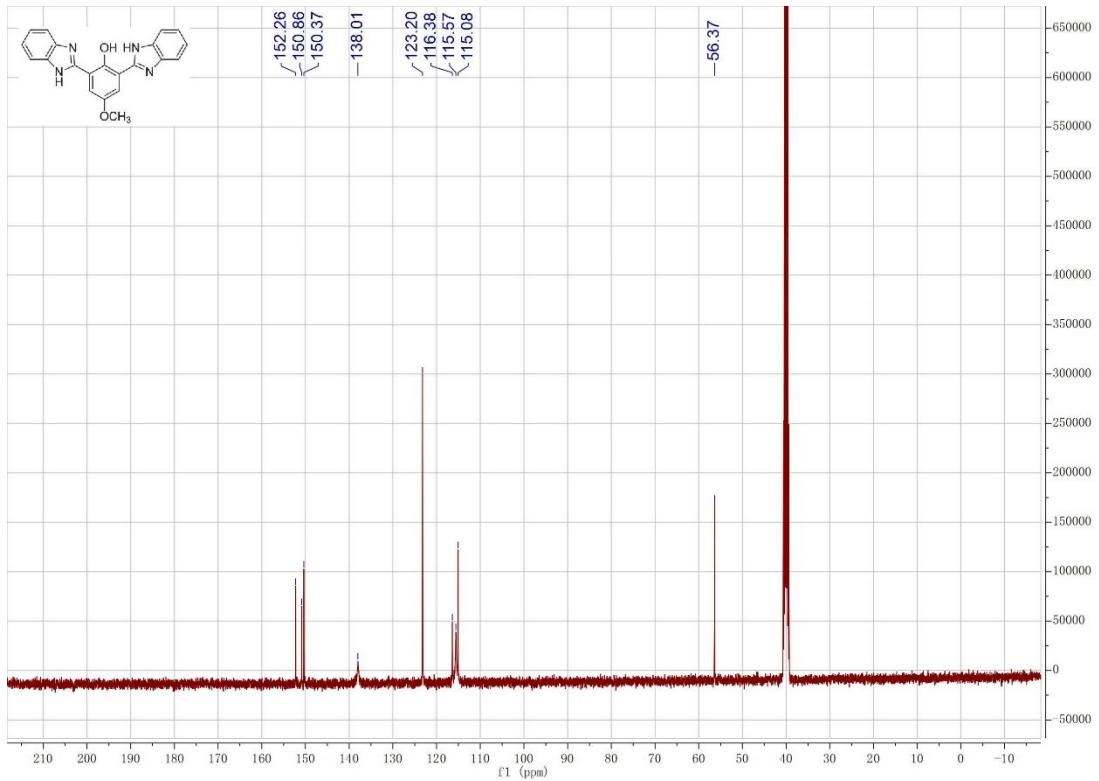


Fig. S7. ¹³C NMR spectra of BBMP.

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Not active			Set Dry Heater	250 °C
Scan Begin	100 m/z	Set Capillary	4000 V	Set Dry Gas	4.0 l/min
Scan End	1000 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste

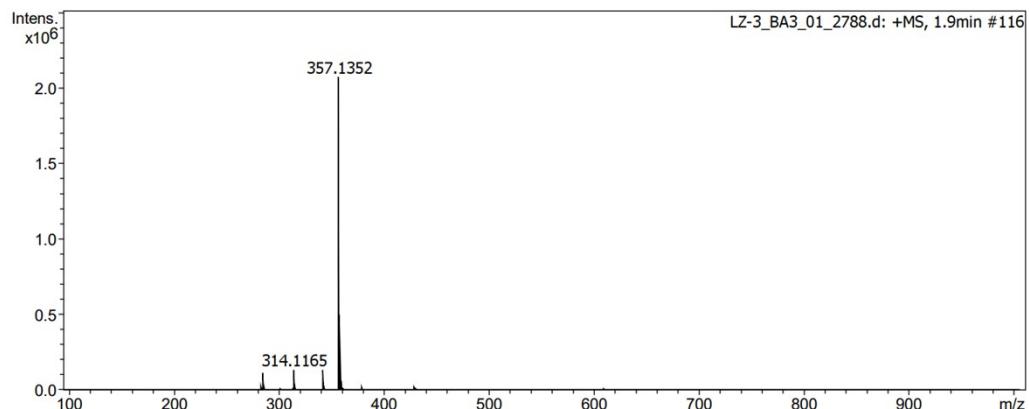


Fig. S8. HRMS spectra of BBMP.