

## Supplementary Information

### Archetypical 2D Sheet-Like $\text{Cu}_2\text{MoS}_4$ for All-Solid-State Symmetric Pseudocapacitors with Ultra-Steady Performance Efficiency

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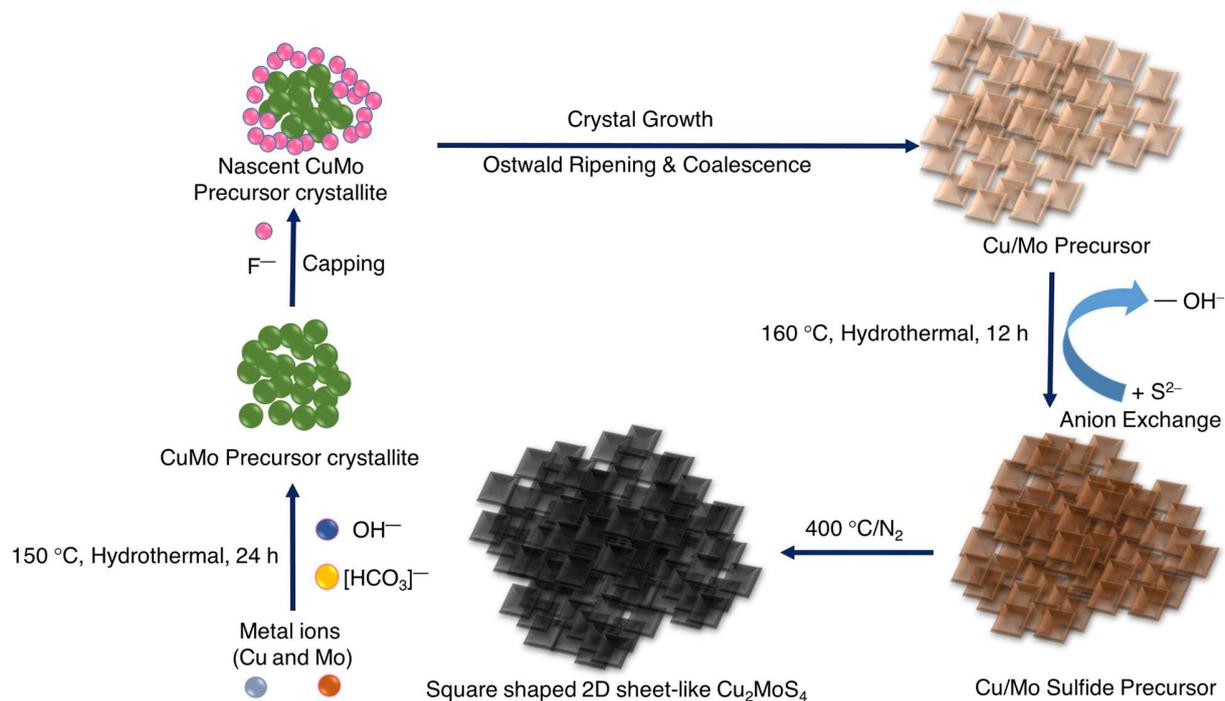
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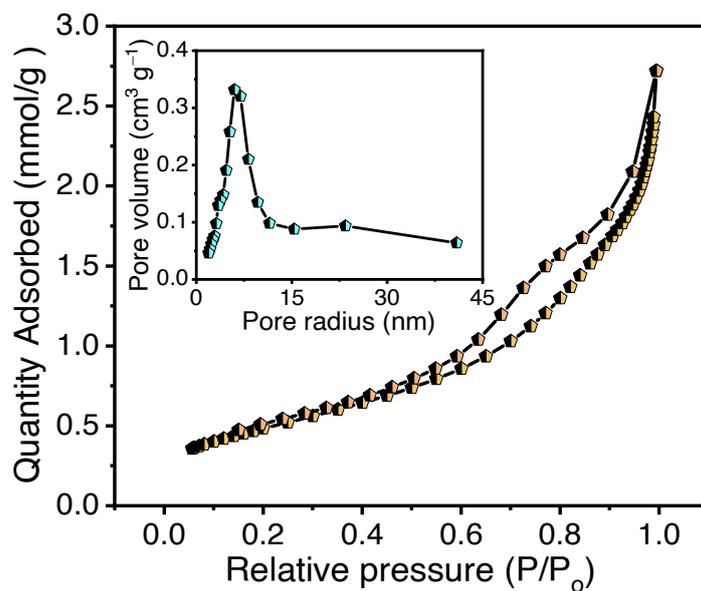
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**Table S1.** Attributions of the peaks in the survey XPS spectrum of  $\text{Cu}_2\text{MoS}_4$ .

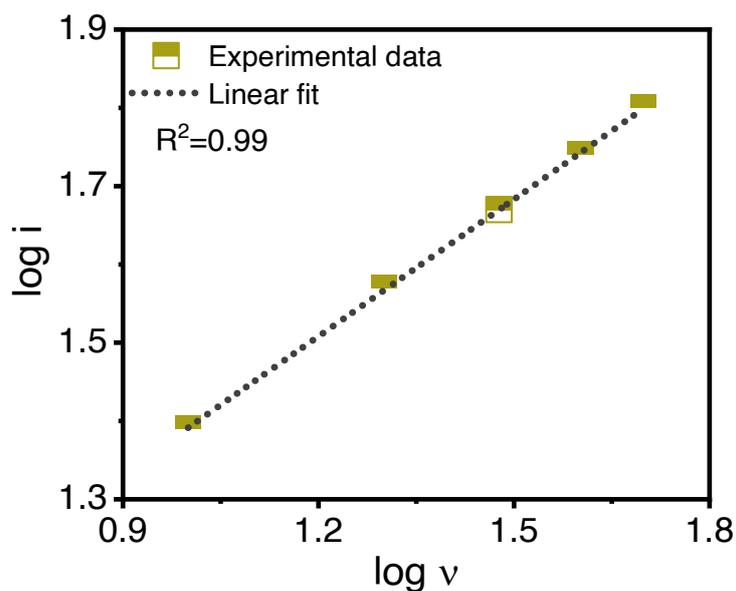
Sl. No.	Binding Energy (eV)	Attributions
1	952.30	Cu 2p <sub>1/2</sub>
2	932.50	Cu 2p <sub>3/2</sub>
3	565.00	Cu 3p
4	531	O 1s
5	506	Mo 3s
6	411	Mo 3p
7	392	Mo 3p
8	284	S 2s
9	228.6	Mo 3d
10	161	S 2s
11	120	Cu 3s
12	74.25	Cu 3p
13	37.26	Mo 4p



**Scheme S1.** Plausible growth mechanism and formation of square shaped 2D sheet-like Cu<sub>2</sub>MoS<sub>4</sub>.<sup>s1</sup>



**Fig. S1** BET N<sub>2</sub> adsorption-desorption isotherm of Cu<sub>2</sub>MoS<sub>4</sub>; inset shows the BJH pore size distribution plot of Cu<sub>2</sub>MoS<sub>4</sub>.

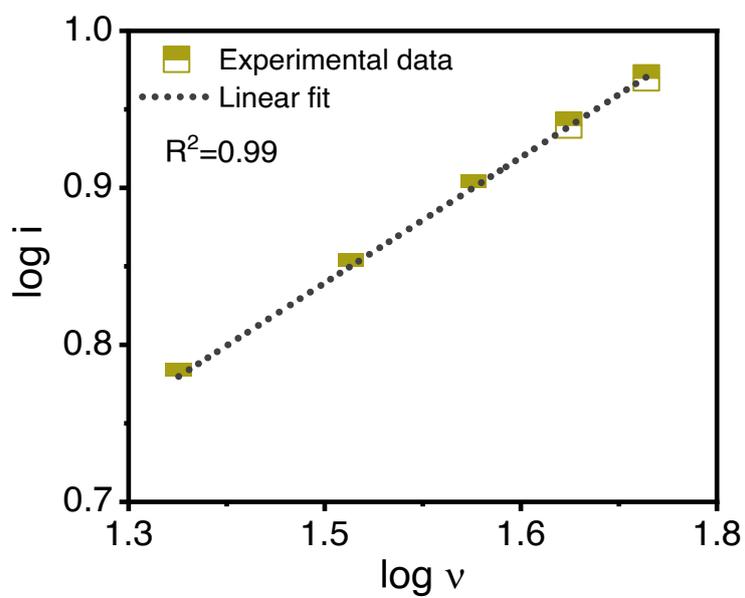


**Fig. S2**  $\log i$  vs.  $\log v$  plot of  $\text{Cu}_2\text{MoS}_4$ , when studied as the positive electrode material.

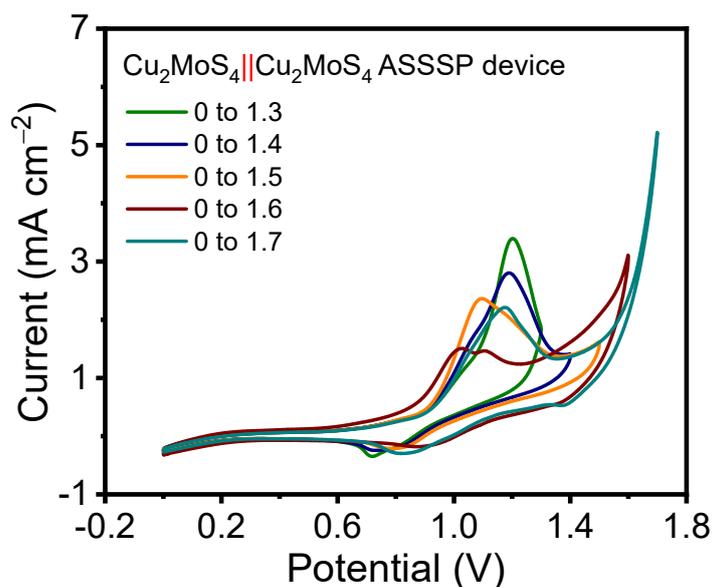
**Table S2:** Comparison of the specific capacitance of  $\text{Cu}_2\text{MoS}_4$  with the reported Cu and Mo based oxides/sulfides in 3 electrode set up.

Sl. No.	Sample Name	Specific Capacitance @ Current density	References
1	$\text{Cu}_2\text{O}$	144 $\text{F g}^{-1}$ @1 $\text{A g}^{-1}$	s2
2	RGO/ $\text{Cu}_2\text{O}$ /Cu	98 $\text{F g}^{-1}$ @1 $\text{A g}^{-1}$	s3
3	$\text{Cu}_2\text{O}$ -GN	416 $\text{F g}^{-1}$ @1 $\text{A g}^{-1}$	s4
4	CNT@CuS	122 $\text{F g}^{-1}$ @1.2 $\text{A g}^{-1}$	s5
5	H-CuS	536 $\text{F g}^{-1}$ @8 $\text{A g}^{-1}$	s6
6	CuS	237 $\text{F g}^{-1}$ @0.5 $\text{A g}^{-1}$	s7
7	PPy/ $\text{MoO}_3$	687 $\text{F g}^{-1}$ @1 $\text{A g}^{-1}$	s8
8	$\text{MoO}_3$ /C	331 $\text{F g}^{-1}$ @1 $\text{A g}^{-1}$	s9

9	MoO <sub>3</sub>	603 F g <sup>-1</sup> @1 A g <sup>-1</sup>	s10
10	MoS <sub>2</sub>	366 F g <sup>-1</sup> @0.5 A g <sup>-1</sup>	s11
11	MoS <sub>2</sub>	231 F g <sup>-1</sup> @1 A g <sup>-1</sup>	s12
12	MoS <sub>2</sub> -rGO	387 F g <sup>-1</sup> @1.2 A g <sup>-1</sup>	s13
<b>13</b>	<b>Cu<sub>2</sub>MoS<sub>4</sub></b>	<b>1055 F g<sup>-1</sup>@1 A g<sup>-1</sup></b>	<b>This work</b>



**Fig. S3**  $\log i$  vs.  $\log v$  plot of  $\text{Cu}_2\text{MoS}_4$ , when studied as the negative electrode material.



**Fig. S4** Current vs. potential plots of  $\text{Cu}_2\text{MoS}_4||\text{Cu}_2\text{MoS}_4$  ASSSP device at different potential windows.

**Table S3.** Comparison of the charge transfer resistance of  $\text{Cu}_2\text{MoS}_4$  with the reported electrode materials (for symmetric supercapacitors) in 3 electrode set up; comparison of the energy density, power density and cyclic stability of the  $\text{Cu}_2\text{MoS}_4 || \text{Cu}_2\text{MoS}_4$  ASSSP device with the reported symmetric supercapacitor devices.

SI. No.	Symmetric device	Charge transfer resistance in 3 electrode set up ( $R_{ct}$ )	Energy density ( $Wh\ kg^{-1}$ )	Power density ( $W\ kg^{-1}$ )	Retention (%) in Cyclic Stability (No. of cycles)	References
1	$\text{Fe}_3\text{O}_4/\text{Graphene}    \text{Fe}_3\text{O}_4/\text{Graphene}$	–	19.2	800.2	94% (3200)	s14
2	$\text{MnO}_2    \text{MnO}_2$	1.2 $\Omega$	12.7	87	83% (3000)	s15
3	$\text{N, S co-doped PCFF}    \text{N, S co-doped PCFF}$	–	16.3	147	79% (10000)	s16
4	$\text{NiS}_2    \text{NiS}_2$	–	7.97	500	90% (1500)	s17

5	MoS <sub>2</sub> /CC    MoS <sub>2</sub> /CC	4.3 Ω	5.42	128	96.5% (5000)	s18
6	MnO <sub>2</sub>    MnO <sub>2</sub>	31.42 Ω	23	1923	92% (2200)	s19
7	MoS <sub>2</sub> /CC    MoS <sub>2</sub> /CC	0.51 Ω	11.1	250	87.9% (1000)	s20
8	NiCo <sub>2</sub> O <sub>4</sub>    NiCo <sub>2</sub> O <sub>4</sub>	–	30.50	750	86% (500)	s21
9	CMS/Ni    CMS/Ni	0.48 Ω	23.61	1000	87.7% (3000)	s22
10	CuCo <sub>2</sub> O <sub>4</sub>    CuCo <sub>2</sub> O <sub>4</sub>	1.89 Ω	16.87	8200	100.94 (3000)	s23
11	FeS    FeS	5.15 Ω	2.56	726	91% (1000)	s24
12	f-CFP    f-CFP	–	5.2	326	99% (5000)	s25
13	PGBC    PGBC	0.51 Ω	6.68	100.2	84% (5000)	s26
14	HCP    HCP	–	9.67	-	90.2% (10000)	s27
15	NCF    NCF	–	1.35	2900	95.8% (1000)	s28
16	rGO-PMo <sub>12</sub>    rGO-PMo <sub>12</sub>	–	17.20	130	89% (5000)	s29
17	PANI@CNFs    PANI@CNFs	10.5 Ω	10.04	225	81% (1000)	s30
18	HN-CNFs/GNs    HN-CNFs/GNs	–	6.3	344.1	99% (5000)	s31
19	CuS/MnS@NF    CuS/MnS@NF	–	2.54	174.7	78% (2500)	s32
20	Cu <sub>2</sub> MoS <sub>4</sub>    Cu <sub>2</sub> MoS <sub>4</sub>	0.45 Ω	30.53	5649	97.1 (10,500)	This work

**Table S4.** Comparison of the relaxation time of the  $\text{Cu}_2\text{MoS}_4 \parallel \text{Cu}_2\text{MoS}_4$  ASSSPC device with the reported symmetric/asymmetric supercapacitor devices.

Sl. No.	Electrode Material	Symmetric/Asymmetric supercapacitor device	Relaxation time ( $\tau_0$ )	Reference No.
1	Carbon/Carbon supercapacitors	Carbon $\parallel$ Carbon	4.9 seconds	s33
2	$\text{CoFe}_2\text{O}_4$ thin film	$\text{CoFe}_2\text{O}_4 \parallel \text{CoFe}_2\text{O}_4$	174 milliseconds	s34
3	Ti-rich $\text{TiO}_2$ tubular nanolettuces	Ti-rich $\text{TiO}_2 \parallel \text{Ti-rich TiO}_2$	1.7 seconds	s35
4	3D cross-linked graphene	NPFG-0.3 $\parallel$ NPFG-0.3	28.4 milliseconds	s36
5	Manganese oxide	$\text{MnO}_2 \parallel \text{NiCo}_2\text{O}_4$	14 milliseconds	s37
6	$\alpha\text{-Fe}_2\text{O}_3$ thin film	$\alpha\text{-Fe}_2\text{O}_3 \parallel \alpha\text{-Fe}_2\text{O}_3$	9 milliseconds	s38
7	$\text{V}_2\text{O}_5$ encapsulated MWCNTs	$\text{V}_2\text{O}_5/\text{MWCNT} \parallel \text{V}_2\text{O}_5/\text{MWCNT}$	5.6 milliseconds	s39
8	$\text{MnO}_x/\text{Au}$ multilayers	$\text{MnO}_x/\text{Au} \parallel \text{MnO}_x/\text{Au}$	5 milliseconds	s40
9	Silicon nanowires (SiNWs)	SiNWs $\parallel$ SiNWs	3.5 milliseconds	s41
10	Graphene on metal template	Metal graphene $\parallel$ Metal graphene	1.8 milliseconds	s42
11	Polymer-derived carbyne	Carbyne $\parallel$ Carbyne	1.2 milliseconds	s43
12	Graphene	Printable graphene $\parallel$ Printable graphene	1 millisecond	s44
13	$\text{Cu}_2\text{MoS}_4$	$\text{Cu}_2\text{MoS}_4 \parallel \text{Cu}_2\text{MoS}_4$	0.5 millisecond	This work

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