Parameter/ Sample	Unit	2MgA1	20K <sub>2</sub> O/2MgAl			
SiO <sub>2</sub>	wt.%	< 0.1	0.3			
$Al_2O_3$	wt.%	38.3	28.1			
MgO	wt.%	59.7	43.9			
K <sub>2</sub> O	wt.%	0.1	24.7			
Other impurities <sup>a</sup>	wt.%	1.9	3.0			
Mg/Al	mol/mol	2.0	2.0			

Table S1. Elemental Composition

<sup>a</sup> other impurities including Cl, SO<sub>3</sub>, P<sub>2</sub>O<sub>5</sub>, CaO, Na<sub>2</sub>O, MnO, CuO, ZnO, Fe<sub>2</sub>O<sub>3</sub>, Ga<sub>2</sub>O<sub>3</sub>, NiO

## Table S2. Physicochemical properties of Hydrotalcite (HTC) catalysts

		Base and Acid site loading (mmol g <sup>-1</sup> )									
	Catalyst	T <sub>max</sub>	Basic site loading <sup>a</sup>	T <sub>max</sub>	Acid site loading <sup>b</sup>	${{S_{BET}}^{c}}^{c}$ $(m^{2}$ $g^{-1})$	${S_{ m MIC}}^{ m d}$ $(m^2$ $g^{-1})$	${\mathop{S_{\rm EXT}}\limits^{{\rm e}}}^{{\rm e}}_{{ m (m}^2}$	$V_p^{f}$ (cm <sup>3</sup> g <sup>-1</sup> )	$V_{mic}^{d}$ (cm <sup>3</sup> g <sup>-1</sup> )	V <sub>mes</sub> <sup>g</sup> (cm <sup>3</sup> g <sup>-1</sup> )
		(°C)	(mmol g <sup>-1</sup> )	(°C)	(mmol g <sup>-1</sup> )						
2	20K <sub>2</sub> O/2MgAl	100, 404, 642	0.498	300	1.200	34	0	4	0.45	0	0.45
	2MgAl	140-230, 648	0.566	400, 536	3.100	54	0	35	0.57	0	0.57

<sup>a</sup> from CO<sub>2</sub>-TPD; <sup>b</sup> from propylamine-TGA-MS; <sup>c</sup> S<sub>BET</sub> total surface area determined using BET equation with  $P/P_0 < 0.05$ ; <sup>d</sup> S<sub>MIC</sub> micropore surface area and V<sub>MIC</sub> micropore volume determined from t-plot analysis <sup>e</sup> S<sub>EXT</sub> external surface area was calculated from the difference between S<sub>BET</sub>-S<sub>MIC</sub>, <sup>f</sup> V<sub>p</sub> total pore volume obtained from isotherm at P/P<sub>0</sub>=0.99 (Error = ±10%) and <sup>g</sup> V<sub>MES</sub> mesopore volume obtained from the difference between V<sub>p</sub>-V<sub>MIC</sub>.



Fig S1. Wide-angle XRD of parent and modified 2MgAl Hydrotalcite



Fig S2. Adsorption–desorption isotherms of  $N_2$  molecules at 77 K and pore size distribution (inset) of a) SBA-15, b) 12MgAl-SBA-15, c)  $8K_2O/12MgAl-SBA-15$ , d)  $20K_2O/12MgAl-SBA-15$ , (e) 2MgAl and (f)  $2K_2O/2MgAl$ 



Fig S3. (a) CO<sub>2</sub>-TPD and (b) TGA-MS-propylamine of parent and modified 2MgAl





Fig. S4 (a) Comparison of furfural conversion for 2MgAl and 20K<sub>2</sub>O/2MgAl and (b and c) product selectivity as a function of furfural conversion achieved over 24 h for 2MgAl and 20K<sub>2</sub>O/2MgAl catalysts



Fig. S5 TGA-DTG measurement of regenerated 20K\_2O/12MgAl-SBA-15 by thermal treatment under an  $\rm N_2$