

Supporting Information

An Efficient Particulate Photocatalyst for Overall Water Splitting Based on Scandium and Magnesium Co-Doped Strontium Titanate

Shigeru Ikeda,^{*a,b} Riku Okamoto,^a Akira Kimura,^a Yuhi Nakayasu,^c Akira Yamakata,^c
Ryota Tomizawa,^d Taizo Masuda,^d Koichiro Nakatani^d

^a Department of Chemistry, Konan University, 9-1 Okamoto, Higashinada-ku, Kobe, Hyogo 658-8501, Japan.

^b Institute for Energy Conversion Materials, Konan University, 9-1 Okamoto, Higashinada-ku, Kobe, Hyogo 658-8501, Japan.

^c Graduate School of Natural Science & Technology, Okayama University, 3-1-1 Tsushima-naka, Kita-ku, Okayama 700-8530 Japan.

^d Carbon Neutral Development Division, Toyota Motor Corporation, 1200 Mishuku, Susono, Shizuoka 410-1193, Japan.

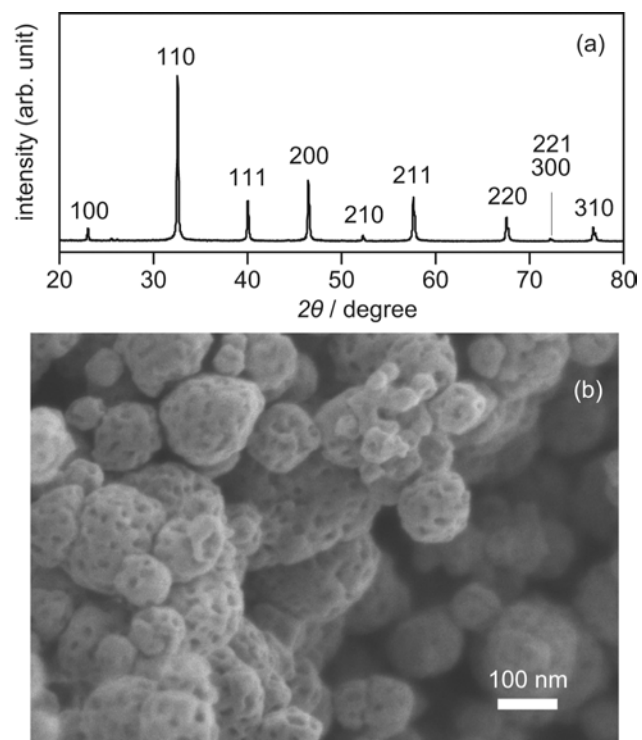


Fig. S1 (a) an XRD pattern and an SEM image of the commercial SrTiO₃ powder supplied from FUJIFILM Wako.

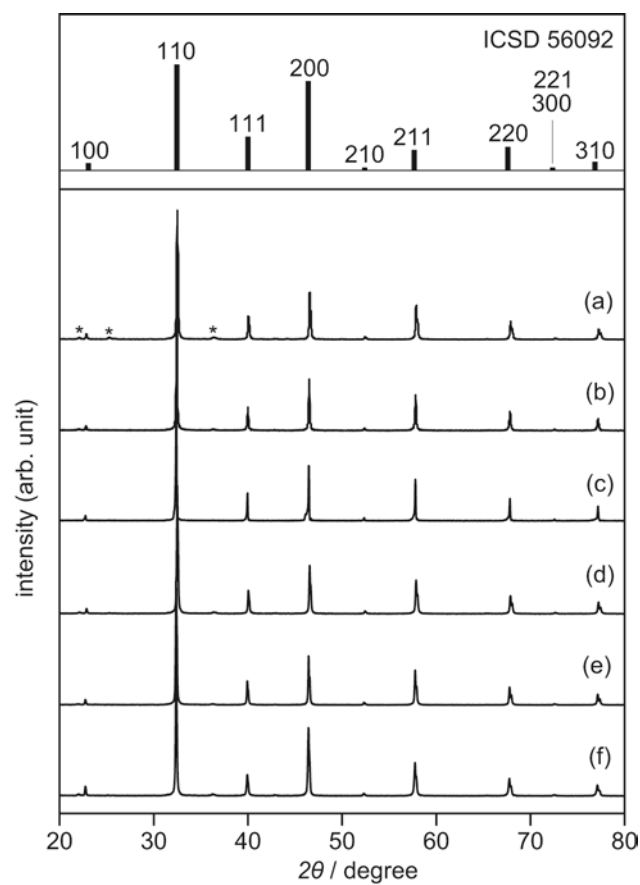


Fig. S2 XRD patterns of (a) SrTiO₃:Al, (b) SrTiO₃:Sc, (c) SrTiO₃:Mg, (d) SrTiO₃:Al,Sc, (e) SrTiO₃:Al,Mg, and (f) SrTiO₃:Sc,Mg samples. An XRD profile of SrTiO₃ (ICSD 56092) obtained by VESTA software¹ was also shown as a reference. Asterisks (*) denote reflections of SrCO₃ derived from the SrCl₂ flux.

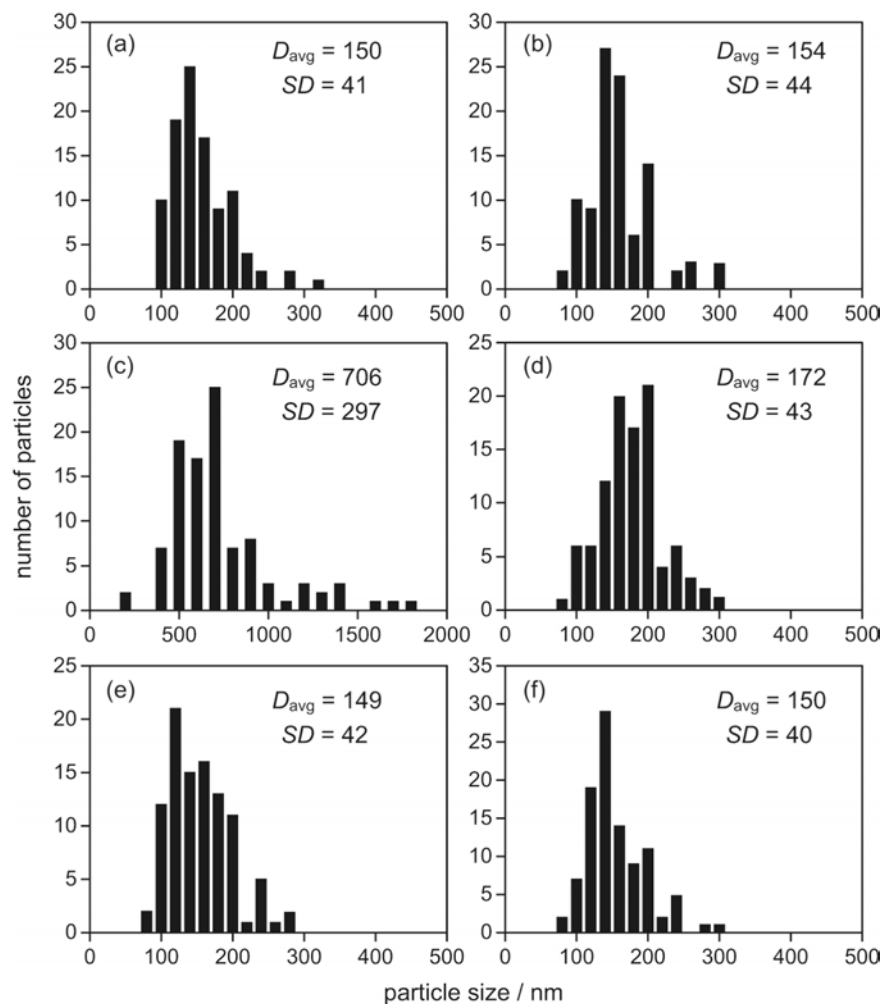


Fig. S3 Particle-size distributions of (a) SrTiO₃:Al, (b) SrTiO₃:Sc, (c) SrTiO₃:Mg, (d) SrTiO₃:Al,Sc, (e) SrTiO₃:Al,Mg, and (f) SrTiO₃:Sc,Mg samples. Insets of these plots indicate average diameters (D_{avg} s) and their standard deviations (SD s) in a nanometer unit.

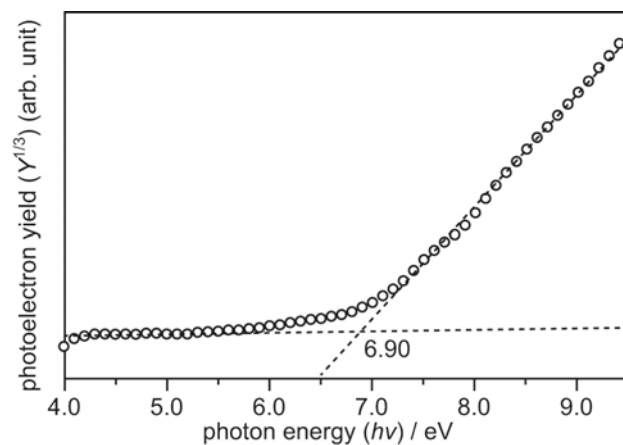


Fig. S4 A photoemission yield ($Y^{1/3}$)-energy ($h\nu$) plot of the $\text{SrTiO}_3\text{:Sc,Mg}$ sample.

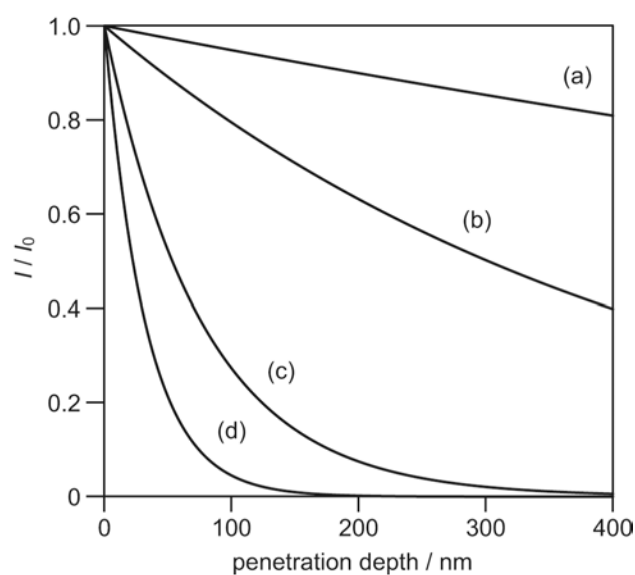


Fig. S5 Transmittance depth profiles of an SrTiO_3 bulk crystal at various photon energies obtained from photoabsorption coefficients (α) reported by Zollner et al.² (a) 3.2 eV (365 nm), (b) 3.4 eV (344 nm), (c) 3.8 eV (326 nm), and (d) 4.0 eV (310 nm).

Table S1 Ratios of intensities of the photoabsorption at 10,000 cm⁻¹ measured in O₂ and H₂O after a 100-μs delay relative to that measured in N₂

Sample	$I_{O_2}/I_{N_2}^{a)}$	$I_{O_2}/I_{H_2O}^{b)}$
SrTiO ₃ :Sc,Mg	0.76	1.76
SrTiO ₃ :Al	0.81	1.36

^{a)} Ratio of the photoabsorption intensity at 10,000 cm⁻¹ measured in O₂ after the 100 μs delay relative to that measured in N₂.

^{b)} Ratio of the photoabsorption intensity at 10,000 cm⁻¹ measured in H₂O after the 100 μs delay relative to that measured in N₂.

References

- 1 K. Momma, F. Izumi, *J. Appl. Crystallogr.* 2011, **44**, 1272.
- 2 S. Zollner, A. A. Demkov, R. Liu, P. L. Fejes, R. B. Gregory, P. Alluri, *J. Vac. Sci. Technol. B* 2000, **18**, 2242.