

Z-scheme water splitting utilizing $\text{CuLi}_{1/3}\text{Ti}_{2/3}\text{O}_2$ as a hydrogen-evolving photocatalyst with photo-response up to 600 nm

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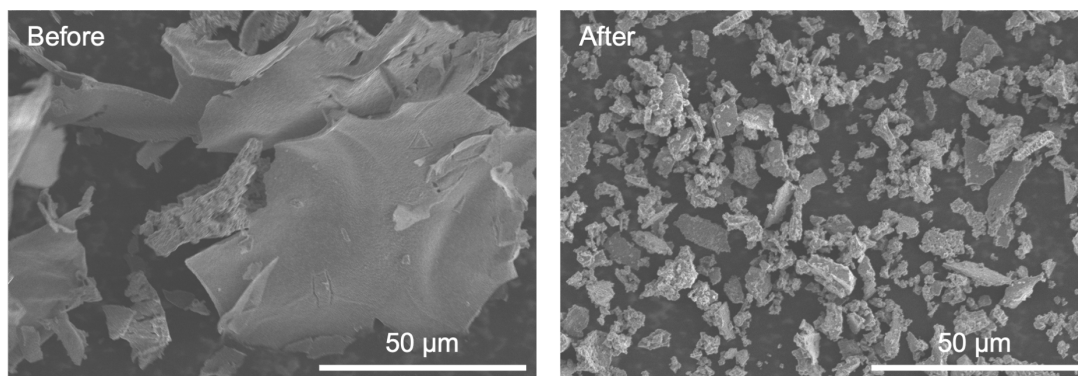


Fig. S1 SEM images of $\text{Li}_4\text{Ti}_5\text{O}_{12}$ before and after the pot-mill treatment.

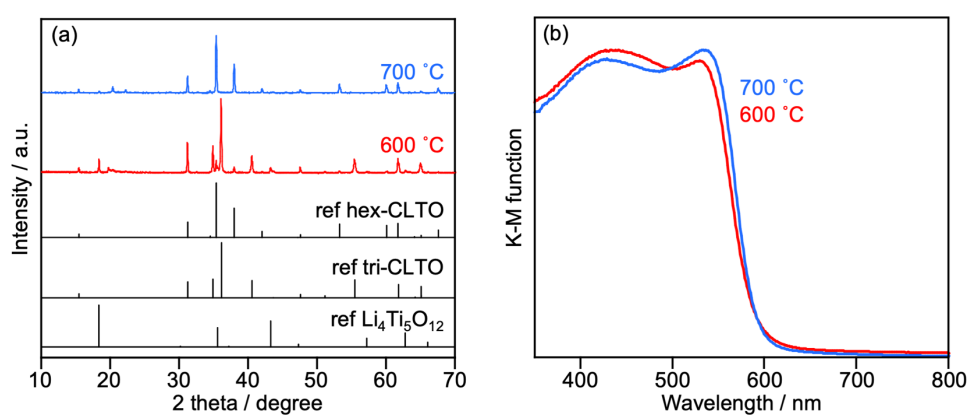


Fig. S2 (a) XRD patterns and (b) UV-vis spectra of CLTO prepared at 600 and 700 °C.

Reference patterns of hexagonal and trigonal CLTO were simulated from structure models of CuGaO_2 modified with composition and lattice constants of CLTO.

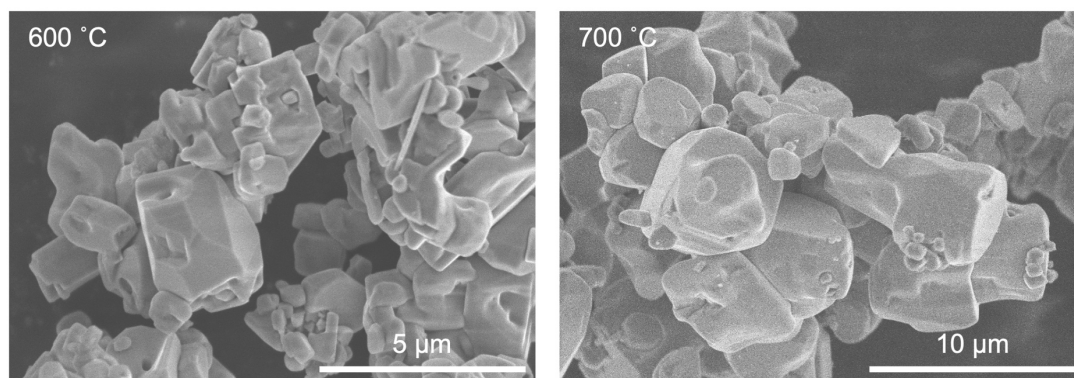


Fig. S3 SEM images of CLTO prepared at 600 and 700 °C.

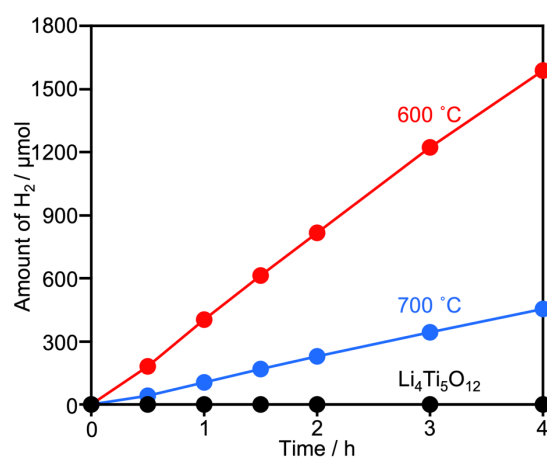


Fig. S4 Sacrificial H₂ evolution on Ru(0.5 wt%)/CLTO and Li₄Ti₅O₁₂ from an aqueous solution containing 0.1 M Na₂S and 0.5 M Na₂SO₃. Photocatalyst: 0.3 g; light source: 300 W Xe lamp ($\lambda > 420$ nm).

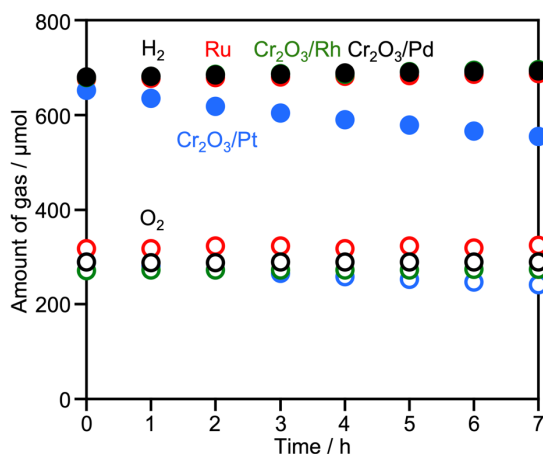


Fig. S5 Water formation over Ru/CLTO and Cr₂O₃/M/CLTO (M: Rh, Pd and Pt) in the dark. Catalyst: 0.05 g; H₂: 10 kPa; O₂: 5 kPa.

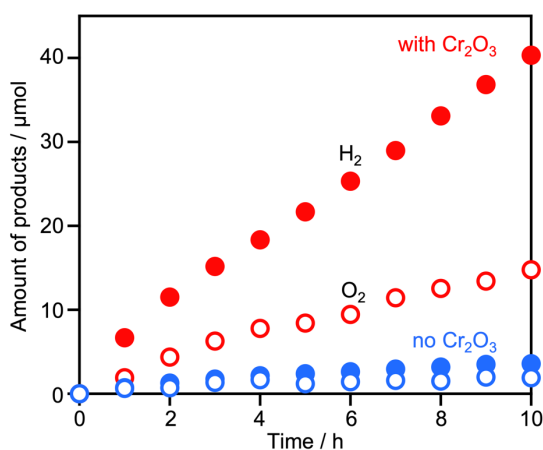


Fig. S6 Influences of Cr₂O₃ modification of Ru/CLTO upon Z-WS by Ru/CLTO-BiVO₄-Co(bpy)₃SO₄ system. Photocatalyst: 0.05 g each; reactant solution: 0.1 mM Co(bpy)₃SO₄, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm).

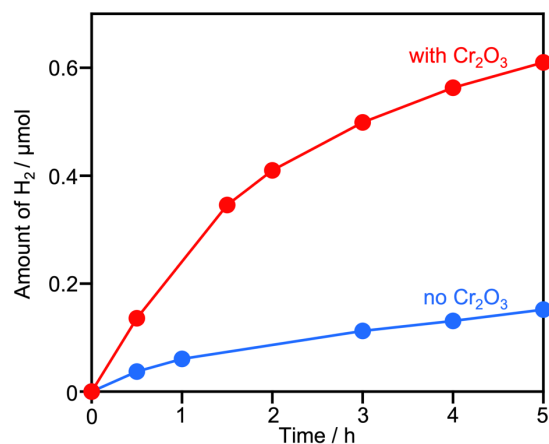


Fig. S7 Influences of Cr₂O₃ modification of Ru/CLTO upon H₂ evolution from a Co(bpy)₃SO₄ solution. Photocatalyst: 0.1 g; reactant solution: 0.5 mM Co(bpy)₃SO₄, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm).

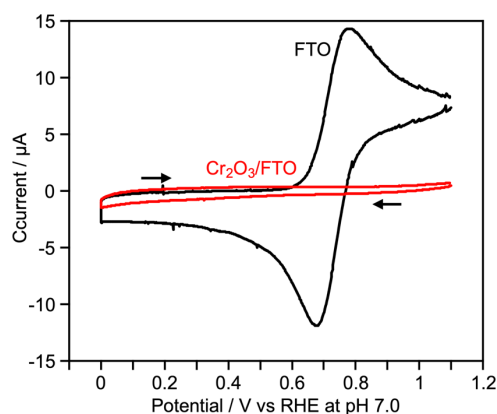


Fig S8. Current–potential curves of FTO and Cr₂O₃/FTO electrodes measured in a solution containing 0.1 M K₂SO₄ and 0.1 mM Co(bpy)₃SO₄ under Ar.

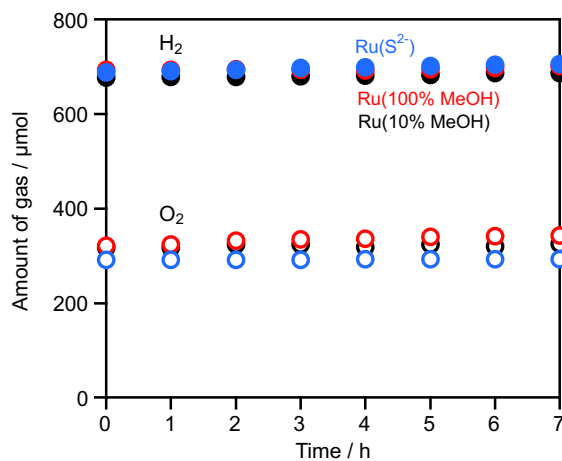


Fig. S9 Water formation over Ru/CLTO in the dark. Catalyst: 0.05 g; H₂: 10 kPa; O₂: 5 kPa.

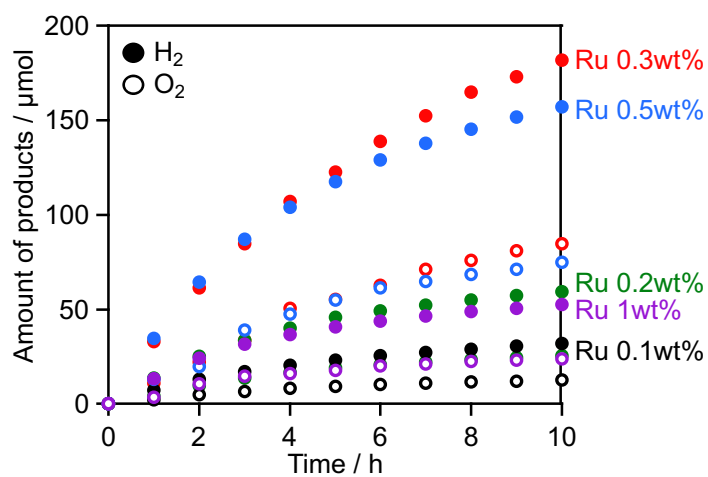


Fig. S10 Influences of Ru amounts upon Z-WS by the Cr₂O₃/Ru(100% MeOH)/CLTO-BiVO₄-Co(bpy)₃SO₄ system. Photocatalyst: 0.05 g each; reactant solution: 0.1 mM Co(bpy)₃SO₄, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm).

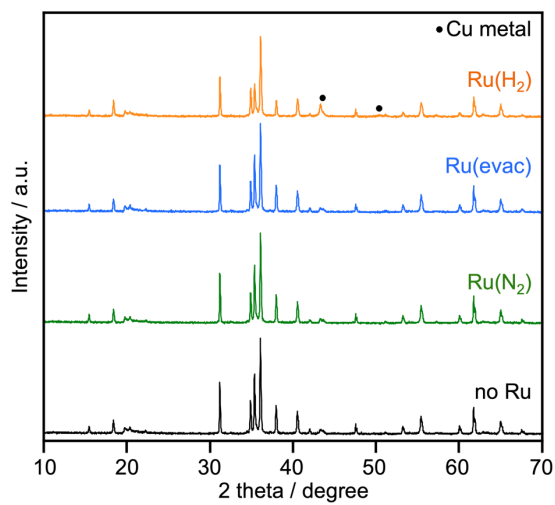


Fig. S11 XRD patterns of Ru/CLTO prepared by an impregnation method with different heat-treatment conditions.

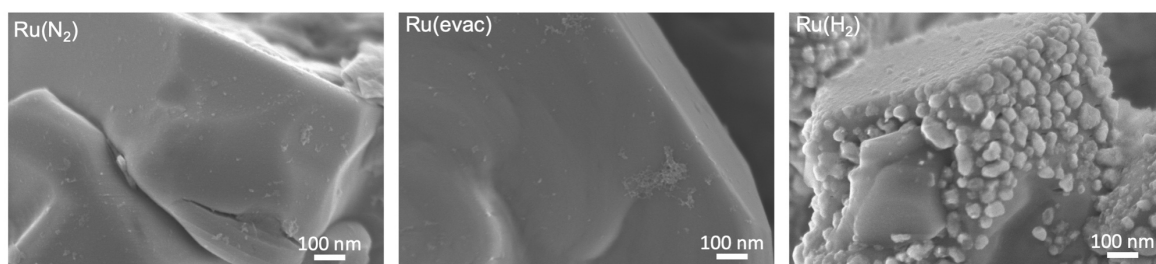


Fig. S12 SEM images of Ru/CLTO prepared by the impregnation method with different heat-treatment conditions.

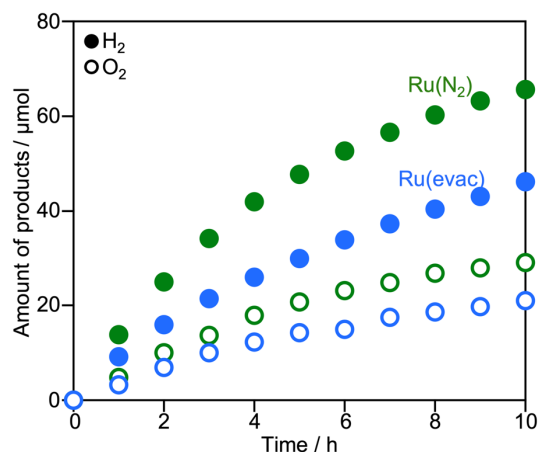


Fig. S13 Photocatalytic Z-WS using $\text{Cr}_2\text{O}_3/\text{Ru}/\text{CLTO}$ prepared via the impregnation method with different heat treatment conditions. Photocatalyst: 0.05 g each; reactant solution: 0.1 mM $\text{Co}(\text{bpy})_3\text{SO}_4$, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm).

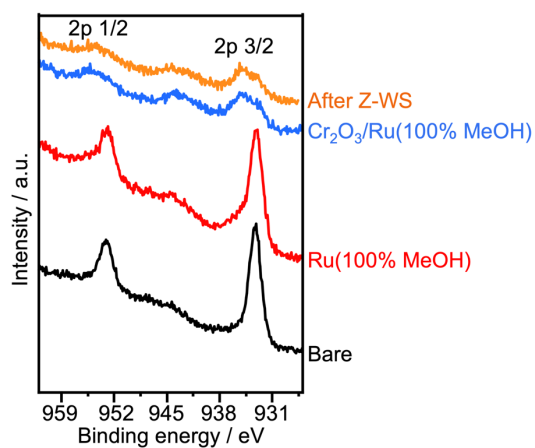


Fig. S14 XPS at Cu 2p of CLTO samples after photoirradiation including Z-WS.

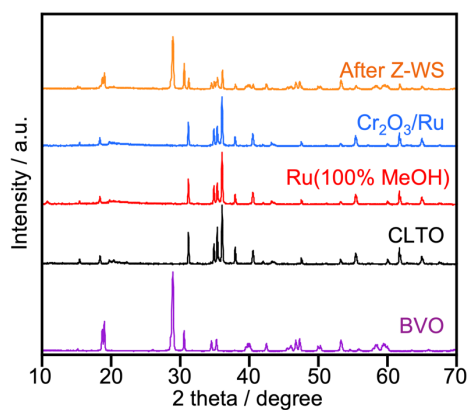


Fig. S15 XRD patterns of CLTO samples after photoirradiation including Z-WS.

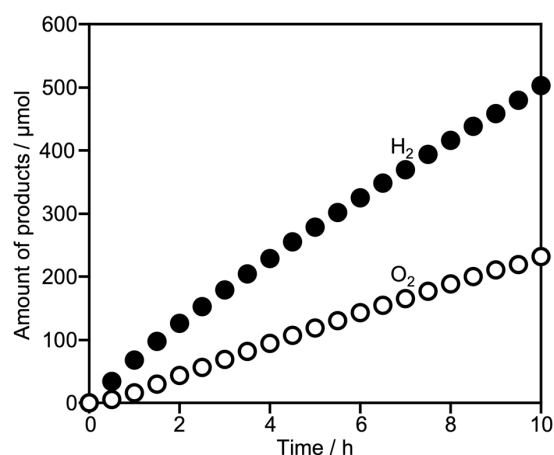


Fig. S16 Z-WS using the Ru/SrTiO₃:Rh-BiVO₄-Co(bpy)₃SO₄ system. Photocatalyst: 0.05 g each; reactant solution: 0.5 mM Co(bpy)₃SO₄, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm).

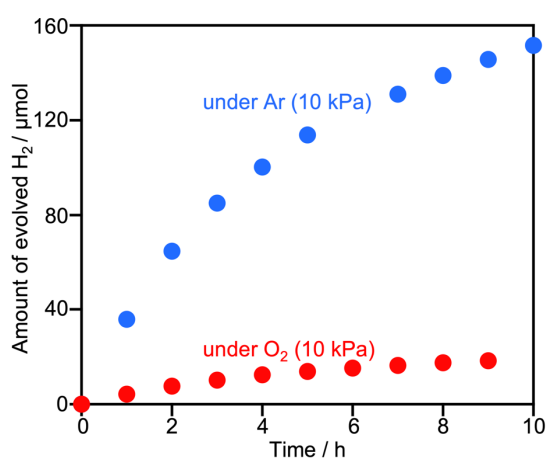


Fig. S17 Influence of O₂ upon H₂ evolution caused by water splitting using the Cr₂O₃/Ru/CLTO-BiVO₄-Co(bpy)₃SO₄ system. Photocatalyst: 0.05 g each; reactant solution: 0.1 mM Co(bpy)₃SO₄, 160 mL; light source: 300 W Xe lamp ($\lambda > 420$ nm). Only H₂ evolution is displayed because O₂ evolution was not detectable in the reaction under O₂ due to the quite large basic signal.